

Addition of Difluorocarbene to 4',5'-Unsaturated Nucleosides: Synthesis and Deoxygenation Reactions of Difluorospirocyclopropane Nucleosides¹

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Received August 2, 2006



Synthetic routes to 4'-(2,2-difluorospirocyclopropane) analogues of adenosine, cytidine, and uridine are described. Treatment of 2',3'-O-isopropylidene-4',5'-unsaturated compounds derived from adenosine and uridine with difluorocarbene (generated from PhHgCF₃ and NaI) gave diastereomeric mixtures of the 2,2-difluorospirocyclopropane adducts. Stereoselectivity resulting from hindrance by the isopropylidene group favored addition at the β face. Removal of base and sugar protecting groups gave new difluorospirocyclopropane nucleoside analogues. The protected uridine analogue was converted into its cytidine counterpart via a 4-(1,2,4-triazol-1-yl) intermediate. Stannyl radical-mediated deoxygenation of the 3'-O-TBS-2'-thionocarbamate derivatives gave the 2'-deoxy products of direct hydrogen transfer. In contrast, identical treatment of the 2'-O-TBS-3'-thionocarbamate isomers resulted in opening of the vicinal difluorocyclopropane ring upon generation of a C3' radical followed by homoallylic hydrogen transfer to give 4'-(1,1-difluoroethyl)-3',4'-unsaturated nucleoside derivatives. Structural aspects and biological effect considerations are discussed.

Introduction

Relationships between structure and biological response activity are fundamental considerations in medicinal chemistry. Regioisomers, diastereomers, and enantiomers usually have increasingly subtle structural differences, but the contrasts in biological response effects can be striking. Conformational preferences are in an even more muted realm of structural effects. Small but meaningful binding energy differences can exist among molecular conformer ranges and targeted binding sites. Nucleic acid component analogues have been employed clinically for many decades, and major favorable drug versus side effects differences exist between nucleoside regioisomers,² diastereomers,³ and enantiomers.⁴

Eyring, Robins, and co-workers systematically investigated nucleoside conformational effects with CD and ORD spectros-

8876 J. Org. Chem. **2006**, 71, 8876–8883

copy.⁵ However, the pseudorotational mobility of furanosyl rings coupled with small energy differences among rotamers limited the analyses to tentative identification of rotational conformation populations about the glycosyl bond. Altona and Sundralingham defined pseudorotational parameters and correlated them with highly populated solid-state nucleoside/nucleotide conformations.⁶ Altona,⁷ Chattopadhyaya,⁸ Marquez,⁹ and others have refined and added additional parametrization to the fundamental

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Altona–Sundralingham⁶ approach for analysis of predominant conformer populations of furanosyl derivatives.

Nucleoside analogues normally are prodrug forms of active pharmaceutical agents. They must undergo interactions with proteins in cellular and compartmental membranes (including active and/or facilitated transfer by transmembrane nucleoside transporters), nucleoside kinases, and mono- and/or dinucleotide kinases and then bind with target proteins (e.g., DNA polymerases, reverse transcriptases, and/or other nucleoside triphosphate metabolizing enzymes). It would be fortuitous if binding studies with a specific target protein (without consideration of the requisite multiple interactions with other proteins in the prior transport and activation pathways) gave strong positive correlations with biological response activities attributable to selective inhibition of the targeted protein. Therefore, continuingand successful-searches for new nucleoside drug candidates necessarily involve intuitive design, synthesis, and empirical testing of novel candidate structures.

Marquez and co-workers have prepared and evaluated numerous "carbocyclic nucleosides" with a cyclopentane ring in place of the furanosyl moiety.¹⁰ Bicyclo[3.1.0]hexane ring analogues of carbocyclic nucleosides have restricted cyclopentane (furanose surrogate) ring conformations. Selective interactions of such analogues with enzymes have been noted, although it is impossible to rigorously exclude binding effects that might result from the fused cyclopropane moiety that accompany conformational restriction of the cyclopentane ring.¹¹ A favorable pseudorotational bias can be amplified with (deoxy)oligonucleotide structures in which "locked-ring" conformations are produced in the monomers by generation of analogues with bicyclo[x.y.z] surrogates of the furanosyl moiety. Altered binding effects resulting from restricted conformations have been measured.¹²

Nucleoside analogues containing a difluorocyclopropane moiety deserve attention because simple nucleobase derivatives have shown biological activity.^{13,14} Conformational constraints in fused or spiro modifications might contribute to increased potency. Fluorine substituents enhance hydrophobicity, and their potent electronegativity might contribute to long-range electronic effects such as increased anomeric bond stability. They also can function as precursors for strategically fluorinated ringopened derivatives. Convenient methods for their preparation would be advantageous.

Fluorocarbenes have singlet ground states and usually add stereospecifically to carbon-carbon double bonds without concomitant insertion into carbon-hydrogen bonds.¹⁵ A number of methods have been reported for synthesis of gem-difluorocyclopropanes via carbene addition to C=C bonds of simple, volatile alkenes.^{15–18} However, those alkenes were often used in large excess to trap the weakly electrophilic difluorocarbene, and harsh reaction conditions and/or carbene sources that are impractical from a preparative standpoint were employed. Seyferth's reagent [phenyl(trifluoromethyl)mercury] was chosen as our source of difluorocarbene¹⁹ because it is a nonhygroscopic solid that permits mild reaction conditions and easy workup. However, its preparation is not trivial,19b,20 and appropriate CAUTION should be exercised with organomercury compounds. Cech and co-workers had used bis(trifluoromethyl)mercury in their studies on difluorocarbene insertion into the oxygen-silicon bond of 4-O-silylated nucleosides.21

Results and Discussion

Our strategy for synthesis of difluorocyclopropanated nucleosides involved the preparation of suitably protected vinyl ether precursors²² followed by addition of difluorocarbene to the electron-rich double bond²³ and deprotection. Protecting groups would serve two roles: (1) replace mobile hydrogen atoms that are detrimental to the generation and lifetime of difluorocarbene and (2) exert a β -facial stereochemical bias for the carbene addition. Synthesis of the uridine and cytidine derivatives commenced with 1-(5-deoxy-2,3-O-isopropylidene-\beta-D-erythropent-4-enofuranosyl)uracil²⁴ (1) (Scheme 1). Attempted addition of difluorocarbene to 1 failed, presumably because of the presence of the acidic N3 proton. The 3-benzoyl derivative 2 underwent carbene addition smoothly to give the diastereomers 3a/3b (5:1) in 91% yield. Elucidation of the structure of the major isomer 3a by X-ray crystallography (Supporting Information) confirmed that the addition of difluorocarbene occurred predominantly at the less hindered β face of the furanose ring. The isopropylidene protecting group was removed, and the major adduct 4 was separated from the mixture. Mild basic hydrolysis gave the crystalline uracil nucleoside analogue 5 (63% overall). Aqueous ammonolysis of 3a removed the benzoyl group to give 6 (99%), which was treated with 1,2,4triazole/phosphoryl chloride/triethylamine in acetonitrile²⁵ to give the 4-(1,2,4-triazol-1-yl) derivative 7 (97%). Treatment of

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^{*a*} Reagents and conditions: (a) BzCl/Et₃N/CH₂Cl₂; (b) PhHgCF₃/NaI/ DME/90 °C; (c) HCl/H₂O/THF; (d) NH₃/H₂O/MeOH; (e) 1,2,4-triazole/ POCl₃/Et₃N/CH₃CN; (f) NH₃/H₂O/1,4-dioxane; (g) TFA/H₂O (9:1).

SCHEME 2^a



^{*a*} Reagents and conditions: (a) 2,5-hexanedione/150 °C; (b) TsCl/ pyridine/ $-20 \rightarrow 15$ °C; (c) KO'Bu/THF/0 °C; (d) PhHgCF₃/Nal/DME/90 °C; (e) TFA/H₂O (9:1); (f) NaHCO₃/H₂O.

7 with aqueous ammonia followed by removal of the isopropylidene group with TFA/H₂O (9:1), neutralization of the TFA salt, and extraction gave the cytosine nucleoside analogue **8**. The hydrochloride—hemihydrate salt of **8** (79%, overall for three steps) was obtained by acidification with 5% HCl/H₂O.

