

Subscriber access provided by Binghamton University | Libraries

Total Synthesis of Biselyngbyaside

Eisuke Sato, Miho Sato, Yurika Tanabe, Naoya Nakajima, Akifumi Ohkubo, and Kiyotake Suenaga J. Org. Chem., Just Accepted Manuscript • Publication Date (Web): 07 Jun 2017 Downloaded from http://pubs.acs.org on June 8, 2017

Just Accepted

Article

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



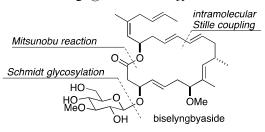
The Journal of Organic Chemistry is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Total Synthesis of Biselyngbyaside

Eisuke Sato, Miho Sato, Yurika Tanabe, Naoya Nakajima, Akifumi Ohkubo, Kiyotake Suenaga* Department of Chemistry, Faculty of Science and Technology, Keio University, Hiyoshi 3-14-1, Kohoku-ku, Yokohama 223-8522, Japan

E-mail: suenaga@chem.keio.ac.jp



Abstract: The first total synthesis of biselyngbyaside, an 18-membered macrolide glycoside, was achieved. The glycoside bond was introduced using the Schmidt method before construction of the 18-membered ring due to the instability of the conjugated diene and the β -hydroxy ester moiety. The macrolactone ring was constructed using the Mitsunobu reaction followed by intramolecular the Stille coupling reaction.

Introduction:

Biselyngbyaside (1, Figure 1), an 18-membered macrolide glycoside, was isolated from the marine cyanobacterium Lyngva sp. collected at Okinawa¹. Biselyngbyaside and its aglycone biselyngbyolide B $(2)^2$ show growth-inhibitory activity against HeLa and HL60 cells. Furthermore, 1 inhibited RANKL-induced osteoclastogenesis and induced apoptosis of mature osteoclasts at a low concentration³. Recently, we investigated whether they inhibit the ATPase activities of SERCA1a and 2a, and determined the X-ray crystal structures with SERCA1a⁴. The X-ray crystal structures showed that the 1,3-diene moiety and the side chain of biselyngbyasides play important roles in their interaction with SERCA. In fact, the activities of biselyngbyaside $C^{2,5}$, in which the 1,3-diene moiety is modified, against HeLa cells and SERCA1a are much weaker than those of 1 and 2 (Figure 1). The growth-inhibitory activity may depend on the affinity to SERCAs. However, the role of the sugar moiety and the differences is IC_{50} values and K_i values between biselyngbyaside and biselyngbyolide B have not been clarified. Because of the instability of the 18-membered ring structure of biselyngby asides, it is difficult to synthesize their artificial analogs using natural products. Therefore, little is known about the structure-activity relationships, especially on the sugar moiety. Synthetic studies of biselyngbyasides have been reported by several groups⁶ and total syntheses of biselyngbyolide B were achieved by Goswami's group^{7a} and our group^{7b}. However, the total synthesis of biselyngbyaside and its analogs with a sugar moiety has not yet been achieved. Herein, we report the first total synthesis of biselyngbyaside.

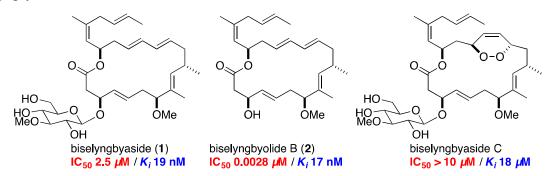


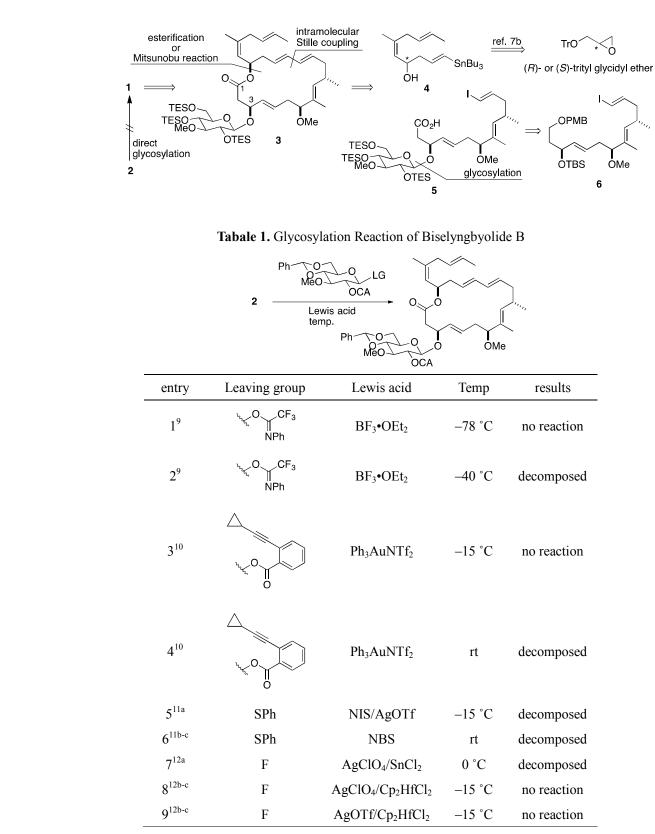
Figure 1. Structures and biological activities of biselyngbyasides. IC_{50} is the growth-inhibitory activities against HeLa cells. K_i is the ATPase-inhibitory activity against SERCA1a.

Result and Discussion:

To synthesize biselyngbyaside, we first tried a direct conversion to glycoside from its aglycone biselyngbyolide B. However, even with the use of various conditions⁹⁻¹² (see Table 1) for the glycosylation reaction the glycoside bond could not be constructed. First, we employed imidate sugar as a glycosyl donor⁹. No desired products were obtained and the starting material was recovered at low temperature or decomposed at high temperature. Although we tested mild activator (Au catalysts¹⁰ (entry 3 and 4), NIS^{11a} (entry 5), NBS^{11b-c} (entry 6), Ag catalysts¹² (entry 7 to 9)), no desired compounds were detected. Possible explanations for why direct glycosylation did not work include the sensitivity of the macrolactone ring system under the reaction conditions and the low reactivity of the C3 hydroxy group. Especially, intramolecular hydrogen bonding between the C3 hydroxy group and the C1 carbonyl oxygen interfere with functionalization of alcohol¹³.

Based on the results described above, our retrosynthetic analysis is shown in Scheme 1. The macrolactone ring was planned to be constructed using an intramolecular Stille coupling reaction. The cyclization precursor **3** would be obtained from stannane 4^{7b} and carboxylic acid **5**. Their two components could be connected *via* esterification or a Mitsunobu reaction. The stannane **4** was derived from the chiral glycidol derivative^{7b}. The glycoside moiety could be introduced before the connection of stannane and vinyl iodide.

