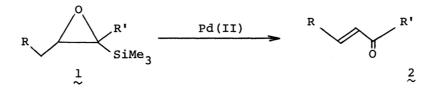
PALLADIUM-PROMOTED TRANSFORMATION OF α , β -EPOXYSILANES TO α , β -UNSATURATED CARBONYL COMPOUNDS

Toshikazu HIRAO,* Naoto YAMADA, Yoshiki OHSHIRO, and Toshio AGAWA Department of Petroleum Chemistry, Faculty of Engineering Osaka University, Yamada-oka, Suita, Osaka 565

Treatment of α , β -epoxysilanes with palladium(II) salt in DMF affords α , β -unsaturated carbonyl compounds, in which the carbonyl group appears on the carbon originally bearing the silyl group.

 $\alpha,\beta\text{-}\textsc{Epoxysilanes}$ have been used as versatile intermediates in organic synthesis.¹⁾ Open chain α,β -epoxysilanes are hydrolyzed in acid to give aldehydes or ketones,²⁾ which indicates α , β -epoxysilane is a synthetic equivalent to aldehyde or ketone. Herein, we wish to describe a novel transformation of α , β -epoxysilanes to α , β -unsaturated carbonyl compounds with palladium(II) salt.



Typical procedure is as follows. To a solution of palladium acetate (1 mmol) in DMF (4 ml) was added (Z)-l-trimethylsilyl-l-hexene oxide³⁾ (la, 1 mmol) at room temperature. The resultant mixture was stirred at 60°C for 7 h under nitrogen. The reaction proceeded with deposition of metallic palladium. After removing the deposit, aqueous saturated sodium chloride solution was added to the filtrate, which was extracted with ether. The combined extract was washed with aqueous saturated sodium chloride solution and concentrated. The residue was chromatographed on a silica gel column to give (E)-2-hexenal in 76% yield. Formation of the Z isomer was not observed in the NMR spectrum of the crude product.

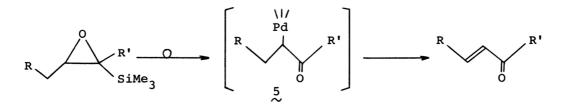
The reaction of la at room temperature also produced (E)-2-hexenal. The

transformation was well achieved only in DMF, and the reaction in acetonitrile gave a poor result. Use of $PdCl_2(PhCN)_2$ instead of $Pd(OAc)_2$ gave a moderate yield of 2a, being contaminated with hexanal (3a) and 1-chloro-1-trimethylsilyl-2-hexanol (4a). The cationic palladium complex, $[Pd(acac)(1,5-COD)]BF_4$, also promoted the present reaction. The results are shown in Table 1.

Similarly, (Z)-1-trimethylsilyl-1-hexene oxide was converted to (E)-2-hexenal in 98% yield. In the case of internal α,β -epoxysilanes, the yield of 2 was decreased. The reason might be explained by the difficulty of their hydrolysis.^{2c,4)} It is of interest that only (E)- α,β -unsaturated carbonyl compounds were produced selectively regardless of the stereochemistry of α,β -epoxysilanes $la\sim ld$.

A catalytic process of the present transformation was attempted by adding CuCl₂ into the mixture of la and a catalytic amount of Pd(OAc)₂ in DMF, but the product obtained was the α -chloro- β -hydroxysilane 4a (62%). The reaction only in the presence of CuCl₂ also gave 4a (80%), which corresponds to the product in magnesium bromide assisted rearrangement of α , β -epoxysilanes.⁵⁾ Reoxidation of the reduced Pd(0) was performed by using Ag₂O as an oxidant. The reaction of la was carried out in the presence of Pd(OAc)₂ (0.1 equiv.) and Ag₂O (1 equiv.) at 60°C for 20 h to produce 2a in 31% yield.

The present reaction involves a carbon-silicon bond cleavage. Although the reaction path has not been clarified, the bond cleavage might be induced by palladium(II) salt. The epoxy ring undergoes rearrangement to form the palladium complex 5^{6} , which is subjected to β -elimination of Pd-H species to give the α,β -unsaturated carbonyl compound 2.



