

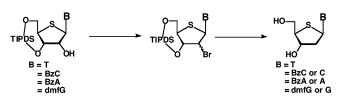
Practical Synthesis of 2'-Deoxy-4'-thioribonucleosides: Substrates for the Synthesis of 4'-ThioDNA

Naonori Inoue, Daisuke Kaga, Noriaki Minakawa,* and Akira Matsuda*

> Graduate School of Pharmaceutical Sciences, Hokkaido University, Kita-12, Nishi-6, Kita-ku, Sapporo 060-0812, Japan

> > matuda@pharm.hokudai.ac.jp; noriaki@pharm.hokudai.ac.jp

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We report herein a practical synthesis of 4'-thiothymidine (15) and appropriately protected 2'-deoxy-4'-thiocytidine (16), -thioadenosine (27), and -thioguanosine (29) derivatives, substrates for the synthesis of 4'-thioDNA, from the corresponding 4'-thioribonucleosides. 2'-Deoxy-4'-thiopyrimidine nucleosides were synthesized using a radical reaction of the corresponding 2'- α -bromo derivatives, which were prepared via 2,2'-O-anhydro derivatives. 2'-Deoxy-4'-thiopurine nucleosides were synthesized using the same radical reaction of the corresponding 2'- β -bromo derivatives.

We have recently been studying the synthesis of a series of 4'-thioribonucleosides with the aim of developing a nucleoside antimetabolite as well as a functional RNA molecule.^{1,2} In our preceding papers, we reported the stereoselective synthesis, via the Pummerer reaction,³ of 4'-thioribonucleosides, which we have since incorporated into oligoribonucleotides by chemical and enzymatic approaches to give 4'-thioRNA.² Because the 4'-thioRNA showed high nuclease resistance and hybridization properties,^{2a} we thought that the RNA molecule would be a promising candidate for functional RNAs such as antisense, ribozyme, RNA aptamer,2b and short interfering RNA.2c

Unlike the 4'-thioRNAs, the properties of oligodeoxyribonucleotides containing 2'-deoxy-4'-thioribonucleosides (4'-thioDNA) have not been well elucidated. Walker and co-workers reported the synthesis of 4'-thioDNAs consisting of 2'-deoxy-4'-thiopyrimidine nucleoside units, and their preliminary properties, including nuclease resistance and hybridization ability.⁴ However since then, nothing has been published on the use of 4'-thioDNA as a functional DNA molecule despite the favorable properties of 4'-thioDNA. This is probably due in large part to the difficulty of synthesizing 2'-deoxy-4'-thioribonucleosides.^{5,6} Haraguchi et al. have recently reported the stereoselective synthesis of 2'-deoxy-4'-thioribonucleosides based on electrophilic glycosidation of 4-thiofuranoid glycals.⁷ Although this method involves interesting chemistry, the reactions were all carried out on a very small scale.

In view of the above background information, we decided to develop a practical synthesis of 2'-deoxy-4'thioribonucleoside derivatives, which are substrates for the synthesis of 4'-thioDNA. Since our synthetic method using the Pummerer reaction can now be carried out on a large scale, ^{1a,8} deoxygenation of the 2'-hydroxyl groups of the resulting 4'-thioribonucleoside derivatives seemed to be the most straightforward and promising method for our purpose.

Two groups have independently reported the synthesis of 2'-deoxy-4'-thioribonucleosides from the corresponding 4'-thioribonucleosides using the radical deoxygenation. Jeong et al. reported the synthesis of the 2'-deoxy-4'thiouridine derivative in a brief communication.⁹ In their paper, the 2'-deoxy derivative was obtained in 79% yield by treatment of the corresponding 4'-thioribo derivative with phenyl chlorothionoformate [PhOC(S)Cl], followed by tributyltin hydride (Bu₃SnH) and triethylborane (Et₃B).¹⁰ In contrast, the reaction with the 2-chloro-4'thioadenosine derivative under the following conditions [N,N'-thiocarbonyldiimidazole, and then Bu₃SnH, 2,2'azobisisobutyronitrile (AIBN), toluene, reflux] afforded the 2'-deoxy congener in poor yield.¹¹ With these results in mind, we first examined the radical deoxygenation of

^{*} To whom correspondence should be addressed. (A.M.) Phone: +81-11-706-3228. Fax: +81-11-706-4980. (N.M.) Phone: +81-11-706-3230. Fax: +81-11-706-4980.