Synthesis of the adenosine analogue **14** (Scheme 2) became our next focus. A procedure reported²⁶ for *N*,*O*-peracylation of 2',3'-O-isopropylideneadenosine (**9**) followed by selective deprotection of O5' was not readily reproducible, and other methods for transient protection—deprotection were not pursued because of the sensitivity of the second *N*-acyl group. Prolonged heating of **9** in 2,5-hexanedione gave the 2,5-dimethylpyrrole derivative



FIGURE 1. X-ray crystal structure of the monomeric 14-pyridine unit.



FIGURE 2. X-ray crystal structure of the supramolecular 14-pyridine complex.

10²⁷ in reasonable yield (56%). The 2,5-dimethylpyrrole substituent at C6 is stable in the presence of basic and organomercury reagents, but it is easily cleaved with release of the 6-amino group in TFA/H₂O.²⁷ Tosylation of 10 in pyridine at low temperature followed by immediate subjection of the sensitive tosylate 11 to base-promoted elimination gave 12. Addition of difluorocarbene to the enol ether 12 proceeded less readily (\sim 60%) but with higher diastereoselectivity (**13a**/**13b**, \sim 15:1, 19 F NMR) than had the analogous reaction with 2. Starting 12 that remained in the product mixture ($\sim 20\%$) could be converted into 13 by addition of fresh PhHgCF₃, but this was not advantageous because product loss (presumably resulting from complexation with mercury species) occurred during each treatment. Target compound 14 was obtained by hydrolysis (TFA/H₂O) of the crude 13a/13b mixture followed by neutralization and chromatography. Crystallization of 14 did not occur with the usual solvents, but yellow-green blades (14-pyridine) were obtained from a pyridine-containing solution. X-ray analysis identified the structure (Figures 1 and 2), which includes hydrogen bonding between the amino group of 14 and the pyridine nitrogen atom (N-N distance: 3.6 Å, Figure 2) and π -stacking interactions between pyridine and the imidazole ring of the nucleobase. The resulting supramolecular "ladder" has parallel strands of alternating rings-a structural feature reminiscent of DNA. The crystalline complex retained pyridine, even upon prolonged heating at 90 °C in vacuo, and melting/ decomposition in a capillary tube began at ~160 °C. Analogous

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SCHEME 3^{*a*}



 a Reagents and conditions: (a) TBS-Cl/pyridine; (b) Im₂CS/toluene/70 °C; (c) Bu₃SnH/AIBN/toluene/90 °C; (d) Bu₄NF/THF; (e) 1,2,4-triazole/ POCl₃/Et₃N/acetonitrile; (f) NH₃/H₂O/1,4-dioxane.

treatment of the cytosine compound **8** did not result in formation of such a pyridine complex.

Transformations into deoxygenated derivatives were then investigated. Diols **5** and **14** were treated with TBS chloride in pyridine, and the 2'-O-TBS ethers **15** (98%) and **16** (72%) were isolated (Scheme 3). Treatment of **15** and **16** with 1,1'thiocarbonyldiimidazole in hot toluene gave the O2' and O3' thiocarbamates **17a/18a** and **17b/18b** (~1:2 ratios), respectively. The imidazole byproduct apparently catalyzed O2'/O3' equilibration of the TBS groups prior to thiocarbamate formation. Barton deoxygenation of **17a** and **18a** gave the 2'-deoxy products **19a** and **20a**, respectively (direct hydrogen transfer to the C2' radical). In contrast, the 3'-thiocarbamate derivatives **17b** and **18b** underwent deoxygenative radical generation at C3' followed by an expected²⁸ difluorocyclopropylcarbinyl to allylcarbinyl radical rearrangement and hydrogen transfer to the homoallylic center to give alkenes **19b** and **20b**, respectively.



FIGURE 3. X-ray crystal structure of 23.

Dolbier had shown that ring opening of such difluorocyclopropylcarbinyl radicals is very rapid and that the C–C bond distal to the CF₂ group is cleaved.^{28,29}

The TBS ethers 19a, 19b, 20a, and 20b were deprotected (Bu₄NF/THF) and purified [chromatography on Dowex 1×2 (OH⁻)] to give the respective uracil, **21a** and **21b**, and adenine, 22a and 22b, analogues (Scheme 3). Treatment of 19a and 19b with 1,2,4-triazole/POCl₃/Et₃N/acetonitrile²⁵ and ammonolysis of the 4-(1,2,4-triazoyl) intermediates gave the protected 4-amino compounds, which were deprotected (Bu₄NF/THF) and purified [Dowex 1×2 (OH⁻) chromatography] to give the cytosine nucleoside analogues 23 and 24. The C6 signal in ^{13}C NMR spectra of the uracil 21a and cytosine 23 compounds is split into a doublet ($J = \sim 2$ Hz), which presumably results from through space (ts) coupling with one of the fluorine atoms (F1) on the cyclopropane ring. Such coupling indicates spatial proximity between F1 and H6. It has been well documented that ¹⁹F-¹³C ts couplings can be transmitted via overlap of fluorine lone-pair and C-H bond orbitals,³⁰⁻³⁴ and weak hydrogen bonding attractions have been suggested.³³ The 2.53 Å distance between H6 and F1 in our X-ray crystal structure of 23 (Figure 3) is slightly less than the sum of the van der Waals radii, 1.20 Å for hydrogen and 1.35 Å for fluorine, ³⁵ in harmony with the observed ts couplings. It is noteworthy that \sim 2 Hz ts coupling is not present in our NMR spectra of 5 and 8, which contain the same respective nucleobases-but have hydroxyl groups at both C2' and C3'. The most highly populated solution conformation ranges must be different for the ribose-like (5 and 8) and 2'-deoxy (21a and 23) compounds.

Conformational analysis of 14•pyridine indicates that the furanosyl ring adopts a C3'-exo (₃E) pucker with a pseudorotational phase angle of $P = 185.4^{\circ}$ and a maximum puckering amplitude of $\nu_{\text{max}} = 33.8^{\circ}.9^{.36}$ Comparison of the X-ray crystal structures of 14•pyridine and 2',3'-dideoxyadenosine³⁷ (ddA)

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TABLE 1. Comparison of Solid-State Conformational Parameters for 14-Pyridine and ddA

compd	P^{a}	$ u_{ m max}{}^a$	χ^a
14	185.4	33.8	-99.7
ddA	190.4	35.7	-95.9
^a Values in degr	rees.		

(Table 1) reveals striking similarities in their pseudorotational parameters. Both are located in the southern hemisphere (*S*) of the pseudorotation cycle, they have almost equal ring puckering, and the base rotations about the glycosyl bond (characterized by χ) are nearly equivalent. It is amazing that this level of conformational similarity exists for two nucleoside analogues with such dissimilar sugar moieties, especially with respect to the cis-vicinal OH groups present in **14** and absent in ddA. Alternative substrate activity and enzyme inactivation have been observed upon incubation of 5'-deoxy-4',5'-unsaturated adenosine analogues with this enzyme are in progress.

In conlusion, we have developed synthetic routes to new difluoro-substituted spirocyclopropane derivatives of uracil 5, cytosine 8, and adenine 14 nucleosides as well as their 2'-deoxy and ring-opened 3'-deoxy analogues. Development of suitable protecting group strategies was crucial for the successful and steroselective addition of difluorocarbene to unsaturated sugar nucleosides. The heteroatom array in compounds 2 and 12 is more extensive than in any alkenes shown to undergo difluorocyclopropanation previously. We employed the base-stable 6-(2,5-dimethylpyrrol-1-yl) substituent on a nucleoside purine ring for the first time, and mild acid treatment released the masked 6-amino group of adenine. Barton deoxygenation at C2', and with concomitant cyclopropylcarbinyl radical ring-opening at C3', provided access to the structurally unique deoxynucleosides 21–24. An X-ray crystal analysis of our supramolecular 14-pyridine complex revealed similarities with DNA secondary structures. Intriguing through-space F····H-C interactions were detected by ¹³C NMR and supported by X-ray crystal distances in 21a and 23. Studies of new analogues derived from a 2-oxabicyclo[3.1.0]hexane template and biological response properties will be reported separately.