Scheme 1. Retrosynthetic Analysis of Biselyngbyaside



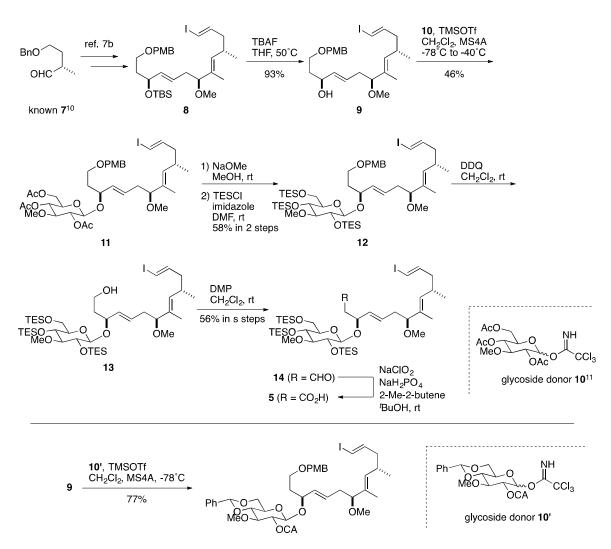
All reactions were performed with activated MS4A. LG: leaving group, CA: chloroacetyl

The synthesis of carboxylic acid 5 was started from known aldehyde 7¹⁴, which was synthesized from

1,3-propanediol in 7 steps (Scheme 2). Aldehyde 7 was converted to vinyl iodide 8 in 14 steps^{7b}. The TBS group was removed using tetrabutylammonium fluoride (93%) to obtain the alcohol 9. Next, we tried the glycosylation reaction. At first, we used glycoside donor 10', which was protected by chloroacetyl group (C2 position) and benzylidene acetal (C4 and C6 position), and the desired glycoside was obtained in good yield (77%). Glycosylation of alcohol 9 was much faster than that of 2 because of the effect of the β -carbonyl group (see Tabale 1 entry 1). Unfortunately, the chloroacetyl group could not be removed at the last stage. Thus, we selected to change the protecting group after glycosylation reaction.

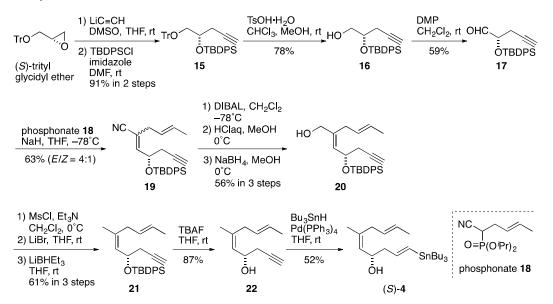
The alcohol **9** and glycoside donor 10^{15} were connected using Schmidt conditions⁸ (46%). In this glycosylation reaction, only β -glycoside was obtained due to the effect of the neighboring acetyl group, and the stereochemistry of the anomeric carbon was determined by the coupling constant of the anomer proton ($J_{1-2} = 7.8$ Hz). At this stage, the acetyl protecting groups on the sugar moiety should be exchanged with TES groups, which can be removed under mild conditions in the last stage of synthesis. So, the acetyl groups were removed by methanolysis and the resulting triol was converted to TES ether **12** (58% in 2 steps). The PMB group was cleaved by DDQ and the primary alcohol **13** was oxidized in two steps, using Dess-Martin periodinane¹⁶ (56% in 2 steps) and Pinnick conditions to give carboxylic acid **5**.

Scheme 2. Synthesis of Carboxylic Acid 5



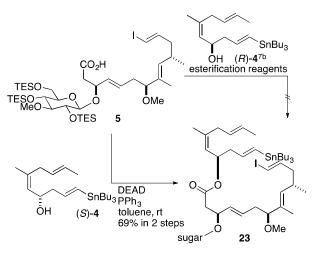
The synthetic route to stannane (*S*)-4 is shown in Scheme 3. Previously, we synthesized stannane (*R*)-4 from (*R*)-trityl glycidyl ether^{7b}. Therefore, we prepared (*S*)-4 from (*S*)-trityl glycigyl ether using same method. The commercially available glycidol derivative was treated with lithium acetylide followed by protection of the secondary alcohol to give TBDPS ether 15 (91% in 2 steps). The trityl group was removed (78%) and the obtained alcohol 16 was oxidized by Dess-Martin periodinane to synthesize aldehyde 17 (59%). Next, the side chain moiety was introduced using the corresponding phosphonate 18 with sodium hydride (63%) to afford alkene 19 as an inseparable mixture of isomers (*E*/*Z* = 4:1). The nitrile group in 19 was converted by a three-step procedure: i) DIBAL reduction, ii) hydrolysis in the presence of acid catalyst, and iii) sodium borohydride reduction to give alcohol 20. In our previous study^{7b}, partial isomerization of the olefin occurred, but an examination of the reaction conditions greatly improved the results without isomerization of the olefin. The undesired isomer could be completely removed in this stage. Deoxygenation of the alcohol 20 gave good results in the reaction with methanesulfonyl chloride, lithium bromide and lithium triethyl borohydride (61% in 3 steps) to provide 21. Finally, the TBDPS group in 21 was removed by tetrabutylammonium fluoride (87%) and the hydrostannation of 22 using tributyltin hydride with palladium catalyst gave (*S*)-4.

Scheme 3. Synthesis of Stannane 4



To connect the two components [carboxylic acid **5** and stannane (R)-**4**^{7b} (Scheme 4)], we tried the esterification reaction. In the synthesis of biselyngbyolide B^{7b}, Shiina esterification¹⁷ proceeded smoothly and the corresponding ester was obtained in good yield. However, with the use of carboxylic acid **5** with a sugar moiety, the desired ester could not be obtained at all. Although we tried various conditions for esterification (using Yamaguchi reagent, EDCI or CDI as a condensation reagent, and DMAP, DMAPO or DMAP•HCl as a catalyst), only the starting material was recovered. Therefore, we selected the Mitsunobu reaction¹⁸ between stannane (S)-**4** and carboxylic acid **5** to obtain ester **23**. As a result, the Mitsunobu reaction proceeded smoothly under the usual reaction conditions using diethyl azodicarboxylate with triphenylphosphine (69% in 2 steps).

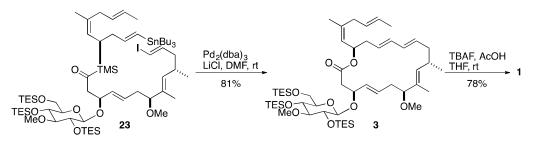
Scheme 4. Connection of carboxylic acid 5 and stannane 4



Finally, the 18-membered ring structure was constructed using an intramolecular Stille coupling reaction¹⁹, similar to the synthesis of biselyngbyolide B, to obtain TES-protected biselyngbyaside 3 (81%)

The Journal of Organic Chemistry

Scheme 5. Completion of the Synthesis



We next investigated the bioactivities of the synthetic compounds (Table 1). Synthetic biselyngbyaside (1) exhibited growth-inhibitory activity against HeLa cells (IC₅₀ 0.72 μ M). In contrast, the protected biselyngbyaside **3** completely lost bioactivity (IC₅₀ >30 μ M). The results showed that the bulky TES groups lowered the affinity with SERCA. This result was supported by a docking simulation study using the Glide program²⁰. The protected compounds **3** did not provide any docking poses using SERCA1a as a template (PDB ID: 4YCM⁴).

Table 1. Bioactivities of Synthetic Compounds

compounds	IC ₅₀ values (HeLa)
natural 1 4	2.5 μM
synthetic 1	$0.63\pm0.13~\mu M$
3	$> 30 \ \mu M$

Conclusion:

In conclusion, we achieved the first total synthesis of biselyngbyaside using a Schmidt glycosylation reaction in the early stage followed by the Mitsunobu reaction and the intramolecular Stille coupling reaction. In addition, we found that biselyngbyaside with protected sugar did not inhibit the growth of HeLa cells. The results showed that the sugar moiety also plays important roles in bioactivity.

