

PII: S0960-894X(97)10067-1

DESIGN AND SYNTHESIS OF NOVEL 2.7-DIALKYL SUBSTITUTED 5(S)-AMINO-4(S)-HYDROXY-8-PHENYL-OCTANECARBOXAMIDES AS IN VITRO POTENT PEPTIDOMIMETIC INHIBITORS OF HUMAN RENIN

Richard Göschke*, Y. Nissim Claude Cohen⁵, Jeanette M. Wood, Jürgen Maibaum** Novartis Pharma AG, Metabolic and Cardiovascular Diseases, CH-4002 Basel, Switzerland

Abstract: Novel low-molecular weight transition-state peptidomimetic renin inhibitors characterized by an allcarbon 8-phenyl substituted octanecarboxamide skeleton have been discovered based on a tonographical design approach. The in vitro most potent inhibitors 21, 25 and 26 incorporating a strong H-bond acceptor group linked to the benzyl spacer of the (P_3-P_1) -unit had IC_{50} s in the low nanomolar range against human renin. © 1997 Elsevier Science Ltd.

Renin inhibitors which specifically block the first, rate limiting step of the blood-pressure regulating renin-angiotensin system (RAS) are an attractive target in drug discovery as novel agents for the treatment of hypertension and other cardiovascular diseases.^{1a,b} However, development of substrate-based transition-state renin inhibitors usually spanning the P_4 - P_1 minimal sequence has been hampered in most cases by limited oral bioavailability, rapid metabolism or extensive first-pass biliary excretion.²

Problems associated with unfavourable pharmacokinetic properties could be potentially overcome by generating structurally distinct renin inhibitors. Recently, a novel approach based on the extension of the P_1 cyclohexyl group of peptide-based inhibitors directly towards P3 and partial truncation of the N-terminal backbone has been reported^{3,4} as a strategy in order to reduce molecular size. The crystal structure of a (P_a-P_1) spanning inhibitor complexed to rh-renin confirmed that its (P3-P1)-modified side chain could indeed be accomodated by the S₃-S₁ site.⁴ Furthermore, elimination of the P₄-P₃ portion led only to a small drop in binding affinity,⁴ with IC₅₀s of these inhibitors in the micromolar range. On the other hand, the significance of the H-bond accepting and -donating P_4 - P_3 amide group for strong binding interactions has been emphasized.^{3b}



This has prompted us to disclose, in a preliminary communication,⁵ part of our efforts towards a similar topographical⁶ design concept based on the enzyme-bound conformation of the peptide-based renin inhibitor CGP 38560 (1),⁷ as predicted by modeling⁸ and confirmed by X-ray crystallography.⁹ Initially, our strategy was guided by the paradigm that the key binding forces between the enzyme and a ligand result from interactions involving the amino acid side chains rather than the amide backbone of the peptide-like inhibitors.^{10,11} Several approaches were extensively investigated to design various hydrophobic conformationally rigid (P₃-P₁)moieties appending a transition state mimetic, that would optimally fill the large, contiguous S_3/S_1 binding ** Fax: ++4161/69 65966; E-Mail: Juergen Klaus.Maibaum@pharma.novartis.com

pocket of the enzyme, and which would allow to sacrifice the P_4 - P_2 spanning backbone and side chains of a topological inhibitor such as 1.⁵ As a first result, the hydroxyethylene mimetic 3 bearing a free NH₂ group was found to inhibit human renin in the sub-micromolar range, thus being 100 times more potent than compound 2.¹² We report herein the synthesis and *in vitro* potency of a novel class of peptidomimetics incorporating a benzyl spacer that directly links the P_3 residue to an alkyl P_1 group of the dipeptide isostere. Attachment of H-bond acceptor groups to the phenyl template led to inhibitors with low nanomolar binding affinities.