Experimental Section³⁹

3-Benzoyl-1-[(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)-2,3-O-isopropylidene-β-D-erythrofuranosyl]uracil (3a). Et₃N (4.7 mL, 3.4 g, 34.1 mmol) and BzCl (2.6 mL, 2.9 g, 22.7 mmol) were added to a stirred solution of 1^{24} (3.00 g, 11.4 mmol) in dried CH₂Cl₂ (30 mL) under N₂, and stirring was continued for 4 h. The mixture was concentrated and chromatographed (EtOAc/hexanes, $1:4 \rightarrow 1:1$) to give 3-benzoyl-1-(5-deoxy-2,3-O-isopropylidene-β-D-erythropent-4-enofuranosyl)uracil (2) (2.87 g, 68%) as a yellow syrup: UV_{max} 253 nm (\$\epsilon 20 800), min 225 nm (\$\epsilon 5600); ¹H NMR (CDCl₃) δ 1.37 (s, 3H), 1.51 (s, 3H), 4.38 (d, J = 1.8 Hz, 1H), 4.60 (d, J= 1.8 Hz, 1H), 5.06 (d, J = 6.1 Hz, 1H), 5.24 (d, J = 6.1 Hz, 1H), 5.68 (s, 1H), 5.85 (d, J = 8.3 Hz, 1H), 7.28 (d, J = 8.3 Hz, 1H), 7.49-7.53 (m, 2H), 7.65-7.69 (m, 1H), 7.91-7.94 (m, 2H); ¹³C NMR (CDCl₃) δ 25.1, 26.4, 79.2, 82.6, 87.2, 96.5, 102.2, 113.6, 129.1, 130.3, 130.8, 135.2, 142.5, 148.9, 161.7, 162.3, 168.1; MS *m*/*z* 370 ([M⁺], 7), 355, 295, 265, 105 ([PhCO⁺], 100), 77; HRMS (C₁₉H₁₈N₂O₆) calcd 370.1164, found 370.1155.

Powdered NaI (1.50 g, 10.0 mmol) was stirred and heated (170 °C, oil bath) under vacuum for 1 h in a flask (250 mL) equipped with a Teflon valve. The bath was allowed to cool to 110 °C, and a solution of 2 (0.86 g, 2.34 mmol) in dried CH_2Cl_2 (2 mL) was injected through a septum (under N₂). Volatiles were evaporated in vacuo, and heating at 110 °C was continued for 1 h. The bath was allowed to cool to ambient temperature, and solid PhHgCF₃ (1.22 g, 3.51 mmol) was added (under N₂). The flask was evacuated and cooled to -78 °C. Dried DME (~10 mL) was transferred under vacuum from a blue solution containing sodium and benzophenone. The reaction mixture was heated at 90 °C overnight, and volatiles were evaporated. The concentrated mixture was deposited on silica gel and chromatographed (EtOAc/hexanes, $1:4 \rightarrow 1:1$) to give **3a/3b** (~5:1; 0.89 g, 91%) as a yellow oil that was recrystallized (MeOH) to give an analytical sample of 3a: UV max 253 nm (*e* 18 800), min 225 nm (*e* 5800); ¹H NMR (CDCl₃) δ 1.36 (s, 3H), 1.53 (s, 3H), 1.65–1.80 (m, 2H), 5.00 (d, J = 6.1Hz, 1H), 5.38 (d, *J* = 6.1 Hz, 1H), 5.51 (s, 1H), 5.84 (dd, *J* = 2.2, 8.1 Hz, 1H), 7.26 (d, J = 8.1 Hz, 1H), 7.50 (t, J = 7.8 Hz, 2H), 7.67 (t, J = 7.8 Hz, 1H), 7.93 (d, J = 7.8 Hz, 2H); ¹⁹F NMR $(CDCl_3) \delta 136.2 \text{ (dm, } J = 164.8 \text{ Hz}, 1\text{F}), 144.9 \text{ (ddd, } J = 7.3,$ 14.6, 164.8 Hz, 1F); ¹³C NMR (Me₂CO- d_6) δ 17.4 (t, J = 10.7Hz), 25.1, 26.3, 73.8 (t, J = 9.1 Hz), 79.4, 85.0, 98.1, 102.9, 109.5 (dd, *J* = 291.5, 299.8 Hz), 113.6, 129.1, 130.6, 130.9, 135.3, 143.3, 149.6, 161.8, 168.1; MS m/z 420 ([M⁺], 100), 405, 377, 361, 241, 205, 188, 162; HRMS (C₂₀H₁₈F₂N₂O₆) calcd 420.1133, found 420.1130. Anal. Calcd for C₂₀H₁₈F₂N₂O₆: C, 57.14; H, 4.32; N, 6.66. Found: C, 57.30; H, 4.50; N, 6.64.

1-[(S)-4,4-C-(1,1-Difluoroethane-1,2-diyl)-β-D-erythrofuranosyl]uracil (5). To a stirred solution of 3a/3b (0.75 g, 1.79 mmol) in THF (10 mL) was added HCl/H₂O (3 M, 10 mL), and stirring was continued until the deprotection was complete (~ 6 h; TLC, EtOAc). Volatiles were evaporated in vacuo, and the residue was neutralized (solid NaHCO₃) and extracted (EtOAc, 5×10 mL). The combined extract was applied to a silica gel column, and elution (EtOAc/hexanes, $1:1 \rightarrow$ EtOAc) gave 3-benzoyl-1-[(S)-4,4-C-(1,1difluoroethane-1,2-diyl)- β -D-erythrofuranosyl]uracil (4) (0.49 g, 72%). This material was dissolved (MeOH, 12 mL), NH₃/H₂O (30%, 4 mL) was added, and the solution was stirred at ambient temperature for 1 h. Volatiles were evaporated, and the residue was chromatographed (EtOAc, then MeOH) to give a syrup that was recrystallized (EtOH) to give 5 (0.31 g, 87%) as colorless needles: mp 225–226 °C; UV_{max} 260 nm (ϵ 10 200), min 230 nm $(\epsilon 2800)$; ¹H NMR (Me₂CO-d₆) δ 1.60–1.66 (m, 1H), 1.85–1.91 (m, 1H), 2.88 (s, 1H), 2.91 (s, 1H), 4.39 (d, J = 4.9 Hz, 1H), 4.67 (dd, J = 4.9, 5.9 Hz, 1H), 5.69 (d, J = 8.3 Hz, 1H), 6.07 (d, J =5.9 Hz, 1H), 7.65 (d, J = 8.3 Hz, 1H); ¹⁹F NMR (Me₂CO- d_6) δ 134.4 (dm, J = 163.9 Hz, 1F), 141.9 (ddd, J = 7.3, 14.5, 163.0 Hz, 1F); ¹³C NMR (Me₂CO- d_6) δ 18.5 (t, J = 9.8 Hz), 68.8, 71.4 (t, *J* = 9.6 Hz), 74.9, 91.1, 103.5, 111.3 (dd, *J* = 292.5, 298.4 Hz), 141.5, 151.6, 163.4; FAB-MS *m*/*z* 321 ([M⁺ - H + 2Na], 15), 299 ([M⁺ + Na], 10), 277, 271, 259, 237 (100), 217, 213; HRMS (C₁₀H₁₀F₂N₂O₅Na) calcd 299.0455, found 299.0448. Anal. Calcd for C₁₀H₁₀F₂N₂O₅: C, 43.49; H, 3.64; N, 10.14. Found: C, 43.49; H, 3.82; N, 10.19.

1-[(*S*)-4,4-*C*-(1,1-Difluoroethane-1,2-diyl)-β-D-erythrofuranosyl]cytosine (8). A solution of NH₃/H₂O (30%, 5 mL) was added to a stirred solution of **3a** (0.71 g, 1.71 mmol) in MeOH (20 mL), and stirring was continued at ambient temperature for 1 h. Volatiles were evaporated, and the residue was chromatographed (EtOAc/ hexanes, 1:2) to give 1-[(*S*)-4,4-*C*-(1,1-difluoroethane-1,2-diyl)-2,3-*O*-isopropylidene-β-D-erythrofuranosyl]uracil (**6**) (0.53 g, 99%) as a syrup.

Et₃N (2.34 mL, 1.71 g, 16.9 mmol) was added dropwise to a stirred, cooled (\sim 0 °C) mixture of 1,2,4-triazole (1.22 g, 17.8 mmol), POCl₃ (0.35 mL, 0.57 g, 3.72 mmol), and acetonitrile (10.5 mL). A solution of **6** (0.53 g, 1.69 mmol) in acetonitrile (6.6 mL) was added to this mixture, and stirring was continued at ambient temperature for 2 h. Et₃N (1.59 mL, 1.16 g, 11.5 mmol) and H₂O

⁽³⁸⁾ Yuan, C.-S.; Liu, S.; Wnuk, S. F.; Robins, M. J.; Borchardt, R. T. In *Advances in Antiviral Drug Design*; De Clercq, E., Ed.; JAI Press: Greenwich, CT, 1996; Vol. 2, pp 41–88.

