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In vitro inhibition effect and structure–activity relationships of some saccharin derivatives on erythrocyte carbonic anhydrase I and II

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Abstract

In this study, *in vitro* inhibitory effects of some saccharin derivatives on purified carbonic anhydrase I and II were investigated using CO₂ as a substrate. The results showed that all compounds inhibited the hCA I and hCA II enzyme activities. Among the compounds, 6-(*p*-tolylthiourenyl) saccharin (**6m**) was found to be the most active one for hCA I activity (IC₅₀ = 13.67 μ M) and 6-(*m*-methoxyphenylurenyl) saccharin (**6b**) was found to be the most active one for hCA II activity (IC₅₀ = 6.54 μ M). Structure-activity relationships (SARs) study showed that, generally, thiourea derivatives (**6I–v**) inhibited more hCA I and hCA II than urea derivatives (**6a–k**). All compounds (excluding **6c** and **6r**) have higher inhibitory activity on hCA II than on hCA I.

Keywords

Carbonic anhydrase, inhibition, saccharin, thiourea, urea

History

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Introduction

Carbonic anhydrase (CA, EC 4.2.1.1) is a ubiquitous zinc enzyme. Basically, there are several cytosolic forms (CA-I, CA-II, CA-III and CA-VII), four membrane-bound forms (CA-IV, CA-IX, CA-XII and CA-XIV), one mitochondrial form (CA-V), as well as a secreted CA form (CA-VI)^{1,2}. They all catalyze a very simple physiological reaction, the interconversion between carbon dioxide and the bicarbonate ion, and are thus involved in crucial physiological processes connected with respiration and transport of CO₂/ bicarbonate between metabolizing tissues and the lungs, pH and CO₂ homeostasis, electrolyte secretion in a variety of tissues/ organs, biosynthetic reactions (such as the gluconeogenesis, lipogenesis and ureagenesis), bone resorption, calcification, tumorigenicity and many other physiologic or pathologic processes¹⁻³. CA inhibitors have now been a mainstay of human clinical intervention for several decades, with at least 25 clinically used drugs that are CA inhibitors⁴. Although there are many studies on this enzyme, the CA enzyme family continues to capture the attention of drug discovery scientists and clinicians as the knowledge regarding the therapeutic implications associated with this enzyme class continues to grow^{4,5}.

Saccharin, 1,2-benzisothiazole-3-one-1,1-dioxide, is a wellknown heterocyclic compound and has been used as a sweetener in the form of its sodium salt since 1885. Yet it is also a heterocycle of pharmaceutical importance, being a key structural element of certain CNS-active drugs⁶. Chemically, saccharin consists of a sulfimide with a lactam and cyclic sulfonamide moiety. The latter functionality is responsible for the acidic character of the molecule and suggests its potential to interact with the zinc ion at the floor of the binding pocket of carbonic anhydrases. Köhler et al. presented saccharin as zinc-binding portion based on the X-ray structure⁷.

Sulfonamides are the best known inhibitors of CA enzymes and are used for the treatment of glaucoma in medicinal chemistry⁸. Acetazolamide (AAZ), dorzolamide (DZA) and brinzolamide (BRZ) are sulfonamide derivatives and are used in the treatment of glaucoma. However, these drugs have several side effects such as numbness and tingling in the fingers and toes, blurred vision, kidney stones, an increase in urination, upset stomach, dry eye and headache or dizziness^{2,9}. There are reports^{10,11} of many such aromatic sulfonamides or

There are reports^{10,11} of many such aromatic sulfonamides or bis-sulfonamide moieties incorporating urea or thiourea groups as very potent inhibitors against three isozymes, human CA I, human CA II and bovine isozyme CA. A small series of five ureidosubstituted benzenesulfonamide derivatives were recently investigated as inhibitors of the cytosolic isoform hCA II by one of these groups. It has been observed that their potency varied between 3.3 and 226 nM, and by means of X-ray crystallography a highly variable orientation of the R-ureido moieties was evidenced when the inhibitor was bound within the enzyme active site¹¹.

In this study, we evaluated 6-(phenylurenyl/thiourenyl) saccharin derivatives (6a-v), synthesized in the previous work¹², effects on hCA I and hCA II purified from human erythrocytes. Additionally, we presented SAR analyses.

Materials and methods

General

Sepharose 4B, L-tyrosine, sulfonamide, synthetic starting material, reagents and solvents were purchased from Merck (Darmstadt, Germany), Alfa Easer (Ward Hill, MA), Sigma-Aldrich (Taufkirchen, Germany) and Fluka (Taufkirchen, Germany).

