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Structure-based design of novel quinoxaline-2-carboxylic acids and analogues as Pim-1 inhibitors

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2	analogues as Pim-1 inhibitors
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#### 34 Abstract

#### 35

36 We identified a new series of quinoxaline-2-carboxylic acid derivatives, targeting the 37 human proviral integration site for Moloney murine leukemia virus-1 (HsPim-1) 38 kinase. Seventeen analogues were synthesized providing useful insight into structure-39 activity relationships studied. Docking studies realized in the ATP pocket of HsPim-1 40 are consistent with an unclassical binding mode of these inhibitors. The lead 41 compound 1 was able to block *Hs*Pim-1 enzymatic activity at nanomolar 42 concentrations (IC<sub>50</sub> of 74 nM), with a good selectivity profile against a panel of 43 mammalian protein kinases. In vitro studies on the human chronic myeloid leukemia 44 cell line KU812 showed an antitumor activity at micromolar concentrations. As a 45 result, compound 1 represents a promising lead for the design of novel anticancer 46 targeted therapies.

47

#### 48 Keywords

49 quinoxaline; Pim-1; kinase inhibitor; anticancer targeted therapy.

- 50
- 51

# 52 Abbreviations<sup>1</sup>

<sup>&</sup>lt;sup>1</sup> IC<sub>50</sub>, 50% inhibitory concentration; SAR, structure-activity relationships; Pim, proviral integration site of Moloney murine leukemia virus; CML, chronic myeloid leukemia; DYRK1A, dual specificity tyrosine phosphorylation regulated kinase 1A; CDK, cyclin-dependent kinase; Haspin, haploid germ cell-specific nuclear protein kinase; CLK1, CDC2-like kinase 1; CK1, casein kinase 1; GSK3, glycogen synthase kinase 3.

#### 54 1. Introduction

#### 55

Proviral integration site for Moloney murine leukemia virus (Pim) kinases belong to a 56 57 small family of constitutively activated proto-oncogenic serine/threonine protein 58 kinases, constituted of three isoforms: Pim-1, Pim-2 and Pim-3 [1]. These 59 oncoproteins control many cellular functions like cell cycle regulation, apoptosis, cell survival, proliferation and differentiation [2,3], and are overexpressed in a large 60 61 number of human cancer types, such as hematopoietic malignancies [4,5] and solid 62 cancers (e. g. bladder [6], prostate [7], breast [8] or oral cancers [9]). These kinases 63 are positive regulators of cell cycle progression at G1/S and G2/M checkpoints, and 64 inhibit apoptosis, acting as oncogenic survival factors [10]. Interestingly, it has been demonstrated that  $PimI^{-/-}2^{-/-}3^{-/-}$  triple knockout mice were viable and fertile, which 65 make these kinases very interesting for targeted cancer therapies [11]. 66

Recently, Pim-1 has been shown to play a significant role in cancer stem cells growth, and in resistance to chemotherapy drugs, promoting multiple drug resistance [12,13]. This kinase is thus considered as a relevant target for cancer therapy and a large variety of small molecule inhibitors have been developed [14-18]. Many of these Pim-1 kinase inhibitors demonstrated significant *in vitro* activity in cancer cell lines and in different *in vivo* tumor xenograft models, and clinical trials are currently ongoing for the most promising candidates [14,18].

74 A remarkable characteristic of Pim-1 active site in comparison to other protein 75 kinases is the presence of an original hinge region (region containing backbone 76 peptide atoms that forms hydrogen bond interactions (H-bonds) with the adenine 77 moiety of ATP). Indeed, this region contains a proline residue (Pro123), which has no H-bond donor property and precludes the formation of one of the conserved H-bond 78 79 involving the hinge backbone and the ATP adenine ring, as it can be observed in other 80 kinases. Thus, Pim-1 bounds ATP via only one hinge H-bond between the ATP 81 adenine amino moiety and the backbone carbonyl of glutamate 121 (Glu121). 82 Moreover, the insertion in the hinge of a valine (Val126), absent in other kinases, 83 changes the hinge conformation, enlarging the catalytic pocket. This unique feature 84 can be exploited for the design of selective inhibitors [19].

The vast majority of Pim-1 inhibitors mainly act as ATP competitive inhibitors, targeting the ATP-binding pocket. They can be classified into two categories: ATPmimetics, which bind to the Glu121 residue of the hinge region, and non-ATP

mimetics, which interact with the ATP binding cleft in a different manner from ATP[20].

90 In a continuing effort to develop new small molecule inhibitors with anticancer 91 properties, our laboratory has been recently focusing on the study of new inhibitors of 92 the signal transducer and activator of transcription 5 (STAT5) activation and 93 expression and their interest in chronic myeloid leukemia (CML) [21]. Indeed, the 94 STAT family transcription factors are commonly activated in cancer by upstream 95 mutations or cell surface signaling molecules. It has been demonstrated that the Pim 96 kinases are induced by the STAT family transcription factors (particularly STAT 3/5) 97 [14]. Regarding the potential of Pim-1 as target in cancer therapy and particularly in 98 leukemia [22,23], we decided to further explore the STAT signaling pathway, by 99 developing new Pim-1 kinase specific inhibitors. In this purpose, we first performed a 100 target-based approach, by realizing a focused in vitro screening of our chemical library on a limited panel of kinases, comprising Homo sapiens Pim-1 (HsPim-1), 101 102 allowing the identification of the quinoxaline-2-carboxylic acid 1 as a new lead compound (Fig. 1). This molecule was able to inhibit the in vitro enzymatic activity of 103 104 HsPim-1 with an IC<sub>50</sub> of 74 nM.

Docking studies, using program GOLD (GOLD version 4.0; CCDC, Cambridge, UK), 105 were performed to understand the binding interactions between the lead compound 1 106 and the ATP pocket of HsPim-1 (PDB ID 3A99) (Fig. 2). Data analysis suggests that 107 the carboxylate group of this molecule can form a key salt bridge with the protonated 108 109 amino group side chain of catalytic Lys67, as it has already been described in other 110 Pim-1 inhibitors [24, 25], and shares also a H-bond interaction with the backbone NH of Asp186 belonging to the DFG motif. Additionally, an H-bond interaction between 111 112 the 3-hydroxyphenyl moiety and the carboxylate group of residue Asp186 can be observed. These studies suggested that compound 1 could act as an ATP competitive 113 114 inhibitor, with a non-ATP mimetic binding mode.

Sixteen new analogues were then synthesized, exploiting the unique sequence of*Hs*Pim-1 ATP-binding cleft.

117 We report herein the design, synthesis, structure-activity relationships (SAR) and in

118 *vitro* evaluations of this new class of Pim-1 inhibitors.

119

## 120 **2. Chemistry**

122 The preparation of quinoxaline-2-carboxylic acids **1**, **5c-e**, and **5h-i** and potassium 123 carboxylate salts **5b**, and **5g** was performed as shown in Scheme 1 by amination of the 124 intermediate ethyl 3-chloroquinoxaline-2-carboxylate **3** with the appropriate amine 125 derivatives.