Design Concept

Intrigued by the close spatial proximity of both the P₁ cyclohexyl and P₃ phenyl sidechains of 1 in its predicted binding mode within the S₃/S₁ site,⁵ we envisaged the possibility to extend P₁ by annulating a phenyl ring, for tethering P₃, to the C3'-C4' bond of the cyclohexyl, which was predicted to be distal to the surface of the S₁ site, as illustrated in Figure 1 (Structure **A**). Modeling of such a tetrahydronaphthalene substituted dipeptide isostere, as well as its 'ring-opened' congener (Figure 1, structure **B** with R¹=H, R²=Et) suggested that in both cases the rigid phenyl spacer would direct a hydrophobic substituent R³, such as aryl or bulky alkyl, towards the S₃ pocket, and that the P₁ alkyl group would be well accomodated by the S₁ site. Furthermore, these initial design considerations predicted the β-configuration at P₁ (**B**, R¹=H, R²=alkyl) to be preferred over the α-configuration (**B**, R¹=alkyl, R²=H) with respect to a better fit in the S₃/S₁ pocket.¹³

Figure 1. Design Approach towards Novel Peptidomimetic Hydroxyethylene Mimetics extended at P1



Synthesis

The novel (P₃-P₁)-extended transition-state mimetic renin inhibitors **10-26** with varying P₁ side chains and tethered P₃ residues (Table) were prepared by the linear route as exemplified for compounds **20** and **21** in the Scheme.^{14,15} Commercially available **4** (Aldrich Ltd.) was protected as its acetate and subsequently treated with NBS to give the corresponding benzylbromide which was used for the diastereoselective alkylation (de \geq 95%) of the N-acyl-oxazolidinone chiral auxiliaries (*R*)-**27b** and (*R*)-**27c** according to the method of Evans *et al.*¹⁶ Reductive cleavage of the chiral auxiliary with LiAlH₄ afforded alcohol (*S*)-**5b** in good (70%), but (*R*)-**5c** in only poor (12%) yields, the major side products resulting from reductive endocyclic cleavage of the oxazolidinone. Similarly, the regioselectivity of the LiAlH₄ reduction dropped dramatically with increasing steric hindrance^{16b} in the case of the *para* mono-substituted analogues **8a-d**¹⁷ and **9a** (for example, 94% yield for **8a** vs. 43% for **8c** vs. 0% for **8e**; *cf*. Scheme). On the other hand, removal of the auxiliary through reaction with lithium benzyl mercaptide and LiAlH₄ reduction of the intermediate thioester in one pot by the method of Damon and Coppola¹⁸ gave enantiomerically pure **5c** in excellent **86%** yield and even moderate to good yields for **8e** (37%). In the case of the phenyl-substituted alcohol **8f**, the presence of the nucleophilic and poorly basic benzyl mercaptide caused complete racemisation at the benzylic stereocenter.¹⁹





Reagents: (a) Ac₂O (93%); (b) NBS, CCl₄ (95%); (c) 3-butyroyl-4(*R*)-benzyl-oxazolidin-2-one (**27b**) or 3-isovaleroyl-4(*R*)-benzyl-oxazolidin-2-one (**27c**), LiHMDS, THF, -78 °C to 0 °C (46-52%); (d) BnSH, *n*-BuLi, LiAlH₄, THF (**5c**:86%); (e) **5c**, BnBr, Cs₂CO₃, DMF (78%); (f) NBS, PPh₃, CH₂Cl₂ (98%); (g) (2*R*)-2,5-dihydro-3,6-dimethoxy-2-isopropylpyrazine, *n*-BuLi, THF, -75 to -20 °C (91%); (h) 1N HCl, MeCN (96%); (i) (BOC)₂O, NEt(*i*Pr)₂, CH₂Cl₂ (96%); (j) DIBAH, toluene (quant.); (k) i. N-butylmethacryl-amide, *n*-BuLi, TiCl(O*i*Pr)₃, THF, -78 °C; ii. RCHO, THF, -78 °C; iii. silica gel chromatography (19%); (l) H₂, Pd/C, MeOH (88%); (m) allylbromide, Cs₂CO₃, acetone (90%) or BrCH₂CO₂Me, Cs₂CO₃, acetone, Nal (80%); (n) 4N HCl-dioxane (**20**:32%; **21**:74%).

The N-BOC-protected α -amino ester 2(S),4(S)-6c was expediciously prepared from 5c (after reprotection of the phenolic OH group) in 62% overall yield following the procedure of Schöllkopf *et al.*²⁰ Subsequent DIBAH reduction afforded the N-BOC amino aldehyde, which was reacted with the dianion generated from *n*butyl acrylamide in the presence of Ti(O*i*Pr)₃Cl similar to the method by Kempf,^{21,22} to give separable mixtures of both 4(S)/4(R)-configured diastereoisomers 7c in good yields (ratio 4(S),5(S),7(R)-7c:4(R),5(S),7(R)-7c ca. 1:2).²³ Hydrogenation (10% Pd/C) of the alkene with concomitant removal of the benzyl protecting group yielded a pair of 2(R)/2(S)-diastereomers which were difficult to separate by silica gel chromatography and thus further transformed as epimeric mixtures.²⁴ O-Alkylation of the phenol intermediates with the corresponding alkylhalogenides (iodoacetamide for 25) was followed by N-deprotection to afford inhibitors 20, 21 and 25,²⁵ whereas the carboxylic acid 24 (Table) was obtained by hydrogenolysis of the benzyl ester precursor. The methylsulfone 26 was prepared by oxidation (oxone, MeOH-H₂O, 16h, r.t, 54%) of the corresponding thioether obtained by alkylation of the phenol precursor with methylthiomethyl chloride (66%).

