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Tetrahedron

Tetrahedron 61 (2005) 6610-6613

New pyrano[3,4-*b*]indoles from 2-hydroxymethylindole and L-dehydroascorbic acid

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Received 30 November 2004; revised 1 April 2005; accepted 21 April 2005

Available online 23 May 2005

Abstract—2-Hydroxymethylindole reacts with L-dehydroascorbic acid under mild conditions to give (3R,3aR,10cS)-3-[(1S)-1,2-dihydroxyethyl]-3a,10c-dihydroxy-3a,5,6,10c-tetrahydrofuro[3',4':5,6]pyrano[3,4-b]indol-1(3H)-one. Its tosyl derivative undergoes cyclization to form a pentacyclic ketal derivative.

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1. Introduction

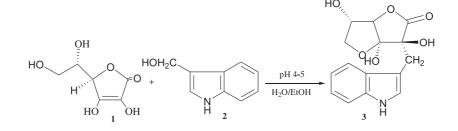
L-Ascorbic acid easily undergoes 2-C-alkylation by 3-hydroxymethylindole, 4-hydroxybenzyl alcohol, and their analogues, which are able to produce quinone methide type structures under mild acidic conditions.^{1,2} Ascorbigen, which is 2-C-[(indol-3-yl)methyl]- α -L-xylo-hex-3-ulofuranosono-4-lactone **3** is a product of the reaction of L-ascorbic acid **1** with the intermediate quinone methide formed from 3-hydroxymethylindole **2**.² Ascorbigen **3** is the main decomposition product of alkaloid glucobrassicin from cruciferous vegetables, and animals and humans receive it with meals. Recently, it was found that it is a strong non-specific immunomodulator with valuable properties (Scheme 1).³

Several ascorbigens have been synthesized in our laboratory

and their unique chemical and biological properties have been investigated,³ but the reaction of isomeric 2-hydroxymethylindole with L-ascorbic acid poses synthetic challenges and would help us to further elucidate the synthetic potential of this unique compound and its derivatives. The goal of our work was to study the preparation of a compound isomeric to ascorbigen by the reaction of 2-hydroxymethylindole **4** with L-ascorbic acid **1**.

2. Results and discussion

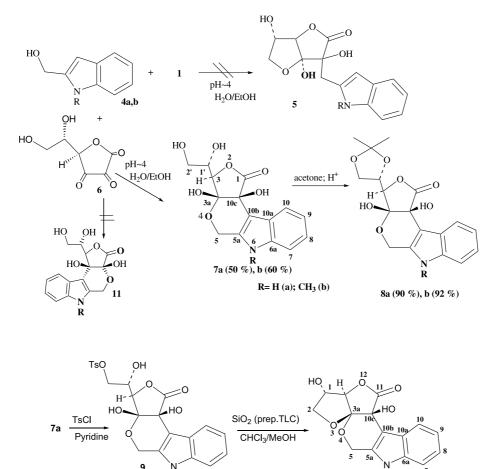
Unfortunately, 2-hydroxymethylindole 4a does not react with 1 with the formation of an ascorbigen-like structure (5) at pH 1–6. It can be explained by the inability of 4a to form a quinone methide structure under these conditions. However, on incubation of the reaction mixture (1+4a in



Scheme 1.

Keywords: L-Ascorbic acid; L-Dehydroascorbic acid; Ascorbigen; Indolopyrane; 2-Hydroxymethylindole. * Corresponding author. Tel.: +7 095 245 37 53; fax: +7 095 245 02 95; e-mail: mnp@space.ru

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Scheme 2.

Scheme 3.

H₂O–MeOH mixture at pH ~ 4.7) for more than two weeks, we succeeded in the isolation of a new product in 20% yield. It was identified as (3R,3aR,10cS)-3-[(1S)-1,2-dihydroxyethyl]-3a,10c-dihydroxy-3a,5,6,10c-tetrahydrofuro[3',4': 5,6]pyrano[3,4-*b*]indol-1(3*H*)-one **7a** (Scheme 2).