⁽³⁹⁾ See the Supporting Information for general experimental details.

(0.66 mL) were added, and stirring was continued for 10 min. Volatiles were evaporated, and the residue was partitioned [icecold, saturated NaHCO₃/H₂O (30 mL)//CH₂Cl₂ (30 mL)]. The aqueous phase was extracted (CH₂Cl₂, 30 mL), and the combined organic phase was washed (brine, 50 mL) and dried (MgSO₄). Volatiles were evaporated to give 1-[(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)-2,3-O-isopropylidene- β -D-erythrofuranosyl]-4-(1,2,4-triazol-1-yl)pyrimidin-2-one (7) (0.60 g, 97%) as a yellow foam: UV max 314, 251 nm (\$\epsilon 5500, 14000), min 280, 232 nm (\$\epsilon 2300, \$\epsilon 14000), min 280, 232 nm (\$\epsilon 2300, \$\epsilon 14000), \$\epsilon 14000, \$\epsilon 140000, \$\epsilon 14000, \$\epsilon 14000 6600); ¹H NMR (CDCl₃) δ 1.43 (s, 3H), 1.57 (s, 3H), 1.74–1.81 (m, 1H), 1.84-1.92 (m, 1H), 5.12 (d, J = 5.9 Hz, 1H), 5.35 (d, J= 5.9 Hz, 1H), 5.73 (s, 1H), 7.08 (d, J = 7.3 Hz, 1H), 7.91 (d, J= 7.3 Hz, 1H), 8.13 (s, 1H), 9.27 (s, 1H); ¹⁹F NMR (CDCl₃) δ 135.8 (ddd, J = 5.5, 14.6, 164.8 Hz, 1F), 143.6 (ddd, J = 7.3, 14.6, 164.8 Hz, 1F); ¹³C NMR (CDCl₃) δ 17.7 (t, J = 11.1 Hz), 25.4, 26.4, 74.6 (t, J = 9.2 Hz), 79.1, 85.3, 95.2, 99.1, 109.5 (dd, J = 292.2, 299.8 Hz), 113.6, 143.4, 149.4, 154.0, 154.5, 160.0; EI-MS m/z 367 ([M⁺], 8), 352, 308, 216, 204, 188, 164 (100), 162, 148; HRMS (C₁₅H₁₅F₂N₅O₄) calcd 367.1092, found 367.1099.

A solution of NH₃/H₂O (30%, 2.2 mL) was added to a stirred solution of 7 in 1,4-dioxane (7 mL), and stirring was continued at ambient temperature for 19 h. Volatiles were evaporated, and the residue was dissolved (TFA/H₂O, 3.0:0.4 mL). The solution was stirred overnight, and volatiles were evaporated in vacuo (<30 °C). H₂O (20 mL) and NaHCO₃ were added to the residue, and the mixture was extracted (EtOAc, 10×100 mL). Evaporation of volatiles gave 8 as yellow oil. A solution of this material in MeOH was treated with 5% HCl/H2O and filtered. The filtrate was concentrated, and excess HCl was removed by repetitive addition and evaporation of MeOH/H2O. The resulting red oil was recrystallized (EtOH) to give 8·HCl·0.5H₂O (0.40 g, 79%) as a white powder: mp 218-220 °C; UV_{max} 278 nm (\$\epsilon\$ 10 600), min 245 nm $(\epsilon 4600)$; ¹H NMR (D₂O) δ 1.79–1.87 (m, 1H), 1.92–2.01 (m, 1H), 4.40 (d, J = 4.9 Hz, 1H), 4.74 (t, J = 5.3 Hz, 1H), 6.04 (d, J = 5.9 Hz, 1H), 6.25 (d, J = 8.3 Hz, 1H), 7.86 (d, J = 8.3 Hz, 1H); ¹⁹F NMR (D₂O) δ 135.0 (ddd, J = 5.5, 14.6, 164.8 Hz, 1F), 142.0 (ddd, J = 7.3, 16.5, 164.8 Hz, 1F); ¹³C NMR (D₂O) δ 17.5 (t, J = 10.3 Hz), 68.5, 70.9 (dd, J = 8.6, 11.1 Hz), 74.5, 91.8,95.8, 109.7 (dd, J = 290.6, 299.6 Hz), 144.9, 148.5, 159.4; FAB-MS (thioglycerol) m/z 298 ([M⁺ + Na], 100); HRMS (C₁₀H₁₁F₂N₃O₄-Na) calcd 298.0615, found 298.0616. Anal. Calcd for $C_{10}H_{11}F_2N_3O_4$. HCl·0.5H₂O: C, 37.45; H, 4.09; N, 13.10. Found: C, 37.14; H, 4.40; N, 12.92.

9-(2,3-*O*-**Isopropylidene**-*β*-**D**-**ribofuranosyl**)-**6**-(**2,5**-**dimeth-ylpyrrol-1-yl)purine**²⁷ (**10**). A solution of 2',3'-*O*-isopropylideneadenosine (**9**) (1.00 g, 3.26 mmol) in 2,5-hexanedione (4 mL) was heated at 150 °C for 2 days. Volatiles were evaporated in vacuo, and the oily residue was chromatographed (EtOAc/hexanes, 1:1) to give **10** (0.70 g, 56%) as an orange oil: UV_{max} 283 nm (*ε* 12 100), min 247 nm (*ε* 4200); ¹H NMR (CDCl₃) *δ* 1.42 (s, 3H), 1.69 (s, 3H), 2.22 (s, 6H), 3.82–3.88 (m, 1H), 4.02 (dd, *J* = 1.5, 12.7 Hz, 1H), 4.60 (s, 1H), 5.17 (dd, *J* = 1.5, 5.8 Hz, 1H), 5.31 (dd, *J* = 4.4, 5.8 Hz, 1H), 5.46 (d, *J* = 10.7 Hz, 1H), 6.00 (s, 2H), 6.03 (d, *J* = 4.4 Hz, 1H), 8.22 (s, 1H), 8.93 (s, 1H); ¹³C NMR (CDCl₃) *δ* 13.0, 24.7, 26.9, 62.2, 81.2, 83.6, 86.2, 92.3, 108.6, 113.6, 128.8, 129.2, 144.2, 149.8, 151.6, 152.1; MS *m*/z 385 ([M⁺], 80), 370, 354, 296, 213 (100), 198; HRMS (C₁₉H₂₃N₅O₄) calcd 385.1750, found 385.1753.

9-[(*S*)-4,4-*C*-(1,1-Difluoroethane-1,2-diyl)- β -D-erythrofuranosyl]adenine (14). A solution of 10 (4.5 g, 11.7 mmol) in dried pyridine (22 mL) was cooled to -20 °C, and TsCl (3.4 g, 17.8 mmol) was added. Stirring was continued for 6 h at -20 °C, and the mixture was allowed to warm to 15 °C overnight. H₂O (100 mL) was added with stirring, and the solution was extracted (CH₂-Cl₂, 2 × 50 mL). The combined organic phase was dried (MgSO₄), and volatiles were evaporated to give 9-(2,3-*O*-isopropylidene-5-*O*-tosyl- β -D-ribofuranosyl)-6-(2,5-dimethylpyrrol-1-yl)purine (11) (6.0 g, 95%) as a yellow syrup: UV_{max} 284 nm (ϵ 12 500), min 247 nm (ϵ 4300); ¹H NMR (CDCl₃) δ 1.40 (s, 3H), 1.63 (s, 3H), 2.22 (s, 6H), 2.42 (s, 3H), 4.25–4.31 (m, 2H), 4.50–4.54 (m, 1H), 5.08 (dd, J = 2.9, 6.3 Hz, 1H), 5.42 (dd, J = 2.4, 6.3 Hz, 1H), 6.00 (s, 2H), 6.23 (d, J = 2.9 Hz, 1H), 7.29 (d, J = 8.0 Hz, 2H), 7.70 (d, J = 8.0 Hz, 2H), 8.19 (s, 1H), 8.87 (s, 1H); ¹³C NMR (CDCl₃) δ 12.9, 20.8, 24.5, 26.3, 68.6, 80.6, 83.4, 83.6, 90.0, 108.3, 113.8, 127.1, 128.3, 129.0, 129.2, 131.4, 134.8, 144.7, 149.3, 151.4, 152.1; MS m/z 539 ([M⁺], 40), 524, 403, 367, 212 ([M⁺ – C₁₁H₁₀N₅], 100); HRMS (C₂₆H₃₁N₅O₆S) calcd 539.1838, found 539.1829.

