



Tetrahedron Letters 44 (2003) 5807-5810

TETRAHEDRON LETTERS

A short and efficient synthesis of 2'-deoxybenzo- and pyridoimidazole C-nucleosides

Mohamed Jazouli,^a Dominique Guianvarc'h,[†] Mohamed Soufiaoui,^{a,*} Khalid Bougrin,^a Pierre Vierling^b and Rachid Benhida^{b,*}

^aLaboratoire de Chimie des Plantes et de Synthèse Organique et Bioorganique, Université Mohamed V-Agdal, Faculté des Sciences, Avenue Ibn Batouta, BP 1014, Rabat, Morocco

^bLaboratoire de Chimie Bioorganique UMR-CNRS 6001, Université de Nice-Sophia Antipolis, Parc Valrose,

06108 Nice cedex 2, France

Received 3 June 2003; accepted 5 June 2003

Abstract—A short route to a series of 2'-deoxy-C-nucleosides featuring substituted nucleobases has been developed. The key step is the formation of the cyclized products following Mukaiyama's type amide coupling and a simple dehydration, starting from readily accessible synthons. The epimerization of the Cl'-stereogenic center was avoided under mild and controlled conditions. © 2003 Elsevier Ltd. All rights reserved.

C-Nucleosides are important targets in organic synthesis due to their potential value as therapeutic agents and biochemical probes. They have been also found in a number of natural and synthetic products (such as thiazofurin, showdomycin, pyrazomycin, etc...) with potent antiviral or antitumor activities.¹ Recently, several synthetic C-nucleosides have been studied as chemotherapeutic agents,^{1,2} as potential universal bases,3 or as building blocks in artificial DNA and RNA syntheses⁴ for the control of gene expression. Their properties were also investigated in a number of other biochemical applications.⁵ Another attractive feature of C-nucleosides arises from the presence of the C-C glycosidic bond which confers to these analogs a greater resistance towards chemical and enzymatic hydrolysis than to the N-nucleosides. Therefore, a large number of synthetic approaches to C-ribofuranosyl nucleosides have recently been explored.6 However, less attention has been focused on their 2'-deoxy analogs.⁷ One of the most well known strategy is the conversion of the C-nucleosides to their 2'-deoxy derivatives by using the radical 2'-deoxygenation process (Barton-McCombie reaction) which, however, requires additional steps and, thus, decreases the overall yield.⁸ The standard nucleophilic substitution of an appropriate halogenated sugar has also been largely used but was in some cases low yielding.⁹

In the continuation of our studies aimed at overcoming the chemical limitations of triple helix formation by using modified synthetic oligonucleotides,¹⁰ we report herein a short synthesis (three key steps) of 2'-deoxy-*C*nucleosides **6** featuring a heterocyclic nucleobase, starting from the readily accessible *O*-Tol protected 2'-deoxyribonic acid **3** with a given α or β anomeric configuration (Scheme 1). These three steps consist into (i) a Mukaiyama's type coupling between 2'-deoxyribonic acid **3** and *ortho*-aryl disubstituted diamines (**a**–**d**) to give the conjugates **4** followed by sequential (ii) acid catalyzed ring-closure, and (iii) *O*-Tol deprotection to afford the target 2'-deoxybenzimidazole or pyridoimidazole *C*-nucleosides **6**.

The starting known 2'-deoxyribonic acids 3α and 3β were best obtained in two steps from the *O*-Tol protected α -chlorosugar 1 by using TMSCN and BF₃Et₂O as catalyst (Scheme 1), followed by the hydrolysis of the resulting cyano-derivative 2α and 2β , respectively (~90% overall yields). The cyano-derivative 2 was obtained as a mixture of α/β diastereomers (α/β molar ratio = 1/3) which were easily separated by standard silica gel column chromatography. It should be noted that, when we performed the Cl/CN substitution as

^{*} Corresponding authors. Tel.: +33-(0)4-9207-6153; fax: +33-(0)4-9207-6151; e-mail: benhida@unice.fr

[†] Present address: Laboratoire de Biophysique, UR 565 INSERM, UMR 8646 CNRS, Muséum National d'Histoire Naturelle, 43 rue Cuvier 75231 Paris Cedex 05, France.



Scheme 1. Synthetic strategy to the 2'-deoxy-C-nucleosides 6β .

described in literature, i.e. by using NaCN in different solvents¹¹ or Et₂AlCN in toluene,¹² we isolated **2** in much lower yields (below 45%) and with a non-controlled α/β anomeric ratio. These lower yields were due in part to the observed 3'- and 5'-toluoyl cleavage side reaction.

