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Original article

Substituted 4-phenyl-2-aminoimidazoles and 4-phenyl-4,5-dihydro-2-aminoimidazoles as voltage-gated sodium channel modulators



Nace Zidar^a, Žiga Jakopin^a, David J. Madge^b, Fiona Chan^b, Jan Tytgat^c, Steve Peigneur^c, Marija Sollner Dolenc^a, Tihomir Tomašić^a, Janez Ilaš^a, Lucija Peterlin Mašič^a, Danijel Kikelj^a,*

^a University of Ljubljana, Faculty of Pharmacy, Aškerčeva 7, 1000 Ljubljana, Slovenia ^b Xention Limited, Iconix Park, London Road, Pampisford, Cambridge CB22 3EG, UK ^c University of Leuven (KU Leuven), Toxicology & Pharmacology, O&N2, PO Box 922, Herestraat 49, 3000 Leuven, Belgium

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ABSTRACT

Voltage-gated sodium channels play an integral part in neurotransmission and their dysfunction is frequently a cause of various neurological disorders. On the basis of the structure of marine alkaloid clathrodin, twenty eight new analogs were designed, synthesized and tested for their ability to block human Na_V1.3, Na_V1.4 and Na_V1.7 channels, as well as for their selectivity against human cardiac isoform Na_V1.5, using automated patch clamp electrophysiological assay. Several compounds exhibited promising activities on different Na_V channel isoforms in the medium micromolar range and some of the compounds showed also moderate isoform selectivities. The most promising results were obtained for the Na_V1.3 channel, for which four compounds were found to possess IC₅₀ values lower than 15 μ M. All of the active compounds bind to the open-inactivated states of the channels and therefore act as state-dependent modulators. The obtained results validate the approach of using natural products driven chemistry for drug discovery starting points and represent a good foundation for future design of selective Na_V modulators.

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1. Introduction

Voltage-gated sodium channels (VGSC, Na_V channels) are integral membrane proteins which play essential roles in the initiation and propagation of action potentials in neurons and other electrically excitable cells. They are composed of a single α -subunit which forms a voltage-sensing pore and one or more auxiliary β -subunits. To date, nine different α -subunits (Na_V1.1–Na_V1.9) and four different β -subunits have been identified. The VGSC isoforms are distributed differentially throughout the electrically excitable cells of the body, and have different functional properties according to their role in each tissue type. Na_V1.1–1.3 and Na_V1.6 are mainly expressed in the central nervous system, Na_V1.4 in the skeletal

muscle, $Na_V 1.5$ in the heart, and $Na_V 1.7-1.9$ are characteristic of the peripheral nervous system [1-7].

The abnormally increased activity of sodium channels leads to over-excitation of specific groups of cells which can cause different neurodegenerative diseases, chronic pain, epilepsy, arrhythmias and spasticity. Mutations of Na_V channel genes and their encoded proteins have been identified in heart, brain, skeletal muscle and peripheral nerves [8–11]. Both, gain and loss of function mutations in Na_V1.1 and Na_V1.2 can give rise to epilepsy [12,13] and migraine [14]. A number of preclinical studies have implicated Na_V1.3, Na_V1.7, Na_V1.8 and Na_V1.9 in nociceptive processing [15,16] and mutations in their genes were linked to spontaneous chronic pain. Mutations of Na_V1.4 that result in hyperactive skeletal sodium channels have been shown to cause myotonia or flaccid paralysis [17], whereas mutations of Nav1.5 have profound effects on cardiac function. leading to a range of cardiac abnormalities [18]. Owing to the extremely broad therapeutic potential of VGSC modulators, and to the recently disclosed first X-ray crystal structures of bacterial Nav channels [1-3], discovery of VGSC modulators has become one of the most attractive topics in medicinal chemistry.

Although there are several drugs acting at Na_V channels, e.g. local anesthetics (lidocaine, procaine), antiarrhythmics (lidocaine,

Abbreviations: VGSC, voltage-gated sodium channel; Nav channel, voltage-gated sodium channel; TBTU, *N.N.V.N'*-tetramethyl-O-(benzotriazol-1-yl)uronium tetra-fluoroborate; DMAP, 4-(dimethylamino)pyridine; NMM, *N*-methylmorpholine; Pro, proline; THF, tetrahydrofuran; DMF, *N.N*-dimethylformamide; DMAP, 4-dimethylaminopyridine; TFA, trifluoroacetic acid; ESI, electrospray ionization; Boc₂O, di-*tert*-butyl dicarbonate; Boc, *tert*-butyloxycarbonyl.

Corresponding author. Tel.: +386 1 4769561; fax: +386 1 4258031.

E-mail address: danijel.kikelj@ffa.uni-lj.si (D. Kikelj).

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tocainide, mexiletine) and antiepileptics (phenytoin, carbamazepine, lamotrigine), a more rational approach is required to exploit the full therapeutic potential of these drug targets. Current drugs acting on Nav channels have low potency and are relatively nonselective, therefore there is a need for the development of isoform-selective modulators which offers the promise of significant advantage over current therapeutic agents [9,19,20]. Recent studies show promise that small molecules can be made selective for different Na_V channel isoforms [8,9,21–25]. Relatively favorable side effect profiles of some currently used drugs acting on Nav channels are generally attributed to their state-dependent action. They usually have a much higher affinity for inactivated channels compared to resting channels, therefore their IC₅₀ values are several hundred times higher on hyperpolarized VGSCs than on VGSCs in depolarized membranes [26,27]. This state-dependency of action is therefore considered to be a key predictor of a high therapeutic index for a VGSC blocker.

