

INTRAMOLECULAR DIELS-ALDER REACTION WITH FURAN-DIENE.
TOTAL SYNTHESIS OF (\pm)-11-KETOTESTOSTERONE AND (\pm)-ADRENOSTERONE.

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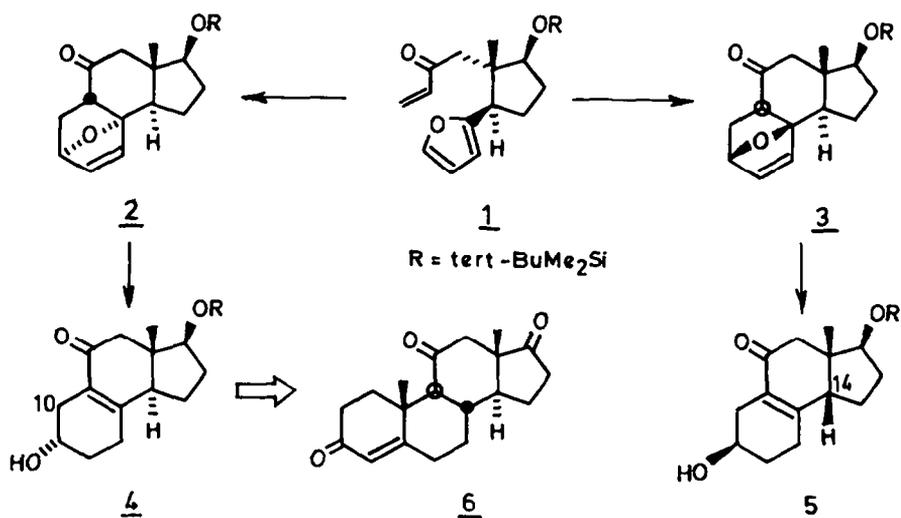
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ABSTRACT

A novel D \rightarrow BCD \rightarrow ABCD route to 11-keto steroids is reported involving a high yield stereoselective intramolecular Diels-Alder reaction of furan-diene **12** in water as a key-step. The dienophilic side chain is readily introduced starting from **8** via a sequence involving alkylation with ethyl (E)-3-ethoxy-4-iodo-2-butenate, reduction and acid hydrolysis. The reduced adduct **14** is further converted into (\pm)-adrenosterone (**6**) via **24**, the dienediolate equivalent of which is a known intermediate in corticosteroid synthesis.

Steroids, and in particular corticosteroids, are challenging targets for total synthesis, not only due to their biological importance, but also because they constitute a rich area for the development of novel strategies in synthesis^{1,2}. In a previous report we have described a novel approach to the BCD-ring system of 11-keto steroids based on the intramolecular Diels-Alder reaction of furan-enone **1** (scheme 1)³. The following conclusions could be drawn from this exploratory study. Among the investigated dienophilic side chains the unsubstituted enone **1** is, at the same time, a suitable substrate for cycloaddition and of further potential interest for corticosteroid synthesis. Depending on the reaction conditions either adduct **2** (kinetic control) or **3** (thermodynamic control) is formed predominantly, albeit in low yield (~40-50 %). After the hydrogenation of the double bond in **2**, the opening of the oxygen bridge is readily effected with base yielding enone **4**; however, the more forceful conditions required for the opening of the product obtained by hydrogenation of adduct **3** cause epimerization at C-14 leading to the *cis*-fused enone **5**.

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SCHEME 1

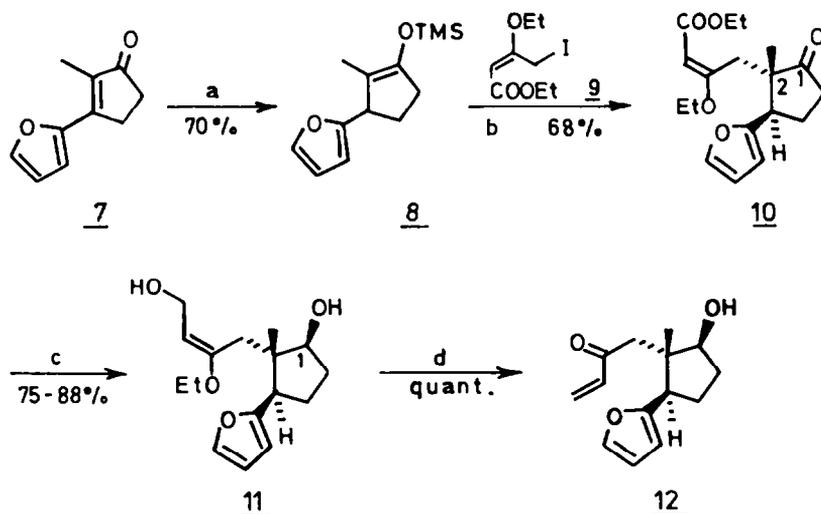
In view of the above results an efficient synthesis of 11-keto steroids along the proposed Diels-Alder strategy still requires: (1) the development of a short and direct route to enone 1; (2) the finding of appropriate conditions upon which the Diels-Alder reaction would proceed stereoselectively and in high yield to 2; (3) the stereoselective appendage of the A-ring with an angular methyl group at C-10. In the present paper we report a solution to these problems which eventually led to a 16-step synthesis of (\pm)-adrenosterone (6)^{4,5}.

The synthesis of enone 12 (scheme 2).

Our previous synthesis of 1 required six steps from the trimethylsilyl enol ether 8, which is readily available from cyclopentenone 7 in 69% yield via reduction with lithium in liquid ammonia followed by trapping with trimethylsilyl chloride³. Not unexpectedly, the direct introduction of the 2-oxo-3-butenyl side chain by alkylation of the enolate anion derived from 8 (methylolithium) failed; various experiments with 1-iodo-3-buten-2-one⁶ in tetrahydrofuran involving variations in temperature, the presence of hexamethylphosphoramide and the mode of addition only led to complex reaction mixtures. In view of the kinetic preference of β -alkoxy allylic alcohols (as in 11) to yield the corresponding enone in acid⁷, we decided to use a 3-alkoxy-4-halo-2-butenate ester as the electrophile⁸ for the introduction of the enone side chain in masked form. Thus, alkylation of the enolate anion derived from 8 (methylolithium) with ethyl (*E*)-3-ethoxy-4-iodo-2-butenate (9)⁹ in tetrahydrofuran-hexamethylphosphoramide gave the desired *trans*-ketone 10 in 68% yield next to 13% of 2,5-dialkylated product (undefined stereochemistry at C-5). The *trans*-stereochemistry of 10 is inferred from the shielded angular methyl resonance (*cis* with furyl group) in the ¹H NMR spectrum (δ 0.75); no *cis*-product was isolated in this experiment¹⁰. We should note here that the double bond configuration in 10 and 11, which

eventually originates from ethyl (*E*)-3-ethoxy-2-butenate^{9c}, has been misassigned in our preliminary report^{4,11}.

Reduction of keto-ester **10** with lithium aluminum hydride in ether and alkaline work-up¹² gave diol **11** as a single diastereomer¹⁰ in yields ranging from 75 to 88 % depending on the scale. The assigned β -orientation of the hydroxyl group at C-1 is in accord with the observed resonance for H-1 which exhibits a triplet pattern with a coupling constant value of 9.5 Hz^{3b}. Further exposure of **11** to Danishefsky's dilute acid conditions (0.005 N hydrochloric acid-tetrahydrofuran, 1:4)¹³ led quantitatively to enone **12** which is further used without purification. Alternatively, **12** can be produced in two steps from cyclopentenone **7**. A reductive alkylation sequence involving lithium in liquid ammonia followed by the in situ addition of iodide **9** gave ester **10** in 34 % yield after HPLC purification (next to 7 % of dialkylated product)¹⁴; subsequent lithium aluminum hydride reduction and acid work-up (dilute sulfuric acid)¹⁵ led directly to enone **12** isolated in 41 % yield after chromatography on florisil.



^aLi, NH₃, *t*-BuOH, THF; NH₃↑; Me₃SiCl, Et₃N, THF; ^bMeLi, THF-ether; **9**, HMPA; ^cLiAlH₄, ether; H₂O-NaOH; ^d0.005 N HCl-THF (1:4).

