# SUCCESSIVE METHYL MIGRATIONS OCCURRING IN THE ACID TREATMENT OF 1-EPOXYETHYL-1,2,2-TRIMETHYL-CYCLOPENTANE DERTVATIVE 

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Summary: Acid treatment of (1R,3S,1'S)-1-(1',2'-Epoxyethyl)-1,2,2-trimethyl-3-acetoxymethyl-
 1-oxa-bicyclo[3,3.0]octane derivatives formed via successive methyl migrations followed by an oxorane ring closure.

During the course of our synthetic study on naturally occurring bioactive compounds starting from $d$-camphor (1), ${ }^{1)}$ we have found that $\mathrm{BF}_{3}$-etherate treatment of an epoxide (6) furnishes three products ( $8 \sim \sim \sim \sim \sim 10$ ) among which $\underset{\sim}{8}$ and $\underset{\sim}{9}$ are l-oxa-bicyclo[3.3.0]octane derivatives formed through successive migrations of methy1 residues (cf. i) and an oxorane ring formation. This paper provides supporting evidence for this interesting finding.

The epoxide ( $\underset{\sim}{6}$ ) was synthesized from $d$-camphoric acid (2) ${ }^{2}$ ) which was prepared by nitric acid oxidation of $\underset{\sim}{1}$. Methylation followed by $\mathrm{LiAlH}_{4}$ reduction of $\underset{\sim}{2}$ gave a diol (3), ${ }^{3 \text { ) }}$ which, on treatment with $\mathrm{Ac}_{2} \mathrm{O}-\mathrm{AcONa}$, furnished a monoacetate (4), oil, $\mathrm{C}_{12} \mathrm{H}_{22}{ }^{\mathrm{O}} 3^{4},{ }^{4}$ ) $[\alpha]_{\mathrm{D}}+39^{\circ}\left(\mathrm{CHCl}_{3}\right.$ ), $\operatorname{IR}(f i 1 m) \mathrm{cm}^{-1}: 3440,1737,{ }^{1}{ }_{\mathrm{H}-\mathrm{NMR}}\left(\mathrm{CCl}_{4}\right) \delta: 0.81,0.95,1.00,2.00(\mathrm{a} 113 \mathrm{H}, \mathrm{s}), 3.37,3.50(2 \mathrm{H}$, $\mathrm{ABq}, \mathrm{J}=9,-\mathrm{C}-\mathrm{CH}_{2} \mathrm{OH}$ ), $3.8-4.1\left(2 \mathrm{H}, \mathrm{AB}\right.$ in $\mathrm{ABX},-\mathrm{CH}-\mathrm{CH}_{2} \mathrm{OAC}$ ). Pyridinium chlorochromate (PCC) oxidation ${ }^{5}$ ) of 4 gave an unstable aldehyde which was immeidately subjected to methylenation to furnish a vinyl-acetate (5), oil, $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{2}$, $\operatorname{IR}$ (film): $1747,1645,910, \delta\left(\mathrm{CCl}_{4}\right): 3.98(2 \mathrm{H}, \mathrm{d}$, $\mathrm{J}=6$ ) , 4.8-5.9 (3H, ABC, vinyl). m-Chloroperbenzoic acid oxidation of 5 gave two epoxides, 6 (70\%), oil, $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{3},[\alpha]_{\mathrm{D}}+44^{\circ}\left(\mathrm{CHCl}_{3}\right)$, $\operatorname{IR}$ (film): 1744, 870, 814, $\delta\left(\mathrm{CCl}_{4}\right): 0.87,0.90,1.00$,
 oil, $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{3},[\alpha]_{\mathrm{D}}+43^{\circ}\left(\mathrm{CHCl}_{3}\right)$, IR (film): $1741,850,805, \delta\left(\mathrm{CCl}_{4}\right): 0.86,0.92,1.02,1.96$ (all $3 \mathrm{H}, \mathrm{s}$ ), $2.4-2.9(3 \mathrm{H}, \mathrm{ABC}), 3.8-4.3(2 \mathrm{H}, \mathrm{AB}$ in ABX$)$.

The $C-1$ ' configurations of both epoxides (S for $\underset{\sim}{6,} \mathrm{R}$ for 7 ) were elucidated by application of the Horeau's method ${ }^{6}$ ) for respective diol-monoacetates ( $6 \mathrm{~b}, 7 \mathrm{~b}$ ), which were prepared from epoxides ( $6, \underset{\sim}{7}$ ) by $\operatorname{LiAlH}_{4}$ reduction (giving $\underset{\sim}{6 a}, \underset{\sim}{7 a}$ ) followed by partial acetylation: $\underset{\sim}{6 b}, ~ \delta$ $\left(\mathrm{CDCl}_{3}\right): 1.14(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=6), 3.76(1 \mathrm{H}, \mathrm{q}, \mathrm{J}=6)\left(-\mathrm{CH}(\mathrm{OH})-\mathrm{CH}_{3}\right), 3.8-4.3\left(2 \mathrm{H}, \mathrm{AB}\right.$ in $\mathrm{ABX},-\mathrm{CH}-\mathrm{CH}_{2}-$ $\mathrm{OAc}) ; \underset{\sim}{7 \mathrm{~b}}, \delta\left(\mathrm{CDCI}_{3}\right): 1.08(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=6), 3.7-4.2(3 \mathrm{H}, \mathrm{m})$; recovered $\alpha$-phenylbutyric acid: $[\alpha] \mathrm{D}$ $+4.8^{\circ}$ ( $c=1.59$, benzene) in preparation of $\underset{\sim}{c}$ and $-0.9^{\circ}(c=2.19)$ for 7 c .