Results and Discussion

The hydroxyethylene mimetic 10 bearing a hydrophobic biphenyl residue at P_1 assumed to be directed towards the S_3 enzyme sub-pocket according to modeling showed weak binding affinity for purified human renin at the micromolar level (Table). Thus, 10 had a 10 fold increased activity in this assay (determined at pH=7.2)⁷ as compared to the non-extended dipeptide isostere 2.¹² Replacement of the terminal phenyl with the bulky *tert*-butyl group gave inhibitor 11 with similar *in vitro* potency. This is in agreement with previous results for 3 and some analogues, that an appended *tert*-butyl is equally well tolerated at P_3 as phenyl which is the commonly more preferred P_3 residue in peptide-based inhibitors.¹² Further optimisation of these initial

target inhibitors was performed within the P_3 tert-butyl substituted series, partially in view of more readily available precursors bearing additional substituents on the phenyl ring.

$ \begin{array}{cccccccccccccccccccccccccccccccccccc$					
No. ^{a)}	R ²	R ³	R ⁴	R ⁵	Binding Affinity IC ₅₀ , μM (pH 7.2)
3					30
10	CH ₃	phenyl	Н	Н	3
11	CH ₃	<i>tert</i> -butyl	Н	Н	2
12	C ₂ H ₅	<i>tert</i> -butyl	Н	Н	0.8
13	$C_2H_5^{(b)}$	<i>tert</i> -butyl	Н	Н	3
14	CH(CH ₃) ₂	<i>tert</i> -butyl	Н	Н	0.1
15	CH ₂ CH(CH ₃) ₂	tert-butyl	Н	Н	4
16	$C(CH_3)_3$	tert-butyl	Н	Н	1.5
17	phenyld)	tert-butyl	Н	Н	39
18 ^{C)}	CH(CH ₃) ₂	tert-butyl	ОН	Н	0.13
19	C ₂ H ₅	tert-butyl	OC ₄ H ₉	Н	0.24
20 <i>c)</i>	CH(CH ₃) ₂	tert-butyl	OCH ₂ CH=CH ₂	Н	0.11
21	$CH(CH_3)_2$	tert-butyl	OCH ₂ CO ₂ CH ₃	Н	0.006
22	$CH(CH_3)_2$	tert-butyl	н	OCH2CO2C2H5	0.29
23 ^{c)}	$CH(CH_3)_2$	Н	OCH ₂ CO ₂ CH ₃	Н	0.037
24	CH(CH ₃) ₂	<i>tert</i> -butyl	OCH ₂ COOH	Н	0.120
25	CH(CH ₃) ₂	tert-butyl	OCH ₂ CONH ₂	Н	0.020
26	CH(CH ₃) ₂	tert-butyl	OCH ₂ SO ₂ CH ₃	Н	0.013

Table. IC 505 of 5(S)-Amino-4(S)-hydroxy-8-phenyl-octanecarboxamides 10-26 against Purified Human Renin

он сн. ..

a) Tested (single determination) as ca. 1:1-mixtures of C2(R,S)-diastereomers; b) (R)-configured at C7; c) Pure diastereomer of 2(R),4(S),5(S),7(S) absolute configuration; d) (R,S)-configured at C7.

In the next step, the structure-activity relationship (SAR) of the putative P_1 residue as well as the required stereochemistry at this position was investigated. Increasing the steric bulkiness by replacing Me with Et and *i*Pr improved binding affinity 20 fold (11 vs. 12 vs. 14), which was attributed to an increase in the van der Waals contacts to the hydrophobic S_1 subsite. In order to prove that the C7(S) absolute stereochemistry at P_1 is required for strong binding of 12, the corresponding (7*R*)-epimer 13 was prepared by the same route shown in the Scheme. Inhibitor 13 was found to be 3-4 times less active than 12, as was also suggested by comparative docking analyses for both epimers.¹³ Extension of the P_1 side chain by incorporation of isobutyl (compound 15) led to a 15 fold drop of the IC₅₀ value compared to 14. Also, the additional CH₃ of the *tert*-butyl group in 16 appeared to interfere unfavourably with the S₁ binding site. A phenyl group at P_1 was not well tolerated as demonstrated by the markedly decreased IC₅₀ value for 17.

