RESEARCH ARTICLE



Novel Azole-Functionalited Flouroquinolone Hybrids: Design, Conventional and Microwave Irradiated Synthesis, Evaluation as Antibacterial and Antioxidant Agents



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Abstract: *Background:* The synthesis of new hybrid molecules consisting of several heterocyclic pharmacophores namely fluoroquinolone, 1,2,4-triazole, 1,3,4-oxadiazole and piperazine was carried out by conventional and successfully optimized microwave mediated techniques.

ARTICLEHISTORY

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DOI: 10.2174/1570180814666170823163540 *Methods*: The structures of new compounds were confirmed using spectroscopic techniques. These compounds were screened for their antibacterial activity against *E. coli*, *P. aeruginosa*, *Y. pseudo-tuberculosis*, *S. aureus*, *E. faecalis*, *B. cereus* and *M. smegmatis*.

Results and Conclusion: Fourteen of these hybrids exhibited excellent antibacterial activities on the test microorganisms when compared with ampicillin with the mic values varying between 0.03- $0.25 \ \mu$ g/mL.

Keywords: Fluoroquinolone, 1,2,4-triazole, 1,3,4-oxadiazole, piperazine, antibacterial activity, antioxidant activity, microwave.

1. INTRODUCTION

The increasing community- and hospital- acquired infectious diseases caused by resistant bacteria to most classes of antibacterial drugs resulted in a pressing and urgent need for designing of new antibiotic candidates. The declaration of The European Centre for Disease Prevention and Control (ECDC) reporting "Every year, the infections caused by resistant bacteria gives rise to 25,000 deaths with a cost of over 1.5 billion Eur because of healthcare spending and labor losses in the Europa" reveals that this is a public health problem with also socio-economic loses [1-5]. Mortality in these patients infected by resistant bacteria was reported as two times higher than the patients infected by nonresistant bacteria [2]. It is clear that pharmaceutical industry today is not able to meet the urgent need for discovery of new and effective antibiotics, despite much efforts aiming to deal with resistance problem by development of new class of antibacterial agents with different mode of action. Some of the main reasons of this undesirable situation are the increasing expenses of antibacterial resistant drugs, strict legal requirements and high prices of antibiotics. In order to overcome this situation, the collaboration of all sectors including regulations, society and researchers is necessary [6]. In order to overcome the drug resistance, the mostly implemented approach aiming the development of new antibacterial agents is to improve the pharmacodynamics of the existing drugs by changing their structural units or producing new derivatives. This approach has been accepted as an attractive strategy by scientific communities since this method does not require the discovery of new antibacterial scaffolds or validation of novel biological targets, which has proven to be an extremely difficult and time consuming task [7].

More recently, the concept of hybrid molecules, which contain two or more pharmacophore groups binding together covalently in one molecular framework, has been introduced in the medicinal chemistry field. These compounds that are obtained by molecular hybridization of several pharmacophore groups, act by inhibiting two or more conventional targets simultaneously, and this multiple target strategy has resulted in the development of a number of bioactive hybrid molecules [8].

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Novel Azole-Functionalited Flouroquinolone Hybrids

DNA gyrase and topoisomerase IV are known as essential and immanent enzymes responsible for bacterial cell growth and division [9]. Fluoroquinolones, proceeding as DNA gyrase and bacterial topoisomerase IV inhibitors, which can inhibit DNA synthesis by forming complexes with DNA gyrase or topoisomerase II enzyme, were discovered as broad-spectrum antibiotics in 1980s. Since their introduction to clinical use, they were extensively applied for the treatment of infectious diseases. But unfortunately, the wide use of fluoroquinolones in clinical settings resulted in the emergence of resistant bacterial strains. The resistance problem along with the decline of the antibiotic discovery and development studies made mandatory the design of new antibiotics [8, 10, 11].

Several successful syntheses of fluoroquinolone hybrids including arylfuran [8], benzofuroxan [12], piperazine [13, 14], thiadiazole [15], triazole [16] or quinolone [17] unit have been reported, most of which were obtained by substitution on piperazine ring at C-7 position of the basic structure of quinolones.

Since some preferable properties including flexibility, convergence and economic nature, multicomponent reactions have received considerable attention in organic, combinatorial and medicinal chemistry leading to the formation of several compounds possessing elaborate biological activities [18-20]. Moreover, the growing emphasis of "green chemistry techniques" in new drug discovery protocols has increased the popularity of the scope of microwave assisted organic synthesis due to some superior properties such as mild reaction conditions, shortened reaction times, increased reaction yields, relatively less decomposition, less formation of by-products etc [21-23]. Thus, the development of new strategies including a combination of multicomponent reactions and microwave irradiation techniques without any harmful organic solvent and catalyst has become one of the most important strategies of synthetic organic chemistry.

Discovery of new agents with antioxidant properties has become another extraordinary active area of preventive medicinal chemistry [24-26] and some fluoroquinolones and related compounds were reported as antioxidant capacity in addition to antimicrobial activity [27-30]. Moreover, synthesizing some fluoroquinolone hybrid compounds with their antioxidant capacity and antibacterial activity are currently continuing in our laboratory [31, 32].

In this context, we reported here the conventional and microwave irradiated green synthesis, antimicrobial and antioxidant activity screening studies of novel fluoroquinoloneazole hybrids by the molecular hybridization strategy. The rational design of the targeted compounds was depicted in Fig. (1).

2. EXPERIMENTEL

2.1. General

All the chemicals were purchased from Fluka Chemie AG Buchs (Switzerland) and used without further purification. Microwave-assisted syntheses were carried out using monomode CEM-Discover microwave apparatus. Reactions were monitored by thin-layer chromatography (TLC) on silica gel 60 F254 aluminium sheets. Melting points of the synthesized compounds were determined in open capillaries on a Büchi B-540 melting point apparatus and are uncorrected. The mobile phase was ethyl acetate:diethyl ether (1:1), and detection was made using UV light. FT-IR spectra were recorded using a Perkin Elmer 1600 series FTIR spectrometer. ¹H NMR and ¹³C NMR spectra were registered in DMSO-d₆ on a BRUKER AVENE II 400 MHz NMR Spectrometer (400.13 MHz for ¹H and 100.62 MHz for ¹³C). The chemical shifts are given in ppm relative to Me₄Si as an internal reference, J values are given in Hz. The elemental analysis was performed on a Costech Elemental Combustion System CHNS-O elemental analyzer. All the compounds gave C, H and N analysis within $\pm 0.4\%$ of the theoretical values. The Mass spectra were obtained on a Quattro LC-MS (70 eV) Instrument.

2.1.1. Ethyl 2-[4-(4-nitrophenyl)piperazin-1-yl]acetate (2)

<u>Method 1</u>

Ethyl bromoacetate (10 mmol) was added to the solution of 1-(4-nitrophenyl) piperazine (10 mmol) in tetrahydrofuran drop wise and the mixture was stirred at room temperature in the presence of triethylamine (30 mmol) for 3 h. The solid formed was removed by filtration, and the solvent was

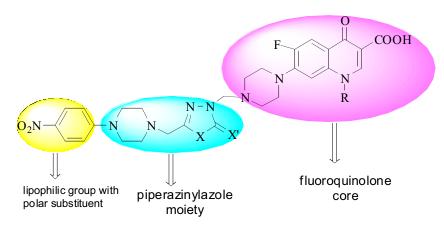


Fig. (1). The rational design of the new fluoroquinolone-azole hybrids.

evaporated under reduced pressure. The yellow solid obtained was purified by crystallization from ethyl acetate to afford the desired compound. Yield 84%.

Method 2

The mixture of ethyl bromoacetate (10 mmol), triethylamine (30 mmol) and 1-(4-nitrophenyl)piperazine (10 mmol) was irradiated in monomode microwave reactor in closed vessel with pressure control at 80 W maximum power for 8 min. The formed yellow product was washed with water and purified by crystallization from ethyl acetate. Yield 99%. mp.115-116°C.

FT-IR (ν_{max} , cm⁻¹): 3080 (ar-H), 1742 (C=O), 1313 and 1587 (NO₂), 1207 (C-O). ¹H NMR (DMSO-*d*₆, δ ppm): 1.24 (t, 3H, CH₃, *J*= 8.0 Hz), 2.69 (s, 4H, 2CH₂), 3.34 (s, 2H, CH₂), 3.50 (s, 4H, 2CH₂), 4.14 (q, 2H, CH₂, *J*= 8.0 Hz), 7.06 (d, 2H, ar-H, *J*= 8.0 Hz), 8.09 (d, 2H, ar-H, *J*=8.0 Hz). ¹³C NMR (DMSO-*d*₆, δ ppm): 14.70 (CH₃), 46.90 (2CH₂), 52.00 (2CH₂), 58.64 (CH₂), 60.48 (CH₂), arC: [113.20 (CH), 113.24 (CH), 126.28 (CH), 126.30 (CH), 137.44 (C), 155.29 (C)], 170.41 (C=O). LC-MS m/z (%): 316.12 ([M+ Na]⁺, 6), 188.16 (100). Elemental Anal. Calcd. for C₁₄H₁₉N₃O₄ (%):C, 57.33; H, 6.53; N, 14.33, Found (%): C, 57.34; H, 6.52; N, 14.33.

2.1.2. [4-(4-Nitrophenyl)piperazin-1-yl]acetohydrazide (3)

<u>Method 1</u>

Hydrazine hydrate (30 mmol) was added to the solution of compound 2 (10 mmol) in absolute ethanol and the mixture was refluxed for 5 h. On removing the solvent under reduced pressure, a solid obtained. This was recrystallized from ethanol to give the target compound. Yield 75%.

Method 2

The mixture of hydrazine hydrate (30 mmol) and compound **2** (10 mmol) was irradiated in monomode microwave reactor in closed vessel with pressure control at 120 W for 5 min. The crude product obtained was recrystallized from ethanol to give the pure compound. Yield: 97%. Mp, 163-164°C.

FT-IR (ν_{max} , cm⁻¹): 3328 and 3248 (NH, NH₂), 3006 (ar-H), 1631 (C=O), 1474 and 1322 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 2.58-2.63 (m, 4H, 2CH₂+DMSO), 3.06 (s, 2H, CH₂), 3.55 (s, 4H, 2CH₂), 4.32 (s, 2H, NH₂), 7.09 (s, 2H, ar-H), 8.11 (s, 2H, ar-H), 9.06 (s, 1H, NH). ¹³C NMR (DMSO-*d*₆, δ ppm): 46.88 (2CH₂), 52.89 (2CH₂), 60.16 (CH₂), arC: [113.22 (2CH), 126.36 (2CH), 137.39 (C), 155.31 (C)], 168.65 (C=O). LC-MS m/z (%): 303.11 ([M+1+ Na]⁺, 15), 302.11 ([M+ Na]⁺, 100). Elemental Anal. Calcd. For C₁₂H₁₇N₅O₃ (%):C, 51.60; H, 6.14; N, 25.08, Found (%): C, 51.61; H, 6.13; N, 25.08.

2.1.3. 5-{[4-(4-Nitrophenyl)piperazin-1-yl]methyl}-1,3,4oxadiazole-2-thiol (4)

<u>Method 1</u>

The solution of KOH (10 mmol) in water was added to the solution of compound **3** (10 mmol) in ethanol and the mixture was refluxed in the presence of CS_2 (20 mmol) for 10 hours. Then, it was cooled to room temperature and acidified to pH 6 with 37 % HCl. On cooling the mixture in cold overnight, a solid obtained. This was recrystallized from acetone to give the title compound **4** as a white solid. Yield: 76%.

<u>Method 2</u>

The mixture of compound **3** (10 mmol), KOH (10 mmol) and CS_2 (20 mmol) in ethanol was irradiated in the monomode microwave reactor in closed vessel at 200 W for 12 min. Then, it was acidified to pH 6 with 37 % HCl. On cooling the mixture in cold overnight, a solid obtained. This was recrystallized from acetone to give the title compound **4** as a white solid. Yield: 97%, mp. 184-186°C.

FT-IR (ν_{max} , cm⁻¹): 3373 (NH), 1587-1314 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 2.58-2.61 (m, 4H, 2CH₂), 2.70 (t, 2H, CH₂, *J*= 8.0 Hz)3.44-3.48 (m, 4H, 2CH₂), 7.00 (d, 2H, ar-H, *J*= 4.0 Hz), 8.02 (d, 2H, ar-H, *J*= 4.0 Hz) 8.04 (s, 1H, NH). ¹³C NMR (DMSO-*d*₆, δ ppm): 46.40 (CH₂), 46.65 (CH₂), 51.46 (CH₂), 51.85 (CH₂), 58.37 (CH₂), arC: [113.10 (CH), 113.16 (CH), 126.14 (2CH), 137.40 (C), 155.05 (C)], 161.05 (oxadiazole C-5), 178.51 (oxadiazole C-2). LC-MS m/z (%): 322.24 ([M+1]⁺, 10), 294.34 (100). Elemental Anal. Calcd. for C₁₃H₁₅N₅O₃S (%): C, 48.59; H, 4.70; N, 21.79, Found (%): C, 48.59; H, 4.70; N, 21.78.

2.1.4. General Method for The Synthesis of Compounds 5a-b

Method 1

The solution of ciprofloxacin (for **5b**) or norfloxacin (for **5a**) (10 mmol) in dimethyl formamide was stirred at room temperature in the presence of formaldehyde (30 mmol) for 15 min. Then, compound **4** was added into it and stirred at room temperature for additional 24 h. The reaction mixture was poured into ice-water and a solid obtained. This crude product was collected by filtration and recrystallized from dimethyl sulfoxide to afford the desired product.

<u>Method 2</u>

The mixture of ciprofloxacin (for 5b) or norfloxacin (for 5a) (10 mmol), compound 4 and formaldehyde (30 mmol) was irradiated in monomode microwave reactor in closed vessel with pressure control at 80 W for 15 min. The solid obtained was recrystallized from dimethyl sulfoxide to afford the desired product.

