Synthesis and Structures of Novel Multi-armed Molecules Involving Benzene as a Core and 4-Phenylthiazole, 4-Pyrazolylthiazole, or Thiadiazole Units as Arms

Mostafa E. Salem, Ahmed F. Darweesh, Ahmad M. Farag, and Ahmed H. M. Elwahy*

Chemistry Department, Faculty of Science, Cairo University, Giza 12613, Egypt *E-mail: aelwahy@hotmail.com Received December 4, 2015 DOI 10.1002/jhet.2629 Published online 00 Month 2016 in Wiley Online Library (wileyonlinelibrary.com).

A synthesis of novel three-, four-, and sixfold branched 4-phenylthiazolylhydrazones, 4-pyrazolylthiazolyl hydrazones, and thiadiazoles which are linked to a benzene core *via* phenoxymethyl spacers was reported. The synthetic methodology includes initially formation of poly(aldehyde thiosemicarbazones) **9**, **14**, and **15** by acid catalyzed condensation of thiosemicarbazide (**8**) with the appropriate poly(aldehydes) **3**, **5**, and **7**, respectively. Subsequent reaction of **9**, **14**, and **15** with each of 2-bromo-1-phenylethanone (**10a**) and 2-bromo-1-(**4**-chlorophenyl)ethanone (**10b**) in refluxing ethanol in the presence of few drops of TEA afforded **11**, **16**, and **18**, respectively, in good yields. On the other hand, the synthesis of the novel poly(4,5-dihydro-1,3,4-thiadiazolyl) derivatives **20**, **21a**,**b**, and **22** was performed by cyclization of **9b**, **14a**,**b**, and **15a**, respectively, in refluxing acetic anhydride.

J. Heterocyclic Chem., 00, 00 (2016).

INTRODUCTION

Since the pioneering work of Vögtle and Weber [1] as well as that of MacNicol *et al.* [2] on multi-armed molecules, much attention have been paid to the synthesis of such compounds for their wide range of applications [3–8]. This class of compounds is considered important hosts for constructing microporous networks possessing selective inclusion properties and their applications in supramolecular host–guest chemistry have been recently reported [9]. Some related compounds have also been used for the formation of discotic mesogens [10], organic electronic and optoelectronic materials [11–13]. They have also been used as building units for dendrimers [14]. Furthermore, thiazole derivatives were

reported to exhibit numerous pharmacological and biological applications, such as anti-HIV, anti-inflammatory, antimicrobial, antihypertensive, analgesic, and herbicidal activity [15].

Moreover, pyrazole system is also considered as an important class of heterocyclic compounds not only for being the core unit in a variety of drugs such as celecobix (Celebrex) [16], sildenafil (Viagra) [17], and rimonabant (Acomplia) [18], but also for possessing a wide range of activities such as antifungal [19], anti-inflammatory [20], antimicrobial [21], antidepressant [22], antiparasitic [23], antiviral [24], and antitumor activities [25].

In addition, 1,3,4-thiadiazole derivatives have also attracted much attention in the last decades for their wide spectrum of biological activities including anti-inflammatory [26], antimicrobial [27], antituberculosis [28], anticancer [29], antidepressant and anxiolytic [30], antihypertensive [31], antioxidant [32], anticonvulsants [33], and antifungal activities [34]. In connection with these findings, we report herein on the synthesis of novel three-, four-, and sixfold branched 4phenylthiazolylhydrazones, 4-pyrazolylthiazolylhydrazones,



Figure 1. Tetrakis- and hexakis(formylphenoxymethyl)benzenes 5a,b and 7a,b.



Scheme 2. Synthesis of tris(aldehyde thiosemicarbazones) 9a and 9b.

and thiadiazoles which are linked to a benzene core *via* phenoxymethyl spacers.

RESULTS AND DISCUSSION

We first synthesized the required aldehydes **3a,b**, **5a,b**, and **7a,b** following reported methods, described by our

group or after modification of literature procedure described by other groups. Thus, tris(formylphenoxymethyl)benzenes **3a,b** were prepared by reacting the potassium salt of 4hydroxybenzaldehyde (**2a**) or salicylaldehyde (**2b**) with tris (bromomethyl)benzene (**1**) in refluxing DMF (Scheme 1) [35].

Similarly, the tetrakis(formylphenoxymethyl)benzenes **5a** and **5b** were prepared, by fourfold substitution of tetrakis

Scheme 3. Reaction of compounds 9a,b with the appropriate 2-bromo-Ethanone derivatives.



a; *p* isomer **b**; *o* isomer

Figure 2. Tetrakis- and hexakis (aldehyde thiosemicarbazones) 14a,b and 15a,b.

Journal of Heterocyclic Chemistry DOI 10.1002/jhet

(bromomethyl)benzenes (4) with four equivalents of the potassium salt of 4-hydroxybenzaldehyde (2a) or salicylaldehyde (2b), respectively. With the same reaction sequence, the hexakis(formylphenoxymethyl)benzenes 7a and 7b were prepared, by sixfold substitution of hexakis(bromomethyl) benzenes (6) with six equivalents of the potassium salt of 4-hydroxybenzaldehyde or salicylaldehyde, respectively (Fig. 1) [35].

The synthetic utility of aldehydes 3, 5, and 7 as building blocks for novel tris-, tetrakis-, and hexakis(thiazoles), in which the 4-phenylthiazolylhydrazone or 4-pyrazolylthiazolylhydrazone is linked to benzene core *via* phenoxymethyl group, was investigated. Thus, the tris(aldehyde

thiosemicarbazones) **9a** and **9b** were first synthesized in 77 and 72% yields, respectively, by acid catalyzed condensation of thiosemicarbazide (**8**) with the appropriate tris (aldehydes) **3a** and **3b**, respectively (Scheme 2).

Reaction of the latter compounds with each of 2-bromo-1-phenylethanone (**10a**) and 2-bromo-1-(4-chlorophenyl) ethanone (**10b**) in refluxing ethanol in the presence of few drops of TEA afforded **11a** and **11b** in 67 and 69% yields, respectively (Scheme 3). In analogy, reaction of compounds **9a,b** with 2-bromo-1-(5-methyl-1-phenyl-1*H*pyrazol-4-yl)ethanone (**12**) in refluxing ethanol in the presence of few drops of TEA afforded **13a** and **13b** in 70 and 64% yields, respectively (Scheme 3). Compound **12** was



a; *p* isomer **b**; *o* isomer

Figure 3. Tetrakis- and hexakis(thiazoles) 16a,b, 17a,b, 18a,b and 19a,b.

synthesized by the reaction of phenylhydrazine with ((dimethylamino)methylene)pentane-2,4-dione, obtained upon treatment of acetylacetone with dimethylformamide dimethylacetal (DMFDMA), followed by bromination upon treatment with Br_2 in AcOH [36].