To improve the moderate binding affinity of 14, we envisaged³ introducing appropriate substituents having the potential of forming additional hydrogen bonds to the enzyme cleft at different positions of the benzyl spacer. In the case of peptide-like enzyme inhibitors, the contribution of H-bonding interactions of the backbone amide groups to the overall binding energy may be outweighed by unfavourable desolvation enthalpies,²⁶ however such interactions may play an important role in providing a proper inhibitor alignment within the enzyme active site.^{11,3b} Previous reports on the SAR of peptide-based inhibitors of renin² and crystal structures of enzyme-inhibitor complexes,^{27,28} indicated the importance of the conserved hydrogen bonding between a P_2/P_3 carbonyl group of various inhibitors to the backbone amide NH of the amino acid

corresponding to Ser219 of human renin for strong inhibitor binding. The overlap of the enzyme-docked inhibitors 14 and CGP38560 (1) within the human renin model revealed the phenyl of 14 to be in a remote position several bond lengths distant from the Ser219 main-chain amide bond of the enzyme. Accordingly, and due to its large distance to other H-bonding groups of the enzyme, the additional phenolic OH of inhibitor 18 does not lead to an increase in binding affinity as compared to 14. However, we realized that attaching an ester group via a two-atom spacer preferentially *ortho* to the P₃ residue would position the ester carbonyl in reasonable hydrogen bonding geometry to the NH of Ser219, similar to the P₃ carbonyl of 1 (Figure 2).²⁹ Inhibitor 21 with a low nanomolar IC₅₀ value revealed a 15 fold enhancement in binding affinity over 14 (Table), which would be in agreement with the formation of an additional hydrogen bonding interaction. On the other hand, the *meta* substituted regioisomer 22 was much weaker in binding, indicating the requirement of the proper spatial orientation of the ester group. The markedly reduced affinities of the alkoxy derivatives 19 and 20, being only equipotent to their unsubstituted congeners 12 and 14, appeared to further support the validity of our design model. The importance of the van der Waals forces of the P₃ tert-butyl group within the hydrophobic S₃ subsite for strong binding is demonstrated by the 6 fold drop of the IC₅₀ of 23 (Table).

Figure 2. Stereoview of the overlay of the energy-minimized Monte Carlo conformations of CGP38560 (1) and inhibitor 21 within the human renin active site model.^{8,11} Both centers of the enzyme S₃ and S₁ contiguous binding pockets are indicated by red colored spots. The close distances of the P₂/P₃ carbonyl of 1, as well as of the ester carbonyl of 21, to the Ser219 (orange colored) amide NH suggested in both cases a strong H-bond.



Replacement of the terminal carboxylic ester with a carboxamide or methylsulfonyl residue as strong Hbonding acceptors led to inhibitors 25 and 26 with comparable *in vitro* potencies, whereas the carboxylic acid 24 was only a weak inhibitor of purified human renin. Finally, N-acetylation of the amino group of several inhibitors in the Table resulted in a loss in binding affinity by 2 to 3 orders of magnitude (data not shown).

In summary, a novel class of small-sized, *in vitro* highly potent peptidomimetic transition-state renin inhibitors which incorporate a constrained benzyl-spacer-linked (P_3-P_1) -moiety has been designed using a topographical approach. Introduction of additional hydrogen bond acceptor/donor residues at the position *ortho* to the P₃ substituent and in appropriate distance to the phenyl template, as suggested by previous SAR data of peptide-based inhibitors and computational modeling, resulted in a remarkable enhancement in binding affinities as compared to the parent inhibitor 14. The most potent compounds within this series, the carboxylic ester 21, the carboxamide 25 and the methylsulfonyl derivative 26, showed IC₅₀ values at the low nanomolar level towards purified human renin. Further optimization of the *in vitro* potency of this compound class and evaluation of the oral activities on blood pressure in Na⁺-depleted marmosets will be reported in due course.