The marine ecosystem is a rich source of chemically and functionally diverse toxins that interact with Na_V channels and provide inspiration for drug design [28,29]. Alkaloids from the Caribbean sponges of the genus Agelas, e.g. monomers clathrodin, hymenidin and oroidin, and dimers sceptrin and dibromosceptrin (Fig. 1), have been shown to interact with muscle and nerve membrane receptors and channels, including VGSCs. Electrophysiological studies suggested that clathrodin and dibromosceptrin act on Na_V channels by influencing channel ion conductance, and by modifying the channel inactivation characteristics, respectively [30]. These alkaloids belong to the pyrrole-2-aminoimidazole structural class of marine alkaloids, with intriguing structural complexity and diverse biological activities [31,32]. In contrast to the majority of neurotoxins, which have high molecular weights and many chiral centers leading to complex 3D architectures, the structures of clathrodinlike alkaloids are relatively simple. This makes them suitable candidates for optimization using established medicinal chemistry strategies. Due to their favorable size and physicochemical properties, they have the potential to permeate further into the channel and possibly bind to sites which are not available to large toxin molecules [9].

The design of the present series of compounds was based on the structure of the alkaloid clathrodin and is schematically presented in Fig. 2. The molecule of clathrodin consists of a 2aminoimidazole moiety that is connected to a pyrrole ring with a 5-atom long, partially flexible, linker. This linker gives clathrodin a certain extent of conformational freedom which is not optimal for the binding affinity. In addition, the linker contains in its structure a C=C double bond conjugated to the 2-aminoimidazole ring that can undergo various transformations, such as additions of different nucleophiles [33] or Diels–Alder cycloadditions [34,35]. The aim of the present study was to design derivatives with a conformationally restricted central part of the molecule that would at the same time lack the potentially unstable alkenyl

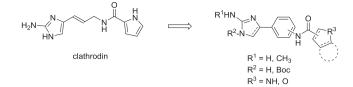


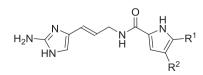
Fig. 2. Design of substituted 4-phenyl-2-aminoimidazoles and 4-phenyl-4,5-dihydro-2-aminoimidazoles as conformationally restricted clathrodin analogs.

double bond. To achieve these goals, we decided to introduce a phenyl ring at position 5 of the 2-aminoimidazole moiety so that the length of the molecule would remain unchanged. To explore the correct spatial arrangement of functional groups we designed two sets of compounds, one with 1,3- and the other with 1,4substitution patterns on the phenyl ring. In addition to the 2substituted pyrrole ring present in the clathrodin molecule, analogs with pyrrol-3-yl, furan-2-yl, indol-2-yl, indol-3-yl, (R)-pyrrolidin-2-yl or (S)-pyrrolidin-2-yl substituents were prepared. To assess the importance of the free amino group of the 2aminoimidazole ring, we designed compounds 24-32 with an N-methyl substituent in position 2 (Schemes 4 and 5). In addition, to examine whether aromaticity and planarity of the imidazole ring are important for biological activity, a set of 4,5-dihydro-2aminoimidazoles 27-32 with a reduced imidazole C=C bond was prepared (Scheme 5).

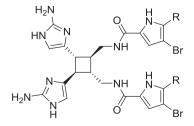
2. Results and discussion

2.1. Chemistry

The synthetic route towards the key intermediates 4a, 4b and compound 6 is presented in Scheme 1, 2-Bromo-3-nitroaceto phenone or 2-bromo-4-nitroacetophenone was reacted with 2aminopyrimidine and a catalytic amount of 4-dimethylamino pyridine in acetonitrile to afford the pyrimidinium salts **1a** or **1b**, respectively. A dehydrated compound 1c was formed instead of 1a, when the reaction of 2-aminopyrimidine and 2-bromo-3nitroacetophenone was carried out at 85 °C instead of at room temperature. This can probably be attributed to the electron withdrawing effect of the nitro group on the phenyl ring, which is more prominent when the nitro group is in the para position (compound **1b**) than in the *meta* position (compound **1a**). Hydrazinolysis of compounds **1a-c** in ethanol using microwave irradiation, followed by a Dimroth-type rearrangement resulted in formation of substituted 2-aminoimidazoles 2a and 2b. Both steps were accomplished following a literature procedure [36] where detailed mechanisms of the reactions are described. In a first attempt to prepare derivatives with carboxamido substituents on the phenyl ring, after reduction of the nitro group of **2a** by catalytic



clathrodin (*E-conf.*, $R^1 = H$, $R^2 = H$) hymenidin (*E-conf.*, $R^1 = H$, $R^2 = Br$) oroidin (*E-conf.*, $R^1 = Br$, $R^2 = Br$)



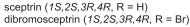
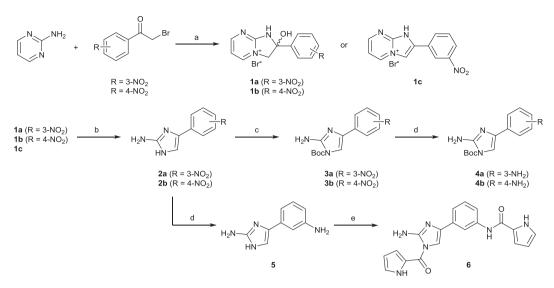


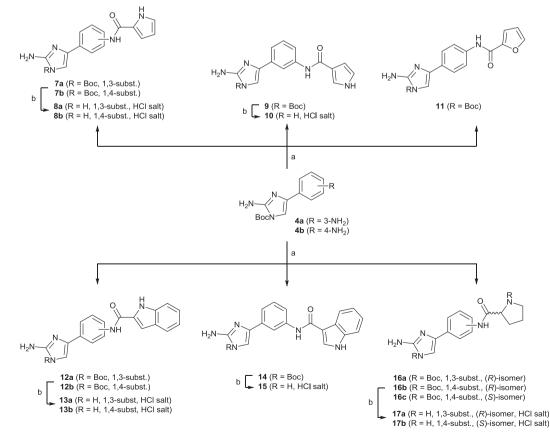
Fig. 1. Structures of Agelas alkaloids clathrodin, hymenidin, oroidin, sceptrin and dibromosceptrin.