The same methodology was extended to the preparation of tetrakis- and hexakis(thiazoles) **16–19**. Thus, compounds **16–19** were successfully prepared, from **5a,b** and **7a,b** firstly by reaction with thiosemicarbazide (**8**) in refluxing EtOH containing few drops of AcOH to give **14a,b** and **15a,b** in 70, 73 and 68, 64% yields, respectively (Fig. 2).

Subsequent reaction of the latter compounds with the appropriate bromoacetyl compounds **10a,b** and **12** in refluxing ethanol in the presence of few drops of TEA afforded the corresponding tetrakis(thiazoles) **16a,b** and **17a,b** in 65, 69 and 66, 68% yields, respectively, as well as hexakis(thiazoles) **18a,b** and **19a,b** in 63, 65 and 66, 67% yields, respectively (Fig. 3).

Our study was extended to include the synthesis of the novel tris(4,5-dihydro-1,3,4-thiadiazolyl) derivative **20** in which the thiadiazolyl moiety is linked to the benzene core *via* phenoxymethyl group, in good yield, by cyclization of

tris(aldehyde thiosemicarbazone) **9b** in refluxing acetic anhydride (Scheme 4).

Encouraged by the above results, we expanded the scope of this reaction to prepare novel fourfold branched (4,5-dihydro-1,3,4-thiadiazolyl) derivatives **21a** and **21b** and sixfold branched (4,5-dihydro-1,3,4-thiadiazolyl) derivative **22** (Fig. 4) by cyclization of the appropriate tetrakis(aldehyde thiosemicarbazones) **14a** and **14b** and hexakis(aldehyde thiosemicarbazone) **15a**, respectively, with acetic anhydride.

The structures of the new synthesized compounds were confirmed by IR, NMR, and mass spectra as well as elementary analyses. The IR spectrum of tris (thiazolylhydrazone) **11a** as a representative example of these class of compounds revealed an absorption band at 3431 cm^{-1} because of (NH). Its ¹H NMR spectra showed the presence of a characteristic singlet signal at 7.99 ppm because of one methine proton (—N=CH—). Mass spectrum of compound **11a** showed the molecular ion peaks at m/z 999 (M⁺) in agreement with its respective molecular formula. The spectra of other bis(thiazoles) **11b**, **16a,b**,

Scheme 4. Synthesis of tris(4,5-dihydro-1,3,4-thiadiazolyl) derivative 20 from 9b.



Figure 4. Tetrakis- and hexakis (4,5-dihydro-1,3,4-thiadiazolyl) derivatives 21a,b and 22.

and **18a**,**b** showed similar spectral data which are listed in the experimental part.

The symmetry of compounds **19a,b** is manifested by a single set of signals characteristic of the six equivalent OCH₂ and six methyl groups in the ¹H-NMR spectra. Similarly, compounds **21a,b** are characterized by a single set of signals characteristic of the four equivalent OCH₂, four acetyl (CH₃CO), four acetamido (NHCOCH₃), and four methine (CH) groups.

CONCLUSIONS

We developed a simple method for the preparation of novel three-, four-, and sixfold branched 4-phenylthiazolylhydrazones, 4-pyrazolylthiazolylhydrazones, and thiadiazoles which are linked to a benzene core *via* phenoxymethyl spacers. Full characterization of these compounds is reported. The new synthesized compounds are interesting both in their own right as unusual molecules and for their promising pharmacological and biological activities as well as for their expected inclusion properties. The extension of the scope of this method to cover additional multi-armed heterocyclic compounds is now under study.

EXPERIMENTAL

General. Melting points were determined in open glass capillaries with a Gallenkamp apparatus. The infrared spectra were recorded in potassium bromide disks on a Pye Unicam SP 3-300 and Shimadzu FTIR 8101 PC infrared spectro photometer. NMR spectra were recorded in deuterated dimethyl sulfoxide (DMSO-d6) with a Varian Mercury VXR-300 NMR spectrometer at 300 MHz (¹H NMR) and at 75 MHz (¹³C NMR). Mass spectra (EI) were obtained at 70 eV with a type Shimadzu GCMQP 1000 EX spec trometer. Analytical thin-layer chromatography was per formed using pre-coated silica gel 60778 plates (Fluka), and the spots were visualized with UV light at 254 nm. The elemental analyses were performed at the Micro analytical center, Cairo University.

Synthesis of poly(thiosemicarbazones) 9a,b, 14a,b, and 15a,b. General procedure. To a solution of poly(aldehydes) 3a,b or 5a,b or 7a,b (1 mmol) in absolute ethanol (25 mL) containing few drops of acetic acid, thiosemicarbazide (8) (3 or 4 or 6 mmol) was added. The reaction mixture was heated under reflux for 3 h. The solid formed upon cooling was collected and recrystallized from ethanol/DMF to give the corresponding.poly(thiosemicarbazones) 9a,b, 14a,b, and 15a,b.

2,2',2''-(4,4',4''-(Benzene-1,3,5-triyltris(methylene))tris(oxy) tris(benzene-4,1-diyl))-tris(methan-1-yl-1-ylidene)tris

(*hydrazinecarbothioamide*) *9a.* Pale yellow powder, (77% yield), mp. 204–205°C; IR: (potassium bromide) 3421,

3367 (NH₂), 3254 (NH), 1601 (C=N) cm⁻¹; ¹H-NMR: δ 5.19 (s, 6H, 3 CH₂O), 7.03–7.99 (m, 21H, ArH + 3 NH₂), 8.07 (s, 3H, 3 CH=N), 11.26 (s, 3H, 2 NH); ms: *m/z* (%) 699 (9.5, M⁺) 389 (54.2), 151 (100), 135 (29.3), 57 (93.5), 43 (82.6). *Anal.* Calcd. for C₃₃H₃₃N₉O₃S₃: C, 56.63; H, 4.75; N, 18.01; S, 13.74. Found: C, 56.55; H, 4.66; N, 17.91; S, 13.68.

2,2',2''-(2,2',2''-(Benzene-1,3,5-triyltris(methylene))tris(oxy) tris(benzene-2,1-diyl))tris(methan-1-yl-1-ylidene)tris

(hydrazinecarbothioamide) 9b. Pale yellow powder, (72% yield), mp. 200–202°C; IR: (potassium bromide) 3421, 3366 (NH₂), 3251 (NH), 1595 (C=N) cm⁻¹; ¹H-NMR: δ 5.26 (s, 6H, 3 CH₂O), 6.96–8.13 (m, 21H, ArH+3 NH₂), 8.51 (s, 3 CH=N), 11.39 (s, 3H, 3 NH); ms: *m/z* (%) 699 (10.2, M⁺), 630 (60.2), 321 (100), 310 (47.2), 55 (40.3). *Anal.* Calcd. for C₃₃H₃₃N₉O₃S₃: C, 56.63; H, 4.75; N, 18.01; S, 13.74. Found: C, 56.52; H, 4.69; N, 17.95; S, 13.66.