The coupling of 3β with the *ortho*-diamines (**a**–**d**) using Mukaiyama's reagent (chloromethylpyridinium iodide)¹³ in CH₂Cl₂ afforded the corresponding β -conjugates **4a**–**d** β in good yields (75–92%). These compounds, apart from the pyridine analog **4d** β (vide infra), were then cleanly cyclized by simple heating in

Table 1.



Entry 1	ortho-Diamine $(\mathbf{a}-\mathbf{d})$ X=C, R=Ac	Temp. (°C)/time (h) 40/12	5 α/β ratio (yield %) ^a	
			5a	0/100 (60)
2	X = C, R = Ac	95/16	5a	50/50 (67)
3	$X = C, R = NO_2$	40/12	5b	0/100 (61)
4	$X = C, R = NO_2$	95/16	5b	45/55 (50)
5	X = C, R = H	45/12	5c	0/100 (55)
6	X = N, R = H	45/12	5d	0/100 (25)
7	X = N, R = H	110/14	5d	25/75 (51)
8	X = N, R = H	50/14	5d	20/80 (50) ^b

^a Yield of pure isolated products (two steps).

^b 1 equiv. of TFA was added.

AcOH, yielding the protected 2'-deoxy-C-nucleoside derivatives 5 β (Table 1, entries 1, 3 and 5), providing however that the temperature was kept at 40–45°C. Indeed, epimerization at the C1'-stereocenter occurred when this reaction was performed in AcOH at 95-110°C (see Table 1, entries 2 and 4).¹⁴ This epimerization likely took place before or during cyclization. That it did not occur from the cyclized 5 β anomer ($\beta \rightarrow \alpha$) is strongly supported by the fact that no epimerization was observed when the pure β -anomers (and also the α -anomers) of **5a** or **5b** were heated in AcOH, the starting materials remaining unchanged (although partial degradation was detected). These results are further in line with our recent findings in the ribo-series with similar analogs¹⁵ and outline that the epimerization process in C-nucleosides is highly dependent on the nature of the aglycone part.^{9,16} In the case of the pyridine analog $4d\beta$, the cyclization was very slow at 40°C and gave a much lower yield of the desired compound 5d β (25%), the starting material being mainly recovered (70%). Conversion of $4d\beta$ into $5d\beta$ could be improved by heating at higher temperature (110°C) or by using additional TFA (1 equiv.) (entries 7 and 8, respectively). However, partial epimerization was observed under these more drastic conditions.

The *O*-Tol-deprotection in **5a**–d β was performed with MeONa in MeOH and afforded quantitatively the target 2'-deoxy-*C*-nucleosides **6a**–d β (Scheme 1).

Where the synthesis sequence of the α anomers was concerned, the coupling step $3(\mathbf{a}-\mathbf{d})\alpha \rightarrow 4(\mathbf{a}-\mathbf{d})\alpha$ was accomplished in a similar way as for their β -analogs, but required longer reaction times. Unfortunately, when $4(\mathbf{a}-\mathbf{d})\alpha$ were heated in AcOH, we observed not only the cyclization into their respective derivatives $5(\mathbf{a}-\mathbf{d})\alpha$ but also the intramolecular migration of the toluoyl group from the 3'-position to the *ortho*-NH₂ group. Indeed, and as shown in Scheme 2, derivative $7\mathbf{b}\alpha$ together with the expected $5\mathbf{b}\alpha$ (in nearly 1/1 molar ratio) was also isolated after AcOH treatment of $4\mathbf{b}\alpha$.



only $6b\alpha$ (87%)

Scheme 2. Cyclization/cleavage steps of the α -derivative.

The *O*- to *N*-toluoyl migration, which was not detected in the case of the β -anomers, is probably favored in the case of α -analogs by (i) the close *cis* relationship of the two substituents in 1'- and 3'-positions (Scheme 2) and (ii) the known 2'-*endo* sugar conformation of such analogs. This migration could be avoided by reversing the order of the reaction sequence. Thus, when **4b** α was first submitted to MeONa/MeOH (deprotection) and then cyclized (AcOH/40°C), the desired product **6b** α was obtained in 87% overall yield.¹⁷

In summary, we have developed a short and straightforward synthesis of 2'-deoxy-C-nucleosides featuring substituted benzimidazole or pyridoimidazole nucleobases. The sequential coupling-cyclization steps could be performed in one operation making the methodology more attractive. Furthermore, this process is highly flexible and does not require the protection of the heterocycles and allows conventional post-synthetic transformations.