<u>1-Ethyl-6-fluoro-7-{4-(5-([4-(4-nitrophenyl)piperazin-1-yl]methyl)-2-thioxo-1,3,4-oxadiazol-3(2H)-yl}methyl]</u> piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (5a)

mp.: 220-221°C, Yield: 70% (Method 1), 88% (Method 2). FT-IR (ν_{max} , cm⁻¹): 3365 (COOH), 3071 (ar-H), 2829 (C=S), 1724 (C=O), 1626 (C=O), 1496 and 1329 (NO₂), 1224 (C-O). ¹H NMR (DMSO- d_6 , δ ppm): 1.39- 1.43 (m, 3H, CH₃), 2.64-92 (m, 8H, 4CH₂), 3.34-3.50 (m, 8H, 4CH₂+ H₂O), 3.79 (s, 2H, CH₂), 4.58 (s, 2H, CH₂), 5.05 (s, 2H, CH₂), 7.01 (d, 2H, ar-H, J= 8.0 Hz), 7.17 (d, 1H, ar-H, J= 8.0 Hz), 7.17 (d, 1H, ar-H), 8.95 (s, 1H, quinolone =CH), 15.36 (s, 1H, COOH). ¹³C NMR (DMSO- d_6 , δ ppm): 14.79 (CH₃), 43.30 (CH₂), 49.54

(CH₂), 49.79 (2CH₂), 51.98 (CH₂), 69.92 (CH₂), arC: [113.19 (2CH), 126.15(2CH), 137.71 (C), 154.59 (C),], 106.26 and 106.45 (d, CH, J= 9.0 Hz), 107.54 (C), 111.80 (CH), 119.81 (C), 137.45(C), 145.85 (C), 148.99 (CH), 152.65 (C), 155.59 (oxadiazole C-5), 166.63 (C=O), 173.99 (oxadiazole C-2), 176.86 (C=O). LC-MS m/z (%):653,22 ([M+1] ⁺, 100). Elemental Anal. Calcd. for C₃₀H₃₃FN₈O₆S (%): C, 55.21; H, 5.10; N, 17.17, Found (%): C, 55.28; H, 5.17; N, 17.21.

<u>1-Cyclopropyl-6-fluoro-7-{4-(5-[4-(4-nitrophenyl)piperazin-1-yl]methyl)-2-thioxo-1,3,4-oxadiazol-3(2H)-yl[methyl)</u> piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (5b)

mp.: 215-216°C, Yield: 75% (Method 1), 89% (Method 2). FT-IR (v_{max}, cm⁻¹): 3333 (COOH), 3022 (ar-H), 1718 (C=O), 1679 (C=O), 1491 and 1356 (NO₂), 1229 (C-O). ¹H NMR (DMSO- d_6 , δ ppm): 1.18 (s, 2H, CH₂), 1.31 (d, 2H, CH_2 , J= 4.0 Hz), 2.65-2.73 (m, 4H, 2CH₂), 2.89-2.94 (m, 4H, 2CH₂), 3.32 (bs, 6H, 3CH₂+ H₂O), 3.47 (s, 3H, CH + CH₂), 3.80 (s, 2H, CH₂), 5.06 (s, 2H, CH₂), 7.05 (d, 2H, ar-H, J= 12.0 Hz), 7.54-7.60 (m, 1H, ar-H), 7.87-7.95 (m, 2H, ar-H), 8.03 (d, 1H, ar-H, J= 8.0 Hz), 8.65 (s, 1H, quinolone =CH), 15.17 (bs, 1H, COOH). 13 C NMR (DMSO- d_6 , δ ppm): 8.04 (2CH₂), 36.30 (CH), 46.68 (2CH₂), 48.35 (CH₂), 49.73 (2CH₂), 51.06 (CH₂), 51.47 (CH₂), 51.96 (2CH₂), 69.91 (CH₂), arC: [113.18 (2CH), 126.12 (2CH), 139.59 (C),], 107.03 (CH), 107.22 (C), 111.31 and 111.54 (d, CH, J= 23.0 Hz), 119.09 and 119.17 (d, C, J= 8.0 Hz), 137.43 (C), 145.41 (C), 148.45 (CH), 152.19 and 154.67 (d, C-F, J_{C-F} = 248.0 Hz), 155.06 (oxadiazole C-5), 166.37 (C=O), 176.82 (C=O), 178.53 (oxadiazole C-2). LC-MS m/z (%): 665,22 ([M+1]⁺, 100). Elemental Anal. Calcd. for C₃₁H₃₃FN₈O₆S (%):C, 56.02; H, 5.00; N, 16.86, Found (%): C, 56.08; H, 5.07; N, 16.91.

2.1.5. General Method for The Synthesis of Compounds 6a-f

<u>Method 1</u>

A mixture of compound 3 (10 mmol) and the corresponding iso(thio)cyanate (10 mmol) was stirred in dichloromethane at room temperature for 12 h (for 6b), 21 h (for 6d and 6e) or 20 h (for 6a, 6c and 6f). On evaporating the solvent under reduced pressure, a solid obtained. The crude product was recrystallized from ethanol (for 6a, 6c, 6d and 6f) or ethyl acetate (for 6b and 6e) to afford the desired product.

<u>Method 2</u>

A mixture of compound **3** (10 mmol) and the corresponding iso(thio)cyanate (20 mmol) was irradiated in monomode microwave reactor in closed and vessel with pressure control at 80 W for 12 min. The crude product obtained was recrystallized from ethanol to (for **6a**, **6c**, **6d** and **6f**) or ethyl acetate (for **6b** and **6e**) to give the pure compound.

<u>N-Benzyl-2-{[4-(4-nitrophenyl]piperazin-1-yl]acetyl}</u> hydrazincarbothioamide (6a)

mp. 209-210°C Yield: 59% (Method 1), 73% (Method 2). FT-IR (v_{max} , cm⁻¹): 3285, 3268, 3231, 3119 (NH), 1743 (C=O), 1484 and 1307 (NO₂).¹H NMR (DMSO- d_6 , δ ppm): 2.64 (s, 4H, 2CH₂), 3.15 (s, 2H, CH₂), 3.38 (s, 2H, CH₂ + H₂O), 3.70 (s, 4H, 2CH₂), 7.02 (d, 2H, ar-H, J= 8.0 Hz), 7.06-7.43 (m, 5H, ar-H), 8.05 (d, 2H, ar-H, J= 8.0 Hz), 9.59 (bs, 2H, 2NH), 9.94 (bs, 1H, NH). ¹³C NMR (DMSO- d_6 , δ ppm): 46.73(2CH₂), 51.88 (CH₂), 52.72 (2CH₂), 59.99 (CH₂), arC: [113.10 (4CH), 126.19 (4CH), 128.62 (CH), 137.28 (C), 139.58 (2C)], 155.15 (C=O), 181.34 (C=S).LC-MS m/z (%): 430.16 ([M+2]⁺, 24), 429.16 ([M+1]⁺, 94), 220.10 (100). Elemental Anal. Calcd. forC₂₀H₂₄N₆O₃S (%):C, 56.06; H, 5.65; N, 19.61, Found (%): C, 56.05; H, 5.67; N, 19.61.

<u>N-Benzyl-2-{[4-(4-nitrophenyl]piperazin-1-yl]acetyl}</u> hydrazincarboxamide (6b)

mp. 203-204°C Yield: 78% (Method 1), 97% (Method 2). FT-IR (v_{max} , cm⁻¹): 3298, 3231, 3183 (NH), 1697 (C=O).¹H NMR (DMSO- d_6 , δ ppm): 2.62 (s, 4H, 2CH₂), 3.09 (s, 2H, CH₂), 3.48 (s, 4H, 2CH₂ + H₂O), 4.25 (s, 2H, CH₂), 6.97 (d, 3H, ar-H, *J*= 32.0 Hz), 7.28 (s, 4H, ar-H), 7.91 (s, 1H, NH), 8.05 (s, 2H, ar-H), 9.53 (s, 2H, 2NH).¹³C NMR (DMSO- d_6 , δ ppm): 43.13(CH₂), 50.36 (2CH₂), 51.54 (2CH₂), 57.67 (CH₂), arC: [112.41 (2CH), 126.40 (2CH), 126.92 (CH), 127.59 (2CH), 128.43 (CH), 138.88 (C), 139.85 (C), 158.23 (C)], 156.89 (C=O), 171.95 (C=O).LC-MS m/z (%): 412.19 ([M]⁺, 100), 413.19 ([M+1]⁺, 22.1). Elemental Anal. Calcd. ForC₂₀H₂₄N₆O₄ (%): C, 58.24; H, 5.87; N, 20.38, Found (%): C, 57.85; H, 5.67; N, 19.44.

<u>N-Ethyl-2-{[4-(4-nitrophenyl)piperazin-1-yl]acetyl}</u> <u>hydrazincarbothioamide (6c)</u>

mp.124-125°C Yield: 75% (Method 1), 96% (Method 2). FT-IR (v_{max} , cm⁻¹): 3286, 3228, 3202, 3120 (NH), 1743 (C=O), 1485 and 1319 (NO₂).¹H NMR (DMSO-*d*₆, δ ppm): 1.19 (t, 3H, CH₃, *J*= 8.0 Hz), 2.63-2.65 (m, 4H, 2CH₂), 3.29 (s, 2H, CH₂), 3.45-3.49 (m, 4H, 2CH₂), 4.09 (q, 2H, CH₂, *J*= 8.0 Hz), 7.01-7.04 (m, 2H, ar-H, *J*= 8.0 Hz), 7.85 (bs, 1H, NH), 8.03-8.08 (m, 2H, ar-H), 9.12 (bs, 1H, NH), 9.68 (bs, 1H, NH).¹³C NMR (DMSO-*d*₆, δ ppm): 14.58 (CH₃), 46.79(2CH₂), 51.87 (2CH₂), 58.54 (CH₂), 60.36 (CH₂), arC: [113.08 (CH), 113.11 (CH), 126.16 (2CH), 137.32 (2C)], 155.16 (C=O), 170.30 (C=S). LC-MS m/z (%): 367.16 ([M+1]⁺, 100). Elemental Anal. Calcd. for C₁₅H₂₂N₆O₃S (%):C, 49.17; H, 6.05; N, 22.93, Found (%): C, 49.14; H, 6.08; N, 22.96.

<u>N-Phenyl-2-{[4-(4-nitrophenyl)piperazin-1-yl]acetyl}</u> <u>hydrazincarbothioamide (6d)</u>

mp.192-194°C Yield: 73% (Method 1), 92% (Method 2). FT-IR (v_{max} , cm⁻¹): 3314, 3277, 3216 (NH), 3053 (ar-H), 1744 (C=O). ¹H NMR (DMSO-*d*₆, δ ppm): 2.48 (s, 4H, 2CH₂), 3.29 (s, 2H, CH₂), 3.44 (s, 4H, 2CH₂), 6.92 (d, 2H, ar-H, *J*= 4.0 Hz), 7.13 (m, 1H, ar-H), 7.22 (m, 2H, ar-H), 7.34 (m, 2H, ar-H), 7.99 (d, 2H, ar-H, *J*= 4.0 Hz), 8.03 (bs, 1H, NH), 8.84 (bs, 2H, NH). ¹³C NMR (DMSO-*d*₆, δ ppm): 52.00 (2CH₂), 54.90 (2CH₂), 61.84 (CH₂), arC: [112.30 (2CH), 124.80 (2CH), 126.56 (2CH),128.41 (CH), 129.35 (2CH), 137.42 (C), 138.54 (C), 155.76 (C)], 170.33 (C=O), 181.12 (C=S). LC-MS m/z (%): 415.15 ([M+1]⁺, 100), 223.18 (32). Elemental Anal. Calcd. For C₁₉H₂₂N₆O₃S (%): C, 55.06; H, 5.35; N, 20.28; Found (%): C, 54.14; H, 5.02; N, 19.98.

<u>N-Phenyl-2-{[4-(4-nitrophenyl)piperazin-1-yl|acetyl}</u> <u>hydrazincarboxamide (6e)</u>

mp. 211-212°C Yield: 76% (Method 1), 93% (Method 2). FT-IR (v_{max} , cm⁻¹): 3258, 3223, 3199 (NH), 1689 (C=O), 1481 and 1332 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 2.62 (s, 4H, 2CH₂), 3.13 (s, 2H, CH₂), 3.49 (s, 4H, 2CH₂), 6.94-7.02 (m, 3H, ar-H), 7.26 (d, 2H, ar-H, *J*=4.0 Hz), 7.46-7.52 (m, 2H, ar-H), 8.05 (d, 2H, ar-H, *J*=16.0 Hz), 8.09 (s, 1H, NH), 8.82 (s,1H, NH), 9.66 (s, 1H, NH). ¹³C NMR (DMSO-*d*₆, δ ppm): 52.16 (2CH₂), 53.89 (2CH₂), 60.80 (CH₂), arC: [113.30 (2CH), 121.60 (2CH), 124.88 (2CH), 128.00 (CH), 128.90 (2CH), 136.92 (C), 139.40 (C), 155.70 (C)], 153.80 (C=O), 170.52 (C=O). LC-MS m/z (%): 398.17 ([M]⁺, 100), 400.19 ([M+2]⁺, 27). Elemental Anal. Calcd. For C₁₉H₂₂N₆O₄ (%):C, 57.28; H, 5.57; N, 21.09, Found (%): C, 56.67; H, 5.14; N, 20.84.

<u>N-(4-fluorophenyl)-2-{[4-(4-nitrophenyl)piperazin-1-yl]</u> acetyl}hydrazincarbothioamide (6f)

mp.201-202°C Yield: 79% (Method 1), 94% (Method 2). FT-IR (v_{max} , cm⁻¹): 3329, 3254, 3221 (NH), 3081 (ar-H), 1736 (C=O), 1488 and 1311 (NO₂).¹H NMR (DMSO-*d*₆, δ ppm): 2.51 (s, 2H, CH₂), 2.65 (s, 4H, 2CH₂), 3.34 (s, 4H, 2CH₂), 7.05 (d, 2H, ar-H, *J*= 8.0 Hz), 7.18 (m, 2H, ar-H), 7.42 (s, 2H, ar-H), 8.06 (d, 2H, ar-H, *J*= 8.0 Hz), 9.61 (bs, 2H, 2NH), 9.92 (bs, 1H, NH). ¹³C NMR (DMSO-*d*₆, δ ppm): 50.87 (2CH₂), 54.93 (2CH₂), 56.90 (CH₂), arC: [111.33 (2CH), 113.79 (2CH), 121.41 (2CH), 126.29 (2CH), 137.65 (C), 150.87 (C), 155.18 (C)], 168.75 (C=O), 181.10 (C=S). LC-MS m/z (%): 433.14 ([M+1]⁺, 100), 434.15 (45), 220.18 (22). Elemental Anal. Calcd. For C₁₉H₂₁FN₆O₃S (%):C, 52.77; H, 4.89; N, 19.43, Found (%): C, 51.94; H, 4.32; N, 19.01.