A solution of potassium tert-butoxide (1.8 g, 16.7 mmol) in dried THF (40 mL) was added to a stirred, cooled (-10 °C) solution of 11 (6.0 g, 11.1 mmol) in dried THF (40 mL), and the dark solution was protected from moisture and stirred for 30 min at -10 °C. The reaction mixture was deposited on silica gel and chromatographed (EtOAc/hexanes, 1:4) to give 9-(5-deoxy-2,3-O-isopropylidene-β-D-erythro-pent-4-enofuranosyl)-6-(2,5-dimethylpyrrol-1yl)purine (12) (2.6 g, 68%) as a pale yellow oil: UV_{max} 284 nm (ϵ 13 500), min 246 nm (ϵ 4400); ¹H NMR (CDCl₃) δ 1.46 (s, 3H), 1.61 (s, 3H), 2.21 (s, 6H), 4.56 (d, J = 2.4 Hz, 1H), 4.69 (d, J =2.4 Hz, 1H), 5.38 (d, J = 5.9 Hz, 1H), 5.61 (d, J = 5.9 Hz, 1H), 5.99 (s, 2H), 6.36 (s, 1H), 8.12 (s, 1H), 8.92 (s, 1H); ¹³C NMR $(CDCl_3)$ δ 13.0, 25.1, 26.2, 79.1, 82.1, 88.0, 90.0, 108.6, 113.6, 128.4, 129.2, 143.6, 149.7, 151.9, 152.1, 161.2; MS m/z 367 ([M⁺], 60), 352, 226, 212 ([$M^+ - C_{11}H_{10}N_5$], 100); HRMS ($C_{19}H_{21}N_5O_3$) calcd 367.1644, found 367.1656.

Treatment of **12** (1.14 g, 3.1 mmol) in dried DME (8 mL) with the reagent prepared from NaI (1.4 g, 9.3 mmol) and PhHgCF₃ (2.19 g, 6.3 mmol) (according to the procedure described for **2** \rightarrow **3a/3b**) gave a mixture that was deposited on silica gel and chromatographed (EtOAc/hexanes, 1:4 \rightarrow 1:1) to give recovered **12** (0.26 g, 20%) plus a mixture of 9-[(*R/S*)-4,4-*C*-(1,1-difluoroethane-1,2-diyl)-2,3-*O*-isopropylidene- β -D-erythrofuranosyl]-6-(2,5dimethylpyrrol-1-yl)purine (**13a/13b**) (~15:1; 0.78 g, 60%) as a yellow oil: ¹H NMR (CDCl₃, major isomer) δ 1.50 (s, 3H), 1.65 (s, 3H), 1.65–1.95 (m, 2H), 2.21 (s, 6H), 5.43 (d, *J* = 5.9 Hz, 1H), 5.74 (d, *J* = 6.1 Hz, 1H), 6.00 (s, 2H), 6.29 (s, 1H), 8.13 (s, 1H), 8.94 (s, 1H); ¹⁹F NMR (CDCl₃) δ 135.6 (ddd, *J* = 5.5, 14.6, 163.0 Hz, 1F), 145.3 (ddd, *J* = 7.3, 14.6, 163.0 Hz, 1F); EI-MS *m/z* 417 ([M⁺], 80), 212, 77 (100); HRMS (C₂₀H₂₁F₂N₅O₃) calcd 417.1612, found 417.1617.

A solution of 13a/13b (0.85 g, 2.0 mmol) in TFA/H₂O (9:1, 20 mL) was stirred at ambient temperature for 3 h. Volatiles were evaporated in vacuo (<30 °C), and saturated NaHCO₃/H₂O (20 mL) was added. The mixture was extracted (EtOAc, 10×100 mL), and volatiles were evaporated from the combined organic phase to give a red oil. Chromatography (EtOAc/MeOH, 20:1) gave 14 (0.35 g, 57%) as a syrup: UV_{max} 260 nm (ϵ 12 000), min 229 nm (ϵ 3300); ¹H NMR (Me₂CO-d₆) δ 1.59-1.67 (m, 1H), 1.87-1.95 (m, 1H), 2.87 (br s, 2H), 4.63 (d, J = 5.4 Hz, 1H), 5.26 (t, J = 5.4 Hz, 1H), 6.22 (d, J = 5.4 Hz, 1H), 6.64 (br s, 2H), 8.20 (s, 1H), 8.23 (s, 1H); ¹⁹F NMR (Me₂CO- d_6) δ 134.5 (ddd, J = 5.5, 16.5, 163.0 Hz, 1F), 141.8 (ddd, J = 7.3, 16.5, 163.0 Hz, 1F); ¹³C NMR (DMSO d_6) δ 17.9 (t, J = 8.8 Hz), 68.1, 70.3 (t, J = 9.2 Hz), 73.2, 87.6, 110.9 (dd, *J* = 292.2, 299.1 Hz), 119.4, 140.2, 149.9, 153.0, 156.3; MS m/z 299 ([M⁺], 7), 270, 178 (100), 148, 136; HRMS (C11H11F2N5O3) calcd 299.0830, found 299.0828. This material was recrystallized (pyridine/MeOH/Et₂O) to give yellow-green blades of 14 pyridine: mp \sim 160–166 °C dec; ¹H NMR (CD₃OD) δ 1.64 (ddd, J = 4.9, 10.3, 15.6 Hz, 1H), 1.89 (ddd, J = 7.3, 9.8, 16.1 Hz, 1H), 4.45 (d, J = 5.4 Hz, 1H), 5.13 (t, J = 5.9 Hz, 1H), 6.19 (d, J = 6.3 Hz, 1H), 7.40–7.43 (m, 2H), 7.81–7.85 (m, 1H), 8.20 (s, 1H), 8.23 (s, 1H), 8.51–8.53 (m, 2H); ¹⁹F NMR (CD₃OD) δ 134.7 (ddd, J = 4.9, 17.1, 163.6 Hz, 1F), 142.3 (ddd, J = 7.3, 17.1, 163.6 Hz, 1F); ¹³C NMR (CD₃OD) δ 17.3 (t, J = 10.3 Hz), 68.4, 70.7 (dd, J = 8.4, 10.7 Hz), 74.5, 88.7, 110.2 (dd, J = 293.0, 297.5 Hz), 119.3, 124.4, 137.2, 140.0, 148.9, 149.8, 152.9, 156.2. X-ray crystallography confirmed the 1:1 complex stoichiometry.

1-[2-Deoxy-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)-β-D-glycerotetrofuranosyl]uracil (21a). TBS-Cl (0.73 g, 4.8 mmol) was added

to a stirred solution of 5 (0.44 g, 1.6 mmol) in dried pyridine (5 mL), and stirring was continued at ambient temperature for 2 days. Volatiles were evaporated in vacuo, and the residue was chromatographed (EtOAc/hexanes, $1:4 \rightarrow 1:1$) to give 1-[2-O-tert-butyldimethylsilyl-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)- β -D-erythrofuranosyl]uracil (15) (0.60 g, 97%): mp 175-177 °C; UV_{max} 260 nm (e 10 400), min 229 nm (ϵ 2600); ¹H NMR (Me₂CO-d₆) δ 0.10 (s, 3H), 0.13 (s, 3H), 0.90 (s, 9H), 1.58-1.70 (m, 1H), 1.82-1.95 (m, 1H), 4.29 (t, J = 5.4 Hz, 1H), 4.42 (d, J = 5.8 Hz, 1H), 4.82 (t, J = 5.7 Hz, 1H), 5.75 (d, J = 8.1 Hz, 1H), 6.15 (d, J = 5.8 Hz, 1H), 7.71 (d, J = 8.3 Hz, 1H), 10.25 (s, 1H); ¹⁹F NMR (Me₂CO d_6) δ 134.1 (dm, $J = \sim 163$ Hz, 1F), 142.0 (dm, $J = \sim 163$ Hz, 1F); ¹³C NMR (Me₂CO- d_6) δ -4.9, -4.7, 18.5 (t, J = 9.8 Hz), 18.7, 26.1, 69.0, 71.2 (dd, J = 8.2, 11.5 Hz), 76.2, 90.0, 103.6, 111.1 (dd, J = 292.5, 298.4 Hz), 141.1, 151.5, 163.6; FAB-MS m/z 435 ([M⁺ – H + 2Na], 50), 413 ([M⁺ + Na], 100), 401, 391; HRMS (C₁₆H₂₄F₂N₂O₅SiNa) calcd 413.1320, found 413.1322. Anal. Calcd for C₁₆H₂₄F₂N₂O₅Si: C, 49.22; H, 6.20; N, 7.17. Found: C, 49.40; H, 6.06; N, 7.07.