The synthesis of ethyl 3-chloroquinoxaline-2-carboxylate **3** was achieved in two steps from commercial *o*-phenylenediamine according to literature procedures [26,27] (Scheme 1). First, the *o*-phenylenediamine was condensed with diethyl 2oxomalonate in the presence of citric acid ( $3 \mod \%$ ) at room temperature in ethanol to give ester **2**, which was then chlorinated using *N*,*N*-dimethylformamide (DMF) as a catalyst in refluxing phosphorous oxychloride, affording the intermediate **3** in quantitative yield.

Access to quinoxaline-2-carboxylic acids, and potassium carboxylate salts was then 133 134 performed using a two-step synthetic pathway. Thus, intermediate 3 undergoes initial 135 nucleophilic aromatic substitution with the appropriate amine in presence of p-TSA as 136 a catalyst in refluxing absolute ethanol to give esters 4a-f and 4h-i [28]. For compound 4f, a supplementary step of *tert*-butyloxycarbonyl (Boc) group 137 deprotection, using trifluoroacetic acid in dichloromethane (DCM), was necessary to 138 139 obtain the derivative 4g. Then, hydrolysis of the intermediate ethyl esters 4a-c, 4e, 140 and 4g-i with potassium carbonate in refluxing 80% aqueous methanol was 141 performed. The potassium salts 5b and 5g were thus obtained without any treatment. 142 A subsequent acidification with a citric acid aqueous solution was realized to afford 143 acids 1, 5c-e, and 5h-i.

Ester 4d was saponified using a 10% aqueous sodium hydroxide solution in refluxing
ethanol, leading, after acidification with a citric acid aqueous solution, to the
corresponding carboxylic acid 5d.

Finally, synthesis of the carboxamide **6** was achieved from acid **5c**, using *N*methylmorpholine and ethyl chloroformate in dichloromethane at 0  $^{\circ}$ C, followed by the addition of a 28% ammonium hydroxide solution.

- 150
- 151 **3. Results and discussion**

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153 3.1. Enzymatic Assays

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155 *3.1.1. Pim-1 enzymatic activity inhibition* 

156 Compounds were first evaluated for their efficacy to inhibit the *in vitro* enzymatic 157 activity of *Hs*Pim-1, using a luminescence-based kinase assay [29]. Compounds that 158 displayed *Hs*Pim-1 IC<sub>50</sub> > 10  $\mu$ M were considered inactive.

159 To get closer insight into the potential binding mode of our compounds within the 160 ATP binding pocket of *Hs*Pim-1, we first decided to structurally vary the substitution 161 patterns of the quinoxaline scaffold of **1** in position 3 (Table 1). Docking analysis 162 revealed that the hydroxyl moiety of the phenyl ring in this position could be able to 163 form an H-bond interaction with the carboxylate side chain of residue Asp186 of the *Hs*Pim-1 ATP binding pocket. Interestingly, it appears that position of the hydroxyl 164 group on the phenyl ring strongly modulates compound activity, since modification 165 from meta (1) to para (5e) or ortho (5i) position reduced significantly the inhibitory 166 167 potency (Table 1, entries 2, 11, and 17). Thus, compounds 5e and 5i maintained a submicromolar activity on HsPim-1 (IC<sub>50</sub> of 0.29  $\mu$ M and 0.76  $\mu$ M, respectively) but 168 169 were less potent (4-fold, 10-fold, respectively) than lead compound 1 (IC<sub>50</sub> of 74 nM). 170 Surprisingly, the replacement of the hydroxyl moiety of compound 1 by an amino 171 group, able to form an H-bond with Asp186, led to a significant loss of potency (5b,  $IC_{50}$  of 2.80  $\mu$ M, 38-fold lower). However, as expected, the substitution by a 172 morpholino group, suppressing the formation of an H-bond, was not favorable for the 173 activity, as shown by derivative 5c (IC<sub>50</sub> of  $1.01 \mu$ M). Again, the para substitution on 174 the phenyl ring was deleterious for the activity, as shown by derivative 5g which was 175 > 3.5-fold less active than its "meta" analogue **5b**, and by the inactive compound **5h** 176 177 (Table 1, entries 4, 13 and 15).

Finally, replacement of the 3-hydroxyphenyl moiety of compound 1 by an 1*H*-indol5-yl group (5d) led to a drastic loss of potency (Table 1, entry 9).

180 Docking analysis also suggested that the carboxylate function in position 2 of the quinoxaline scaffold of compound **1** was crucial for the activity, establishing, notably, 181 182 a key salt bridge with the catalytic Lys67. However, carboxylic acids are known to be 183 responsible for limited permeability across biological membranes, metabolic 184 instability, and potential adverse effects [30]. To circumvent these issues, and to 185 confirm the results of the modeling studies, we evaluated ethyl ester derivatives (4a-e and 4g-i) of all synthesized acids and carboxylate salts, and we replaced the acid 186 group of compound 5c by a carboxamide isosteric moiety (6). Both ester and 187 carboxamide functions are not able to form a salt bridge like the carboxylic acid 188 189 group, resulting in a complete loss of *Hs*Pim-1 enzymatic activity inhibition (Table 1,

entries 1, 3, 5, 7, 8, 10, 12, 14 and 16), highlighting the highly critical role of this typeof interaction in position 2 of the quinoxaline ring.

192

#### 193 *3.1.2. Selectivity over a panel of mammalian protein kinases*

194 A selectivity profile of the most active HsPim-1 inhibitors was performed. In that 195 purpose, most promising candidates were further evaluated in an expanded panel of 196 mammalian protein kinases such as RnDYRK1A, HsCDK2/CyclinA, 197 HsCDK9/CyclinT, HsHaspin, MmCLK1, SscCK1 $\delta$ / $\epsilon$  and SscGSK3 $\alpha$ / $\beta$ . Inhibition 198 values were determined using a luminescence-based kinase assay [29].

Similar inhibition trends were observed with 6 of the mammalian kinases tested (*Hs*CDK2/CyclinA, *Hs*CDK9/CyclinT, *Hs*Haspin, *Mm*CLK1, *Ssc*CK1\delta/ $\epsilon$  and *Ssc*GSK3 $\alpha/\beta$ ), with IC<sub>50</sub> > 10  $\mu$ M in every case, for each compound evaluated, suggesting an interesting selectivity profile against these potential off-target kinases (Table 2). Notably, we observed > 130-fold differences between IC<sub>50</sub> values for *Hs*Pim-1 over these mammalian kinases for our lead inhibitor **1**.

- In contrast, quinoxalines **1**, **5b**, **5e** and **5i** displayed a micromolar to submicromolar inhibition of RnDYRK1A (table 2, entries 1, 2, 4 and 5). Lead **1** exhibited nevertheless an IC<sub>50</sub> value at least 3.5-fold higher for RnDYRK1A than for *Hs*Pim-1. Interestingly, compound **5c**, despite a less potent activity profile against *Hs*Pim-1, was more than 10-fold selective with respect to this kinase (Table 2, entry 3).
- 210

211 3.2. In vitro cell-based assays

212 Most active HsPim-1 inhibitors were then tested in vitro on the human CML cell line

213 KU812, overexpressing Pim-1. Cytotoxic effects were evaluated using a MTT assay,

and living cells were also counted with the trypan blue dye exclusion method.

