



Structure–Activity Relationships of New 1-substitutedmethyl-4-[5-(*N*-methyl-*N*-propylamino) pentyloxy]piperidines and Selected 1-[(*N*-substituted-*N*-methyl)-3-propyloxy]-5-(*N*-methy-l-*N*-propyl)-pentanediamines as H₃-Antagonists

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Novel, potent non-imidazole histamine H₃ receptor antagonists have been prepared and *in vitro* tested as H₃-receptor antagonists (the electrically evoked contraction of the guinea-pig jejunum). The present compounds contain a 4-hydroxypiperidine core, which behaves as a conformationally restricted version of the 3-amino-1-propanol moiety common to the many previously described non-imidazole H₃ ligands. Detailed structure-activity studies revealed that 1-(2-benzofuranlylmethyl)- 5c (pA₂ = 8.47 ± 0.05) and 1-(3-benzofuranlylmethyl)-4-[5-(*N*-methyl-*N*-propyl)pentyloxy]piperidine 5d (pA₂ = 8.15 ± 0.07) exhibit high potency for the H₃ histamine receptor. In addition, the potency of selected 1-[(*N*-substituted-*N*-methyl)-3-propyloxy]-5-(*N*-methyl-*N*-propyl)pentanediamines as antagonist of the H₃ histamine receptor was also evaluated. Replacement of the 4-hydroxypiperidine of the leads 7 and 5c by a highly flexible 3-(methylamino)propyloxy chain yields compounds 6a (pA₂ = 8.02) and 6b (pA₂ = 6.23) with higher and lower potency than their piperidine analogues (7, pA₂ = 7.79; 5c, pA₂ = 8.47), respectively. The histaminergic H₁ antagonism of selected compounds 5c, 5d and 6a has been established on the isolated guinea-pig ileum by conventional methods; the pA₂ values have compared with the potency of pyrilamine. None of them showed any H₁-antagonistic activity (pA₂ < 4; for pyrilamine pA₂ = 8.5).

Key words: histamine H₃-receptor, H₃ non-imidazole antagonists, 1-substitutedmethyl-4-[5-(*N*-methyl-*N*-propylamino)pentyloxy]piperidines, 1-[(*N*-substituted-*N*-methyl)-3-propyloxy]-5-(*N*-methyl-*N*-propyl)pentanediamines

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Histamine is a biogenic amine involved in a large variety of physiological functions. It exerts its actions through four distinct G protein-coupled receptors, named H₁, H₂, H₃ and H₄ (1,2). H₁ and H₂ receptor antagonists are well-known therapeutic agents and are in use for the treatment of allergic disease (3) and peptic ulcer (4), respectively. H₄ receptor is involved in inflammatory processes and takes an immunomodulatory role. Ligands of these receptors could be effective in the regulation of the immune response (5).

The histamine H₃ receptor was discovered in 1983 by Arrang *et al.* (6) as a presynaptic autoreceptor, and the gene was successively cloned in 1999 by Lovenberg *et al.* (7). Based on tissue distribution analysis, it has been proved that the expression of the receptor is predominantly restricted to the brain (8). The H₃ receptor does not only mediate the inhibition of synthesis and release of histamine from histaminergic neurons via a negative feedback loop (9,10), but also exerts modulatory effects on other neurotransmitter systems, for example, the cholinergic (11,12), dopaminergic (13), noradrenergic (14) and serotonergic (15) systems, in both the central and peripheral nervous system. A variety of potential therapeutic application for H₃ receptor antagonists/inverse agonists has been proposed to be potential drugs for the treatment of several CNS disorders, such as attention-deficit hyperactivity disorder (ADHD) (16,17), Alzheimer's disease (18), epilepsy (19), schizophrenia (20) and obesity (21). However, emerging novel therapeutic concepts have been introduced and some indication in the H₃ receptor field, for example, migraine, pain or allergic rhinitis, might take advantage of peripherally acting ligands. For example, kojic acid analogues of benzyl-1-(4-(3-(piperidine-1-yl)propoxy)phenyl)-methanamine derivatives might act peripherally (22). Another approach for the treatment of allergic rhinitis is the search of new dual H₁H₃ receptor antagonists (23). Recently, scientists at the GlaxoSmithKlein Medicines Research Centre have published dual H₁H₃ receptor antagonists based on phthalazinone core (24).

The physiological and pathophysiological implications of histamine H₃ receptors increase the need for potent and

selective ligands as pharmacological tools and potential drugs development. The first generation of H₃ antagonists was characterized by the presence of an imidazole ring as in histamine, many of which have found utility as pharmacological tools (25,26). In contrast to the early work in the field, most chemical series of current interest appear to be non-imidazole compounds because of major disadvantages of 4-substituted imidazole moiety, including poor brain penetration and issue related to hepatic cytochrome P₄₅₀ enzymes inhibition, such as drug–drug interactions, liver toxicity and inhibition of adrenal synthesis (27–30). Additionally, non-imidazoles H₃ antagonists/inverse agonists tend to be more selective versus H₁, H₂ and H₄ receptors (31).

Since 1994, when the marine natural product aplysamine-1 was patented as a weak H₃ histamine receptor antagonist (32), diamine-based ligands, containing the characteristic aminopropoxyphenyl structural pharmacophore, have become an important chemical class of H₃ histamine receptor antagonists. This motif has been repeated in a number of different series of compounds from several laboratories, for example, compounds **1** (33) and **2** (34) (Chart 1). Later on, the successful replacement of the highly flexible propyloxy link with 4-phenoxy piperidine moiety **3** (35) (Chart 1) or the partially rigid 2-aminoethylbenzofuran substructure **4** (36) (Chart 1) was demonstrated.

Previously, we reported the synthesis and preliminary pharmacological investigation of new series of 1-benzyl-

4-(3-aminopropoxy)- and 1-benzyl-4-(5-amino)pentoxypiperidine derivatives (37). It appeared that by comparison of homologous pairs, the 1-benzyl-4-(5-aminopentoxy)piperidines have slightly higher potency than their 1-benzyl-4-(3-aminopropoxy)piperidines analogues. 1-Benzyl-4-[5-(*N*-methyl-*N*-propylaminopentoxy)]piperidine was the most potent compound of these series ($pA_2 = 7.79$) and was chosen as the lead compound for further structural modification. As the H₃ receptor antagonist potency could not be increased by modifying the *N*-substituent of 1-benzyl-4-[5-(*N*-methyl-*N*-substitutedaminopentoxy)]piperidine, attention was paid to replacement benzene ring by various substituents consist of benzene fused six-, five-membered heterocycles containing carbonyl group and oxygen or only oxygen atom and benzene fused six-, five-membered hydrocarbons rings like 3,4-dihydro-4-oxo-2H-chromen-2-yl- and -3-yl-, 4-oxo-2H-chromen-2-yl- and -3-yl-, 2- and 3-benzofuranyl, 2-indenyl or 2-naphthyl one. With the aim of investigation the effect of increased flexibility of the central core on antagonistic potency, the 4-hydroxypiperidine was replaced by 3-(methylamino)propyloxy chain, and two of the most active analogues of both 4-hydroxypiperidine series, that is, benzyl (**6a**) and 2-benzofuranyl (**6b**) derivative were synthesized and pharmacological evaluated. In addition, structure–activity relationships (SAR) resulting from variation of the position of 4-[5-(*N*-methyl-*N*-propylaminopentoxy)]piperidinyl moiety attached by methylene linker in benzene fused six-, five-membered heterocyclic and the presence or absence of double bond at heterocycle ring are discussed.

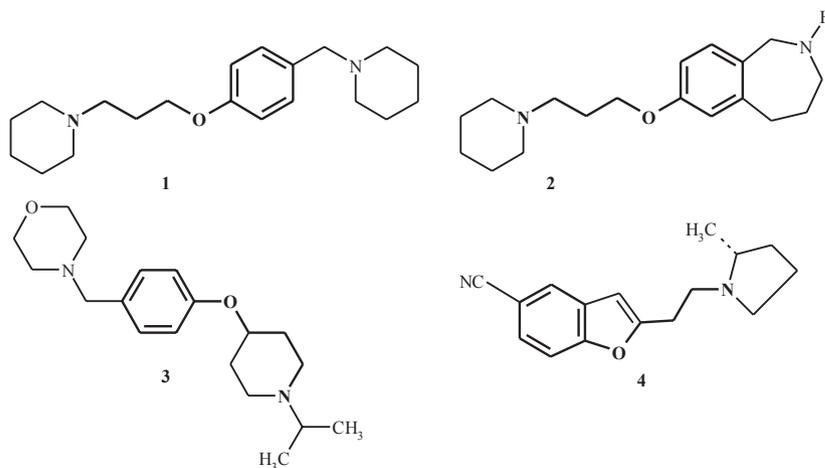
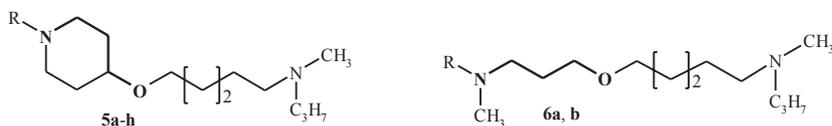


Chart 1: Representative non-imidazole H₃-histamine receptor antagonists containing the characteristic aminopropoxyphenyl structural pharmacophore, its rigid analogues and the target molecules of this study.

