

LETTERS TO THE EDITOR

Aminolysis of 2-Amino-5-benzylidene-1,3-thiazol-4(5*H*)-one in Cyclic Amines: Recyclization to 2-Amino-5-benzylidene-1,5-dihydro-4*H*-imidazol-4-one Derivatives

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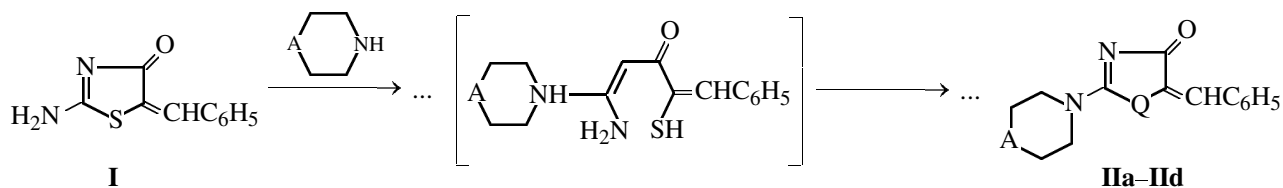
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It is known [1] that the amino group in 2-amino-5-benzylidene-1,3-thiazol-4(5*H*)-one (5-benzylidene-pseudothiohydantoin, **I**) is replaced by phenylamino group by the action of aniline. We have found that aminolysis of compound **I** in piperidine or morpholine involves not only expected transamination to give

2-piperidino (**IIa**) or 2-morpholino derivative (**IIb**) of 5-benzylidene-1,3-thiazol-4(5*H*)-one but also replacement of the endocyclic sulfur atom by nitrogen with formation of 2-piperidino- or 2-morpholino-5-benzylidene-1,5-dihydro-4*H*-imidazol-4-one **IIc** or **IId**, respectively.



Q = S (**a**, **b**), NH (**c**, **d**); A = CH₂ (**a**, **c**), O (**b**, **d**).

The formation of compounds **IIa** and **IIb** is likely to occur without opening of the thiazole ring [2], while dihydroimidazole derivatives **IIc** and **IId** can be formed only as a result of recyclization of open-chain intermediate arising from opening of the thiazole ring at the C²–S or C⁵–S bond in a way similar to the reaction of **I** with hydroxide ion [3] or reactions of amines with 4,5-dihydrothiazol-2-amine [4, 5]. The structure of the alkaline hydrolysis [6], alcoholysis [2], and aminolysis products [7] of 2-amino-1,3-thiazol-4(5*H*)-one derivatives suggests that these reactions involve cleavage of the C²–S bond.

2-Amino-5-benzylidene-1,3-thiazol-4(5*H*)-one (**I**) was synthesized by the procedure described in [8]. Aminolysis of 2-amino-5-benzylidene-1,3-thiazol-4(5*H*)-one (**I**) in piperidine. A mixture of 1.00 g of compound **I** and 9.57 g of piperidine was heated

under reflux until it became homogeneous (about 10 min) and was then left to stand for 24 h at room temperature. The precipitate was filtered off, washed with petroleum ether–ethanol (10:1), and recrystallized from toluene to isolate 0.250 g (20%) of **5-benzylidene-2-piperidino-1,5-dihydro-4*H*-imidazol-4-one** (**IIc**) with mp 208–210°C (published data [9]: mp 198°C for the *Z* isomer). The filtrate was combined with the washings, and a solid separated from the solution in several days. The precipitate was filtered off and recrystallized first from toluene and then from ethanol. We thus isolated 0.253 g (19%) of **5-benzylidene-2-piperidino-1,3-thiazol-4(5*H*)-one** (**IIa**) with mp 202–204°C; published data: mp 202–204°C [10], 209–211°C [11].

Aminolysis of 2-amino-5-benzylidene-1,3-thiazol-4(5*H*)-one (I**) in morpholine.** A mixture of

2.00 g of compound **I** and 9.20 g of morpholine was heated under reflux until it became homogeneous (about 5 min) and was left to stand overnight at room temperature. The precipitate was filtered off, washed with petroleum etherethanol (1:1), and treated in succession with boiling acetone and propan-2-ol. After 2 days, a solid separated from the solution in propan-2-ol; it was filtered off and recrystallized from propan-2-ol. Yield of **5-benzilydene-2-morpholino-1,5-dihydro-4H-imidazol-4-one (IIc)** 0.501 g (19.9%), mp 220–221°C; published data [9]: mp 232°C, *Z* isomer). The filtrate was combined with the washings; after 30 min, the precipitate was filtered off and recrystallized twice from toluene. Yield of **5-benzylidene-2-morpholino-1,3-thiazol-4(5H)-one (IIb)** 0.284 g (10.6%), mp 202–204°C; published data: mp 202–204°C [10], 203–204°C [11].

The purity of compounds **IIa–IIc** was checked by TLC and elemental analysis, and their structure was confirmed by the IR and ¹H NMR spectra. The *m/z* values for the molecular ions in the mass spectra of **IIc** and **IIc** coincided with the calculated values.

REFERENCES

1. Culvenor, C.C.J., Davies, W., Maclaren, J.A., Nelson, P.F., and Savige, W.E., *J. Chem. Soc.*, 1949, p. 2573.
2. Omar, M.T. and Kasem, M.A., *J. Heterocyclic Chem.*, 1981, vol. 18, no. 7, p. 1413.
3. Fedoseev, V.M., Mandrugina, A.A., and Semenenko, M.N., *Khim. Geterotsikl. Soedin.*, 1984, no. 1, p. 44.
4. Rodyunin, A.A., Mandrugina, A.A., and Fedoseev, V.M., Abstracts of Papers, *I Vsesoyuznaya konferentsiya po teoreticheskoi organicheskoi khimii* (I All-Union Conf. on Theoretical Organic Chemistry), Volgograd, 1991, p. 159.
5. Mandrugina, A.A., Fedoseev, V.M., Khomutov, S.M., and Roshchina, T.M., Abstracts of Papers, *I Vsesoyuznaya konferentsiya po teoreticheskoi organicheskoi khimii* (I All-Union Conf. on Theoretical Organic Chemistry), Volgograd, 1991, p. 158.
6. Ramsh, S.M., Ivanenko, A.G., Shpilevii, V.A., Medvedskii, N.L., and Kushakova, P.M., *Khim. Geterotsikl. Soedin.*, 2005, no. 7, p. 1089.
7. El'tsov, O.S., Mokrushin, V.S., Bel'skaya, N.P., and Kozlova, N.M., *Izv. Ross. Akad. Nauk, Ser. Khim.*, 2003, no. 2, p. 440.
8. Liberman, D., Himbert, J., and Hengl, L., *Bull. Soc. Chim. Fr.*, 1948, p. 1120; *Chem. Abstr.*, 1949, vol. 43, p. 3819.
9. Khodair, A.I., El-Subbagh, H.I., and Al-Obaid, A.M., *Phosph., Sulfur Silicon*, 1998, vol. 140, p. 159.
10. Raouf, A.R.A., Omar, M.T., Omran, S.M.A., and El-Bayoumy, K.E., *Acta Chim. (Budapest)*, 1974, vol. 83, nos. 3–4, p. 359.
11. Kutschy, P., Dzurilla, M., Kristian, P., and Kutschyova, K., *Coll. Czech. Chem. Commun.*, 1981, vol. 46, no. 2, p. 436.