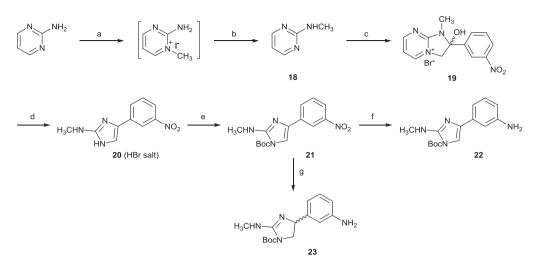
Scheme 1. Synthesis of key intermediates 4a and 4b, and compound 6. Reagents and conditions: (a) method 1: DMAP, CH₃CN, rt, 5 h (for the synthesis of 1a and 1b); method 2: DMAP, CH₃CN, 85 °C, 5 h (for the synthesis of 1c); (b) hydrazine hydrate, EtOH, MW: 120 °C, 40 min; (c) Boc₂O, 2:1 MeOH/H₂O, rt, 3 h; (d) H₂/Pd-C, THF, rt, 5 h; (e) 1 equiv pyrrole-2-carboxylic acid, TBTU, NMM, DMF, rt, 15 h.

hydrogenation, the obtained amine **5** was coupled with pyrrole-2carboxylic acid in a TBTU-promoted reaction. We found that coupling occurred at two possible sites, the aromatic amino group and the imidazole *N*-1, to give the product **6**. Interestingly, the reaction did not occur at the 2-amino group of the imidazole ring, probably because of its low reactivity. Therefore, the imidazole *N*-1 nitrogen was Boc-protected by reacting compounds **2a** and **2b** with Boc-anhydride in a mixture of methanol and water. The reduction of the nitro group of Boc-protected derivatives **3a** and **3b** was accomplished through catalytic hydrogenation in tetrahydrofuran using Pd–C as a catalyst, to give the target amines **4a** and **4b**.

The 4-phenyl-2-aminoimidazole analogs **7–17** were prepared according to Scheme 2. TBTU-promoted coupling of **4a** or **4b** and the corresponding carboxylic acid (pyrrole-2-carboxylic acid,



Scheme 2. Synthesis of 4-phenyl-2-aminoimidazoles 7–17. Reagents and conditions: (a) corresponding carboxylic acid, TBTU, NMM, CH₂Cl₂, 35 °C, 24 h; (b) method 1: HCl_(g), THF/ EtOH, rt, 5 h (for the synthesis of 8a, 10, 13a, 15 and 17a); method 2: acetyl chloride, EtOH, 0 °C, 1 h (for the synthesis of 8b, 13b and 17b).



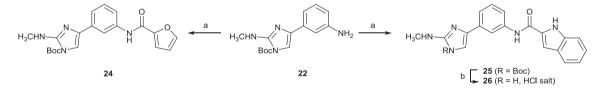
Scheme 3. Synthesis of *N*-methylated intermediates 22 and 23. *Reagents and conditions*: (a) CH₃I, acetone, 75 °C, 15 h; (b) NaOH, EtOH/H₂O, 80 °C, 0.5 h; (c) 2-bromo-3nitroacetophenone, DMAP, CH₃CN, 50 °C, 15 h; (d) hydrazine hydrate, EtOH, MW: 120 °C, 40 min; (e) Boc₂O, 2:1 MeOH/H₂O, rt, 3 h; (f) H₂/Pd–C, THF, rt, 5 h; (g) H₂/Pd–C, THF, rt, 24 h.

pyrrole-3-carboxylic acid, furan-2-carboxylic acid, indole-2carboxylic acid, indole-3-carboxylic acid, *N*-Boc-L-proline or *N*-Boc-D-proline) afforded Boc-protected derivatives **7a**, **7b**, **9**, **11**, **12a**, **12b**, **14** and **16a**–**c** which were converted into final compounds upon cleavage of the Boc protecting group with either gaseous hydrochloric acid or acetyl chloride–ethanol as a convenient reagent for *in situ* HCl formation [37].

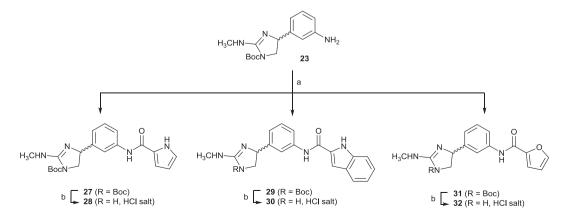
N-Methylated intermediates **22** and **23** were prepared according to Scheme 3. 2-(*N*-Methylamino)-pyrimidine (**18**) was prepared by heating 2-aminopyrimidine and iodomethane in acetone followed by treatment with 10% aqueous sodium hydroxide. This was then reacted with 2-bromo-3-nitroacetophenone in acetonitrile and a catalytic amount of 4-dimethylaminopyridine to afford 2-hydroxy-

1-methyl-2-(3-nitrophenyl)-2,3-dihydro-1*H*-imidazo[1,2-*a*]pyrimidin-4-ium bromide (**19**). After hydrazinolysis in ethanol using microwave irradiation and subsequent Dimroth-type rearrangement, compound **19** was converted into 2-(*N*-methylamino)imidazole **20** [36]. Boc protection of the imidazole *N*-1 of compound **20**, followed by reduction of the nitro group afforded the amine **22**. A prolonged catalytic hydrogenation (24 h) enabled both, the reduction of the nitro group and the imidazole C==C bond, to give the 4,5-dihydro-2-(*N*-methylamino)-imidazole derivative **23**.

4-Phenyl-2-(*N*-methylamino)-imidazoles **24–26**, and 4-phenyl-4,5-dihydro-2-(*N*-methylamino)-imidazoles **27–32** were prepared according to Schemes 4 and 5. TBTU-promoted coupling of amines **22** or **23**, and the corresponding carboxylic acid (pyrrole-2-



Scheme 4. Synthesis of 4-phenyl-2-(*N*-methylamino)-imidazoles 24–26. Reagents and conditions: (a) corresponding carboxylic acid, TBTU, NMM, CH₂Cl₂, 35 °C, 24 h; (b) HCl_(g), THF/ EtOH, rt, 5 h.



Scheme 5. Synthesis of 4-phenyl-4,5-dihydro-(*N*-methylamino)-imidazoles 27–32. Reagents and conditions: (a) corresponding carboxylic acid, TBTU, NMM, CH₂Cl₂, 35 °C, 24 h; (b) HCl_(g), EtOH, rt, 5 h.