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⁽⁵⁾ A preferable β -anomer synthesis of the 2'-deoxy-4'-thioribonucleoside derivative has been achieved by glycosidation of 3-Ocarbamoyl-2-deoxythiosugar and 5-ethyluracil with assistance of the 3-O-carbamoyl group; see: Shaw-Ponter, S.; Mills, G.; Robertson, M.; Bostwick, R. D.; Hardy, G. W.; Young, R. J. Tetrahedron Lett. 1996, 37, 1867-1870.

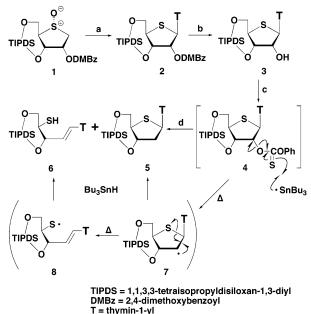
⁽⁶⁾ For examples: (a) Secrist, J. A.; Tiwari, K. N.; Riordan, J. M.; Montgomery, J. A. J. Med. Chem. **1991**, 34, 2361–2366. (b) Dyson, M. R.; Coe, P. L.; Walker, R. T. J. Med. Chem. 1991, 34, 2782-2786.

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⁽¹⁰⁾ The radical reaction was presumably conducted at room temperature because Et₃B was used as a radical initiator.

SCHEME 1^a



^a Key: (a) thymine, TMSOTf, Et₃N, CH₂Cl₂-toluene; (b) MeNH₂ in MeOH; (c) PhOC(S)Cl, DMAP, CH₃CN; (d) Bu₃SnH, AIBN, toluene, 80 °C.

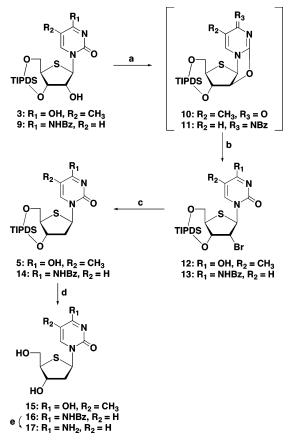
the 2'-hydroxyl group of 1-[3,5-O-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-4-thio- β -D-ribofuranosyl]thymine (**3**) (Scheme 1).

When the Pummerer reaction of the sulfoxide 1 with silvlated thymine was carried out, the desired 4'-thioribosylthymine derivative 2 was stereoselectively obtained in 81% yield. Treatment of 2 with methylamine gave 3, a substrate for the radical deoxygenation. After conversion of **3** into the thionocarbonate [PhOC(S)Cl, (dimethylamino)pyridine (DMAP), CH₃CN, rt], the resulting 4, after a water workup, was subjected to the standard radical deoxygenation conditions (Bu₃SnH, AIBN, toluene, 80 °C) to furnish the 4'-thiothymidine derivative 5 in 48% yield along with 35% of the unexpected compound **6**. In previous papers,^{9,11} compounds such as **6** were not obtained.^{12,13} Treatment of 3 with N,N'-thiocarbonyldiimidazole, followed by the radical deoxygenation under heating conditions also afforded the same results. In contrast, when the radical deoxygenation of 4 was conducted at room temperature using Et₃B as a radical initiator, 5 was obtained in only 10% yield after 24 h in our hands.^{9,10} In the above reaction, approximately 40% of 4 was recovered, while the ring opening product 6 was not observed. From these results, it might be postulated that formation of the radical intermediate 7 from the thionocarbonate 4 through homolytic cleavage of the C-O bond required higher temperature; however, this reaction is accompanied by homolytic cleavage of the C-S bond

(12) After the report of ref 11, Secrist et al. presented the structure of the byproduct obtained in the reaction described in ref 11. However, this type of byproduct was not observed in our substrate; see: Secrist, J. A.; Parker, W. B.; Tiwari, K. N.; Messini, L.; Shaddix, S. C.; Rose, L. M.; Bennett, L. L.; Montgomery, J. A. *Nucleosides Nucleotides* **1995**, *14*, 675–686.