A stirred solution of 15 (0.13 g, 0.3 mmol) and 1,1'-thiocarbonyldiimidazole (0.09 g, 0.5 mmol) in dried toluene (2 mL) was heated at 70 °C for 30 min. Volatiles were evaporated in vacuo, and chromatography of the residue (EtOAc/hexanes, $1:2 \rightarrow 1:1$) separated two isomers. The more polar 1-{3-O-tert-butyldimethylsilyl-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)-2-O-[(imidazol-1-yl)thiocarbonyl]- β -D-erythrofuranosyl}uracil (17a) (38 mg, 23%) had: UV_{max} 259 nm (ϵ 17 500), min 229 nm (ϵ 6200); ¹H NMR (CDCl₃) δ -0.06 (s, 3H), 0.06 (s, 3H), 0.83 (s, 9H), 1.67-1.76 (m, 1H), 1.81-1.90 (m, 1H), 5.02 (d, J = 4.9 Hz, 1H), 5.80 (d, J= 7.8 Hz, 1H), 5.99 (t, J = 4.9 Hz, 1H), 6.18 (d, J = 4.9 Hz, 1H), 7.07 (s, 1H), 7.29 (d, J = 7.8 Hz, 1H), 7.63 (s, 1H), 8.38 (s, 1H), 10.54 (br s, 1H); ¹⁹F NMR (CDCl₃) δ 135.1 (dm, J = 169 Hz, 1F), 141.6 (dm, J = 169 Hz, 1F); ¹³C NMR (CDCl₃) δ -5.3, -5.1, 17.7, 17.9 (t, J = 10.1 Hz), 25.3, 67.7, 70.2 (dd, J = 8.1, 12.1 Hz), 82.6, 89.3, 103.7, 108.9 (dd, J = 294.1, 297.6 Hz), 117.9, 131.2, 137.1, 140.2, 150.1, 163.4, 182.5; FAB-MS (glycerol) m/z 501 ([M+ + H], 52), 373 (100), 261; HRMS ($C_{20}H_{27}F_2N_4O_5SSi$) calcd 501.1439, found 501.1454. The less polar 1-{2-O-tert-butyldimethylsilyl-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)-3-O-[(imidazol-1yl)thiocarbonyl]- β -D-erythrofuranosyl}uracil (17b) (65 mg, 40%) had: mp 205-206 °C dec; UV_{max} 260 nm (ε 16 500), min 232 nm (ϵ 6200); ¹H NMR (CDCl₃) δ 0.01 (s, 6H), 0.73 (s, 9H), 1.82– 1.97 (m, 2H), 4.90 (t, J = 5.9 Hz, 1H), 5.86 (dd, J = 2.2, 8.3 Hz, 1H), 5.99 (d, J = 6.3 Hz, 1H), 6.28 (d, J = 5.4 Hz, 1H), 7.11 (s, 1H), 7.22 (d, J = 8.3 Hz, 1H), 7.68 (s, 1H), 8.15 (s, 1H), 8.41 (br s, 1H); ¹⁹F NMR (CDCl₃) δ 134.7 (ddd, J = 4.3, 15.0, 164.5 Hz, 1F), 143.8 (ddd, J = 6.4, 15.0, 164.5 Hz, 1F); ¹³C NMR (Me₂CO d_6) δ -5.1, -4.9, 18.2, 19.6 (t, J = 10.3 Hz), 25.7, 69.0 (t, J =10.3 Hz), 74.8, 79.7, 92.2, 104.0, 110.4 (dd, *J* = 293.4, 298.0 Hz), 119.4, 132.1, 138.0, 142.0, 151.5, 163.2, 185.3; FAB-MS (glycerol) m/z 501 ([M⁺ + H], 20), 443, 373, 353, 261 (100); HRMS (C₂₀H₂₇F₂N₄O₅SSi) calcd 501.1439, found 501.1437.

Bu₃SnH (0.6 g, 2.0 mmol) was added to a solution of 17a (0.25 g, 0.5 mmol) and AIBN (20 mg, 0.1 mmol) in dried, deoxygenated toluene. The mixture was heated at 90 °C for 50 min, and volatiles were evaporated. The residual oil was chromatographed (EtOAc/ hexanes, $1:4 \rightarrow 1:2$) to give 1-[3-O-tert-butyldimethylsilyl-2-deoxy-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)- β -D-glycero-tetrofuranosyl]uracil (19a) (0.12 g, 64%): UV_{max} 260 nm (\$\epsilon\$ 10 200), min 229 nm (ε 2500); ¹H NMR (CDCl₃) δ 0.07 (s, 3H), 0.10 (s, 3H), 0.91 (s, 9H), 1.68–1.76 (m, 2H), 2.35 (dt, J = 6.3, 14.2 Hz, 1H), 2.59 (ddd, J = 2.4, 6.3, 14.2 Hz, 1H), 4.51 (dd, J = 2.4, 5.9 Hz, 1H),5.79 (d, J = 8.3 Hz, 1H), 6.38 (t, J = 6.3 Hz, 1H), 7.32 (d, J = 8.3Hz, 1H), 8.47 (br s, 1H); ¹⁹F NMR (CDCl₃) δ 135.0 (dm, J = 164Hz, 1F), 142.1 (dm, J = 165 Hz, 1F); ¹³C NMR (CDCl₃) δ -5.2, -4.9, 17.7 (t, J = 10.1 Hz), 17.8, 25.5, 42.3, 69.0, 72.2 (dd, J =7.8, 10.5 Hz), 86.5, 103.0, 110.1 (dd, J = 294.8, 299.3 Hz), 139.0, 150.3, 163.6; FAB-MS (thioglycerol) m/z 397 ([M⁺ + Na], 100), 375, 263, 243, 237; HRMS ($C_{16}H_{24}F_2N_2O_4SiNa$) calcd 397.1371, found 397.1375.

Bu₄NF/THF (1 M; 2.0 mL, 2.0 mmol) was added to a stirred solution of 19a (0.60 g, 1.6 mmol) in THF (10 mL), and stirring was continued for 1 h. Volatiles were evaporated, and the residue was dissolved (H₂O, 20 mL) and applied to a column of Dowex 1 \times 2 (OH⁻) resin (in H₂O). Elution [H₂O \rightarrow MeOH \rightarrow AcOH/MeOH (1:3)] and evaporation of volatiles from UV-active fractions gave **21a** (0.36 g, 87%) as colorless crystals. An analytically pure sample was obtained by several recrystallizations (EtOH): mp 205-206 °C; UV_{max} 260 nm (ϵ 9700), min 229 (ϵ 3600); ¹H NMR (Me₂-CO- d_6) δ 1.65–1.72 (ddd, J = 4.9, 10.3, 16.6 Hz, 1H), 1.83–1.90 (ddd, J = 6.8, 9.8, 17.1 Hz), 2.54-2.59 (ddd, J = 2.4, 6.8, 14.2)Hz, 1H), 2.66–2.72 (dt, J = 14.2, 6.6 Hz, 1H), 4.59 (dd, J = 2.0, 6.3 Hz, 1H), 5.67 (d, J = 8.3 Hz, 1H), 6.48 (t, J = 6.8 Hz, 1H), 7.57 (d, J = 8.3 Hz, 1H); ¹⁹F NMR (CD₃OD) δ 134.6 (ddd, J =5.4, 16.5, 163.0 Hz, 1F), 141.9 (ddd, *J* = 7.3, 16.5, 163.0 Hz, 1F); ¹³C NMR (CD₃OD) δ 17.9 (t, J = 10.3 Hz), 41.8, 69.9, 73.9 (dd, J = 7.6, 10.7 Hz), 88.4, 103.3, 112.2 (dd, J = 292.2, 299.1 Hz), 142.2 (d, J = 1.8 Hz), 152.5, 166.6; EI-MS m/z 260 ([M⁺], 5), 232, 188, 167, 149, 128, 113, 96 (100); HRMS ($C_{10}H_{10}F_2N_2O_4$) calcd 260.0608, found 260.0598. Anal. Calcd for C₁₀H₁₀F₂N₂O₄: C, 46.16; H, 3.87; N, 10.77. Found: C, 46.30; H, 3.85; N, 10.60.