Experimental Section:

General Information:

Chemicals and solvent were the best grade available and were used as received from commercial sources. Optical rotations were measured with a JASCO DIP-1000 polarimeter. ¹H NMR spectra were recorded on a JEOL JNM-AL400 (400 MHz), a JEOL JNM-A400 (400 MHz) or a JEOL JNM-ECX400 (400 MHz) instrument. Chemical shifts are reported δ values in parts per million relative to the residual solvent signal (CHCl₃: δ = 7.26 ppm; CD₃OD: δ = 3.31 ppm) and coupling constants are in hertz (Hz). The following abbreviations are used for spin multiplicity: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet and

br = broad. ¹³C NMR spectra were recorded on a JEOL JNM-AL400 (100 MHz), a JEOL JNM-A400 (100 MHz) or a JEOL JNM-ECX400 (100 MHz) instruments using CDCl₃ or CD₃OD as a solvent. Chemical shifts are reported in parts per million from the solvent signal (CDCl₃ :77.16 ppm; CD₃OD: 49.00 ppm). IR spectra were recorded on a recorded on a JASCOFT/IR-4200 instrument and reported in wavenumbers (cm⁻¹). High-resolution mass spectra were recorded by electrospray ionization (ESI) using time-of-flight (TOF) on a LCT premier EX spectrometer (Waters). Both TLC analysis and preparative TLC were conducted on E. Merck precoated silica gel 60 F254. Wako gel 60N and Nacalai Tesque silica gel 60 were used for column chromatography unless otherwise noted. Organic solvents for moisture-sensitive reactions were distilled from the following drying agents: THF (Na-benzophenone ketyl), toluene (Na), CH₂Cl₂ (P₂O₅), MeOH (calcium hydride). Anhydrous DMF was used as obtained from commercial supply. All moisture-sensitive reactions were performed under an atmosphere of nitrogen, and the starting materials were azeotropically dried with benzene befor use.

Synthesis of Biselyngbaside:

(35,4E,7S,8E,10S,12E)-13-iodo-7-methoxy-1-((4-methoxybenzyl)oxy)-8,10-dimethyltrideca-4,8,12-trien-3-ol (9): To a solution of TBS ether **8** (27.4 mg, 43.6 μmol) in THF (0.3 mL) was added 1 M solution of TBAF (0.1 mL, 0.1 mmol). The reaction was stirred at 50 °C for 14 h, then quenched by addition of saturated aqueous NH₄Cl and extracted with EtOAc (3 × 10 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 3:1 to 2:1) to give alcohol **9** (20.8 mg, 40.4 μmol, 93%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.25 (d, J = 8.3 Hz, 2H), 6.87 (d, J = 8.3 Hz, 2H), 6.42 (dt, J = 14.7, 7.3 Hz, 1H), 6.00 (d, J = 14.7Hz, 1H), 5.61-5.50 (m, 2H), 5.10 (d, J = 9.3 Hz, 1H), 4.44 (s, 2H), 4.30 (m, 1H), 3.80 (s, 3H), 3.66 (m, 1H), 3.61 (m, 1H), 3.43 (t, J = 6.8 Hz, 1H), 3.15 (s, 3H), 2.79 (brs, 1H, OH), 2.51 (m, 1H), 2.34 (m, 1H), 2.18 (m, 1H), 2.03 (m, 1H), 1.97 (m, 1H), 1.83-1.78 (m, 2H), 1.52 (s, 3H), 0.98 (d, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 159.4, 145.1, 134.5, 134.1, 133.4, 130.2, 129.5, 127.4, 114.0, 87.1, 75.6, 73.1, 71.8, 68.4, 55.8, 55.4, 43.6, 36.9, 36.8, 31.9, 20.9, 11.0; IR (neat) 3447, 2922, 2858, 2357, 2341, 1611, 1510, 1457, 1363, 1300, 1246, 1173, 1092, 1035, 950, 820 cm⁻¹; HRMS-ESI: Exact mass calcd for C₂₄H₃₅INaO₄ [M+Na]⁺: 537.1478; found 537.1440; [α]₀^{24.5} +14.5 (c 0.88, CHCl₃).

(2R, 3R, 4S, 5R, 6R)-2-(acetoxymethyl)-6-(((3S, 4E, 7S, 8E, 10S, 12E)-13-iodo-7-methoxy-1-((4-methoxybenz yl)oxy)-8, 10-dimethyltrideca-4, 8, 12-trien-3-yl)oxy)-4-methoxytetrahydro-2H-pyran-3, 5-diyl diacetate (11): To a mixture of alcohol **9** (45.5 mg, 88.4 µmol), imidate **10** (48.6 mg, 105 µmol) and MS4A (339.3 mg) were added CH₂Cl₂ and stirred at room temperature for 30 min, then cooled to -78 °C. The solution was added 55 mM solution of TMSOTf (0.15 mL, 8.3 µmol). The reaction was stirred at -78 °C for 1 h and at -40 °C for 1.5 h, then quenched by addition of Et₃N and filtered. The filtrate was concentrated *in vacuo* and the residue was purified column chromatography on SiO₂ (hexane/EtOAc 3:1 to 2:1) to give glycoside **11** (33.3 mg, 40.8 µmol, 46%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.25 (d, J = 8.8 Hz, 2H) 6.86 (d, J = 8.8 Hz, 2H), 6.42

(dt, J = 14.2, 7.1 Hz, 1H), 5.98 (d, J = 14.2 Hz, 1H), 5.63 (m, 1H), 5.27 (dd, J = 8.3 Hz, 15.6 Hz, 1H), 5.11 (d, J = 9.8 Hz, 1H), 5.06 (dd, J = 9.8 Hz, 9.8 Hz, 1H), 4.98 (dd, J = 7.8, 9.3 Hz, 1H), 4.45 (d, J = 7.8 Hz, 1H), 4.40 (d, J = 7.8 Hz, 1H), 4.29 (m, 1H), 4.20 (dd, J = 4.9, 12.2 Hz, 1H), 4.09 (dd, J = 2.4, 12.2 Hz, 1H), 3.80 (s, 3H), 3.55-3.43 (m, 5H), 3.38 (s, 3H), 3.15 (s, 3H), 2.51 (m, 1H), 2.37 (m, 1H), 2.14 (m, 1H), 2.09 (s, 3H), 2.09 (s, 3H), 2.05 (s, 3H), 2.03-1.94 (m, 2H), 1.87 (m, 1H), 1.74 (m, 1H), 1.54 (s, 3H), 0.99 (d, J = 6.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 170.9, 170.0, 169.3, 159.2, 144.9, 133.8, 133.7, 131.6, 131.0, 130.9, 129.3, 113.9, 113.8, 97.7, 86.9, 81.5, 76.1, 75.7, 72.8, 72.0, 71.9, 69.1, 66.4, 62.6, 58.4, 55.9, 55.4, 43.6, 37.6, 35.8, 31.8, 21.1, 21.0, 20.9, 20.8; IR (neat) 2953, 2932, 2868, 1750, 1612, 1513, 1456, 1437, 1373, 1302, 1226, 1173, 1155, 1092, 1038, 970, 904, 823, 599 cm⁻¹ HRMS-ESI: Exact mass calcd for C₃₇H₅₃INaO₁₂ [M+Na]⁺: 839.2479; found 839.2462; [α]_D^{24.6} +0.1 (c 1.37, CHCl₃).