(13) Quite recently, Dong and Paquette reported a similar reaction, Dong, S.; Paquette, L. A. J. Org. Chem. **2005**, 70, 1580–1596.

SCHEME 2^a



 a Key: (a) Tf_2O, DMAP, CH_2Cl_2; (b) LiBr, BF_3·Et_2O, 1,4-dioxane, 50 °C; (c) Bu_3SnH, V70-L, CH_2Cl_2; (d) TBAF, THF; (e) MeNH_2 in MeOH.

to give the radical intermediate 8. Consequently, the radical deoxygenation of 4 afforded 6 along with the desired 5.

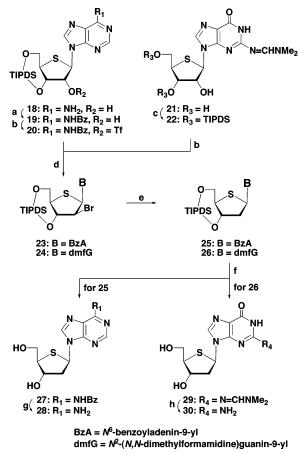
As an alternative method to prepare 2'-deoxy-4'-thioribonucleosides, we next envisioned the strategy of bromination at the 2'-position of 4'-thioribonucleoside derivatives, followed by radical reduction of the bromo group. As shown in Scheme 2, compound 3 was first converted into the 2,2'-O-anhydro derivative 10 by treatment with trifluoromethanesulfonic anhydride (Tf₂O) in CH₂Cl₂ in the presence of DMAP. The resulting 10 was then subjected to bromination conditions (LiBr, BF₃·Et₂O in dioxane)¹⁴ to give 12 in 75% yield over two steps. When 12 was treated with Bu₃SnH in CH₂Cl₂ at room temperature in the presence of racemic 2,2'-azobis(2,4-dimethyl-4-methoxyvaleronitrile) (V70-L), the desired 5 was obtained in 94% vield without formation of 6. Deprotection of the silvl group of 5 by tetrabutylammonium fluoride (TBAF) gave 4'-thiothymidine (15) in good yield. Likewise, compound 91a was converted into 2'-deoxy-4'thiocytidine derivative 16^{15} via the bromide 13. All reactions were conducted on a gram scale, and compound 15 and the N-benzoyl derivative 16 were finally obtained in more than 1 g. Analytical data of 15 and 17 prepared

⁽¹¹⁾ Tiwari, K. N.; Secrist, J. A.; Montgomery, J. A. Nucleosides Nucleotides 1994, 13, 1819–1828.

⁽¹⁴⁾ Aoyama, Y.; Sekine, T.; Iwamoto, Y.; Kawashima, E.; Ishido, Y. Nucleosides Nucleotides 1996, 15, 733–738.

⁽¹⁵⁾ Kumar, S.; Horton, J. R.; Jones, G. D.; Walker, R. T.; Roberts, R. J.; Cheng, X. Nucleic Acids Res. **1997**, 25, 2773–2783.

SCHEME 3^a



 a Key: (a) BzCl, pyridine, then NaOMe in MeOH; (b) Tf₂O, DMAP, CH₂Cl₂; (c) TIPDSCl₂, pyridine; (d) Bu₄NBr, benzene; (e) Bu₃SnH, V70-L, CH₂Cl₂; (f) TBAF, THF; (g) MeNH₂ in MeOH; (h) NH₃ in EtOH, 80 °C.