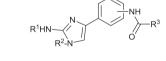
carboxylic acid, indole-2-carboxylic acid or furan-2-carboxylic acid) afforded Boc-protected derivatives **24**, **25**, **27**, **29** and **31** which were converted into end products upon cleavage of the Boc protecting group.

2.2. Biological evaluation

In total, twenty eight novel compounds were synthesized and studied for their ability to block the human sodium channel isoforms Na_V1.3, Na_V1.4 and Na_V1.7, as well as their selectivity against the human cardiac isoform Na_V1.5. The screening was performed using an automated patch clamp electrophysiology assay on the Sophion QPatch HT system (Sophion Bioscience A/S). Compounds of three different structural classes, (i) 4-phenyl-2-aminoimidazoles 6-17, (ii) 4-phenyl-2-(N-methylamino)-imidazoles 24-26, and (iii) 4-phenyl-4,5-dihydro-2-(N-methylamino)-imidazoles 27–32, were tested and their IC₅₀ values, *i.e.* concentrations of compounds that inhibit sodium channel currents by 50%, are presented in Tables 1 and 2. The IC₅₀ values of synthesized compounds against each Na_V isoforms were determined from concentration-response relationships established by cumulatively applying four escalating concentrations (0.3–10 μ M) of test compound to a cell, as detailed in the Experimental section. All of the active compounds acted as state-dependent VGSC modulators, blocking only the openinactivated state of the channels (depolarization from ~ -70 mV) and possessing no activity on the closed state of the channels, with IC_{50} values higher than 30 μ M (depolarization from -100 mV, data

Table 1

Modulatory activities of 4-phenyl-2-aminoimidazoles **6–17** and 4-phenyl-2-(*N*-methylamino)-imidazoles **24–26** on the open-inactivated state of the Na_V channel isoforms.



$Compd \ R^1$		R ²	R ³	Subst	IC ₅₀ (μM) ^a			
					Na _V 1.3	Na _V 1.4	Na _V 1.5	Na _V 1.7
6	Н	Pyrrol- 2-oyl	Pyrrol-2-yl	1,3	25	_b	27	19
7a	Н	Boc	Pyrrol-2-yl	1,3	25	25	_b	_b
7b	Н	Boc	Pyrrol-2-yl	1,4	_ ^b	_b	_b	_ ^b
8a ^c	Н	Н	Pyrrol-2-yl	1,3	_ ^b	_b	_b	_ ^b
8b ^c	Н	Н	Pyrrol-2-yl	1,4	_b	_b	_b	27
9	Н	Boc	Pyrrol-3-yl	1,3	23	_b	_b	_b
10 ^c	Н	Н	Pyrrol-3-yl	1,3	12	15	_b	_b
11	Н	Boc	Furan-2-yl	1,4	27	_b	_b	22
12a	Н	Boc	Indol-2-yl	1,3	26	_b	_b	_b
12b	Н	Boc	Indol-2-yl	1,4	22	28	_b	_b
13a ^c	Н	Н	Indol-2-yl	1,3	_b	_b	_b	_b
13b ^c	Н	Н	Indol-2-yl	1,4	_b	_b	_b	_b
14	Н	Boc	Indol-3-yl	1,3	27	26	_b	_b
15 ^c	Н	Н	Indol-3-yl	1,3	12	14	_b	26
16a	Н	Boc	(R)-Pyrrolidin-2-yl	1,3	_b	_b	_b	29
16b	Н	Boc	(R)-Pyrrolidin-2-yl	1,4	_b	_b	20	_b
16c	Н	Boc	(S)-Pyrrolidin-2-yl	1,4	25	_b	_b	19
17a ^c	Н	Н	(R)-Pyrrolidin-2-yl	1,3	8	_b	_b	28
17b ^c	Н	Н	(S)-Pyrrolidin-2-yl	1,4	13	_b	_b	_b
24	CH_3	Boc	Furan-2-yl	1,3	_b	_b	_b	26
25	CH_3	Boc	Indol-2-yl	1,3	_b	_b	_b	_ ^b
26 ^c	CH_3	Н	Indol-2-yl	1,3	_b	_b	20	_ ^b
Lidocair	Lidocaine					2	N/A ^d	29

^a The concentration of compound that inhibits a sodium channel current by 50%. Each IC_{50} value is the mean of three independent experiments.

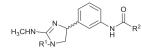
^b IC₅₀ value higher than 30 μ M.

^c HCl salt.

^d N/A: not available.

Table 2

Modulatory activities of 4-phenyl-4,5-dihydro-2-(*N*-methylamino)-imidazoles 27-32 on the open-inactivated state of the Na_V channel isoforms.



Compd	\mathbb{R}^1	R ²	$IC_{50} (\mu M)^{a}$				
			Na _v 1.3	Na _V 1.4	Na _v 1.5	Na _V 1.7	
27	Boc	Pyrrol-2-yl	24	_b	24	23	
28 ^c	Н	Pyrrol-2-yl	21	_b	_b	26	
29	Boc	Indol-2-yl	_b	_b	3	18	
30 ^c	Н	Indol-2-yl	_b	_b	b	_b	
31	Boc	Furan-2-yl	_b	_b	_b	_b	
32 ^c	Н	Furan-2-yl	22	_b	_b	27	
Lidocaine		•	10.5	2	N/A ^d	29	

 $^{\rm a}$ The concentration of compound that inhibits a sodium channel current by 50%. Each IC_{50} value is the mean of three independent experiments.

^b IC₅₀ value higher than 30 μM.

^c HCl salt.

^d N/A: not available.

not shown). IC₅₀ values for lidocaine as reference substance were 10.5 μ M against Na_V1.3, 2.1 μ M against Na_V1.4 and 29 μ M against Na_V1.7 channel. Activity of lidocaine against Na_V1.5 isoform was not determined.