2.1.6. General Method For The synthesis of Compounds (7a-f)

<u>Method 1</u>

A solution of corresponding carbo(thio)amide **6** (10 mmol) in ethanol/water (1:1) was refluxed in the presence of 2N NaOH (20 mmol) for 7 h (for 7a), 8 h (for 7b), 12 h (for 7c) or 13 h (for 7d and 7e). The resulting solution was then cooled to room temperature and acidified to pH 5 with 37% HCl. The precipitate formed was collected by filtration, washed with water, and recrystallized from dimethylformamide to give the target compounds.

Method 2

A solution of corresponding carbo(thio)amide **6** (10 mmol) and 2N NaOH (20 mmol) in ethanol/water (1:1) was irradiated in monomode microwave reactor in open vessel at 150 W for 14 min. The resulting solution was then acidified to pH 5 with 37% HCl. The precipitate formed was collected by filtration, washed with water, and recrystallized from dimethylformamide to give the target compounds.

<u>Benzyl-5-{[4-(4-nitrophenyl)piperazin-1-yl]methyl}-4H-</u> 1,2,4-triazole-3-thiol (7a)

mp.218-220 °C, Yield: 87% (Method 1), 95% (Method 2). FT-IR (v_{max} , cm⁻¹): 3113 (NH), 3035 (ar-CH), 2821 (SH), 1482 and 1305 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 1.04-

1.07 (m, 4H, 2CH₂), 2.50-3.46 (m, 6H, 3CH₂ + H2O), 5.34 (s, 2H, CH₂), 6.94-6.97 (m, 2H, ar-H), 6.98-7.57 (m, 5H, ar-H), 8.03 (d, 2H, ar-H, J= 8.0 Hz), 13.88 (bs, 1H, SH). ¹³C NMR (DMSO- d_6 , δ ppm): 46.37 (CH₂), 46.67 (CH₂), 46.77 (CH₂), 51.88 (CH₂), 52.06 (CH₂), 52.24 (CH₂), arC: [113.09 (CH), 113.14 (CH), 126.14 (CH), 127.37 (CH), 127.79 (CH), 128.75 (CH), 128.85 (CH), 129.40 (CH), 129.65 (CH), 134.50 (C), 136.69 (C), 149.57 (C)], 155.05 (triazole C-5), 181.34 (triazole C-3). LC-MS m/z (%): 433.13 ([M+ Na]⁺, 10), 412.16 ([M+2]⁺, 22), 411.15 ([M+1]⁺, 100). Elemental Anal. Calcd. forC₂₀H₂₂N₆O₂S (%):C, 58.52; H, 5.40; N, 20.47, Found (%): C, 58.57; H, 5.43; N, 20.49.

5-{[4-(4-Nitrophenyl)piperazin-1-yl|methyl}-4-benzyl-4H-1,2,4-triazole-5-one (7b)

mp.242-243 °C, Yield: 73% (Method 1), 85% (Method 2). FT-IR (ν_{max} , cm⁻¹): 3164 (NH), 3027 (ar-CH), 1689 (C=O), 1479 and 1313 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 2.41 (d, 2H, CH₂, *J*=4.0 Hz), 2.50 (t, 2H, CH₂, *J*=4.0 Hz), 3.24 (s, 2H, CH₂), 3.32 (s, 2H, CH₂), 3.36 (s, 4H, 2CH₂ +H₂O), 4.89 (s, 2H, CH₂), 6.97 (d, 2H, ar-H, *J*=4.0 Hz), 7.26 (d, 3H, ar-H, *J*=8.0 Hz), 7.32 (d, 2H, CH₂, *J*=4.0 Hz), 8.03

(d, 2H, ar-H, J= 12.0 Hz), 11.75 (s, 1H, NH). ¹³C NMR (DMSO- d_6 , δ ppm): 42.43 (CH₂), 51.27 (2CH₂), 51.78 (2CH₂), 56.11 (CH₂), arC: [112.41 (2CH), 126.45 (2CH), 128.01 (CH), 128.53 (2CH), 128.76 (2CH), 137.06 (C), 138.88 (C), 146.90 (C)], 155.70 (triazole C-5), 158.23 (triazole C-3). LC-MS m/z (%): 417.18 ([M+ Na]⁺, 43), 395.19 ([M+1]⁺, 100). Elemental Anal. Calcd. For C₂₀H₂₂N₆O₃ (%): C, 60.90; H, 5.62; N, 21.31, Found (%): C, 59.97; H, 5.13; N, 20.89.

<u>5-{[4-(4-Nitrophenyl)piperazin-1-yl]methyl}-4-ethyl-4H-</u> <u>1,2,4-triazole-3-thiol (7c)</u>

mp.215-216°C, Yield: 75% (Method 1), 88% (Method 2). FT-IR (v_{max} , cm⁻¹): 3088 (ar-CH), 2830 (SH), 1589 and 1336 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 1.27 (t, 3H, CH₃, *J*= 8.0 Hz), 2.48-3.54 (m, 2H, CH₂ + DMSO), 2.63 (d, 2H, CH₂, *J*= 8.0 Hz), 3.12 (s, 2H, CH₂), 3.43-3.46 (m, 2H, CH₂ + H2O), 3.62 (s, 2H, CH₂), 4.02 (q, 2H, CH₂, *J*= 8.0 Hz), 6.99 (d, 2H, ar-H, *J*= 12.0 Hz), 8.02 (d, 2H, ar-H, *J*= 12.0 Hz), 13.60 (bs, 1H, SH). ¹³C NMR (DMSO-*d*₆, δ ppm): 13.75 (CH₃), 46.65 (CH₂), 46.73 (CH₂), 51.89 (2CH₂), 52.07 (CH₂), 58.68 (CH₂), arC: [113.09 (CH), 113.19 (CH), 126.14 (CH), 126.16 (CH), 137.29 (C), 149.26 (C)], 155.11 (triazole C-5), 171.78 (triazole C-3). LC-MS m/z (%): 389.43 ([M+ K+2]⁺, 100), 375.47 ([M+ Na+4]⁺, 85). Elemental Anal. Calcd. forC₁₅H₂₀N₆O₂S (%):C, 51.71; H, 5.79; N, 24.12, Found (%): C, 51.75; H, 5.82; N, 24.10.

5-{[4-(4-Nitrophenyl)piperazin-1-yl|methyl}-4-phenyl-4H-1,2,4-triazole-3-thiol (7d)

mp.198-200 °C, Yield: 81% (Method 1), 92% (Method 2). FT-IR (v_{max} , cm⁻¹): 3109 (NH), 3018 (ar-CH), 2850 (SH), 1480 and 1310 (NO₂). ¹H NMR (DMSO- d_6 , δ ppm): 2.37 (t, 4H, 2CH₂, *J*=12.0 Hz), 3.28 (t, 4H, 2CH₂, *J*=12.0 Hz), 3.36 (s, 2H, CH₂), 6.97 (d, 2H, ar-H, *J*=8.0 Hz), 7.36-7.41 (m, 3H, ar-H), 7.53-7.57 (m, 2H, ar-H), 8.03 (d, 2H, ar-H, *J*= 8.0

Hz), 13.90 (s, 1H, SH). ¹³C NMR (DMSO- d_6 , δ ppm): 48.84 (CH₂), 50.50 (2CH₂), 52.19 (2CH₂), arC: [113.35 (2CH), 126.29 (2CH), 127.21 (2CH), 129.35 (2CH), 132.60 (CH), 137.65 (C), 143.18 (C), 143.18 (C)], 156.79 (triazole C-5), 174.33 (triazole C-3). LC-MS m/z (%): 419.14 ([M+ Na]⁺, 100), 420.15 ([M+2]⁺, 31). Elemental Anal. Calcd. For C₁₉H₂₀N₆O₂S (%): C, 57.56; H, 5.08; N, 21.20, Found (%): C, 57.02; H, 4.89; N, 20.79.

<u>5-{[4-(4-Nitrophenyl)piperazin-1-yl]methyl}-4-phenyl-4H-</u> 1,2,4-triazole-5-one (7e)

mp.242-243 °C, Yield: 68% (Method 1), 81% (Method 2). FT-IR (v_{max} , cm⁻¹): 3184 (NH), 3046 (ar-CH), 1676 (C=O), 1475 and 1305 (NO₂). ¹H NMR (DMSO-*d*₆, δ ppm): 2.64 (t, 4H, 2CH₂, *J*=12.0 Hz), 3.28 (t, 4H, 2CH₂, *J*=12.0 Hz), 3.35 (s, 2H, CH₂), 6.92-7.06 (m, 3H, ar-H), 7.23-7.25 (m, 1H, ar-H), 7.27-7.51 (m, 3H, ar-H), 8.03 (d, 2H, ar-H, *J*= 8.0 Hz), 9.62 (s, 1H, NH). ¹³C NMR (DMSO-*d*₆, δ ppm): 51.16 (2CH₂), 51.74 (2CH₂), 53.16 (CH₂), arC: [113.33 (2CH), 125.52 (2CH), 126.29 (2CH), 127.01 (CH), 129.21 (2CH), 133.28 (C), 137.65 (C), 151.98 (C)], 151.85 (triazole C-3), 157.15 (triazole C-5). LC-MS m/z (%): 419.16 ([M+ K]⁺, 48), 381.17 ([M+1]⁺, 100). Elemental Anal. Calcd. For C₁₉H₂₀N₆O₃ (%): C, 59.99; H, 5.30; N, 22.09, Found (%): C, 59.18; H, 5.04; N, 21.88.

<u>5-{[4-(4-Nitrophenyl)piperazin-1-yl]methyl}-4-(4-</u> fluorophenyl)-4H-1,2,4-triazole-3-thiol (7f)

mp.234-235 °C, Yield: 82% (Method 1), 93% (Method 2). FT-IR (v_{max} , cm⁻¹): 3183 (NH), 3056 (ar-CH), 2831 (SH). ¹H NMR (DMSO- d_6 , δ ppm): 2.37 (t, 4H, 2CH₂, *J*=12.0 Hz), 3.28 (t, 4H, 2CH₂, *J*=12.0 Hz), 3.42 (s, 2H, CH₂), 6.97 (d, 2H, ar-H, *J*=8.0 Hz), 7.36-7.41 (m, 2H, ar-H), 7.53-7.57 (m, 2H, ar-H), 8.03 (d, 2H, ar-H, *J*= 8.0 Hz), 13.90 (s, 1H, SH). ¹³C NMR (DMSO- d_6 , δ ppm): 50.36 (2CH₂), 51.54 (2CH₂), 56.98 (CH₂) arC: [112.41 (2CH), 116.54 (2CH), 126.40 (2CH), 129.13 (2CH), 134.49 (C), 138.88 (C), 158.23 (C), 162.88 (d, *J*_{C-F}=42.0 Hz], 156.82 (triazole C-5), 174.45 (triazole C-3). LC-MS m/z (%): 414.13 ([M+1]⁺, 100), 425.15 ([M+2]⁺, 83), 436.13 ([M+Na]⁺, 24). Elemental Anal. Calcd. For C₁₉H₁₉FN₆O₂S (%):C, 55.06; H, 4.62; F, 4.58; N, 20.28, Found (%): C, 54.72; H, 4.13; N, 19.89.

2.1.7. General Method for The synthesis of Compounds (8a-l)

<u>Method 1</u>

The solution of ciprofloxacin (for **8g-l**) or norfloxacin (for **8a-f**) (10 mmol) in dimethyl formamide was stirred at room temperature in the presence of formaldehyde (37 %, 30 mmol) for 15 min. Then, the corresponding compound 7 was added into it and stirred at room temperature for additional 24 h. The reaction mixture was poured to ice-water and a solid obtained. This crude product was collected by filtration and recrystallized from dimethyl sulfoxide to afford the desired product.

<u>Method 2</u>

The mixture of ciprofloxacin (for **8g-l**), norfloxacin (for **8a-f**) (10 mmol), the corresponding compound 7 and formaldehyde (30 mmol) was irradiated in monomode

microwave reactor in closed vessel with pressure control at 80 W for 15 min. The solid obtained was recrystallized from dimethyl sulfoxide to afford the desired product.

<u>1-Ethyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl)piperazin-1-yl]methyl}-4-benzyl-5-thioxo-4,5-dihydro-1H-1,2,4-triazole-1-yl}methyl[piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (8a)</u>

mp. 216 °C (dec.), Yield: 87% (Method 1), 94% (Method 2). FT-IR (v_{max} , cm⁻¹): 3060 (ar-CH), 1716 (C=O), 1630(C=O), 1447 and 1332 (NO₂), 1248 (C-O). ¹H NMR (DMSO- d_6 , δ ppm): 1.43 (t, 3H, CH₃, J= 4.0 Hz) 2.44 (s, 4H, 2CH₂), 2.95 (s, 4H, 2CH₂), 3.21-3.24 (m, 4H, 2CH₂+H₂O), 3.37 (s, 4H, 2CH₂), 3.54 (s, 2H, CH₂), 4.58 (s, 2H, CH₂), 5.20 (s, 2H, CH₂), 5.42 (s, 2H, CH₂), 6.92 (d, 2H, ar-H, J= 8.0 Hz), 7.16-7.33 (m, 5H, ar-H), 7.91 (d, 2H, ar-H, J= 12.0 Hz), 8.01 (d, 2H, ar-H, J= 8.0 Hz), 8.91 (s, 1H, quinolone =CH), 15.24 (s, 1H, COOH). ¹³C NMR (DMSO- d_6 , δ ppm): The spectrum could not be obtained, due to the slight solubility in any NMR solvent. LC-MS m/z (%): 764.29 $([M+Na]^+, 10), 744.31 ([M+3]^+, 15), 743.31 ([M+2]^+, 45),$ 742.30 ($[M+1]^+$, 100). Elemental Anal. Calcd. For C₃₇H₄₀FN₉O₅S (%): C, 59.91; H, 5.43; N, 16.99, Found (%): C, 59.94; H, 5.47; N, 17.03.