from 16 were identical with those for the authentic samples.⁶

For the synthesis of the 4'-thiopurine derivatives, the synthetic route is illustrated in Scheme 3. Starting with 18,^{2a} 19 was prepared by treatment with an excess amount of benzoyl chloride, followed by brief treatment with sodium methoxide. After introduction of a trifluoromethanesulfonyl group on the 2'-hydroxyl group of 19, bromination of **20** was examined. Among our attempts, treatment of **20** with Bu₄NBr in benzene at room temperature gave the best result, and 23 was obtained in 66% yield in two steps. Reduction of the bromo group proceeded smoothly under the radical conditions, and the 2'-deoxy-4'-thioadenosine derivative 27 was finally obtained in good yield. The 4'-thioguanosine derivative 29 was also prepared from 21 via protection, bromination, reduction, and deprotection steps. In these cases, all reactions were also conducted on gram scale to give 27 and 29 in sufficient quantities. The amino protective groups of 27 and 29 were deprotected to give 2'-deoxy-4'-thioadenosine (28) and 2'-deoxy-4'-thioguanosine (30), respectively. The analytical data of 28 and 30 were identical with the authentic data.¹⁶

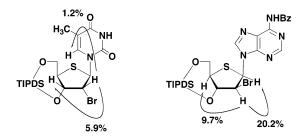


FIGURE 1. Structures of 12 and 23 confirmed by NOE experiments.

The configurations of the bromo groups of the pyrimidine (12 and 13) and purine (23 and 24) derivatives were confirmed by NOE experiments of 12 and 23 (Figure 1). Thus, NOEs were observed at H-6 (1.2%) and H-3' (5.9%) upon irradiation of H-2' of 12. Therefore, it was determined that 12 had an α -bromo group at the 2'-position. In contrast, NOEs were observed at H-1' (20.2%) and H-4' (9.7%) upon irradiation of H-2' of 23, which indicated that there is a β -bromo group at the 2'-position. Conversion of 3 into the 2'-bromo derivative 12 proceeded with overall retention of 2,2'-anhydro derivative 10. In the reaction with 18, the substitution proceeded via an S_N2type reaction to give 23 with inversion of configuration.

Thus far, a number of reactions have been reported for 4'-thionucleosides derivatives, and unexpected results arising from the participation of the sulfur atom were observed in some cases.^{13,17} In our reactions reported here, unexpected cleavage of the C–S bond in the sugar moiety to give **6** was observed during the radical deoxygenation on heating. This unfavorable C–S bond cleavage was avoided, however, when the radical reaction was carried out at room temperature. In contrast, participation of the sulfur atom would be negligible in the nucleophilic substitution at the C2'-position of 4'-thioribonucleoside derivative. These results agreed with those of the reactions of the O-congeners.

In conclusion, we have investigated the practical synthesis of the 2'-deoxy-4'-thioribonucleosides derivatives **15**, **16**, **27**, and **29** which would be easily converted into the corresponding phosphoramidite units for 4'thioDNA synthesis. The desired compounds were synthesized from 4'-thioribonucleoside derivatives via bromination, followed by radical reduction at room temperature. Since four kinds of 2'-deoxy derivatives are now available on a gram scale, investigations of the synthesis and properties of 4'-thioDNA are in progress.

Experimental Section

1-[2-α-Bromo-2-deoxy-3,5-O-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-4-thio-β-D-ribofuranosyl]thymine (12). To a solution of **3** (1.4 g, 2.6 mmol) in dry CH_2Cl_2 (26 mL) containing DMAP (1.3 g, 11 mmol) was added Tf_2O (0.88 mL, 5.2 mmol) at 0 °C, and the mixture was stirred at room temperature for 2 h. The reaction was quenched by addition of saturated aqueous NaHCO₃. The reaction mixture was partitioned between AcOEt

^{(16) (}a) Montgomery, J. A.; Secrist, J. A. 1991 PCT/US90/05252. (b) Draanen, N. A. V.; Freeman, G. A.; Short, S. A.; Harvey, R.; Jansen, R.; Szczech, G.; Koszalka, G. W. J. Med. Chem. **1996**, *39*, 538–542.