(((2R,3R,4S,5R,6R)-2-(((3S,4E,7S,8E,10S,12E)-13-iodo-7-methoxy-1-((4-methoxybenzyl)oxy)-8,10-dime thyltrideca-4,8,12-trien-3-yl)oxy)-4-methoxy-6-(((triethylsilyl)oxy)methyl)tetrahydro-2H-pyran-3,5-diyl)bis(ox y))bis(triethylsilane) (12): To a solution of glycoside 11 (33.3 mg, 40.8 µmol) in MeOH (0.5 mL) was added 2 M solution of NaOMe in MeOH (0.5 mL, 1 mmol) and stirred at room temperature for 12 h, then guenched by addition of DOWEX 50W and filtered. The filtrate was concentrated in vacuo to give triol (26.8 mg) as a colorless oil. The solution of triol in DMF (0.3 mL) was added imidazole (48.3 mg, 0.71 mmol) and TESCI (0.05 mL, 0.30 mmol). After stirring at room temperature for 3.5 h, the reaction was quenched by addition of H_2O and extracted with EtOAc (3 \times 10 mL). The combined organic layers were washed with brine, dried over Na_2SO_4 and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 15:1 to 10:1) to give TES ether 12 (24.3 mg, 23.5 μmol, 58% in 2 steps) as a colorless oil: ¹H NMR (400 MHz, $CDCl_3$) δ 7.25 (d, J = 8.8 Hz, 2H), 6.86 (d, J = 8.8 Hz, 2H), 6.43 (dt, J = 14.6, 7.8 Hz, 1H), 5.98 (d, J = 14.6 Hz, 1H), 5.58 (dt, J = 14.6, 6.8 Hz, 1H), 5.28 (dd, J = 8.8, 14.6 Hz, 1H), 5.10 (d, J = 9.8 Hz, 1H), 4.41 (s, 2H), 4.29 (dd, J = 7.8, 14.1 Hz, 1H), 4.23 (d, J = 7.3 Hz, 1H), 3.80 (s, 3H), 3.69 (dd, J = 4.9, 11.2 Hz, 1H), 3.52 (s, 3H), 3.52-3.44 (m, 3H), 3.39 (dd, J = 5.9, 7.3 Hz, 1H), 3.34 (dd, J = 7.8, 7.8 Hz, 1H), 3.15 (s, 3H), 3.06 (m, 1H), 2.93 (dd, J = 8.8, 8.8 Hz, 1H), 2.50 (m, 1H), 2.36 (m, 1H), 2.14 (m, 1H), 2.04-1.95 (m, 3H), 1.77 (m, 1H), 1.52 (s, 3H), 0.99-0.93 (m, 27H), 0.67-0.54 (m, 21H); 13 C NMR (100 MHz, CDCl₃) δ 159.2, 145.0, 133.8, 133.7, 132.3, 131.1, 130.9, 129.4, 129.3, 113.8, 98.2, 88.5, 86.9, 77.1, 75.8, 74.4, 72.7, 70.9, 67.3, 62.4, 61.8, 55.8, 55.4, 43.6, 37.3, 35.8, 31.9, 20.7, 11.2, 7.1, 7.0, 5.3, 5.2, 4.7 IR (neat) 2952, 2911, 2875, 2359, 1614, 1540, 1513, 1457, 1417, 1375, 1302, 1246, 1095, 1041, 1006, 971, 852, 815, 741 cm⁻¹ HRMS-ESI: Exact mass calcd for $C_{49}H_{89}INaO_9Si_3$ [M+H]⁺: 1055.4757; found 1055.4751; $[\alpha]_D^{26.8}$ –2.4 (c 1.22, CHCl₃)

(3S,4E,7S,8E,10S,12E)-13-iodo-7-methoxy-3-(((2R,3R,4S,5R,6R)-4-methoxy-3,5-bis((triethylsilyl)oxy)-6 -(((triethylsilyl)oxy)methyl)tetrahydro-2H-pyran-2-yl)oxy)-8,10-dimethyltrideca-4,8,12-trien-1-ol (13): To a solution of TES ether 12 (26.6 mg, 25.7 μmol) in CH₂Cl₂ was added 1 M solution of pH 7 phosphate buffer (1 mL) and DDQ (13.6 mg, 59.9 μmol) and stirred at room temperature for 20 min. The reaction mixture was added DDQ (13.7 mg, 60.4 µmol) and stirred at room temperature for 1.5 h and added DDQ (28.9 mg, 127 µmol). After stirring at room temperature for 1 h, the reaction was quenched by addition of saturated aqueous solution of NaHCO₃ and extracted with CH₂Cl₂ (3×10 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 10:1 to 5:1) to give alcohol **13** (17.4 mg) as a mixture with anisaldehyde: ¹H NMR (400 MHz, CDCl₃) δ 6.43 (dt, *J* = 14.6, 7.8 Hz, 1H), 5.98 (d, *J* = 14.6 Hz, 1H), 5.58 (dt, *J* = 15.6, 7.3 Hz, 1H), 5.42 (dd, *J* = 8.3, 15.6 Hz, 1H), 5.11 (d, *J* = 9.3 Hz, 1H), 4.31 (dt, *J* = 13.7, 7.8 Hz, 1H), 4.21 (d, *J* = 7.8 Hz, 1H), 3.85 (dd, *J* = 2.0, 11.2 Hz, 1H), 3.65-3.61 (m, 2H), 3.52 (s, 3H), 3.42 (dd, *J* = 6.8, 6.8 Hz, 1H), 3.37 (dd, *J* = 8.8, 9.0 Hz, 1H), 3.35 (dd, *J* = 7.8, 8.8 Hz, 1H), 3.18 (m, 1H), 3.16 (s, 3H), 2.95 (dd, *J* = 9.0, 9.0 Hz, 1H), 2.72 (brs, 1H, OH), 2.51 (m, 1H), 2.36 (m, 1H), 2.16 (m, 1H), 2.05-1.95 (m, 2H), 1.73-1.71 (m, 2H), 1.53 (d, *J* = 1.0 Hz, 3H), 0.99-0.94 (m, 27H), 0.69-0.58 (m, 21H); ¹³C NMR (100 MHz, CDCl₃) δ 145.0, 133.8, 133.5, 132.8, 131.1, 130.3, 99.6, 88.2, 86.8, 77.4, 76.1, 75.6, 62.9, 61.9, 59.4, 55.9, 55.7, 43.7, 38.3, 37.1, 31.9, 20.7, 11.2, 7.1, 6.9, 5.3, 5.3, 4.5; HRM-ESI: Exact mass calcd for C₄₁H₈₁INaO₈Si₃ [M+Na]⁺: 935.4182; found 935.4214.

