

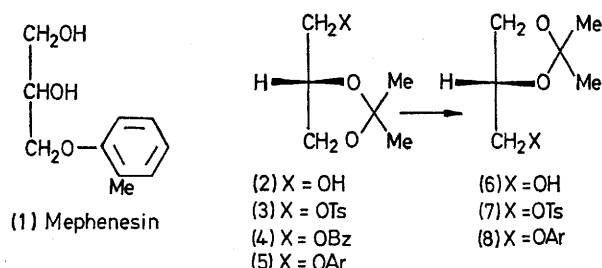
Absolute Configuration of 3-Aryloxypropane-1,2-diols and Derivatives: Mephenesin Isomers

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Summary The optical isomers of mephenesin (3-*o*-tolylpropane-1,2-diol) were prepared from 2*R*- and 2*S*-3-tosyloxypropane-1,2-diol acetonide by reaction with *o*-cresol (NaOH) and the configurations confirmed on the basis of c.d. spectra in Cupra A solution.

MANY important drugs demonstrate stereospecificity in their pharmacological actions and/or metabolism, sometimes showing enantiomeric potency differences of several hundred-fold, or even different pharmacological effects.¹ Demonstration of these significant differences is dependent upon finding methods to obtain materials of known absolute stereochemistry and high optical purity.



We report the preparation of the optical isomers of 3-*o*-tolylpropane-1,2-diol, mephenesin, (1),² suitable for study of muscle relaxants of known absolute configuration, and an intermediate suitable for obtaining the optical isomers of 1-alkylamino-3-aryloxypropan-2-ols,[†] important selective adrenergic agonists and antagonists. The acetonide (2) was prepared by the method of Baer from (+)-2*R*,3*R*,4*R*,5*R*-mannitol.³ 2*S*-Glycerol-2,3-acetonide (2) was converted into the 2*S*-tosylate ester (7), by formation of the benzyl ether, hydrolysis of the acetal, tosylate ester formation, hydrogenolysis and subsequent reformation of the acetonide acetal.⁴ The respective tosylates, (3) and (7), were converted into aryl ethers (5) and (8) respectively using sodium *o*-cresolate [MeOH; 110° (sealed bomb); 30 h] followed by hydrolysis, (0.5*N* aq. HCl; 70°; 2 h) to afford respectively 2*R*-mephenesin, m.p. 89–90° (H₂O) (70% yield), and 2*S*-mephenesin, m.p. 89–90° (H₂O) (65% yield). The i.r. spectra (KBr) are not significantly different than that recorded for mephenesin, m.p. 70–72°.⁵ High-resolution mass spectral fragmentation were essentially identical with the spectrum of *rac*-mephenesin.

[†] Following completion of this work a report appeared using a related derivative in the synthesis of an isomer of practalol, thus establishing its absolute configuration: J. C. Danilewicz and J. E. G. Kemp, *J. Medicin. Chem.*, 1973, **16**, 168. Similar results have been obtained in this laboratory.

[‡] We thank Dr. L. A. Mitscher and Mr. M. S. Bathala, Ohio State University, for these spectra.

[§] Incomplete separation of the half-acid phthalates of mephenesin has been reported, although no assignments of absolute stereochemistry were made: K. A. Thaker and S. H. Patel, *J. Sci. Ind. Res., India*, 1961, **20B**, 327.

[¶] C.d. measurements (in deg cm²/mol) were recorded on a JASCO UV/ORD/5 instrument with a c.d. attachment (Sproul Instrument Corporation, Model SS20-2). Intensities are not absolute since the reaction between glycols and Cupra A is an equilibrium process. Solutions were prepared in Cupra A solution [0.01*M*-Cu²⁺ dissolved in 0.34*M*-NH₃ (H₂O-EtOH)].⁶

Optical rotations, Na_D line, were small and not useful. Differentiation between isomers using o.r.d. spectra or n.m.r. spectra in asymmetric solvents failed. The c.d. spectra, in Cupra A solution (Figure), provided clear distinction between optical isomers,^{‡§} and provided data consistent with the known absolute configurations based on synthesis.