Several compounds exhibited promising activities on different Nav channel isoforms in medium micromolar range and some of the compounds showed also moderate isoform selectivities, e.g., compounds 8b, 9, 12a, 16a, 16b, 17b, 24 and 26. The most promising results were obtained for the Nav1.3 channel where four compounds had IC₅₀ values lower than 15 µM. The most potent compound was the (R)-proline derivative **17a** blocking Na_V1.3 with an IC₅₀ value of 8 µM. Compound **17a** was selective towards Na_V1.4 and Na_V1.5, and weakly active against the Na_V1.7 isoform $(IC_{50} = 28 \ \mu M)$. Channel-blocking activity against the Na_V1.3 with an IC₅₀ value of 13 μ M was observed also for compound **17b**, an (S)proline-containing analog of **17a** with a 1,4-substitution pattern on the phenyl ring. Moreover, compound 17b did not possess any activity on other isoforms and could thus represent a starting point for the development of selective modulators of Na_V1.3. Two other compounds of the series, pyrrol-3-yl derivative 10 and indol-3-yl derivative 15, also possessed Nav1.3 modulatory activity, both with an IC_{50} value of 12 μ M. Both compounds were also active against the Na_V1.4 isoform with IC_{50} values of 15 μM and 14 $\mu M,$ respectively. Interestingly, isomers of compounds 10 and 15 containing 2- instead of 3-substituted pyrrole or indole rings, pyrrol-2yl derivative 8a and indol-2-yl derivative 13a were devoid of activity on all Nav channel isoforms, suggesting the importance of the orientation of pyrrole or indole rings for antagonist activity. Furthermore, based on comparison of the isomeric pyrrol-3-yl derivatives $\textbf{7a}~(IC_{50} \text{ value of } 25 ~\mu\text{M} \text{ against } Na_V 1.3)$ and $\textbf{7b}~(IC_{50}$ value $>30 \mu$ M against Na_V1.3), the 1,3-substitution pattern on the phenyl ring is preferred over 1,4-substitution pattern for the Na_V1.3 modulatory activity.

The most active modulators of the skeletal muscle isoform Na_V1.4 were 4-phenyl-2-aminoimidazoles **10** (IC₅₀ = 15 μ M) and **15** (IC₅₀ = 14 μ M) which were also active against the Na_V1.3 channel with IC₅₀ values of 12 μ M. The introduction of *N*-methyl substituent on the primary amino group of 2-aminoimidazole ring in derivatives **24–32** resulted in complete loss of activity against the Na_V1.4 isoform.

The $Na_V 1.5$ channel isoform is located in the heart. Blockade of this channel is associated with increased risk of a cardiac side-effect

liability, and it is regarded as an unwanted activity in any therapeutic molecule. Fortunately, the majority of tested compounds were almost inactive against the Na_V1.5, with the exception of compound **29** with an IC₅₀ of 3 μ M.

The Na_V1.7 isoform is mainly located in the peripheral nervous system and has been identified as a promising drug target for the treatment of neuropathic pain [15]. The most active compound against the Nav1.7 channel was the indol-2-vl derivative 29 $(IC_{50} = 18 \ \mu M)$, but as it also displayed activity against the Na_V1.5 isoform (IC₅₀ = 3 μ M) it is not suitable for further development. A moderate activity against the Nav1.7 isoform was observed also for 4-phenyl-2-aminoimidazoles 6 and 16c, both having IC₅₀ values of 19 µM. Compound **16c** displayed also a weak Na_V1.3 modulatory activity with IC₅₀ value of 25 µM. As both, Na_V1.3 and Na_V1.7 channels have been linked to nociceptive processing, finding compounds of dual activity against both isoforms could lead to synergistic effects and contribute to favorable therapeutic profiles of such compounds. Interestingly, compound 16b, the R isomer of 16c, was not active neither against the Na_V1.3 nor the Na_V1.7 channels

3. Conclusion

We have designed and synthesized a series of clathrodin analogs and evaluated their effects on human Na_V1.3–1.5 and Na_V1.7 channels using an automated patch clamp electrophysiology assay. Several compounds exhibited promising activities with moderate isoform selectivities. All of the active compounds acted as statedependent VGSC modulators. The most promising results were obtained for the Na_V1.3 channel, against which four compounds were found to possess IC₅₀ values lower than 15 μ M. The results provide valuable information on structure–activity relationships and offer good prospect for the development of isoform selective VGSC modulators.

4. Experimental section

4.1. Electrophysiology

Cells were prepared by dissociation from T175 cell culture flasks using trypsin–EDTA (0.05%), cells were kept in serum free media in the cell hotel on board the QPatch HT. These cells were sampled, washed and re-suspended in extracellular recording solution by the QPatch HT immediately before application to well site on the chip. Once in whole-cell configuration, vehicle (0.1% DMSO v/v) was applied to the cells to achieve a stable control recording (4-min total). This was followed by application of test concentrations as a single bolus addition (4-min incubation per test concentration). Compounds were prepared in extracellular recording solution from a 10 mM (100% DMSO) stock to yield a final 10 μ M (0.1% DMSO) test concentration from which subsequent serial dilutions in extracellular solution were performed (0.3–10 µM). Voltage protocols for the sodium channels being screened were designed to reflect the high-frequency, pathophysiological state of the channels that may be therapeutic targets (Nav1.3, Nav1.4 and Nav1.7), and the lowfrequency, physiological state of the safety target (Nav1.5). Currents were elicited from Na_V1.3, Na_V1.4 and Na_V1.7 cell lines using a standard two-pulse voltage protocol. From a holding potential of -100 mV, a 20 ms activating step to -20 mV was applied to assess the effect of compounds on resting (closed) state block. The second activating pulse was applied following a 5-s pre-pulse to half inactivation potential (variable depending on the sodium channel studied, -65 to -75 mV) to assess block on the openinactivated state of the channel. This protocol was applied at a sweep interval of 0.067 Hz throughout the duration of the experiment. To study Nav1.5 currents, a pulse train consisting of 10 repetitive activating test pulses to -20 mV from a holding potential of -100 mV were applied at a 1 Hz frequency until 10 pulses were reached, this sequence was repeated at a sweep interval of 0.016 Hz throughout the duration of the experiment. For Na_V1.3, Na_V1.4 and Na_V1.7 channels the peak inward current was determined for both the closed and open-inactivated test pulses from each sweep applied to the cells and for Nav1.5 from the tenth pulse of each pulse train recorded. Data was captured using QPatch assay software (v5.0). The % inhibition of peak current was calculated as the mean peak current value for the last three sweeps measured in each concentration test period relative to the last three sweeps recorded during the control vehicle period. Sigmoidal concentration response curves (four parameter logistic curves) were fitted to the % inhibition data using Xlfit (IDBS) from which the IC₅₀ was determined. Fits were constrained at 0 and 100%. Data are presented as mean \pm SD for minimum of 3 independent observations.