(3S,4E,7S,8E,10S,12E)-13-iodo-7-methoxy-3-(((2R,3R,4S,5R,6R)-4-methoxy-3,5-bis((triethylsilyl)oxy)-6 -(((triethvlsilvl)oxy)methvl)tetrahvdro-2H-pvran-2-vl)oxy)-8,10-dimethvltrideca-4,8,12-trienal (14): To a solution of alcohol 13 (17.4 mg, mixture with anisaldehyde) in CH_2Cl_2 (0.5 mL) was added Dess-Martin periodinane (23.1 mg, 54.5 µmol). After stirring at room temperature for 30 min, the mixture was added Dess-Martin periodinane (21.4 mg, 50.5 µmol) and stirred at room temperature for 1 h. The reaction was quenched by addition of saturated aqueous solution of $Na_2S_2O_3$ and extracted with EtOAc (3 × 10 mL). The combined organic layers were washed with saturated aqueous solution of NaHCO₃ and brine, dried over Na_2SO_4 and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 10:1 to 5:1) to give aldehyde 14 (13.2 mg, 14.5 μ mol, 56% in 2 steps) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 9.71 (t, J = 2.9 Hz, 1H), 6.42 (dt, J = 14.6, 7.3 Hz, 1H), 5.98 (d, J = 14.6 Hz, 1H), 5.71 (dt, J = 15.1, 7.3 Hz, 1H), 5.35 (dd, J = 8.8, 15.1 Hz, 1H), 5.11 (d, J = 9.3 Hz, 1H), 4.77 (ddd, J = 5.4, 8.8, 13.7 Hz, 1H), 4.27 (d, J = 7.8 Hz, 1H), 3.81 (dd, J = 1.5, 10.7 Hz, 1H), 3.69 (dd, J = 5.4, 10.7 Hz, 1H), 3.52 (s, 3H), 3.48-3.39 (m, 2H), 3.34 (dd, *J* = 7.8, 8.8 Hz, 1H), 3.15 (s, 3H), 3.10 (m, 1H), 2.94 (dd, *J* = 8.8, 8.8 Hz, 1H), 3.15 (s, 3H), 3.10 (m, 2H), 3.14 (dd, *J* = 8.8, 8.8 Hz, 1H), 3.15 (s, 3H), 3.10 (m, 2H), 3.14 (dd, *J* = 8.8, 8.8 Hz, 1H), 3.15 (s, 3H), 3.10 (m, 2H), 3.14 (dd, *J* = 8.8, 8.8 Hz, 1H), 3.15 (s, 3H), 3.10 (m, 2H), 3.14 (dd, *J* = 8.8, 8.8 Hz, 1H), 3.15 (s, 3H), 3.10 (m, 2H), 3.14 (s, 3H), 3.15 (s, 3H), 3.16 (s, 3H) 1H), 2.65 (m, 1H), 2.53-2.48 (m, 2H), 2.38 (m, 1H), 2.16 (m, 1H), 2.03 (m, 1H), 1.96 (m, 1H), 1.53 (d, J = 1.0Hz, 3H), 0.99-0.90 (m, 27H), 0.69-0.55 (m, 21H); ¹³C NMR (100 MHz, CDCl₃) δ 201.4, 145.0, 133.9, 133.6, 133.2, 129.4, 98.6, 88.3, 77.4, 77.2, 75.7, 75.6, 72.5, 70.9, 62.4, 61.9, 55.9, 49.2, 43.7, 37.3, 31.9, 20.8, 11.2, 7.1, 6.9, 5.3, 5.2, 4.7; IR (neat) 2954, 2911, 2876, 2356, 2338, 1732, 1716, 1698, 1558, 1540, 1520, 1507, 1472, 1456, 1081, 1008, 969, 808, 738 cm⁻¹; HRMS-ESI: Exact mass calcd for C₄₁H₇₉INaO₈Si₃ [M+Na]⁺: 933.4025; found 933.4037; $[\alpha]_D^{24.0}$ –5.5 (c 0.66, CHCl₃).

(3S,4E,7S,8E,10S,12E)-13-iodo-7-methoxy-3-(((2R,3R,4S,5R,6R)-4-methoxy-3,5-bis((triethylsilyl)oxy)-6 -(((triethylsilyl)oxy)methyl)tetrahydro-2H-pyran-2-yl)oxy)-8,10-dimethyltrideca-4,8,12-trienoic acid (5): To a

The Journal of Organic Chemistry

solution of aldehyde 14 (13.2 mg, 14.5 μ mol) in ^{*i*}BuOH (1 mL) was added 2-Me-2-butene (0.5 mL), 1 M aqueous solution of NaH₂PO₄ (1 mL) and 1 M aqueous solution of NaClO₂ (0.5 mL). After stirring at room temperature for 40 min, the mixture was diluted with EtOAc and H₂O and extracted with EtOAc (3 × 10 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue (15.5 mg) was used to the next reaction without further purification.

(*S*)-tert-butyldiphenyl((1-(trityloxy)pent-4-yn-2-yl)oxy)silane (**15**): To a stirred suspension of lithium acetylide ethylenediamine complex (1.5 g, 16.3 mmol) in DMSO (10 mL) was added a solution of (*S*)-(-)-trityl glycidyl ether (2.4 g, 7.6 mmol) in THF (10 mL) at room temperature. After stirring for 2 h, the mixture was diluted with saturated aqueous solution of NH₄Cl at 0 °C, and extracted with EtOAc (3×50 mL). The combined organic layers were washed with water and brine, dried over Na₂SO₄ and concentrated *in vacuo* to give crude alcohol (2.82 g) and the crude alcohol was used for the next reaction without further purification. To a solution of the crude alcohol in DMF (5 mL) was added imidazole (1.01 g, 14.8 mmol) and TBDPSCI (2.2 mL, 8.6 mmol). The reaction was stirred at room temperature for 2 h, then quenched by addition of water and extracted with EtOAc (3×100 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 30:1 to 25:1) to give TBDPS ether 15 (4.0 g, 6.9 mmol, 91% in 2 steps) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(S)-2-((tert-butyldiphenylsilyl)oxy)pent-4-yn-1-ol (16): To a solution of TBDPS ether 15 (4.0 g, 6.9 mmol) in CHCl₃ (10 mL) and MeOH (10 mL) was added TsOH•H₂O (247.9 mg, 1.30 mmol). The reaction was stirred at room temperature for 1 h, then quenched by addition of saturated aqueous solution of NH₄Cl and extracted withe EtOAc (3×100 mL). The combined organic layers were washed with saturated aqueous solution of NaHCO₃, water and brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 20:1 to 10:1 to 5:1) to give alcohol 16 (1.83 g, 5.41 mmol, 78%) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(*S*)-2-((tert-butyldiphenylsilyl)oxy)pent-4-ynal (**17**): To a solution of alcohol **16** (777.7 mg, 2.30 mmol) in CH₂Cl₂ (7 mL) was added Dess-Martin periodinane (1.95 g, 4.60 mmol). After stirring at room temperature for 30 min, the mixture was added Dess-Martin periodinane (1.88 mg, 4.43 mmol) and stirred at room temperature for 10 min. The reaction was quenched by addition of saturated aqueous solution of Na₂S₂O₃ and extracted with EtOAc (3 × 100 mL). The combined organic layers were washed with saturated aqueous solution of NaHCO₃ and brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 30:1 to 20:1) to give aldehyde **17** (455.7 mg, 1.35 mmol, 59%) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(S,E)-2-((E)-but-2-en-1-yl)-4-((tert-butyldiphenylsilyl)oxy)hept-2-en-6-ynenitrile (19): To a solution of