 α,β -Epoxysilanes are easily prepared by oxidation of vinylsilanes or homologation⁷⁾ of ketones or aldehydes with α -chloro- α -trimethylsilyl carbanion. Combination of this homologation reaction and the present palladium(II)-promoted transformation provides a versatile process for the preparation of α,β -unsaturated aldehydes.

1	D-11-3/		Reaction			
	Palladium Salt	Solv.	Temp.	Time	Products ^a	(8)
	Pd(OAc) ₂	DMF	60°C	7 h	^∕∕∿₀	90 (76) ^b
SiMe ₃					2a	
$\stackrel{\texttt{la}}{\sim}$	Pd(OAc) ₂	DMF	rt	3 days	2a ~	71
	Pd (OAc) 2	сн _з си	rt	3 days	2a	10
	Pd (OAc) 2	DMSO	rt	4 days		2
	PdCl ₂ (PhCN) ₂	DMF	rt	2.5 h	2a ~	40
					$\sim\sim\sim_0$	27
					^{3a} OH SiM€	21 3
	PdCl ₂ (PhCN) ₂	CH ₃ CN	80°C	3.5 h	4a 2a ∼	trace
	2	3			∼ 4a	38
	$Pd(NO_3)_2$	DMF	rt	3 h		58
	2 4 (11 - 3 / 2	212	10	5 	2a ~	7
	[Pd(acac)(1,5-COD)]E	את האד	rt	3 h	3a ∼	, 47
<u>_</u> 0		-1			2a ~	
SiMe ₃	Pd(OAc) ₂	DMF	60°C	7 h	∕∕∕∕0 2b	80
	$\frac{Pd(OAc)}{2}$	DMF	60°C	l h	~ 2a ~	98
	Pd (OAc)	DMF	100°C	38 h	\sim	20
					\sim^{0}	20
	DdCl (DhCNI)	DMF	60°C	7 h	3d ~	32
	$PdCl_2(PhCN)_2$	Drif		/ 11	2d ∼ 3d	32
€ ^{SiMe} 3	le Pd(OAc) ₂	DMF	100°C	15 h	O 2e	20

Table 1. Reaction of α , β -Epoxysilanes with Palladium Salt

a. Yields were determined by glc. b. Isolated yield. c. Formation of 2a was not observed.

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References

- P. F. Hudrlik, D. Peterson, and R. J. Rona, J. Org. Chem., <u>40</u>, 2263 (1975);
 P. F. Hudrlik, A. M. Hudrlik, R. J. Rona, R. N. Misra, and G. P. Withers, J. Am. Chem. Soc., <u>99</u>, 1993 (1977); M. C. Croudace and N. E. Schore, J. Org. Chem., <u>46</u>, 5357 (1981).
- a) G. Stork and E. Colvin, J. Am. Chem. Soc., <u>93</u>, 2080 (1971); b) R. K. Boeckman and K. J. Bruza, Tetrahedron Lett., <u>1974</u>, 3365; c) P. F. Hudrlik, J. P. Arcoleo, R. H. Schwartz, R. N. Misra, and R. J. Rona, ibid., <u>1977</u>, 591.
- 3) (Z)- α , β -Epoxysilanes were prepared by the Cp₂TiCl₂-catalyzed Grignard exchange reaction with trimethylsilylacetylenic compounds, followed by oxidation with MCPBA; F. Sato, H. Ishikawa, and M. Sato, Tetrahedron Lett., <u>22</u>, 85 (1981).
- 4) C. M. Robbins and G. H. Whitham, J. Chem. Soc., Chem. Commun., <u>1976</u>, 697.
- 5) P. F. Hudrlik, R. N. Misra, G. P. Withers, A. M. Hudrlik, R. J. Rona, and J. P. Arcoleo, Tetrahedron Lett., 1976, 1453.
- 6) Y. Ito, T. Hirao, and T. Saegusa, J. Org. Chem., <u>43</u>, 1011 (1978).
- 7) C. Burford, F. Cooke, E. Ehlinger, and P. Magnus, J. Am. Chem. Soc., <u>99</u>, 4536 (1977).

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