SCHEME 2

The Diels-Alder reaction of enone **12** (scheme 3)

Previously, we had observed the predominant formation of adduct **2** upon treatment of enone **1** in methylene chloride at room temperature for six days (¹H NMR; 61 % conversion; ratio 8:1 of **2** and **3**, respectively)³. In contrast, under the same conditions only minor amounts of the desired adduct **13** (< 10 %) were obtained from the corresponding alcohol **12**. This lack of reactivity (compared to **1**) can be ascribed to an intramolecular hydrogen bond¹⁶ between the 17-hydroxyl group (cf. steroid numbering) and the dienophilic side chain (cf. **15**) making the latter spatially unavailable for the cycloaddition reaction¹⁷. This is revealed by comparison of the H-12 resonances in the ¹H NMR spectrum (360 MHz, CDCl₃) of **1** (2H : 2.61

ppm, singlet) ^{3b} and **12** (1H : 3.34 ppm, doublet; 1H : 2.42 ppm, doublet; [2J] = 17 Hz). A further indication is provided by the results of the cycloaddition of **12** in chloroform-d, (dimethyl sulfoxide)-d₆ and water-d₂ (table). In the dipolar aprotic DMSO, where the alcohol should be completely hydrogen-bonded to the solvent ¹⁶, a similar reactivity is noticed as in the case of the protected hydroxy-enone **1**. Most notably, however, after dissolving enone **12** in deuterium oxide (10 min; ultrasonic assistance) immediate and complete conversion to adduct **13** is observed. Undiluted enone **12** has also been found to yield the desired adduct **13** upon standing at -20°C for prolonged periods of time.

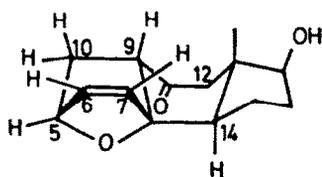
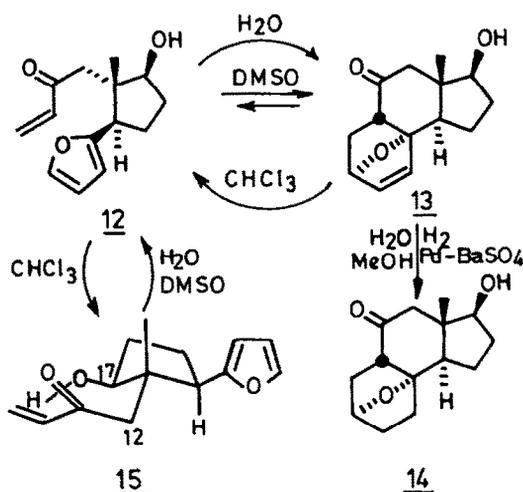


TABLE : Diels-Alder reaction of **12**

Conditions	Ratio (%) ^a	
	12	13
CDCl ₃ , 25°C, 3d ^b	>90	<10
(CD ₃) ₂ SO, 25°C, 3d ^b	50	50
D ₂ O, 25°C, 10 min ^c	-	>95
neat, -20°C, >2 months	-	>95

^aFrom ¹H NMR; ^bSame ratio's after 6 days;

^cDissolution assisted by Ultrasonic vibration.



SCHEME 3

For preparative purposes enone **12** is shaken vigorously in water (~0.07 M) for 15-30 min at room temperature. The resulting milky emulsion, which by TLC contains only the desired adduct, is separated by decantation from undissolved starting material and unidentified side products which remain as an oily film. Interestingly, upon isolation of **13** by extraction from the aqueous phase, the gradual formation of enone **12** is observed indicating a facile cycloreversion in the organic phase. Therefore, adduct **13** is kept in the aqueous phase and, after the addition of methanol, directly hydrogenated (Pd-BaSO₄; 2.2 bar). In this way pure reduced adduct **14** is obtained in overall yields of up to 85 % (from diol **11**). Lower yields, typically 50-60 %, are observed when the cycloaddition is performed on larger scales (see experimental) or in more concentrated solution; in these cases vigorous mechanical stirring or ultrasonic assistance did not give rise to any reproducible yield improvement. Also, retreating the oily residue (*vide supra*) in the manner as described above does not lead to substantial amounts of more **14**.

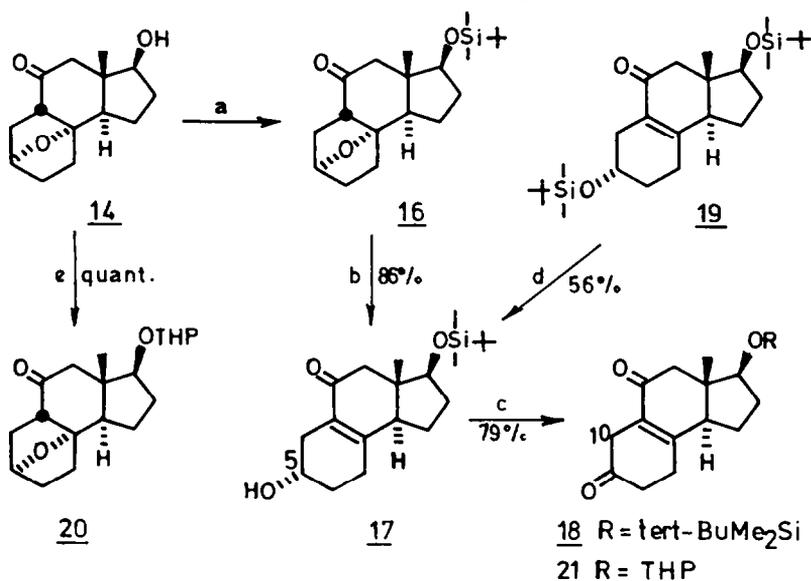
As in the case of adduct **23^b**, the stereochemistry of **13** is inferred from the following ¹H NMR spectral data (CDCl₃): (i) a doublet of doublets coupling pattern (11.5 and 8.0 Hz) for

the at high field resonating β -hydrogen at C-10 (1.42 ppm), typical of an exo-type adduct (oxygen bridge and H-9 trans)¹⁸; (ii) a large geminal coupling constant value for the H-12 protons (18.5 Hz) in accord with a boat-type conformation for ring C with a small endocyclic torsion angle at the 11,12-bond¹⁹; (iii) a diagnostic long range coupling between H-6 and H-14 (0.75 Hz)²⁰. Eventually, the structure **13** was fully confirmed by an X-ray diffraction study of the reduced adduct **14**²¹. The preferred formation of the exo-adduct with an α -oriented bridge (cf. **2** and **13**) under kinetic conditions has been discussed previously^{3b}.

Recently Breslow²² has reported important rate accelerations of some intermolecular Diels-Alder reactions when performed in water (also in the presence of cyclodextrins), the principal effect being ascribed to hydrophobic association of the diene with the dienophile^{23,24}. In view of the intramolecular mode of our reaction, the observed rate enhancement in water compared to dimethyl sulfoxide (cf. table) was unexpected²⁵. Finally, the enhanced stability of adduct **13** in water compared to organic solvents is also noteworthy and will be the subject of further study.

The synthesis of (+)-11-ketotestosterone (27) and (+)-adrenosterone (6)

Our next objective required the opening of the oxygen bridge of the reduced adduct. This would release the hydroxyl group at C-5 and, after oxidation, provide for a BCD-ring system correctly functionalized for the introduction at C-10 of the 19-methyl group and of the necessary side chain for the construction of the A-ring (scheme 4).

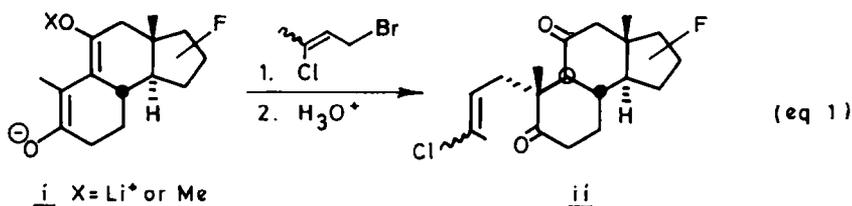


^asee text; ^bNaOMe, MeOH, 2 h, r.t.; ^cSO₃·C₅H₅N, Et₃N, DMSO, r.t.; ^dIM HF in CH₃CN, C₆H₆, 30 min, r.t.; ^eDHP, p-TsOH, CH₂Cl₂.