phosphonate **18** (404.6 mg, 1.56 mmol) in THF (12 mL) was added NaH (60% in oil, 64.2 mg, 1.61 mmol) at 0 °C and warmed to room temperature. After stirring for 15 min at room temperature, the mixture was cooled to -78 °C and added the solution of aldehyde **17** (455.7 mg, 1.35 mmol) in THF (3 mL, 1 mL). The reaction mixture was stirred at -78 °C for 3 h, diluted with saturated aqueous NH₄Cl, and extracted with EtOAc (3×20 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc = 30:1) to give nitrile **19** (349.9 mg, 0.85 mmol, 63%, E/Z = ca. 4:1) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(S,E)-2-((E)-but-2-en-1-yl)-4-((tert-butyldiphenylsilyl)oxy)hept-2-en-6-yn-1-ol (**20**): To a solution of nitrile **19** (349.9 mg, 0.85 mmol) in CH₂Cl₂ (10 mL) cooled at -78 °C was added DIBAL (1.0 M solution in hexane, 2.5 mL, 2.5 mmol). After stirring for 10 min, the mixture was diluted with MeOH (2 mL) and warmed to room temperature. The precipitate was filtered by Celite pad and the filtrate was concentrated to give imine as a colorless oil. The solution of the imine in THF (10 mL) coold at 0 °C was added HClaq (1.0 M, 1 mL, 1 mmol). After stirring for 15 min, the mixture was diluted with saturated aqueous solution of NaHCO₃ and extracted with EtOAC (3 × 20 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo* to give aldehyde. To the solution of the aldehyde in MeOH (5 mL) cooled at 0 °C was added NaBH₄ (43.9 mg, 1.16 mmol). The reaction mixture was stirred for 1 h, diluted with saturated aqueous solution of NaHCO₃ then extracted with EtOAc (3 × 20 mL). The combined organic layers was washed with brine, dried over Na₂SO₄ and concentrated *in vacuo* to give aldehyde. To the solution of the aldehyde in MeOH (5 mL) cooled at 0 °C was added NaBH₄ (43.9 mg, 1.16 mmol). The reaction mixture was stirred for 1 h, diluted with saturated aqueous solution of NaHCO₃ then extracted with EtOAc (3 × 20 mL). The combined organic layer was washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc = 10:1 to 8:1) to give alcohol **20** (181.3 mg, 0.48 mmol, 56% in 3 steps) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

tert-butyl(((S,5Z,8E)-6-methyldeca-5,8-dien-1-yn-4-yl)oxy)diphenylsilane (21): To a solution of alcohol 20 (181.3 mg, 0.48 mmol) in CH₂Cl₂ (2 mL) cooled at 0 °C was added Et₃N (0.3 mL, 2.16 mmol) and MsCl (0.1 mL, 1.29 mmol). After stirring for 2.5 h, the mixture was diluted with water and extracted with EtOAc ($3 \times 10 \text{ mL}$). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo* to give mesylate. The mesylate was dissolved to THF (2 mL) and added LiBr (143.3 mg, 1.65 mmol) at room temperature. The mixture was stirred for 1 h, diluted with water and extracted with EtOAc ($3 \times 10 \text{ mL}$). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo* to give bromide. To a solution of the bromide in THF (3 mL) cooled at 0 °C was added lithium triethylborohydride solution (1.0 M in THF, 1.5 mL, 1.5 mmol). The reaction mixture was extracted with EtOAc ($3 \times 10 \text{ mL}$). The combined organic layers were washed with saturated aqueous solution of NaHCO₃ and brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc = 50:1) to give diene 21 (117.6 mg, 0.29 mmol, 61% in 3 steps) as a colorless oil: The

The Journal of Organic Chemistry

analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(S,5Z,8E)-6-methyldeca-5,8-dien-1-yn-4-ol (22): To a solution of diene 21 (117.6 mg, 0.29 mmol) in THF (1.5 mL) was added TBAF solution (1.0 M in THF, 0.6 mL, 0.6 mL). After stirring for 18 h at room tempetature, the mixture was diluted with saturated aqueous solution of NH₄Cl and extracted with EtOAc (3 × 10 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc = 10:1 to 5:1) to give alcohol 22 (41.5 mg, 0.25 mmol, 87%) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(S, 1E, 5Z, 8E)-6-methyl-1-(tributylstannyl)deca-1, 5, 8-trien-4-ol ((S)-4): To a degassed solution of alcohol **22** (41.5 mg, 0.25 mmol) in THF (1 mL) was added Pd(PPh₃)₄ (18.1 mg, 15.7 µmol) and Bu₃SnH (0.1 mL, 0.37 mmol). After stirring at 0 °C for 30 min, the solvent was removed *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/ EtOAc 20:1 to 10:1) to give stannane (S)-4 (59.3 mg, 0.13 mmol, 52%) as a colorless oil: The analytical data are identical with that of enantiomer^{7b}, except for the specific rotation.

(R, 1E, 5Z, 8E)-6-methyl-1-(tributylstannyl)deca-1, 5, 8-trien-4-yl

(3S,4E,7S,8E,10S,12E)-13-iodo-7-methoxy-3-(((2R,3R,4S,5R,6R)-4-methoxy-3,5-bis((triethylsilyl)oxy)-6-(((tri ethylsilyl)oxy)methyl)tetrahydro-2H-pyran-2-yl)oxy)-8,10-dimethyltrideca-4,8,12-trienoate (23): To a solution of stannane (S)-4 (7.7 mg, 16.9 μ mol), carboxylic acid 5 (15.5 mg, crude) and PPh₃ (7.8 mg, 29.7 μ mol) in toluene (0.3 mL) was added 2.2 M solution of DEAD in toluene (0.02 mL, 44 µmol). After stirring at room temperature for 16 h, the mixture was diluted with EtOAc and H₂O and extracted with EtOAc (3×10 mL). The combined organic layers were washed with brine, dried over Na_2SO_4 and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 15:1 to 10:1) to give ester 23 (13.6 mg, 10.0 µmol, 69% in 2 steps) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 6.43 (dt, J = 14.6, 7.8 Hz, 1H) 5.98 (d, J = 14.6 Hz, 1H), 5.97 (d, J = 19.0 Hz, 1H), 5.78 (dt, J = 19.0, 6.3 Hz, 1H), 5.68 (dt, J = 14.6, 6.8 Hz, 1H), 5.55 (m, 1H), 5.44 (m, 1H), 5.31-5.25 (m, 2H), 5.11 (d, J = 9.3 Hz, 1H), 5.11 (d, J = 9.3 Hz, 1H), 4.63 (m, 1H), 4.25 (d, J = 7.8 Hz, 1H), 3.81 (dd, J = 1.0, 11.2 Hz, 1H), 3.67 (dd, J = 5.4, 11.2 Hz, 1H), 3.52 (s, 3H), 3.44-3.32 (m, 3H), 3.14 (s, 3H), 3.08 (m, 1H), 2.93 (dd, *J* = 8.8, 8.8 Hz, 1H), 2.83 (m, 1H), 2.71-2.66 (m, 2H), 2.53-2.41 (m, 3H), 2.37-2.29 (m, 2H), 2.13 (m, 1H), 2.03-1.97 (m, 2H), 1.67-1.41 (m, 15H), 1.35-1.21 (m, 9H), 1.00-0.80 (m, 27H), 0.71-0.59 (m, 21H); ¹³C NMR (100 MHz, CDCl₃) δ 169.6, 145.0, 143.6, 139.6, 133.8, 133.8, 133.5, 132.0, 129.5, 128.3, 126.6, 124.0, 98.3, 88.5, 86.7, 77.4, 75.9, 75.7, 73.7, 71.0, 70.4, 62.4, 61.8, 55.9, 43.7, 41.4, 37.6, 36.1, 31.9, 29.9, 29.3, 29.2, 27.4, 23.4, 20.7, 18.0, 13.9, 11.3, 9.5, 7.1, 7.0, 5.3, 5.2, 4.7; IR (neat) 2954, 2925, 2875, 2854, 2359, 2340, 1733, 1457, 1417, 1376, 1338, 1239, 1178, 1151, 1082, 1006, 965, 852, 813, 741, 689, 669 cm⁻¹; HRMS-ESI C₆₄H₁₂₁INaO₉Si₃Sn [M+Na]⁺: 1387.6283; found 1387.6252; $[\alpha]_D^{22.2}$ –0.68 (c 0.68, CHCl₃).