<u>1-Ethyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl)piperazin-1yl]methyl}-4-benzyl-5-oxo-4,5-dihydro-1H-1,2,4-triazole-1yl)methyl]piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3carboxylic acid (8b)</u>

mp. 235 °C, Yield: 88% (Method 1), 96% (Method 2). FT-IR (v_{max}, cm⁻¹): 3053 (ar-CH), 1744 (C=O), 1660(C=O), 1442 and 1328 (NO₂), 1241 (C-O). ¹H NMR (DMSO-*d*₆, δ ppm): 1.41 (t, 3H, CH₃, J= 12.0 Hz), 2.44 (s, 4H, 2CH₂), 2.51 (s, 4H, 2CH₂), 2.82 (s, 4H, 2CH₂), 3.23 (s, 4H, 2CH₂), 3.39 (s, 2H, CH₂), 4.58 (d, 2H, CH₂, J=4.0 Hz), 4.69 (s, 2H, CH₂), 4.96 (s, 2H, CH₂), 6.94 (d, 2H, ar-H, J= 8.0 Hz), 7.17 (d, 1H, ar-H, J=8.0 Hz), 7.25-7.33 (m, 5H, ar-H), 7.91 (d, 1H, ar-H, J= 16.0 Hz), 8.02 (d, 2H, ar-H, J= 8.0 Hz), 8.94 (s, 1H, quinolone =CH), 15.33 (s, 1H, COOH). ¹³C NMR (DMSO-*d*₆, δ ppm): 12.41 (CH₃), 42.69 (CH₂), 48.86 (CH₂), 50.36 (2CH₂), 50.76 (2CH₂), 51.43 (2CH₂), 51.54 (2CH₂), 56.37 (CH₂), 63.93 (CH₂), arC: [112.41 (2CH), 114.25 and 114.45 (d, CH, J=4.0 Hz), 126.43 (2CH), 128.01 (CH), 128.53 (2CH), 128.76 (2CH), 138.88 (C), 158.23 (C)], 102.55 (CH), 111.26 (C), 118.19 (C), 137.01 (C), 138.84 (triazole C-5), 146.55 and 146.90 (d, C, J= 9.0 Hz), 148.80 (CH), 150.10 and 152.20 (d, C-F, $J_{C-F} = 105.0$ Hz), 149.78 (C=O), 153.02 (triazole C-3), 167.02 (C=O), 176.57 (C=O). LC-MS m/z (%): 748.31 ([M+Na]⁺, 23), 727.32 ([M+2]⁺, 39), 726.31 ([M+1]⁺, 100). Elemental Anal. Calcd. For C₃₇H₄₀FN₉O₆ (%): C, 61.23; H, 5.56; N, 17.37, Found (%): C, 60.91; H, 5.17; N, 17.02.

<u>1-Ethyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl)piperazin-1-yl]methyl}-4-ethyl-5-thioxo-4,5-dihydro-1H-1,2,4-triazole-1-yl}methyl]piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (8c)</u>

mp. 265-266 °C, Yield: 90% (Method 1), 97% (Method 2). FT-IR (v_{max} , cm⁻¹): 3037 (ar-CH), 1708 (C=O), 1626

(C=O), 1463 and 1363 (NO₂). ¹H NMR (DMSO- d_6 , δ ppm): 1.30 (s, 3H, CH₃), 1.38 (s, 3H, CH₃), 2.48-2.55 (m, 8H, 4CH₂ + DMSO), 2.67-2.71 (s, 2H, CH₂), 2.86 (s, 2H, CH₂), 3.31-3.40 (m, 4H, 2CH₂+H₂O), 3.68 (s, 2H, CH₂), 4.08 (bs, 2H, CH₂), 4.56 (s, 2H, CH₂), 5.09 (s, 2H, CH₂), 6.97 (d, 2H, ar-H, *J*= 8.0 Hz), 7.14 (s, 1H, ar-H), 7.89 (s, 1H, ar-H), 8.00 (d, 2H, ar-H, *J*= 8.0 Hz), 8.92 (s, 1H, quinolone =CH), 15.24 (s, 1H, COOH). ¹³C NMR (DMSO- d_6 , δ ppm): The spectrum could not be obtained, due to the slight solubility in any NMR solvent. LC-MS m/z (%): 680.77 ([M+1]⁺, 100). Elemental Anal. Calcd. For C₃₂H₃₈FN₉O₅S (%):C, 56.54; H, 5.63; N, 18.54, Found (%): C, 56.58; H, 5.67; N, 18.53.

<u>1-Ethyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl)piperazin-1-yl]methyl}-4-phenyl-5-thioxo-4,5-dihydro-1H-1,2,4-triazol-1-yl}methyl[piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (8d)</u>

mp. 248 °C, Yield: 84% (Method 1), 91% (Method 2). FT-IR (v_{max}, cm⁻¹): 3372 (OH), 3044 (ar-H), 1716 (C=O), 1625(C=O), 1494 and 1325 (NO₂), 1226 (C-O). ¹H NMR (DMSO-*d*₆, δ ppm): 1.37 (t, 3H, CH₃), 2.77 (s, 4H, 2CH₂), 2.93 (s, 4H, 2CH₂), 3.29 (s, 4H, 2CH₂), 3.36 (bs, 4H, 2CH₂+ H₂O), 4.22 (s, 2H, CH₂), 4.56 (s, 2H, CH₂), 5.20 (s, 2H, CH₂), 5.85 (s, 1H, ar-H), 6.30-6.35 (m, 1H, ar-H), 6.73 (s, 3H, ar-H), 7.95 (s, 2H, ar-H), 7.40-7.46 (m, 3H, ar-H), 7.93 (d, 1H, ar-H, J= 8.0 Hz), 8.90 (s, 1H, quinolone =CH), 15.33 (bs, 1H, COOH). ¹³C NMR (DMSO- d_6 , δ ppm): 14.36 (CH₃), 48.70 (2CH₂), 49.56 (2CH₂), 49.91 (2CH₂), 51.53 (2CH₂), 51.85 (2CH₂), 69.74 (CH₂), arC: [101.40 and 101.64 (d, CH, J= 24.0 Hz), 115. 76 (2CH), 115.72 (CH), 116.05 (CH), 119.54 (CH), 129. 13 (2CH), 129.25 (C), 130.81 (2CH), 144.67 (C), 149.69 (C)], 106. 97 (CH), 108.74 (C), 111.38 and 111.62 (d, CH, J= 24.0 Hz), 120. 95 (C), 139.94 (C), 144.51 (C), 148.49 (CH), 151.81 (triazole C-5), 152.53 and 155.01 (d, C-F, J_{C-F} = 248.0 Hz), 166.79 (C=O), 173.04 (triazole C-3), 176.79 (C=O). LC-MS m/z (%): 728,27 ([M+1] $^+$, 100). Elemental Anal. Calcd. ForC₃₆H₃₈FN₉O₅S (%): C, 59.41; H, 5.26; N, 17.32, Found (%):C, 59.38; H, 5.27; N, 17.31.

<u>1-Ethyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl)piperazin-1-yl]methyl}-4-phenyl-5-oxo-4,5-dihydro-1H-1,2,4-triazol-1-yl]methyl[piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (8e)</u>

mp. 190-192 °C, Yield: 83% (Method 1), 92% (Method 2). FT-IR (v_{max}, cm⁻¹): 3046 (ar-CH), 1748 (C=O), 1679 (C=O), 1492 and 1320 (NO₂), 1251 (C-O). ¹H NMR $(DMSO-d_6, \delta ppm)$: 1.44 (s, 3H, CH₃), 2.44 (s, 4H, 2CH₂), 2.52 (s, 4H, 2CH₂), 2.72 (s, 4H, 2CH₂), 2.89 (s, 4H, 2CH₂), 3.26 (bs, $CH_2 + H_2O$), 4.61 (s, 2H, CH_2), 4.73 (s, 2H, CH_2), 6.96 (s, 2H, ar-H), 7.20 (s, 2H, ar-H), 7.55 (s, 4H, ar-H), 7.93 (s, 1H, ar-H), 8.03 (s, 2H, ar-H), 8.96 (s, 1H, quinolone =CH), 15.36 (s, 1H, COOH). ¹³C NMR (DMSO-*d*₆, δ ppm): 12.42 (CH₃), 48.86 (CH₂), 50.36 (2CH₂), 50.76 (2CH₂), 51.44 (2CH₂), 51.56 (2CH₂), 55.68 (CH₂), 63.93 (CH₂), arC: [112.41 (CH), 114.24 and 114.45 (d, CH, J=10.5 Hz), 118.19 (CH), 126.40 (2CH), 126.58 (2CH), 128.91 (2CH), 129.32 (CH), 134.33 (C),], 102.55 (CH), 111.26 (C), 138.74 (C), 138.89 (C), 148.43 (triazole C-5), 146.66 and 146.88 (d, C, J= 11.0 Hz), 150.12 and 152.21 (d, C-F, $J_{C-F} = 105.0$ Hz),

158.28 (C), 148.86 (CH), 150.31 (triazole C-3), 167.05 (C=O), 177.18 (C=O). LC-MS m/z (%): 734.30 ([M+Na]⁺, 28), 712.29 ([M+1]⁺, 100), 666.28 (36). Elemental Anal. Calcd. for $C_{36}H_{38}FN_9O_6$ (%): C, 60.75; H, 5.38; N, 17.71; Found (%): C, 60.11; H, 5.05; N, 16.96.

<u>1-Ethyl-6-fluoro-7-{4-[4-(4-fluorophenyl)]-[(3-{[4-(4nitrophenyl)piperazin-1-yl]methyl}-5-thioxo-4,5-dihydro-1H-1,2,4-triazol-1-yl)methyl]piperazin-1-yl}-4-oxo-1,4dihydroquinoline-3-carboxylic acid (8f)</u>

mp. 242 °C, Yield: 83% (Method 1), 93% (Method 2). FT-IR (v_{max}, cm⁻¹): 3070 (ar-H), 1728 (C=O), 1656 (C=O), 1492 and 1323 (NO₂), 1228 (C-O). ¹H NMR (DMSO- d_6 , δ ppm): 1.42 (s, 3H, CH₃), 2.38 (s, 4H, 2CH₂), 2.51 (s, 4H, 2CH₂), 2.70 (s, 2H, CH₂), 2.99 (s, 2H, CH₂), 3.26 (s, 4H, 2CH₂), 4.59 (s, 4H, 2CH₂), 5.20 (s, 2H, CH₂), 6.95 (s, 2H, ar-H), 7.19 (s, 1H, ar-H), 7.40 (s, 2H, ar-H), 7.58 (s, 2H, ar-H), 7.92 (d, 1H, ar-H, J= 12.0 Hz), 8.02 (s, 2H, ar-H), 8.95 (s, 1H, quinolone =CH), 15.37 (bs, 1H, COOH). 13 C NMR (DMSO-d₆, δ ppm): 12.45 (CH₃), 48.05 (CH₂), 50.35 (2CH₂), 50.94 (2CH₂), 51.85 (2CH₂), 52.05 (2CH₂), 55.74 (CH₂), 63.51 (CH₂), arC: [112.41 (2CH), 114.25 and 114.48 (d, CH, J= 11.5 Hz), 115.85 and 116.07 (d, 2CH, J= 11.0 Hz), 126.40 (2CH), 129.84 (2CH), 133.36 (C), 138.75 (C), 158.23 (C), 161.32 and 163.41(d, C-F, $J_{C-F} = 104.5$ Hz)], 102.45 (CH), 111.38 (C), 118.19 (C), 138.45 (C), 146.77 (C), 148.81 (CH), 150.22 and 152.15 (d, C-F, $J_{C-F} = 96.5$ Hz), 154.35 (triazole C-5), 167.02 (C=O), 171.13 (triazole C-3), 178.41 (C=O). LC-MS m/z (%): 746.27 ([M+1]⁺, 100), 747.25 ([M+2]⁺, 37). Elemental Anal. Calcd. For C₃₆H₃₇F₂N₉O₅S (%):C, 57.98; H, 5.00; N, 16.90, Found (%): C, 56.97; H, 4.83; N, 16.32.

<u>1-Cyclopropyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl] piperazin-1-yl]methyl}-4-benzyl-5-thioxo-4,5-dihydro-1H-</u> <u>1,2,4-triazole-1-yl}methyl]piperazin-1-yl}-4-oxo-1,4-</u> <u>dihydroquinoline-3-carboxylicacid (8g)</u>

mp. 225-226 °C, Yield: 91% (Method 1), 97% (Method 2). FT-IR (v_{max} , cm⁻¹): 3086 (ar-CH), 1732 (C=O), 1668(C=O), 1495 and 1322 (NO₂), 1256 (C-O). ¹H NMR (DMSO-d₆, δ ppm): 1.16 (d, 2H, CH₂, J=8.0 Hz), 1.34 (d, 2H, CH_2 , J= 8.0 Hz) 2.47 (s, 4H, 2CH₂), 2.98 (s, 4H, 2CH₂), 3.16 (s, 4H, 2CH₂), 3.38 (s, 4H, 2CH₂), 3.55 (s, 2H, CH₂), 3.81 (s, 1H, CH), 5.20 (s, 2H, CH₂), 5.43 (s, 2H, CH₂), 6.92 (d, 2H, ar-H, J= 8.0 Hz), 7.30 (s, 5H, ar-H), 7.55 (d, 1H, ar-H, J= 4.0 Hz), 7.88 (d, 1H, ar-H, J= 12.0 Hz), 8.01 (d, 2H, ar-H, J= 8.0 Hz), 8.66 (s, 1H, quinolone =CH), 15.01 (s, 1H, COOH). ¹³C NMR (DMSO-*d*₆, δ ppm): 8.04 (2CH₂), 36.24 (CH), 46.57 (2CH₂), 47.97 (2CH₂), 49.99 (2CH₂), 50.26 (2CH₂), 52.13 (2CH₂), 69.22 (CH₂), arC: [109.37 (C), 113.17 (2CH), 126.01 (2CH), 127.35 (2CH), 127.87 (2CH), 128.87 (CH), 136.48 (C), 139.68 (C), 148.49 (C)], 106. 84 (CH), 107.52 (C), 111.39 and 111.62 (d, CH, J= 23.0 Hz), 119.22 (C), 137.77(C), 145.64 and 145.84 (d, C, J= 10.0 Hz), 148.29 (CH), 149.46 (triazole C-5), 152.53 and 155.01 (d, C-F, $J_{C-F} = 248.0$ Hz), 166.20 (C=O), 170.47 (triazole C-3), 176.87 (C=O). LC-MS m/z (%): 776.31 ($[M+Na]^+$, 10), 754.33 ([M+1]⁺, 100). Elemental Anal. Calcd. for C₃₈H₄₀FN₉O₅S (%):C, 60.54; H, 5.35; N, 16.72, Found (%): C, 60.58; H, 5.37; N, 16.69.