Acknowledgements: The main author would like to thank Dr. Walter Fuhrer and Dr. Vittorio Rasetti for the critical discussions in the course of this work. We are grateful to H.-P. Baum for performing the *in vitro* assay and Dr. Vincenzo Tschinke for providing the stereoview presented in Figure 2. R. G. Is indebted to Ms. Nicole Hasler and Ms. Florence Lugrin for their excellent technical assistance.

2740

References and Notes

^Y Present address of main author: Dr. Richard Göschke, Felixhägli 21, CH-4103 Bottmingen, Switzerland. ^E Present address: Synergix Drug Design Ltd., Technology Park Malha, Jerusalem 91487, Israel.

1. a) Kleinert, H.D. Cardiovasc. Drugs Ther. 1995, 9, 645; b) Wood, J.M.; Cumin, F.; Maibaum, J. Pharmacol. Ther. 1994, 61, 325.

2. For a recent review, see Greenlee, W. Med. Chem. Rev. 1990, 10, 173.

3. (a) Plummer, M.; Hamby, J.M.; Hingorani, G.; Batley, B.L.; Rapundalo, S.T. *Bioorg. Med. Chem. Lett.* **1994**, *3*, 2119. (b) Plummer, M.S.; Shahripour, A.; Kaltenbronn, J.S.; Lunney, E.A.; Steinbaugh, B.A.; Hamby, J.M.; Hamilton, H.W.; Sawyer, T.K.; Humblet, C.; Doherty, A.M.; Taylor, M.D.; Hingorani, G.; Batley, B.L.; Rapundalo, S.T. *J. Med. Chem.* **1995**, *38*, 2893.

4. Lefker, B.A.; Hada, W.A.; Wright, A.S.; Martin, W.H.; Stock, I.A.; Schulte, G.K.; Pandit, J.; Danley, D.E.; Ammirati, M.J.; Sneddon, S.F. Bioorg. Med. Chem. Lett. 1995, 5, 2623.

5. (a) Maibaum, J.; Rasetti, V.; Rüeger, H.; Cohen, N.C.; Göschke, R.; Mah, R.; Rahuel, J.; Grütter, M.G.; Cumin, F.; Wood, J.M. In: Medicinal Chemistry: Today and Tomorrow. Proceedings of the AFMC International Medicinal Chemistry Symposium, Tokyo, 3-8 September 1995; Mikio Yamazaki (Ed.), Blackwell Science UK, 1997, p.155-162. (b) Rahuel, J.; Grütter, M.G.; Cohen, N.C.; Maibaum, J.; Rasetti, V.; Rüeger, H.; Göschke, R.; Mah, R.; Cumin, F.; Wood, J. Poster Abstract, XVIIth Congress of the International Union of Crystallography, Seattle (USA); August 8-17th, 1996.

6. Farmer, P.S., Ariëns, E.J. Trends in Pharmaceutical Sciences 1982, 3, 362.

7. Bühlmayer, P.; Caselli, A.; Fuhrer, W.; Göschke, R.; Rasetti, V.; Rüeger, H.; Stanton, J. L.; Criscione, L.; Wood, J. M. J. Med. Chem. 1988, 31, 1839.

8. Cohen, N.C. Trends in Med. Chem. '88; van der Goot, H.; Dománi, G.; Pallos, L.; Timmerman, H., Eds.; Elsevier Science Publishers: Amsterdam, 1989; pp. 13-28.

9. Rahuel, J.; Priestle, J.P.; Grütter, M.G. J. Struct. Biol. 1991, 107, 227.

10. Freidinger, R.M. Trends in Pharmaceutical Sciences 1989, 10, 270.

11. Sali, A.; Veerapandian, B.; Cooper, J.B.; Foundling, S.I.; Hoover, D.J.; Blundell, T.L. The EMBO J. 1989, 8, 2179.

12. Rasetti, V.; Cohen, N.C.; Rüeger, H.; Göschke, R.; Maibaum, J.; Cumin, F.; Fuhrer, W.; Wood, J.E. Bioorg. Med. Chem. Lett. 1996, 6, 1589

13. Preliminary conformational and docking analyses were based on the calculation of the energy profiles of the C6-C7 bond torsion angles and indicated that inhibitor 11 (C7(S)-configured) should allow more favourable van der Waals interactions with the S_3/S_1 site than the corresponding C7(R)-stereoisomer, in agreement with the *in vitro* binding data for the ethyl homologues 12, 13.