Acknowledgements

We are grateful to the CNRS and the Université de Nice-Sophia Antipolis for financial support (Convention CNRS-CNR 11492).

References

 For reviews on the chemistry and biochemistry of Cnucleoside analogs, see: (a) Hacksell, U.; Daves, G. D., Jr. Prog. Med. Chem. 1985, 22, 1–65; (b) Postema, M. H. D. Tetrahedron 1992, 48, 8545–8599; (c) Watanabe, K. A. *Chemistry of Nucleosides and Nucleotides*; Townsend, L. B., Ed.; Plenum Press: New York, 1994; Vol. 3, 421–535; (d) Jaramillo, C.; Knapp, S. *Synthesis* **1994**, 1–20; (e) Togo, H.; He, W.; Waki, Y.; Yokoyama, M. *Synlett* **1998**, 700–716.

- For recent articles, see: (a) Franchetti, P.; Marchetti, S.; Cappellaci, L.; Jayaram, H. N.; Yalowitz, J. A.; Goldstein, B. M.; Barascut, J.-L.; Dukhan, D.; Imbach, J.-L.; Grifantini, M. J. Med. Chem. 2000, 43, 1264–1270; (b) Bojack, G.; Earnshaw, C. G.; Klein, R.; Lindell, S. D.; Lowinski, C.; Preuss, R. Org. Lett. 2001, 3, 839–842.
- (a) Millican, T. A.; Mock, G. A.; Chauncey, M. A.; Patel, T. P.; Eaton, M. A. W.; Gunning, J.; Cutbush, S. D.; Neidele, S.; Mann, J. *Nucleic Acids Res.* **1984**, *12*, 7435– 7453; (b) Kool, E. T. *Acc. Chem. Res.* **2002**, *35*, 936–943 and references cited therein.
- (a) Ogawa, A. K.; Wu, Y.; McMinn, D. L.; Liu, J.; Schultz, P.; Romesberg, F. E. J. Am. Chem. Soc. 2000, 122, 3274–3287; (b) Morales-Rojas, H.; Kool, E. T. Org. Lett. 2003, 4, 4377–4380; (c) Li, J.-S.; Fan, Y.-H.; Zhang, Y.; Marky, L. A.; Gold, B. J. Am. Chem. Soc. 2003, 125, 2084–2093.
- (a) Harusawa, S.; Imazu, T.; Takashima, S.; Araki, L.; Ohishi, H.; Kurihara, T.; Sakamoto, Y.; Yamamoto, Y.; Yamatodani, A. J. Org. Chem. 1999, 64, 8606–8615; (b) Parsch, J.; Engels, J. W. J. Am. Chem. Soc. 2000, 124, 5664–5672; (c) Wenska, G.; Skalski, B.; Gdaniec, Z.; Adamiak, R. W.; Matulic-Adamic, J.; Beigelman, L. J. Photochem. Photobiol. 2000, 133, 169–176; (d) Tanaka, K.; Tasaka, M.; Cao, H.; Shionoya, M. Eur. J. Pharm. Sci. 2001, 13, 77–83.
- (a) Wichai, U.; Woski, S. A. Org. Lett. 1999, 1, 1173– 1175; (b) Franchetti, P.; Cappellaci, L.; Marchetti, S.; Martini, C.; Costa, B.; Varani, K.; Borea, P. A.; Grifantini, M. Bioorg. Med. Chem. 2000, 8, 2367–2373; (c) Ramasamy, K. S.; Bandaru, R.; Averett, D. J. Org. Chem. 2000, 65, 5849–5851; (d) Franchetti, P.; Marchetti, S.; Cappellaci, L.; Yalowitz, J. A.; Jayaram, H. N.; Goldstein, B. M.; Grifantini, M. Bioorg. Med. Chem. Lett. 2001, 11, 67–69; (e) Chun, B. K.; Song, G. Y.; Chu, C. K. J. Org. Chem. 2001, 66, 4852–4858; (f) Manferdini, M.; Morelli, C. F.; Veronese, A. C. Tetrahedron 2002, 58, 1005–1010.
- (a) Calter, M. A.; Zhu, C. J. Org. Chem. 1999, 64, 1415–1419; (b) Chen, D.-W.; Beuscher, A. E., IV; Stevens, R. C.; Wirsching, P.; Lerner, R. A.; Janda, K. D. J. Org. Chem. 2001, 66, 1725–1732; (c) Griesang, N.; Richert, C. Tetrahedron Lett. 2002, 43, 8755–8758; (d) Takase, M.; Morikawa, T.; Abe, H.; Inouye, M. Org. Lett. 2003, 5, 625–628.
- For recent examples, see: (a) Tanaka, K.; Shionoya, M. J. Org. Chem. 1999, 64, 5002–5003; (b) Seela, F.; Debelak, H. J. Org. Chem. 2001, 66, 3303–3312.
- See for example: Aketani, S.; Tanaka, K.; Yamamoto, K.; Ishihama, A.; Cao, H.; Tengeiji, A.; Hiraoka, S.; Shiro, M.; Shionoya, M. J. Med. Chem. 2002, 45, 5594– 5603 and also Refs. 7b and 7c.
- (a) Guianvarc'h, D.; Benhida, R.; Fourrey, J.-L.; Maurisse, R.; Sun, J. S. *J. Chem. Soc., Chem. Commun.* **2001**, 1814–1815; (b) Guianvarc'h, D.; Fourrey, J.-L.; Maurisse, R.; Sun, J. S.; Benhida, R. *Org. Lett.* **2002**, *4*, 4209–4212; (c) Guianvarc'h, D.; Fourrey, J.-L.; Maurisse, R.; Sun, J. S.; Benhida, R. *Bioorg. Med. Chem.* **2003**, *11*, 2751–2759.