<u>1-Cyclopropyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl]) piperazin-1-yl]methyl}-4-benzyl-5-oxo-4,5-dihydro-1H-1,2,4-triazole-1-yl)methyl]piperazin-1-yl}-4-oxo-1,4dihydroquinoline-3-carboxylicacid (8h)</u>

mp. 151-153 °C, Yield: 88% (Method 1), 94% (Method 2). FT-IR (v_{max}, cm⁻¹): 3046 (ar-CH), 1725 (C=O), 1674 (C=O), 1491 and 1319 (NO₂), 1251 (C-O). ¹H NMR (DMSO-d₆, δ ppm): 1.20 (s, 2H, CH₂), 1.33 (s, 2H, CH₂) 2.44 (s, 4H, 2CH₂), 2.51 (s, 4H, 2CH₂), 2.73 (s, 2H, CH₂), 3.25 (s, 4H, 2CH₂), 3.39 (s, 4H, 2CH₂), 3.81 (s, 1H, CH), 4.94 (s, 2H, CH₂), 5.04 (d, 2H, CH₂,J=8.0 Hz), 6.96-6.98 (m, 2H, ar-H), 7.28-7.35 (m, 5H, ar-H), 7.56-7.60 (m, 1H, ar-H), 7.91 (d, 1H, ar-H, J= 12.0 Hz), 8.04 (d, 2H, ar-H, J= 8.0 Hz), 8.66 (s, 1H, quinolone =CH), 15.22 (s, 1H, COOH). ¹³C NMR (DMSO-*d*₆, δ ppm): 9.15 (2CH₂), 37.62 (CH), 42.69 (CH₂), 50.35 (2CH₂), 50.78 (2CH₂), 51.44 (2CH₂), 51.58 (2CH₂), 57.38 (CH₂), 65.95 (CH₂), arC: [112.44 (CH), 114.67 (C), 126.43 (2CH), 128.11 (2CH), 128.53 (2CH), 128.76 (2CH), 136.30 (C), 138.85 (C), 158.23 (C)], 107.59 (CH), 109.37 (C), 113.23 and 113.33 (d, CH, J= 10.5 Hz), 137.06 (C), 147.79 (CH), 148.95 and 150.86 (d, C, J= 95.5 Hz), 150.78 (triazole C-3), 153.02 (triazole C-5), 167.05 (C=O), 178.57 (C=O). LC-MS m/z (%): 776.31 ([M+K]⁺, 27), 738.31 ([M+1]⁺, 100). Elemental Anal. Calcd. for C₃₈H₄₀FN₉O₆ (%): C, 61.86; H, 5.46; N, 17.09, Found (%): C, 61.08; H, 5.11; N, 16.73.

<u>1-Cyclopropyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl) piperazin-1-yl]methyl}-4-ethyl-5-thioxo-4,5-dihydro-1H-1,2,4-triazole-1-yl)methyl]piperazin-1-yl}-4-oxo-1,4dihydroquinoline-3-carboxylicacid (8i)</u>

mp. 240-241 °C (dec.), Yield: 86% (Method 1), 92% (Method 2). FT-IR (v_{max}, cm⁻¹): 3019 (ar-CH), 1719 (C=O), 1626 (C=O), 1464 and 1330 (NO₂). ¹H NMR (DMSO-d₆, δ ppm): 1.15 (d, 2H, CH₂, *J*=8.0 Hz), 1.30 (s, 5H, CH₃ +CH₂), 2.54 (s, 4H, 2CH₂), 2.87 (s, 2H, CH₂), 3.06 (s, 2H, CH₂), 3.31-3.40 (m, 8H, 4CH₂ + H₂O), 3.68 (s, 2H, CH₂), 3.80 (s, 1H, CH₂), 4.09 (s, 1H, CH), 5.09 (s, 2H, CH₂), 6.96-7.01 (m, 2H, ar-H), 7.51-7.56 (m, 1H, ar-H), 7.84-7.89 (m, 2H, ar-H), 8.01 (d, 1H, ar-H, J= 8.0 Hz), 8.64 (s, 1H, quinolone =CH), 15.11 (s, 1H, COOH). ¹³C NMR (DMSO-*d*₆, δ ppm): 8.04 (2CH₂), 36.24 (CH), 46.57 (2CH₂), 47.97 (2CH₂), 49.99 (2CH₂), 50.26 (2CH₂), 52.13 (2CH₂), 69.22 (CH₂), arC: [111.62 (C), 113.17 (2CH), 126.01 (2CH), 139.68 (C)], 106. 84 (CH), 107.52 (C), 111.39 (CH), 119.22 (C), 137.77(C), 145.64 and 145.84 (d, C, J= 10.0 Hz), 148.29 (CH), 152.53 and 155.01 (d, C-F, $J_{C-F} = 248.0$ Hz) 155.21 (triazole C-5), 166.20 (C=O), 170.47 (triazole C-3), 176.87 (C=O). LC-MS m/z (%): 692.78 ([M+1]⁺, 100). Elemental Anal. Calcd. For C₃₃H₃₈FN₉O₅S (%): C, 57.30; H, 5.54; N, 18.22, Found (%): C, 57.33; H, 5.58; N, 18.26.

<u>1-Cyclopropyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl) piperazin-1-yl|methyl}-4-phenyl-5-thioxo-4,5-dihydro-1H-1,2,4-triazole-1-yl)methyl]piperazin-1-yl}-4-oxo-1,4dihydroquinoline-3-carboxylicacid (8j)</u>

mp. 225-226 °C (dec.), Yield: 90% (Method 1), 95% (Method 2). FT-IR (ν_{max} , cm⁻¹): 3355 (OH), 3070 (ar-H), 1721 (C=O), 1624(C=O), 1494 and 1324 (NO₂), 1225 (C-O). ¹H NMR (DMSO-*d*₆, δ ppm): 1.04 (d, 2H, CH₂, *J*=8.0 Hz),

1.30 (d, 2H, CH_2 , J= 8.0 Hz) 2.77 (s, 4H, 2 CH_2), 2.95 (s, 4H, 2CH₂), 3.24 (s, 4H, 2CH₂), 3.40 (bs, 4H, 2CH₂+ H₂O), 3.76 (s, 1H, CH), 4.22 (s, 2H, CH₂), 5.20 (s, 2H, CH₂), 5.73 (d, 1H, ar-H, J= 8.0 Hz), 6.34-6.79 (m, 4H, ar-H), 7.95 (d, 2H, ar-H, J= 8.0 Hz), 7.40-7.47 (m, 3H, ar-H), 7.95 (d, 1H, ar-H, J= 8.0 Hz), 8.58 (s, 1H, quinolone =CH), 15.18 (bs, 1H, COOH). ¹³C NMR (DMSO- d_6 , δ ppm): 8.03 (2CH₂), 36.04 (CH), 48.77 (2CH₂), 49.06 (2CH₂), 49.96 (2CH₂), 51.35 (2CH₂), 51.58 (CH₂), 68.74 (CH₂), arC: [101.48 and 101.72 (d, CH, J= 24.0 Hz), 115.67 (2CH), 115.79 (CH), 116.72 (CH), 129.31 (2CH), 129.40 (C), 130.83 (2CH), 144.35 (C), 149.39 (C),], 106. 79 (CH), 108.64 (C), 111.36 and 111.60 (d, CH, J= 24.0 Hz), 119.45 and 120.59 (d, C, J=114.0 Hz), 139.49 (C), 144.46 (C), 148.24 (CH), 151.52 and 155.28 (d, C-F, J_{C-F} = 376.0 Hz), 151.19 (triazole C-5), 166.43 (C=O), 169.79 (triazole C-3), 176.57 (C=O). LC-MS m/z (%): 740,27 ($[M+1]^+$, 100). Elemental Anal. Calcd. For C₃₇H₃₈FN₉O₅S (%): C, 60.07; H, 5.18; N, 17.04, Found (%):C, 60.08; H, 5.17; N, 17.01.

<u>1-Cyclopropyl-6-fluoro-7-{4-[(3-{[4-(4-nitrophenyl]) piperazin-1-yl]methyl}-4-phenyl-5-oxo-4,5-dihydro-1H-1,2,4-triazole-1-yl]methyl]piperazin-1-yl}-4-oxo-1,4dihydroquinoline-3-carboxylicacid (8k)</u>

mp. 165-166 °C, Yield: 79% (Method 1), 88% (Method 2). FT-IR (v_{max}, cm⁻¹): 3034 (ar-CH), 1728 (C=O), 1668 (C=O), 1489 and 1321 (NO₂), 1250 (C-O). ¹H NMR (DMSO-d₆, δ ppm): 1.18 (s, 2H, CH₂), 1.33 (s, 2H, CH₂) 2.42 (s, 4H, 2CH₂), 2.51 (s, 4H, 2CH₂), 2.89 (s, 4H, 2CH₂), 3.25 (s, 4H, 2CH₂), 3.70 (s, 2H, CH₂), 3.81 (s, 1H, CH), 4.73 (s, 2H, CH₂), 6.91-7.02 (m, 2H, ar-H), 7.23-7.27 (m, 1H, ar-H), 7.48 (s, 1H, ar-H), 7.53 (d, 4H, ar-H, J=4.0 Hz), 7.83 (d, 1H, ar-H, J= 12.0 Hz), 8.00 (t, 2H, ar-H, J= 16.0 Hz), 8.62 (s, 1H, quinolone =CH), 15.16 (s, 1H, COOH). 13 C NMR (DMSO-*d*₆, δ ppm): 9.15 (2CH₂), 37.62 (CH), 50.24 (2CH₂), 50.81 (2CH₂), 51.45 (2CH₂), 51.62 (2CH₂), 55.68 (CH₂), 63.95 (CH₂), arC: [109.26 (C), 112.53 (2CH), 113.17 and 114.68 (C), 126.40 (2CH), 128.15 (2CH), 128.54 (2CH), 128.76 (2CH), 136.36 (C), 138.79 (C), 158.58 (C)], 107.55 (CH), 109.41 (C), 113.42 (d, CH, J= 12.5 Hz), 137.16 (C),147.81 (CH), 148.76 and 150.97 (d, C, J= 110.5 Hz), 151.78 (triazole C-5), 153.52 (triazole C-3), 167.46 (C=O), 176.77 (C=O). LC-MS m/z (%): 746.30 ($[M+Na]^+$, 31), 724.29 ([M+1]⁺, 100). Elemental Anal. Calcd. for C₃₇H₃₈FN₉O₆ (%): C, 61.40; H, 5.29; N, 17.42; Found (%): C, 61.02; H, 5.08; N, 16.93.

<u>1-Cyclopropyl-6-fluoro-7-{4-[4-(4-fluorophenyl)]-[(3-{[4-(4-nitrophenyl)piperazin-1-yl]methyl}-5-thioxo-4,5dihydro-1H-1,2,4-triazol-1-yl)methyl[piperazin-1-yl}-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (81)</u>

mp. 201-202 °C, Yield: 80% (Method 1), 91% (Method 2). FT-IR (v_{max} , cm⁻¹): 3043 (ar-CH), 1744 (C=O), 1665 (C=O), 1491 and 1320 (NO₂), 1254 (C-O). ¹H NMR (DMSO-*d*₆, δ ppm): 1.08 (s, 2H, CH₂), 1.33 (s, 2H, CH₂), 2.54 (s, 4H, 2CH₂), 2.72 (d, 8H, 4CH₂, *J*=16.0 Hz), 2.82 (s, 2H, CH₂), 3.36 (s, 4H, 2CH₂), 4.12 (s, 1H, CH), 5.23 (s, 2H, CH₂), 6.23 (s, 1H, ar-H), 6.85 (d, 2H, ar-H, *J*=8.0 Hz), 7.05 (d, 2H, ar-H, *J*=12.0 Hz), 7.37-7.45 (m, 4H, ar-H), 8.01-8.08 (m, 2H, ar-H), 8.66 (s, 1H, quinolone =CH), 15.21 (s, 1H, COOH). ¹³C NMR (DMSO-*d*₆, δ ppm): 7.78 (2CH₂), 36.81

(CH), 52.05 (4CH₂), 52.71 (2CH₂), 54.82 (2CH₂), 60.71 (CH₂), 65.77 (CH₂), arC: [112.34 (2CH), 115.53 and 116.09 (d, 2CH, J= 28.0 Hz), 126.42 (2CH), 129.88 (2CH), 133.54 (C), 138.75 (C), 158.35 (C), 161.42 and 163.42 (d, C-F, J_{C-F} = 100.0 Hz)], 102.73 (CH), 111.41 (C), 114.14 and 114.37 (d, CH, J= 11.5 Hz), 118.21 (C), 138.21 (C), 146.44 (C), 148.75 (CH), 150.16 and 152.32 (d, C-F, J_{C-F} = 108.0 Hz), 154.65 (triazole C-5), 167.74 (C=O), 173.25 (triazole C-3), 177.19 (C=O). LC-MS m/z (%):758.27 ([M+1] ⁺, 100), 712.26 (37), 680.25 (26). Elemental Anal. Calcd. for C₃₇H₃₈FN₉O₆ (%):C, 58.64; H, 4.92; N, 16.63; Found (%): C, 58.01; H, 4.23; N, 16.02.