(4S, 5E, 8S, 9E, 11S, 13E, 15E) - 8 - methoxy - 4 - (((2R, 3R, 4S, 5R, 6R) - 4 - methoxy - 3, 5 - bis) ((triethylsilyl) oxy) - 6 - (((triethylsilyl) oxy) - 6 - ((triethylsilyl) oxy) - 6 - (triethylsilyl) oxy) - (triethylsilyl) oxy) - (triethylsilyl) oxy) -

riethylsilyl)oxy)methyl)tetrahydro-2H-pyran-2-yl)oxy)-9,11-dimethyl-18-((1Z,4E)-2-methylhexa-1,4-dien-1-yl) oxacyclooctadeca-5,9,13,15-tetraen-2-one (**3**): To a degassed solution of ester **23** (13.6 mg, 10.0 µmol) in DMF (6 mL) was added LiCl (3.8 mg, 90 µmol) and Pd₂(dba)₃ (0.6 mg, 0.7 µmol). After stirred at room temperature for 4 h, the mixture was diluted by Et₂O and H₂O and extracted with Et₂O (3 × 10 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by column chromatography on SiO₂ (hexane/EtOAc 15:1 to 10:1) to give macrolactone **3** (7.7 mg, 8.1 µmol, 81%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 6.03-5.89 (m, 2H), 5.59-5.25 (m, 6H), 5.07-5.00 (m, 2H), 4.54 (dt, *J* = 15.6, 7.8 Hz, 1H), 4.20 (d, *J* = 7.8 Hz, 1H), 3.81-3.74 (m, 2H), 3.66-3.55 (m, 1H), 3.52 (s, 3H), 3.38-3.31 (m, 2H), 3.15 (s, 3H), 3.05 (m, 1H), 2.96-2.09 (m, 2H), 2.75-2.63 (m, 3H), 2.43-2.14 (m, 6H), 1.89-1.81 (m, 1H), 1.70-1.42 (m, 9H), 1.34-1.17 (m, 1H), 1.08-0.83 (m, 27H), 0.73-0.57 (m, 21H); ¹³C NMR (100 MHz, CDCl₃) δ 169.7, 138.7, 136.6, 135.1, 133.7, 132.3, 131.8, 131.3, 130.4, 128.2, 126.9, 126.6, 123.6, 99.1, 88.2, 87.7, 77.4, 75.6, 74.2, 70.8, 70.4, 62.0, 61.8, 55.4, 41.6, 40.3, 38.0, 36.7, 35.9, 32.6, 29.9, 23.6, 22.1, 18.0, 10.2, 9.5, 7.1, 7.0, 5.3, 5.2, 4.7; IR (neat) 2953, 2932, 2915, 2878, 2360, 2342, 1733, 1558, 1540, 1507, 1456, 1088, 1007, 969, 815, 741 cm⁻¹; HRMS-ESI: Exact mass calcd for C₅₂H₉₄NaO₉Si₃ [M+Na]⁺: 969.6103; found 969.6115; [α]_D^{25.8} –20.3 (c 0.39, CHCl₃).

Biselyngbyaside (1): To a solution of macrolactone 3 (7.2 mg, 7.6 µmol) in THF (0.3 mL) was added 1.8 M solution of AcOH in THF (0.04 mL, 72 µmol) and 1 M solution of TBAF in THF (0.06 mL, 60 µmol). The reaction was stirred at room temperature for 11.5 h and added 1 M solution of TBAF in THF (0.05 mL, 50 µmol). After stirring at room temperature for 4 h, the reaction was quenched by addition of saturated aqueous solution of NH₄Cl and extracted with EtOAc (3×10 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated *in vacuo*. The residue was purified by PTLC on SiO₂ [200 x 100 x 0.5, CHCl₃/MeOH 5:1] to give biselyngbyaside (3.6 mg, 6.0 µmol, 78%) as a colorless oil: ¹H NMR $(400 \text{ MHz, CD}_3\text{OD}) \delta 6.08-5.98 \text{ (m, 2H)}, 5.59-5.38 \text{ (m, 7H)}, 5.14 \text{ (d, } J = 8.8 \text{ Hz, 1H)}, 5.12 \text{ (d, } J = 8.8 \text{ Hz, 1H)},$ 4.51 m, 1H, 4.26 (d, J = 6.8 Hz, 1H), 3.86 (dd, J = 2.4, 11.7 Hz, 1H), 3.74 (dd, J = 4.6, 11.7 Hz, 1H), 3.64 (s, 3.86 (dd, J = 2.4, 11.7 Hz, 1H), 3.74 (dd, J = 4.6, 11.7 Hz, 1H), 3.64 (s, 3.86 (dd, J = 2.4, 11.7 Hz, 1H), 3.74 (dd, J = 4.6, 11.7 Hz, 1H), 3.64 (s, 3.86 (dd, J = 2.4, 11.7 Hz, 1H), 3.74 (dd, J = 4.6, 11.7 Hz, 1H), 3.64 (s, 3.86 (dd, J = 2.4, 11.7 Hz, 1H), 3.74 (dd, J = 4.6, 11.7 Hz, 1H), 3.64 (s, 3.86 (dd, J = 2.4, 11.7 Hz, 110 (dd, J = 4.6, 11.7 (dd, J = 4.6, 13H), 3.47 (m, 1H), 3.45 (dd, *J* = 9.3, 9.8 Hz, 1H), 3.22 (dd, *J* = 6.8, 9.3 Hz, 1H), 3.19 (m, 1H), 3.16 (s, 3H), 3.05 (dd, J = 9.3, 9.3 Hz, 1H), 2.93 (m, 1H), 2.76-2.68 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.34-2.23 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.34-2.23 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.34-2.23 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.34-2.23 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.34-2.23 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.34-2.23 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.57 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.57 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.54 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.54 (m, 2H), 2.57 (dd, J = 8.0, 14.9 Hz, 1H), 2.54 (m, 2H), 2.57 (m, 2H), 2.56H), 1.97 (m, 1H), 1.68 (s, 3H), 1.65 (d, J = 5.9 Hz, 3H), 1.56 (s, 3H), 1.04 (d, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CD₃OD) δ 172.1, 140.1, 138.5, 135.3, 133.3, 133.0, 132.8, 132.1, 131.9, 129.3, 127.9, 127.5, 124.9, 100.9, 89.1, 87.7, 77.7, 77.4, 74.6, 72.6, 70.7, 62.3, 61.0, 55.6, 43.0, 41.5, 39.6, 36.8, 36.6, 34.0, 23.6, 22.4, 18.0, 10.1; IR (neat) 2365, 2342, 1559, 1508, 1073 cm⁻¹; HRMS-ESI: Exact mass calcd for C₃₄H₅₂O₉Na $[M+Na]^+$: 627.3509; found 627.3510; $[\alpha]_D^{26.3}$ –42.9 (c 0.11, CHCl₃).

Cell Growth Analysis:

All cells were obtained from RIKEN Cell Bank. HeLa cells were cultured at 37° C with 5% CO₂ in DMEM (Nissui) supplemented with 10% heat-inactivated fetal bovine serum (FBS), 100 units/mL penicillin, 100 µg/mL streptomycin, 0.25 µg/mL amphotericin, 300 µg/mL L-glutamine, and 2.25 mg/mL NaHCO₃.