It is suggested that 7a is a product of the reaction of L-dehydroascorbic acid 6 with 4a. When L-ascorbic acid was oxidized to L-dehydroascorbic acid by the action of oxygen in the presence of activated charcoal in $H_2O-MeOH^4$ before addition of 4a, the yield of compound 7a increased to 50% and the reaction time was reduced to seven days. Similarly, compound 7b was obtained in 60% yield from 1-methyl-2-hydroxymethyindole 4b and L-dehydroascobic acid 6. It suggests that the nucleophilic attack of 3-C indole atom on carbonyl group at 2-C of dehydroascorbic acid takes place followed by hemiacetal formation with the participation of 3-CO of dehydroascorbic acid and hydroxy group of 4. The reaction of 7a or 7b with acetone in the presence of *p*-toluenesulfonic acid gave crystalline 1', 2'-Oisopropylidene derivatives 8a or 8b, respectively. By the reaction of 7a with an excess of *p*-toluenesulfonyl chloride in pyridine 2'-O-monotosyl derivative **9** was obtained. On purification by preparative TLC, 9 partially underwent cyclization into acetal 10 (Scheme 3).

The stereochemistry of compounds 7a,b was suggested by NOE experiments and appears to be in accord with the

previously observed dominant mode of alkylation in such systems.^{1,3,5} An isomeric structure with a different type of annelation of the pyranoindole and furanone rings than in compound **7** could be proposed for the product of reaction of **4** with **6**. A double resonance experiment allowed the identification in the ¹H NMR spectrum of **8a** of the signal of the OH-group (10c-OH), which is nearest to the carbonyl group. In the ¹H coupled ¹³C NMR spectrum of compound **8a** a coupling constant ³*J*_{1C,10c-OH} ≈ 4 Hz was observed. The NOE for this 10c-OH proton was determined (η_{10c-OH} {10-H} ≈ 3.5%) by saturation of 10-H. An analogous effect was observed in the reverse experiment (η_{10-H} {10c-OH}) in accordance with structure **7** and **8**.

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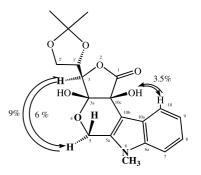


Figure 1. NOE experiments for 8b.

The reciprocal orientation of two hydroxyl groups (10c-OH and 3a-OH) was elucidated based on rather high NOE values $\eta_{3-H}{5-Ha} \approx 9\%$ and $\eta_{5-Ha}{3-H} \approx 6\%$), in compound **8b** 5-Ha is the highfield doublet of AB system 5-H₂ (Fig. 1). Molecular modeling of compounds **7** and **8** led to a conclusion, that the distance between atoms 5-Ha and 3-H may be <2 Å only in the structure in, which the two hydroxyl groups in the tetrahydrofuranone cycle are *cis*-oriented.

The S-configuration at 10c atom follows from analysis of relevant literature, as there are no examples of attack on the 2C atom of L-ascorbic acid from the side *cis* to the bulky substituent at C-4 (CHOH–CH₂OH).³

The compounds obtained have a structural fragment of 1,3,4,9-tetrahydropyrano[3,4-b]indole, which represents a framework of some anti-inflammatory drugs such as etodolac.⁶

3. Experimental

3.1. General

NMR spectra were recorded on a Varian Unity +400 instrument (400 MHz for ¹H, 100.6 MHz for ¹³C). NOE values (η_{Hi} {H_j, %) were measured as an increase of H_i signal intensity when H_j signal was saturated. Analytical TLC was performed on Kieselgel F₂₅₄ plates (Merck), preparative TLC chromatography on plates (20×20 cm, 0.5 mm) with Kieselgel 60 F₂₅₄ (Merck), and column chromatography on Kieselgel 60 (Merck), using the following systems of solvents: CHCl₃/MeOH 5:1 (A); 10:1 (B) and EtOAc/petroleum ester, 1:1 (C). Optical rotations were measured on a Perkin-Elmer 241 polarimeter. Infrared spectra were recorded with a Nicolet Avatar 330 FT-IR spectrometer using KBr discs.