SCHEME 4

We have previously described the ready conversion of **16** (obtained from **2** via catalytic hydrogenation) into hydroxy-enone **17** (sodium methoxide in methanol, 2 h at room

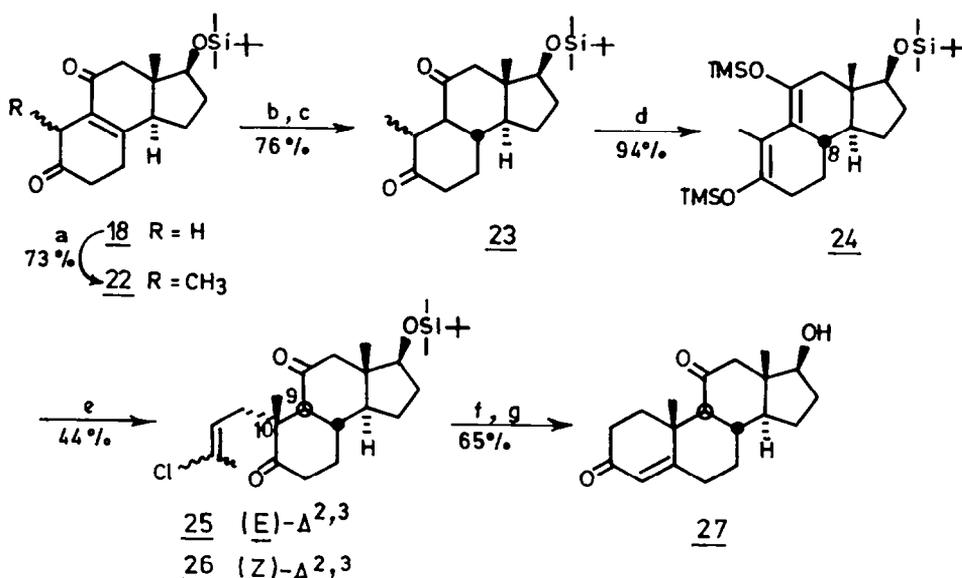
temperature; 86 % yield) and the subsequent oxidation to **18** (sulfur trioxide pyridine complex in triethylamine-dimethyl sulfoxide²⁶; 79 % yield)³⁰. The protection of the hydroxyl group in **14** with *tert*-butyldimethylsilyl chloride-imidazole²⁷ (ratio 2:1; 10 equiv) in dimethylformamide at room temperature offered no problem on a small scale but required a large excess of reagent (83 % yield of **16**). Enlarging the scale, however, gave rise to the formation of ring opened product **17** next to protected **16**; e.g. on a 3g-scale, a 89 % yield of **16** and **17** (ratio 4:1, respectively) was obtained (see experimental). On still larger scales substantial amounts of the disilyl ether **19** were also formed. Although the latter may be recycled via selective mono-desilylation to **17** with 1 M hydrogen fluoride in acetonitrile (56 % yield)²⁸, we examined a variety of alternative conditions for converting alcohol **14** into protected **16** in a reproducible way. However, the investigated methods led either to the recovery of starting material or to the formation of varying amounts of opened **17** and disilylated **19** next to the desired **16** (see experimental)²⁹. Lewis acid catalysis is probably responsible for the facile bridge opening since **16** was found unaffected by triethylamine alone. In contrast with the above result, protection of the hydroxyl group in **14** as the tetrahydropyranyl ether **20** occurs smoothly and in quantitative yield. Although **20** can be converted in a similar way to the corresponding keto-enone **21** (60 % overall yield), we preferred to use the silylated derivative **18** in the final sequence for reasons of stereohomogeneity.



Although the unsaturated diketone **18** offers in principle diverse possibilities for steroid synthesis³⁰, we have so far been mainly concerned with its conversion into a classical corticosteroid derivative. This necessitates both the reduction of the $\Delta^{8,9}$ -double bond to the requisite *trans*-BC-ring junction and the timely introduction of the angular 19-methyl group and of the side chain for eventual appendage of the A-ring. Crucial to the choice of our further strategy was the finding of Stork^{2d,f} that, upon alkylation of 11-oxygen substituted $\Delta^{9,11}, \Delta^{5,10}$ -dienolates **1**, the electrophile is introduced from the α -face leading to the derivative **11** with the group in equatorial position (equation 1).

Consequently, we set out to study the conversion of **18** into a derivative as **1**. This was readily accomplished as follows (scheme 5). Alkylation of the enolate anion derived from **18** (lithium diisopropylamide, tetrahydrofuran) with methyl iodide in hexamethylphosphoramide gave **22** as a 1:1 diastereomeric mixture in 73 % yield. Subsequent reduction with lithium in

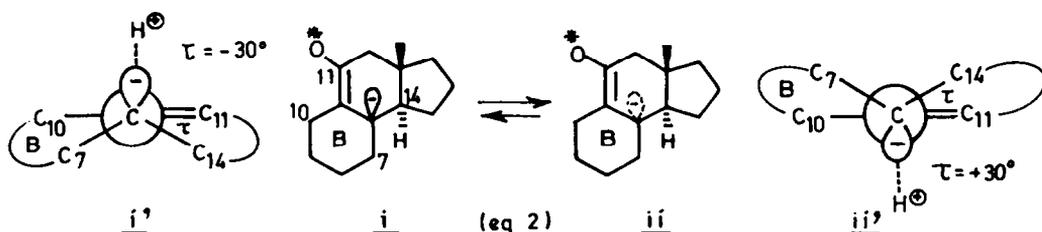
liquid ammonia in the presence of *tert*-butanol, directly followed by Jones oxidation, led to a mixture of diketone **23** (76 % yield) in which two diastereomers predominate (¹H NMR). Treatment of this mixture with trimethylsilyl iodide and hexamethyldisilazane in methylene chloride³¹ followed by fast elution on florisil finally yielded a single dienedioldisilylether **24** (94 % yield) with the required stereochemistry at C-8 as substantiated by the further synthesis of (±)-adrenosterone (**6**). This crucial stereochemical outcome results from the dissolved metal-ammonia reduction of enone **22**. It is of interest here to analyze the possible pathways for reduction with the results obtained with the microcomputer program for systematic conformational analysis (SCA) that has been recently reported by one of us³².



^a(*i*-Pr)₂NLi, THF; CH₃I, HMPT; ^bLi, NH₃, THF, *t*-BuOH; ^cJones, acetone; ^dMe₃SiI, (Me₃Si)₂NH, CH₂Cl₂; ^eCH₃Li, THF; ICH₂CHC(CH₃)Cl, HMPT; ^f70 % HClO₄-HCOOH; ^g2 N HCl-THF

SCHEME 5

Assuming that the stereochemical outcome of the enone reduction at C-8 is dictated by the thermodynamic stability of the intermediates **i** and **ii**³³, the SCA program computes a free energy difference ΔG° 16 kJ/mol (1 cal = 4.186 J) for the equilibrium **i** \rightleftharpoons **ii**, in line



with the observed result (equation 2). On the other hand, consideration of the minimum energy transition states in which overlap of the developing p-orbital on C-8 with the $\Delta^{9,11}$ -

bond of the enolate system is maintained during protonation³⁴, leads ideally to 1' (β -protonation at C-8) or 11' (α -protonation at C-8) with a torsion angle at C-8, C-9 in the C-ring of -30° and $+30^\circ$, respectively. In view of the deduced possible basic conformations³² for the C-ring of 1, i.e., 13+/HC, 13+/E and 14-/E with torsion angles at C-8, C-9 of -16° , 0° , and -29° , respectively, and of 11, i.e. 13+/HC and 13+/E with torsion angles of -16° and 0° , respectively, only in the former case can the stereoelectronic requirements for preferred protonation be accommodated. Hence, also on this basis would the observed β -configuration at C-8 be expected.

The eventual annelation of the A-ring proceeded via Wichterle's³⁵ method^{24,f}. In view of the reported enhanced reactivity towards nucleophiles of (E)-1,3-dichloro-2-butene, compared with the (Z)-derivative, we decided to study the alkylation with both pure isomers³⁶. Treatment of the dienediolate generated from 24 (methylolithium) with both (E)- and (Z)-1-iodo-3-chloro-2-butene³⁷ in hexamethylphosphoramide led, however, in both cases to a disappointing 44 % yield of the expected isomers 25 and 26, respectively³⁸. The assigned configuration at C-9 and C-10 in both compounds mainly follows from the observed ¹H NMR coupling pattern of H-9 (d, $J = 12-12.5$ Hz) and from the low field resonance³⁹ of the 19-methyl group (1.32-1.37 ppm). This preference for undergoing alkylation leading to equatorial products (cf. 24 to 25 and 26; eq. 2) has been rationalized before^{2f,39}. Finally the full steroid nucleus was realized according to the modified Wichterle's conditions (perchloric acid-formic acid)⁴⁰, directly followed by acidic hydrolysis of the intermediate formate to (\pm)-11-ketotestosterone (27) in 65 % yield. Final confirmation was obtained through oxidation²⁶ to (\pm)-adrenosterone (6; mp $185-186^\circ\text{C}$)⁴¹, whose spectral properties (IR, ³⁶⁰ ¹H NMR) and TLC behavior were identical with those of an authentic sample⁴².