2.2. Antimicrobial Activity

The test microorganisms were obtained from the Hifzissihha Institute of Refik Saydam (Ankara, Turkey) and were as follows: Escherichia coli (E. coli) ATCC35218, Yersinia pseudotuberculosis (Y. pseudotuberculosis) ATCC911, Pseudomonas aeruginosa (P. aeruginosa) ATCC43288, Enterococcus faecalis (E. faecalis) ATCC29212, Staphylococcus aureus (S. aureus) ATCC25923, Bacillus cereus (B. cereus) 709 Roma, Mycobacterium smegmatis (M. smegmatis) ATCC607. All the newly synthesized compounds were weighed and dissolved in hexane to prepare extract stock solution of 20.000 microgram/milliliter (µg/mL). The antimicrobial effects of the substances were tested quantitatively in respective broth media by using double microdilution and the minimal inhibition concentration (MIC) values (±g/mL) were determined. The antibacterial and antifungal assays were performed in Mueller-Hinton broth (MH) (Difco, Detroit, MI) at pH.7.3 and buffered Yeast Nitrogen Base (Difco, Detroit, MI) at pH 7.0, respectively. The microdilution test plates were incubated for 18-24 h at 35 °C. Brain Heart Infusion broth (BHI) (Difco, Detriot, MI) wasused for M. smegmatis, and incubated for 48-72 h at 35 °C [39]. Ciprofloxacin, Norfloxacin, Ampicillin and Strepromicin were used as standard antibacterial and antimicobacterial drugs, respectively. Dimethylsulfoxide with dilution of 1:10 was used as solvent control. The results obtained were submitted in Table 2.

2.3. Antioxidant Activity Studies

2.3.1. DPPH (2,2-diphenyl-1-picrylhydrazyl) Radical Scavenging Activity

The scavenging activity of different chemicals was determined using the free radical DPPH (2,2-diphenyl-1picrylhydrazyl), as described by Blois [40]. A 100- μ L chemical solution was mixed with 1 ml of freshly prepared methanolic DPPH solution. The reaction mixture was incubated for 30 min at room temperature in the dark and was then measured at 520 nm. The activity was expressed as μ mol Trolox equivalent.

2.3.2. FRAP (The Ferric Reducing Ability of Plasma)

FRAP was measured using the method described by Benzie & Strain [41] with some modification. To 100 μ l of each sample was added 2900 μ l freshly prepared FRAP reagent containing 300 mM acetate buffer (pH 3.6), 10 mM TPTZ (2,4,6-tripyridyle-s-triazine) and 20 mM FeCl.6H₂O in proportions of 10:1:1 (v/v). The mixture was incubated for 30 min at 37°C and measured at 593 nm. The values were expressed as μ mol of Trolox/g.

2.3.3. CUPRAC (Cupric Ion Reducing Antioxidant Capacity)

CUPRAC was measured following the procedure described by Apak *et al.* [42] with some modification. Briefly, 100 μ L of each chemical solution was mixed with 900 μ L bidistilled water, 1 ml acetate buffer solution (1 mM, pH: 7.0), 1 ml CuCl₂ (10 mM) and 1 ml 7.5 mM neocuproine to a final volume of 4 ml. The reaction mixture was then incubated in the dark for 30 min at room temperature, and the absorbance of the reaction mixture was measured at 450 nm against water blank. Trolox was used as the standard calibration curves, and the results were expressed as μ mol Trolox equivalent per g.

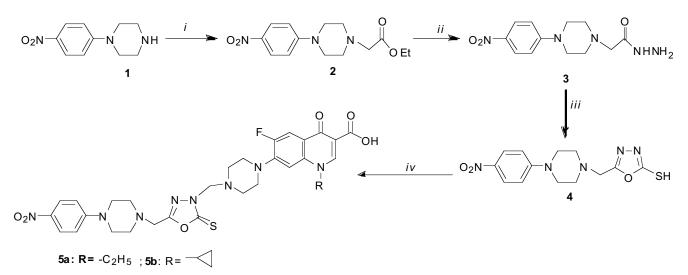
3. RESULTS AND DISCUSSION

3.1. Chemistry

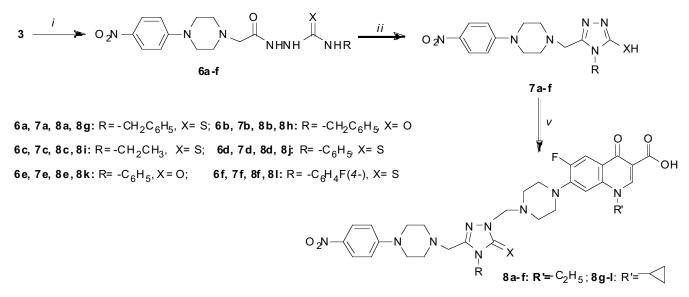
In this study, we attempted to conventional and MW mediated green synthesize of new fluoroquinolone-azole hybrids as possible antimicrobial drug candidates. On the basis of ¹H, ¹³C NMR, FT IR and EI- MS data, the structures of the target products were established. All synthesized compounds were checked for purity and identity using elemental analysis. The MICs against clinically important pathogens were determined as well. The synthetic methodologies adopted to obtain the target compounds were depicted in Schemes **1** and **2**.

The treatment of the starting compound, 1-(4nitrophenyl) piperazine (1) with ethyl bromoacetate to give ethyl 2-[4-(4-nitrophenyl)piperazin-1-yl]acetate (2) was carried out under conventional and also microwave (MW) irradiated conditions with a view to maximizing the yield of the product and minimizing the reaction time. With the assessing of MW irradiated method, the yield of the reaction was improved to good level (99 %) however, more significantly, the reaction time for complete consumption of starting materials was lowered from 3 h with conventional heating to a remarkable 8 min. Moreover, MW irradiated method with no solvent supplied more ecofriendly way. Compound 2 was confirmed by the disappearance of broad singlet for NH and the presence of triplet at the region 1.24 ppm and a quartet at 4.14 ppm due to the presence of ethyl group proton in the ${}^{1}H$ NMR spectrum of compound 2. This group appeared at 14.70 ppm (CH₃) and 60.48 ppm (CH₂) in the ^{f3}C NMR spectrum. Further, in the FT IR, the appearance of the stretching band due to C=O group at the region 1741 cm⁻¹ confirmed the formation of ester (2).

The substitution of ester group by hydrazide function generated [4-(4-nitrophenyl)piperazin-1-yl]acetohydrazide (3), which were confirmed by the appearance of broad signal for -NHNH₂ group in FT IR. Further, the protons of hydrazide function were resonated at 4.32 (NH₂) and 9.06 (NH) as D_2O exchangeable singlets confirming the formation of ace-



Scheme 1. i: BrCH₂COOEt, ii: NH₂NH₂, iii: KOH, CS₂, EtOH-H₂O, iv: HCHO and norfloxacin (for 5a) or ciprofloxacin (for 5b).



Scheme 2. i: suitable isocyanate or isothiocyanate, ii: KOH, EtOH-H2O, v: HCHO and norfloxacin (for 8a-f) or ciprofloxacin (for 8g-l).

tohydrazide (3). In order to optimize microwave (MW) irradiation conditions, MW was applied at different power values of 120 and 150 W without any solvent, while the conventional synthesis of compound **3** required ethanol as reaction solvent. The complete conversion of the compound **2** in best yield was observed after microwave irradiation at 120 W maximum power for 5 min. Higher MW power or longer reaction time caused to lower yield.

Compound **3** produced $5-\{[4-(4-nitrophenyl)piperazin-1$ $yl]methyl<math>\}-1,3,4$ -oxadiazole-2-thiol (**4**) on the treatment with CS₂in basic media. The ¹³C NMR observations reveal the appearance 1,3,4-oxadiazole C-2 and C-5 carbons at the region 161.05 ppm and 178.51 ppm confirming the formation of 1,3,4-oxadiazole. On the other hand, carbo(thio)amides (**6a-f**), which were obtained from the reaction of **3** with alkyliso(thio)cyanates, were confirmed by the presence of additional signals at the related chemical shift values originated from alkyliso(thio)cyanate moiety. These compounds exhibited mass fragmental and elemental analysis data confirming the assigned structures. Two different procedures containing conventional and microwave prompted techniques were applied for the syntheses. In the conventional heating, the reaction yielding compounds **6a-f** was completed in 12-21 h with 59-79 % yield, while reaction time was 12 min with the yield 73-97 %.

The basic treatment of compounds **6a-f** yielded the corresponding 1,2,4-triazoles (**7a-f**), which can be considered as important tools for further condensation reactions leading to the formation of new bioactive molecules. The reaction was carried out in water as a none toxic ecofriendly solvent under reflux and also microwave conditions. With the use of MW, higher yield was assessed, however, the important effect of MW irradiation was on reaction time. Microwave irradiation decreased the reaction time from 7-13 h to 14 min and increased the yields from 73-87 % to 81-95%. The optimum reaction condition was assessed at 150 Watt maximum

| Entry | Solvent | MW (Watt) | Time (min) | Yield (%) |
|-------|-----------------------------|-----------|------------|-----------|
| 1 | EtOH | 100 | 10 | 63 |
| 2 | EtOH-H ₂ O (2:1) | 100 | 10 | 65 |
| 3 | H ₂ O | 100 | 10 | 62 |
| 4 | DMF | 100 | 12 | 84 |
| 5 | DMF | 100 | 14 | 87 |
| 6 | - | 80 | 15 | 94 |
| 7 | - | 75 | 12 | 94 |
| 8 | MeCN | 100 | 12 | 74 |
| 9 | THF | 100 | 12 | 78 |
| 10 | DCM | 100 | 14 | 79 |

Table 1. Optimization of the model reaction conditions for compound 8a under MW conditions.

power. The FT IR spectra of compounds **7a-f** have -SH or -NH stretching bands at 2821-2850 cm⁻¹ and C=O stretching band at 1676-1689 of 1,2,4-triazole nucleus. In the ¹H NMR spectra, resonances assigned to the -SH or NH protons on 1,2,4-triazole ring were detected at 13.60-13.90 ppm (SH) and 9.62-11.75 ppm (NH), respectively which are supported by the literature findings [32-38]. ¹³C NMR spectra of these compounds have resonances of triazole C-5 and C-3 at 155.05-157.15 ppm and 158.23-181.34 ppm, respectively.

The one-pot, three-component Mannich type reaction of compounds 4 and 7a-f with norfloxacin and ciprofloxacin yielded the corresponding fluoroquinolone-azole hybrids (5a, b and 8a-l). This reaction proceeds via the formation of immonium salt which subsequently attacks the N-1 of triazole or oxadiazole N-3 giving rise to the corresponding Mannich bases. In the preliminary experiment, to optimize the conditions for this condensation, the synthesis of compound 8a was selected as model reaction and some reaction parameters including time and MW power were screened on the model reaction (Table 1). With the aim to provide further improvement for this synthetic approach, the model reaction was also performed in the presence of different organic solvents such as acetonitrile, EtOH, EtOH-H2O, H2O, THF, MeCN and DMF, however, the best result was assessed in solvent free media and the corresponding product was obtained in nearly quantitative yields within 15 min (Table 1, entry 6) in model reaction under microwave irradiation. In comparison with the long refluxing time in hazardous solvent, microwave irradiation provided more efficient and green way for one pot Mannich type condensation with relatively higher product yield. The number of signals and their chemical shifts are in accordance with the assigned structures for compounds 8a-l. In the ¹H and ¹³C NMR spectra, additional signals corresponding tonorfloxacin or ciprofloxacin were recorded at the related chemical shift values, while the spectra of these compounds showed the disappearance of the characteristic bands of triazole (or oxadiazole)-NH or SH. Moreover, the preparation of Mannich bases (8a-l) was verified by registration of their mass spectrums which were in accordance with their molecular masses and the elemental

analysis data (carbon, hydrogen and nitrogen) were $\pm 0.4\%$ of the theoretical values.

3.2. Biological Activity

3.2.1. Antimicrobial Activity

All the compounds were tested for their antimicrobial activities and the results were presented in Table 2. This reveals that the newly synthesized compounds, exhibited good to slight activities against some of the test microorganisms with the mic values varying between $<0.03-250 \mu g/mL$. All the newly synthesized compounds except 6a-f exhibited activity on Mycobacterium smegmatis (Ms), a nonpigmented rapidly growing mycobacterium leading to mortality with the mic values varying between 0.03-62.50 µg/mL. In fact, the activity on Ms of **5a**,**b** and **8a-1** was better than standard drug streptomycin with mic 4 μ g/mL It is evident from Table 2 that the aminoalkylation of triazoles 7a-f and oxadiazole, 4 with fluoroquinolones namely norfloxacin and ciprofloxacin caused to excellent activity towards the test bacteria with the mic values between $<0.03-0.99 \ \mu g/mL$. It can be concluded that the good activity of these compounds is due to the presence of fluoroquinolone core in their structures. In fact, the mic values of compounds **5a,b** and **8a-I** are the same or very close to mic values of norfloxacin and ciprofloxacin.