The Journal of Organic Chemistry

HeLa cells were seeded at 2×10^4 cells/well in 96-well plates (Iwaki) and cultured overnight. Various concentrations of compounds were then added, and cells were incubated for 72 h. Cell proliferation was measured by the MTT assay. Adriamycin was used as positive control (IC₅₀ value 0.5 μ M (HeLa cells)).

Docking Simulation:

The PDB structure 4YCM was prepared with the Protein Preparation Wizard program assuming a pH 7 and used as the starting structure for docking analysis.

Acknowledgement:

This work was supported by JSPS KAKENHI Grant No. 16H03285. We thank Sanyo Fine Co., Ltd. for their gift of chiral trityl glycidyl ether.

Supporting Information:

The Supporting Information is available free of charge on the ACS Publication website at DOI: XXXXX.

¹H and ¹³C NMR spectra of all new compounds.

Reference:

1) Teruya, T.; Sasaki, H.; Kitamura, K.; Nakayama, T.; Suenaga, K. Org. Lett. 2009, 11, 2421-2424.

2) Ohno, O.; Watanabe, A.; Morita, M.; Suenaga, K.; Chem. Lett. 2014, 43, 287-289.

3) Yonezawa, T.; Mase, N.; Sasaki, H.; Teruya, T.; Hasegawa, S.; Cha, B. Y.; Yagasaki, K.; Suenaga, K.; Nagai, K.; Woo, J. T. *J. Cell. Biochem.* **2012**, *113*, 440-448.

4) Morita, M.; Ogawa, H.; Ohno, O.; Yamori, T.; Suenaga, K.; Toyoshima, C. FEBS Lett. 2015, 589, 1406-1411.

5) Morita, M.; Ohno, O.; Teruya, T.; Yamori, T.; Inuzuka, T.; Suenaga, K. *Tetrahedron* **2012**, *68*, 5984-5990.

6) (a) Sawant, P.; Maier, M. E. Synlett 2011, 2011, 3002-3004. (b) Chandrasekhar, S.; Rajesh, G.; Naresh, T. *Tetrahedron Lett.* 2013, *54*, 252-255. (c) Tanabe, Y.; Sato, E.; Nakajima, N.; Ohkubo, A.; Ohno, O.; Suenaga, K. *Org. Lett.* 2014, *16*, 2858-2861.

7) (a) Das, S.; Paul, D.; Goswami, R. K. *Org. Lett.* **2016**, *18*, 1908-1911. (b) Sato, E.; Tanabe, Y.; Nakajima, N.; Ohkubo, A.; Suenaga, K. *Org. Lett.* **2016**, *18*, 2047-2049.

8) (a) Schmidt, R. R.; Michel, J. Angew. Chem. Int. Ed. 1980, 19, 731-732. (b) Grundler, G.; Schmidt, R. R. Carbohydr. Res. 1985, 135, 203-218. (c) Schmidt, R. R. Angew. Chem. Int. Ed. 1986, 25, 212-235. (d) Zhu, X.; Schmidt, R. R. Angew. Chem. Int. Ed. 2009, 48, 1990-1934.

9) (a) Yu, B.; Tao, H. *Tetrahedron Lett.* **2001**, *42*, 2405-2407. (b) Yu, B.; Sun, J. *Chem. Commun.* **2010**, *46*, 4668-4679.

10) (a) Kashyap, S.; Hotha, S. Tetrahedron Lett. 2006, 47, 2021-2023. (b) Li, Y.; Yang, X.; Liu, Y.; Zhu,

C.; Yang, Y.; Yu, B. *Chem. Eur. J.* **2010**, *16*, 1871-1882. (c) Tang, Y.; Li, J.; Zhu, Y.; Li, Y.; Yu, B. *J. Am. Chem. Soc.* **2013**, *135*, 18396-18405.

11) (a) Ferrier, R.; Hay, R.; Vethaviyasar, N. *Carbohydr. Res.* 1973, 27, 55-61. (b) Nicolaou, K. C.; Seitz, S. P.; Papahatjis, D. P. J. Am. Chem. Soc. 1983, 105, 2430-2424. (c) Veeneman, G. H.; Leeuwen, S. H.; Boom, J. H. *Tetrahedron Lett.* 1990, 31, 1331-1334.

12) (a) Mukaiyama, T.; Murai, Y.; Shoda, S. *Chem. Lett.* **1981**, *10*, 431-432. (b) Suzuki, K.; Maeta, H.; Matsumoto, T.; Tsuchihashi, G. *Tetrahedron Lett.* **1988**, *29*, 3571-3574. (c) Suzuki, K.; Maeta, H.; Matsumoto, T. *Tetrahedron Lett.* **1989**, *30*, 4853-4856.

13) (a) Toshima, K.; Misawa, M.; Ohta, K.; Tatsuta, K.; Kinoshita, M. *Tetrahedron Lett.* 1989, *30*, 6417-6420. (b) Blanchard, N.; Roush, W. R. *Org. Lett.* 2003, *5*, 81-84. (c) Kusumi, S.; Tomono, S.; Okuzawa, S.; Kaneko, E.; Ueda, T.; Sasaki, K.; Takahashi, D.; Toshima, K. *J. Am. Chem. Soc.* 2013, *135*, 15909-15912.

14) (a) Coppi, A.; Ricci, A.; Taddei, M. J. Org. Chem. 1988, 53, 911-913. (b) Heapy, A. M.; Wagner, T. W.; Brimble, M. M. Synlett 2007, 2007, 2359-2362. (c) Schomaker, J. M.; Pulgam, V. R.; Borhan, B. J. Am. Chem. Soc. 2004, 126, 13600-13601. (d) Roberts, S. W.; Rainier, J. D. Org. Lett. 2007, 9, 2227-2230.

15) (a) Helferich, V. B.; Lang, O. *J. Prakt. Chem.* **1932**, *132*, 321-334. (b) Aspinall, G. O.; Crane, A. M.; Gammon, D. W.; Ibrahim, I. H.; Khare, N. K. *Carbohydr. Res.* **1991**, *216*, 337-355.

16) (a) Dess, D. B.; Martin, J. C. J. Org. Chem. **1983**, 48, 4155-4156. (b) Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. **1991**, 113, 7277-7287.

17) (a) Shiina, I.; Kubota, M.; Ibuka, R.; *Tetrahedron Lett.* **2002**, *43*, 7535-7539. (b) Shiina, I.; Hashizume, M.; Yamai, Y.; Oshiumi, H.; Shimazaki, T.; Takasuna, Y.; Ibuka, R. *Chem. Eur. J.* **2005**, *11*, 6601-6608.

18) (a) Mitsunobu, O. *Synthesis* **1981**, *1981*, 1-28. (b) Swamy, K. C. K.; Kumar, N. N. B.; Balaraman, E.; Kumar, K. V. P. P. *Chem. Rev.* **2009**, *109*, 2551-2651.

19) (a) Brodmann, T.; Janessen, D.; Kalesse, M. J. Am. Chem. Soc. **2010**, 132, 13610-13611. For review, see: (b) Stille, J. K. Angew. Chem. Int. Ed. **1986**, 25, 508-524. (c) Nicolaou, K. C.; Bulger, P. G.; Sarlah, D. Angew. Chem. Int. Ed. **2005**, 44, 4442-4489. (d) Pattenden, G.; Sinclair, D. J. J. Organomet. Chem. **2002**, 653, 261-268.

20) Glide, version 6.4, Schrödinger, LLC, New York, NY, 2014.