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A stereospecific synthesis of L-ribose and L-ribosides from D-galactose

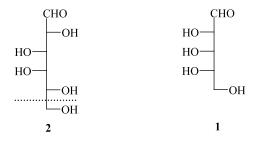
Zhen-Dan Shi, Bing-Hui Yang* and Yu-Lin Wu

State Key Laboratory of Bio-organic and Natural Products Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai 200032, China

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Abstract—An inexpensive D-galactose was converted into L-ribose and its derivatives via mild reaction conditions. The L-ribosyl donor was submitted to a glycosidation according to Vorbrüggen's conditions to give L-ribosides in high yields. © 2001 Elsevier Science Ltd. All rights reserved.

Recently, the use of L-carbohydrates and their corresponding nucleosides in medicinal applications has greatly increased. In particular, several modified nucleosides derived from L-sugars, such as (-)-(2'R,5'S)-1-(2-hydroxymethyloxathiolan-5-yl)-cytosine (3TC),¹ L-thymidine (L-T),² L-3'-thiacytidine (L-3-TC),^{3,4} L-5fluoro-3'-thia-cytidine (L-FTC),⁵ L-2',3'-dideoxycytidine (L - ddC),⁶ L - 5 - fluoro - 2',3' - dideoxy - cytidine (L - 5-FddC),^{7,8} and L-2'-fluoro-5-methylarabinofuranosyl uracil (L-FMAU),9 have shown great potential as useful antiviral agents. In addition, due to the stereoselectivity of enzymes, L-ribose modified oligoribonucleotides become attractive candidates for diagnostic and therapeutic uses because L-RNA ligands remain uncleaved in biological fluids.¹⁰ For these reasons, L-carbohydrates, modified L-nucleosides, especially L-ribose and its derivatives are of interest. Up to now, several syntheses of L-ribose and L-ribosides from L-arabinose,^{11–13} D-glucose,¹⁴ D-ribose,¹⁵ L-xylose¹⁶ have been reported and herein we report a stereospecific synthesis of L-ribose 1 and L-ribosides from D-galactose 2.



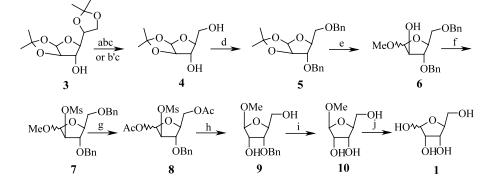
^{*} Corresponding author.

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According to our observations, there exists some useful information of D-galactose 2 in relation to L-ribose 1, i.e. D-galactose is a hexose while L-ribose is a pentose without C-6 and with configurations at both C-3 and C-4 the same, while C-2 is different. Therefore, the key conversion of D-galactose into L-ribose in our synthetic approach includes oxidation cleavage and reduction at 5,6-diol of galactose and the configuration inversion at 2-hydroxy group of resulting arabinose. The synthetic route to L-ribose is depicted in Scheme 1.

At first, we obtained compound **3**, 1,2,5,6-di-O-isopropylidene-D-galactofuranose from D-galactose as described in the literature.¹⁷ Chemoselective cleavage of the 5,6-O-isopropylidene diol of **3** with NaIO₄/HIO₄ (1.0 eq./0.5 eq.)-ether in one operation or with 10% AcOH followed by NaIO₄ cleavage of the resulting glycol and then reduction of the aldehyde with sodium borohydride in one-pot furnished L-arabinose derivative **4** in 85–92% yield. After some conversions including the protection of 3,5-dihydroxyl groups with benzyl chloride, methanolysis, and methanesulfonylation, a substrate (7) for configuration inversion was obtained.

Configuration inversion of the 2-hydroxyl group in compound 7 was attempted by several methods including Mitsunobu method and oxidation/reduction procedure, but all of these were unsuccessful. We therefore reacted 7 with Ac₂O, AcOH and H₂SO₄, this gave only the 1,5-diacetate **8**, but not the 1-acetate, in 89% yield. The inversion of 2-OH configuration and the hydrolysis of the diacetate took place with NaOMe/MeOH for the intramolecular $S_N 2$ reaction of methanesulfonyloxy group with C-1 alkoxide. Then, L-ribose was prepared