- Kolb, A.; Huynh Dinh, T.; Igolen, J. Bull. Soc. Chim. Fr. 1973, 3447.
- 12. Iyer, R. P.; Phillips, L. R.; Egan, W. Synth. Commun. 1991, 21, 2053.
- Mukaiyama, T.; Usui, M.; Shimada, E.; Saigo, K. Chem. Lett. 1975, 1045–1048.
- The anomeric configuration at the Cl'-stereocenter of purified products was assigned on the basis of ¹H NMR (2D COSY-NOESY experiments).
- (a) Guianvarc'h, D.; Benhida, R.; Fourrey, J.-L. *Tetrahedron Lett.* **2001**, *42*, 647–650; (b) Guianvarc'h, D.; Fourrey, J.-L.; Tran Huu Dau, M.-E.; Guérineau, V.; Benhida, R. *J. Org. Chem.* **2002**, *67*, 3724–3732.
- 16. For the conversion of α to β anomer, see: (a) Chaudhuri, N. C.; Ren, R. X.-F.; Kool, E. T. *Synlett* **1997**, 341–347; (b) Jiang, Y. L.; Stivers, J. T. *Tetrahedron Lett.* **2003**, *44*, 85–88.
- 17. Spectral data of selected compounds.

4aβ. Mp (AcOEt)=181–183°C. ¹H NMR (CDCl₃, 200 MHz) δ (ppm) 2.28–2.41 (m, 1H), 2.37 (s, 3H), 2.43 (s, 3H), 2.46 (s, 3H), 2.71–2.86 (m, 1H), 4.20 (br s, 2H), 4.55–4.93 (m, 4H), 5.56 (d, 1H, J=4.8 Hz), 6.75 (d, 1H, J=8.5 Hz), 7.18 (d, 2H, J=7.9 Hz), 7.28 (d, 2H, J=7.9 Hz), 7.71 (dd, 1H, J=2.0 and 8.5 Hz), 7.85 (d, 1H, J=2.0 Hz), 7.87 (d, 2H, J=8.2 Hz), 7.95 (d, 2H, J=8.2 Hz), 8.53 (br s, 1H). ¹³C NMR (CDCl₃, 50 MHz) δ 21.98, 26.43, 37.83, 64.60, 76.24, 79.38, 85.48, 116.83, 126.93, 128.58, 129.62, 129.68, 130.07, 144.83, 145.92, 166.6, 167.40, 170.47. MS (ES⁺) m/z=531 [M+H]⁺, 553 [M+ Na]⁺, 569 [M+K]⁺.

