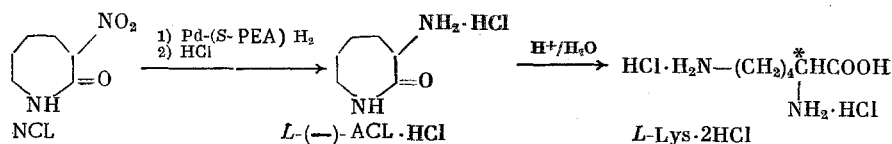


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UDC 541.128:542.91:547.466.46

We have accomplished the asymmetric hydrogenation of 3-nitrocaprolactam (NCL) by the action of a chiral catalytic system containing PdCl_2 and *S*- α -phenylethylamine (PEA) previously used for the asymmetric reductive aminolysis of azlactones [1]. The hydrogenation using 1 mmole NCL, 0.18 mmole PdCl_2 , 1.5 mmole amine, and 15 ml dimethoxyethane proceeds at about 20°C at atmospheric pressure over 5 h. Treatment with HCl gave the hydrochloride salt of (-)-3-aminocaprolactam (ACL•HCl) with an 11% excess of the L-(-) enantiomer, $[\alpha]_D^{20} -2.7^\circ$ (C 5.2, 6 N HCl). Acid hydrolysis of this salt gave L-(+)-lysine•2HCl with the same optical purity



The hydrogenation of NCL, as described in a patent [2], in the presence of $\text{Ru}_2\text{Cl}_4[(-)\text{-DIOP}]_2$ at 75°C and 30 atm H_2 for 70 h gave (+)-ACL•HCl with $[\alpha]_D^{20} +10.3^\circ$ and 39% excess of the optically active enantiomer and, after hydrolysis, to L-lysine•2HCl, mp 177-178°C (the excess of the optically active enantiomer was not indicated).

LITERATURE CITED

1. E. I. Karpeiskaya, L. F. Godunova, E. S. Neupokoeva, and E. I. Klabunovskii, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1104 (1978).
2. European Patent No. 0083332 A2 (1983); *Chem. Abstr.*, **99**, 158860j (1983).

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Translated from *Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya*, No. 9, p. 2157, September, 1985. Original article submitted April 18, 1985.