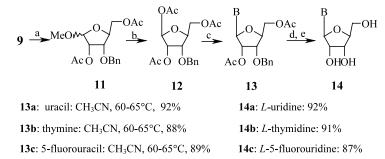
Scheme 1. Reagents and conditions: (a) 10% AcOH-H₂O, rt, 24 h; (b) NaIO₄, MeOH, H₂O, rt, 1h; (c) NaBH₄, MeOH, H₂O, 3 h, three steps in 92% yield; (b') NaIO₄/HIO₄ (1.5 eq.), Et₂O, rt, 4h; (d) KOH, BnCl, 1,4-dioxane, reflux, 2 h, 95%; (e) 10% HCl-MeOH, rt, 3h, 96%; (f) MsCl, Et₃N, rt, overnight, 98%; (g) Ac₂O, AcOH, H₂SO₄, 4°C, overnight, 89%; (h) NaOMe, MeOH, rt, 6 h, 83%; (i) 10% Pd-C, MeOH, H₂, 2 h; (j) Dowex [H⁺], H₂O, 50°C, 24 h, two steps in 95% yield.

by debenzylation of **9** with 10% palladium-carbon in methanol followed by hydrolysis of methyl glycoside with ion-exchange resin (H⁺ form) in 95% yield. The resulting structure was confirmed by comparison with a commercial sample (Aldrich). In conclusion we have synthesized L-ribose from D-galactose, an inexpensive material, by using cheap reagents under mild reaction conditions.

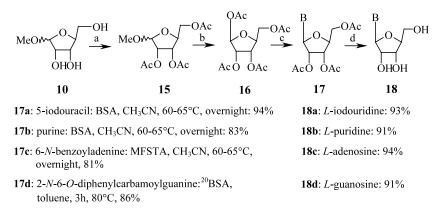
Scheme 2 shows the syntheses of L-ribosides 14. Diacetate 11 was prepared by the treatment of 9 with acetic anhydride and pyridine and converted to 1,2,5-tri-*O*-acetyl-3-*O*-benzyl-L-ribofuranoside 12 as an isolable mixture ($\alpha:\beta=1:11$). According to the Vorbrüggen

method,¹⁸ the β -*N*-glycosidic bond linkages are made by the L-ribosyl donor **12** and the protected bases. We therefore obtained **13a**, **b**, **c** by treatment of **12** with the protected bases, respectively, in the presence of TMSOTf and BSA (*N*,*O*-bis-trimethylsilylacetamide) in good yields (92%, 88%, 89% for **13a**, **b**, **c**, respectively). The L-ribosides were obtained by deacetylation with NH₃-H₂O/MeOH and then debenzylation using 10% palladium-carbon in methanol in high yield (92%, 91%, 87% for **14a**, **b**, **c**, respectively).¹⁹

For some bases, sensitive to debenzylation, like purine, 5-iodouracil, N^6 -benzoyl-adenine, N^2 - O^6 -diphenylcarbamoylguanine, we resorted to another synthetic route



Scheme 2. *Reagents and conditions*: (a) Ac₂O, py, overnight, rt, 92%; (b) AcOH, Ac₂O, H₂SO₄, 4°C, overnight, 83%; (c) Base, TMSOTf, BSA, solvent, overnight; (d) NH₃-H₂O, MeOH, 60°C, overnight; (e) 10% Pd-C, H₂, rt, 3 h



Scheme 3. Reagents and conditions: (a) Ac_2O , py, overnight, rt, 90%; (b) Ac_2O , AcOH, H_2SO_4 , 4°C, overnight, 74%; (c) B, TMSOTf, solvents; (d) NH_3/H_2O , MeOH, 60°C, overnight.

(Scheme 3). Protection of the triol **10** with acetic anhydride and pyridine furnished the triacetate **15** and treatment of **15** with Ac₂O/AcOH/H₂SO₄ afforded a separable mixture **16** (α : β =1:7) of tetra-*O*-acetyl-Lribose. Using the same procedures for glycosidation, we acquired the L-ribosides (**17a**, **b**, **c**, **d**) in good yield (83–94%) on different bases, selecting BSA (*N*,*O*-bistrimethylsilylacetamide), MFSTA (*N*-methyl-*N*-trimethylsilyl-trifluoroacetamide) and different solvents (acetonitrile or toluene). The deprotected products **18a**, **b**, **c**, **d** were obtained in high yields (91–94%).

Thus we have synthesized L-ribose from the easily available 3 in ten steps and in 57% overall yield. These procedures provide a practical synthesis of L-ribose and its derivatives. The biological activity of the L-ribosides and their derivatives are being assessed.