As expected, a same trend was observed between *Hs*Pim-1 enzymatic activity inhibition and *in vitro* cytotoxic potency. Indeed, quinoxalines with a good level of activity on *Hs*Pim1 (IC<sub>50</sub> of 0.074 to 2.80  $\mu$ M) also exhibited *in vitro* cytotoxic effects on KU812 cell line with EC<sub>50</sub> values ranging from 38.9 ± 3.4  $\mu$ M to 177.5 ± 13.1  $\mu$ M (Table 2). Moreover, the best *Hs*Pim-1 inhibitor **1** (IC<sub>50</sub> of 74 nM), was also the most cytotoxic compound (EC<sub>50</sub> of 38.9 ± 3.4  $\mu$ M).

#### **4. Conclusion**

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223 In this study, we identified a new series of quinoxaline-2-carboxylic acids and 224 analogues, exhibiting a potent activity against the HsPim-1 oncoprotein. Among the 17 225 compounds synthesized, 5 significantly blocked HsPim-1 with IC<sub>50</sub> values in the 226 submicromolar to low micromolar range. In particular, lead compound 1 showed the 227 best inhibitory effect against HsPim-1, with an  $IC_{50}$  value of 74 nM. SAR in positions 2 228 and 3 of the quinoxaline scaffold confirmed the molecular modeling studies, 229 highlighting the crucial role of the carboxylic acid function in position 2 for the HsPim-230 1 inhibitory activity of these compounds. In vitro studies of the 5 most potent inhibitors 231 on the human CML cell line KU812, confirmed their interest, with antitumor activities 232 at micromolar concentrations.

This series of compounds, and particularly lead **1**, could therefore represent new attractive drug candidates for extending further pharmacomodulation studies in a way to improve their potency and selectivity profile.

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#### 237 **5. Experimental section**

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239 5.1. General remarks

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241 All solvents were anhydrous reagents from commercial sources. Unless otherwise 242 noted, all chemicals and reagents were obtained commercially and used without 243 purification. Microwave heating was carried out with a single-mode Initiator Alstra 244 (Biotage) unit. Melting points (Mp) were determined on a Stuart capillary apparatus and 245 are uncorrected. High-resolution mass spectra (HRMS) were performed in positive 246 mode with an ESI source on a Q-TOF mass spectrometer (Bruker maXis) with an accuracy tolerance of 2 ppm. NMR spectra were recorded at 300 MHz (<sup>1</sup>H) or 75 MHz 247 (<sup>13</sup>C) on a Bruker Avance (300 MHz) spectrometer. The chemical shifts are reported in 248 249 parts per million (ppm,  $\delta$ ) relative to residual deuterated solvent peaks. The 250 abbreviations s = singlet, d = doublet, t = triplet, q = quadruplet, m = multiplet and bs = doublet251 broad signal were used throughout. Known compounds were prepared according to

- 252 literature procedures: *tert*-butyl (3-aminophenyl)carbamate, and *tert*-butyl (4253 aminophenyl)carbamate [31], 4-(1-methylpiperidin-4-yl)aniline [32].
- 254

255 5.2. Chemistry

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257 5.2.1. Ethyl 3-oxo-3,4-dihydroquinoxaline-2-carboxylate (2)

- A mixture of *o*-phenylenediamine (708 mg, 6.55 mmol), diethyl 2-oxomalonate (1.14 g, 6.55 mmol) and citric acid (41 mg, 0.20 mmol) in ethanol (13 mL) was stirred magnetically at room temperature for 10 min. Ethanol was then evaporated under reduced pressure, and the residue was stirred with crushed ice for 5 min, filtered and dried under vacuum to give compound 2 (1.13 g, 79%) as a beige solid.
- 263 Mp 168.8 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  12.85 (bs, 1H, NH), 7.97 (dd, 1H, J = 8.1, 264 1.2 Hz), 7.64 (ddd, 1H, J = 8.1, 7.2, 1.2 Hz), 7.48 (dd, 1H, J = 8.1, 1.2 Hz), 7.42 (ddd, 265 1H, J = 8.1, 7.2, 1.2 Hz), 4.56 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.49 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). 266 <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  163.4, 154.6, 148.4, 132.8, 132.2, 132.1, 130.2, 125.0, 267 116.5, 62.6, 14.2.
- 268

# 269 5.2.2. Ethyl 3-chloroquinoxaline-2-carboxylate (3)

Into a dry three-neck round bottom flask was introduced compound **2** (218 mg, 1.00 mmol) in phosphorous oxychloride (2 mL) at ice bath temperature. Dimethylformamide (0.1 mL) was then added at 0 °C and the reaction mixture was refluxed for 30 min. After cooling, the resulting mixture was diluted with ethyl acetate and washed with a 10% sodium hydroxide solution (2  $\times$  5 mL), and brine (2  $\times$  10 mL). The combined organic layers were dried over MgSO<sub>4</sub>, filtered, and evaporated under reduced pressure to obtain derivative **3** (237 mg, 100%) as a beige solid.

- 277 Mp 46.4 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 8.18 (m, 1H), 8.07 (m, 1H), 7.92-7.81 (m,
- 278 2H), 4.58 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.49 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz,
- $279 \quad CDCl_3) \, \delta \, 163.9, \, 144.7, \, 143.9, \, 142.2, \, 139.7, \, 132.6, \, 131.0, \, 129.6, \, 128.3, \, 63.0, \, 14.1.$
- 280

# 281 5.2.3. Ethyl 3-((3-hydroxyphenyl)amino)quinoxaline-2-carboxylate (4a)

- 282 Method A: a solution of compound **3** (1.11 g, 4.70 mmol), 3-aminophenol (622 mg,
- 283 5.70 mmol) and *p*-TSA, as a catalyst, in absolute ethanol (40 mL) was refluxed for 110

h. Ethanol was then evaporated under reduced pressure, and the resulting residue was purified by silica column chromatography using cyclohexane with ethyl acetate gradient (0-50%) as eluent to give the desired compound **4a** (1.0 g, 69%) as a red powder.

- 287 Mp 233.3 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6) δ 10.08 (bs, 1H, NH), 9.52 (bs, 1H, OH),
- 288 7.99 (dd, 1H, J = 8.4, 0.6 Hz), 7.85-7.75 (m, 2H), 7.61-7.54 (m, 2H), 7.24-7.14 (m, 2H),
- 289 6.51 (ddd, 1H, J = 7.4, 2.4, 1.5 Hz), 4.48 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.41 (t, 3H, J = 7.2
- 290 Hz, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, DMSO-d6) δ 166.1, 158.3, 148.8, 142.4, 140.7, 136.0,
- 291 133.4, 132.7, 130.0 (2 × C), 126.7, 126.6, 111.1, 110.6, 107.3, 62.8, 14.5.
- 292
- 293 5.2.4. Ethyl 3-((3-aminophenyl)amino)quinoxaline-2-carboxylate (4b)

The title compound was synthesized according to the general method A from compound 3 (330 mg, 1.40 mmol) and *tert*-butyl (3-aminophenyl)carbamate (312 mg, 1.50 mmol) in absolute ethanol (10 mL). The reaction mixture was refluxed for 110 h. Compound **4b** was obtained (139 mg, 32%) as a red powder.