The target molecules of this study



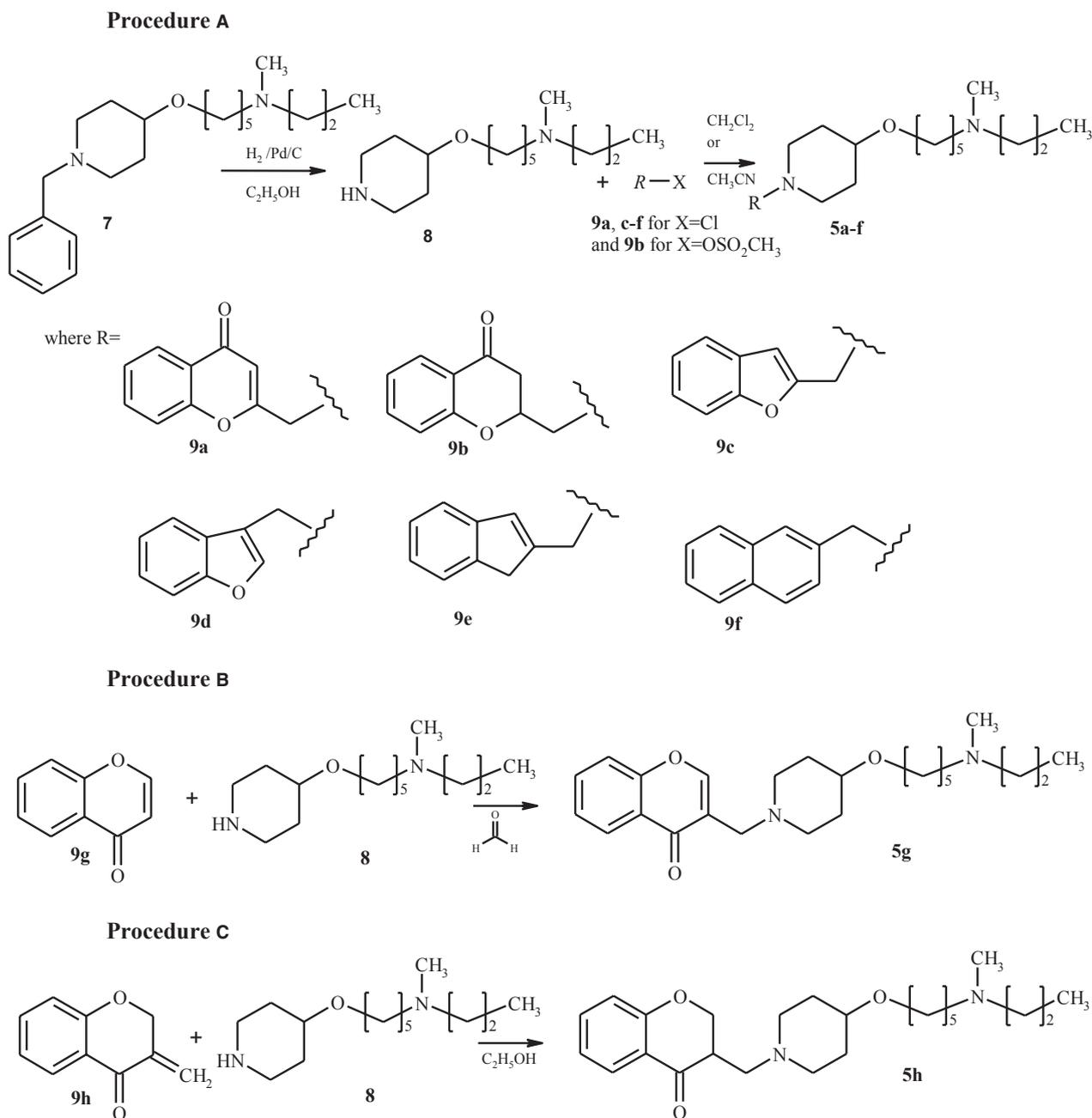
In the present work, we report the synthesis and preliminary pharmacological investigation [functionally on *in vitro* test system using guinea-pig jejunum preparations (38)] of new series of 1-[(substitutedmethyl)]-4-[5-(*N*-methyl-*N*-propyl)pentoxy]piperidines and selected 1-[(*N*-substituted-*N*-methyl)-3-propyloxy]-5-(*N*-methyl-*N*-propyl)pentandiamines as H₃ histamine receptor antagonists.

Furthermore, to determine whether the most active compounds of both series are or are not dual H₁H₃ receptor antagonists, compounds **5c**, **5d** and **6a** have been tested

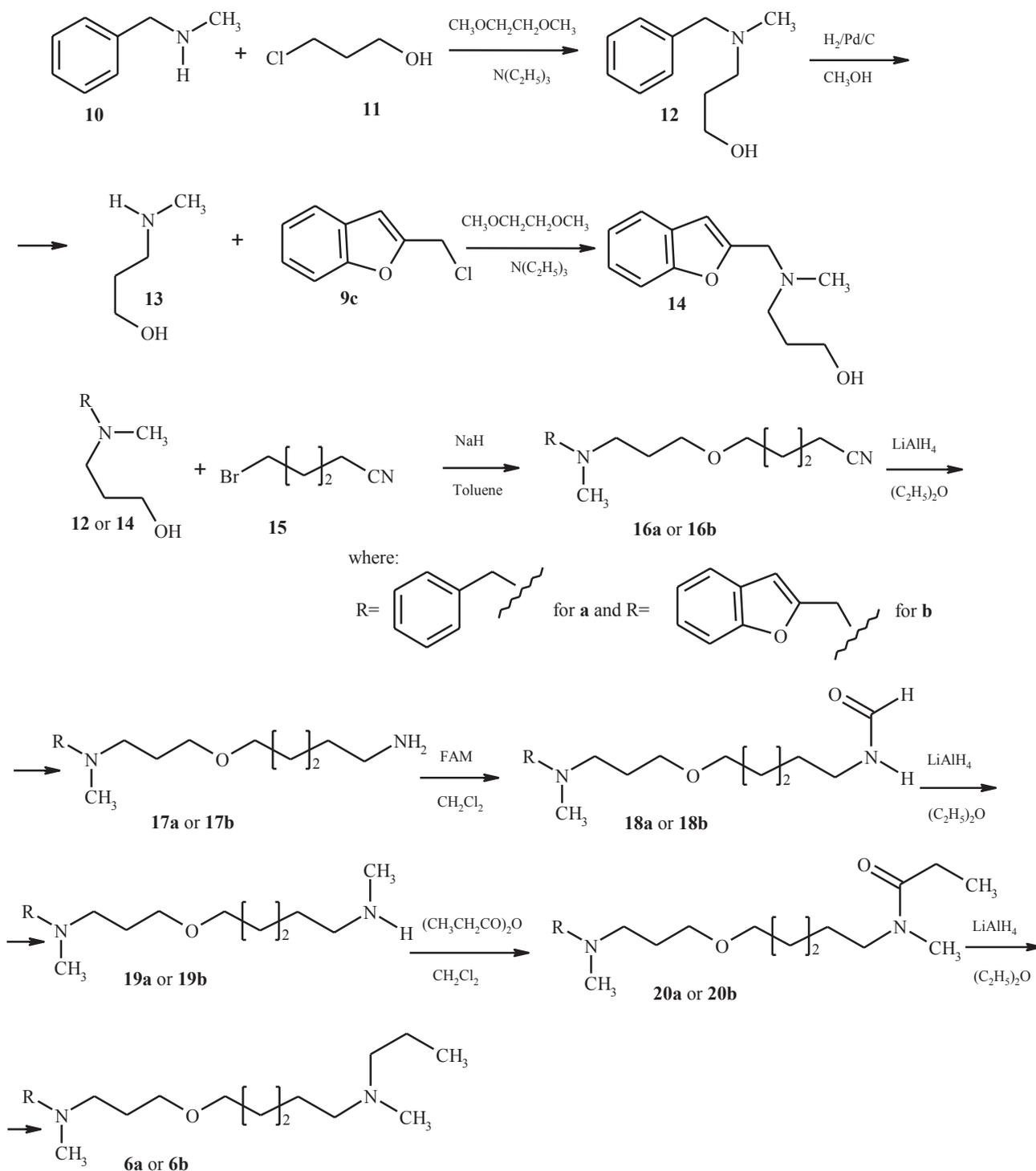
for H₁ antagonistic effects *in vitro*, following standard methods, using the guinea-pig ileum (39).