High resolution mass spectra were registered on a MAT 8430 Finnigan instrument (USA) with data operating system SS-300 (EI, 70 eV, direct introduction, temperature of ion source 250 °C). Electron impact (EI) mass-spectra were registered on a SSQ 710 Finnigan MAT instrument (USA), (EI: 70 eV, direct introduction). Melting points were determined on a Buchi SMP-20 apparatus and are uncorrected.

3.1.1. (*3R*,3*aR*,10*cS*)-3-[(1*S*)-1,2-Dihydroxyethyl]-3a,10cdihydroxy-3a,5,6,10c-tetrahydrofuro[3',4':5,6]pyrano[3,4-*b*]indol-1(3*H*)-one (7a). To a solution of 4a (1.2 g, 7.36 mmol) in CH₃OH (10 mL) was added buffer solution prepared by dissolving of 2.7 g of citric acid and 6 g of Na₂HPO₄ in 300 mL of H₂O and solution of L-dehydroascorbic acid (6), obtained from 1 (6.5 g, 36.8 mmol) in 100 mL of CH₃OH. The reaction mixture was stirred for seven days at rt, saturated with NaCl, extracted with CHCl₃ (1×50 mL) and then with EtOAc (3×40 mL). EtOAc extract was dried (Na₂SO₄), evaporated and after column chromatography (A) gave 7a (1.2 g, 50%) as a light-brown powder (as minimum 96% purity by NMR data); R_f 0.26 (A); $[\alpha]_D^{20}$ +4 (MeOH); ν_{max} : 3266, 1767, 1455, 1104, 741 cm⁻¹; HRMS Calcd for C₁₅H₁₅NO₇: 321.0849. Found: 321.0830; ¹H NMR: δ (DMSO- d_6) 3.54 (m, 2H, 2'-H_AH_B); 3.91 (m, 1H, 1'-H); 4.52 (d, J=3.8 Hz, 1H, 3-H); 4.82 (d, J=15.4 Hz, 1H, 5-H_B); 4.90 (br s, 1H, 2'-OH); 4.96 (d, J=15.4 Hz, 1H, 5-H_A); 5.97 (br s, 1H, 1'-OH); 6.50 (br s, 1H, 10-OH); 7.01 (ddd, J=7.8, 7.0, 1.0 Hz, 1H, 9-H); 7.08 (ddd, J=8.1, 7.0, 1.2 Hz, 1H, 8-H); 7.19 (br s, 1H, 3a-OH); 7.35 (ddd, J=8.1, 1.0, 0.7 Hz, 1H, 7-H); 7.73 (ddd, J=7.8, 1.2, 0.7 Hz, 1H, 10-H); 11.14 (s, 1H, 6-H); ¹³C NMR: δ (DMSO- d_6) 58.0 (5-C); 61.5 (2'-C); 68.8 (1'-C); 70.9 (10c-C); 81.2 (3-C); 99.4 (3a-C); 104.1 (10b-C); 111.4 (7-C); 119.2 (9-C); 120.6 (10-C); 121.4 (8-C); 125.8 (10a-C); 134.3 (5a-C); 136.5 (6a-C); 174.3 (1-C).