2,2',2'',2'''-(4,4',4'',4'''-(Benzene-1,2,4,5-tetrayltetrakis(methylene)) tetrakis (oxy)-tetrakis(benzene-4,1-diyl))tetrakis(methan-1-yl-1-ylidene) tetrakis (hydrazinecarbo-thioamide) 14a. Pale yellow powder, (73% yield), mp. 246–248°C; IR: (potassium bromide) 3425, 3364 (NH₂), 3254 (NH), 1594 (C=N) cm⁻¹; ¹H-NMR: δ 5.29 (s, 8H, 4 CH₂O), 7.03–7.98 (m, 26H, ArH+4 NH₂), 8.07 (s, 4H, 4 CH=N), 11.29 (s, 4H, 4 NH); ms: *m*/*z* (%) 906 (24.9, M⁺), 815 (100), 801 (31.3), 481 (21.9), 249 (12.2), 57 (11.8). Anal. Calcd. for C₄₂H₄₂N₁₂O₄S₄: C, 55.61; H, 4.67; N, 18.53; S, 14.14. Found: C, 55.51; H, 4.53; N, 18.39; S, 14.01.

2,2',2'',2'''-(2,2',2''',2'''-(Benzene-1,2,4,5-tetrayltetrakis (methylene))tetrakis (oxy)tetrakis(benzene-2,1-diyl))tetrakis(methan-1-yl-1-ylidene)tetrakis (hydrazine-carbothioamide) 14b. Pale yellow powder, (70% yield), mp. 203–205°C; IR: (potassium bromide) 3430, 3275 (NH₂), 3150 (NH), 1595 (C=N) cm⁻¹; ¹H-NMR: δ 5.41 (s, 8H, 4 CH₂O), 6.93–8.11 (m, 26H, ArH+4 NH₂), 8.51 (s, 4H, CH=N), 11.42 (s, 4H, 4 NH); ms: *m/z* (%) 906 (14.9, M⁺), 816 (47.6), 814 (100), 801 (31.3), 481 (21.9), 250 (12.2), 59 (12.8). Anal. Calcd. for C₄₂H₄₂N₁₂O₄S₄: C, 55.61; H, 4.67; N, 18.53; S, 14.14. Found: C, 55.44; H, 4.45; N, 18.41; S, 13.88.

2,2',2'',2''',2'''',2'''''-(4,4',4''',4'''',4'''',4''''-(Benzene-1,2,3,4,5,6hexaylhexakis (methylene))hexakis(oxy)hexakis(benzene-4,1-diyl)) hexakis(methan-1-yl-1-ylidene)hexakis(hydrazinecarbothioamide) 15a. Pale yellow powder, (68% yield), mp. > 300°C; IR: (potassium bromide) 3423, 3271 (NH₂), 3159 (NH), 1601 (C=N) cm⁻¹; ¹H-NMR: δ 5.32 (s, 12H, 6 CH₂O), 6.93–7.94 (m, 36H, ArH+6 NH₂), 8.05 (s, 6H, CH=N), 11.26 (s, 6H, 6 NH). Anal. Calcd. for C₆₀H₆₀N₁₈O₆S₆: C, 54.53; H, 4.58; N, 19.08; S, 14.56. Found: C, 54.40; H, 4.43; N, 18.91; S, 14.44.

15b. Pale yellow powder, (64% yield), mp. 240–242°C; IR: (potassium bromide) 3418, 3260 (NH₂), 3145 (NH), 1600

(C=N) cm⁻¹; ¹H-NMR: δ 5.52 (s, 12H, 6 CH₂O), 6.95–8.11 (m, 36H, ArH+6 NH₂), 8.39 (s, 6H, 6 CH=N), 11.31 (s, 6H, 6 NH). *Anal.* Calcd. for C₆₀H₆₀N₁₈O₆S₆: C, 54.53; H, 4.58; N, 19.08; S, 14.56. Found: C, 54.40; H, 4.42; N, 19.19; S, 14.72.

Synthesis of poly(thiazoles) 11a,b, 16a,b, and 18a,b. General procedure. To a solution of poly(thiosemicarbazones) 9a,b or 14a,b or 15a,b (1 mmol) in absolute ethanol (25 mL) containing few drops of TEA, 2-bromo-1phenylethanone (10a) or 2-bromo-1-(4-chlorophenyl) ethanone (10b) (3 or 4 or 6 mmol) was added. The reaction mixture was heated under reflux for 5 h. The solid product formed upon cooling was collected and recrystallized from ethanol/DMF to give the corresponding poly (thiazoles) 11a,b, 16a,b, and 18a,b.

1,3,5-Tris((*4-(2-(4-phenylthiazol-2-yl)hydrazono)methyl*) *phenoxy)methyl)benzene 11a.* Orange powder, (67% yield), mp. 210–212°C; IR: (potassium bromide) 3431 (NH), 1603 (C=N) cm⁻¹; ¹H-NMR: δ 5.20 (s, 6H, 3 CH₂O), 7.01–7.85 (m, 33H, ArH+3 thiazole-H), 7.99 (s, 3H, 3 CH=N), 11.98 (s, 3H, 3 NH); ms: *m/z* (%) 999 (4.5, *M*⁺), 992 (100), 815 (67.1), 481 (21.9), 249 (12.2), 57 (11.8). *Anal.* Calcd. for C₅₇H₄₅N₉O₃S₃: C, 68.45; H, 4.53; N, 12.60; S, 9.62. Found: C, 68.29; H, 4.38; N, 12.45; S, 9.44.

1,3,5-Tris((2-((2-(4-(4-chlorophenyl)thiazol-2-yl)hydrazono) methyl)phenoxy)-methyl)benzene 11b. Yellow powder, (69% yield), mp. 180–182°C; IR: (potassium bromide) 3436 (NH), 1599 (C=N) cm⁻¹; ¹H-NMR: δ 5.30 (s, 6H, 3 CH₂O), 7.01–7.95 (m, 30H, ArH+3 thiazole-H), 8.47 (s, 3H, 3 CH=N), 12.11 (s, 3H, 3 NH); ms: m/z (%) 1101 (5.6, M^+), 1067 (9.1), 603 (100), 339 (59.4), 264 (62.6), 55 (64.5). Anal. Calcd. for C₅₇H₄₂Cl₃N₉O₃S₃: C, 62.04; H, 3.84; N, 11.42; S, 8.72. Found: C, 61.95; H, 3.79; N, 11.31; S, 8.62.