298 Mp 207.7 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  10.00 (bs, 1H, NH), 7.97 (d, 1H, J =

8.1 Hz), 7.81-7.77 (m, 2H), 7.58-7.51 (m, 1H), 7.26 (bs, 1H), 7.09-6.97 (m, 2H), 6.32
(m, 1H), 5.17 (bs, 2H, NH<sub>2</sub>), 4.48 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.41 (t, 3H, J = 7.2 Hz,

- 301 CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, DMSO-d6)δ 166.1, 149.7, 149.0, 142.6, 140.3, 135.9, 133.3,
- 302 132.5, 130.0, 129.7, 126.7, 126.4, 109.7, 108.2, 105.8, 62.8, 14.5.
- 303

304 5.2.5. *Ethyl 3-((3-morpholinophenyl)amino)quinoxaline-2-carboxylate (4c)* 

The title compound was synthesized according to the general method A from compound 306 **3** (361 mg, 1.53 mmol) and 3-morpholinoaniline (299 mg, 1.68 mmol) in absolute 307 ethanol (10 mL). The reaction mixture was refluxed for 110 h. Compound **4c** was 308 obtained (439 mg, 76%) as an orange powder.

- 309 Mp 170.6 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  10.09 (bs, 1H, NH), 7.98 (d, 1H, J =
- 310 8.1 Hz), 7.87-7.70 (m, 2H), 7.67-7.47 (m, 2H), 7.34 (d, 1H, *J* = 7.8 Hz), 7.24 (t, 1H, *J* =
- 311 8.1 Hz), 6.71 (d, 1H, J = 7.8 Hz), 4.48 (q, 2H, J = 6.9 Hz, CH<sub>2</sub>), 3.77 (m, 4H,  $2 \times$  CH<sub>2</sub>O
- 312 morpholine), 3.16 (m, 4H,  $2 \times CH_2N$ , morpholine), 1.41 (t, 3H, J = 6.9 Hz, CH<sub>3</sub>). <sup>13</sup>C
- 313 NMR (75 MHz, DMSO-d6) δ 166.0, 152.2, 148.9, 142.4, 140.4, 136.0, 133.5, 132.6,
- 314 130.0, 129.8, 126.7 (2 × C), 111.5, 110.7, 107.1, 66.6 (2 × C), 62.8, 48.9 (2 × C), 14.5.
- 315

- 316 5.2.6. Ethyl 3-((1H-indol-5-yl)amino)quinoxaline-2-carboxylate (4d)
- 317 The title compound was synthesized according to the general method A from compound
- 318 3 (237 mg, 1.00 mmol) and 5-aminoindole (397 mg, 3.00 mmol) in absolute ethanol (10
  319 mL). The reaction mixture was refluxed for 36 h. Compound 4d was obtained (245 mg,
- 320 73%) as a red powder.
- 321 Mp 204.7 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  10.24 (bs, 1H, NH), 8.80 (bs, 1H, NH),
- 322 8.26 (s, 1H, indolyl), 8.00 (dd, 1H, *J* = 8.4, 0.9 Hz), 7.77 (dd, 1H, *J* = 8.4, 0.9 Hz), 7.66
- 323 (ddd, 1H, J = 8.4, 6.9, 0.9 Hz), 7.50-7.38 (m, 3H), 7.23 (t, 1H, J = 2.4 Hz, indolyl),
- 324 6.56 (bs, 1H, indolyl), 4.60 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.53 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). <sup>13</sup>C
- 325 NMR (75 MHz, CDCl<sub>3</sub>) δ 166.4, 150.0, 143.6, 136.2, 133.0, 132.8, 131.5, 130.6, 130.1,
- 326 128.1, 126.6, 125.4, 125.1, 117.2, 112.8, 111.2, 102.6, 62.9, 14.3.
- 327

328 5.2.7. Ethyl 3-((4-hydroxyphenyl)amino)quinoxaline-2-carboxylate (4e)

- The title compound was synthesized according to the general method A from compound 3 (302 mg, 1.28 mmol) and 4-aminophenol (418 mg, 3.83 mmol) in absolute ethanol (16 mL). The reaction mixture was refluxed for 20 h. Compound **4e** was obtained (155 mg, 39%) as a red powder.
- 333 Mp 231.8 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  9.84 (bs, 1H, NH), 9.30 (bs, 1H, OH), 334 7.94 (d, 1H, J = 8.1 Hz), 7.80-7.60 (m, 4H), 7.51 (m, 1H), 6.80 (d, 2H, J = 8.7 Hz),
- 335 4.47 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.41 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz,
- 336 DMSO-d6) δ 166.0, 154.0, 149.2, 142.8, 135.9, 133.3, 132.5, 131.0, 130.0, 126.5,
  337 126.1, 122.7 (2 × C), 115.8 (2 × C), 62.7, 14.5.
- 338
- 339 5.2.8. Ethyl 3-((4-((tert-butoxycarbonyl)amino)phenyl)amino)quinoxaline-2-carboxylate
  340 (4f)
- The title compound was synthesized according to the general method A from compound 342 **3** (95 mg, 0.40 mmol), *tert*-butyl (4-aminophenyl)carbamate (250 mg, 1.20 mmol) in 343 absolute ethanol (6.5 mL). The reaction mixture was refluxed for 64 h in a sealed tube. 344 After purification by silica column chromatography using  $CH_2Cl_2$  with MeOH gradient 345 (0-2%) as eluent, compound **4f** was obtained (152 mg, 93%) as an orange powder.

346 Mp 198.1 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  10.00 (bs, 1H, NH), 9.35 (bs, 1H, NH), 347 8.00-7.40 (m, 8H), 4.48 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.49 (s, 9H, 3 × CH<sub>3</sub>), 1.41 (t, 3H, J = 348 7.2 Hz, CH<sub>3</sub>CH<sub>2</sub>O).

349

350 5.2.9. Ethyl 3-((4-aminophenyl)amino)quinoxaline-2-carboxylate (4g)

To a solution of compound **4f** (37 mg, 0.09 mmol) in  $CH_2Cl_2$  (5 mL) was added dropwise trifluoroacetic acid (1 ml, 13.06 mmol). The mixture was stirred at room temperature for 6 h. The resulting mixture was made alkaline with a saturated sodium carbonate solution and extracted with  $CH_2Cl_2$ . The combined organic layers were dried over MgSO<sub>4</sub>, filtered, and evaporated under reduced pressure to give the desired derivative **4g** (27 mg, 96%) as a red powder.

357 Mp 203.3 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  10.06 (bs, 1H, NH), 8.01 (dd, 1H, J = 8.4, 358 1.2 Hz), 7.80-7.64 (m, 4H), 7.44 (ddd, 1H, J = 8.1, 6.6, 1.2 Hz), 6.77 (d, 2H, J = 8.7359 Hz), 4.61 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 3.70 (bs, 2H, NH<sub>2</sub>), 1.55 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). 360 <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  166.4, 149.7, 143.6, 142.6, 136.2, 132.8, 130.6, 130.5, 361 130.2, 126.6, 125.4, 122.4 (2 × C), 115.5 (2 × C), 62.9, 14.3.