4.2. Chemistry – general

Chemicals were obtained from Acros, Aldrich Chemical Co., and Fluka and used without further purification. Analytical TLC was performed on silica gel Merck 60 F₂₅₄ plates (0.25 mm), using visualization with UV light and ninhydrin. Column chromatography was carried out on silica gel 60 (particle size 240-400 mesh). HPLC analyses were performed on an Agilent Technologies 1100 instrument with a G1365B UV-VIS detector, a G1316A thermostat and a G1313A autosampler using a Phenomenex Luna 5 µm C18 column $(4.6 \times 150 \text{ mm or } 4.6 \times 250 \text{ mm})$ and flow rate of 1.0 mL/min. The eluent consisted of trifluoroacetic acid (0.1% in water) or ammonia (0.1% in water) as solvent A and methanol as solvent B. Microwaveassisted reactions were performed using a CEM Discover microwave reactor (CEM Corp.). Melting points were determined on a Reichert hot stage microscope and are uncorrected. ¹H NMR and ¹³C NMR spectra were recorded at 400 and 100 MHz, respectively, on a Bruker AVANCE III 400 spectrometer in DMSO-d₆ or MeOH-d₄ solution, with TMS as the internal standard. IR spectra were recorded on a Perkin-Elmer Spectrum BX FT-IR spectrometer or Thermo Nicolet Nexus 470 ESP FT-IR spectrometer. Mass spectra were obtained using a VGAnalytical Autospec Q mass spectrometer. Optical rotations were measured on a Perkin-Elmer 241 MC polarimeter. The reported values for specific rotation are average values of 10 successive measurements using an integration time of 5 s. The purity of the tested compounds was established to be >95%.

4.3. Synthetic procedures

4.3.1. General procedure A. Synthesis of compounds 1a-c (with 1c as an example)

To a solution of 2-aminopyrimidine (325 mg, 3.41 mmol) and 2bromo-4-nitroacetophenone (1.00 g, 4.10 mmol) in acetonitrile (15 mL) 4-dimethylaminopyridine (4.2 mg, 0.034 mmol) was added. After being stirred at 85 °C for 5 h, the reaction mixture was filtered, washed with acetonitrile (10 mL) and ether (2 × 10 mL), and dried to give **1c**.

4.3.1.1. 2-(3-Nitrophenyl)-1H-imidazo[1,2-a]pyrimidin-4-ium bromide (**1c**). Yield, 66% (0.726 g); orange solid; mp 251–255 °C; IR (KBr) ν = 3075, 3023, 2674, 2579, 2557, 1642, 1526, 1483, 1341, 1304, 1234, 1109, 1083, 928, 915, 805, 790, 732 cm⁻¹. ¹H NMR (DMSO-d₆) δ 7.41 (dd, 1H, ³*J*₁ = 6.8 Hz, ³*J*₂ = 4.4 Hz, Ar-H-6), 7.86 (t, 1H, ³*J* = 8.0 Hz, Ar-H-5'), 8.29–8.32 (m, 1H, Ar-H-4'/6'), 8.45–8.48 (m, 1H, Ar-H-4'/6'), 8.83–8.85 (m, 2H, Ar-H-3, Ar-H-7), 8.86 (t, 1H, ⁴*J* = 2.0 Hz, Ar-H-2'), 9.20 (dd, 1H, ³*J* = 6.8 Hz, ⁴*J* = 2.0 Hz, Ar-H-5), signal for the NH proton not seen; ¹³C NMR (DMSO-d₆) δ 110.46, 112.40, 120.53, 124.14, 130.71, 130.94, 132.20, 137.08, 137.18, 145.82, 148.45, 155.45; MS (ESI) m/z (%) = 241 ([M - Br]⁺, 100). HRMS for C₁₂H₉N₄O₂: calculated 241.0726; found 241.0724.

4.3.2. General procedure *B*. Synthesis of compounds **2a** and **2b** (with **2a** as an example)

To a suspension of **1a** or **1c** (0.590 mmol) in ethanol (3 mL) in a 10 mL glass vessel hydrazine hydrate (0.369 mL, 35% hydrazine in solution, 4.13 mmol) was added, the vessel was sealed, placed in a microwave reactor, and heated at 120 °C for 40 min (maximum power = 50 W, ramp time = 3 min). The mixture was cooled to rt, the solvent evaporated and the resulting residue was purified by flash column chromatography using dichloromethane/ methanol = 10/1 with dissolved NH_{3(g)} as an eluent, to afford 2-aminoimidazole **2a**.

4.3.2.1. 4-(3-Nitrophenyl)-1H-imidazol-2-amine (**2a**). Yield, 90%; orange solid; mp 180–182 °C; IR (KBr) ν = 3446, 3361, 3079, 2854, 2674, 1622, 1548, 1516, 1482, 1342, 1260, 1173, 1146, 1033, 867, 799, 725 cm⁻¹. ¹H NMR (DMSO-*d*₆) δ 5.45 (s, 2H, NH₂, D₂O exchangeable), 7.27 (s, 1H, Ar-H-5), 7.55 (t, 1H, ³*J* = 8.0 Hz, Ar-H-5'), 7.90–7.93 (m, 1H, Ar-H-4'/6'), 8.01–8.04 (m, 1H, Ar-H-4'/6'), 8.42 (s, 1H, Ar-H-2'), 10.59 (br s, 1H, NH, D₂O exchangeable); ¹³C NMR (MeOH-*d*₄) δ 111.54, 117.78, 119.72, 129.24, 129.29, 132.87, 135.84, 148.76, 151.14; MS (ESI) *m/z* (%) = 205 (MH⁺, 100). HRMS for C₉H₉N₄O₂: calculated 205.0726; found 205.0721.