1,2,4,5-Tetrakis((4-((2-(4-phenylthiazol-2-yl)hydrazono) methyl)phenoxy)methyl)-benzene 16a. Orange powder, (65% yield), mp. > 300°C; IR: (potassium bromide) 3433 (NH), 1602 (C=N) cm⁻¹; ¹H-NMR: δ 5.32 (s, 8H, 4 CH₂O), 7.08–7.84 (m, 42H, ArH+4 thiazole-H), 7.98 (s, 4H, 4 CH=N), 11.99 (s, 4H, 4 NH). ¹³C-NMR: δ 66.9, 104.5, 115.1, 127.9, 128.4, 129.2, 129.3, 130.6, 131.3, 131.6, 134.6, 135.0, 135.3, 155.7, 159.4. *Anal.* Calcd. for $C_{74}H_{58}N_{12}O_4S_4$: C, 67.97; H, 4.47; N, 12.85; S, 9.81. Found: C, 67.83; H, 4.31; N, 12.67; S, 9.69.

1,2,4,5-Tetrakis((2-((2-(4-(4-chlorophenyl)thiazol-2-yl) hydrazono)methyl) phenoxy)methyl)benzene 16b. Yellow powder, (69% yield), mp. 190–192°C; IR: (potassium bromide) 3438 (NH), 1598 (C=N) cm⁻¹; ¹H-NMR: δ 5.45 (s, 8H, 4 CH₂O), 7.00–7.86 (m, 38H, ArH+4 thiazole-H), 8.46 (s, 4H, 4 CH=N), 12.10 (s, 4H, 4 NH). ¹³C-NMR: δ 67.5, 104.2, 112.8, 121.1, 122.7, 125.1, 127.1, 128.0, 130.5, 131.8, 133.4, 135.1, 136.8, 149.2, 155.9, 162.2, 168.2. Anal. Calcd. for C₇₄H₅₄Cl₄N₁₂O₄S₄: C, 61.49; H, 3.77; N, 11.63; S, 8.87. Found: C, 61.36; H, 3.65; N, 11.44; S, 8.73. 1,2,3,4,5,6-Hexakis((4-((2-(4-phenylthiazol-2-yl)hydrazono) methyl)phenoxy) methyl)benzene 18a. Orange powder, (63% yield), mp. > 300°C; IR: (potassium bromide) 3428 (NH), 1602 (C=N) cm⁻¹; ¹H-NMR: δ 5.32 (s, 12H, 6 CH₂O), 7.01–7.80 (m, 60H, ArH+6 thiazole-H), 7.95 (s, 6H, 6 CH=N), 11.96 (s, 6H, 6 NH). Anal. Calcd. for C₁₀₈H₈₄N₁₈O₆S₆: C, 67.48; H, 4.40; N, 13.12; S, 10.01. Found: C, 67.28; H, 4.18; N, 12.90; S, 9.87.

1,2,3,4,5,6-Hexakis((2-((2-(4-(4-chlorophenyl)thiazol-2-yl) hydrazono) methyl)phenoxy)methyl)benzene 18b. Yellow powder, (65% yield), mp. 208–210°C; IR: (potassium bromide) 3433 (NH), 1599 (C=N) cm⁻¹; ¹H-NMR: δ 5.55 (s, 12H, 6 CH₂O), 6.94–7.77 (m, 54H, ArH+6 thiazole-H), 8.35 (s, 6H, 6 CH=N), 11.88 (s, 6H, NH); ¹³C-NMR: δ 64.3, 104.2, 112.8, 120.5, 120.7, 123.0, 127.0, 128.3, 129.3, 130.3, 131.7, 133.4, 149.2, 155.8, 162.2, 168.0. *Anal.* Calcd. For C₁₀₈H₇₈Cl₆N₁₈O₆S₆: C, 60.93; H, 3.69; N, 11.84; S, 9.04. Found: C, 60.67; H, 3.52; N, 11.71; S, 8.89.

Synthesis of Poly(pyrazolylthiazoles) 13a,b, 17a,b and 19a,b. *General procedure*. To a solution of poly (thiosemicarbazones) 9a,b or 14a,b or 15a,b (1 mmol) in absolute ethanol (25 mL) containing few drops of TEA, 2bromo-1-(5-methyl-1-phenyl-1*H*-pyrazol-4-yl)ethanone (12) (3 or 4 or 6 mmol) was added. The reaction mixture was heated under reflux for 5 h. The solid product formed upon cooling was collected and recrystallized from ethanol/DMF to give the corresponding poly(pyrazolylthiazoles) 13a,b, 17a,b, and 19a,b.

1,3,5-Tris((*4*-((*2*-(*4*-(*5*-*methyl-1-phenyl-1H-pyrazol-4-yl)thiazol-2-yl)hydrazono) methyl)phenoxy)methyl)benzene 13a.* Orange powder, (70% yield), mp. 220–222°C; IR: (potassium bromide) 3424 (NH), 1601 (C=N) cm⁻¹; ¹H-NMR: δ 2.49 (s, 9H, 3 CH₃), 5.20 (s, 6H, 3 CH₂O), 6.84 (s, 3H, 3 thiazole-H), 7.08–7.61 (m, 30H, ArH), 7.91 (s, 3H, CH=N), 7.98 (s, 3H, 3 pyrazole-H), 11.95 (s, 3H, 3 NH); ¹³C-NMR: δ 11.9, 69.1, 101.4, 115.2, 116.7, 120.2, 124.8, 125.2, 126.4, 127.4, 129.3, 135.4, 137.4, 138.7, 139.2, 141.0, 144.3, 159.1, 168.1; ms: *m/z* (%) 1239 (9.5, M⁺), 992 (100), 815 (67.1), 481 (21.9), 249 (12.2), 57 (11.8). *Anal.* Calcd. for C₆₉H₅₇N₁₅O₃S₃: C, 66.81; H, 4.63; N, 16.94; S, 7.75. Found: C, 66.69; H, 4.52; N, 16.85; S, 7.58.

1,3,5-Tris((2-(2-(4-(5-methyl-1-phenyl-1H-pyrazol-4-yl) thiazol-2-yl)hydrazono)-methyl)phenoxy)methyl)benzene 13b.

Orange powder, (64% yield), mp. > 300°C; IR: (potassium bromide) 3433 (NH), 1597 (C=N) cm⁻¹; ¹H-NMR: δ 2.50 (s, 9H, 3 CH₃), 5.30 (s, 6H, 3 CH₂O), 6.86 (s, 3H, 3 thiazole-H), 7.01–7.83 (m, 30H, ArH), 7.90 (s, 3H, 3 CH=N), 8.46 (s, 3H, 3 pyrazole-H), 12.08 (s, 3 NH); ms: m/z (%) 1239 (11.4, M⁺), 992 (100), 874 (10.7), 249 (12.2), 57 (11.8). Anal. Calcd. for C₆₉H₅₇N₁₅O₃S₃: C, 66.81; H, 4.63; N, 16.94; S, 7.75. Found: C, 66.71; H, 4.51; N, 16.79; S, 7.93.