3.2.2. Antioxidant Activity

To determine antioxidant capacity (AC, μ mol TE/g) value of the newly synthesized compounds, DPPH, FRAP and CUPRAC assays were assayed. The AC values of the compounds are given in Table **3** and their statisticall processing (one-way ANOVA) indicated that AC values significantly (P<0.05) differed with in and among measured tests. Compound **8j** (134.32 μ mol TE/g), followed **8d** and **8c** for DPPH, which are fluoroquinolone-triazole-piperazine hybrids, **6c** (1595.4 μ mol TE/g), followed **6d** (784.61 μ mol TE/g) and **6e** (505.17 μ mol TE/g), which are carbothioamides, for FRAP and **8c** (940.20 μ mol TE/g) followed **8j** (582.38 μ mol TE/g) and **8d** (442.36 μ mol TE/g) for CUPRAC had the highest AC values, while compounds **7b** (2.94 and 201.62 μ mol TE/g)

| Compound No. | | Microorganizms and Minimal Inhibition Concentrations(µg/mL) | | | | | | | |
|--------------|------|---|-------|------|-------|-------|------|--|--|
| Compound No. | Ec | Pa | Yp | Sa | Ef | Bc | Ms | | |
| | 250 | 250 | 250 | 250 | 250 | 250 | 35 | | |
| | 250 | 250 | 250 | 250 | 250 | 250 | 35 | | |
| | 250 | 250 | 250 | 250 | 250 | 250 | 62.5 | | |
| | 0,25 | <0,25 | <0,25 | 0,25 | <0,25 | <0,25 | 1.0 | | |
| | 0.03 | 0.03 | 0.03 | 0.03 | 0.03 | 0.03 | 0.03 | | |
| | - | 125 | 250 | 250 | 125 | 250 | - | | |
| | - | - | 250 | - | 125 | 250 | - | | |
| | 250 | 250 | 125 | 250 | 125 | - | - | | |
| | 250 | - | 125 | - | 125 | 250 | - | | |
| | 250 | - | - | 250 | 125 | - | - | | |
| | 250 | 250 | 125 | 250 | 125 | - | - | | |

Table 2. Screening for their antimicrobial activity of the newly synthesized 29 compounds.

(Table 2) contd....

| Compound No. | Microorganizms and Minimal Inhibition Concentrations(µg/mL) | | | mL) | | | |
|--|---|------|------|------|-------|------|-------|
| Compound No. | Ec | Pa | Yp | Sa | Ef | Bc | Ms |
| $O_2 N \longrightarrow N $ | 250 | 250 | 250 | 250 | 250 | 250 | 62.5 |
| | - | 125 | 250 | 250 | 250 | - | 62.5 |
| O_2N N N N N N N N N N | 250 | 250 | 250 | 125 | 250 | 250 | 35 |
| O ₂ N-N-N N-N N SH 7d | - | 250 | 250 | - | 250 | - | 62.5 |
| | 125 | 250 | 250 | 250 | 250 | 250 | 35 |
| O ₂ N-N-N N-N N SH F | 125 | 125 | 250 | 250 | 250 | 250 | 62.5 |
| | 0.03 | 0.48 | 0.06 | 0.24 | 0.488 | 0.48 | 0.244 |
| O_2N N N N N N N N N N | 0,99 | 0,50 | 0,25 | 0,25 | 0,50 | 0,25 | 0,99 |

(Table 2) contd....

Novel Azole-Functionalited Flouroquinolone Hybrids

| Compound No. | Mic | roorganizm | ns and Mini | mal Inhibit | ion Concent | rations(µg/ | mL) |
|--|-------|------------|-------------|-------------|-------------|-------------|-------|
| | Ec | Pa | Yp | Sa | Ef | Bc | Ms |
| O_2N N N N N N N N N N | 0,99 | 0,50 | 0,25 | 0,99 | 0.122 | 0,25 | 0.244 |
| | 0.03 | 0.24 | 0.06 | 0.48 | 0.122 | 0.24 | 0.244 |
| | 0.12 | 0.48 | 0,25 | 0.48 | 0.122 | 0,25 | 0.25 |
| | 0.12 | 0.48 | 0.24 | 0.48 | 0.25 | 0.24 | 0.244 |
| | <0.03 | 0.06 | <0.03 | 0.12 | 0.122 | 0.12 | 0.244 |
| | <0.03 | 0.24 | <0.03 | 0.24 | 0.48 | 0.24 | 0.244 |

(Table 2) contd....

| Compound No. | | Microorganizms and Minimal Inhibition Concentrations(µg/mL) | | | | | | |
|--------------|---------|---|---------|---------|---------|---------|-------|--|
| | Ec | Pa | Yp | Sa | Ef | Bc | Ms | |
| | 0.25 | 0.25 | 0.25 | 0.25 | 0.122 | 0.122 | 0.244 | |
| | <0.03 | 0.03 | 0.06 | 0.24 | 0.25 | 0.12 | 0.25 | |
| | <0.03 | 0.24 | 0.03 | 0.24 | 0.122 | 0.03 | 0.244 | |
| | <0.03 | 0.24 | 0.12 | 0.48 | 0.122 | 0.12 | 0.244 | |
| Cip. | < 0.03 | < 0.03 | <0.03 | 0.122 | 0.122 | 0.06 | - | |
| Norf. | < 0.041 | < 0.041 | < 0.041 | < 0.041 | < 0.041 | < 0.041 | - | |
| Amp. | 10 | 18 | >128 | 35 | 10 | 15 | - | |
| Strep. | - | - | - | - | - | - | 4 | |

Ec: Escherichia coli ATCC 25922, Pa; Pseudomonas aeruginosa ATCC 43288, Yp; Yersinia pseudotuberculosis ATCC 911, Sa; Staphylococcus aureus ATCC 25923, Ef; Enterococcus faecalis ATCC 29212, Bc; Bacillus cereus 702 Roma, Ms; M. smegmatis ATCC607, Cip.; Ciprofloxacin, Norf.; Norfloxacin Amp.; Ampicillin, Strep.; Streptomycin, (—); no activity.

for DPPH and CUPRAC and **2** (37.95µmol TE/g) for FRAP had the lowest AC values among the synthesized compounds (Table **3**). In the table, for the compounds **6a**, **7a** and **7d**, no AC value for FRAP and CUPRAC was determined (Table **3**).

tion (79.78%). Respectively, compounds 6c, 6d, 8a, 8e and 8f at the right upper plan and compounds 5a, 5b, 8b, 8c, 8d, 8j and 8g at the right lower plan were low associated and correlated with FRAP, CUPRAC and DPPH ranging from 0.097 to 0.387 (Fig. (2) at the left upper plan). The remaining compounds 6b, 6e, 6f, 7b, 8h, 8i, 8k and 8l at the left

positive scoreson PC1 explained 48.75% of the total varia-

PCA was carried out by treating data obtained from the three different antioxidant capacity tests. This separation was mainly across PC1, with FRAP, CUPRAC and DPPH having

| Compd. No. | DPPH (µmol TE/g) | FRAP (µmol TE/g) | CUPRAC (µmol TE/g) |
|---------------|---------------------------------------|-----------------------------|--------------------------------|
| 2 | $9.48\pm0.20^{\text{b}}$ | 37.95 ± 2.55^{a} | $269.50\pm1.68^\circ$ |
| 3 | $7.87\pm0.29^{\text{a}}$ | $82.61 \pm 3.63^{\circ}$ | 221.62 ± 2.54^{a} |
| 4 | $3.94\pm0.18^{\text{a}}$ | 91.34 ± 4.52^{ab} | 215.44 ± 6.51^{a} |
| 5a | $56.49\pm2.32^{\rm h}$ | $182.70 \pm 11.90^{\circ}$ | $329.15 \pm 11.35^{\rm g}$ |
| 5b | $64.32\pm5.08^{\rm j}$ | $258.46 \pm 3.15^{\rm d}$ | $382.34 \pm 12.42^{\rm h}$ |
| 6a | $64.33\pm0.05^{\text{g}}$ | n.d.** | n.d.** |
| 6b | $26.09\pm0.33^{\circ}$ | $283.65 \pm 35.42^{\rm ef}$ | $254.30\pm4.47^{\rm c}$ |
| 60 | $63.56\pm0.00^{\text{g}}$ | $1595.4 \pm 67.49^{\circ}$ | 287.32 ± 7.95^{d} |
| 6d | $64.18\pm0.01^{\text{g}}$ | $784.61 \pm 34.46^{\rm h}$ | 309.77 ± 8.79^{e} |
| 6e | $44.98\pm0.32^{\rm d}$ | $505.17 \pm 42.45^{\rm g}$ | $261.91 \pm 16.99^{\circ}$ |
| 6f | $35.63\pm0.28^{\rm d}$ | $408.22 \pm 36.49^{\rm g}$ | $289.74 \pm 13.43^{\circ}$ |
| 7a | $61.60\pm0.03^{\rm fg}$ | n.d.** | n.d.** |
| 7b | $2.94\pm0.29^{\text{a}}$ | 71.61 ± 4.63^{ab} | $201.62 \pm 7.54^{\mathrm{a}}$ |
| 7c | $59.28\pm0.22^{\rm f}$ | $116.70 \pm 4.92^{\rm bc}$ | 299.20 ± 6.76^{de} |
| 7d | 54.97 ± 0.05^{e} | n.d**. | n.d.** |
| 7e | $53.46\pm0.15^{\text{e}}$ | $38.76\pm2.74^{\rm a}$ | $221.33\pm6.39^{\text{b}}$ |
| 7f | $36.32\pm0.19^{\rm f}$ | 125.36 ± 3.88^{bc} | 278.20 ± 4.65^{de} |
| 8a | 42.66 ± 4.33^{d} | $325.64 \pm 19.21^{\rm f}$ | $424.08 \pm 4.89^{\rm f}$ |
| 8b | $62.29\pm1.30^{\rm h}$ | $118.84 \pm 3.78^{\circ}$ | 337.56 ± 12.17^{e} |
| 8c | $73.26\pm3.60^{\rm h}$ | 245.23 ± 21.97^{de} | $940.20 \pm 12.42^{\circ}$ |
| 8d | 120.82 ± 2.32^{1} | $272.82 \pm 21.90^{\circ}$ | $442.36 \pm 12.75^{\text{g}}$ |
| 8e | $82.47\pm2.09^{\rm e}$ | 137.64 ± 18.53° | $347.10 \pm 11.18^{\rm f}$ |
| 8f | 69.63 ± 3.04^{d} | 225.19 ± 15.20° | $332.54 \pm 6.79^{\circ}$ |
| 8g | $70.45\pm0.30^{\rm h}$ | $126.45\pm4.80^{\circ}$ | 315.32 ± 12.40^{e} |
| 8h | $61.59\pm2.76^{\scriptscriptstyle 1}$ | $172.24 \pm 14.98^{\circ}$ | $312.47 \pm 10.57^{\rm g}$ |
| 8i | $60.28\pm2.42^{\rm h}$ | $218.46 \pm 4.43^{\circ}$ | 268.46 ± 11.67^{e} |
| 8j | $134.32\pm5.98^{\rm j}$ | $204.96 \pm 4.17^{\rm d}$ | $582.38 \pm 16.33^{\rm h}$ |
| 8k | 53.75 ± 2.21^{h} | $235.32 \pm 18.86^{\circ}$ | 744.75 ± 11.89^{1} |
| 81 | 59.66 ± 4.87^{d} | $195.90 \pm 12.39^{\rm f}$ | $284.19 \pm 5.67^{\rm f}$ |

| Table 3. | Antioxidant capacity (AC) values of 29 synthesized novel compounds. Values represent the mean ± SD of three determi- |
|----------|--|
| | nations. |

An analysis of variance (SPSS version 11.5. one-way ANOVA) was used for comparisons among the means. Values with the same letter at superscript within a column are not significantly different at P<0.05. * Not detected.

upper and **6a**, **7a**, **7c**, **7d**, **7e**, **7f**, **4**, **3** and **2** at the left lower plan located on PC2 with negative loadings, accounting 31.04% of the variance, were not associate and correlate with any measuring tests (Fig. **2**).

3.2.2.1. Statistical Analysis

Triplicate measurements (n = 3) were performed during the measurements of antioxidant capacity (AC) of the compounds. The AC values were presented as mean \pm pooled



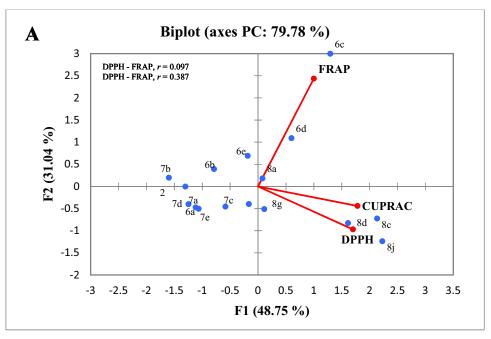


Fig. (2). Loading bi-plot of principle component analysis (PCA) of 29 synthesized novel compounds contributing to values obtained from three antioxidant capacity assays measured by FRAP, CUPRAC and DPPH.

standard deviation (mean \pm SD). For statistical evaluation, one-way ANOVA and Duncan's multiple range test for comparison among the means at significance level of P < 0.05 (IBM SPSS Statistics V22.0) were employed. A statistical software package (XLSTAT version 2015.6) using ADDINSOFT (Damremont, Paris, France) was also employed to perform principal component analysis (PCA).

CONCLUSION

This study reports the conventional and successfully developed microwave assisted synthesis of some new Norfloxacin and Ciprofloxacin derivatives incorporating several other heterocyclic moieties namely 1.3.4-oxadiazole, 1.2.4triazole and piperazine. Hence, we combined all these potential chemotherapeutic units by Mannich reaction, a one pot three component condensation. The structures of new compounds were confirmed by IR, ¹H-, ¹³C NMR, mass spectroscopic and elemental analysis techniques. In addition, the newly synthesized compounds were screened for their antimicrobial and antioxidant activities. The antimicrobial screening suggests that among the newly synthesized compounds, the ones containing fluoroquinolone moiety in their structures exhibited excellent activities against most of the test microorganisms. Among the synthesized compounds, the fluoroquinolone derivatives 8j, 8d and 8c displayed antioxidant activity for DPPH, carbo(thio)amides 6c, 6d and 6e for FRAP and fluoroquinolones 8c, 8j and 8d exhibited antioxidant activity for CUPRAC, while triazole compound 7b for DPPH and CUPRAC and ester 2 for FRAP had the lowest AC values. For the synthesis of target compounds, microwave irradiation was considered more efficient and eco-friendly.

CONSENT FOR PUBLICATION

Not applicable.

CONFLICT OF INTEREST

The authors declare no conflict of interest, financial or otherwise.