3.1.2. (3R,3aR,10cS)-3-[(1S)-1,2-Dihydroxyethyl]-3a,10cdihydroxy-6-methyl-3a,5,6,10c-tetrahydrofuro[3',4': 5,6]pyrano[3,4-b]indol-1(3H)-one (7b). Compound 7b was obtained from 4b (1 g, 6.25 mmol) and purified by the same way as 7a, as a light-brown powder (1.25 g, 60%); $R_{\rm f}$ 0.44 (A); $[\alpha]_{\rm D}^{20}$ +6.5 (MeOH); $\nu_{\rm max}$: 3368, 1786, 1455, 1031, 759 cm⁻¹; HRMS Calcd for C₁₆H₁₇NO₇: 335.1005. Found: 335.0998; ¹H NMR: δ (CD₃OD) 3.64 (s, 3H, $6-CH_3$; 3.81 (m, 2H, 2'-H_AH_B); 4.08 (m, 1H, 1'-H); 4.63 (d, J = 5.3 Hz, 1H, 3-H); 5.04 (d, J = 15.6 Hz, 1H, 5-H_B); 5.13 $(d, J=15.6 \text{ Hz}, 1\text{H}, 5\text{-H}_{A}); 7.10 (ddd, J=7.9, 7.1, 1.0 \text{ Hz}, 1.0 \text{ Hz}); 7.10 (ddd, J=7.9, 7.1, 1.0 \text{ Hz}); 7.10 (ddd, J=$ 1H, 9-H); 7.19 (ddd, J=8.3, 7.1, 1.2 Hz, 1H, 8-H); 7.37 (ddd, J=8.3, 1.0, 0.7 Hz, 1H, 7-H); 7.88 (ddd, J=7.9, 1.2, 1.2)0.7 Hz, 1H, 10-H); ¹³C NMR: δ (CD₃OD) 29.9 (N–CH₃); 59.8 (5-CH₂); 63.3 (2'-C); 70.9 (1'-C); 72.5 (10c-C); 81.2 (3-C); 100.9 (3a-C); 104.3 (10b-C); 110.1 (7-C); 121.0 (9-C); 121.6 (10-C); 123.0 (8-C); 126.7 (10a-C); 136.3 (5a-C); 139.8 (6a-C); 175.9 (1-C).

3.1.3. (3R,3aR,10cS)-3-[(4S)-2,2-Dimethyl-1,3-dioxolan-4-yl]-3a,10c-dihydroxy-3a,5,6,10c-tetrahydrofuro[3', 4':5,6]pyrano[3,4-b]indol-1(3H)-one (8a). To a solution of 7a (500 mg, 1.56 mmol) in dry acetone (10 mL) was added p-toluenesulfonic acid (10 mg). The reaction mixture was stirred for 40 min at rt, diluted with 100 mL of 10% NaHCO₃ solution, extracted with EtOAc $(2 \times 30 \text{ mL})$. After drying (Na₂SO₄) of the extract, evaporating, and recrystallization from acetone 8a (500 mg, 90%) was obtained as colorless crystals. Mp 190–196 °C (decomp.); $R_{\rm f} 0.16$ (C); $[\alpha]_{\rm D}^{20}$ +6.7 (MeOH); $\nu_{\rm max}$: 3329, 1785, 1085, 1059, 753 cm⁻¹; HRMS Calcd for C₁₈H₁₉NO₇: 361.1161. Found: 361.1143; ¹H NMR: δ (DMSO- d_6) 1.29 and 1.33 $(2 \text{ s}, 2 \times 3\text{H}, 1', 2' - OC(CH_3)_2); 3.90 \text{ (dd}, J = 8.9, 6.5 \text{ Hz}, 1\text{H},$ $2'-H_{\rm B}$); 4.14 (dd, J=8.9, 6.3 Hz, 1H, $2'-H_{\rm A}$); 4.34 (m, 1H, 1'-H); 4.39 (dd, J=8.0, 1.0 Hz, 1H, 3-H); 4.96 (d, J=15.8 Hz, 1H, 5-H_B); 5.02 (d, J = 15.8 Hz, 1H, 5-H_A); 6.44 (s, 1H, 10c-OH); 6.96 (d, J = 1.0 Hz, 1H, 3a-OH); 7.02 (ddd, J=7.9, 7.2, 1.1 Hz, 1H, 9-H); 7.10 (ddd, J=8.1, 7.2,1.3 Hz, 1H, 8-H); 7.36 (ddd, *J*=8.1, 1.1, 0.8 Hz, 1H, 7-H); 7.72 (ddd, J=7.9, 1.3, 0.8 Hz, 1H, 10-H); 11.26 (s, 1H, 6-H); ¹³C NMR: δ (DMSO- d_6) 25.5 and 26.6 (1',2'-OC(CH₃)₂); 59.6 (5-C); 65.1 (2'-C); 70.7 (10c-C); 74.7 (1'-C); 79.3 (3-C); 99.7 (3a-C); 104.0 (10b-C); 108.8 (1',2'-OC(CH₃)₂); 111.5 (7-C); 119.4 (9-C); 120.2 (10-C); 121.8 (8-C); 125.4 (10a-C); 134.0 (5a-C); 136.8 (6a-C); 173.3 (1-C).