Previous corticosteroid syntheses have often introduced the 11-oxygen function at a late stage^{1a,c,d,2a,b,d}. In contrast, in the present synthesis the same function plays a central role throughout the sequence. In masked form it served to activate the electrophile 9 upon alkylation of the enolate anion derived from 8 and to unravel the enone moiety in 12 upon hydrolysis of 11. It was found essential in activating the dienophile for the key Diels-Alder reaction to adduct 13³. It further served as a handle for the opening of the oxygen bridge (cf. 16 to 17) and enabled the obtention of the desired C-8 stereochemistry via dissolved metal reduction of enone 22. Eventually, it allowed for the generation of a dienediolate (cf. 24) and, hence, for the stereoselective introduction of the modified Wichterle reagent, necessary for the appendage of the A-ring. Finally we note that in this 16-step sequence the 19 carbon atoms of the steroid originated from furan, 2-methyl-1,3-cyclopentanedione, methyl iodide and 2 moles of acetoacetic ester.

EXPERIMENTAL

IR Spectra were recorded on a Beckmann IR 4230 or a Unicam SP 1000 spectrometer, mass spectra on a AEI MS-50 spectrometer. The ¹H NMR spectra were recorded at 90 MHz (Varian EM-390) or at 360 MHz (WH-Brucker) with TMS as internal standard. Rf values are

quoted for Merck silica gel 60 GF₂₅₄ plates of thickness 0.25 mm. HPLC separations were performed on Waters LC/System 500 or Waters 6000 A, both with RI-detection. M.ps. are uncorrected.

Ethyl(+)-(2' α ,3' β)-3-ethoxy-4-(3'-(2"-furyl)-2'-methylcyclopentan-1'-on-2'-yl)-2(E)butenoate (10). **Method a:** Cyclopentenone **7** (6 g; 37 mmol) is reduced with lithium (0.778 g; 0.111 mol) in liq ammonia (200 mL) containing *tert*-butanol (2.8 mL; 29.6 mmol) and THF (40 mL), as previously described^{3b}. After addition of isoprene the mixture is diluted with THF (100 mL) and treated in one portion with a soln of iodide **9** (31.6 g; 0.111 mol) in THF (60 mL). After stirring for 2.5 hrs at r.t., the reaction mixture is poured into a satd NH₄Cl soln. After usual work-up the residue is purified by HPLC with 15 % EtOAc-hexane as eluent, yielding 4.0 g of **10** (33.7 %) next to 1.3 g of 2,5-dialkylated product (7.4 %). **Method b:** To a cooled (0°C) soln of enol ether **8** (28.4 g; 0.12 mol) in dry THF (1 L) is added dropwise a soln of methyl lithium in ether (103.4 mL of a 1.16 M soln). After completion of the reaction (TLC) a soln of iodide **9** (104 g; 0.367 mol) in HMPA (182 mL) is added. After stirring for 3 hrs at r.t. the reaction mixture is poured into a satd NH₄Cl soln (700 mL). After the usual extractive (ether) work-up the residue (160 g) is dissolved in ether-hexane (1:1, 1.5 L) and the organic phase washed with water (4 x 75 mL). After drying (MgSO₄) and concentration in vacuo, the residue (100 g) is purified by HPLC with 20 % EtOAc-hexane as eluent, yielding 26 g of **10** (67.5 %) next to 7.4 g of 2,5-dialkylated product (13 %). For ester **10**: m.p. 76°C (EtOAc-hexane). Rf 0.19 (20 % EtOAc-isooctane). UV(MeOH): λ_{max} 220-240 nm. IR (KBr): 3110, 1740, 1710, 1700, 1620 cm⁻¹. NMR (360 MHz, CDCl₃): 7.31 (1H, dd: 2, 1 Hz), 6.30 (1H, dd: 3, 2 Hz), 6.06 (1H, dt: 3, 0.5 Hz), 5.10 (1H, s), 4.15 (2H, q: 7. Hz), 3.80 (2H, m), 3.68 (1H, d: 15 Hz), 3.36 (1H, dd: 11, 6.5 Hz), 2.88 (1H, dd: 15, 1 Hz), 2.5-2.0 (4H, m), 1.29 (3H, t: 7 Hz), 1.28 (3H, t: 7 Hz), 0.75 (3H, s) ppm. MS: m/z 320 (M⁺, 2), 276 (4), 275 (20), 164 (26), 163 (100), 162 (43), 158 (23). Exact mass. calcd for C₁₈H₂₄O₅: 320.1624. Found: 360.1612. For the dialkylated derivative: Rf 0.22 (20 % EtOAc-isooctane). IR (film): 1745, 1715, 1620 cm⁻¹. NMR (360 MHz, CDCl₃): 7.28 (1H, dd: 2, 1 Hz), 6.27 (1H, dd: 3, 2 Hz), 6.02 (1H, brd: 3 Hz), 5.12 (1H, s), 5.05 (1H, s), 4.13 (4H, m), 3.95-3.77 (4H, m), 3.55 (1H, t: 7.5 Hz), 3.43 (1H, d: 14.5 Hz), 3.33 (1H, dd: 13.5, 11.5 Hz), 2.98 (1H, d: 14.5 Hz), 3.04 (1H, dd: 13.5, 4 Hz), 2.3-2.0 (3H), 1.33 (3H, t: 7 Hz), 1.31 (3H, t: 7 Hz), 1.28 (3H, t: 7 Hz), 1.24 (3H, t: 7 Hz), 0.83 (3H, s) ppm. MS: m/z 185 (15), 178 (14), 163 (10), 162 (15), 84 (58), 79 (30), 77 (20), 75 (100).

(+)-(1' β ,2' α ,3' β)-3-Ethoxy-4-(3'-(2"-furyl)-1'-hydroxy-2'-methylcyclopentan-2'-yl)-2(E)-butenol (11). A soln of ester **10** (15.67 g; 49 mmol) in ether (155 mL) is added dropwise to a suspension of LAH (6.5 g; 0.17 mol) in ether (190 mL). After reflux for 1 h the suspension is cooled to 0°C and treated consecutively with water (6.5 mL), NaOH (15 % soln; 6.5 mL) and water (19.5 mL). After stirring for 30 min at r.t. the precipitate is filtered and thoroughly washed with ether. After concentration in vacuo the residue is purified by HPLC (55 % EtOAc-hexane) yielding 10.33 g of crystalline diol **11**. Typical yields are in the range 75-88 %. M.p. 84-86°C (EtOAc-isooctane). Rf 0.28 (60 % EtOAc-isooctane). IR (KBr): 3600-3200, 1640, 1620 cm⁻¹. NMR (360 MHz, CDCl₃): 7.36 (1H, dd: 2, 1 Hz), 6.33 (1H, dd: 3, 2 Hz), 6.11 (1H, dt: 3, 0.5 Hz), 4.87 (1H, t: 8 Hz), 4.07 (2H, m), 4.0 (1H, brt: 9.5 Hz), 3.78 (2H, m), 3.47 (1H, br s), 3.07 (1H, t: 9.8 Hz), 2.50 (1H, d: 14 Hz), 2.23 (1H, d: 14 Hz), 2.13-1.60 (5H, m), 1.33 (3H, t: 7 Hz), 0.64 (3H, s) ppm. MS: m/z 262 (M⁺-H₂O, 38), 217 (24), 205 (24), 55 (100).