362

363 5.2.10. Ethyl 3-((4-(1-methylpiperidin-4-yl)phenyl)amino)quinoxaline-2-carboxylate
364 (4h)

The title compound was synthesized according to the general method A from compound 366 **3** (45 mg, 0.19 mmol) and 4-(1-methylpiperidin-4-yl)aniline (40 mg, 0.21 mmol) in absolute ethanol (1.5 mL). The reaction mixture was refluxed for 112 h. After purification by silica column chromatography using  $CH_2Cl_2$  with MeOH gradient (0-10%) as eluent, compound **4h** was obtained (47 mg, 63%) as an orange powder.

370 Mp 127 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  10.28 (bs, 1H, NH), 8.02 (dd, 1H, J = 8.1,

371 1.2 Hz), 7.84 (d, 2H, J = 8.4 Hz), 7.77 (dd, 1H, J = 8.1, 1.2 Hz), 7.69 (ddd, 1H, J = 8.1,

372 6.6, 1.2 Hz), 7.47 (ddd, 1H, J = 8.1, 6.6, 1.2 Hz), 7.26 (d, 2H, J = 8.4 Hz), 4.59 (q, 2H,

373 *J* = 7.2 Hz, CH<sub>2</sub>), 3.30 (d, 2H, *J* = 11.7 Hz), 2.70-2.45 (m, 6H), 2.24-2.09 (m, 2H), 1.96

374 (d, 2H, J = 11.7 Hz), 1.53 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ 

375 166.4, 149.3, 143.2, 139.4, 137.7, 136.4, 132.9, 130.5, 130.2, 127.3 (2 × C), 126.7,
376 125.9, 120.5 (2 × C), 63.0, 55.7 (2 × C), 45.1, 40.4, 31.9 (2 × C), 14.3.

- 378 5.2.11 Ethyl 3-((2-hydroxyphenyl)amino)quinoxaline-2-carboxylate (4i)
- 379 The title compound was synthesized according to the general method A from compound
- 380 3 (239 mg, 1.01 mmol) and 2-aminophenol (121 mg, 1.11 mmol) in absolute ethanol
  (10 mL). The reaction mixture was refluxed for 110 h. Compound 4i was obtained (156
  mg, 50%) as an orange powder.
- 383 Mp 166.6 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  10.66 (bs, 1H, NH), 10.16 (bs, 1H, 384 OH), 8.89 (m, 1H), 7.99 (d, 1H, J = 8.1 Hz), 7.80 (m, 2H), 7.56 (m, 1H), 7.00-6.82 (m, 385 3H), 4.49 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 1.43 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz,
- 386 DMSO-d6) δ 165.8, 149.0, 147.0, 142.6, 135.9, 133.5, 132.3, 130.1, 128.4, 126.6,
- 387 126.5, 123.1, 119.7, 119.6, 114.7, 62.7, 14.6.
- 388

# 389 5.2.12. 3-((3-Hydroxyphenyl)amino)quinoxaline-2-carboxylic acid (1)

Method B: to ester **4a** (72 mg, 0.23 mmol) in aqueous methanol (80%, 10 mL), was added potassium carbonate (97 mg, 0.70 mmol) and the reaction mixture was refluxed for 4 h. After cooling, the solvent was removed under reduced pressure. Then, the residue was acidified with a saturated citric acid aqueous solution, and extracted with ethyl acetate. The combined organic layers were dried over MgSO<sub>4</sub>, filtered, and evaporated under reduced pressure to yield the acid **1** (33 mg, 50%) as a red powder.

- 396 Mp 199.6 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  10.52 (bs, 1H, NH), 9.51 (bs, 1H, OH),
- 397 7.98 (d, 1H, J = 8.4 Hz), 7.84-7.76 (m, 2H), 7.60-7.54 (m, 2H), 7.25-7.14 (m, 2H), 6.50
- 398 (d, 1H, J = 7.8 Hz). <sup>13</sup>C NMR (75 MHz, DMSO-d6) δ 168.1, 158.3, 149.3, 142.5, 140.7,
  399 136.0, 133.2, 132.9, 130.1, 130.0, 126.6, 126.5, 111.0, 110.5, 107.2. HRMS (ESI) m/z:
- 400  $[M+H]^+$  calcd for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub>O<sub>3</sub>, 282.2742; found, 282.0873.
- 401

# 402 5.2.13. Potassium 3-((3-aminophenyl)amino)quinoxaline-2-carboxylate (5b)

- Method C: to ester 4b (82 mg, 0.26 mmol) in aqueous methanol (80%, 10 mL) was
  added potassium carbonate (37 mg, 0.26 mmol), and the reaction mixture was refluxed
  for 4 h. After cooling, the solvent was removed under reduced pressure, and freezedried to obtain compound 5b (84 mg, 100%) as an orange powder.
- 407 Mp > 375 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  13.15 (bs, 1H, NH), 7.82 (dd, 1H, J =
- 408 8.4, 1.5 Hz), 7.65 (dd, 1H, J = 7.2, 1.8 Hz), 7.58 (td, 1H, J = 7.2, 1.5 Hz), 7.38 (ddd,
- 409 1H, J = 8.4, 7.2, 1.8 Hz), 7.21 (t, 1H, J = 1.8 Hz), 7.14 (dd, 1H, J = 7.8, 1.8 Hz), 6.97

410 (t, 1H, J = 7.8 Hz), 6.22 (dd, 1H, J = 7.8, 1.8 Hz), 5.04 (bs, 2H, NH<sub>2</sub>). <sup>13</sup>C NMR (75 411 MHz, DMSO-d6)  $\delta$  166.7, 150.4, 149.6, 143.2, 141.6, 141.5, 136.6, 130.2, 129.5 (2 × 412 C), 125.9, 124.4, 108.5, 107.5, 104.9. HRMS (ESI) m/z: [M+H]<sup>+</sup> calcd for C<sub>15</sub>H<sub>13</sub>N<sub>4</sub>O<sub>2</sub>, 413 281.2895; found, 281.1032.

414

## 415 *5.2.14. 3-((3-Morpholinophenyl)amino)quinoxaline-2-carboxylic acid (5c)*

The title compound was synthesized according to the general method B from compound
4c (150 mg, 0.40 mmol) and potassium carbonate (218 mg, 1.58 mmol) in aqueous
methanol (80%, 5 mL). The reaction mixture was refluxed for 4 h. Compound 5c was
obtained (139 mg, 100%) as an orange powder.

420 Mp 186.6 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  10.49 (bs, 1H, NH), 7.97 (m, 1H), 421 7.82-7.74 (m, 2H), 7.67 (s, 1H), 7.56 (ddd, 1H, *J* = 8.4, 6.3, 2.1 Hz) 7.32 (d, 1H, *J* = 8.7 422 Hz), 7.23 (t, 1H, *J* = 8.1 Hz), 6.69 (dd, 1H, *J* = 8.1, 1.5 Hz), 3.77 (m, 4H, 2 × CH<sub>2</sub>O 423 morpholine), 3.16 (m, 4H, 2 × CH<sub>2</sub>N, morpholine). <sup>13</sup>C NMR (75 MHz, DMSO-d6)  $\delta$ 424 168.0, 152.2, 149.4, 142.5, 140.5, 135.9, 133.2, 132.9, 129.9, 129.7, 126.7, 126.4, 425 111.4, 110.5, 106.9, 66.6 (2 × C), 49.0 (2 × C). HRMS (ESI) m/z: [M+H]<sup>+</sup> calcd for 426 C<sub>19</sub>H<sub>19</sub>N<sub>4</sub>O<sub>3</sub>, 351.3793; found, 351.1453.