3.1.4. (3*R*,3a*R*,10cS)-3-[(4S)-2,2-Dimethyl-1,3-dioxolan-4-yl]-3a,10c-dihydroxy-6-methyl-3a,5,6,10c-tetrahydrofuro[3',4':5,6]pyrano[3,4-b]indol-1(3H)-one (8b). Compound **8b** was obtained and purified by the same way as **8a** from **7b** (500 mg, 1.5 mmol) as colorless crystals (517 mg, 92%). Mp 161–165 °C (decomp.); $R_f 0.19$ (C); $[\alpha]_D^{20} + 5$ (c 2.5, MeOH); HRMS Calcd for C₁₉H₂₁NO₇: 375.1318. Found: 375.1324; ¹H NMR: δ (DMSO- d_6) 1.29 and 1.33 $(2 \text{ s}, 2 \times 3\text{H}, 1', 2' - OC(CH_3)_2); 3.62 \text{ (s}, 3\text{H}, 6 - CH_3); 3.88 \text{ (dd,})$ J=8.8, 6.8 Hz, 1H, 2'-H_B,); 4.15 (dd, J=8.8, 6.5 Hz, 1H, $2'-H_A$; 4.32 (m, 1H, 1'-H); 4.43 (dd, J=8.0, 1.2 Hz, 1H, 3-H); 5.04 (d, J = 15.8 Hz, 1H, 5-H_B); 5.11 (d, J = 15.8 Hz, 1H, 5-H_A); 6.47 (s, 1H, 10c-OH); 6.98 (d, J=1.2 Hz, 1H, 3a-OH); 7.06 (ddd, J=8.0, 7.1, 1.0 Hz, 1H, 9-H); 7.17 (ddd, J=8.2, 7.1, 1.2 Hz, 1H, 8-H); 7.48 (ddd, J=8.2, 1.0,0.8 Hz, 1H, 7-H); 7.73 (ddd, *J*=8.0, 1.2, 0.8 Hz, 1H, 10-H); 11.26 (s, 1H, 6-H); ¹³C NMR: δ (DMSO- d_6) 25.4 and 26.5 $(1',2'-OC(CH_3)_2)$; 29.7 (N-CH₃); 59.1 (5-C); 65.0 (2'-C); 70.5 (10c-C); 74.7 (1'-C); 79.0 (3-C); 99.8 (3a-C); 103.3 $(10b-C); 108.8 (1', 2'-OC(CH_3)_2); 109.6 (7-C); 119.6 (9-C);$ 120.2 (10-C); 121.6 (8-C); 125.0 (10a-C); 135.2 (5a-C); 137.5 (6a-C); 173.1 (1-C).

3.1.5. (3R,3aR,10cS)-3a,10c-Dihydroxy-1-oxo-1,3,3a,5, 6,10c-hexahydrofuro[3',4':5,6]pyrano[3,4-b]indol-3-yl]-2-hydroxyethyl tosylate (9), and (1S,3aS,10cS,12aR)-1,10c-dihydroxy-1,6,10c,12a-tetrahydro-2*H*-furo[2", 3":4',5']furo[3',4':5,6]pyrano[3,4-b]indol-11(5*H*)-one (10). To a stirred solution of 7a (500 mg, 1.56 mmol) in dry pyridine (10 mL) was added tosylchloride (305 mg, 1.6 mmol) and after 4 h incubation the reaction mixture was diluted by citric acid solution and extracted by EtOAc (2×30 mL). After evaporation 600 mg of brown oil was obtained. It consisted of 80% of 9 and not contained 10. After preparative TLC chromatography (system B) 60 mg of 9 (8% yield,) and 300 mg of 10 (63% yield) were obtained. The latter was formed during chromatography.