Treatment of $\underset{\sim}{6}$ in benzene with $\mathrm{BF}_{3}$-etherate at room temp. ( $20^{\circ}$ ) for 50 min . furnished three products: $\underset{\sim}{8}(38 \%), \underset{\sim}{9}(13 \%)$, and $\underset{\sim}{10}(42 \%): \underset{\sim}{8}$, oil, $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{3},\left[{ }^{[\alpha]}\right]_{\mathrm{D}}+12^{\circ}\left(\mathrm{CHCl}_{3}\right)$, IR (film) $: 1738, \delta\left(\mathrm{CCl}_{4}\right): 0.90(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=7), 0.96,1.09,1.93(\mathrm{all} 3 \mathrm{H}, \mathrm{s}), 3.15(1 \mathrm{H}, \mathrm{d} . \mathrm{d}, \mathrm{J}=8,12,2-$
 $+37^{\circ}\left(\mathrm{CHCl}_{3}\right), \operatorname{IR}(f i l \mathrm{~m}): 1738, \delta\left(\mathrm{CCl}_{4}\right): 0.88(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=7), 0.87,1.02,1.96(\mathrm{all} 3 \mathrm{H}, \mathrm{s}), 3.27$ $\left(1 \mathrm{H}, \mathrm{d} . \mathrm{d}, \mathrm{J}=8,9,2-\mathrm{H}_{A}\right), 3.83$ (1H, d.d, $\mathrm{J}=8,6,2-\mathrm{H}_{\mathrm{B}}$ ), $3.6-4.2$ ( $2 \mathrm{H}, \mathrm{AB}$ in $A B X$ ).

Since $\underset{\sim}{8}$ and $\underset{\sim}{9}$ were respectively shown by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ to possess one sec. and two tert. methyls and two methylene protons attached to a carbon adjacent to an ether oxygen, they were presumed to be products formed from $\underset{\sim}{6}$ via successive methyl migrations as shown in $\underset{\sim}{i}$ (via a, c giving $\underset{\sim}{8}$, while via b, c giving 9). The C-3R configurations of both $\underset{\sim}{8}$ and $\underset{\sim}{9}$ were presumed on the basis of CD analysis ${ }^{7}$ ) of their lactone derivatives, 11 and 12 , prepared by $\mathrm{RuO}_{4}$ oxidation $^{8)}$ of $\underset{\sim}{8}$ and
 $1776,1747, \mathrm{CD}(\mathrm{MeOH}):[\theta]_{217}^{20}-6800$ (neg. max.) ; 12, oil, $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{4},[\alpha]_{\mathrm{D}}+27^{\circ}\left(\mathrm{CHCl}_{3}\right)$, IR ( $\mathrm{CC1} 1_{4}$ ) : 1771. 1748, CD (dioxane): $\left\lceil 01_{223}^{20}-2500\right.$ (neg. max.). Therefore, $8 \underset{\sim}{8}$ and 9 were presumed to be isomeric at their angular configurations. Finally X-ray analysis of the lactone 11 was carried out to obtain a direct proof of the structure.

CRYSTAL DATA: $\mathrm{C}_{13} \mathrm{H}_{20} 0_{4}$, M.W. $=240.30$, monoclinic space group $\mathrm{P}_{1}, \mathrm{a}=8.827(1), \mathrm{b}=7.467$ (1), $c=10.713(1) \AA, B=112.52(1)^{\circ}, \mathrm{z}=2, \mathrm{U}=652.2 \AA^{0}{ }^{3}, \mathrm{D}_{\mathrm{x}}=1.21 \mathrm{~g} \cdot \mathrm{~cm}^{-3}, \mathrm{D}_{\mathrm{o}}=1.20 \mathrm{~g} \cdot \mathrm{~cm}^{-3} . \mathrm{A}$ total of 1043 non-zero independent reflections with $2 \Theta \leq 120^{\circ}$ were measured on an automated diffractometer using 20-w scan technique and Ni-filtered Cu Ko radiation. The structure was solved by direct method using "MULTAN" program. ${ }^{\text {9 }}$ ) Block-diagonal least-squares refinement with anisotropic nonhydrogen atoms and isotropic hydrogens reduced $R$ to $0.071 .{ }^{10 \text { ) The figure is a }}$ computer generated ORTEP stereoview of the molecule ( $30 \%$ ellipsoids).

The third product 10 , oil, $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{3},[\alpha]_{\mathrm{D}}+9^{\circ}\left(\mathrm{CHCl}_{3}\right)$, IR (film), 3400, 1732, $\delta\left(\mathrm{CCl}_{4}\right)$ : $0.88,1.11,2.00(\mathrm{all} 3 \mathrm{H}, \mathrm{s}), 1.07(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=7), 3.1-3.6(2 \mathrm{H}, \mathrm{AB}$ in ABX$), 4.07(2 \mathrm{H}, \mathrm{m}), 5.29$






Figure
( 1 H , narrow $\mathrm{m}, \mathrm{W}_{\mathrm{h} / 2}=5,=\mathrm{CH}=\mathrm{C}=$ ), was shown to possess one sec. and two tert. methyls and one olefinic proton and was presumed to be derived from $\underset{\sim}{i} v i a$ deprotonation (route d). The structure 10 was further substantiated by a fact that $\mathrm{BF}_{3}$-etherate treatment of 10 in benzene at room temp. for 24 hr gave $\underset{\sim}{8}(51 \%)$ and $\underset{\sim}{9}$ (9\%) with recovery of $\underset{\sim}{10}$ (33\%).

The successive methyl migrations presented here are reminiscent of a biogenetic pathway from cuparane-type (ii) to trichothecane-type (iii) sesquiterpenes. ${ }^{11 \text { ) The behavior of minor }}$ epoxide (7) against acid is under investigation.

## References and Footnotes

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(Received in Japan 16 February 1980)