5aβ. Mp (AcOEt)=129–131°C. ¹H NMR (CDCl₃, 200 MHz) δ (ppm) 2.30–2.50 (m, 1H), 2.37 (s, 3H), 2.44 (s, 3H), 2.68 (s, 3H), 2.88 (ddd, 1H, *J*=13.9, 5.8 and 1.2 Hz), 4.59–4.88 (m, 3H), 5.59 (m, 1H), 5.68 (dd, 1H, *J*=10.1 and 5.8 Hz), 7.18 (d, 2H, *J*=7.9 Hz), 7.29 (d, 2H, *J*=7.9 Hz), 7.61 (d, 1H, *J*=8.5 Hz), 7.74 (br s, 1H), 7.91 (d, 2H, *J*=8.5 Hz), 7.93 (dd, 1H, *J*=8.5 and 1.6 Hz), 7.97 (d, 2H, *J*=8.5 Hz), 8.24 (d, 1H, *J*=1.6 Hz). ¹³C NMR (CDCl₃, 50 MHz) δ 21.69, 21.75, 26.79, 39.15, 64.77, 76.03, 76.15, 84.58, 114.69, 123.29, 126.35, 126.47, 128.28, 129.68, 132.34, 144.54, 144.70, 156.79, 166.25, 167.84. MS (IC) *m*/*z*=513 [M+H]⁺, 119 [Tol]⁺.

6aβ. ¹H NMR (CDCl₃, 200 MHz) δ (ppm) 2.30 (m, 1H), 2.43 (m, 1H), 2.65 (s, 3H), 3.65–3.85 (2dd, 2H, J=12.0, 6.4 and 4.8 Hz), 4.05 (m, 1H), 4.32 (m, 1H), 5.40 (dd, 1H, J=9.6 and 6.3 Hz), 7.59 (d, 1H, J=8.5 Hz), 7.92 (dd, 1H, J=8.5 and 1.5 Hz), 8.21 (d, 1H, J=1.5 Hz). ¹³C NMR (CD₃OD, 50 MHz) δ 26.95, 42.81, 63.63, 73.68, 75.90, 89.80, 128.85, 130.12, 130.23, 130.65, 133.44, 159.87. MS (ES⁻) m/z=275 [M–H]⁻.

4a α . ¹H NMR (CDCl₃, 200 MHz) δ (ppm) 2.24 (s, 3H), 2.26 (s, 3H), 2.34 (s, 3H), 2.73 (m, 2H), 4.29 (br s, 2H), 4.44 (d, 2H, *J*=4.6 Hz), 4.70 (t, 1H, *J*=4.6 Hz), 4.83 (t, 1H, *J*=6.5 Hz), 5.49 (br t, 1H, *J*=3.2 Hz), 6.67 (d, 1H, *J*=8.5 Hz), 6.99 (d, 2H, *J*=7.9 Hz), 7.19 (d, 2H, *J*=7.9 Hz), 7.60–7.67 (m, 2H), 7.74 (d, 2H, *J*=8.2 Hz), 7.88 (d,

2H, J=8.2 Hz), 8.40 (s, 1H). ¹³C NMR (CDCl₃, 50 MHz) δ 21.97, 22.02, 26.27, 36.09, 64.40, 76.04, 79.28, 84.54, 116.98, 122.07, 126.58, 126.94, 127.01, 128.59, 129.51, 129.64, 130.03, 130.11, 144.56, 144.78, 146.22, 166.21, 166.55, 171.69. MS (ES⁺) m/z = 531 [M+H]⁺, 553 [M+Na]⁺.