1,2,4,5-Tetrakis((4-((2-(4-(5-methyl-1-phenyl-1H-pyrazol-4-yl) thiazol-2-yl)hydrazono)methyl)phenoxy)methyl)benzene 17a.

Orange powder, (66% yield), mp. > 300°C; IR: (potassium bromide) 3447 (NH), 1588 (C=N) cm⁻¹; ¹H-NMR: δ 2.49 (s, 12H, 4 CH₃), 5.32 (s, 8H, 4 CH₂O), 6.83 (s, 4H, 4 thiazole-H), 7.08–7.61 (m, 38H, ArH), 7.90 (s, 4H, 4 CH=N), 7.98 (s, 4H, 4 pyrazole-H), 11.95 (s, 4H, 4 NH). ¹³C-NMR: δ 11.9, 66.8, 101.4, 115.2, 116.7, 124.8, 127.5, 127.7, 129.1, 134.8, 135.4, 138.7, 139.2, 140.9, 144.3, 159.0, 168.1. *Anal.* Calcd. for C₉₀H₇₄N₂₀O₄S₄: C, 66.40; H, 4.58; N, 17.21; S, 7.88. Found: C, 66.55; H, 4.66; N, 17.07; S, 8.01.

1,2,4,5-Tetrakis((2-((2-(4-(5-methyl-1-phenyl-1H-pyrazol-4-yl)thiazol-2-yl)hydrazono)methyl)phenoxy) methyl)benzene *17b.* Orange powder, (68% yield), mp. > 300°C; IR: (potassium bromide) 3434 (NH), 1599 (C=N) cm⁻¹; ¹H-NMR: δ 2.49 (s, 12H, 4 CH₃), 5.45 (s, 8H, 4 CH₂O), 6.84 (s, 4H, 4 thiazole-H), 7.00–7.80 (m, 38H, ArH), 7.88 (s, 4H, 4 CH=N), 8.46 (s, 4H, 4 pyrazole-H), 12.06 (s, 4H, 4 NH). *Anal.* Calcd. for C₉₀H₇₄N₂₀O₄S₄: C, 66.40; H, 4.58; N, 17.21; S, 7.88. Found: C, 66.17; H, 4.34; N, 17.39; S, 7.65.

1,2,3,4,5,6-Hexakis((4-((2-(4-(5-methyl-1-phenyl-1H-pyrazol-4-yl)thiazol-2-yl)hydrazono)methyl)phenoxy)methyl)benzene 19a.

Orange powder, (66% yield), mp. > 300°C; IR: (potassium bromide) 3432 (NH), 1600 (C=N) cm⁻¹; ¹H-NMR: δ 2.49 (s, 18H, 6 CH₃), 5.35 (s, 12H, 6 CH₂O), 6.78 (s, 6H, 6 thiazole-H), 7.01–7.53 (m, 54H, ArH), 7.89 (s, 6H, 6 CH=N), 7.94 (s, 6H, 6 pyrazole-H), 11.93 (s, 6H, 6 NH). *Anal.* Calcd. for C₁₃₂H₁₀₈N₃₀O₆S₆: C, 65.98; H, 4.53; N, 17.49; S, 8.01. Found: C, 65.80; H, 4.69; N, 17.25; S, 7.78.

1,2,3,4,5,6-Hexakis((2-((2-(4-(5-methyl-1-phenyl-1H-pyrazol-4-yl)thiazol-2-yl)hydrazono)methyl)phenoxy)methyl)benzene 19b. Orange powder, (67% yield), mp. 286–288°C; IR: (potassium bromide) 3431 (NH), 1597 (C=N) cm⁻¹; ¹H-NMR: δ 2.39 (s, 18H, 6 CH₃), 5.57 (s, 12H, 6 CH₂O), 6.76 (s, 6H, 6 thiazole-H), 6.79–7.80 (m, 54H, ArH), 7.95 (s, 6H, 6 CH=N), 8.35 (s, 6H, 6 pyrazole-H), 11.89 (s, 6H, NH). Anal. Calcd. for C₁₃₂H₁₀₈N₃₀O₆S₆: C, 65.98; H, 4.53; N, 17.49; S, 8.01. Found: C, 66.21; H, 4.76; N, 17.64; S, 8.26.

Synthesis of poly(4,5-dihydro-1,3,4-thiadiazolyl) derivatives 20b, 21a,b, and 22a. *General procedure*. A solution of each of poly(thiosemicarbazones) 9b, 14a,b, or 15a in acetic anhydride (20 mL) was heated under reflux for 7–10 h. The reaction mixture was then cooled and poured into crushed ice. The solid residue was collected and recrystallized from ethanol/ethylacetate to give poly(4,5-dihydro-1,3,4-thiadi azolyl) derivatives 20, 21a,b, and 22.

N,*N'*,*N''*-(5,5',5''-(2,2',2''-(Benzene-1,3,5-triyltris(methylene) tris(oxy)tris-(benzene-2,1-diyl))tris(4-acetyl-4,5-dihydro-1,3,4-thiadiazole-5,2-diyl))-triacetamide 20. Colorless crystal, (64% yield), mp. 198–200°C; IR: (potassium bromide) 3427 (NH), 1710 (C=O), 1662 (C=O), 1606 (C=N) cm⁻¹; ¹H-NMR: δ 2.03, 2.38 (2 s, 18H, 6 CH₃CO), 5.14 (s, 6H, 3

CH₂O), 6.77 (s, 3H, 3 thiadiazole-H), 6.96–7.47 (m, 15H, ArH), 11.69 (s, 3H, 3 NH); ms: m/z (%) 951 (11, M⁺), 920 (66.4), 800 (100), 400 (24.6), 232 (19), 139 (7.6). Anal. Calcd. for C₄₅H₄₅N₉O₉S₃: C, 56.77; H, 4.76; N, 13.24; S, 10.10. Found: C, 56.65; H, 4.52; N, 13.09; S, 9.87.