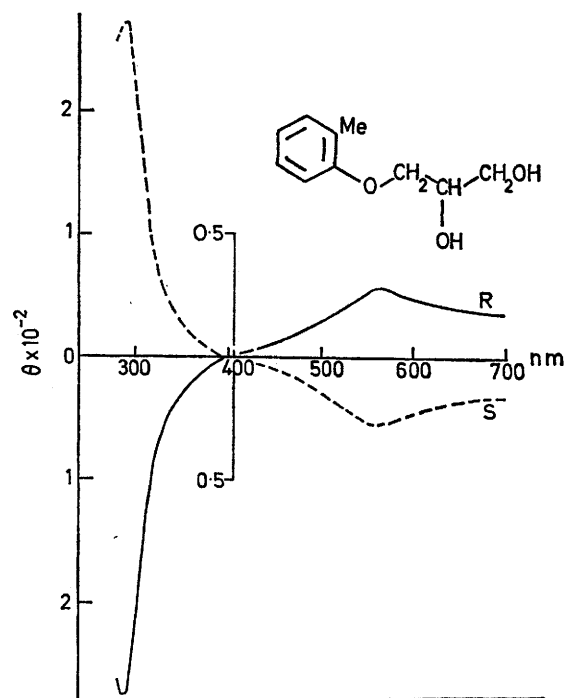
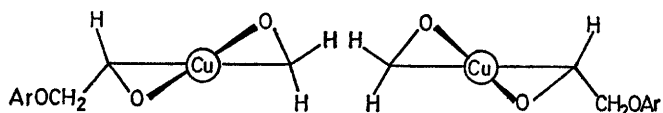


FIGURE. C.d. spectra of mephenesin isomers in Cupra A solution.

The c.d. spectra of 2*R*- and 2*S*-mephenesin show positive and negative bands respectively in the 600 nm region and Cotton effects of opposite sign in the 300 nm region consistent with their absolute configurations:[¶] 2*R* (*c* 0.220 Cupra A) [θ]₅₆₀ 30°, [θ]₃₈₀ 0°, [θ]₂₉₀ –275°; (*c* 0.232 Cupra A) [θ]₅₆₅ –30°, [θ]₃₈₀ 0°, [θ]₂₉₀ 270°.

The Cotton effects are similar to those consistently observed for the chloramphenicol isomers, and assigned to the chiral centre at C-1.⁷ The signs of the bands in the



λ 2S-Mephenesin
Ar = *o*-Tolyl

δ 2R-Mephenesin
Ar = *o*-Tolyl

400—700 nm region are consistent with stereochemistry assigned from Cupra A spectra of other 1,2-glycols,⁸ and to the chiral phenylethane-1,2-diols.⁹ Conformations of δ and λ ^{8,10} are, therefore, assigned to the diol-Cupra A complexes of 2R- and 2S-mephenesin, respectively.

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¹ For recent reviews, see A. F. Casy, 'Medicinal Chemistry,' 3rd edn., ed. A. Burger, Wiley, New York, 1970, pp. 81—107; W. L. Alworth, 'Stereochemistry and Its Applications in Biochemistry—The Relations Between Symmetry and Biological Stereospecificity,' Wiley-Interscience, New York, 1972.

² The glycol muscle relaxants have been reviewed: H. B. Donahoe and K. K. Kimura, 'Drugs Affecting the Central Nervous System,' Medicinal Research Series, vol. 2, ed. A. Burger, Dekker, New York, 1968, pp. 265—326; E. L. Engelhart and C. A. Stone, 'Medicinal Chemistry,' 3rd edn., ed. A. Burger, Wiley, New York, 1970, pp. 1525—1537.

³ E. Baer, *Biochem. Prep.*, 1952, **2**, 31; E. Baer and Fischer, *J. Amer. Chem. Soc.*, 1955, **67**, 944, 2031.

⁴ O. T. Schmidt and W. Blank, *Chem. Ber.*, 1956, **89**, 283; J. S. Brimacombe, A. B. Foster, and A. H. Haines, *J. Chem. Soc.*, 1960, 2582; B. Belleau and J. Puranen, *J. Medicin. Chem.*, 1963, **6**, 325; D. Triggle and B. Belleau, *Canad. J. Chem.*, 1962, **40**, 1201.

⁵ E. C. G. Clarke, 'Isolation and Identification of Drugs,' Pharmaceutical Press, London, 1969, p. 744.

⁶ R. E. Reeves, *Methods Carbohydrate Chem.*, 1965, **5**, 203; R. E. Reeves, *Adv. Carbohydrate Chem.*, 1957, **6**, 107.

⁷ L. A. Mitscher, P. W. Howison, and T. D. Sokolowski, *J. Medicin. Chem.*, 1973, **16**, 98.

⁸ S. T. K. Bukhari, R. D. Guthrie, A. I. Scott, and A. D. Wrixon, *Tetrahedron*, 1969, **26**, 3653.

⁹ L. A. Mitscher, G. Clark, and M. S. Bathala, unpublished results.

¹⁰ J. I. Legg and B. E. Douglas, *J. Amer. Chem. Soc.*, 1966, **88**, 2697; *IUPAC Information Bulletin*, 1968, **33**, 68.