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REFERENCES

- Tran, T.P.; Ellsworth, E.L.; Sanchez, J.P.; Watson, B.M.; Stier, M.A.; Showalter, H.D.H.; Domagala, J.M.; Shapiro, M.A.; Joannides, E.T.; Gracheck, S.J.; Nguyen, D.Q.; Bird, P.; Yip, J.; Sharadendu, A.; Ha, C.; Ramezani, S.; Wuc, X.; Singh, R. Structure-activity relationships of 3-aminoquinazolinediones, a new class of bacterial type-2 topoisomerase (DNA gyrase and topo IV) inhibitors. *Bioorg. Med. Chem. Lett.*, **2007**, *17*, 1312-1320.
- [2] Zidar, N.; Tomasic, T.; Macut, H.; Sirc, A.; Brvar, M.; Montalvao, S.; Tammela, P.; Ilas, J.; Kikelj, D. New N-phenyl-4,5dibromopyrrolamides and N-Phenylindolamides as ATPase inhibitors of DNA gyrase. *Eur. J. Med. Chem.*, 2016, *117*, 197-211.
- [3] Sherer, B.A.; Hull, K.; Green, O.; Basarab, G.; Hauck, S.; Hill, P.; Loch, J.T.; Mullen, G.; Bist, S.; Bryant, J.; Boriack-Sjodin, A.; Read, J.; DeGrace, N.; Uria-Nickelsen, M.; Illingworth, R. N.; Eakin, A. E. Pyrrolamide DNA gyrase inhibitors: Optimization of antibacterial activity and efficacy. *Bioorg. Med. Chem. Lett.*, 2011, 21, 7416-7420.
- [4] Demirci, S.; Demirbas, A.; Ulker, S.; Alpay-Karaoglu, S.; Demirbas, N. Synthesis of some heterofunctionalized penicillanic acid derivatives and investigation of their biological activities. *Arch. Pharm. Chem. Life Sci.*, **2013**, *346*, 1-21.
- [5] Ceylan, S.; Bektas, H.; Bayrak, H.; Demirbas, N.; Alpay-Karaoglu, S.; Ulker, S. Syntheses and biological activities of new hybrid molecules containing different heterocyclic moieties. *Arch. Pharm. Chem. Life Sci.*, 2013, 346, 743-756.
- Bisacchi, G.S.; Manchester, J.I. A new-class antibacterial-almost. Lessons in drug discovery and development: A critical analysis of

more than 50 years of effort toward ATPase inhibitors of DNA gyrase and topoisomerase IV. ACS Infect. Dis., 2015, 1, 4-41.

- [7] Siva, S.P.; Liaqat, S.; Girgis, S.A.; Samir, A.; Hall, C.D.; Katritzky, R.A. Novel antibacterial active quinolone-fluoroquinolone conjugates and 2D-QSAR studies. *Bioorg. Med. Chem. Lett.*, 2015, 25, 3816-3821.
- [8] Wang, X.D.; Wei, W.; Wangb, P.F.; Tang, Y.T.; Deng, R. C.; Li, B.; Zhou, S.S.; Zhang, J.W.; Zhang, L.; Xiao, Z.P.; Ouyang, H.; Zhu, H.L. Novel 3-arylfuran-2(5H)-one-fluoroquinolone hybrid: Design, synthesis and evaluation as antibacterial agent, *Bioorg. Med. Chem.*, **2014**, *22*, 3620-3628.
- [9] Xiao, Z.P.; Wang, X.D.; Wang, P.F.; Zhou, Y.; Zhang, J.W.; Zhang, L.; Zhou, J.; Zhou, S.S.; Ouyang, H.; Lin, X.Y.; Mustapa, M.; Reyinbaike, A.; Zhu, H.L. Design, synthesis, and evaluation of novel fuoroquinolone-flavonoid hybrids as potent antibiotics against drug-resistant microorganisms. *Eur. J. Med. Chem.*, 2014, 80, 92-100.
- [10] Basoglu, S.; Ulker, S.; Alpay-Karaoglu, S.; Demirbas, N. Microwave-assisted synthesis of some hybrid molecules containing penicillanic acid or cephalosporanic acid moieties and investigation of their biological activities. *Med. Chem. Res.*, 2014, 23, 3128-3143.
- [11] Panda, S.S.; Liaqat, S.; Girgis, A.S.; Samir, A.; Hall, C.D.; Katritzky, A.R. Novel antibacterial active quinolone-fluoroquinolone conjugates and 2D-QSAR studies. *Bioorg. Med. Chem. Lett.*, 2015, 25, 3816-3821.
- [12] Chugunova, E.; Akylbekov, N.; Bulatova, A.; Gavrilov, N.; Voloshina, A.; Kulik, N.; Zobov, V.; Dobrynin, A.; Syakaev, V.; Burilov, A. Synthesis and biological evaluation of novel structural hybrids of benzofuroxan derivatives and fluoroquinolones. *Eur. J. Med. Chem.*, **2016**, *116*, 165-172.
- [13] Abdel-Aziz, M.; Park, S.E.; Abuo-Rahma, G.E.D.A.A.; Sayed, M.A.; Kwon, Y. Novel N-4-piperazinyl-ciprofloxacin-chalcone hybrids: Synthesis, physicochemical properties, anticancer and topoisomerase I and II inhibitory activity. *Eur. J. Med. Chem.*, 2013, 69, 427-438.
- [14] Panda, S.S.; Detistov, O.S.; Girgis, A.S.; Mohapatra, P.P.; Samir, A.; Katritzky, A.R. Synthesis and molecular modeling of antimicrobial active fluoroquinolone-pyrazine conjugates with amino acid linkers. *Bioorg. Med. Chem. Lett.*, **2016**, *26*, 2198-2205.
- [15] Plech, T.; Wujec, M.; Kosikowska, U.; Malm, A.; Rajtar, B.; Polz-Dacewicz, M. Synthesis and *in vitro* activity of 1,2,4-triazoleciprofloxacin hybrids againstdrug-susceptible and drug-resistant bacteria. *Eur. J. Med. Chem.*, **2013**, *60*, 128-134.
- [16] Plech, T.; Kapron, B.; Paneth, A.; Kosikowska, U.; Malm, A.; Strzelczyk, A.; Staczek, P.; Swiatek, L.; Rajtar, B.; Polz-Dacewicz, M. Search for factors affecting antibacterial activity and toxicity of 1,2,4-triazole-ciprofloxacin hybrids. *Eur. J. Med. Chem.*, **2015**, *97*, 94-103.
- [17] Panda, S.S.; Liaqat, S.; Girgis, A.S.; Samir, A.; Hall, C.D.; Katritzky, A.R. Novel antibacterial active quinolone-fluoroquinolone conjugates and 2D-QSAR studies. *Bioorg. Med. Chem. Lett.*, 2015, 25, 3816-3821.
- [18] Yamunaa, E.; Zeller, M.; Prasad, K.J.R. InCl₃-catalyzed fourcomponent reaction: A novel synthesis of N-carbazolyl dihydropyridines. *Tetrahedron Lett.*, 2011, 52, 6805-6808.
- [19] Dongare, S.B.; Chavan, H.V.; Bhale, P.S.; Mule, Y.B.; Kotmale, A.S.; Bandgar, B.P. A catalyst- and solvent-free multicomponent synthesis of 7-azagramine analogues *via* a Mannich type reaction. *Chinese Chem. Lett.*, **2016**, *27*, 99-103.
- [20] Alizadeh, A.; Ghanbaripour, R.; Zhu, L.G. A Re-engineering approach: Synthesis of Pyrazolo[1,2-a]Pyrazoles and Pyrano[2,3-c]Pyrazoles via an Isocyanide-based four-component reaction under solvent-free conditions. *Tetrahedron*, 2014, 70, 2048-2053.
- [21] Bhuyan, D.; Sarma, R.; Prajapati, D. Microwave-assisted efficient synthesis of spiroquinoline derivatives *via* a catalyst- and solventfree aza-Diels-Alder reaction. *Tetrahedron Lett.*, **2012**, *53*, 6460-6463.
- [22] Sivakumar, K.K.; Rajasekaran, A.; Senthilkumar, P.; Wattamwar, P.P. Conventional and microwave assisted synthesis of pyrazolone Mannich bases possessing anti-inflammatory, analgesic, ulcero-

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genic effect and antimicrobial properties. *Bioorg. Med. Chem. Lett.*, **2014**, *24*, 2940-2944.

- [23] Tala, R.S.; Schnell, M.S.; Haskell, L.C. Microwave-assisted solidphase synthesis of side-chain to side-chain lactam-bridge cyclic peptides. *Bioorg. Med. Chem. Lett.*, 2015, 25, 5708-5711.
- [24] Nayak, P.S.; Narayana, B.; Sarojini, B.K.; Sheik, S.; Shashidhara, K.S.; Chandrashekar K.R. Design synthesis, molecular docking and biological evaluation of imides, pyridazines, and imidazoles derived from itaconic anhydride for potential antioxidant and antimicrobial activities. J. Taibah Univ. Sci., 2016, 10, 823-838.
- [25] Rostom, S.A.F.; Bekhit, A.A. Microwave-assisted synthesis of certain pyrrolyl pyridines, some derived ring systems and their evaluation as anticancer and antioxidant agents. *Eur. J. Med. Chem.*, 2015, 92, 712-722.
- [26] Balabani, A.; Hadjipavlou-Litina, D.J.; Litinas, K.E.; Mainou, M.; Tsironi, C.C.; Vronteli, A. Synthesis and biological evaluation of (2,5-dihydro-1H-pyrrol-1-yl)-2H-chromen-2-ones as free radical scavengers. *Eur. J. Med. Chem.*, **2011**, *46*, 5894-5901.
- [27] Sheppard, J.G.; Long, T.E. Allicin-inspired thiolated fluoroquinolones as antibacterials against ESKAPE pathogens. *Bioorg. Med. Chem. Lett.*, 2016, 26, 5545-5549.
- [28] Vázquez, M.M.P.; Vázquez, P.P.; Galera, M.M.; Gil García, M.D. Determination of eight fluoroquinolones in groundwater samples with ultrasound-assisted ionic liquid dispersive liquid-liquid microextraction prior to high-performance liquid chromatography and fluorescence detection. *Analytica Chimica Acta*, 2012, 748, 20-27.
- [29] Greeff, J.; Joubert, J.; Malan, S.F.; van Dyk, S. Antioxidant properties of 4-quinolones and structurally related flavones. *Bioorg. Med. Chem.*, 2012, 20, 809-818.
- [30] Venepally, V.; Prasad, R.B.N.; Poornachandra, Y.; Kumar, G.; Jala, R.C.R. Synthesis of novel ethyl 1-ethyl-6-fluoro-7-(fatty amido)-1,4-dihydro-4-oxoquinoline-3-carboxylate derivatives and their biological evaluation. *Bioorg. Med. Chem. Lett.*, 2016, 26, 613-617.
- [30] Mermer, A.; Demirci, S.; Ozdemir, S.B.; Demirbas, A.; Ulker, S.; Ayaz, F.A.; Aksakal, F.; Demirbas, N. Conventional and microwave irradiated synthesis, biological activity evaluation and molecular docking studies of highly substituted piperazine-azole hybrids. *Chinese Chem. Lett.*, 2017, 28, 995-1005.
- [31] Demirci, S.; Mermer, A.; Ak, G.; Aksakal, F.; Colak, N.; Demirbas, A.; Ayaz, F.A.; Demirbas, N. Conventional and microwaveassisted total synthesis, antioxidant capacity, biological activity, and molecular docking studies of new hybrid compounds. J. Heterocyclic Chem., 2017, 54(3), 1785–1805.
- [32] Yolal, M.; Basoglu, S.; Bektas, H.; Demirci, S.; Alpay-Karaoglu, S.; Demirbas, A. Synthesis of eperezolid-like molecules and evaluation of their antimicrobial activities. *Russian J. Bioorg. Chem.*, 2012, 38, 539-549.
- [33] Demirci, S.; Demirbas, A.; Ulker, S.; Alpay-Karaoglu, S.; Demirbas, N. Synthesis of some heterofunctionalized penicillanic acid derivatives and investigation of their biological activities. *Arch. Pharm. Chem. Life Sci.*, 2014, 347, 200-220.
- [34] Basoglu, S.; Demirbas, A.; Ulker, S.; Alpay-Karaoglu, S.; Demirbas, N. Design, synthesis and biological activities of some 7aminocephalosporanic acid derivatives. *Eur. J. Med. Chem.*, 2013, 69, 622-631.
- [35] Bayrak, H.; Demirbas, A.; Karaoglu, S.A.; Demirbas, N. Synthesis of some new 1, 2, 4-triazoles, their Mannich and Schiff bases and evaluation of their antimicrobial activities. *Eur. J. Med. Chem.*, 2009, 44, 1057-1066.
- [36] Bayrak, H.; Demirbas, A.; Demirbas, N.; Karaoglu, S.A. Synthesis of some new 1, 2, 4-triazoles starting from isonicotinic acid hydrazide and evaluation of their antimicrobial activities. *Eur. J. Med. Chem.*, 2009, 44, 4362-4366.
- [37] Demirci, S.; Basoglu, S.; Bozdereci, A.; Demirbas, N. Preparation and antimicrobial activity evaluation of some new bi- and triheterocyclic azoles. *Med. Chem. Res.*, 2013, 22, 4930-4945.
- [38] Basoglu, S.; Ulker, S.; Alpay-Karaoglu, S.; Demirbas, N. Microwave-assisted synthesis of some hybrid molecules containing penicillanic acid or cephalosporanic acid moieties and investigation of their biological activities. *Med. Chem. Res.*, 2013, 23, 3128-3143.

- [39] Villanova, P.A. National committee for clinical laboratory standards. 3rd ed; NCCLS Document **1993**, *13*, 25.
- [40] BLOIS, M.S. Antioxidant determinations by the use of a stable free radical, *Nature*, **1958**, *181*, 1199-1200.
- [41] Iris F.F.; Benzie, J.J.S. Ferric reducing/antioxidant power assay: Direct measure of total antioxidant activity of biological fluids and modified version for simultaneous measurement of total antioxi-

dant power and ascorbic acid concentration. *Method Enzymol.*, 1999, 299, 15-27.

[42] Apak, R.; Güçlü, K.; Özyürek, M.; Karademir, S.E. Novel total antioxidant capacity index for dietary polyphenols and vitamins C and E, using their cupric ion reducing capability in the presence of neocuproine: CUPRAC method. J. Agricul. Food Chem., 2004, 52, 7970-7981.