4.3.3. General procedure C. Synthesis of compounds **3a** and **3b** (with **3a** as an example)

To a suspension of **2a** (0.500 g, 2.45 mmol) in a mixture of methanol (10 mL) and water (5 mL) Boc-anhydride (1.07 g, 2.90 mmol) was added and the mixture was stirred at rt for 3 h. The precipitate was filtered off, washed with methanol (2×5 mL) and dried to give **3a**.

4.3.3.1. tert-Butyl 2-amino-4-(3-nitrophenyl)-1H-imidazole-1carboxylate (**3a**). Yield, 95%; yellow solid; mp 178–180 °C; IR (KBr) ν = 3413, 3278, 3122, 2982, 1739, 1637, 1526, 1477, 1447, 1344, 1320, 1273, 1257, 1212, 1152, 1119, 1066, 903, 836, 773, 716 cm⁻¹. ¹H NMR (DMSO-*d*₆) δ 1.60 (s, 9H, *t*-Bu), 6.73 (br s, 2H, NH₂, D₂O exchangeable), 7.64 (t, 1H, ³*J* = 8.0 Hz, Ar-H-5'), 7.67 (s, 1H, Ar-H-5), 8.06–8.09 (m, 1H, Ar-H-4'/6'), 8.19–8.22 (m, 1H, Ar-H-4'/6'), 8.55 (t, 1H, ⁴*J* = 2.0 Hz, Ar-H-2'); ¹³C NMR (DMSO-*d*₆) δ 28.00 (CCH₃), 85.45 (CCH₃), 108.88, 119.33, 121.80, 130.40, 131.31, 135.28, 135.78, 148.70, 149.25, 151.18; MS (ESI) *m*/*z* (%) = 305 (MH⁺, 10), 249 ([MH – *t*-Bu]⁺, 90), 205 ([MH – Boc]⁺, 100). HRMS for C₁₄H₁₇N₄O₄: calculated 305.1250; found 305.1255.

4.3.4. General procedure D. Synthesis of compounds **4a**, **4b** and **5** (with **4a** as an example)

Compound **3a** (1.00 g, 3.29 mmol) was dissolved in THF (50 mL), Pd/C (200 mg) was added and the reaction mixture was stirred under hydrogen atmosphere for 5 h. The catalyst was filtered off and the solvent removed under reduced pressure to give **4a**.

4.3.4.1. tert-Butyl 2-amino-4-(3-aminophenyl)-1H-imidazole-1carboxylate (**4a**). Yield, 95%; white solid; mp 160–162 °C; IR (KBr) ν = 3424, 3345, 3127, 2972, 1737, 1636, 1617, 1546, 1452, 1359, 1326, 1267, 1239, 1207, 1159, 1119, 1062, 847, 769, 721 cm⁻¹. ¹H NMR (DMSO-*d*₆) δ 1.58 (s, 9H, *t*-Bu), 5.02 (br s, 2H, NH₂, D₂O exchangeable), 6.42–6.45 (m, 1H, Ar-H-4'/6'), 6.55 (br s, 2H, NH₂, D₂O exchangeable), 6.86–6.88 (m, 1H, Ar-H-4'/6'), 6.95–6.99 (m, 2H, Ar-H-2', Ar-H-5'), 7.12 (s, 1H, Ar-H-5); ¹³C NMR (DMSO-*d*₆) δ 27.50 (CCH₃), 84.48 (CCH₃), 105.28, 110.32, 112.60, 112.80, 128.76, 133.78, 137.69, 148.57, 148.91, 150.15; MS (ESI) m/z (%) = 275 (MH⁺, 100). HRMS for C₁₄H₁₉N₄O₂: calculated 275.1508; found 275.1504.

4.3.5. General procedure E. Synthesis of compounds **7a**, **7b**, **9**, **11**, **12a**, **12b**, **14**, **16a**–**c**, **24**, **25**, **27**, **29** and **31** (with **7a** as an example)

To a suspension of pyrrole-2-carboxylic acid (122 mg, 1.09 mmol) and TBTU (380 mg, 1.18 mmol) in dichloromethane (5 mL) *N*-methylmorpholine (0.501 mL, 4.56 mmol) was added, and the mixture stirred at rt for 0.5 h upon which a clear solution formed. Compound **4a** (250 mg, 0.91 mmol) was added and the mixture stirred at 35 °C overnight. The solvent was evaporated in vacuo, the residue dissolved in ethyl acetate (30 mL), and washed successively with water (2 × 10 mL), saturated aqueous NaHCO₃ solution (2 × 10 mL), and brine (1 × 10 mL). The organic phase was dried over Na₂SO₄, filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography using ethyl acetate/petroleum ether or dichloromethane/methanol as an eluent, to afford **7a**.

4.3.5.1. tert-Butyl 4-(3-(1H-pyrrole-2-carboxamido)phenyl)-2amino-1H-imidazole-1-carboxylate (7a). Yield, 45% (150 mg); white solid; mp 175–177 °C; IR (KBr) *ν* = 3404, 3354, 3142, 2976, 1731, 1644, 1555, 1517, 1431, 1361, 1311, 1255, 1199, 1157, 1122, 1089, 1038, 758, 716 cm⁻¹. ¹H NMR (DMSO-*d*₆) δ 1.59 (s, 9H, *t*-Bu), 6.16-6.18 (m, 1H, Pyrr-H), 6.63 (br s, 2H, NH₂, D₂O exchangeable), 6.96-6.98 (m, 1H, Pyrr-H), 7.09-7.11 (m, 1H, Pyrr-H), 7.26 (s, 1H, Ar-H-5), 7.29 (t, 1H, ${}^{3}I = 8.0$ Hz, Ar-H-5'), 7.41–7.43 (m, 1H, Ar-H-4'/6'), 7.67-7.69 (m, 1H, Ar-H-4'/6'), 8.07 (t, 1H, ${}^{4}J = 2.0$ Hz, Ar-H-2'), 9.77 (s. 1H, NH, D₂O exchangeable), 11.63 (s. 1H, NH, D₂O exchangeable); ¹³C NMR (DMSO-*d*₆) δ 27.51 (CCH₃), 84.67 (<u>C</u>CH₃), 105.93, 108.89, 111.25, 116.28, 118.54, 119.41, 122.47, 126.06, 128.57, 133.69, 137.03, 139.47, 148.88, 150.39, 159.08; MS (ESI) m/z (%) = 368 (MH⁺, 20), 312 ($[MH - t-Bu]^+$, 100), 268 ($[MH - Boc]^+$, 60). HRMS for C19H22N5O3: calculated 368.1723; found 368.1724. HPLC: Phenomenex Luna 5 μ m C18 column (4.6 mm \times 150 mm); mobile phase: 30-80% of MeOH in TFA (0.1%) in 25 min; flow rate 1.0 mL/min; injection volume: 20 µL; retention time: 9.813 min (99.6% at 254 nm, 99.3% at 280 nm).