Acknowledgements

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- 19. All new compounds gave satisfactory spectral and microanalytical data. Selected data for compound 9: $[\alpha]_{D}^{20} =$ +9.1(c 0.5, MeOH); IR (film, cm⁻¹): 3415, 3032, 1455; ¹H NMR (300 MHz, CDCl₃): δ 7.41–7.31 (m, 5H, Ph), 4.87 (s, 1H, H-1), 4.56 (s, 2H, OBn), 4.23-4.11 (m, 2H, H-3, H-4), 4.04 (d, 1H, J_{2,3}=4.2 Hz, H-2), 3.77 (dd, 1H, $J_{4,5a} = 2.2$ Hz, $J_{5a,5b} = 12.9$ Hz, H-5a), 3.55 (dd, 1H, $J_{4,5b} = 3.3$ Hz, $J_{5a,5b} = 12.0$ Hz, H-5b), 3.40 (s, 3H, OMe); HRMS (m/z) calcd. for C₁₃H₁₈O₅: 254.1177; found: 254.1154. Compound 1: $[\alpha]_D^{20} = +19.2$ (c 2.0, H₂O); IR (KBr): v_{max} 3500 (brs) cm⁻¹; ¹H NMR (300 MHz, CD₃OD): δ 4.93 (d, 0.57H, $J_{1,2}$ = 5.0 Hz, β -anomer), 4.78 (d, 0.43H, $J_{1,2}=1.5$ Hz, α -anomer), 3.94–3.82 (m, 2H, H-2, H-3), 3.77-3.61 (m, 2H, H-4, H-5a), 3.48 (m, 1H, H-5b); MS (EI): m/z 151 (M⁺+1), 133 (M⁺-OH); Anal. calcd for C5H10O5.0.1H2O: C, 39.53 H, 6.72; found: C, 39.47, H, 6.79. Compound **13b**: $[\alpha]_{D}^{20} = -23.4$ (c 0.60, MeOH); IR (KBr): v_{max} 3197 (brs), 1748 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 8.59 (s, 1H, NH), 7.38-7.30 (m, 5H, Ph), 7.14 (s, 1H, H-6), 5.85 (d, 1H, $J_{1',2'} = 3.3$ Hz, H-1'), 5.37 (dd, 1H, $J_{1',2'}=3.3$ Hz, $J_{2',3'}=5.2$ Hz, H-2'), 4.62 (d, 1H, J=11.3 Hz, OBn), 4.46 (d, 1H, J=11.3 Hz, OBn), 4.31 (dd, 1H, J_{4'5'}=4.1 Hz, J_{5a',5b'}=13.1 Hz, H-5a'), 4.25–4.19 (m, 2H, H-3', H-4'), 4.15 (dd, 1H, $J_{4',5b'}$ = 6.3 Hz, $J_{5a',5b'} = 11.8$ Hz, H-5b'), 2.14 (s, 3H, OAc), 2.04 (s, 3H, OAc), 1.91 (s, 3H, CH₃); MS (EI): m/z 307 (M⁺+1-base); Anal. calcd for C₂₁H₂₄O₈N₂: C, 58.33, H, 5.56, N, 6.48; found: C, 58.22, H, 5.56, N, 6,32. Compound 14b: $[\alpha]_D^{20} = +15.6$ (c 1.1, MeOH); ¹H NMR (300 Hz, DMSO-d₆): δ 11.31 (s, 0.25H, NH), 7.74 (s, 1H, H-6), 5.78 (d, 1H, $J_{1',2'} = 5.6$ Hz, H-1'), 4.03 (t, 1H, $J_{1',2'} =$ 5.5 Hz, J _{2',3'}=5.3 Hz, H-2'), 3.96 (t, 1H, J_{2',3'}=4.9 Hz, $J_{3'4'} = 3.9$ Hz, H-3'), 3.80 (q, 1H, $J_{3',4'} = 3.9$ Hz, $J_{4',5a'} = 3.4$ Hz, $J_{4',5a'}$ = 3.4 Hz, $J_{4',5b'}$ = 3.4 Hz, H-4'), 3.63 (dd, 1H, $J_{4',5a'} = 3.4$ Hz, $J_{5a',5b'} = 12.1$ Hz, H-5a'), 3.53 (dd, 1H, $J_{4',5b'} = 3.4$ Hz, $J_{5a',5b'} = 12.1$ Hz, H-5b'), 1.77 (s, 3H, CH₃); MS (EI): m/z 258 (M⁺); Anal. calcd for C₁₀H₁₄O₆N₂·0.25H₂O: C, 45.71, H, 5.52, N, 10.67; found: C, 45.89, H, 5.44, N, 10.33.
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