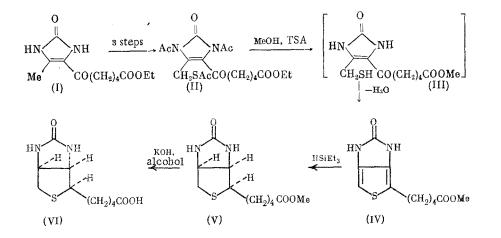
## S. I. Zav'yalov and O. V. Dorofeeva

The three-step synthesis of the ethyl ester of N,N'-diacetyl- $\alpha$ -keto- $\alpha$ '-acetylthiodehydrodesthiobiotin (II) from the ethyl ester of  $\alpha$ -ketodehydrodesthiobiotin (I) was accomplished previously [1-3]. In this paper we describe the preparative conversion of thioacetate (II) to racemic biotin (VI) by the following scheme:



The methyl ester of 2,3,4,5-tetradehydrobiotin (IV) was obtained in 75% yield by heating thioacetate (II) with MeOH in the presence of p-toluenesulfonic acid (TSA). The N-deacetylation of (II), transesterification, and cyclization of the intermediate mercapto ketoester (III) all occur during reaction [3].

The subsequent cis-hydrogenation of the thiophene ring of ester (IV) using  $HSiEt_3$  in CF<sub>3</sub>COOH as described in [4], in the presence of BF<sub>3</sub> etherate and alkaline  $Al_2O_3$ , at 45-50°C, led to the methyl ester of biotin (V), which after hydrolysis with alcoholic KOH solution gave technical (VI) in 80% yield when based on (IV). Running the hydrogenation of (IV) without the  $Al_2O_3$  lowered the yield of technical (VI) to 40%, and down to 5-8% without the BF<sub>3</sub> etherate.

The growth stimulating activity of technical (VI) toward *Saccharomyces cerevisiae* was 40% of the activity of d-biotin.

Technical (VI) was purified by heating with dilute  $H_2SO_4$  solution, and then with aqueous Na<sub>2</sub>CO<sub>3</sub> solution in the presence of resin AV-172 P, followed by recrystallization from 80% alcohol. The yield of the purified (VI) was 50% when based on (IV). The purified (VI) had 50% of the activity of d-biotin when tested on *Saccharomyces cerevisiae* and *Lactobacterium plantarum*. The obtained (VI) coincided with an authentic specimen in its melting point, R<sub>f</sub>, and the IR, PMR, and mass spectra.

## EXPERIMENTAL

The UV spectrum was taken in alcohol solution on a Specord UV-VIS instrument, and the IR spectrum was taken as a KBr pellet on a UR-20 instrument. The PMR spectra were obtained on a Varian DA-60-IL instrument at an operating frequency of 60 MHz (internal standard = HMDS), and the mass spectrum was obtained on a Varian MAT CH-6 instrument with direct insertion of the substance into the ion source at an ionizing energy of 70 eV and an ionization chamber temperature of 180°. The TLC was run on Silufol UV-254.

<u>Methyl Ester of 2,3,4,5-Tetradehydrobiotin (IV)</u>. A solution of 2 g of the Et ester of N,N'-diacetyl- $\alpha$ -keto- $\alpha$ '-acetylthiodehydrodesthiobiotin (II) and 0.2 g of TSA in 10 ml of

N. D. Zelinskii Institute of Organic Chemistry, Academy of Sciences of the USSR, Moscow. Translated from Izvestiya Akademii Nauk SSSR, Khimicheskaya, No. 3, pp. 692-694, March, 1983. Original article submitted July 30, 1982. MeOH was refluxed for 4 h, cooled to  $\sim 20^{\circ}$ , treated with aqueous Na<sub>2</sub>CO<sub>3</sub> solution to pH 7.5-8, and extracted with ethyl acetate (EA). The extract was dried over MgSO<sub>4</sub>, filtered through either an SiO<sub>2</sub> ( $\sim 100$  mesh) or an Al<sub>2</sub>O<sub>3</sub> (II activity) bed, and evaporated in vacuo. We obtained 0.92 g (75%) of (IV), mp 147-148° (from benzene), R<sub>f</sub> 0.32 (EA, the spots were detected either with iodine vapors or in UV light). Ultraviolet spectrum:  $\lambda_{max}$  260 nm. PMR spectrum (C<sub>5</sub>D<sub>5</sub>N,  $\delta$ , ppm): 1.56 m (CH<sub>2</sub>CH<sub>2</sub>), 2.16 m (CH<sub>2</sub>COOCH<sub>3</sub>), 2.63 m (CH<sub>2</sub>C=C), 3.47 s (CH<sub>3</sub>O), 6.18 s (HC-C), 11.40 s (2NH).

Racemic Biotin (VI). With stirring, to 20 ml of CF<sub>3</sub>COOH were successively added 1 ml of BF<sub>3</sub> etherate, 2 g of (IV), 5 ml of HSiEt<sub>3</sub>, and 0.2 g of Al<sub>2</sub>O<sub>3</sub> (II activity). The mixture was heated for 30 h at 45-50° and then evaporated in vacuo. The residue was treated with 7 ml of conc. HCl in 33 ml of alcohol, let stand for 12 h at  $\sim$ 20°, and then evaporated in vacuo. With stirring, the residue was treated with a solution of 3.3 g of KOH in 40 ml of alcohol and the mixture was let stand for 12 h at  $\sim 20^\circ$ . The alcohol was vacuum-distilled, and the residue was dissolved in 10 ml of water and, with cooling ( $0^{\circ}$ ), was acidified with dilute HCl solution (1:1) to pH  $\sim$ 1. The obtained precipitate was filtered, washed in succession with water and acetone, and dried in the air. We obtained 1.6 g (80%) of technical (VI), mp 226-227°. A mixture of 1.6 g of technical (VI) and 15 ml of 10% H<sub>2</sub>SO<sub>4</sub> solution was refluxed for 1 h. After cooling to  $\sim 20^\circ$  the precipitate was filtered, washed in succession with water and acetone, and dissolved in a solution of 2 g of Na2CO3 in 20 ml of water. To decolorize the obtained solution it was treated in portions with 2 g of resin AV-172 P, heated at the boil for 5 min, filtered hot, and the filtrate was cooled to ~20° and acidified with dilute HCl solution (1:1). The precipitate was filtered, washed in succession with water and acetone, and dried to constant weight at 100°. We obtained 0.96 g (50%) of pure (VI), mp 231-232° (from 80% alcohol). PMR spectrum (CF<sub>3</sub>COOH, δ, ppm): 1.30 m (CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 2.10 m (CH<sub>2</sub>CO), 2.48 m (CH<sub>2</sub>S), 2.95 m (CHS), 4.36 and 4.22 m (HC-CH). Rf 0.56 (EA-alcohol, 3:2, the spots were detected first with K<sub>3</sub>Fe(CN)<sub>6</sub> solution and then with iodine vapors). Mass spectrum, m/z: 244 (M<sup>+</sup>), 184, 112, 97, 85.

The authentic (VI) racemate had the same characteristics.

The paper is dedicated to Prof. E. Lederer (France) on the occasion of his 75th birthday.

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## CONCLUSIONS

Starting with the ethyl ester of N,N'-diacetyl- $\alpha$ -keto- $\alpha$ '-acetylmercaptodehydrodesthiobiotin, the three-step synthesis of racemic biotin was accomplished in 37% yield.

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