Chemistry

The general synthetic procedures used in this study are illustrated in Schemes 1 and 2. The key intermediate for all novel synthesized 4-hydroxypiperidines **5a-h** is 4-[5-(*N*-methyl-*N*-propyl)pentoxy]piperidine **8**, which was prepared by hydrogenation of 1-benzyl-4-[5-(*N*-methyl-*N*-pro-



Scheme 1: Synthesis of 1-substitutedmethyl-4-[5-(*N*-methyl-*N*-propylamino)]piperidines **5a-h**.



Scheme 2: Synthesis of 1-[N-substituted-N-methyl]-3-propyloxy]-5-(N-methyl-N-propyl)-pentanediamines **6a** and **6b**.

pyl)pentyl]piperidine **7** with a catalytic amount of palladium on charcoal in ethanol.

The 4-hydroxypiperidines **5a–f** (Scheme 1; Procedure A) were obtained from **8** by alkylation with the corresponding mesyl chloride **9b** and chlorides **9a, c–f** in dichlorome-

thane or acetonitrile followed by purification with column chromatography. The product **5g** (Scheme 1; Procedure B) was synthesized by Mannich reaction involving the condensation of 1,4-benzopyrone **9g** with formaldehyde and **8** in ethanol. The compound **9h** in ethanol solution was treated with **8** to yield **5h**.

The 3-(methylamino)propyloxy derivatives **6a**, **b** were obtained from compound **12** or **14** by a six-step synthesis (Scheme 2) including the following: O-alkylation with 5-bromopentanenitrile in dry toluene in the presence of sodium hydride and 1,4,7,10,13-pentaoxycyclopentadecane (15-crown-5 ether) to compounds **16a** or **16b**, reduction with LiAlH_4 in dry ethyl ether to compound **17a** or **17b**, formylation with formic acid-acetic anhydride (FAM) to yield **18a**, **b**, reduction with LiAlH_4 in dry diethyl ether to obtain **19a** or **19b**, acetylated with propionic anhydride to amides **20a** or **20b** and reduction with LiAlH_4 in dry diethyl ether to the desired compounds **6a** or **6b**, each step was followed by purification with column chromatography. 3-Benzylmethylamino-1-propanol **10** was N-alkylated with **11** in 1,2-dimethoxyethane (DME) in the presence of triethylamine (TEA) to 3-(N-benzyl-N-methylamino)-1-propanol **12** (40). Hydrogenation of (**12**) with a catalytic amount of palladium on charcoal in methanol yield to **13** (40). The compound **14** was prepared from 2-chloromethylbenzofuran **9c** through nucleophilic substitution of the chlorine atom by 3-methylamino-1-propanol **13** in DME in the presence of TEA. All obtained final free bases were treated with methanolic oxalic acid solution, and oxalic acid salts were precipitated with dry diethyl ether and crystallized twice from ethanol.

Experimental Section

General methods

All melting points (mp) were taken in open capillaries on an electrothermal apparatus and are uncorrected. For all compounds, ^1H NMR spectra were recorded on Varian Gemini 200 MHz, Varian Mercury VX 300 MHz. Chemical shifts are expressed in p.p.m. downfield from internal TMS as reference. ^1H NMR data are reported in order: multiplicity (br, broad; s, singlet; d, doublet; t, triplet; m, multiplet; *, exchangeable by D_2O) number of protons and approximate coupling constant in Hertz. Elemental analysis (C, H, N) for all compounds was measured on Perkin Elmer Series II CHNS/O Analyzer 2400 (PerkinElmer, Inc., Waltham, MA, USA) and are within $\pm 0.4\%$ of the theoretical values. TLC was performed on silica gel 60 F_{254} plates (Merck, Darmstadt, Germany). Flash column chromatography was carried out using silica gel 60 Å 50 μm (J. T. Baker B. V.), employing the same eluent as was indicated by TLC. All obtained final free bases were treated with methanolic oxalic acid solution, and oxalic acid salts were precipitated with dry diethyl ether and crystallized twice from ethanol.

Description of the synthetic methods for preparing intermediates **9a–e** and **9h** has been given in Appendix S1 also graphic illustrations of synthetic pathways for these intermediates (Schemes 3 and 4) have been presented in supporting information section.

The 1,4-benzopyrone, 5-bromopentanenitrile, propionic anhydride, 3-benzylmethylamino-1-propanol, 3-chloro-1-propanol, 4-oxo-4H-1-benzopyran-2-carboxylic acid, 2-

hydroxy-benzaldehyde, 3-bromo-1-propene, 2H-1-Benzopyran-2-one, 2-hydroxyacetophenone, ethyl chloroacetate and indan-1-one were all purchased from commercial source - (Sigma-Aldrich Corp, St. Louis, MO, USA).

Synthesis of 4-[5-(N-methyl-N-propyl)pentyl]piperidine **8**

A mixture of **7** (0.007 mol) and Pd/C (10%; 70 mg) in 100 mL of ethanol was shaken in autoclave under 3 atm. of hydrogen at 50 °C temperature for 5 h. The catalyst was filtered off, and the solvent was removed in vacuo. The residue was purified by column chromatography with ethyl acetate:methanol:TEA: 9:4:1 to give **8** as colourless oil.

8. $\text{C}_{14}\text{H}_{30}\text{N}_2$, (M = 242.4); yield 98.0%; ^1H NMR(CDCl_3) δ p.p.m.: 0.863–0.912 (t, 3H, CH_2CH_3 , J = 7.33 Hz); 1.315–1.635 (m, 11H, $^{\text{alif}}\text{CH}_2$, $^{\text{pip}}\text{CH}_2$, NH); 1.84–1.88 (m, 2H, $^{\text{pip}}\text{CH}_2$); 2.077–2.17 (m, 2H, $^{\text{pip}}\text{CH}_2$); 2.2 (s, 3H, NCH_3); 2.246–2.335 (m, 4H, $^{\text{alif}}\text{CH}_2$); 2.722–2.759 (m, 2H, $^{\text{pip}}\text{CH}_2$); 3.223–3.306 (m, 1H, $^{\text{pip}}\text{CHO}$); 3.394–3.438 (t, 2H, $^{\text{alif}}\text{CH}_2\text{O}$, J = 6.6 Hz); 3.485 (s, 2H, CH_2Ph); R_f = 0.17.

Elemental analysis for dioxalic acid salt $\text{C}_{14}\text{H}_{30}\text{N}_2 \cdot 2\text{C}_2\text{H}_2\text{O}_4$ (M = 422,48); mp_{dioxalic acid salt} = 110–111 °C.

	C (%)	H (%)	N (%)
Calculated	51.17	8.11	6.63
Found	51.07	8.45	6.54

General method for the preparation of 1-substitutedmethyl-4-[5-(N-methyl-N-propylamino)pentyl]piperidines **5a–f**

To a solution of 4-[5-(N-methyl-N-propyl)pentyl]piperidine **8** (0.0024 mol) in 60 mL of dichloromethane (for **5a**) or in 120 mL of acetonitrile (for **5b**) was added the corresponding chloride **9a** or mesyloxide **9b** (0.0012 mol). In case of compounds **5c–f** to solution of compound **8** (0.0011 mol) in 10 mL of acetonitrile, in the presence of potassium carbonate (0.0011 mol), was added the corresponding chloride **9c–f** (0.0011 mol). The mixture was stirred at room temperature for 72–96 h and next 24 h at 40 °C. The solvent was evaporated and residue was washed with saturated aqueous solution of potassium carbonate and extracted with ethyl acetate (3 × 50.0 mL). The organic extracts were dried (Na_2SO_4) and filtered. The solvent was evaporated to give the crude products as a sticky oil, which was purified by column chromatography.

5a. $\text{C}_{24}\text{H}_{36}\text{N}_2\text{O}_3$, (M = 400.52); yield 24%; ^1H NMR(CDCl_3) δ p.p.m.: 0.87–0.89 (t, 3H, CH_2CH_3 , J = 7.35 Hz); 1.31–1.36 (m, 2H, $^{\text{alif}}\text{CH}_2$); 1.46–1.54 (m, 4H, $^{\text{alif}}\text{CH}_2$); 1.55–1.59 (m, 2H, $^{\text{alif}}\text{CH}_2$); 1.61–1.67 (m, 2H, $^{\text{pip}}\text{CH}_2$); 1.86–1.89 (m, 2H, $^{\text{pip}}\text{CH}_2$); 2.23 (s, 3H, NCH_3); 2.31–2.37 (m, 6H, $^{\text{alif}}\text{CH}_2$, $^{\text{pip}}\text{CH}_2$); 2.79–2.82 (m, 2H, $^{\text{pip}}\text{CH}_2$); 3.28–3.31 (m, 1H,



^{pip}CHO); 3.39–3.42 (t, 2H, OCH₂, J = 6.6 Hz); 3.46 (s, 2H, NCH₂); 6.41 (s, 1H, CO-CH); 7.29–7.38 (m, 1H, ^{arom}CH); 7.4–7.45 (m, 1H, ^{arom}CH); 7.62–7.64 (m, 1H, ^{arom}CH); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 89:10:1) *R_f* = 0.37.