4.3.6. General procedure F. Synthesis of compounds **8a**, **10**, **13a**, **15**, **17a**, **26**, **28**, **30** and **32** (with **8a** as an example)

Solution of compound **7a** (108 mg, 0.250 mmol) in a 1:1 mixture of THF and EtOH (15 mL) was saturated with gaseous HCl and stirred at rt for 5 h. The solvent was removed under reduced pressure, the solid was filtered off and washed with diethyl ether (3×5 mL), to afford **8a**.

4.3.6.1. 4-(3-(1H-Pyrrole-2-carboxamido)phenyl)-2-amino-1H-imidazol-3-ium chloride (8a). Yield, 85% (76 mg); brown foam; mp 170–174 °C; IR (KBr) ν = 3232, 3142, 2972, 2766, 1676, 1625, 1607, 1547, 1529, 1486, 1422, 1324, 1227, 1158, 1114, 1044, 883, 742 cm⁻¹. ¹H NMR (DMSO- d_6) δ 6.17–6.19 (m, 1H, Pyrr-H), 6.97–6.99 (m, 1H, Pyrr-H), 7.09-7.11 (m, 1H, Pyrr-H), 7.28 (s, 1H, Ar-H-5), 7.34 (d, 1H, ${}^{3}J = 8.0$ Hz, Ar-H-4'/6'), 7.41 (t, 1H, ${}^{3}J = 8.0$ Hz, Ar-H-5'), 7.48 (s, 2H, NH₂, D₂O exchangeable), 7.67 (d, 1H, ${}^{3}J = 8.0$ Hz, Ar-H-4[']/6[']), 8.03 (s, 1H, Ar-H-2'), 10.04 (s, 1H, NH, D₂O exchangeable), 11.80 (s, 1H, NH, D₂O exchangeable), 12.18 (br s, 1H, NH, D₂O exchangeable), 12.87 (br s, 1H, NH, D₂O exchangeable); ¹³C NMR (DMSO- d_6) δ 108.95, 109.32, 111.86, 116.11, 119.22, 120.00, 122.65, 125.86, 126.48, 128.01, 129.20, 139.83, 147.77, 159.12; MS (ESI) m/z (%) = 268 ([M - Cl]⁺, 100). HRMS for C₁₄H₁₄N₅O: calculated 268.1198; found 268.1193. HPLC: Phenomenex Luna 5 μ m C18 column (4.6 mm \times 150 mm); mobile phase: 10-90% of MeOH in TFA (0.1%) in 20 min; flow rate 1.0 mL/min; injection volume: 20 µL; retention time: 13.386 min (95.1% at 254 nm, 95.1% at 280 nm).

4.3.7. General procedure *G*. Synthesis of compounds **8b**, **13b** and **17b** (with **8b** as an example)

To an ice cold solution of **7b** (200 mg, 0.544 mmol) in ethanol (4 mL) acetyl chloride (600 μ L, 8.4 mmol) was added dropwise. After being stirred at 0 °C for 1 h the reaction mixture was allowed to warm to rt, the solvent and the excess HCl were evaporated and the solid residue was triturated with diethyl ether, yielding **8b**.

4.3.7.1. 4-(4-(1H-Pyrrole-2-carboxamido)phenyl)-2-amino-1H-imidazol-3-ium chloride (**8b**). Yield, 42% (69 mg); white solid; mp 114– 116 °C; IR (KBr) $\nu = 3421, 1656, 1412, 1333, 1124, 836, 746 \text{ cm}^{-1}, {}^{1}\text{H}$ NMR (DMSO-*d*₆) δ 6.16–6.19 (m, 1H, Pyrr-H), 6.98 (br s, 2H, NH₂, D₂O exchangeable), 7.08 (s, 1H, Pyrr-H), 7.29 (s, 1H, Pyrr-H), 7.41 (s, 1H, Ar-H-5), 7.61 (d, 2H, ${}^{3}J = 8.7$ Hz, Ar-H-2',6'/3',5'), 7.84 (d, 2H, $^{3}J = 8.7$ Hz, Ar-H-2',6'/3',5'), 9.94 (s, 1H, NH, D₂O exchangeable), 11.73 (s, 1H, NH, D₂O exchangeable), 12.02 (s, 1H, NH, D₂O exchangeable), 12.76 (s, 1H, NH, D₂O exchangeable); ¹³C NMR $(DMSO-d_6) \delta$ 108.44, 108.96, 111.73, 119.88, 122.24, 122.73, 124.60, 125.89, 126.45, 139.20, 147.46, 159.08; MS (ESI) $m/z = 268 [M - Cl]^+$. HRMS for C₁₄H₁₄N₅O: calculated 268.1198; found 268.1194. HPLC: Phenomenex Luna 5 μ m C18 column (4.6 mm \times 150 mm); mobile phase: 50-80% of methanol in trifluoroacetic acid (0.1%) in 30 min; flow rate 1.0 mL/min; injection volume: 10 µL; retention time: 12.880 min (95.1% at 254 nm).

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Appendix A. Supplementary data

Supplementary data related to this article can be found at http://dx.doi.org/10.1016/j.ejmech.2013.12.034.

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