⁽¹⁷⁾ For examples: (a) Hancox, E. L.; Walker, R. T. Nucleosides Nucleotides 1996, 15, 135–148. (b) Otter, G. P.; Elzagheid, M. I.; Jones, G. D.; MacCulloch, A. C.; Walker, R. T.; Oivanen, M.; Klika, K. D. J. Chem. Soc., Perkin Trans. 2 1998, 2343–2349. (c) Miller, J. A.; Pugh, A. W.; Ullah, G. M. Nucleosides Nucleotides Nucleic Acids 2000, 19, 1475–1486.

and H₂O. The separated organic layer was washed with saturated aqueous NaHCO₃ (three times), followed by brine. The organic layer was dried (Na₂SO₄) and concentrated in vacuo to give crude 10. The resulting 10 was dissolved in dry 1,4-dioxane (26 mL), then BF₃·Et₂O (47%, an Et₂O solution, 0.49 mL, 3.9 mmol) and LiBr (340 mg, 3.9 mmol) were added to the solution. The mixture was heated at 50 °C for 30 min. The reaction was quenched by addition of H₂O. The reaction mixture was partitioned between AcOEt and H₂O. The separated organic layer was washed with saturated aqueous NaHCO₃, followed by brine. The organic layer was dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by a silica gel column, eluted with hexane/AcOEt (9:1-5:1), to give 12 (1.1 g, 75%, as a white solid): ¹H NMR (CDCl₃) δ 9.11 (s, 1 H), 8.09 (s, 1 H), 6.04 (s, 1 H), 4.37 (d, 1 H, J = 4.8 Hz), 4.13 (m, 2 H), 4.03 (d, 1 H, J =12.9 Hz), 3.72 (d, 1 H, J = 9.1 Hz), 1.93 (s, 3 H), 0.96–1.12 (m, 28 H); ¹³C NMR (CDCl₃) δ 163.6, 150.6, 136.3, 110.6, 71.2, 65.9, 58.9, 58.0, 51.7, 17.5, 17.4, 17.3, 17.1, 16.9, 16.8, 13.3, 13.1, 12.6, 12.5; FAB-LRMS m/z 581 (MH+); FAB-HRMS calcd for C₂₂H₄₀⁷⁹BrN₂O₅SSi₂ (MH⁺) 579.1380, found 579.1387.

1-[2-Deoxy-3,5-O-(1,1,3,3-tetraisopropyldisiloxane-1,3diyl)-4-thio- β -D-riofuranosyl]thymine (5). To a solution of 12 (1.1 g, 1.9 mmol) in dry CH_2Cl_2 (9.5 mL) containing 2,2'-azobis-(2,4-dimethyl-4-methoxyvaleronitrile) (racemic form) (120 mg, 0.39 mmol) was added tributyltin hydride (0.80 mL, 2.5 mmol), and the mixture was stirred at room temperature for 15 min. The mixture was concentrated in vacuo, and the residue was purified by silica gel column, eluted with hexane/AcOEt (9:1-1:1), to give 5 (1.0 g, 94%, as a yellow form): ¹H NMR (CDCl₃) δ 9.78 (s, 1 H), 7.89 (s, 1 H), 6.07 (d, 1 H, J=7.6 Hz), 4.45 (m, 1 H), 4.13 (dd, 1 H, J = 3.1 and 12.9 Hz), 3.95 (d, 1 H, J = 12.9Hz), 3.32 (m, 1 H), 2.48 (m, 1 H), 2.25 (m, 1 H), 1.92 (s, 3 H), 0.88-1.15 (m, 28 H); ¹³C NMR (CDCl₃) δ 163.8, 150.8, 136.5, 110.7, 70.9, 58.0, 57.1, 54.8, 43.0, 17.5, 17.3, 17.1, 17.0, 16.9, 13.4, 13.2, 12.9, 12.6, 12.4; FAB-LRMS m/z 501 (MH⁺); FAB-HRMS calcd for $C_{22}H_{41}N_2O_5SSi_2$ (MH⁺) 501.2275, found 501.2265.