Elemental analysis for dioxalic acid salt C₂₄H₃₆N₂O₃ · 2C₂H₂O₄ (M = 580.06); mp_{dioxalic acid salt} = 155–158 °C.

	C (%)	H (%)	N (%)
Calculated	57.92	6.94	4.83
Found	58.10	7.06	4.85

5b. C₂₄H₃₈N₂O₃, (M = 402.57); yield 74%; ¹H NMR(CDCl₃) δ p.p.m.: 0.88–0.90 (t, 3H, CH₂CH₃, J = 7.5 Hz); 1.32–1.38 (m, 2H, ^{alif}CH₂); 1.46–1.52 (m, 4H, ^{alif}CH₂); 1.56–1.63 (m, 4H, ^{alif}CH₂, ^{pip}CH₂); 1.84–1.86 (m, 2H, ^{pip}CH₂); 2.22 (s, 3H, NCH₃); 2.27–2.35 (m, 6H, ^{alif}CH₂, ^{pip}CH₂); 2.72–2.77 (m, 2H, CO-CH₂); 2.78–2.83 (m, 4H, ^{alif}CH₂, ^{pip}CH₂); 3.26–3.29 (m, 1H, ^{pip}CHO); 3.41–3.43 (t, 2H, OCH₂, J = 6.6 Hz); 4.58–4.62 (m, 1H, OCHCH₂); 6.97–7.01 (m, 2H, ^{arom}CH); 7.44–7.47 (m, 1H, ^{arom}CH); 7.87–7.88 (m, 1H, ^{arom}CH); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 39:10:1) *R_f* = 0.39.

Elemental analysis for dioxalic acid salt C₂₄H₃₈N₂O₃ · 2C₂H₂O₄ (M = 582.65); mp_{dioxalic acid salt} = 135–137 °C.

	C (%)	H (%)	N (%)
Calculated	57.72	7.27	4.81
Found	57.80	7.34	4.74

5c. C₂₃H₃₆N₂O₂, (M = 372.51); yield 41.48%; ¹H NMR, (CDCl₃) δ p.p.m.: 0.90–0.95 (t, 3H, J = 7.35, CH₂CH₃); 1.31–1.76 (m, 10H, ^{alif}CH₂, ^{pip}CH₂); 1.90–1.96 (m, 2H, ^{pip}CH₂); 2.24 (s, 3H, NCH₃); 2.26–2.37 (m, 6H, ^{alif}CH₂, ^{pip}CH₂); 2.83–2.87 (m, 2H, ^{pip}CH₂); 3.28–3.36 (m, 1H, ^{pip}CHO); 3.43–3.47 (t, 2H, J = 6.6, OCH₂); 3.72 (s, 2H, NCH₂); 6.61 (s, 1H, 3-H); 7.21–7.32 (m, 2H, ^{arom}CH); 7.50–7.58 (m, 2H, ^{arom}CH); TLC (ethyl acetate:methanol:TEA: 9:1:4) *R_f* = 0.27.

Elemental analysis for dioxalic acid salt C₂₃H₃₆N₂O₂ · 2C₂H₂O₄ (M = 554.64); mp_{dioxalic acid salt} = 168–169 °C.

	C (%)	H (%)	N (%)
Calculated	58.68	7.30	5.07
Found	58.46	7.33	5.04

5d. C₂₃H₃₆N₂O₂, (M = 372.51); yield 53.68%; ¹H NMR, (CDCl₃) δ p.p.m.: 0.86–0.91 (t, 3H, J = 7.5, CH₂CH₃); 1.24–1.65 (m, 10H, ^{alif}CH₂, ^{pip}CH₂); 1.86–1.90 (m, 2H,

^{pip}CH₂); 2.13–2.17 (m, 2H, ^{pip}CH₂); 2.21 (s, 3H, NCH₃); 2.26–2.35 (m, 4H, ^{alif}CH₂); 2.79–2.83 (m, 2H, ^{pip}CH₂); 3.22–3.29 (m, 1H, ^{pip}CHO); 3.39–3.43 (t, 2H, J = 6.6, OCH₂); 3.63 (s, 2H, NCH₂); 7.20–7.32 (m, 2H, ^{arom}CH); 7.44–7.48 (m, 1H, ^{arom}CH); 7.52 (s, 1H, 2-H); 7.70–7.73 (m, 1H, ^{arom}CH); TLC (ethyl acetate:methanol:concentrated ammonium hydroxide: 9:1:1) *R_f* = 0.29.

Elemental analysis for dioxalic acid salt C₂₃H₃₆N₂O₂ · 2C₂H₂O₄ (M = 554.64); mp_{dioxalic acid salt} = 155–157 °C.

	C (%)	H (%)	N (%)
Calculated	58.68	7.30	5.07
Found	58.53	7.42	4.93

5e. C₂₄H₃₈N₂O, (M = 370.54); yield 29%; ¹H NMR, CDCl₃, δ p.p.m.: 0.88–0.93 (t, J = 7.2, 3H, CH₂CH₃); 1.30–1.40 (m, 2H, CH₂); 1.45–1.66 (m, 8H, CH₂); 1.86–1.91 (m, 2H, CH₂^{pip}); 2.09–2.16 (m, 2H, CH₂^{pip}); 2.25 (s, 3H, NCH₃); 2.31–2.40 (m, 4H, CH₂); 2.75–2.79 (m, 2H, CH₂^{pip}); 3.24–3.33 (m, 1H, CHO^{pip}); 3.37–3.45 (m, 6H, OCH₂^{alif}, =C-CH₂-N, CH₂^{inden}); 6.67 (s, 1H, CH=C^{inden}); 7.11–7.16 (m, 1H, CH^{arom}); 7.21–7.31 (m, 2H, CH^{arom}); 7.4–7.42 (m, 1H, CH^{arom}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 18:1:0.5) *R_f* = 0.31.

Elemental analysis for dioxalic acid salt C₂₄H₃₈N₂O · 2C₂H₂O₄ (M = 554.58); mp_{dioxalic acid salt} = 169–172 °C.

	C (%)	H (%)	N (%)
Calculated	60.64	7.65	5.05
Found	60.47	7.36	4.89

5f. C₂₅H₃₈N₂O, (M = 382.65); yield 21%; ¹H NMR, CDCl₃, δ p.p.m.: 0.86–0.91 (t, 3H, J = 7.5, CH₂CH₃); 1.31–1.67 (m, 10H, CH₂^{alif}, CH₂^{pip}); 1.85–1.92 (m, 2H, CH₂^{pip}); 2.13–2.17 (m, 2H, CH₂^{pip}); 2.21 (s, 3H, NCH₃); 2.26–2.35 (m, 4H, CH₂^{alif}); 2.75–2.82 (m, 2H, CH₂^{pip}); 3.27–3.30 (m, 1H, CHO^{pip}); 3.40–3.44 (t, 2H, J = 6.6, OCH₂); 3.65 (s, 2H); 7.41–7.50 (m, 3H, CH^{arom}); 7.72 (s, 1H, CH^{arom}); 7.79–7.83 (m, 3H, CH^{arom}); TLC (eluents:dichloromethane:ethyl acetate and dichloromethane:methanol:concentrated ammonium hydroxide – gradient from 4:1 to 9:1:1) *R_f* = 0.47.

Elemental analysis for dioxalic acid salt C₂₄H₃₈N₂O · 2C₂H₂O₄ (M = 562.73); mp_{dioxalic acid salt} = 161–162 °C.

	C (%)	H (%)	N (%)
Calculated	61.9	7.53	4.98
Found	61.57	7.84	4.71

Synthesis of 1-(4-oxo-2H-chromen-3-ylmethyl)-4-[5-(*N*-methyl-*N*-propyl)pentyl-oxo]-piperidine **5g**

A mixture of 1,4-benzopyrone **9g** (0.045 mol), 4-[5-(*N*-methyl-*N*-propyl)pentyl-oxo]piperidine **8** (0.05 mol), paraformaldehyde (0.09 mol) and 40 mL of absolute ethanol was refluxed for 24 h. The solvent was evaporated to give the crude product as a sticky oil, which was purified by column chromatography.