(+)-(1' β ,2' α ,3' β)-4-(3'-(2"-Furyl)-1'-hydroxy-2'-methylcyclopentan-2'-yl)-1-buten-3-one (12). **Method a:** A soln of diol **11** (10.33 g; 36.9 mmol) and dil HCl (60 mL; 0.005 N) in THF (240 mL) is stirred at r.t. for 4 hrs. The reaction mixture is diluted with ether, washed with a satd NaHCO₃ soln and dried over MgSO₄. After the usual work-up enone **12** (8.63 g) is obtained in quantitative yield and further used without purification. Pure enone is best obtained via chromatography on florisil using 20-30 % EtOAc-isooctane. **Method b:** A soln of ester **10** (100 mg; 0.3 mmol) in ether (1.5 mL) is added dropwise to a suspension of LAH (38 mg; 1 mmol) in ether (1.5 mL). After reflux for 30 min, the suspension is cooled to r.t. and treated with water (38 μ L) and dil sulfuric acid (2 N) until a clear aqueous phase is obtained. After ether extraction the organic phase is washed with NaHCO₃ soln and dried (MgSO₄). After the usual work-up the residue is purified by column chromatography on florisil with 25 % EtOAc-isooctane as eluent yielding 30 mg (41 %) of enone **12**. Rf 0.46 (60 % EtOAc-isooctane). IR (film): 3430, 1685, 1660, 1610 cm⁻¹. IR (CHCl₃ soln): 3420, 1710-

1685, 1665, 1610 cm^{-1} . NMR (360 MHz, CDCl_3): 7.35 (1H, dd: 2, 0.8 Hz), 6.38 (1H, ABd: 17.5, 10.5 Hz), 6.33 (1H, dd: 3, 2 Hz), 6.25 (1H, ABd: 17.5, 1 Hz), 6.09 (1H, br d: 3 Hz), 5.85 (1H, dd: 10, 1.5 Hz), 4.56 (1H, br d: 1 Hz), 4.12 (1H, br t: 8.5 Hz), 3.34 (1H, d: 17 Hz), 2.86 (1H, dd: 11.5, 8.5 Hz), 2.42 (1H, d: 17 Hz), 2.21–1.76 (4H, m), 0.70 (3H, s) ppm. MS: m/z 234 (M^+ ; 2), 216 (4), 183 (15), 164 (54), 149 (87), 94 (100). Exact mass. Calcd for $\text{C}_{14}\text{H}_{18}\text{O}_3$: 234.1255. Found: 234.1110.

The Diels-Alder reaction of enone 12. (+)-(1 α ,2 α ,5 β ,6 β ,9 β ,11 α)-5-Hydroxy-6-methyl-14-oxatetracyclo[7.4.1.1¹¹.0^{2,6}]/tetradecan-8-one (14). A suspension of crude enone **12** (4.28 g; 18 mmol), obtained from diol **11** (5.04 g), in water (240 mL) is shaken vigorously for 30 min. At this stage the adduct can be isolated in pure form by decanting the milky emulsion from the oily residue and extracting the aqueous phase with ether. When kept into the organic phase the adduct **13** reverts to enone **12** within hours. Usually, however, the milky emulsion is transferred to a hydrogenation vessel (1 L) and diluted with methanol (120 mL). After addition of palladium on barium sulfate (180 mg) the mixture is hydrogenated at 2.2 bar. After 2 hrs the catalyst is filtered and washed with methanol. After concentration of the filtrate in vacuo, the residue is salted out with solid sodium chloride and extracted with ether. After usual work-up (MgSO_4) the solid residue is recrystallized from EtOAc-hexane (3:2) yielding 2.23 g of alcohol **14**. Typical overall yields from diol **11** are dependent on the scale and range from 53 % (this procedure) to 85 % (small scale). For adduct **13**: m.p. 88–90.5°C. Rf 0.10 (60 % EtOAc-isooctane). IR (KBr): 3400, 1700, 1635, 1615 cm^{-1} . NMR (360 MHz, CDCl_3): 6.47 (1H, AB, 5.8 Hz), 6.43 (1H, AB, dd: 5.8, 1.5, 0.75 Hz), 4.93 (1H, dd: 4.5, 1.5 Hz), 3.90 (1H, t: 8.5 Hz), 2.59 (1H, AB, 18.5 Hz), 2.57 (1H, dd: 8.0, 4.5 Hz), 2.33 (1H, dt: 11.5, 4.5 Hz), 2.31 (1H, br AB: 18.5 Hz), 2.29–2.19, 2.05–1.87, 1.74–1.66 (5H, m), 1.42 (1H, dd: 11.5, 8.0 Hz), 1.22 (3H, s) ppm. MS: m/z 234 (M^+ , 5), 216 (7), 164 (72), 149 (100), 134 (17), 121 (25), 94 (57), 55 (80), 43 (75). For alcohol **14**: m.p. 120°C (sublimation). Rf 0.12 (60 % EtOAc-isooctane). IR (KBr): 3500, 1680 cm^{-1} . NMR (360 MHz, CDCl_3): 4.48 (1H, t: 5.0 Hz), 3.86 (1H, td: 8.75, 5.5 Hz), 2.62 (1H, dd: 9.0, 6.0 Hz), 2.54 (1H, AB, 19.0 Hz), 2.23 (1H, AB, 19.0 Hz), 2.25–2.05, 1.95–1.54 (11H, m), 1.64 (1H, dd: 12.3, 9.0 Hz), 1.13 (3H, s) ppm. MS: m/z 236 (M^+ , 80), 218 (22), 208 (60), 207 (18), 180 (100), 179 (30). Exact mass. Calcd for $\text{C}_{14}\text{H}_{20}\text{O}_3$: 236.1412. Found: 236.1445.

(+)-(1 α ,2 α ,5 β ,6 β ,9 β ,11 α)-5-tert-Butyldimethylsilyloxy-6-methyl-14-oxatetracyclo[7.4.1.1¹¹.0^{2,6}]/tetradecan-8-one (16). To a soln of tert-butyldimethylsilyl chloride (18.9 g; 0.126 mol) in dry DMF (22.5 mL) is added a soln of alcohol **14** (3 g; 12.6 mmol) and imidazole (4.3 g; 63 mmol) in DMF (30 mL) in one portion. After completion of the reaction (TLC) the mixture is diluted with n-hexane (60 mL) and poured into a satd NH_4Cl soln. The aqueous phase is extracted with hexane and ether. After usual work-up of the combined organic phases (MgSO_4) the residue is purified by gradient elution on silica gel (10 g/g product) yielding the desired **16** (3.2 g; 72 %) with 3 % ether-benzene and the opened alcohol **17** (0.75 g; 17 %) with 50 % EtOAc-hexane. Remaining traces of tert-butyldimethylsilyl alcohol are removed in vacuo (0.05 mm Hg/70°C). On larger scales varying amounts of the corresponding disilyl ether **19** are also formed (vide infra). The base opening of pure silyl ether **16** to the corresponding enone **17** has been described before. For the spectral data of **16** and **17**, see reference 3b. For the disilyl ether **19**: m.p. 112°C. Rf 0.85 (60 % EtOAc-hexane). IR (KBr): 1645, 1620 cm^{-1} . NMR (360 MHz, CDCl_3): 3.9 (2H, m), 2.59 (1H, d: 16.0 Hz), 2.10 (1H, dd: 16.0, 1.0 Hz), 0.87 (9H, s), 0.86 (9H, s), 0.76 (3H, d: 1.0 Hz), 0.055 (3H, s), 0.050 (3H, s), 0.024 (3H, s), 0.018 (3H, s) ppm. MS: m/z 449 (M^+ ; 15, 10), 407 (77), 71 (100). No reaction was observed when using a small excess of tert-butyldimethylsilyl chloride (TBMCS) and imidazole in DMF, e.g., ratio 2.2/1.1, 3/1.5, 2/4, 3/6 (equivalents, respectively). More recent methods involving TBMCS (7.5 equiv) and HMPA^{29a}, TBMCS (6.6 equiv) and Et_3N (30 equiv) in THF^{29b}, TBMCS (1.5 equiv) and Et_3N (3.0 equiv)/DMAP (0.15 equiv) in methylene chloride^{29c}, TBMCS and lithium sulfide in acetonitrile^{29d}, TBMCS (6 equiv) and diisopropylethylamine (3 equiv) in methylene chloride^{29e}, tert-butyldimethylallylsilane and p-toluene sulfonic acid in acetonitrile^{29f}, tert-butyldimethylsilyl perchlorate and pyridine in acetonitrile^{29g}, TBMCS (1.0 equiv) and allquat 336 or 18-Crown-6, potassium carbonate^{29h}, also led to the recovery of starting material. The desired silyl ether **16**, next to varying amounts of **17** and **19** (TLC), was formed using a large excess of TBMCS and imidazole in DMF (vide supra), TBMCS and 3,5-dimethylpyrazole in DMF (ratio 1/1, 2/1 or 2/2, respectively), tert-

butyldimethylsilyl cyanide (1.0 equiv) in DMF at 25°C or 80°C,²⁹ⁱ and TBDMCS/diisopropyl-ethylamine in DMF^{29e}.

