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A Short Stereospecific Synthesis of a 2,6-Diarylmonoepoxylignanolide

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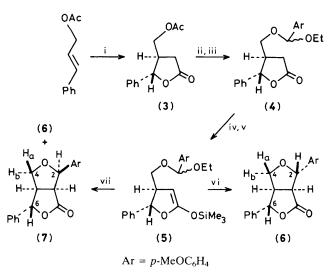
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The 2-(*p*-methoxyphenyl)-6-phenyl-3,7-dioxabicyclo[3.3.0]octan-8-one (**6**), with natural stereochemistry, has been synthesised by a stereospecific route using an intramolecular aldol reaction $[(5) \rightarrow (6)]$.

The 2,6-diaryl-3,7-dioxabicyclo[3.3.0]octane lignans¹ comprise a large group of natural products exhibiting a wide range of biological activity, and there is considerable interest in their synthesis.² The group includes a few lactones (1), 2,6diarylmonoepoxylignanolides in Weinges' terminology,³ showing *e.g.* antitumour action⁴ and plant growth inhibition.⁵ One lactone of type (1) has been prepared, by oxidation of a natural product,⁶ and also by an oxidative coupling approach⁵ which, although biomimetically interesting, is low yielding and not general.

We considered such compounds to be worthwhile targets both in their own right and as potential precursors, *via* reduction, to unsymmetrical compounds (2; $Ar^1 \neq Ar^2$). Various ingenious approaches to the symmetrical compounds (2; $Ar^1 = Ar^2$) have been described⁷ but only one route $\begin{array}{c} 0 & H & Ar^{2} \\ H & 5 & 1 \\ H & 5 & 1 \\ H & 1 & 6 \\ H & R \\ \hline \\ (1) & R_{2} = 0 \\ (2) & R = H \end{array}$

reported⁸ for the important unsymmetrical compounds. We set out here a short and stereospecific route, which should prove general, to the 2-(p-methoxyphenyl)-6-phenylmono-epoxyliganolide (6) with the geometry of the natural series (Scheme 1).



Scheme 1. Reagents: i, $Mn(OAc)_3$, AcOH; ii, H_3O^+ ; iii, ArCH-(Cl)(OEt); iv, LDA, -70 °C; v, Me_3SiCl ; vi, $TiCl_4$, -78 °C; vii, $CF_3SO_3SiMe_3$.

The manganese(III) induced radical addition of acetic acid to olefins⁹ was applied to *trans*-cinnamyl acetate to afford a single *trans*-lactone (**3**), (49%). The primary alcohol function was revealed on acid hydrolysis (78%) and reacted with 1-ethoxy-1-(*p*-methoxyphenyl)chloromethane at 0 °C, with triethylamine, to provide the mixed acetal (**4**), (53%). The lactone enol trimethylsilyl ether (**5**) was generated (92%) by successive treatments with lithium di-isopropylamide (LDA) and trimethylsilyl chloride.

Treatment of (5) with titanium tetrachloride at -78 °C gave the desired compound (6), (40%), as a single stereoisomer. This intramolecular aldol reaction impresses a *cis*-ring fusion, and the stereochemistry at the fourth chiral centre is controlled by transition-state conformation, possibly the chairlike arrangement of an enol titanium intermediate,¹⁰ with the C-2 aryl equatorial. We know of only one related application of the intramolecular aldol reaction to prepare cyclic ethers in this way.¹¹ A competing scission of the acetal to anisaldehyde was also observed; it is planned to reduce this side reaction by direction of the site of co-ordination.

Ring closure of (5) using trimethylsilyl trifluoromethanesulphonate gave a mixture of (6) and its 2-epimer (7). In this case a cyclic transition state cannot be envisaged. The stereochemistry of lactones (6) and (7) was determined by comparison of ¹H and ¹³C n.m.r. spectroscopic measurements with the substantial literature data,^{1,4–7} and by examination of proton nuclear Overhauser enhancement effects. Thus in lactone (6), irradiation of 4-H_a led to signal enhancements (measured by Fourier transform difference methods) at 2-H (5%), 6-H (5%), and 4-H_b (17%), while in lactone (7) irradiation of 4-H_a produced enhancements at 2-H (6%), 5-H (9%), and 4-H_a (23%).

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References

- 1 A. Pelter and R. S. Ward in 'Chemistry of Lignans,' ed. C. S. Rao, Andra University Press, 1978, pp. 227–275.
- 2 R. S. Ward, Chem. Soc. Rev., 1982, 75.
- 3 K. Weinges, F. Nader, and K. Künstler, in ref. 1, p. 1.
- 4 A. Ulubelen, Y. Saiki, H. Lotter, V. M. Chari, and H. Wagner,
- Planta Med., 1978, 34, 403.
 5 R. Cooper, H. E. Gottlieb, D. Lavie, and E. C. Levy, Tetrahedron, 1979, 35, 861.
- 6 A. S. R. Anjaneyula, A. M. Rao, V. K. Rao, L. Row, A. Pelter, and R. S. Ward, *Tetrahedron*, 1977, 33, 133.
- 7 A. Pelter, R. S. Ward, D. J. Watson, P. Collins, and I. T. Kay, J. Chem. Soc., Perkin Trans. 1, 1982, 175; P. Brownbridge and T. H. Chan, Tetrahedron Lett., 1980, 3427; K. K. Mahalanabis, M. Mumtaz, and V. Snieckus, *ibid.*, 1982, 3975.
- 8 A. Pelter, R. S. Ward, P. Collins, R. Venkateswarlu, and I. T. Kay, *Tetrahedron Lett.*, 1983, 523.
- 9 E. I. Heiba, R. M. Dessau, and P. G. Rodewald, J. Am. Chem. Soc., 1974, 96, 7977.
- 10 E. Nakamura, J. Shimada, Y. Horiguchi, and I. Kuwajima, Tetrahedron Lett., 1983, 3341.
- 11 G. S. Cockerill and P. Kocienski, J. Chem. Soc., Chem. Commun., 1983, 705.