

SUPPORTING INFORMATION

Pristine Simple Oxides as Visible Light Driven Photocatalysts: Highly Efficient Decomposition of Organic Compounds over Platinum-loaded Tungsten Oxide

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a) Preparation of Pt-loaded WO₃ photocatalysts

Fine particulate WO₃ with a particle size of 50–200 nm was separated from commercial WO₃ powder (99.99%, Kojundo Chemical Laboratory Co.) as follows. The WO₃ powder (5 g) was suspended in purified water (100 mL) with ultrasonic irradiation for 30 min, and then the suspension was centrifuged at 1000 rpm for 10 min. After the removal of precipitate containing large aggregates, fine particulate WO₃ with a particle size of 50–200 nm was collected by centrifugation (10000 rpm for 30 min). BET specific surface area of obtained WO₃ fine particles, which was determined by nitrogen-adsorption at 77 K, was ca. 10.5 m²g⁻¹. (before the separation: ca. 2.4 m²g⁻¹)

Platinum-loaded WO₃ samples (Pt-WO₃) were prepared as follows. An aqueous suspension containing the fine particulate WO₃ and hexachloroplatinic acid (H₂PtCl₆·6H₂O) was exposed to visible light ($\lambda > 400$ nm) provided by a 300 W Xe lamp (LX-300F, Cermex) fitted with a cutoff filter (L-42, HOYA). After 2 h of irradiation, methanol (10 vol%) was added and the suspension exposed to further irradiation for 2 h. This procedure resulted in a highly uniform dispersion of platinum particles (average size, 5 nm) on the WO₃ surface, as shown in Fig. S1. Titanium (IV) oxide (TiO₂ P25, Nippon Aerosil, Japan, ca. 55.4 m²g⁻¹) was used as a reference photocatalyst. Nitrogen-doped TiO₂ (N-TiO₂) for comparative analysis was prepared by treatment of TiO₂ powder (ST-01, Ishihara Sangyo Kaisha Ltd., ca. 320 m²g⁻¹) under NH₃ flow (50 mL min⁻¹) at 823 K for 3 h. This method is known as relatively reproducible procedure to prepare active N-TiO₂ sample, and applied in a considerable number of studies. We confirmed that the visible-light induced activities of prepared N-TiO₂ sample were comparable to those reported in the literatures. The specific surface area of N-TiO₂ sample was determined to be ca. 88.9 m²g⁻¹.

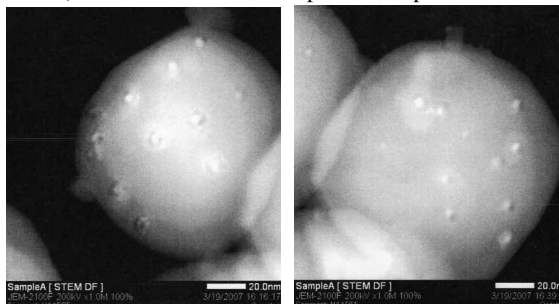


Figure S1 Scanning transmission microscopy images of Pt(1 wt%)-WO₃ samples.

b) Photoreactivity Experiments

Photocatalytic decomposition of acetic acid (AcOH) was carried out in a Pyrex reaction cell containing a suspension of the photocatalyst powder (50 mg) in an aerated aqueous AcOH solution (5 vol%, 250 mL) with continuous agitation using a magnetic stirrer. Photocatalytic decomposition of acetaldehyde (AcH) or isopropyl alcohol (IPA) was performed using a Pyrex reaction cell with internal volume of 330 mL. The photocatalyst powder (50 mg) was spread in a square (15×15 mm) on the bottom of the reactor, and gaseous AcH (1000 ppm, ca. 15 μ mol) or IPA (1200 ppm, ca. 17 μ mol) was introduced into the cell. In both cases, a 300 W Xe lamp was used as the light source. Components in the gas phase were analyzed by gas chromatography. Action spectrum analysis for AcOH decomposition was performed under monochromatic irradiation. In this case, 10 mg of photocatalyst powder was suspended in an aerated aqueous AcOH solution (5 vol%, 2 mL), which was subjected to monochromatic irradiation (with FWHM \sim 17 nm) in the range of 300–600 nm using a diffraction grating type illuminator (Jasco CRM-FD) equipped with a 300-W xenon lamp (Hamamatsu Photonics C2578-02). The intensity of irradiation, measured by a Molecron PM5200 laser power meter, was in the range of 1.98–6.52 10^{-8} einstein s⁻¹.

c) Photoacoustic Spectroscopic Measurements

Photoacoustic spectroscopic measurements were conducted using a home-made PA cell composed of an aluminum body (internal volume, ca. 1 mL) with a Pyrex window and two valves for gas flow. Three atmospheres were used in the reactions: flows of ambient air, nitrogen (N₂), or O₂ containing isopropyl alcohol (IPA) vapor. Monochromatic light (ca. 0.2 mW cm⁻²) chopped at 80 Hz was obtained from a Xe lamp as modulated probe light, and an ultraviolet light-emitting diode (365 nm, ca. 2.8 mW cm⁻²) was employed as continuous irradiation for excitation of WO₃. The PA signal was acquired by a condenser microphone buried in the cell, and amplified and monitored using a digital lock-in amplifier. In time-resolved analyses, a light-emitting diode (625 nm) was used as a probe light, and the output intensity was modulated by a digital function generator at 80 Hz. The PA spectrum of bare WO₃ under UV irradiation in the presence of air and IPA vapor exhibits an upward shift in the visible region at wavelengths longer than 470 nm, as shown in Fig. S2.

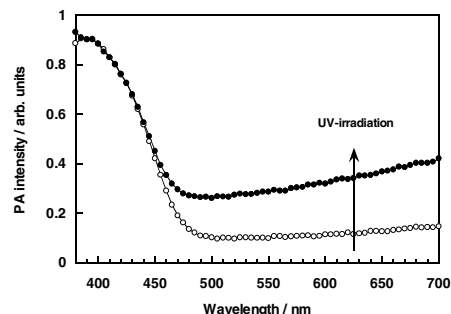


Figure S2 Photoacoustic spectra of a bare WO₃ powder before (open symbols) and after (closed symbols) UV irradiation for 30 min in air containing IPA vapor