(±)-(2.3β,3αβ,6.7α,8.9.9bc)-Octahydro-3-tert-butylidimethylsilyloxy-7-hydroxy-3a-methyl-1H-benz[e]indene-5(4H)-one (17). To a soln of ether **19** (140 mg; 0.3 mmol) in benzene (2 mL) is added a soln of 1 M hydrogen fluoride in acetonitrile (300 μl of a 5 % soln of 40 % HF in CH₃CN). After 25 min (TLC) solid NaHCO₃ is added and the mixture extracted with ether. After the usual work-up the residue is purified by column chromatography on silica gel yielding the alcohol **17** (59 mg; 56 %) next to disilyl ether **19** (20 mg; 14 %).

(±)-(1α,2α,5β,6β,9β,11α)-6-Methyl-5-tetrahydropyranyloxy-14-oxatetracyclo[7.4.1.1^{11,02,6}.6]tetradecan-8-one (20). A soln of alcohol **14** (3.91 g; 16.6 mmol), *p*-toluene-sulfonic acid (20 mg) and dihydropyran (2.08 g; 24.9 mmol) in methylene chloride (70 mL) is stirred for 30 min at 30°C. After the addition of K₂CO₃ the mixture is diluted with methylene chloride, filtered and washed with a satd Na₂CO₃ soln. Usual work-up gave crude crystalline ether **20** (5.3 g; 100 %) sufficiently pure for further work. Rf 0.37 (60 % EtOAc-isooctane). IR (KBr): 1695 cm⁻¹. NMR (360 MHz, CDCl₃): 4.60 (1H, m), 4.47 (1H, t: 5.0 Hz), 3.92-3.78 (2H, m), 3.49 (1H, m), 1.16 and 1.13 (3H, 2 s) ppm. MS: m/z 302 (M⁺-18, 22), 218 (10), 180 (10), 85 (100).

(±)-(2.3β,3αβ,8.9.9bc)-Hexahydro-3a-methyl-3-tetrahydropyranyloxy-1H-benz[e]indene-5,7-(4H,6H)-dione (21). A soln of crude ether **20** (5.3 g; 16.6 mmol) and sodium methoxide (0.894 g; 16.6 mmol) in methanol (250 mL) is stirred under nitrogen at 20°C for 3 hrs. After concentration in vacuo, water is added and the soln extracted with ether. After drying (MgSO₄) and further work-up the residue is purified via column chromatography on silica gel (70 % EtOAc-hexane) yielding 4.5 g (85 %) of hydroxy enone (cf. **17**) that is subsequently oxidized as described for the conversion of **17** into **18**^{3b}. The corresponding diketone **21** (3.13 g) is obtained in 70 % yield. Rf 0.2 (50 % EtOAc-isooctane). IR (KBr): 1720, 1665, 1625 cm⁻¹. NMR (360 MHz, CDCl₃): 4.63 (1H, m), 3.99 (1H, m), 3.88 (1H, m), 3.51 (1H, m), 3.14 (1H, br AB: 21.5 Hz), 3.03 (1H, br AB: 21.5 Hz), 0.87 and 0.85 (3H, d: 1.0 Hz).

(±)-(2.3β,3αβ,8.9.9bc)-Hexahydro-3-tert-butylidimethylsilyloxy-3a,6-dimethyl-1H-benz[e]indene-5,7-(4H,6H)-dione (22). To a cooled soln (0°C) of lithium diisopropylamide, prepared from diisopropylamine (0.905 g; 8.96 mmol) in dry THF (54 mL) and *n*-butyllithium (6.48 mL of a 1.33 M soln in hexane), is added dropwise a soln of enone **18** (3.0 g; 8.62 mmol) in THF (82 mL). After stirring for 30 min at 0°C HMPA (11 mL), followed by methyl iodide (10.03 g; 70.6 mmol), are added. After completion of the reaction (TLC) the mixture is poured into a satd NH₄Cl soln and extracted with ether. After the usual work-up the residue is purified by HPLC with 10 % EtOAc-hexane as eluent, yielding a 1:1 mixture of diastereomeric **22** (2.27 g; 73 %). Rf 0.36 (30 % EtOAc-isooctane). UV (MeOH): λ_{max} 254 nm. IR (KBr): 1710, 1655, 1615 cm⁻¹. NMR (360 MHz, CDCl₃): 3.92/3.89 (1H, t: 8.5 Hz), 3.35/3.29 (1H, br q: 7.5 Hz), 1.24/1.23 (3H, d: 7.3 Hz), 0.89/0.88 (9H, s), 0.82/0.76 (3H, d: 0.8 Hz), 0.06 (6H, s).

(±)-(2.3β,3αβ,5a,8.9.9aβ,9bc)-Octahydro-3-tert-butylidimethylsilyloxy-3a,6-dimethyl-1H-benz[e]indene-5,7-(4H,6H)-dione (23). To a soln of lithium (71 mg; 10 mmol) in liquid ammonia (20 mL; distilled from sodium) is added slowly a soln of enone **23** (618 mg; 1.7 mmol) in THF (13 mL) and *tert*-butanol (0.13 g; 1.7 mmol). After stirring for 45 min at -33°C the reaction is quenched with NH₄Cl. After evaporation of the ammonia, the residue is dissolved in water and extracted with ether. After the usual work-up, the residue is dissolved in acetone (50 mL) and treated at 0°C with Jones reagent. After completion of the reaction (TLC), excess reagent is destroyed with isopropanol and solid sodium carbonate added. After stirring for 15 min the mixture is filtered, the precipitate washed with ether and acetone, and the combined organic phases concentrated in vacuo. The residue is dissolved in water and extracted with ether. After the usual work-up the residue is purified by column chromatography with 10 % EtOAc-isooctane as eluent yielding 469 mg of **23** (76 %) as a diastereomeric mixture in which 2 isomers (NMR) predominate. Rf 0.20, 0.25 (20 % EtOAc-isooctane). IR (KBr): 1710 cm⁻¹. NMR (360 MHz, CDCl₃): 3.83/3.79 (1H, t: 8.5 Hz), 1.19/1.05 (3H, d: 7.0/6.5 Hz), 0.87 (9H, s), 0.75/0.73 (3H, d: 0.5/1.0 Hz) ppm. MS: m/z 307 (M⁺-57, 100), 289 (10), 231 (26), 75 (98). Exact mass. Calcd for C₂₁H₃₆O₃Si: 364.2434. Found: 364.2294.

(±)-(2.3β,3αβ,4.8,9.9αβ,9β)-Octahydro-3-tert-butylidimethylsilyloxy-3α,6-dimethyl-5,7-ditrimethylsilyloxy-1H-benz[e]indene (24). To a cooled (-20°C) soln of diketone **23** (0.369 g; 1.01 mmol) in dry methylene chloride (18 mL) is added consecutively hexamethyldisilazane (0.512 mL; 2.43 mmol) and iodotrimethylsilane (0.317 mL; 2.23 mmol) under nitrogen. After stirring for 15 min at -20°C and for 2 hrs at 25°C, the reaction mixture is diluted with methylene chloride (20 mL) and the organic phase washed with NaHCO₃ soln (5 mL). After the usual work-up, the residue is purified rapidly over florisil (20 g; 1 % acetone-hexane as eluent) yielding 484 mg of a colourless viscous oil (94 %). R_f 0.96 (20 % EtOAc-hexane). IR (film): 1610 cm⁻¹. NMR (360 MHz, CCl₄): 3.68 (1H, t; 8.5 Hz), 2.06 and 1.97 (2H, br AB: 16 Hz), 1.84 (3H, t; 1.8 Hz), 0.89 (9H, s), 0.79 (3H, s) ppm.