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General procedure for 6-(phenylurenyl/thiourenyl) saccharin derivatives

Phenylisocyanate or phenylisothiocyanate derivatives (1 mmol) were added to a solution of 6-aminosaccharin (1 mmol) and triethyl amine (1 mL) in dry DMF. The mixture was stirred at room temperature for 12h and then poured into cold 1 M HCl. The precipitate was filtered and washed with cold water. The crude products were recrystallized from ethanol over 99% purity. The synthetic procedures are depicted in Scheme 1.

Preparation of hemolysate and purification from blood red cells

Preparation of hemolysate and purification from blood red cells made by the literature¹³ was presented in supporting information.

CA enzyme assay

CA activity measured by the Maren method¹⁴ was presented in supporting information.

In vitro inhibition studies

For the inhibition studies of saccharin, different concentrations of these compounds were added to the enzyme. Activity percentage values of CA for different concentrations of each saccharin were determined by regression analysis using Microsoft Office 2000 Excel. CA enzyme activity without a saccharin solution was accepted as 100% activity.

Results and discussion

Chemistry

Х

R

Х

R

Х

R

The synthetic procedures are depicted in Scheme 1 6-(Phenylurenyl/thiourenyl) saccharin compounds (6a-v) were synthesized from 4-nitrotoluene (1) in five steps by known procedures¹².

Biological evaluation of saccharin derivatives for hCA I and hCA II inhibitory activities

For evaluating the hCA I and hCA II inhibitory effects, all compounds were subjected to hCA I and hCA II inhibition assay with CO₂ as a substrate. The result showed that all compounds (6a-v) inhibited the hCA I and hCA II enzyme activity.

The IC₅₀ values and inhibition constants of 6a-v analogues against hCA I and hCA II are summarized in Table 1 and the IC₅₀ graphs are given in Figure 1. The IC₅₀ figures are presented as supporting information.

We have determined the IC₅₀ values of 13.57-74.90 µM for the inhibition of hCA I and $6.54-49.00 \,\mu\text{M}$ for the inhibition of hCA II. Among all compounds, **6m** (IC₅₀ = 13.67 μ M) was found to be the most active one for hCA I inhibitory activity and 6b $(IC_{50} = 6.54 \,\mu\text{M})$ showed the highest hCA II inhibitory activity. 6d (IC₅₀ = $30.81 \,\mu\text{M}$) was found to be the most active one for hCA I inhibitory activity and **6b** (IC₅₀ = $6.54 \,\mu$ M) showed the highest hCA II inhibitory activity for the urea derivatives. Among the thiourea derivatives, **6m** (IC₅₀ = 13.67 μ M) showed the highest hCA I inhibitory activity and 6q (IC₅₀ = 8.10 µM) showed the highest hCA II inhibitory activity.

It was reported⁷ that saccharin most likely coordinates in a deprotonated state through its nitrogen atom to the catalytically active zinc ion. Additionally, ureido-substituted benzenesulfonamide moieties were evidenced when the inhibitor was bound within the enzyme active site^{10,11}. We believe that the synthesized saccharin urea/thiourea derivatives inhibited hCA I and II in the same way.

Structure-activity relationships

Generally, we have seen that all compounds (excluding 6c and 6r) have a higher inhibitory activity on hCA II than hCA I in the SARs study. When same substituents bonded to phenyl ring, most of the thiourea derivatives (61-v) exhibited higher hCA I and hCA

CH3 NH_O CISO₃H $NH_{3(aq)}$, ether CrO₃/H₂SO₄ Pd-C. EtOH 60 °C, 20h 40 °C , 2h rt, 24h NO₂ 'nо, ΝO, 2h, reflux NH2 ŃΟ₂ 1 2 з 5 6a 6b 6c 6d 6e 6f 6g 6h N=C=X 0 0 0 0 0 0 0 0 X=O,S DMF, TEA. Н 3-0CH3 4-OCH₃ 4-CH₃ 3-Cl 4-Cl 3,4-di-Cl 3-NO₂ 12 h, rt **6i** 6k 61 6j 6m 6n 60 6p NH 0 0 0 S S S S S 3-F 4-NO₂ 2-F 4-F Н 4-CH₃ 4-OCH₃ 2-F 6q 6r 6s 6t 6u 6v || X 6a-v S S S S S S 4-F 3-I 3-Cl 2,4-di-Cl 3,5-di-Cl 4-NO₂



