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Stereoselective Palladium-Catalyzed 1,3-Arylboration of Unconjugated Dienes for Expedient Synthesis of 1,3-Disubstituted Cyclohexanes

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ABSTRACT: As a significant pharmacophore, 1,3-disubstituted cyclohexanes are widespread in natural products and synthetic bioactive molecules. A palladium-catalyzed arylboration of 1,4-cyclohexadienes is reported, which allows expeditious access to an array of functionalized 1,3-disubstituted cyclohexanes from the readily available starting materials. Palladium catalysis enables the arylboration to proceed in a reversed regioselectivity compared with earlier nickel catalysis. The most striking feature of this protocol lies in the 1,3-regioselectivity and exclusive cis-diastereoselectivity. Intriguingly, the success of this three-component reaction does not rely on the application of dative ligands, but a cheap ammonium chloride salt instead. The synthetic utility of this method is highlighted by a series of downstream stereospecific transformations and a drug molecule synthesis.

KEYWORDS: Alkenes, 1,3-Arylboration, Palladium, Metal migration, Stereoselectivity

Over the past few decades, fragment-based drug discovery (FBDD) has developed into an important strategy to identify the lead pharmaceutical compounds.¹ However, compared with the sp²-rich, planar compounds, three-dimensional (3D) fragments are highly limited in candidate libraries.^{1a} In this context, the 1,3-disubstituted cyclohexane fragment is widely investigated as a privileged scaffold in many significant, pharmaceutically relevant molecules, such as anti-diabetic agent, cannabinoid receptors, MAO-B inhibitor and so on, as depicted in Figure 1A.² However, general synthetic approach towards this class of compounds is quite scarce.2a, 2d, 3 Therefore, the development of a general, efficient and stereoselective method for direct synthesis of 1,3-disubstituted cyclohexanes from readily available starting materials is a highly desirable yet challenging task. On the other hand, the chemical dearomatization of nonactivated arenes provides a powerful platform for rapid assembly of value-added, complex molecules from inexpensive fundamental aromatic compounds, which has attracted extensive interest in the synthetic community in the past few years.⁴ For example, a great advance in the transformation of simple arenes into cyclohexane derivatives has been achieved recently by the Sarlah group through a sequence of photochemical arene-arenophile cycloaddition and metal catalysis (Figure 1B, left).⁵ With this method, a set of 1,2- and 1,4-disubstituted cyclohexanes were selectively prepared.^{5b,d-f} Herein, we present a protocol which can convert benzene into a diverse range of functionalized 1,3-disubstituted cyclohexanes by combination of the well-known Birch reduction⁶ and an unprecedented arylboration7 (Figure 1B, right). Remarkably, the palladium catalysis allows the arylboration to proceed in an extraordinarily 1,3-regioselective and cis-diastereoselective manner.⁸⁻⁹ Furthermore, the incorporation of a boron group into the products enables further stereospecific transformations to afford a variety of highly functionalized 3D molecules.¹⁰

A) Representive Bioactive Molecules Containing the Fragment of 1,3-Disubstituted Cyclohexane





As an extension of our interest in the application of nickel migration,¹¹⁻¹²we commenced this study by exploring a nickelcatalyzed arylboration of 1,4-cyclohexadienes, with 1,4cyclohexadiene bromobenzene (1a). (2a)and bis(pinacolato)diboron (B_2pin_2 3) as model substrates. As depicted in Table 1, the regio- and diastereo-selectivity would be the main challenge of this arylboration reaction. For example, it was found that with a nickel(II) salt and a 1,10phenanthroline derivative L1 as the catalyst, only the 3,1arylboration product 4a was isolated, albeit in lower yield with a poor diastereoisomeric ratio (1:1) (entry 1). Without the ligand, the 1,2-arylboration product *cis*-**5a** was isolated as a ACS Paragon Plus Environment single isomer in this three-component reaction (Table 1, entry 2), which is consistent with Brown's ligand-free system.⁹ⁱ These results reveal that the epimerization should derive from nickel migration. After extensively screening reaction conditions, the diastereoselectivity could not be improved, therefore we turned our attentions to other transition metal catalysts.