1-[3-Deoxy-4-(1,1-difluoroethyl)-β-D-glycero-tetr-3-enofuranosyl]uracil (21b). Treatment of 17b (0.25 g, 0.5 mmol) with Bu₃-SnH/AIBN/toluene (as described for **17a** → **19a**) gave 1-[2-*O*-tertbutyldimethylsilyl-3-deoxy-4-(1,1-difluoroethyl)-β-D-glycero-tetr-3-enofuranosyl]uracil (**19b**) (0.16 g, 83%): UV_{max} 259 nm (ϵ 10 300), min 231 nm (ϵ 3700); ¹H NMR (CDCl₃) δ 0.07 (s, 3H), 0.10 (s, 3H), 0.87 (s, 9H), 1.81 (t, J = 18.8 Hz, 3H), 4.97 (s, 1H), 5.46 (s, 1H), 5.78 (d, J = 7.8 Hz, 1H), 6.32 (s, 1H), 7.02 (d, J = 7.8 Hz, 1H), 10.09 (br s, 1H); ¹⁹F NMR (CDCl₃) δ 92.9 (dq, J = 18.3, 265.5 Hz, 1F), 97.4 (dq, J = 18.3, 265.5 Hz, 1F); ¹³C NMR (CDCl₃) δ -4.81, -4.76, 17.9, 21.9 (t, J = 26.5 Hz), 25.5, 79.8, 94.0, 102.7, 103.6, 116.2 (t, J = 236.6 Hz), 138.8, 149.8, 155.6 (dd, J = 30.2, 36.6 Hz), 163.4; FAB-MS (thioglycerol) *m*/z 397 ([M⁺ + Na], 100), 375, 359, 339, 317, 263; HRMS (C₁₆H₂₄F₂N₂O₄-SiNa) calcd 397.1371, found 397.1372.

Bu₄NF/THF (1 M; 1.4 mL, 1.4 mmol) was added to a stirred solution of 19b (0.26 g, 0.695 mmol) in THF (5 mL), and stirring was continued for 15 min. Volatiles were evaporated, and the residue was dissolved (H₂O, 10 mL) and applied to a column of Dowex 1 \times 2 (OH⁻) resin (in H₂O). Elution [H₂O \rightarrow MeOH \cdot AcOH/MeOH (1:3)], and evaporation of volatiles from UV-active fractions gave 21b (0.15 g, 82%) as an oil that crystallized upon standing. An analytically pure sample was obtained by several recrystallizations (EtOH): mp 148-150 °C; UV_{max} 260 nm (e 9800), min 231 nm (ϵ 2800); ¹H NMR (CD₃OD) δ 1.82 (t, J = 18.6 Hz, 3H), 5.05 (m, 1H), 5.60 (m, 1H), 5.73 (d, J = 8.3 Hz, 1H), 6.30 (d, J = 2.9 Hz, 1H), 7.34 (d, J = 8.3 Hz, 1H); ¹⁹F NMR $(Me_2CO-d_6) \delta$ 94.1 (dqm, J = 18.8, 264.2 Hz, 1F), 94.7 (dqm, J =18.8, 264.2 Hz, 1F); ¹³C NMR (Me₂CO- d_6) δ 22.4 (t, J = 26.5Hz), 79.0, 95.0, 104.1, 104.2 (t, J = 4.1 Hz), 117.6 (t, J = 236.3 Hz), 151.1, 155.9 (t, J = 32.5 Hz), 163.5; EI-MS m/z 261 (MH⁺, 35), 242, 231, 199, 167, 148, 113 (100); HRMS (C₁₀H₁₁F₂N₂O₄) calcd 261.0687, found 261.0694. Anal. Calcd for $C_{10}H_{10}F_2N_2O_4$: C, 46.16; H, 3.87; N, 10.77. Found: C, 46.31; H, 3.96; N, 10.60.

1-[2-Deoxy-(S)-4,4-*C***-(1,1-difluoroethane-1,2-diyl)**-β-**D**-*glycero***tetrofuranosyl]cytosine (23).** Treatment of **19a** (1.10 g, 2.54 mmol) by the sequence described for the conversion of **6** → **8**, followed by deprotection as described for **19a** → **21a**, gave an acetate salt after chromatography with Dowex 1 × 2 (OH⁻) resin [H₂O → MeOH → AcOH/MeOH (1:3)]. The salt was neutralized (Na₂CO₃/ H₂O), and the mixture was extracted (EtOAc). The combined organic phase was concentrated and subjected to PTLC (EtOAc/ MeOH, 10:1) to give **23** (0.56 g, 74%). Recrystallization (MeOH) gave an analytical sample as colorless needles: mp 217–218 °C dec; UV_{max} 271 nm (*ϵ* 8900), min 255 nm (*ϵ* 8000); ¹H NMR (CD₃-OD) δ 1.65–1.89 (m, 2H), 2.38–2.48 (dt, *J* = 6.5, 14.2 Hz, 1H), 2.57–2.65 (ddd, J = 2.2, 6.3, 14.2 Hz, 1H), 4.56 (dd, J = 2.2, 6.1 Hz, 1H), 4.87 (s, 3H), 5.91 (d, J = 7.6 Hz, 1H), 6.45 (t, J = 6.6 Hz, 1H), 7.56 (d, J = 7.6 Hz, 1H); ¹⁹F NMR (CD₃OD) δ 135.0 (ddd, J = 5.5, 16.5, 163.0 Hz, 1F), 142.2 (ddd, J = 7.3, 16.5, 163.0 Hz, 1F), 142.2 (ddd, J = 7.3, 16.5, 163.0 Hz, 1F); ¹³C NMR (CD₃OD) δ 18.8 (t, J = 10.3 Hz), 42.7, 70.0, 73.8 (dd, J = 7.6, 10.7 Hz), 89.0, 96.6, 112.3 (dd, J = 292.6, 299.5 Hz), 141.9 (d, J = 2.3 Hz), 158.3, 167.8; EI-MS m/z 259 (M⁺, 3), 230, 222, 207, 179, 138, 128 (100); HRMS (C₁₀H₁₁F₂N₃O₃) calcd 259.0768, found 259.0759. Anal. Calcd for C₁₀H₁₁F₂N₃O₃: C, 46.34; H, 4.27; N, 16.21. Found: C, 46.29; H, 4.31; N, 16.12.

1-[3-Deoxy-4-(1,1-difluoroethyl)-β-D-glycero-tetr-3-enofuranosyl]cytosine (24). Treatment of 19b (0.37 g, 1.0 mmol) by the sequence described for conversion of $6 \rightarrow 8$, followed by deprotection as described for $19a \rightarrow 21a$, gave a residue after chromatography with Dowex 1 \times 2 (OH⁻) resin (H₂O \rightarrow MeOH). Extraction (EtOAc) followed by PLTC (EtOAc/MeOH, 10:1) of the concentrated extract gave a product that was recrystallized (EtOH/Et₂O) to give 24 (0.20 g, 76%) as colorless plates: mp 197-198 °C; UV_{max} 270 nm (*\epsilon* 8700), min 259 (*\epsilon* 8400); ¹H NMR (CD₃-OD) δ 1.83 (t, J = 18.8 Hz, 3H), 4.88 (s, 3H), 4.95–4.98 (m, 1H), 5.57-5.60 (m, 1H), 5.91 (d, J = 7.3 Hz, 1H), 6.34 (d, J = 2.9 Hz, 1H), 7.31 (d, J = 7.3 Hz, 1H); ¹⁹F NMR (CD₃OD) δ 93.8 (dq, J= 18.3, 265.5 Hz, 1F), 95.3 (dq, J = 18.3, 265.5 Hz, 1F); ¹³C NMR (CD₃OD) δ 22.5 (t, J = 26.7 Hz), 79.8, 96.5, 97.3, 104.0 (t, J = 3.8 Hz), 117.9 (t, J = 235.4 Hz), 141.8, 156.8 (dd, J = 31.5, 34.1 Hz), 157.8, 167.8; EI-MS m/z 259 (M⁺, 2), 241, 230, 166, 123, 112 (100); HRMS (C₁₀H₁₁F₂N₃O₃) calcd 259.0768, found 259.0780. Anal. Calcd for C10H11F2N3O3: C, 46.34; H, 4.27; N, 16.21. Found: C, 46.12; H, 4.48; N, 16.01.