Compound **9**. Colourless oil; $R_f 0.52$ (B); HRMS Calcd for $C_{22}H_{21}NO_9S$: 475.0937. Found: 475.0930; ¹H NMR: δ (DMSO- d_6) 2.48 (s, 3H, CH₃ of Ts); 4.04 (m, 1H, 1'-OH); 4.16 (dd, J=9.6, 6.4 Hz, 1H, 2'-H_A); 4.31 (dd, J=9.6, 3.2 Hz, 1H, 2'-H_B); 4.38 (d, J=8.1 Hz, 1H, 3-H); 4.97 (d, J=15.1 Hz, 5-H_A); 5.00 (d, J=15.1 Hz, 5-H_B); 5.98 (d, J=6.4 Hz, 1H, 1'-OH); 6.52 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_B); 4.97 (dd, J=15.1 Hz, 5-H_A); 6.92 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_A); 6.92 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_A); 6.92 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_A); 6.92 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_A); 6.92 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_A); 6.92 (s, 1H, 10c-OH); 7.02 (ddd, J=15.1 Hz, 5-H_A); 7.02 (ddd, J=15.1 Hz, 7-H_A); 7.02 (dd, J=15.1 Hz, 7-H_A);

8.1, 7.0, 1.0 Hz, 1H, 9-H); 7.06 (s, 1H, 3a-OH); 7.12 (ddd, J=8.1, 7.0, 1.2 Hz, 1H, 8-H); 7.38 (ddd, J=8.1, 1.0, 0.8 Hz, 1H, 7-H); 7.52 (d, J=9.6 Hz, 2H, Ts) 7.70 (ddd, J=8.0, 1.2, 0.8 Hz, 1H, 10-H); 7.84 (d, J=9.6 Hz, 2H, Ts); 11.25 (s, 1H, 6-H).

Compound **10**. Colourless crystals. Mp 201–205 °C; R_f 0.61 (A); ν_{max} : 3416, 1778, 1477, 1087, 1038, 747 cm⁻¹; HRMS Calcd for C₁₅H₁₃NO₆: 303.0743. Found: 303.0738; ¹H NMR: δ (DMSO- d_6 -benzene- d_6 , 3:1) 4.26 (dd, J=9.3, 4.1 Hz, 1H, 2-H_A); 4.52 (dd, J=9.3, 6.1 Hz, 1H, 2-H_B); 4.60 (dd, J=6.1, 4.1 Hz, 1H, 1-H); 5.06 (s, 1H, 12a-H); 5.20 (d, J=15.6 Hz, 1H, 5-H_A); 5.31 (d, J=15.6 Hz, 1H, 5-H_B); 6.05 (br s, 1H, 1-OH); 6.25 (s, 1H, 10c-OH); 7.15 (t, J= 7.5 Hz, 1H, 9-H); 7.22 (t, J=7.5 Hz, 1H, 8-H); 7.53 (d, J= 7.9 Hz, 1H, 7-H); 8.05 (d, J=7.9 Hz, 1H, 10-H); 11.50 (s, 1H, 6-H); ¹³C NMR: δ (DMSO- d_6 -benzene- d_6 , 3:1) 61.4 (2'-C); 70.5 (10c-C); 73.6 (1'-C); 75.9 (5-C); 82.1 (12a-C); 104.9 (10b-C); 110.2 (3a-C); 111.5 (7-C); 119.5 (9-C); 120.1 (10-C); 121.8 (8-C); 124.9 (10a-C); 134.1 (5a-C); 136.7 (6a-C); 172.9 (11-C).

Acknowledgements

This study was supported by Russian Foundation for Basic Research, grant No. 03-03-32090.

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