Table 1. Reaction Optimization

	chemo- and diastereo-selectivity		
$\frac{\operatorname{ArX}(2)}{1}$	Bpin 3,1-product (4)	1,2-product (5)	1,3-product (6)
Der	ivation from con	ditions A	Results
	no change no ligand	30%(1:1) ^a 4a 33% cis-5a	
Derivation from conditions B			<i>cis-</i> 6a , [%]
Pd((NiBr ₂ T T Na Na D	no change Pd(OAc) ₂ instead of Pd(acac) ₂ NiBr ₂ ·DME instead of Pd(acac) ₂ TBAC instead of TMAC TMAB instead of TMAC TMAI instead of TMAC NaHCO ₃ instead of NaOAc Na ₂ CO ₃ instead of NaOAc DCE instead of CHCl ₃ Dioxane instead of CHCl ₃ no TMAC no NaOAc		$ \begin{array}{r} 82(78)^{\circ} \\ 57 \\ 0 \\ 23 \\ 20 \\ 6 \\ 39 \\ 56 \\ 45 \\ 52 \\ 5 \\ 2 \end{array} $
	ArX (2) B ₂ pin ₂ (3) [M] Der Pd((C NiBr ₂ T T T Na Na Na I Di	ArX (2) B ₂ pin ₂ (3) [M] Bpin (ArX (2) B ₂ pin ₂ (3) [M] Derivation from conditions A no change no ligand Derivation from conditions B no change Pd(OAc) ₂ instead of Pd(acac) ₂ NiBr ₂ ·DME instead of Pd(acac) ₂ NiBr ₂ ·DME instead of TMAC TMAB instead of NaOAc Na ₂ CO ₃ instead of NaOAc Na ₂ CO ₃ instead of CHCl ₃ Dioxane instead of CHCl ₃ no TMAC no TMAC no TMAC no TMAC

Conditions **A**: NiBr₂·DME (5 mol %), **L1** (5 mol %), **1a** (1.0 mmol), **2a** (0.5 mmol), B₂pin₂ (**3**, 0.75 mmol), LiOMe (1.0 mmol), in DMF, 30 °C, 24 h. Isolated yield. ^{*a*} The ratio in parenthese is *cis-/trans*-**4a**. Conditions **B**: Pd(acac)₂ (5 mol %), **1a** (0.4 mmol), **2b** (0.2 mmol), B₂pin₂ (**3**, 0.2 mmol), TMAC (0.2 mmol), NaOAc (0.4 mmol), in anhydrous CHCl₃, 60 °C, 24 h; GC yields against naphthalene. ^{*b*} **1a** (0.6 mmol), isolated yield.



After carefully surveying a series of commonly used metal catalysts, Pd(acac)₂ proved to be most efficient in this threecomponent transformation. As illustrated in Table 1, entry 3, when aryl iodide 2b was used, the 1,3-arylboration product cis-6a was isolated in 78% yield. Notably, only a single cis-isomer was detected in this reaction. In addition, no 1,2-arylboration products were observed in the Pd-catalyzed conditions. Further catalyst evaluation indicated that Pd(acac)₂ gave the best yield (Table 1, entries 3 and 4). Replacing the palladium catalyst with nickel salt resulted in no desired product formation (Table 1, entry 5). The importance of tetramethylammonium chloride (TMAC) additive was highlighted by replacing it with tetrabuthylammonium chloride (TBAC), tetramethylammonium bromide (TMAB) and tetramethylammonium iodide (TMAI) leading to dramatically decreased yields (Table 1, entries 6-8). These results indicate that both the cation and anion of the amounium salt are crucial to this three-component reaction. The additive likely plays a role to generate an active anionic palladium complex and/or stablize the Pd(0) catalyst from precipitation.¹³ Moreover, the examination of bases and solvents indicated that NaOAc and CHCl₃ were the best choices (Table 1, entries 9-12). Control

experiments revealed that both the additive and the base played an essential role for the success of this aryboration reaction (Table 1, entries 13 and 14).



Scheme 1. Proposed Mechanism

Subsequently, a catalytic cycle involving cis-2,3-palladium migration was proposed to rationalize this stereoselective transformation. As shown in Scheme 1, the reaction is initiated by oxidative addition of ArI with a Pd(0) species (I). In the presence of the ammonium salt (TMAC), a new Pd(II) complex III is probably generated from the Pd(II) complex II, which reacts with the unconjugated diene 1a to deliver intermediate IV. Then the intermediate IV rapidly undergoes β -H elimination and migratory insertion. The excellent stereoselectivity observed in products suggest that the Pd-H does not diassociate from the substrate in this process, which is in accordance with Larock's observations in the Pd-catalyzed carboarylation reactions.^{14,15} Therefore, the migratory insertion preceeds in a coplanar manner affording a stable π -allylPd(II) complex VI. Subsequent transmetalation with B₂pin₂ forms intermediate VII, which delivers the arylboration product 6 and the Pd(0) catalyst by reductive elimination. Interestingly, the migratory insertion of Pd-H also proceeds in a highly regioselective manner, as no 1,2- and 1,4-products are observed.