427

# 428 5.2.15. 3-((1H-Indol-5-yl)amino)quinoxaline-2-carboxylic acid (5d)

429 To ester 4d (81 mg, 0.24 mmol) in ethanol (5 mL), was added a 10% sodium hydroxide 430 solution (2 mL) and the reaction mixture was refluxed for 18 h. Ethanol was then 431 evaporated under reduced pressure, and the resulting residue was acidified to pH 2 with 432 a 15% citric acid solution and extracted with ethyl acetate. The organic layers were then 433 washed with water and brine, dried over MgSO<sub>4</sub>, filtered, and evaporated under reduced 434 pressure. The residue was finally purified by silica column chromatography using 435  $CH_2Cl_2$  with MeOH gradient (0-20%) as eluent to give the acid 5d (30 mg, 41%) as a 436 red powder.

437 Mp 244.4 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  12.45 (bs, 1H, NH), 11.07 (bs, 1H, 438 NH), 8.70-6.80 (m, 8H), 6.44 (bs, 1H, indolyl). <sup>13</sup>C NMR (75 MHz, DMSO-d6)  $\delta$ 439 168.6, 151.4, 133.6, 133.1, 132.5, 129.9, 129.4, 129.1 (3 × C), 127.1, 126.9, 125.7, 440 116.6, 112.9, 111.6, 102.5. HRMS (ESI) m/z: [M+H]<sup>+</sup> calcd for C<sub>17</sub>H<sub>13</sub>N<sub>4</sub>O<sub>2</sub>, 305.3109; 441 found, 305.1034.

443 5.2.16. 3-((4-Hydroxyphenyl)amino)quinoxaline-2-carboxylic acid (5e)

The title compound was synthesized according to the general method B from compound
4e (90 mg, 0.29 mmol) and potassium carbonate (80 mg, 0.58 mmol) in aqueous
methanol (80%, 10 mL). The reaction mixture was refluxed for 4 h. Compound 5e was
obtained (79 mg, 97%) as an orange powder.

- 448 Mp 197.4 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6) δ 10.22 (bs, 1H, NH), 9.31 (bs, 1H, OH),
- 449 7.93 (dd, 1H, J = 8.1, 0.9 Hz), 7.78-7.63 (m, 4H), 7.51 (ddd, 1H, J = 8.1, 6.6, 1.5 Hz),
- 450 6.80 (d, 2H, J = 8.7 Hz). <sup>13</sup>C NMR (75 MHz, DMSO-d6) δ 168.1, 153.9, 149.6, 142.9,
- 451 135.8, 133.2, 132.7, 131.1, 129.9, 126.4, 125.9, 122.5 (2  $\times$  C), 115.8 (2  $\times$  C). HRMS

452 (ESI) m/z:  $[M+H]^+$  calcd for  $C_{15}H_{12}N_3O_3$ , 282.2742; found, 282.0873.

453

454 5.2.17. Potassium 3-((4-aminophenyl)amino)quinoxaline-2-carboxylate (5g)

- The title compound was synthesized according to the general method C from compound
  4g (16 mg, 0.05 mmol) and potassium carbonate (7 mg, 0.05 mmol) in aqueous
  methanol (80%, 5 mL). The reaction mixture was refluxed for 4 h. Compound 5g was
  obtained (14 mg, 87%) as a red powder.
- 459 Mp > 375 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  12.80 (bs, 1H, NH), 7.84 (d, 1H, J =460 7.8 Hz), 7.60-7.50 (m, 4H), 7.32 (m, 1H), 6.60 (d, 2H, J = 8.4 Hz), 4.84 (bs, 2H NH<sub>2</sub>). 461 <sup>13</sup>C NMR (75 MHz, DMSO-d6)  $\delta$  167.0, 150.4, 144.2, 143.0, 142.0, 136.4, 130.4, 462 130.2, 129.5, 125.6, 123.8, 120.9 (2 × C), 114.7 (2 × C). HRMS (ESI) m/z: [M+H]<sup>+</sup>
- 463 calcd for  $C_{15}H_{13}N_4O_2$ , 281.2895; found, 281.1032.
- 464

# 465 5.2.18. 3-((4-(1-Methylpiperidin-4-yl)phenyl)amino)quinoxaline-2-carboxylic acid (5h)

The title compound was synthesized according to the general method B from compound
467 4h (42 mg, 0.11 mmol) and potassium carbonate (45 mg, 0.32 mmol) in aqueous
468 methanol (80%, 5 mL). The reaction mixture was refluxed for 4 h. Compound 5h was
469 obtained (19 mg, 49%) as a yellow powder.

- 470 Mp 321 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  13.33 (bs, 1H, NH), 8.00 (d, 1H, J = 7.8
- 471 Hz), 7.85 (d, 2H, J = 8.4 Hz), 7.68-7.58 (m, 2H), 7.54-7.31 (m, 1H), 7.23 (d, 2H, J = 1.0
- 472 8.4 Hz), 2.86 (d, 2H, J = 11.4 Hz), 2.46-2.36 (m, 1H), 2.19 (s, 3H, CH<sub>3</sub>), 1.95 (td, 2H, J
- 473 = 11.4, 1.8 Hz), 1.77-1.62 (m, 4H). <sup>13</sup>C NMR (75 MHz, DMSO-d6)  $\delta$  166.9, 150.4,

474 142.1, 141.7, 140.1, 138.8, 136.4, 130.7, 129.8, 127.5 (2 × C), 125.9, 124.7, 119.3 (2 × C), 56.4 (2 × C), 46.7, 41.2, 33.7 (2 × C). HRMS (ESI) m/z: an abundant fragment ion 476 has been observed at m/z 213.1022 that have been attributed to the loss of the 1-477 methylpiperidin-4-yl)phenyl)amino moiety from the quinoxaline ring to form the ion 478  $[C_9H_6N_2O_2 + K]^+$  (m/z calcd for  $C_9H_6KN_2O_2$ , 213.2545; found, 213.1022).

479

# 480 5.2.19. 3-((2-Hydroxyphenyl)amino)quinoxaline-2-carboxylic acid (5i)

The title compound was synthesized according to the general method B from compound
4i (101 mg, 0.33 mmol) and potassium carbonate (135 mg, 0.98 mmol) in aqueous
methanol (80%, 10 mL). The reaction mixture was refluxed for 4 h. Compound 5i was
obtained (60 mg, 65%) as a red powder.