N,*N*',*N*'''-(5,5',5'',5'''-(4,4',4'',4'''-(Benzene-1,2,4,5tetrayltetrakis-(methylene))tetrakis(oxy)tetrakis(benzene-4,1diyl))tetrakis(4-acetyl-4,5-dihydro-1,3,4-thiadiazole-5,2-diyl)) tetraacetamide 21a. Colorless powder, (68% yield), mp. 242–245°C; IR: (potassium bromide) 3448 (NH), 1705 (C=O), 1644 (C=O), 1605 (C=N) cm⁻¹; ¹H-NMR: δ 1.98, 2.24 (2s, 24H, 8 CH₃CO), 5.43 (s, 8H, 4 CH₂O), 6.92 (s, 4H, 4 thiadiazole-H), 7.26–7.94 (m, 18H, ArH), 11.62 (s, 4H, 4 NH); ¹³C-NMR: δ 21.7, 22.4, 61.6, 67.2, 112.6, 120.8, 124.1, 128.5, 129.3, 146.6, 153.7, 167.3, 169.1; ms: m/z (%) 1242 (10.4, M⁺), 1148 (7.5), 991 (100), 815 (67.1), 249 (12.2), 58 (12.8). Anal. Calcd. for C₅₈H₅₈N₁₂O₁₂S₄: C, 56.02; H, 4.70; N, 13.52; S, 10.32. Found: C, 55.88; H, 4.57; N, 13.28; S, 10.19.

N,N',N'',N'''-(5,5',5'',5'''-(2,2',2'',2'''-(Benzene-1,2,4,5tetrayltetrakis-(methylene))tetrakis(oxy)tetrakis(benzene-2,1diyl))tetrakis(4-acetyl-4,5-dihydro-1,3,4-thiadiazole-5,2-diyl)) tetraacetamide 21b. Colorless powder, (66% yield), mp. 203–205°C; IR: (potassium bromide) 3444 (NH), 1728 (C=O), 1669 (C=O), 1607 (C=N) cm⁻¹; ¹H-NMR: δ 2.17, 2.38 (2 s, 24H, 8 CH₃CO), 5.20 (s, 8H, 4 CH₂O), 6.77 (s, 4H, 4 thiadiazole), 6.95–7.66 (m, 18H, ArH), 11.69 (s, 4H, NH); ms: *m/z* (%) 1242 (23.4, M⁺), 1127 (9.8), 991 (100), 815 (60.9), 297 (11.2), 60 (4.3). Anal. Calcd. for C₅₈H₅₈N₁₂O₁₂S₄: C, 56.02; H, 4.70; N, 13.52; S, 10.32. Found: C, 55.97; H, 4.66; N, 13.44; S, 10.27.

N,*N*'',*N*''',*N*'''',*N*''''',*S*'''',*S*''',*S*'''',*S*'''',*S*'''',*4*''',*4*''',*4*''',*4*'''', *4*'''''-(*Benzene-1,2,3,4,5,6-hexaylhexakis(methylene*))*hexakis(oxy) hexakis(benzene-4,1-diyl)*)*hexakis(4-acetyl-4,5-dihydro-1,3,4-thiadiazole-5,2-diyl)*)*hexa-acetamide* 22. Colorless powder, (67% yield), mp. 230–232°C; IR: (potassium bromide) 3433 (NH), 1702 (C=O), 1662 (C=O), 1607 (C=N) cm⁻¹; ¹H-NMR: δ 2.02, 2.16 (2 s, 36H, 12 CH₃CO), 5.17 (s, 12H, 6 CH₂O), 6.72 (s, 6H, 6 thiadiazole-H), 6.85–7.11 (m, 24H, ArH), 11.67 (s, 6H, 6 NH); ¹³C-NMR: δ 21.8, 22.5, 63.6, 65.4, 114.6, 126.6, 134.0, 137.3, 145.9, 157.9, 167.2, 169.2. *Anal.* Calcd. For C₈₄H₈₄N₁₈O₁₈S₆: C, 55.25; H, 4.64; N, 13.81; S, 10.54. Found: C, 55.18; H, 4.75; N, 13.72; S, 10.43.

Acknowledgments. Professor A. H. M. Elwahy and Dr. A. F. Darweesh gratefully acknowledge the Alexander von Humboldt Foundation for a research fellowship.

REFERENCES AND NOTES

 Vögtle, F.; Weber, E. Angew Chem Int Ed 1974, 13, 814.
 Hardy, A. D. U.; MacNicol, D. D.; Wilson, D. R. J Chem Soc Perkin Trans 1979, 2, 1011. [3] Foces-Foces, C.; Llamas-Saiz, A. L.; Claramunt, R. M.; Jagerovic, N.; Jimeno, M. L.; Elguero, J. J Chem Soc Perkin Trans 1995, 2, 1359.

[4] Christensen, C. A.; Bryce, M. R.; Batsanov, A. S.; Becher, J. Chem Commun 2000 331.

[5] Lambert, C.; Nöll, G. Chem Eur J 2002, 8, 3467.

[6] Yang, C.; Chen, X. M.; Yang, Y. S. Chem Commun 1997, 2041.

[7] Elwahy, A. H. M. Tetrahedron Lett 2001, 42, 5123.

[8] (a) Hoskins, B. F.; Robson, R.; Slizys, D. A. Angew Chem Int Ed 1997, 36, 2752; (b) Al-Smadi, M.; Ratrout, S. J Heterocyclic Chem 2004, 41, 887; (c) Al-Smadi, M.; Ratrout, S. Molecules 2004, 9, 957; (d) Al-Smadi, M. Asian J Chem 2007, 19, 1783; (e) Al-Smadi, M. J Heterocyclic Chem 2007, 44, 915; (f) Al-Smadi, M.; Mohammad, S. J Heterocyclic Chem 2009, 46, 201.

[9] (a) Steed, J. W.; Turner, D. R.; Wallace, K. J. Core Concepts in Supramoecular Chemistry and Nanochemistry; Wiley: Chichester, 2007;
(b) Steed, J. W.; Atwood, J. L. Supramolecular Chemistry, 2nd ed.; Wiley: Chichester, 2009.

[10] (a) Garcia-Frutos, E. M.; Omenat, A.; Barbera, J.; Serrano, J. L.; Gomez-Lor, B. J Mater Chem 2011, 21, 6831; (b) Gomez-Lor, B.; Alonso, B.; Omenat, A.; Serrano, J. L. Chem Commun 2006 5012; (c) Talarico, M.; Termine, R.; Garcia-Frutos, E. M.; Omenat, A.; Serrano, J. L.; Gomez-Lor, B.; Golemme, A. Chem Mater 2008, 20, 6589; (d) Cleuvenbergen, S. V.; Asselberghs, I.; Garcia-Frutos, E. M.; Gomez-Lor, B.; Clays, K.; Perez-Moreno, J. J Phys Chem C 2012, 116, 12312.

[11] Traber, B.; Wolff, J. J.; Rominger, F.; Oeser, T.; Gleiter, R.; Goebel, M.; Wortmann, R. Chem Eur J 2004, 10, 1227.