5g. C₂₄H₃₆N₂O₃, (M = 400.54); yield 27%; ¹H NMR (CDCl₃): δ p.p.m.: 0.87–0.92 (t, 3H, -CH₂CH₃, J = 7.35 Hz); δ = 1.26–1.45 (m, 2H^{pip}), δ = 1.46–1.67 (m, 4H, -CH₂-^{alif}); δ = 1.88–1.94 (m, 2H^{pip}); δ = 2.25 (s, 3H, N-CH₃), δ = 2.31–2.4 (m, 4H, -CH₂-), δ = 2.81–2.85 (m, 2H^{pip}); δ = 3.25–3.34 (m, 1H, -CHO), δ = 3.40–3.45 (t, -CH₂-^{alif}, J = 6.45); δ = 3.49 (s, 2H, -CH₂-); δ = 7.37–7.47 (m, 2H^{chromone}); δ = 7.63–7.69 (m, 1H^{chromone}); δ = 8.02 (s, 1H, OCH = ^{chromone}); δ = 8.21–8.24 (m, 1H^{chromone}); TLC (ethyl acetate:methanol:TEA: 9:1:4), R_f = 0.49.

Elemental analysis for dioxalic acid salt C₂₄H₃₆N₂O₃ · 2C₂H₂O₄ (M = 580.64); mp_{dioxalic acid salt} = 165–167 °C.

	C (%)	H (%)	N (%)
Calculated	57.91	6.93	4.82
Found	57.87	6.88	4.85

Synthesis of 1-(3,4-dihydro-4-oxo-2H-chromen-3-ylmethyl)-4-[5-(*N*-methyl-*N*-propyl)-pentyl-oxo]-piperidine **5h**

To a solution of 2,3-dihydro-3-methylene-4H-benzopyran-4-on **9h** (0.0033 mol) in 15 mL of ethanol was added **8** (0.0033 mol). The mixture was stirred for 168 h at room temperature. The solvent was evaporated, and residue was purified by column chromatography. The title products were obtained as sticky oil.

5h. C₂₄H₃₈N₂O₃, (M = 402.57); yield 30%; ¹H NMR δ p.p.m.: 0.83–0.98 (t, 3H, J = 7.35, CH₂CH₃); 1.29–1.65 (m, 8H, ^{pip}CH₂, ^{alif}CH₂); 1.85–1.99 (m, 2H, ^{pip}CH₂); 2.03–2.15 (m, 2H, ^{pip}CH₂); 2.22 (s, 3H, NCH₃); 2.27–2.47 (m, 4H, ^{alif}CH₂, ^{pip}CH₂); 2.4–2.96 (m, 6H, ^{alif}CH₂, ^{pip}NCH₂); 3.23–3.25 (m, 1H, ^{pip}CH); 3.4–3.45 (t, 2H, ^{alif}OCH₂, J = 6.6 Hz); 4.36–4.43 (dd, 2H, J = 8.7 Hz, J = 2.7 Hz, ^{chrom}OCH₂); 4.57–4.62 (dd, 1H, J = 4.5 Hz, J = 7.2 Hz, ^{chrom}CH); 6.95–7.03 (m, 2H, ^{arom}CH); 7.44–7.51 (m, 1H, ^{arom}CH); 7.86–7.89 (dd, 1H, J = 1.8 Hz, J = 6 Hz, ^{arom}CH); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 19:1:0.5), R_f = 0.4.

Elemental analysis for dioxalic acid salt C₂₄H₃₈N₂O₃ · 2C₂H₂O₄ (M = 582.65); mp_{dioxalic acid salt} = 129–132 °C.

	C (%)	H (%)	N (%)
Calculated	57.72	7.27	4.81
Found	58.09	7.09	4.74

General method for the preparation of 1-[(*N*-substituted-*N*-methyl)-3-propyl-oxo]-5-(*N*-methyl-*N*-propyl)pentanediamines **6a** and **6b**

Synthesis of 3-(*N*-benzyl-*N*-methylamino)-1-propanol **12**

A mixture of 3-benzylmethylamine **10** (0.2 mol), 3-chloro-1-propanol **11** (0.2 mol), triethylamine (0.2 mol) and 1,2-dimethoxyethane (150 mL) was heated at 93 °C for 20 h. The reaction mixture was cooled, and the trimethylamine hydrochloride was filtered off. The solvent was evaporated, and the residue was distilled under reduced pressure to give **12** as colourless oil.

12. C₁₁H₁₇NO, (M = 179.29); yield 51% bp = 152–156 °C/15 torr [lit. bp = 98–100 °C/0.25 torr [40]]; ¹H NMR (CDCl₃) δ p.p.m.: 1.74–1.77 (m, 2H, CH₂CH₂CH₂); 2.22 (s, 3H, NCH₃); 2.61–2.64 (t, J = 6 Hz, 2H, CH₂N); 3.51 (s, 2H, CH₂Ph); 3.74–3.78 (t, J = 6 Hz, 2H, CH₂OH); 4.50 (br, 1H, OH); 7.27–7.32 (m, 5H, H_{aromat.}).

Synthesis of 3-(methylamino)-1-propanol **13.** A mixture of **12** (0.1 mol) and Pd/C (10%; 110 mg) in 100 mL of ethanol was shaken in autoclave under 20 atm. of hydrogen at 60 °C temperature for 4 h. The catalyst was filtered off, and the solvent was removed in vacuo. The residue was distilled under reduced pressure to give **13** as colourless oil.

13. C₄H₁₁NO, (M = 89.16); yield 61.0%; bp = 80–82 °C/13 torr [lit.^o bp = 74–76 °C/10 torr [40]] ¹H NMR(CDCl₃) δ p.p.m.: 1.54–1.58 (m, 2H, CH₂CH₂CH₂); 2.27 (s, 3H, NCH₃); 2.61–2.63 (t, J = 6 Hz, 2H, CH₂N); 3.24 (br, 1H, OH); 3.56–3.58 (t, J = 6 Hz, 2H, CH₂OH).

Synthesis of 3-[(benzofuran-2yl)methyl]amino]-1-propanol **14**

To a solution of 3-(methylamino)-1-propanol **13** (0.022 mol) and TEA (0.022 mol) in 50 mL of DME was added 2-chloromethylbenzofuran **9c** (0.022 mol). The reaction mixture was heated at 80 °C for 24 h. After cooling, the resulting precipitate was filtered off and solvent was evaporated to give the crude product as a sticky oil, which was purified by column chromatography.

14. C₁₃H₁₇NO₂, (M = 219.31); yield 57%; ¹H NMR (CDCl₃) δ p.p.m.: 1.74–1.77 (m, 2H, CH₂CH₂CH₂); 2.34 (s, 3H, NCH₃); 2.66–2.70 (t, J = 6 Hz, 2H, CH₂N); 3.72 (s, 2H, CH₂^{benzofurane}); 3.76–3.0 (t, J = 6 Hz, 2H, CH₂OH); 4.77 (br, 1H, OH); 6.59 (s, 1H, H^{benzofurane}); 7.17–7.27); 7.45–7.54 (m, 1H, H^{benzofurane}); TLC (ethyl acetate:methanol:TEA: 93:1:1), R_f = 0.22.

Synthesis of 1-[(N-substituted-N-methylamino)-3-propyloxy]-5-pentanenitriles 16a and 16b. To a solution of the corresponding 3-(N-substituted-N-methylamino)-1-propanol **12** or **14** (0.009 mol) in 100 mL of dry toluene was added sodium hydride (0.018 mol). The resultant suspension was stirred at room temperature for 1 h and then was added dropwise 15-crown-5 ether (0.054 mol) and then treated in a single portion with 5-bromopentanenitrile (0.054 mol). The reaction mixture was stirred at room temperature for 72 h, and excess of sodium hydride was quenched by dropwise addition of ethanol (30 mL). The solvent was evaporated under reduce pressure, and water (50 mL) was added. The mixture was extracted with dichloromethane (3 × 50.0 mL), and organic layer was dried (Na₂SO₄) and filtered. The solvent was evaporated, and residue was purified by column chromatography. The title products were obtained as sticky oil.

16a. C₁₆H₂₄N₂O, (M = 260.42); yield 50%; ¹HNMR, CDCl₃, δ: 1.66–1.82 (m, 6H, CH₂^{alif}); 2.19 (s, 3H, NCH₃); 2.33–2.38 (t, 2H, J = 7.2, OCH₂); 2.41–2.46 (t, 2H, J = 7.2, OCH₂); 3.40–3.47 (m, 4H, CH₂^{alif}); 3.48 (s, 2H, PhCH₂); 7.22–7.34 (m, 5H, CH^{arom}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 29:10:1), R_f = 0.29.