- 485 Mp 191.1 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6) δ 14.00 (bs, 1H, COOH), 10.84 (bs, 1H,
- 486 NH), 10.11 (bs, 1H, OH), 8.88 (m, 1H), 7.97 (d, 1H, J = 8.1 Hz), 7.80 (m, 2H), 7.56 (m,
- 487 1H), 7.00-6.68 (m, 3H). <sup>13</sup>C NMR (75 MHz, DMSO-d6) δ 167.1, 148.7, 146.5, 142.2,
- 488 135.3, 132.7, 132.4, 129.4, 127.9, 126.1, 125.8, 122.5, 119.2, 119.0, 114.2. HRMS
- 489 (ESI) m/z:  $[M+H]^+$  calcd for  $C_{15}H_{12}N_3O_3$ , 282.2742; found, 282.0871.
- 490

## 491 5.2.20. 3-((3-Morpholinophenyl)amino)quinoxaline-2-carboxamide (6)

To a solution of compound **5c** (50 mg, 0.14 mmol) and *N*-methylmorpholine (31  $\mu$ L, 0.29 mmol) in dichloromethane (5 mL) at 0 °C, was added ethyl chloroformate (21  $\mu$ L, 0.21 mmol). The reaction mixture was stirred magnetically at 0 °C for 1 h, and a 28% solution of ammonium hydroxide (5 mL) was added. The reaction was stirred overnight at room temperature and extracted with dichloromethane. The organic layer was then washed with water and brine, dried over MgSO<sub>4</sub>, filtered, and evaporated under reduced pressure to give the carboxamide **6** (50 mg, 100%) as a red powder.

- 499 Mp 203.9 °C. <sup>1</sup>H NMR (300 MHz, DMSO-d6)  $\delta$  11.52 (bs, 1H, NH), 8.75 (bs, 1H,
- 500 NH<sub>2</sub>), 8.26 (bs, 1H, NH<sub>2</sub>), 7.93 (d, 1H, J = 8.1 Hz), 7.82-7.50 (m, 2H), 7.69 (bs, 1H),
- 501 7.56 (m, 1H), 7.32-7.20 (m, 2H), 6.69 (d, 1H, J = 8.1 Hz), 3.78 (m, 4H,  $2 \times CH_2O$
- 502 morpholine), 3.17 (m, 4H, 2 × CH<sub>2</sub>N, morpholine). <sup>13</sup>C NMR (75 MHz, DMSO-d6)  $\delta$
- 503 168.5, 152.2, 149.4, 142.7, 140.6, 135.3, 133.1, 132.8, 129.8, 129.6, 126.7, 126.3,
- 504 111.1, 110.3, 106.6, 66.6 (2 × C), 49.0 (2 × C). HRMS (ESI) m/z:  $[M+H]^+$  calcd for
- 505  $C_{19}H_{20}N_5O_2$ , 350.3946; found, 350.1613.

# 507 5.3. Molecular Modeling

508

509 Molecular modeling studies were performed using SYBYL-X 1.3 software [33] running 510 on a Dell precision T3400 workstation. The three-dimensional structure of compound 1 511 (under its carboxylate form to imitate physiological conditions) was built from a 512 standard fragments library and optimized using the Tripos force field [34] including the 513 electrostatic term calculated from Gasteiger and Hückel atomic charges. Powell's 514 method available in Maximin2 procedure was used for energy minimization until the gradient value was smaller than 0.001 kcal/(mol\*Å). The crystal structure of Pim-1 in 515 516 complex with AMP-PNP at 1.6 Å resolution (PDB ID 3A99) [35] was used as template 517 for docking. Water molecules were removed from the coordinates set since no 518 information about conserved water molecules is known for this chemical series in Pim-519 1. Flexible docking of compound 1 into ATP-binding site was performed using GOLD 520 software [36]. The most stable docking model was selected according to the best scored 521 conformation predicted by the GoldScore scoring function. Finally, the complexe was 522 energy-minimized using Powell's method available in Maximin2 procedure with the 523 Tripos force field and a dielectric constant of 4.0, until the gradient value reached 0.1 524 kcal/mol.Å.

525

526 *5.4. Biology* 

527

# 528 5.4.1. Mammalian protein kinase assays

Kinase enzymatic activities were assayed in 384-well plates using the ADP-Glo<sup>TM</sup> assay 529 530 kit (Promega, Madison, WI) according to the recommendations of the manufacturer. This assay is a luminescent ADP detection assay that provides a homogeneous and 531 532 high-throughput screening method to measure kinase activity by quantifying the amount 533 of ADP produced during a kinase reaction. Briefly, the reactions were carried out in a 534 final volume of 5 µL for 30 min at 30 °C in ADP-Glo buffer and 10 µM ATP (40 mM Tris pH 7.5, 20 mM MgCl<sub>2</sub> and 0.1 mg/mL of BSA). After that, 5  $\mu$ L of ADP-Glo<sup>TM</sup> 535 536 Kinase Reagent was added to stop the kinase reaction. After an incubation time of 50 min at room temperature (rt), 10 µL of Kinase Detection Reagent was added for one 537

hour at rt. The transmitted signal was measured using the Envision (PerkinElmer, Waltham, MA) microplate luminometer and expressed in Relative Light Unit (RLU). In order to determine the half maximal inhibitory concentration ( $IC_{50}$ ), the assays were performed in triplicate in the absence or presence of increasing doses of the tested compounds. Kinase activities are expressed in % of maximal activity, *i.e.* measured in the absence of inhibitor. Peptide substrates were obtained from Proteogenix (Schiltigheim, France).

545 The following kinases were analyzed during this study: HsPim-1 (human proto-546 oncogene, recombinant, expressed in bacteria) was assayed with 0.8 µg/µL of histone 547 H1 (Sigma #H5505) as substrate; RnDYRK1A-kd (Rattus norvegicus, amino acids 1 to 548 499 including the kinase domain, recombinant, expressed in bacteria, DNA vector 549 kindly provided by Dr. W. Becker, Aachen, Germany) was assayed with 0.033 µg/µL of 550 the following peptide: KKISGRLSPIMTEQ as substrate; HsCDK2/CyclinA (cyclin-551 dependent kinase-2, human, kindly provided by Dr. A. Echalier-Glazer, Leicester, UK) 552 was assayed with 0.8 µg/µL of histone H1 as substrate; HsCDK9/CyclinT (human, 553 recombinant, expressed by baculovirus in Sf9 insect cells) was assayed with 0.27 µg/µL 554 of the following peptide: YSPTSPSYSPTSPSYSPTSPSKKKK, as substrate; HsHaspin-555 kd (human, kinase domain, amino acids 470 to 798, recombinant, expressed in bacteria) 556 0.007 μg/μL of Histone H3 was assayed with (1-21)peptide 557 (ARTKQTARKSTGGKAPRKQLA) as substrate; MmCLK1 (from Mus musculus, 558 recombinant, expressed in bacteria) was assayed with 0.027 µg/µL of the following 559 peptide: GRSRSRSRSRSR as substrate;  $SscCK1\delta/\epsilon$  (casein kinase  $1\delta/\epsilon$ , porcine brain, 560 native, affinity purified) was assayed with 0.022  $\mu g/\mu L$  of the following peptide: 561 RRKHAAIGSpAYSITA ("Sp" stands for phosphorylated serine) as CK1-specific 562 substrate;  $SscGSK-3\alpha/\beta$  (glycogen synthase kinase-3, porcine brain, native, affinity 563 purified) isoforms were assayed with 0.010  $\mu$ g/ $\mu$ L of GS-1 peptide, a GSK-3-selective 564 substrate (YRRAAVPPSPSLSRHSSPHQSpEDEEE). To validate the kinase assay, 565 model inhibitors were used for each tested enzyme: Staurosporine from Streptomyces 566 *sp.* (#S5921, purity  $\geq$ 95%, Sigma-Aldrich) for *Ssc*CK1 $\delta$ / $\epsilon$ ; Indirubin-3'-oxime (#I0404, purity  $\geq 98\%$ , Sigma-Aldrich) for SscGSK-3 $\alpha/\beta$ , HsPim-1, human Cyclin-dependent 567 568 kinases, RnDYRK1A and MmCLK1; CHR-6494 (#SML0648, purity ≥98%, Sigma-569 Aldrich) for Haspin.