1-[2-Deoxy-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)-β-D-glycerotetrofuranosyl]adenine (22a). TBSCl (0.76 g, 5.02 mmol) was added to a stirred solution of 14 (0.50 g, 1.67 mmol) in dried pyridine (5 mL), and stirring was continued at ambient temperature for 1 day. Volatiles were evaporated in vacuo, and the residue was chromatographed (EtOAc/hexanes, $1:2 \rightarrow 1:1$) to give 1-[2-O-tertbutyldimethylsilyl-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)- β -D-erythrofuranosylladenine (16) (0.54 g, 78%). This material (0.45 g, 1.09 mmol) and 1,1'-thiocarbonyldiimidazole (0.29 g, 1.63 mmol) were dissolved in dried toluene/THF (10:4, 14 mL) and heated at 65 °C for 5 h. Volatiles were evaporated in vacuo, and the residue was chromatographed (EtOAc/hexanes, $1:1 \rightarrow EtOAc$) to give 18a/18b (1:2; 0.44 g, 77%). This material and AIBN (34 mg, 0.21 mmol) were dissolved in dried toluene (15 mL), and the solution was deoxygenated (N2). Bu3SnH (0.68 mL, 0.73 g, 2.5 mmol) was added, and the solution was heated at 100 °C for 1 h. Volatiles were evaporated, and the residue was chromatographed (EtOAc/ hexanes, $1:1 \rightarrow \text{EtOAc}$) to give 1-[3-O-tert-butyldimethylsilyl-2deoxy-(S)-4,4-C-(1,1-difluoroethane-1,2-diyl)- β -D-glycero-tetrofuranosyl]adenine (20a) (0.10 g, 31%): ¹⁹F NMR (CDCl₃) δ 134.6 (dm, J = 164 Hz, 1F), 142.2 (dm, J = 164 Hz, 1F); EI-MS m/z397 ([M⁺], 17), 340, 320, 304, 262, 242 (100), 205, 162, 135, 115; HRMS (C₁₇H₂₅F₂N₅O₂Si) calcd 397.1745, found 397.1739; and $1-[2-O-tert-butyldimethylsilyl-3-deoxy-4-(1,1-difluoroethyl)-\beta-D$ glycero-tetr-3-enofuranosyl]adenine (20b) (0.22 g, 65%): UV_{max} 259 nm (ϵ 15 100), min 230 nm (ϵ 3700); ¹H NMR (CDCl₃) δ 0.04 (s, 3H), 0.05 (s, 3H), 0.86 (s, 9H), 1.80 (t, J = 18.5 Hz, 3H), 5.37–5.40 (m, 1H), 5.58 (dd, J = 2.0, 2.9 Hz, 1H), 6.44 (d, J = 2.9 Hz, 1H), 6.62 (br s, 2H), 7.84 (s, 1H), 8.37 (s, 1H); ¹⁹F NMR (CDCl₃) δ 94.1 (dq, J = 18.3, 265.5 Hz, 1F), 96.6 (dq, J = 18.3, 265.5 Hz, 1F); ¹³C NMR (CDCl₃) δ –5.0, –4.9, 17.7, 21.8 (t, J = 26.8 Hz), 25.3, 79.5, 92.4, 102.1, 116.1 (t, J = 236.4 Hz), 119.3, 137.3, 149.2, 153.1, 155.1 (t, J = 33.6 Hz), 156.2; FAB-MS (thioglycerol) m/z 420 ([M⁺ + Na], 75), 398 (100), 382, 340, 266; HRMS (C₁₇H₂₅F₂N₅O₂SiNa) calcd 420.1643, found 420.1647.

Treatment of 20a (0.29 g, 0.73 mmol) with Bu₄NF/THF, as described for $19a \rightarrow 21a$, and chromatography with Dowex 1×2 (OH^{-}) resin $(H_2O \rightarrow MeOH)$ gave a residue that was purified by PTLC (EtOAc/MeOH, 19:1). This material (0.18 g, 87%) was recrystallized (EtOH) to give 22a as tiny colorless needles: mp 208–209 °C; UV_{max} 259 nm (ϵ 14 900), min 226 nm (ϵ 2200); ¹H NMR (CD₃OD) δ 1.65–1.73 (ddd, J = 4.9, 10.3, 16.3 Hz, 1H), 1.79-1.88 (ddd, J = 7.3, 10.3, 16.1 Hz, 1H), 2.70-2.77 (ddd, J =2.9, 7.3, 14.2 Hz, 1H), 3.24-3.31 (ddd, J = 5.4, 6.8, 14.2 Hz, 1H), 4.77 (dd, J = 2.9, 6.8 Hz, 1H), 6.63 (dd, J = 5.4, 7.3 Hz, 1H), 8.20 (s, 1H), 8.21 (s, 1H); ¹⁹F NMR (CD₃OD) δ 135.1 (ddd, *J* = 5.2, 15.7, 162.2 Hz, 1F), 142.1 (ddd, *J* = 6.9, 15.7, 162.3 Hz, 1F); ¹³C NMR (CD₃OD) δ 17.5 (t, J = 10.7 Hz), 41.9, 69.9, 73.5 (dd, J = 8.0, 10.3 Hz), 86.1, 112.2 (dd, J = 293.4, 297.9 Hz),120.6, 140.9, 150.7, 154.1, 157.4; EI-MS m/z 283 ([M⁺], 10), 265, 190, 163, 162, 136, 135 (100%), 108; HRMS (C₁₁H₁₁F₂N₅O₂) calcd 283.0880, found 283.0888. Anal. Calcd for C11H11F2N5O2: C, 46.65; H, 3.91; N, 24.73. Found: C, 46.72; H, 3.89; N, 24.70.

1-[3-Deoxy-4-(1,1-difluoroethyl)- β -D-glycero-tetr-3-enofuranosyl]adenine (22b). Deprotection of 20b (0.57 g, 1.44 mmol) (as described for $19b \rightarrow 21b$) and chromatography with Dowex 1×2 (OH^{-}) resin $[(H_2O \rightarrow MeOH \rightarrow AcOH/MeOH (1:3)]$ gave an acetate salt. The salt was neutralized (Na₂CO₃/H₂O), and the mixture was extracted (EtOAc). PTLC (EtOAc/MeOH, 10:1) of the concentrated organic phase gave material (0.39 g, 96%) that was recrystallized (EtOH) to give 22b as colorless cubes: mp 185-186 °C; UV_{max} 259 nm (*e* 15 300), min 230 nm (*e* 3800); ¹H NMR $(\text{Me}_2\text{CO-}d_6) \delta$ 1.80 (t, J = 18.8 Hz, 1H), 5.29 (br s, 1H), 5.56-5.60 (m, 1H), 5.76–5.80 (m, 1H), 6.53 (d, J = 2.9 Hz, 1H), 6.84 (br s, 2H), 8.12 (s, 1H), 8.23 (s, 1H); 19 F NMR (Me₂CO-d₆) δ 94.5 (qm, J = 18.9 Hz, 2F); ¹³C NMR (Me₂CO- d_6) δ 22.5 (t, J = 26.5Hz), 78.9, 93.4, 104.1 (t, J = 4.1 Hz), 117.7 (t, J = 235.3 Hz), 120.3, 139.4, 150.6, 154.3, 155.4 (t, J = 35.7 Hz), 157.3; EI-MS m/z 283 ([M⁺], 17), 266, 218, 190, 136 (100); HRMS (C₁₁H₁₁F₂N₅O₂) calcd 283.0880, found 283.0876. Anal. Calcd for C₁₁H₁₁F₂N₅O₂: C, 46.65; H, 3.91; N, 24.73. Found: C, 46.80; H, 3.66; N, 24.91.

Acknowledgment. We gratefully acknowledge NIH Grant No. GM029332, pharmaceutical company unrestricted gift funds (M.J.R.), and Brigham Young University for financial support. We thank Professor John F. Cannon for X-ray crystallographic analyses.

Supporting Information Available: Experimental details, NMR spectra of selected intermediates, and X-ray crystal structures and CIF data for compounds **3a**, **14**•pyridine, and **23**. This material is available free of charge via the Internet at http://pubs.acs.org.

JO061606U