16b. C₁₈H₂₄N₂O₂, (M = 300.44); yield 42%; ¹HNMR, CDCl₃, δ: 1.61–1.71 (m, 4H, CH₂^{alif}); 1.76–1.85 (m, 2H, CH₂^{alif}); 2.27–2.32 (m, 2H, CH₂CN); 2.34 (s, 3H, NCH₃); 2.50–2.55 (t, 2H, J = 7.5, NCH₂); 3.36–3.40 (t, 2H, J = 5.7, OCH₂); 3.42–3.46 (t, 2H, J = 6.3, OCH₂); 3.71 (s, 2H, benzofuran-CH₂-N); 6.59 (s, 1H, 3-H^{benzofurane}); 7.17–7.28 (m, 2H, CH^{benzofurane}); 7.45–7.54 (m, 2H, CH^{benzofurane}); TLC (ethyl acetate:methanol:TEA: 39:1:1), R_f = 0.49.

Synthesis of 1-[(N-substituted-N-methylamino)-3-propyloxy]-5-pentanediamines 17a and 17b. To a solution of the corresponding 1-[(N-substituted-N-methylamino)-3-propyloxy]-5-pentanenitrile **16a** or **16b** (0.0046 mol) in 40 mL of anhydrous ethyl ether LiAlH₄ (0.0092 mol) was added. The mixture was stirred at room temperature for 1 h and quenched by dropwise addition of water (0.35 mL), 10% of NaOH solution (0.35 mL) and water (0.35 mL). The suspension was stirred for 30 min and filtered. The filter cake was washed with ether (2 × 30 mL). The organic layers were combined, washed with water (3 × 30 mL), dried (Na₂SO₄) and filtered. The solvent was evaporated, and residue was purified by column chromatography. The title products were obtained as sticky oil.

17a. C₁₆H₂₈N₂O, (M = 264.46); yield 87%; ¹HNMR, CDCl₃, δ: 1.32–1.61 (m, 8H, CH₂^{alif}, NH₂); 1.75–1.84 (q, 2H, J = 6.9, CH₂^{alif}); 2.18 (s, 3H, NCH₃); 2.42–2.47 (t, 2H, J = 7.5, CH₂^{alif}); 2.66–2.70 (t, 2H, J = 6.9, CH₂^{alif}); 3.37–3.41 (t, 2H, J = 6.6, OCH₂); 3.43–3.48 (t, 2H, J = 6.6, OCH₂); 3.48 (s, 2H, PHCH₂); 7.22–7.32 (m, 5H, CH^{arom});

TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 29:10:1), R_f = 0.16.

17b. C₁₈H₂₈N₂O₂, (M = 304.48); yield 42%; ¹HNMR, CDCl₃, δ: 1.21–1.59 (m, 8H, CH₂^{alif}, NH₂); 1.76–1.86 (m, 2H, CH₂^{alif}); 2.32 (s, 3H, NCH₃); 2.50–2.55 (m, 2H, CH₂^{alif}); 2.64–2.67 (t, 2H, J = 6.9, CH₂^{alif}); 3.35–3.40 (t, 2H, J = 6.6, OCH₂); 3.43–3.47 (t, 2H, J = 6.6, OCH₂); 3.69 (s, 2H, benzofuran-CH₂-N); 6.57 (s, 1H, 3-H^{benzofurane}); 7.16–7.27 (m, 2H, CH^{benzofurane}); 7.44–7.53 (m, 2H, CH^{benzofurane}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 9:1:1), R_f = 0.15.

Synthesis of 1-[(N-substituted-N-methyl)-3-propyloxy]-5-(N-formylo)-pentanediamines 18a and 18b. To a solution of the corresponding **17a** or **17b** (0.004 mol) in 25 mL of anhydrous dichloromethane was added FAM (10 mL). The mixture was stirred at 5–10 °C for 0.5 h. Then, water (50.0 mL) and ethyl acetate (50.0 mL) were added, and the mixture was neutralized with K₂CO₃, and water layer was extracted with dichloromethane (2 × 50 mL). The organic layers were combined, washed with water (3 × 50 mL), dried (Na₂SO₄), filtered and evaporated, and residue was purified by column chromatography to give the desired compounds as sticky oil.

18a. C₁₇H₂₈N₂O₂, (M = 292.47); yield 99%; ¹HNMR, CDCl₃, δ: 1.32–1.41 (m, 2H, CH₂^{alif}); 1.42–1.62 (m, 4H, CH₂^{alif}); 1.74–1.83 (q, 2H, J = 6.9, CH₂^{alif}); 2.19 (s, 3H, NCH₃); 2.42–2.47 (t, 2H, J = 7.5, CH₂^{alif}); 3.25–3.32 (m, 2H, CH₂^{alif}); 3.37–3.41 (t, 2H, J = 6.6, OCH₂); 3.43–3.47 (t, 2H, J = 6.6, OCH₂); 3.48 (s, 2H, PHCH₂); 7.21–7.34 (m, 5H, CH^{arom}); 8.13–8.14 (s, 1H, COH); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 29:10:1), R_f = 0.42.

18b. C₁₉H₂₈N₂O₃, (M = 332.67); yield 42%; ¹HNMR, CDCl₃, δ: 1.26–1.40 (m, 2H, CH₂^{alif}); 1.43–1.59 (m, 4H, CH₂^{alif}); 1.70–1.85 (m, 2H, CH₂^{alif}); 2.33 (s, 3H, NCH₃); 2.50–2.55 (t, 2H, J = 7.5, NCH₂); 3.21–3.31 (m, 2H, CH₂NCOH); 3.34–3.39 (m, 2H, OCH₂); 3.43–3.47 (t, 2H, J = 6.3, OCH₂); 3.70 (s, 2H, benzofurane-CH₂-N); 5.53 (br s, 1H, NH*); 6.58 (s, 1H, (s, 1H, 3-H^{benzofurane}); 7.17–7.28 (m, 2H, CH^{benzofurane}); 7.44–7.54 (m, 2H, CH^{benzofurane}); 8.14 (s, 1H, NCOH); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 9:1:1), R_f = 0.49.

Synthesis of 1-[(N-substituted-N-methyl)-3-propyloxy]-5-(N-methyl)pentanediamines 19a and 19b. To a solution of the appropriate **19a** or **19b** (0.004 mol) in 50 mL of anhydrous ethyl ether was added LiAlH₄ (0.008 mol). The mixture was stirred at room temperature for 1 h and quenched by dropwise addition of water (0.3 mL), 10% of NaOH solution (0.3 mL) and water (0.3 mL). The suspension was stirred for 30 min and filtered. The filter cake was washed with ether (2 × 25 mL).

The organic layers were combined, washed with water (3 × 25 mL), dried (Na₂SO₄) and filtered. The solvent was evaporated, and residue was purified by column chromatography. The title products were obtained as sticky oil.

19a. C₁₇H₃₀N₂O, (M = 279.49); yield 72%; ¹HNMR, CDCl₃, δ: 1.25–1.39 (m, 2H, CH₂^{alif}); 1.40–1.60 (m, 5H, CH₂^{alif}, NH*); 1.62–1.84 (m, 2H, CH₂^{alif}); 2.18 (s, 3H NCH₃); 2.42 (s, 3H NCH₃); 2.42–2.47 (t, 2H, J = 7.5, NCH₂); 2.54–2.58 (t, 2H, J = 7.2, NCH₂); 3.37–3.41 (t, 2H, J = 6.6, OCH₂); 3.43–3.47 (t, 2H, J = 6.6, OCH₂); 3.48 (s, 2H, PhCH₂); 7.20–7.34 (m, 5H, CH^{arom}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 29:10:1), R_f = 0.22.

19b. C₁₉H₃₀N₂O₂, (M = 318.69); yield 89%; ¹HNMR, CDCl₃, δ: 1.31–1.39 (m, 2H, CH₂^{alif}); 1.43–1.60 (m, 5H, CH₂^{alif}, NH*); 1.76–1.86 (m, 2H, CH₂^{alif}); 2.32 (s, 3H, NCH₃); 2.42 (s, 3H, NCH₃); 2.50–2.57 (m, 4H, CH₂^{alif}); 3.35–3.40 (t, 2H, J = 6.6, OCH₂); 3.43–3.47 (t, 2H, J = 6.6, OCH₂); 3.70 (s, 2H, ^{benzofurane}-CH₂-N); 6.58 (s, 1H, 3-H^{benzofurane}); 7.17–7.27 (m, 2H, CH^{benzofurane}); 7.45–7.54 (m, 2H, CH^{benzofurane}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 4:1:1), R_f = 0.22.

Synthesis of 1-[(N-substituted-N-methyl)-3-propyloxy]-5-(N-methyl-N-propionylcarbonyl)-pentanediamine amides **20a** and **20b**.

To an ice-cooled solution of the appropriate **19a** or **19b** in 70 mL of dichloromethane was slowly added propionic anhydride, and reaction mixture was stirred at room temperature for 2 h. Saturated aqueous solution of sodium bicarbonate was then added until pH 9, and the reaction mixture was stirred for 24 h. The organic layer was separated, dried (Na₂SO₄) and filtered. Dichloromethane was evaporated to give the crude products **20a** and **20b** as sticky oil, which was purified by column chromatography.