 N^{6} -Benzoyl-9-[2 β -bromo-2-deoxy-3,5-O-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-4-thio- β -D-ribofuranosyl]adenine (23). To a solution of 19 (1.4 g, 2.2 mmol) in dry CH₂Cl₂ (22 mL) containing DMAP (1.1 g, 8.8 mmol) was added Tf₂O (0.73 mL, 4.3 mmol) at 0 °C, and the mixture was stirred at room temperature for 30 min. The reaction was quenched by addition of saturated aqueous NaHCO₃. The reaction mixture was partitioned between AcOEt and H₂O. The separated organic layer was washed with saturated aqueous NaHCO₃ (three times), followed by brine. The organic layer was dried (Na₂SO₄) and concentrated in vacuo. The resulting crude 20 was coevaporated with toluene and then dissolved in dry benzene (37 mL). To this solution, tetrabutylammonium bromide (1.4 g, 4.3 mmol) was added, and the reaction mixture was stirred at room

temperature for 1 h. The reaction was quenched by addition of H₂O. The reaction mixture was partitioned between AcOEt and H₂O. The separated organic layer was washed with saturated NaHCO₃, followed by brine. The organic layer was dried (Na₂-SO₄) and concentrated in vacuo. The residue was purified by silica gel column, eluted with hexane/AcOEt (3:1-1:1), to give **23** (1.0 g, 66%, as a yellow form): ¹H NMR (CDCl₃) δ 9.02 (s, 1 H), 8.84 (s, 1 H), 8.46 (s, 1 H), 8.03 (m, 1 H), 7.63 (m, 1 H), 7.55 (m, 2 H), 6.24 (d, 1 H, J = 6.6 Hz), 4.82 (dd, 1 H, J = 10.7 and 9.4 Hz), 4.41 (dd, 1 H, J = 6.6 and 10.7 Hz), 4.25 (dd, 1 H, J = 2.9 and 11.1 Hz), 4.17 (dd, 1 H, J = 2.8 and 11.1 Hz), 3.48 (ddd, 1 H, J = 9.4, 2.9 and 2.8 Hz), 1.05–1.22 (m, 28 H); ¹³C NMR $({\rm CDCl_3}) \; \delta \; 164.4, \, 152.7, \, 152.6, \, 152.3, \, 149.6, \, 142.4, \, 133.7, \, 132.7, \,$ 128.8, 127.7, 122.6, 60.0, 57.2, 55.4, 50.4, 17.5, 17.4, 17.3, 17.2, 17.1, 13.8, 13.2, 12.6; FAB-LRMS m/z 694 (MH⁺); FAB-HRMS calcd for $C_{29}H_{42}$ ⁷⁹BrN₅O₄SSi₂ (MH⁺) 692.1758, found 692.1761.

N⁶-Benzoyl-9-[2-deoxy-3,5-O-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-4-thio- β -D-ribofuranosyl]adenine (25). In the similar manner as described for 5, 23 (2.6 g, 3.8 mmol) in dry CH₂Cl₂ (20 mL) containing 2,2'-azobis(2,4-dimethyl-4-methoxyvaleronitrile) (racemic form) (0.23 g, 0.76 mmol) was treated with tributyltin hydride (1.5 mL, 5.6 mmol) to give 25 (2.0 g, 84%, as a white form): ¹H NMR (CDCl₃) δ 9.01 (s, 1 H), 8.81 (s, 1 H), 8.54 (s, 1 H), 8.02 (m, 1 H), 7.62 (m, 1 H), 7.53 (m, 2 H), 6.12 (d, 1 H, J = 6.5 Hz), 4.71 (m, 1 H), 4.17 (dd, 1 H, J = 2.9and 12.6 Hz), 4.02 (dd, 1 H, J = 2.6 and 12.6 Hz), 3.47 (ddd, 1 H, J = 8.8, 2.6 and 2.9 Hz), 2.57–2.69 (m, 2 H), 0.91–1.15 (m, 28 H); $^{13}\mathrm{C}$ NMR (CDCl_3) δ 164.4, 152.5, 151.3, 149.4, 142.1, 133.6, 132.7, 128.8, 127.7, 123.4, 72.0, 58.9, 55.2, 54.6, 43.6, 17.5,17.4, 17.3, 17.2, 17.1, 17.0, 13.4, 13.0, 12.6, 12.5; FAB-LRMS m/z 614 (MH⁺); FAB-HRMS calcd for C₂₉H₄₃N₅O₄SSi₂ (MH⁺) 614.2653, found 614.2682.

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Supporting Information Available: Experimental procedures and spectral data for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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