#### 571 5.4.2. Cell Cultures and Reagents

572 KU812 cell lines were obtained from the American Type Culture Collection (ATCC) 573 and maintained according to the supplier's recommendations. Cell lines were cultured 574 in RPMI, with 10% fetal bovine serum, 1% glutamine, and 1% penicillin/streptomycin 575 at 37 °C and 5% CO<sub>2</sub>.

576

577 5.4.3. Cell Proliferation Assays

578 Cell viability and proliferation were studied using a MTT cell proliferation assay. Briefly,  $0.2 \times 10^5$  cells were incubated in 100 µL of X-Vivo red phenol free medium 579 580 (Lonza, Basel, Switzerland) in 96 well plates. In initial screening assays, cells were 581 incubated with 10 µM of each compound (quinoxalines stock solution at 50 mM in 582 DMSO) for 24, 48, and 72 h. Imatinib mesylate (Selleckchem, stock solution at 10 mM 583 in DMSO) was used as reference. To determine the concentration-effect of the 584 molecules, cells were treated with concentrations ranging from 100 nM to 50 µM for 24 585 or 48 h. Cells were incubated with 10 µL of MTT working solution (5 g/L of 586 methylthiazolyldiphenyl-tetrazolium bromide) during 4 h. Cells were then lysed 587 overnight at 37 °C with 100 µL of 10% SDS and 0.003% HCl. Optical density (OD) at 588 570 nm was measured using a spectrophotometer (Dynex, Chantilly, United States). 589 Living cells were also counted with the trypan blue dye exclusion method. When a 590 dose-dependent activity was observed, EC<sub>50</sub> values were calculated using Graphpad 591 PRISM 7 software (n = 3 in triplicate). Data were collected from at least three 592 independent experiments and the values reported are means  $\pm$  standard errors of the 593 mean.

594

#### 595 Acknowledgments

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614	Refe	erences
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- **1** : *Hs*Pim-1 IC<sub>50</sub> = 74 nM
- **Fig. 1.** Chemical structure of compound **1**.



- **Fig. 2.** Binding pose found by the docking program GOLD for compound **1** into the
- 787 ATP pocket of Pim-1 (PDB ID 3A99). Hydrogen bonds are indicated as yellow lines.
- 788



791 Scheme 1. Reagents and conditions: (*i*) diethyl 2-oxomalonate (1 eq), citric acid (3 792 mol%), EtOH, rt, 10 min, 79%; (*ii*) DMF (cat.), POCl<sub>3</sub>, 0°C, and then reflux, 30 min, 793 100%; (*iii*) amine (1.1-3 eq), *p*-toluenesulfonic acid (cat.), EtOH, reflux, 20-112 h, 32-794 93%; (*iv*) TFA (20%), DCM, rt, 6 h, 96%; (*v*) K<sub>2</sub>CO<sub>3</sub> (1-4 eq), MeOH/H<sub>2</sub>O (4/1), reflux, 795 4 h, 49-100%; (*vi*) NaOH (10%), EtOH, reflux, 18 h, 41%; (*vii*) ClCOOEt (1.5 eq), 796 NMM (2 eq), DCM, 0°C, 1 h, and NH<sub>4</sub>OH, rt, overnight, 100%.

# 797 **Table 1**

798 Enzymatic assays on *Hs*Pim-1.

Entry	Compd	$R_1$	$R_2$	<i>Hs</i> Pim-1 IC <sub>50</sub> $(\mu M)^{a}$
1	4a		OEt	> 10
2	1	-	ОН	0.074
3	4b	- NH <sub>2</sub>	OEt	> 10
4	5b		ОК	2.80
5	4c		OEt	> 10
6	5c		ОН	1.01
7	6		$\mathrm{NH}_2$	> 10
8	4d		OEt	> 10
9	5d		ОН	> 10
10	4e	ОН	OEt	> 10
11	5e	— — ОН	ОН	0.29
12	4g		OEt	> 10
13	5g		ОК	> 10
14	4h		OEt	> 10
15	5h		ОН	> 10
16	4i	HO	OEt	> 10
17	5i		ОН	0.76
18	Staurosporine			0.031

<sup>a</sup> Values are a mean of  $n \ge 3$  independent experiments. *Hs*: *Homo sapiens*.

801 Table 2

Kinase selectivity profile and cell-based assays of most active quinoxalines. 802 803



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				Kinase enz	Kinase enzymatic $IC_{50} (\mu M)^a$					$EC_{50} (\mu M)^a$		
Entry	Cpd	$R_1$	$R_2$	HsPim-1	<i>Rn</i> DYRK1A	HsCDK2 /CyclinA	HsCDK9 /CyclinT	<i>Hs</i> Haspin	Mm CLK1	<i>Ssc</i> CK1δ/ε	Ssc GSK3α/β	KU 812 <sup>b</sup>
1	1	- С	Н	0.074	0.27	> 10	> 10	> 10	> 10	> 10	> 10	$38.9\pm3.4$
2	5b		К	2.80	1.67	> 10	> 10	> 10	> 10	> 10	> 10	$63.8 \pm 1.9$
3	5c		Н	1.01	> 10	> 10	> 10	> 10	> 10	> 10	> 10	57.3 ± 6.1
4	5e	— — ОН	Н	0.29	0.098	> 10	> 10	> 10	> 10	> 10	> 10	$41.7\pm3.7$
5	5i	HO	Н	0.76	0.74	> 10	> 10	> 10	> 10	> 10	> 10	$177.5 \pm 13.1$
6	5 Imatinib mesylate			ND	ND	ND	ND	ND	ND	ND	ND	$0.6\ \pm 0.02$

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<sup>a</sup> Values are a mean of  $n \ge 3$  independent experiments. <sup>b</sup> Cells were treated with concentrations ranging from 100 nM to 50  $\mu$ M for 48 h. Cell viability was then determined by MTT assays, and EC<sub>50</sub> values were 807 808 calculated using Graphpad PRISM 7 software (n = 3 in triplicate; data are the mean  $\pm$  SEM).

809 Rn: Rattus norvegicus, Hs: Homo sapiens, Mm: Mus musculus, Ssc: Sus scrofa, DYRK1A: dual specificity tyrosine phosphorylation regulated kinase 1A, CDK:

810 cyclin-dependent kinase, Haspin: haploid germ cell-specific nuclear protein kinase, CLK1: CDC2-like kinase 1, CK1: casein kinase 1, GSK3: glycogen 811 synthase kinase 3, ND: not determined.

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- 17 new quinoxaline-2-carboxylic acid derivatives targeting Pim-1 were synthesized.
- A nanomolar inhibitor of *Hs*Pim-1 (1) was identified.
- Molecular modeling suggested a non-ATP mimetic binding mode.
- Best candidates exhibited *in vitro* antitumor activity at micromolar concentrations.