20a. C₂₀H₃₄N₂O₂, (M = 334.56); yield 93%; ¹HNMR, CDCl₃, δ: 1.11–1.18 (m, 3H, CH₂CH₃); 1.29–1.39 (m, 2H, CH₂^{alif}); 1.43–1.62 (m, 4H, CH₂^{alif}); 1.74–1.84 (m, 2H, CH₂^{alif}); 2.18–2.19 (d, 3H, J = 1.2, PhCH₂NCH₃); 2.27–2.36 (m, 2H, CH₂^{alif}); 2.42–2.47 (t, 2H, J = 7.5, CH₂^{alif}); 2.91–2.95 (d, 3H, J = 12.9, CH₃NCO); 3.33–3.47 (m, 6H, CH₂^{alif}); 3.48 (s, 2H, PhCH₂); 7.21–7.34 (m, 5H, CH^{arom}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 139:10:1), R_f = 0.67.

20b. C₂₂H₃₄N₂O₃, (M = 374.58); yield 83%; ¹HNMR, CDCl₃, δ: 1.10–1.16 (m, 3H, CH₂CH₃); 1.24–1.34 (m, 2H, CH₂^{alif}); 1.43–1.59 (m, 4H, CH₂^{alif}); 2.23–2.35 (m, 5H, NCH₃, CH₂^{alif}); 2.49–2.54 (t, 2H, J = 7.5, CH₂^{alif}); 2.89–2.93 (d, 3H, J = 9.9, CH₃NCO); 3.31–3.38 (m, 4H, CH₂^{alif}); 3.42–3.47 (m, 2H, CH₂^{alif}); 3.69 (s, 2H, benzofuran-CH₂-N); 6.57 (s, 1H, 3-H^{benzofurane}); 7.16–7.30 (m, 2H, CH^{benzofurane}); 7.44–7.53 (m, 2H, CH^{benzofurane}); TLC (dichlo-

romethane:methanol:concentrated ammonium hydroxide: 9:1:1), R_f = 0.57.

Synthesis of 1-[(N-substituted-N-methyl)-3-propyloxy]-5-(N-methyl-N-propyl)-pentanediamines **6a** and **6b**.

To a solution of the appropriate amides **20a** or **20b** (0.002 mol) in 50 mL of anhydrous ethyl ether LiAlH₄ (0.004 mol) was added. The mixture was stirred at room temperature for 2 h and quenched by dropwise addition of water (0.15 mL), 10% of NaOH solution (0.15 mL) and water (0.15 mL). The suspension was stirred for 30 min and filtered. The filter cake was washed with ether (2 × 25 mL). The combined organic layers were washed with water (3 × 25 mL), dried (Na₂SO₄) and filtered. The solvent was evaporated, and residue was purified by column chromatography. The title products were obtained as sticky oil.

6a. C₂₀H₃₆N₂O, (M = 320.58); yield 51%; ¹HNMR, CDCl₃, δ: 0.86–0.91 (t, 3H, J = 7.5, CH₂CH₃); 1.30–1.37 (m, 2H, CH₂^{alif}); 1.41–1.62 (m, 6H, CH₂^{alif}); 1.77–1.84 (q, 2H, J = 6.9, CH₂^{alif}); 2.18 (s, 3H, NCH₃); 2.20 (s, 3H, NCH₃); 2.24–2.33 (m, 4H, CH₂^{alif}); 2.42–2.47 (t, 2H, J = 7.5, CH₂^{alif}); 3.36–3.41 (t, 2H, J = 6.6, OCH₂); 3.43–3.47 (t, 2H, J = 6.6, OCH₂); 3.48 (s, 2H, PhCH₂); 7.22–7.32 (m, 5H, CH^{arom}); TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 139:10:1), R_f = 0.38.

Elemental analysis for dioxalic acid salt C₁₉H₃₆N₂O · 2C₂H₂O₄ (M = 494.66); mp_{dioxalic acid salt} = 123–125 °C.

	C (%)	H (%)	N (%)
Calculated	57.58	8.05	5.58
Found	57.42	7.91	5.66

6b. C₂₂H₃₆N₂O₂, (M = 360.58); yield 72%; ¹HNMR, CDCl₃, δ: 0.86–0.91 (t, 3H, J = 7.5, CH₂CH₃); 1.26–1.35 (m, 2H, CH₂^{alif}); 1.42–1.56 (m, 6H, CH₂^{alif}); 1.76–1.86 (q, 2H, J = 6.9, CH₂^{alif}); 2.12 (s, 3H, NCH₃); 2.26–2.34 (m, 4H, CH₂^{alif}); 2.32 (s, 3H, NCH₃); 2.50–2.55 (t, 2H, J = 7.5, CH₂^{alif}); 3.35–3.39 (t, 2H, J = 6.6, OCH₂); 3.42–3.47 (t, 2H, J = 6.6, OCH₂); 3.67 (s, 2H, ^{benzofurane}-CH₂-N); 6.57 (s, 1H, 3-H^{benzofurane}); 7.16–7.27 (m, 2H, CH^{benzofurane}); 7.44–7.53 (m, 2H, CH^{benzofurane}). TLC (dichloromethane:methanol:concentrated ammonium hydroxide: 9:1:1), R_f = 0.35.

Elemental analysis for dioxalic acid salt C₂₂H₃₆N₂O₂ · 2C₂H₂O₄ (M = 520.66); mp_{dioxalic acid salt} = 103–105 °C.

	C (%)	H (%)	N (%)
Calculated	59.97	7.76	5.38
Found	59.85	7.49	5.57



Pharmacology. The potency of all the obtained compounds was tested for H₃ antagonistic effects *in vitro* on the guinea-pig jejunum (38).

Selected compounds **5c**, **5d** and **6a** were also tested for H₁ antagonistic effects *in vitro*, following standard methods, using the guinea-pig ileum (39).

H₃ antagonistic effects *in vitro* on the guinea-pig jejunum for compounds **5a–h**, **6a,b**, **7** and **8**

Male guinea-pigs weighing 300–400 g were killed by a blow on the head. A portion of the small intestine, 20–50 cm proximal to the ileocaecal valve (jejunum), was removed and placed in Krebs buffer [composition (mM) NaCl 118; KCl 5.6; MgSO₄ 1.18; CaCl₂ 2.5; NaH₂PO₄ 1.28; NaHCO₃ 25; glucose 5.5 and indomethacin (1 × 10⁻⁶ M)]. Whole jejunum segments (2 cm) were prepared and mounted between two platinum electrodes (4 mm apart) in 20 mL Krebs buffer, continuously gassed with 95% O₂:5% CO₂ and maintained at 37 °C. Contractions were recorded isotonicly under 1.0 g tension with Hugo Sachs Hebel-Messvorsatz (TI-2)/HF-modem (Hugo Sachs Elektronik, Hugstetten, Germany) connected to a pen recorder. After equilibration for 1 h with washings every 10 min, the muscle segments were stimulated maximally between 15 and 20 Volt and continuously at a frequency of 0.1 Hz and a duration of 0.5 millisecond, with rectangular-wave electrical pulses, delivered by a Grass Stimulator S-88 (Grass Instruments Co., Quincy, MA, USA). After 30 min of stimulation, 5 min before adding (R)- α -methylhistamine, pyrilamine (1 × 10⁻⁵ M concentration in organ bath) was added, and then cumulative concentration-response curves (half-log increments) of (R)- α -methylhistamine, H₃-agonist, were recorded until no further change in response was found. Five minutes before adding the tested compounds, the pyrilamine (1 × 10⁻⁵ M concentration in organ bath) was added, and after 20 min, cumulative concentration-response curves (half-log increments) of (R)- α -methylhistamine, H₃-agonist, were recorded until no further change in response was found. Statistical analysis was carried out with the Student's *t*-test. In all test, *p* < 0.05 was considered statistically significant. The potency of an antagonist is expressed by its pA₂ value, calculated from the Schild (39) regression analysis where at least three concentrations were used. The pA₂ values were compared with the potency of thioperamide.

H₁ antagonistic activity for **5c**, **5d** and **6a** compounds

Selected compounds were tested for H₁ antagonistic effects *in vitro*, following standard methods, using the guinea-pig ileum (39).