(±)-(2.3β,3αβ,5α,8,9.9αβ,9β)-Octahydro-3-tert-butylidimethylsilyloxy-6α-(3'-chlorobut-2'(E)-en-1'-yl)-3α,6β-dimethyl-1H-benz[e]indene-5,7(4H,6H)-dione (25). To a cooled soln (0°C) of freshly prepared ether **24** (439 mg; 0.86 mmol) in dry THF (20 mL) is added dropwise methylolithium (1.56 mL of a 1.11 M soln in ether). After completion of the reaction (TLC) the reaction mixture is cooled to -40°C and treated with (E)-1-iodo-3-chloro-2-butene (186 mg; 0.86 mmol) in HMPA (3 mL; 17 mmol) in one portion. After 20 min at 20°C the mixture is poured into water and extracted with ether. After the usual work-up the residue is purified by HPLC (5 % EtOAc-hexane) yielding 168 mg of diketone **25** (43 %). R_f 0.40 (20 % EtOAc-isooctane). IR (film): 1710, 1660 cm⁻¹. NMR (360 MHz, CDCl₃): 5.41 (1H, ddq; 10.0, 6.25 and 1.25 Hz), 3.79 (1H, t; 8.5 Hz), 2.70 (1H, dd; 14.0, 10.0 Hz), 2.63 (1H, ddq; 14.0, 6.25 and 1.0 Hz), 2.56 (1H, td; 14.5, 6.25 Hz), 2.34 (1H, d; 12.5 Hz), 2.33 (1H, ddd; 15.0, 4.75, 2.5 Hz), 2.21 (1H, AB; 11.5 Hz), 1.96 (3H, br s; Δ^{1/2} 4 Hz), 1.32 (3H, s), 0.87 (9H, s), 0.73 (3H, br s), 0.008 (6H, s) ppm. MS: m/z 452 (M⁺, 4), 395 (35), 225 (17), 161 (18), 145 (20), 135 (17), 75 (100). Exact mass. Calcd for C₂₅H₄₁O₃SiCl₃³⁵: 452.2513. Found: 452.2513.

The (Z)-isomer 26. Using (Z)-1-iodo-3-chloro-2-butene, there was obtained from ether **24** (429 mg) the corresponding diketone **26** (169 mg; 44 % yield) in the same manner as described for the synthesis of **25**. R_f 0.39 (20 % EtOAc-isooctane). IR (film): 1705 cm⁻¹. NMR (360 MHz, CDCl₃): 5.39 (1H, ddq; 9.5, 5.5 and 1.25 Hz), 3.79 (1H, t; 8.25 Hz), 2.89 (1H, dd; 14.5, 9.5 Hz), 2.54 (2H, m), 2.35 (1H, d; 12.0 Hz), 2.31 (1H, ddd; 15.0, 4.5, 2.5 Hz), 2.21 (1H, AB; 11.5 Hz), 2.04 (3H, br s), 1.37 (3H, s), 0.86 (9H, s), 0.73 (3H, br s), 0.08 (6H, s) ppm. MS: m/z 452 (M⁺, 32), 417 (22), 395 (30), 223 (17), 207 (75), 75 (100).

(±)-11-ketotestosterone (27) and (±)-adrenosterone (6). A soln of the chloride **25** or **26** (or a mixture of both; 71 mg; 0.16 mmol) in formic acid (7.2 mL) and 70 % perchloric acid (0.7 mL) is stirred at 80°C under nitrogen until no starting material is visible on TLC. After concentration in vacuo the residue is diluted with EtOAc (10 mL) and a satd Na₂CO₃ soln (3 mL) and further treated with solid NaHCO₃ until the evolution of CO₂ has ceased. After separation of the organic phase, the aqueous phase is extracted with ethyl acetate. After usual work-up (Na₂SO₄) the residue is diluted with THF (5 mL) and 2 N hydrochloric acid (1.4 mL). After stirring for 35 min at reflux the solution is diluted with an equal volume of ether and further treated consecutively with a satd NaHCO₃ soln (1.5 mL) and solid NaHCO₃. The aqueous phase is extracted with ether and the combined organic phases dried over Na₂SO₄. After the usual work-up the residue is purified by HPLC (60% EtOAc-hexane as eluent) yielding 31 mg (65%) of (±)-11-ketotestosterone (**27**). M.p. 192-195°C (EtOAc). R_f 0.20 (60 % EtOAc-hexane). UV (MeOH): λ_{max} 238 nm. IR (CHCl₃ soln): 3450 (br), 1710, 1670, 1620 cm⁻¹. NMR (360 MHz, CDCl₃): 0.76 (3H, d; 0.75 Hz), 1.43 (3H, s), 2.77 (1H, ddd; 13.75, 5.0, 3.25 Hz), 3.87 (1H, t; 8.5 Hz), 5.73 (1H, br s) ppm. MS: m/z 302 (M⁺, 20), 274 (14), 258 (16), 243 (18), 181 (40), 161 (38), 122 (100). For the purpose of structural confirmation a sample of (±)-11-ketotestosterone (**27**) is oxidized with sulfur trioxide pyridine complex in dimethyl sulfoxide²⁶ to give, after purification on HPLC (60 % EtOAc-hexane), (±)-adrenosterone (**6**). M.p. 185°-186°C (EtOAc). R_f 0.21 (60 % EtOAc-hexane). IR (CHCl₃ soln): 1745, 1710, 1665, 1620 cm⁻¹. UV (MeOH): λ_{max} 236 nm. NMR (360 MHz, CDCl₃): 0.89 (3H, d; 0.75 Hz), 1.44 (3H, s), 2.77 (1H, ddd; 13.75, 5.0, 3.25 Hz), 5.75 (1H, br s). MS: m/z 300 (M⁺, 20), 285 (5), 282 (4), 257 (10), 256 (10), 228 (10), 161 (16), 122 (100).