[12] Shoji, T.; Higashi, J.; Ito, S.; Okujima, T.; Yasunami, M.; Morita, N. Chem Eur J 2011, 17, 5116.

[13] Bruder, F. -K.; Hagen, R.; R lle, T.; Weiser, M. -S.; F cke, T. Angew Chem Int Ed 2011, 50, 4552.

[14] (a) Astruc, D.; Boisselier, E.; Ornelas, C. Chem Rev 2010, 110, 1857; (b) Vögtle, F.; Richardt, G.; Werner, N. Dendrimer Chemistry Concepts, Syntheses, Properties, Applications; Wiley VCH, 2009; (c) Nanjwade, B. K.; Bechra, H. M.; Derkar, G. K.; Manvi, F. V.; Nanjwade, V. K. Eur J Pharm Sci 2009, 38, 185; (d) Newkome, G. R.; Moorefield, C. N.; Vögtle, F. Dendritic Molecules: Concepts, Syntheses, Perspectives; VCH: New York, 1996; (e) Rajakumar, P.; Raja, S. Tetrahedron Lett 2008, 49, 6539; (f) Rajakumar, P.; Raja, S. Synth Commun 2009, 39, 3888.

[15] (a) Bonde, C. G.; Gaikwad, N. J Bioorg Med Chem 2004, 12, 2151; (b) Pontillo, J.; Chen, C. Bioorg Med Chem Lett 2005, 15, 1407; (c) Pontiki, E.; Hadjipavlou, L. D.; Chaviara, A. T.; Bolos, C. A. Bioorg Med Chem Lett 2006, 16, 2234; (d) Küçükgüzel, G.; Kocatepe, A.; Clercq, E. D.; Şahin, F.; Güllüce, M. Eur J Med Chem 2006, 41, 353; (e) Verma, A.; Saraf, S. K. Eur J Med Chem 2008, 43, 897; (f) Abdel-Wahab, B. F.; Mohamed, S. F.; Amr, A.; El-G, E.; Abdalla, M. M. Monatsh Chem 2008, 139, 1083.

[16] Penning, T. D.; Talley, J. J.; Bertenshaw, S. R.; Carter, J. S.; Collins, P. W.; Docter, S.; Graneto, M. J.; Lee, L. F.; Malecha, J. W.; Miyashiro, J. M.; Rogers, R. S.; Rogier, D. J.; Yu, S. S.; Anderson, G. D.; Burton, E. G.; Cogburn, J. N.; Gregory, S. A.; Koboldt, C. M.; Perkins, W. E.; Seibert, K.; Veenhuizen, A. W.; Zhang, Y. Y.; Isakson, P. C. J Med Chem 1997, 40, 1347.

[17] Terrett, N. K.; Bell, A. S.; Brown, D.; Ellis, P. Bioorg Med Chem Lett 1996, 6, 1819.

[18] Seltzmann, H. H.; Carroll, F. I.; Burgess, J. P.; Wyrick, C. D.; Burch, D. F. J Chem Soc Chem Commun 1995 1549.

[19] Mert, S.; Kasimogullari, R.; Ica, T.; Colak, F.; Altun, A.; Ok, S. Eur J Med Chem 2014, 78, 86.

[20] (a) El-Sayed, M. A. A.; Abdel-Aziz, N. I.; Abdel-Aziz, A. A. M.; El-Azab, A. S.; ElTahir, K. E. H. Bioorg Med Chem 2012, 20, 3306; (b) Singh, S. K.; Saibaba, V.; Rao, K. S.; Reddy, P. G.; Daga, P. R.; Rajjak, S. A.; Misra, P.; Rao, Y. K. Eur J Med Chem 2005, 40, 977; (c) Lee, K. Y.; Kim, J. M.; Kim, J. N. Tetrahedron Lett 2003, 44, 6737; (d) Sharma, P. K.; Kumar, S.; Kumar, P.; Kaushik, P.; Kaushik, D.; Dhingra, Y.; Aneja, K. R. Eur J Med Chem 2010, 45, 2650; (e) Tewari, A. K.; Singh, V. P.; Yadav, P.; Gupta, G.; Singh, A.; Goel, R. K.; Shinde, P.; Mohan, G. C. Bioorg Chem 2014, 56, 8.

[21] (a) Bondock, S.; Fadaly, W.; Metwally, M. A. Eur J Med Chem 2010, 45, 3692; (b) Isloor, A. M.; Kalluraya, B.; Shetty, P. Eur J Med Chem

2009, 44, 3784; (c) Prakash, O.; Hussain, K.; Kumar, R.; Wadhwa, D.; Sharma, C.; Aneja, K. R. Org Med Chem Lett 2011, 1, 1; (d) Li, M.; Zhao, B. X. Eur J Med Chem 2014, 85, 311; (e) Sridhar, R.; Perumal, P. T.; Etti, S.; Shanmugan, P. M. N.; Prabavathy, N.; Mathivanan, V. R. Bioorg Med Chem Lett 2004, 14, 6035; (f) Ningaiah, S.; Bhadraiah, U. K.; Doddaramappa, S. D.; Keshavamurthy, S.; Javarasetty, C. Bioorg Med Chem Lett 2014, 24, 245.

[22] Abdel-Aziz, M.; Abuo-Rahma, G.-D.; Hassan, A. A. Eur J Med Chem 2009, 44, 3480.

[23] Rathelot, P.; Azas, N.; El-Kashef, H.; Delmas, F.; Giorgio, C. D.; Timon-David, P.; Maldonado, J.; Vanelle, P. Eur J Med Chem 2002, 37, 671.

[24] Hashem, A. I.; Youssef, A. S. A.; Kandeel, K. A.; Abou-Elmagd, W. S. I. Eur J Med Chem 2007, 42, 934.

[25] (a) Lv, P.-C.; Li, H.-Q.; Sun, J.; Zhou, Y.; Zhu, H.-L. Bioorg Med Chem 2010, 18, 4606; (b) Li, X.; Liu, J. L.; Yang, X. H.; Lu, X.; Zhao, T. T.; Gong, H. B.; Hl, Z. Bioorg Med Chem 2012, 20, 4430; (c) Khloya, P.; Celik, G.; Ram, S.; Vullo, V.; Supuran, C. T.; Sharma, P. K. Eur J Med Chem 2014, 76, 284.

[26] (a) Hafez, H. N.; Hegab, M. I.; Ahmed-Farag, I. S.; El-Gazzar, A. B. A. Bioorg Med Chem Lett 2008, 18, 4538; (b) Schenone, S.; Brullo, C.; Bruno, O.; Bondavalli, F.; Ranise, A.; Filippelli, W.; Rinaldi, B.; Capuano, A.; Falcone, G. Bioorg Med Chem 2006, 14, 1698; (c) Salgm-Goksen, U.; Gokhan-Kelekci, N.; Goktas, O.; Koysal, Y.; Kilic, E.; Isik, S.; Aktay, G.; Ozalp, M. Bioorg Med Chem 2007, 15, 5738.