Male guinea-pigs weighing 300–400 g were killed by a blow on the head. The ileum was excised and placed in phosphate buffer at room temperature (pH 7.4) containing (mM)

Histamine H₃-Receptor; H₃ Non-Imidazole Antagonists

NaCl (136.9); KCl (2.68); NaHPO₄ (7.19). After flushing the intraluminal contents, segments of about 2 cm long were cut and mounted for isotonic contractions in water jacked 20 mL organ baths filled with oxygenated (O₂:CO₂ = 95:5, v/v) Krebs buffer containing (mM) NaCl (117.5); KCl (5.6); MgSO₄ (1.18); CaCl₂ (2.5); NaH₂PO₄ (1.28); NaHCO₃ (25); glucose (5.5) and indomethacin (1 × 10⁻⁶ M) at 37 °C under a constant load of 0.5 g. After a 30-min equilibration period with washings every 10 min, a submaximal priming dose of histamine (1 μ M) was given and washed out (standard washing procedure: three changes of buffer during 30 min). After washing out, the antagonistic activity of given compounds was measured by recording a concentration-response curve (CRC) for histamine in the presence of the testing compounds **5c**, **5d** and **6a**, which was added 5 min before histamine. This procedure was repeated with higher concentrations of the compounds. The antagonism was of a competitive nature causing a parallel shift of the CRC. The pA₂-values were calculated according to Arunlakshana and Schild (39). The pA₂ values were compared with the potency of pyrilamine.

Results and Discussion

The potency of all synthesized compounds was preliminary *in vitro* tested as H₃ receptor antagonists – electrically evoked contraction of the guinea-pig jejunum. 1-Substitutedmethyl-4-[5-(*N*-methyl-*N*-propylamino)pentyl]piperidines (**5a–h**) and selected 1-[(*N*-substituted-*N*-methyl)-3-propyl]oxy]-5-(*N*-methyl-*N*-propyl)pentanediamines (**6a**, **6b**) all showed moderate to high antagonist activity at H₃-receptor, except of derivative **6b** showing weak potency. Results are presented in Table 1.

The 1-benzyl-4-[5-(*N*-methyl-*N*-propylaminopentyl)]piperidine **7** has been selected as the lead structure from the previously examined compounds (37) to allow a direct comparison of the changes in activity of the newly designed compounds with the replacement of the benzyl group of the *N*-piperidine moiety.

It has appeared that by comparison of homologous pairs that **5g** and **5h** have higher potency than their analogous **5a** and **5b**. The differences are observed inside of each series. While it is quite a significant difference in H₃-receptor potency between derivative **5g** (pA₂ = 7.63) – with double bond – and **5h** (pA₂ = 7.1) – without double bond in 2,3-dihydro-4H-pyran-4-one ring – only very small difference in activity between the derivative **5a** (pA₂ = 7.2) and **5b** (pA₂ = 7.11) is observed. Moreover, comparing derivatives **5a** and **5b** with unsubstituted derivative **8** (pA₂ = 7.31), it is seen that H₃ antagonistic activity is almost on the same level, independently on the presence or absence of substituent **9a** or **9b**. These results suggest that in the chromones (**5a** and **5g**) and chromanones (**5b** and **5h**) series, the heterocyclic ring should contain a double bond and 4-[5-(*N*-methyl-*N*-propyl)pentyl]piperidi-

Table 1: H₃ antagonistic activity of 1-substituted-4-[5-(*N*-methyl-*N*-propylamino)pentyl]oxy]piperidines (**5a–h**, **7**, **8**) and 1-[(*N*-substituted-*N*-methyl)-3-propyl]oxy]-5-(*N*-methyl-*N*-propyl)pentanediamines (**6a**, **6b**) as tested on the *in vitro* test system on the guinea-pig jejunum

Cpd.	Structure	The concentrations of the tested compound c ₁ (M)	pA ₂ (sem) H ₃	N (<i>caviae</i>)
5a		1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷ ; 1 × 10 ⁻⁶	7.2 (0.1)	9 (3)
5b		1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷ ; 1 × 10 ⁻⁶	7.11 (0.14)	9 (3)
5c		3 × 10 ⁻⁹ ; 1 × 10 ⁻⁸ ; 3 × 10 ⁻⁸	8.47 (0.05)	12 (4)
5d		1 × 10 ⁻⁸ ; 3 × 10 ⁻⁸ ; 1 × 10 ⁻⁷	8.15 (0.07)	12 (4)
5e		3 × 10 ⁻⁸ ; 1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷	7.76 (0.09)	12 (4)
5f		1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷ ; 1 × 10 ⁻⁶	7.04 (0.11)	9 (3)
5g		3 × 10 ⁻⁸ ; 1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷	7.63 (0.08)	9 (3)
5h		1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷ ; 1 × 10 ⁻⁶	7.1 (0.15)	13 (4)
7		3 × 10 ⁻⁸ ; 1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷	7.79 (0.06)	12 (4)
8		1 × 10 ⁻⁷ ; 3 × 10 ⁻⁷ ; 1 × 10 ⁻⁶	7.31 (0.1)	9 (3)
6a		1 × 10 ⁻⁸ ; 3 × 10 ⁻⁸ ; 1 × 10 ⁻⁷	8.06 (0.05)	12 (4)
6b		3 × 10 ⁻⁷ ; 1 × 10 ⁻⁶ ; 3 × 10 ⁻⁶	6.23 (0.12)	12 (4)
Thioiperamide		3 × 10 ⁻⁹ ; 1 × 10 ⁻⁸ ; 3 × 10 ⁻⁸	8.65 (0.07)	36 (12)

sem, standard error of the mean; N, number of different animal preparations; *caviae*, number of animals.

nyl substituent should be present at a favourable position 3.

To investigate the influence of the size of the heterocyclic ring and the presence of the carbonyl group, on the antagonistic activity on the histamine H₃ receptor, two methylbenzofuran derivatives containing 4-[5-(*N*-methyl-*N*-propylamino)pentyl]oxy]piperidiny moiety at position 2 (**5c**) and 3 (**5d**) in benzofuran ring were synthesized. The

reduction of a heterocyclic ring with a six-membered to five-membered ring and removal of the carbonyl group lead to compound **5c** (pA₂ = 8.47) and **5d** (pA₂ = 8.15) showing the highest potency for all presented 4-hydroxypiperidine derivatives. We observe that the 2-position of 4-[5-(*N*-methyl-*N*-propylamino)pentyl]oxy]piperidiny moiety in the 2-methylbenzofuran ring (**5c**) is slightly favourable for histamine H₃-receptor antagonist activity than position 3. Replacement of the benzyl substituent by 2-methylindeny



group **5e** ($pA_2 = 7.76$) reduced potency. Antagonistic activity was further reduced when the 2-methylindenyl substituent has been replaced by the most lipophilic 2-methylnaphthyl moiety **5f** ($pA_2 = 7.04$).

Surprisingly, replacement of the 4-hydroxypiperidine in the leads **7** and **5c** by a highly flexible 3-(methylamino)propyloxy chain yields compounds **6a** ($pA_2 = 8.02$) and **6b** ($pA_2 = 6.23$) with higher and lower affinity than their piperidine analogues (**7**, $pA_2 = 7.79$; **5c**, $pA_2 = 8.47$), respectively.

In conclusion, it may be noticed that the spatial configurations greatly decide on the activity of these series than the others physico-chemical parameters.

Additionally, selected compounds **5c**, **5d** and **6a** have also been tested for H_1 antagonistic effects *in vitro*, following standard methods, using the guinea-pig ileum. None of them showed any H_1 -antagonistic activity ($pA_2 < 4$; for pyrrolamine $pA_2 = 8.5$).

Due to the lack of antagonistic activity of selected compounds at H_1 receptor, we can conclude that the derivatives of **5c**, **5d** and **6a** are not double H_1H_3 receptor antagonists.

Conclusions and Future Directions

Novel, potent non-imidazole histamine H_3 receptor antagonists have been prepared possessing either a highly flexible 3-(methylamino)propyloxy chain or 4-hydroxypiperidine ring resulting in increased rigidity of these compounds. Investigation of the compounds in an *in vitro* functional assay on the guinea-pig ileum led to the identification of **5c** and **5d** as the most potent compounds *in vitro* in both series ($pA_2 = 8.47$ and $pA_2 = 8.15$, respectively). By comparing pairs of homologous compounds **5c/6b** versus **7/6a**, it was found that the **5c** ($pA_2 = 8.47$) has a higher activity than its 3-(methylamino)propyloxy analogue **6b** ($pA_2 = 6.23$), which is in contrast to a pair of homologues **7/6a**, where 3-(methylamino)propyloxy derivative **6a** ($pA_2 = 8.02$) has a higher activity than its 4-hydroxypiperidine analogue **7** ($pA_2 = 7.79$). For explanation of tentative interpretation of the different SAR for the compounds **5c/6b** versus **7/6a**, we will try to solve this problem by molecular modelling. However, further structural modifications, both in 4-hydroxypiperidine and 3-(methylamino)propyloxy series, are required to explain these unexpected results. At last, based on the SAR of all synthesized compounds, the most active derivatives will be chosen and investigated for their effects on central histaminergic neuron activity *in vivo*.

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Conflict of Interest

The authors have declared no conflict of interest.

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Supporting Information

Additional Supporting Information may be found in the online version of this article:

Appendix S1. Description of the synthetic methods for preparing intermediates **9a–e** and **9h**.