Compound	Х	R	hCA I IC ₅₀ (µM)	hCA II IC ₅₀ (µM)	Compound	Х	R	hCA I IC ₅₀ (µM)	hCA II IC ₅₀ (µM)
5	_	_	54.90	49.00	61	S	Н	46.74	40.24
6a	0	Η	63.06	21.13	6m	S	4-CH ₃	13.67	11.14
6b	0	3-OCH ₃	41.65	6.54	6n	S	4-OCH ₃	43.31	30.32
6c	0	4-OCH ₃	34.55	37.66	60	S	2-F	74.90	30.23
6d	0	4-CH ₃	30.81	24.65	6р	S	3-F	46.70	14.55
6e	0	3-C1	38.99	17.79	6q	S	4-F	55.87	8.10
6f	0	4-C1	63.03	23.67	6r	S	3-I	20.21	30.70
6g	0	3,4-di-Cl	37.37	12.67	6 s	S	3-C1	33.41	17.19
6h	0	3-NO ₂	37.47	32.74	6t	S	2,4-di-Cl	59.38	12.48
6i	0	$4-NO_2$	54.24	38.87	6u	S	3,5-di-Cl	26.10	11.04
6j	0	2-F	44.71	41.61	6v	S	$4-NO_2$	47.73	43.49
6k	0	4-F	42.64	17.14			_		



Figure 1. IC $_{50}$ graphics of saccharin derivatives (5, 6a–v) on hCA I and hCA II.





(a) For urea derivatives:

(i) Although electron-withdrawing groups (nitro and halogens) bonded to *meta* position of the phenyl ring (**6e**, **6g** and **6h**) increased the inhibitory activity on hCA I, electron-donating groups (methoxy)

bonded to *meta* position of phenyl ring (**6b**) had the highest hCA II inhibitory activity ($IC_{50} = 6.54 \mu M$).

(ii) Electron-donating groups (methoxy, methyl) bonded to *para* position of the phenyl ring (**6c** and **6d**) inhibited hCA I activity more than halogens and electron-withdrawing groups bonded. On the other hand, halogen groups bonded to the *para* position of



the phenyl ring $(\mathbf{6f}, \mathbf{6g} \text{ and } \mathbf{6k})$ increased the inhibitory activity on hCA II.

(b) For thiourea derivatives:

- (i) Electron-donating groups bonded to the *para* position of the phenyl ring (**6m** and **6n**) increased the hCA I inhibitory activity.
- (ii) Moving-F group on the phenyl ring from *ortho* (**60**, $IC_{50} = 30.23 \,\mu\text{M}$) to *meta* (**6p**, $IC_{50} = 14.55 \,\mu\text{M}$) and

para (6q, $IC_{50} = 8.10 \,\mu\text{M}$) positions led to major enhancement of hCA II inhibitory activity.

- (iii) In same moving for hCA I activity, **6p** $(IC_{50} = 46.70 \,\mu\text{M})$ was more inhibited than **6o** $(IC_{50} = 74.90 \,\mu\text{M})$ and **6q** $(IC_{50} = 55.87 \,\mu\text{M})$.
- (iv) Halogen series on the *meta* position of the phenyl ring showed a linear relationship for higher hCA I inhibitory activity with increasing size and



Figure 1. Continued.

polarizability (for size and polarizability, I > Cl > F, hCA Ι inhibitory activity, for 6r $(IC_{50} = 20.21 \,\mu M) > 6s$ $(IC_{50} = 33.41 \,\mu M) > 6p$ $(IC_{50} = 46.70 \,\mu\text{M})$. Interestingly, this series showed an inverse relationship for hCA II inhibitory activity with increasing size and polarizability (for hCA II inhibitory activity, 6r $(IC_{50} = 30.70 \,\mu M) < 6s$ $(IC_{50} = 17.19 \,\mu\text{M}) < 6p (IC_{50} = 14.55 \,\mu\text{M}).$

Conclusions

In conclusion, we evaluated 6-(phenylurenyl/thiourenyl) saccharin derivatives (**6a–v**) effects on hCA I and hCA II purified from human erythrocytes and SARs were examined. All compounds inhibited both hCA I and hCA II enzyme activities. Most of the compounds had higher hCA II inhibitory activity than hCA I activity. Most of the thiourea derivatives (**61–v**) exhibited higher hCA I and hCA II inhibitory activities than urea derivatives (**6a–k**). The present study revealed that activity could also be influenced by the type and position of the substituent on the phenyl ring. Among all compounds, **6m** showed the highest hCA II inhibitory activity and **6b** showed the highest hCA II inhibitory activity.

In summary, enzyme inhibition is the most important issue for drug design and biochemical applications^{15–27}. Therefore, our results suggested that saccharin derivatives are likely to be adopted as candidates to treat glaucoma and may be taken for further evaluation in *in vivo* studies.

Declaration of interest

The authors report no conflicts of interest.

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