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REFERENCES AND NOTES

1. For early corticosteroid syntheses, see : (a) R.B. Woodward, F. Sondheimer, D. Taub, K. Heusler and W.M. McLamore, *J. Am. Chem. Soc.*, **74**, 4223 (1952); (b) L.H. Sarett, G.E. Arth, R.M. Lukes, R.E. Beyler, G.I. Poos, W.F. Johns, and J.M. Constantine, *Ibid.*, **74**, 4976 (1952); (c) W.S. Johnson and D.S. Allen, Jr., *Ibid.*, **79**, 1261 (1957); (d) L. Velluz, G. Nomine, and J. Mathieu, *Angew. Chem.*, **72**, 725 (1960)
2. For more recent approaches to corticosteroids, see : (a) J.W. Scott, P. Buchsacher, L. Labler, W. Meier, and A. Fürst, *Helv. Chim. Acta*, **57**, 1217 (1974); (b) U. Eder, G. Sauer, G. Haffer, J. Ruppert, and R. Wiechert, *Ibid.*, **59**, 999 (1976); (c) W.S. Johnson, S. Escher, and B.W. Metcalf, *J. Am. Chem. Soc.*, **98**, 1039 (1976) and W.S. Johnson, B. Frei, and A.S. Gopalan, *Ibid.*, **46**, 1512 (1981); (d) G. Stork and E.W. Logusch, *Ibid.*, **102**, 1219 (1980); (e) G. Stork, G. Clark, and C. Shiner, *Ibid.*, **103**, 4948 (1981) and G. Stork and D.H. Sherman, *Ibid.*, **104**, 3758 (1982); (f) G. Stork, J.D. Winkler, and C. Shiner, *Ibid.*, **104**, 3767 (1983); (g) F. Ziegler and T.-F. Wang, *Ibid.*, **106**, 718 (1984).
3. (a) L.A. Van Royen, R. Mijngheer, and P.J. De Clercq, *Tetrahedron Lett.*, **23**, 3283 (1982); for the full account, see : (b) L.A. Van Royen, R. Mijngheer, and P.J. De Clercq, *Bull. Soc. Chim. Belges*, **93**, 1019 (1984).
4. For a preliminary account, see : L.A. Van Royen, R. Mijngheer and P.J. De Clercq, *Tetrahedron Lett.*, **24**, 3145 (1983).
5. For total syntheses of optically active adrenosterone, see ref. 1d and 2d; for a synthesis of (\pm)-adrenosterone, see ref. 2f.
6. Obtained by a Finkelstein reaction on the corresponding chloride; for the synthesis of the latter compound, see : Y.U. Arbuzov and A.M. Korolev, *Zh. Obsh. Khim.*, **32**, 3674 (1962).
7. M. Stiles and A.L. Longroy, *Tetrahedron Lett.*, 337 (1961); M. Stiles and A.L. Longroy, *J. Org. Chem.*, **32**, 1095 (1967). However, see also : E. Wenkert and D.P. Strike, *J. Am. Chem. Soc.*, **86**, 2044 (1964).
8. For previous alkylations involving 3-alkoxy-4-bromo-2-butenate esters, see : (a) S.M. Weinreb and J. Auerbach, *J. Am. Chem. Soc.*, **97**, 2503 (1975); and with further use as the equivalent of acetoacetate, see e.g. : (b) G. Stork, D.F. Taber, and M. Marx, *Tetrahedron Lett.*, 2445 (1978); (c) G. Stork, R.K. Boeckmann, Jr., D.F. Taber, W.C. Still and J. Singh, *J. Am. Chem. Soc.*, **101**, 7107 (1977); (d) S. Danishefsky, K. Vaughan, R.C. Gadwood, and K. Tsuzuki, *J. Am. Chem. Soc.*, **102**, 4262 (1980); (e) L.A. Paquette, G.D. Annis, H. Schostarez, and J.F. Blount, *J. Org. Chem.*, **46**, 3768 (1981); (f) C. Exon, M. Nobbs, and P. Magnus, *Tetrahedron*, **37**, 4515 (1981). For a similar use of methyl 4-iodo-3-methoxy-2-butenate as a π -electrophilic equivalent of acetoacetate, see : (g) S. Danishefsky, K. Vaughan, R. Gadwood, and K. Tsuzuki, *J. Am. Chem. Soc.*, **103**, 4136 (1981).
9. Obtained from ethyl (*E*)-3-ethoxy-2-butenate via NBS-bromination^{8a,9a}, followed by a Finkelstein reaction^{8b}; (a) E.B. Reid and W.R. Ruby, *J. Am. Chem. Soc.*, **73**, 1054 (1951). Ethyl (*E*)-3-ethoxy-2-butenate was prepared by acid catalyzed reaction of ethyl acetoacetate with triethyl orthoformate and ethanol^{9b}, a procedure known to yield the (*E*)-derivative^{9c}; (b) Blaise, Maire, *Ann. Chim. Phys.*, **8**, 567 (1908); (c) E.E. Smitsman, and A.N. Voldeng, *J. Org. Chem.*, **29**, 3161 (1964). Our ¹H NMR spectral data were consistent with those previously reported : (d) J.-C. Chalchat, F. Theron and R. Vessière, *Ann. Chim.*, **Z**, 269 (1971); see also ref. 9c.
10. For a discussion of the stereochemical outcome of a related reaction, see ref. 3b.
11. Originally this was brought to our attention by a previous referee. The other groups who have employed this type of reagent⁸ probably also have dealt with the (*E*)-derivative, although the method of preparation has not always been explicated. Nevertheless, only Stork et al.^{8b,c} assign the (*E*)-configuration to it; in other cases, the reagent (or a reaction product) is drawn as the (*Z*)-derivative or referred to as a crotonate derivative (methyl and ester *trans*).
12. W.F. Gannon and H.O. House, *Org. Synth.*, **40**, 14 (1960).
13. S. Danishefsky, T. Kitahara, C.F. Yan, and J. Morris, *J. Am. Chem. Soc.*, **101**, 6996 (1973).
14. E.S. Binkley and C.H. Heathcock, *J. Org. Chem.*, **40**, 2156 (1975); see also ref. 8e,f.
15. R.L. Frank and H.K. Hall, *J. Am. Chem. Soc.*, **72**, 1645 (1950).
16. M.D. Joesten and L.J. Schaad, "Hydrogen Bonding", Marcel Dekker, New York, 1974.
17. The reverse effect has been observed in a series of intramolecular Diels-Alder reactions involving furan-diene, see : T. Takebayashi, N. Iwasawa, and T. Mukayama, *Bull. Chem. Soc. Jpn.*, **56**, 1107 (1983).
18. W.L. Nelson and D.R. Allen, *J. Heterocyclic Chem.*, **9**, 561 (1972). See also : P.J. De Clercq and L.A. Van Royen, *Synthetic Commun.*, **9**, 771 (1979).
19. M. Bartfield and D. Grant, *J. Am. Chem. Soc.*, **85**, 1899 (1963).

20. S.H. Grover and J.B. Stothers, *J. Am. Chem. Soc.*, **91**, 4331 (1969). See also: W.M. Grootaert and P.J. De Clercq, *Tetrahedron Lett.*, **23**, 3291 (1982).
21. M. Van Meerssche, J.P. Declercq, L.A. Van Royen and P.J. De Clercq, *Bull. Soc. Chim. Belges*, **92**, 115 (1983).
22. D.C. Rideout and R. Breslow, *J. Am. Chem. Soc.*, **102**, 7816 (1980).
23. Also enhanced endo/exo selectivities have been reported, an effect persisting even with relatively insoluble systems where two phases are present: R. Breslow, U. Maitra, and D. Rideout, *Tetrahedron Lett.*, **24**, 1901 (1983).
24. For recent examples of rate and selectivity enhancements of aqueous intermolecular Diels-Alder reactions involving salts of diene-acids, see: P.A. Grieco, P. Garner, and Z. He, *Tetrahedron Lett.*, **24**, 1897 (1983); P.A. Grieco, K. Yoshida, and P. Garner, *J. Org. Chem.*, **48**, 3137 (1983); P.A. Grieco, K. Yoshida, and Z. He, *Tetrahedron Lett.* **25**, 5715 (1984).
25. For an example where the hydrophobic effect actually slows down an intramolecular Diels-Alder reaction with furan-diene, see: D.D. Sternbach and D.M. Rossana, *J. Am. Chem. Soc.*, **104**, 5853 (1982).
26. J. Parikh and W. von E. Doering, *J. Am. Chem. Soc.*, **89**, 5505 (1967).
27. E.J. Corey and A. Venkateswarlu, *J. Am. Chem. Soc.*, **94**, 6190 (1972).
28. R.F. Newton, D.P. Reynolds, C.F. Webb, and S.M. Roberts, *J. Chem. Soc., Perkin 1*, 2055 (1981); R.F. Newton, D.P. Reynolds, M.A.W. Finch, D.R. Kelly, and S.M. Roberts, *Tetrahedron Lett.*, 3981 (1979); E.W. Collington, M. Finch, and I.J. Smith, *ibid.*, **26**, 681 (1985).
29. (a) R.M. Christie, M. Gill, and A.W. Rickards, *J. Chem. Soc., Perkin 1*, 593 (1981); (b) K.K. Ogilvie, D.P.C. McGee, S.M. Boisvert, G.H. Hakimelahi, and Z.A. Proba, *Can. J. Chem.*, **61**, 1204 (1983); (c) S.K. Chaudhary, O. Hernandez, *Tetrahedron Lett.*, 99 (1979); (d) G.A. Olah, B.G.B. Gupta, S.C. Narang, and R. Malhotra, *J. Org. Chem.*, **44**, 4272 (1979); (e) L. Lombardo, *Tetrahedron Lett.*, **25**, 227 (1984); (f) T. Morita, Y. Okamoto, and H. Sakurai, *Tetrahedron Lett.*, 835 (1980); (g) T.J. Barton and C.R. Tully, *J. Org. Chem.*, **43**, 3649 (1978); (h) M. Lissch and J. Weiffen, *Synthetic Commun.*, **11**, 545 (1981); (i) J.K. Rasmussen and S.M. Heilmann, *Synthesis*, 523 (1979).
30. For the potential use of the corresponding 11-deoxy-derivative, see ref. 1d.
31. R. Muller and D. McKean, *Synthesis*, 730 (1979).
32. P.J. DeClercq, *Tetrahedron*, **40**, 3717, 3729 (1984).
33. D.H.R. Barton and C.H. Robinson, *J. Chem. Soc.*, 3045 (1954).
34. G. Stork and S.D. Darling, *J. Am. Chem. Soc.*, **86**, 1761 (1964).
35. O. Wichterle, *Collect. Czech. Chem. Commun.*, **12**, 93 (1947).
36. L.F. Hatch and R.H. Perry, Jr., *J. Am. Chem. Soc.*, **77**, 1136 (1955).
37. Prepared from the corresponding chlorides via Finkelstein reaction. The latter were obtained from pure methyl (*E*)- and (*Z*)-3-chloro-2-butenoate; see: R.E. Conrow and J.A. Marshall, *Synthetic Commun.*, **11**, 419 (1981). The stereochemical assignment follows that of ref. 9d.
38. See, also: F. Näf and R. Decorzant, *Helv. Chim. Acta*, **57**, 1317 (1974).
39. G. Stork and E.W. Logusch, *J. Am. Chem. Soc.*, **102**, 1218 (1980).
40. M. Kobayashi and T. Matsumoto, *Bull. Chem. Soc. Japan*, **52**, 1978 (1979).
41. Literature mp 167-169°C (ref. 2f).
42. Purchased from Fluka.