14. An alternative synthesis of the prototype 8-phenyl-2(R),7(R)-dimethyl model analogue of 10 and 11 has been recently disclosed: Hanessian, S.; Raghavan, S. Bioorg. Med. Chem. Lett. 1994, 4, 1697.

15. All new compounds have been characterized at least by high resolution ¹H-NMR and/or FAB mass spectrometry.

16. a) Evans, D.A.; Ennis, M.D. J.Am.Chem.Soc. 1982, 104, 1737-1739. b) Evans, D.A.; Britton, T.C.; Ellman J.A., Tetrahedron Lett. 1987, 6141.

17. The precursors of **8a-f** were prepared similarly as **5b**,c from the *mono*-substituted benzylbromides and the corresponding (*R*)-configured N-acyl Evans auxiliaries. (*S*)-**8c** was prepared from 3-isovaleroyl-4(*S*)-benzyl-oxazolidin-2-one ((*S*)-**27c**).

18. Damon, R.E.; Coppola, G.M. Tetrahedron Letters 1990, 31, 2849.

19. Racemisation during cleavage of the oxazolidinone auxiliary by transesterification using the more basic lithium benzyl oxide has been described previously: Trimble, L.A.; Vederas, J.C. J. Am. Chem. Soc. 1986, 108, 6397.

20. Schöllkopf, U.; Groth, U.; Deng, Ch. Angewandte Chemie 1981, 93, 793.

21. Kempf, D.J. J.Org.Chem. 1986, 51, 3921.

22. In our hands, *n*-butylacrylamide proved to be more stable than methyacrylamide on storage at lower temperature, probably due to a reduced susceptibility for polymerisation (cf. Fitt, J.; Gschwend, H.W. J. Org. Chem. 1980, 45, 4257). The compound could be kept for several months at < -10 °C without signs of decomposition.

23. Stereochemical assignments within this series were tentatively based on the relative *in vitro* binding affinities of 2(R,S),4(S)configured 22 (Table) and its 2(R,S),4(R)-epimer obtained after silica gel flash chromatography separation of the C2 methylene
intermediate, with 22 being 10 fold more active towards human renin than the 4(R)-isomers. For 7c and analogues, assignment was
based on the generally observed relative ¹H-NMR up-field shifts of the *BOC*-<u>NH</u> signals (ca. 0.2 ppm, DMSO-d₆) for the 4(S)- vs. 4(R)-isomers. In all cases, the 4(R)-diastereomers corresponding to 7c appeared to be more polar by t.l.c.

24. In the course of this work, it was discovered for related N-BOC protected hydroxyethylene dipeptide isosteres substituted with different (P_3-P_1) -moieties [Ref. 5] that catalytical hydrogenation of the methylene group at the C-2 position occurred highly diastereoselectively in the presence of optically active $[Ru_2Cl_4((S)-BINAP)_2]$ ·NEt₃ [see: Takaya, H.; Noyori, R. et al., J. Am. Chem. Soc. **1987**, 109, 1596] to give the desired 2(R)-methyl isomers (H. Rüeger, personal communication).

25. The regioisomeric inhibitor 22 was obtained accordingly starting from 4-(tert-butyl)-2-hydroxy-benzoic acid.

26. Weber, A.E.; Steiner, M.G.; Krieter, P.H.; Colletti, A.E.; Tata, J.R.; Halgren, T.A.; Ball, R.G.; Doyle, J.J.; Schorn, T.W.; Stearns, R.A.; Miller, R.R.; Siegl, P.K.S.; Greenlee, W.J.; Patchett, A.A. J.Med.Chem. 1992, 35, 3755.

27. For a recent review: Rahuel, J.; Priestle, J.P.; Grütter, M.G. J. Struct. Biology 1991, 107, 227.

28. Foundling, S.I.; Cooper, J.; Watson, F.E.; Cleasby, A.; Pearl, L.H.; Sibanda, B.L.; Hemmings, A.; Wood, S.P.; Blundell, T.L.; Valler, M.J.; Norey, C.G.; Kay, J.; Boger, J.; Dunn, B.M.; Leckie, B.J.; Jones, D.M.; Atrash, B.; Hallett, A.; Szelke, M. Nature 1987, 327, 349.

29. In a parallel effort within different series of peptidomimetic renin inhibitors, a similar strategy was applied successfully to design in vitro highly potent P₃ extended dipeptide isosteres in our laboratories [Ref. 5].

(Received in Belgium 2 July 1997; accepted 29 September 1997)