With the optimal reaction conditions in hand, we next turned our attention to the investigation of the generality of this palladium-catalyzed arylboration reaction. As outlined in Table 2, both electron-rich and electron-deficient aryl iodides successfully yielded the cis-1,3-arylboration products in moderate to good yields. Accordingly, a number of 1,3disubstituted cyclohexene derivatives were prepared. Significantly, in all cases, the *cis*-1,3-arylboration products were isolated as a single constitutional isomer. Remarkably, this reaction exhibited extraordinary functional group compatibility. A broad set of functional groups, such as ethers (OR), chlorides (Cl), bromides (Br), esters (COOR), ketones (C=O), amides (CONHR), cyano (CN), trifluoromethylsulfonates (OTf), olefins (C=C), as well as arylamines (ArNH₂), phenols (ArOH), alcohols (ROH) and indoles are all well tolerated in this reaction. Ortho-substituted (Cl and OBn) aryl iodides can also furnish the desired products in good yields (6f, 6k and 6o). It is worthy highlighting that this palladium-catalyzed arylboration shows good chemoselectivity towards polyhalogenated arenes (6p-6s), which provides further opportunities for iterative crosscouplings.¹⁶ Moreover, this method could be used to directly modify the complex bioactive molecules and their derivatives (6x-6z and 6ab).

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^{*a*} Isolated yields on 0.5 mmol scale; the number in parentheses is NMR yield. ^{*b*} Yield of the corresponding alcohol after oxidation. ^{*c*} Isolated yields on 6.0 mmol scale; ^{*d*} CsF instead of NaOAc, DCE instead of CHCl₃.

Additionally, the carboboration product **6ac** can be obtained in a moderate yield as well, when 2-iodocyclohexenone was used as the substrate. Finally, substituted 1,4-cyclohexadiene **1b** was also able to provide 1,3-arylboration products in a good yield, albeit with a poor regioisomeric ratio (**6ad** and **6ad**'). Fortunately, the two isomers could be separated by flash column chromatography.

To further demonstrate the synthetic value of this palladiumcatalyzed arylboration reaction, the scope of other dienes was explored. As shown in Table 2, when 1,5-cyclooctadiene (1,5-COD) was used as the reactant, surprisingly, a series of *cis*-1,2arylboration products were selectively obtained (**7a-7d**), probably owing to the second double bond plays a role of ligand to facilitate the reductive elimination at the *original* position.¹⁷ Moreover, the products of regioselective 1,2-arylboration only occurred at the terminal double bond, when 4-vinylcyclohexene (4-VCH) was used (**8a** and **8b**). It is noteworthy that this is the first example to achieve a selective 1,2-arylboration with unactivated terminal olefins.⁹ Furthermore, replacing B₂pin₂ with arylboronic acid (**1e**) led to the formation of *cis*-1,3-diarylation product (**9**) in a good yield by slightly optimizing the reaction conditions. Heck product, rather than arylboration product, were isolated from the reaction with allybenzene as the substrate (**10**). This result indicated that the C-B bond could only be constructed from π -allyl-palladium species, rather than from the π -benzyl-palladium intermediates in this catalytic system.^{8c} Additionally, the arylboration products could also be anticipated with acyclic conjugated dienes (**1g** and **1h**), albeit with poor selectivity between 1,2- and 1,4-regioisomers under the current reaction conditions, which offers a different

regioselectivity from the previous dual metal catalysis methods.^{9f-9g} While the 1,3-cyclohexadiene **1i** afforded the single *cis*-1,4-arylboation product **13** with a good yield.



Scheme 2. Stereospecific Transformations

To illustrate the potential of this 1,3-arylboration to create diverse structures, we conducted subsequent transformations of several products. As illustrated in Scheme 2, saturated product 14 was obtained first after hydrogenation of the olefin. Then the C-B bonds were reacted with a benzoquinone and an aryl aldehyde, providing 15^{18} and 16^{19} respectively. The C-B bonds were converted to hydroxyl groups by oxidation with H₂O₂, resulting in a series of cyclic allylalcohols (17). The alcohols were further transformed to 18 by selective oxidation of the double bond, 19 by a Mitsunobo reaction with phthalimide,²⁰ and a plethora of saturated alcohols 20a-20c via hydrogenation. In addition, a 5-arylsubstituted cyclohexenone 21 was provided by a selective oxidation of the alcohol. Notably, except the direct hydrogenation reaction (14), all other transformations are stereospecific processes.



Scheme 3. Synthesis of Bioactive Molecule

Finally, the synthetic utility of this palladium-catalyzed arylboration reaction is further demonstrated in an efficient synthesis of *anti*-diabetic agent 26,^{2f, 21} using commercially available 22 as the starting material (Scheme 3).

In summary, we have developed a novel strategy to access 1,3-disustituted cyclohexanes from benzene. This strategy integrates the classical Birch reduction with a newly developed palladium-catalyzed arylboration. 1,2-Arylboration of 1,5-

COD and 4-VCH, the downstream specific transformation and the rapid synthesis of pharmaceutically relevant molecule address the synthetic potentials of this protocol, and provide valuable additions to the FBDD libraries. Therefore, we believe this chemistry will greatly advance complex molecules synthesis and medicinal chemistry candidates.

ASSOCIATED CONTENT

Supporting Information

This material is available free of charge via the Internet at http://pubs.acs.org." Condition investigation, experimental procedures and compound characterization (PDF), and Crystallographic data (CIF)

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Notes

The authors declare no competing financial interest.

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