[27] (a) Sah, P.; Bidawat, P.; Seth, M.; Gharu, C. P. Arab J Chem 2014, 7, 181; (b) Almajan, G. L.; Barbuceanu, S. F.; Bancescu, G.; Saramet, I.; Saramet, G.; Draghici, C. Eur J Med Chem 2010, 45, 6139; (c) Alagawadi, K. R.; Alegaon, S. G. Arab J Chem 2011, 4, 465; (d) Swamy, S. N.; Priya, B. S.; Prabhuswamy, B.; Doreswamy, B. H.; Prasad, J. S.; Rangapa, K. S. Eur J Med Chem 2006, 41, 531; (e) Lamani, R. S.; Shetty, N. S.; Kamble, R. R.; Khazi, I. A. Eur J Med Chem 2009, 44, 2828; (f) Kadi, A. A.; Al-Abdullah, E. S.; Shehata, I. A.; Habib, E. E.; Ibrahim, T. M.; El-Emam, A. A. Eur J Med Chem 2010, 45, 5006.

[28] (a) Talath, S.; Gadad, A. K. Eur J Med Chem 2006, 41, 918;
(b) Kolavi, G.; Hegde, V.; Khazi, I. A.; Gadad, P. Bioorg Med Chem 2006, 14, 3069.

[29] (a) Rzeski, W.; Matysiak, J.; Kandefer-Szerszen, M. Bioorg Med Chem 2007, 15, 3201; (b) Noolvi, M. N.; Patel, H. M.; Singh, N.; Gadad, A. K.; Cameotra, S. S.; Badiger, A. Eur J Med Chem 2011, 46, 4411; (c) Karki, S. S.; Panjamurthy, K.; Kumar, S.; Nambiar, M.; Ramareddy, S. A.; Chiruvella, K. K.; Raghavan, S. C. Eur J Med Chem 2011, 46, 2109; (d) Yang, X. H.; Wen, Q.; Zhao, T. T.; Sun, J.; Li, X.; Xing, M.; Lu, X.; Zhu, H. L. Bioorg Med Chem 2012, 20, 1181; (e) Rajak, H.; Agarawal, A.; Parmar, P.; Thakur, B. S.; Veerasamy, R.; Sharma, P. C.; Kharya, M. D. Bioorg Med Chem Lett 2011, 21, 5735; (f) Kumar, D.; Maruthi, K. N.; Chang, K. H.; Shah, K. Eur J Med Chem 2010, 45, 4664; (g) Ibrahim, D. A. Eur J Med Chem 2009, 44, 2776.

[30] Clerici, F.; Pocar, D.; Guido, M.; Loche, A.; Perlini, V.; Brufani, M. J Med Chem 2001, 44, 931.

[31] Hasui, T.; Matsunaga, N.; Ora, T.; Ohyabu, N.; Nishigaki, N.; Imura, Y.; Igata, Y.; Matsui, H.; Motoyaji, T.; Tanaka, T.; Habuka, N.; Sogabe, S.; Ono, M.; Siedem, C. S.; Tang, T. P.; Gauthier, C.; De Meese, L. A.; Boyd, S. A.; Fukumoto, S. J Med Chem 2011, 54, 8616.

[32] (a) Sunil, D.; Isloor, A. M.; Shetty, P.; Satyamoorthy, K.; Prasad, A. S. B. Arab J Chem 2010, 3, 211; (b) Khan, I.; Ali, S.; Hameed, S.; Rama, N. H.; Hussain, M. T.; Wadood, A.; Uddin, R.; Ul-Haq, Z.;

Khan, A.; Ali, S.; Choudhary, M. I. Eur J Med Chem 2010, 45, 5200.

[33] Jatav, V.; Mishra, P.; Kashaw, S.; Stables, J. P. Eur J Med Chem 2008, 43, 1945.

[34] (a) Hu, Y.; Li, C.-Y.; Wang, X.-M.; Yang, Y.-H.; Zhu, H.-L. Chem Rev 2014, 114, 5572; (b) Chen, C. J.; Song, B. A.; Yang, S.; Xu, G. F.; Bhadury, P. S.; Jin, L. H.; Hu, D. Y.; Li, Q. Z.; Liu, F.; Xue, W.; Lu, P.; Chen, Z. Bioorg Med Chem 2007, 15, 3981; (c) Liu, F.; Luo, X. Q.; Song, B. A.; Bhadury, P. S.; Yang, S.; Jin, L. H.; Xue, W.; Hu, D. Y. Bioorg Med Chem 2008, 16, 3632; (d) Liu, X. H.; Shi, Y. X.; Ma, Y.; Zhang, C. Y.; Dong, W. L.; Pan, L.; Wang, B. L.; Li, B. J.; Li, Z. M. Eur J Med Chem 2009, 44, 2782; (e) Xu, W. M.; Yang, S.; Bhadury, P.; He, J.; He, M.; Gao, L. L.; Hu, D. Y.; Song, D. Y. Pestic Biochem Physiol 2011, 101, 6; (f) Zoumpoulakis, P.; Camoutsis, C.; Pairas, G.; Sokovic, M.; Glamoclija, J.; Potamitis, C.; Pitsas, A. Bioorg Med Chem 2012, 20, 1569.

[35] (a) Kocyigit, O.; Guler, E. J Incl Phenom Macrocycl Chem 2010, 67, 29; (b) Rajakumar, P.; Srisailas, M. Tetrahedron Lett 2002, 43, 1909; (c) Chand, D. K.; Bharadwaj, P. K. Tetrahedron Lett 1996, 37, 8443; (d) Finocchiaro, P.; Consiglio, G. A.; Imbrogiano, A.; Failla, S. Phosphorus Sulfur Silicon Relat Elem 2007, 182, 1689; (e) Elwahy, A. H. M. Tetrahedron Lett 2001, 42, 5123; (f) Abdelhamid, I. A.; Darweesh, A. F.; Elwahy, A. H. M. Tetrahedron Lett 2015, 56, 7085.

[36] (a) Nagarajan, K.; Arya, V. P.; Shenoy, S. J. J Chem Res 1986,
(S) 166, (M) 1401; (b) Shaaban, M. R.; Eldebss, T. M. A.; Darweesh, A. F.; Farag, A. M. J Heterocyclic Chem 2008, 45, 1739; (c) Shaaban, M. R.; Eldebss, T. M. A.; Darweesh, A. F.; Farag, A. M. J Chem Res 2010 8; (d) Goddard, C. J. J Heterocyclic Chem 1991, 28, 1607.