5aα. ¹H NMR (CDCl₃, 200 MHz) δ (ppm) 2.25 (s, 3H), 2.41 (s, 3H), 2.67 (s, 3H), 2.92–3.13 (m, 2H), 4.53 (m, 2H), 4.68 (m, 1H), 5.62 (m, 1H), 5.76 (dd, 1H, J = 6.5 and 4.4 Hz), 6.83 (d, 2H, J=8.0 Hz), 7.10-7.40 (m, 6H), 7.59 (d, 1H, J=8.5 Hz), 7.95 (d, 2H, J=8.0 Hz), 8.23 (br s, 1H). ¹³C NMR (CDCl₃, 50 MHz) δ 21.84, 21.97, 27.07, 37.96, 64.43, 76.10, 76.35, 83.93, 123.38, 126.50, 126.91, 129.22, 129.58, 129.63, 129.95, 132.39, 144.58, 144.54, 158.69, 166.06, 166.73. MS (ES⁺) m/z = 513 [M+H]⁺. 6aα. ¹H NMR (CD₃OD, 200 MHz) δ (ppm) 2.30 (ddd, 1H, J=12.9, 5.0 and 4.7 Hz), 2.64 (s, 3H), 2.78 (ddd, 1H, J = 13.3, 6.3 and 8.2 Hz), 3.65 (dd, 1H, J = 12.6 and 5.0 Hz), 3.74 (dd, 1H, J=12.6 and 4.0 Hz), 4.17 (dd, 1H, J=8.3 and 4.1 Hz), 4.42 (m, 1H), 5.40 (dd, 1H, J=8.3and 5.0 Hz), 7.57 (d, 1H, J=8.6 Hz), 7.90 (dd, 1H, J=8.6 and 1.5 Hz), 8.19 (d, 1H, J=1.5 Hz). ¹³C NMR (CD₃OD, 50 MHz) δ 27.28, 42.12, 63.83, 73.93, 76.22, 89.32, 124.52, 133.53. MS (ES⁻) m/z = 275 [M–H]⁻. **4b** α . ¹H NMR (CDCl₃, 200 MHz) δ (ppm) 2.33 (s, 3H), 2.41 (s, 3H), 2.80 (dd, 2H, J = 5.7 and 2.9 Hz), 4.49 (d, 2H, J=4.6 Hz), 4.68 (br s, 2H), 4.77 (t, 1H, J=4.6 Hz), 4.89 (t, 1H, J = 6.0 Hz), 5.54 (t, 1H, J = 2.9 Hz), 6.70 (d, 1H, J=8.8 Hz), 7.10 (d, 2H, J=8.1 Hz), 7.25 (d, 2H, J = 8.2 Hz), 7.80 (d, 2H, J = 8.2 Hz), 7.92 (dd, 1H, J = 8.8and 2.5 Hz), 7.93 (d, 2H, J=8.1 Hz), 8.01 (d, 1H, J=2.4 Hz), 8.56 (s, 1H). ¹³C NMR (50 MHz, CDCl₃) δ (ppm) 21.64; 21.69, 35.81, 64.27, 75.93, 79.03, 84.19, 111.71, 115.31, 120.77, 120.86, 122.87, 124.00, 126.22, 126.66, 129.75, 129.78, 138.11, 144.59, 144.86, 148.48, 166.36, 166.78, 172.39. **6b** α . ¹H NMR (CD₃OD, 200 MHz) δ (ppm) 2.34 (ddd, 1H, J = 13.0, 8.2 and 4.9 Hz), 2.80 (ddd, 1H, J = 13.0, 6.2 and 5.0 Hz), 3.70 (m, 2H), 4.19 (dd, 1H, J=4.3 and 8.2

and 5.0 Hz), 3.70 (m, 2H), 4.19 (dd, 1H, J=4.3 and 8.2 Hz), 4.38–4.50 (m, 1H), 5.42 (dd, 1H, J=5.0 and 8.2 Hz), 7.60 (d, 1H, J=8.9 Hz), 8.11 (dd, 1H, J=8.9 and 2.0 Hz), 8.39 (d, 1H, J=2.0 Hz). ¹³C NMR (50 MHz, CD₃OD) δ (ppm) 42.08, 63.86, 73.95, 76.23, 89.84, 119.52, 145.18, 163.52. MS (ES⁻) m/z 278 [M–H]⁻.

8bα. ¹H NMR (DMSO- d_6 , 200 MHz) δ (ppm) 2.08 (dt, 1H, J=12.8 and 3.1 Hz), 2.40 (ddd, 1H, J=13.3, 9.0 and 5.4 Hz), 2.50 (s, 3H), 3.70 (s, 2H), 3.90–3.99 (m, 1H), 4.05–4.17 (m, 1H), 4.54 (dd, 1H, J=9.0 and 3.7 Hz), 4.96 (br s, 1H), 6.39 (t, 1H, J=6.8 Hz), 7.01 (d, 1H, J=8.9 Hz), 7.17 (d, 1H, J=8.9 Hz), 7.45 (m, 1H), 7.86 (dd, 1H, J=2.8 and 8.9 Hz), 7.94 (d, 1H, J=5.7 Hz), 9.20 (d, 1H, J=3.0 Hz), 10.08 (s, 1H). ¹³C NMR (50 MHz, DMSO d_6) δ 26.70, 41.11, 62.08, 71.26, 78.31, 88.84, 107.92, 112.38, 112.90, 117.12, 120.07, 131.24, 131.32, 139.46, 139.71, 141.63, 146.64, 154.23, 171.85, 171.92. MS (ES⁻) m/z=414 [M–H]⁻.