ACS Medicinal Chemistry Letters

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ACS Med. Chem. Lett., Just Accepted Manuscript • DOI: 10.1021/acsmedchemlett.6b00167 • Publication Date (Web): 01 Jun 2016 Downloaded from http://pubs.acs.org on June 5, 2016

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The Discovery of Novel Dot1L Inhibitors through a Structure-Based **Fragmentation Approach**

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Dot1L, protein lysine methyltransferase, inhibitor, mixed lineage leukemia, protein structure-based design

ABSTRACT: Oncogenic MLL fusion proteins aberrantly recruit Dot1L, a histone methyltransferase, to ectopic loci, leading to local hypermethylation of H₃K₇₉ and misexpression of HoxA genes driving MLL-rearranged leukemias. Inhibition of the methyltransferase activity of Dot1L in this setting is predicted to reverse aberrant H3K79 methylation, leading to repression of leukemogenic genes and tumor growth inhibition. In the context of our Dot1L drug discovery program, highthroughput screening led to the identification of 2, a weak Dot1L inhibitor with an unprecedented, induced pocket binding mode. A medicinal chemistry campaign, strongly guided by structure-based consideration and ligand-based morphing, enabled the discovery of 12 and 13, potent, selective, and structurally completely novel Dot1L inhibitors.



Protein lysine methyltransferases (PKMT) are a class of 51 enzymes which catalyze the transfer of a methyl group from co-factor S-adenosylmethionine (SAM) to the lysine ε-amino group of proteins.¹⁻³ Typically, PKMTs mono-, di-, or tri-methylate lysine residues of histone proteins, thereby controlling chromatin structure and transcriptional accessibility of genes. However, a number of other non-histone substrate proteins have also been identified.⁴

Dot1L is a 165 kD protein comprised of 1537 amino acids. The catalytic domain of Dot1L, residing in the Nterminal 400 amino acids, is structurally distinct to the SET-domain containing protein lysine methyltransferases and clusters more closely with the catalytic domain of a related class of proteins, the protein arginine methyltransferases (PRMT).^{3,5} Dot1L is the only known enzyme to methylate lysine 79 of histone 3 (H3K79), with the H3K79me2 mark being associated with active transcription.⁶⁻⁸ Under physiological conditions, Dot1L is critical for normal hematopoiesis,^{9,10} however, misdirected catalytic activity is believed to be causative for a subset of acute leukemias.^{11,12} Several oncogenic fusion proteins including MLL-ENL, MLL-AF4 and MLL-AF9 aberrantly recruit Dot1L to ectopic loci, leading to local hypermethylation of H3K79 and misexpression of genes (including HoxA) which drive the leukemic phenotype. Inhibition of the methyltransferase activity of Dot1L in MLL-rearranged leukemias (mixed lineage leukemia, MLL) is predicted to ACS Paragon Plus Environment

reverse ectopic H₃K₇₉ methylation, leading to repression of leukemogenic genes (HoxA9, Meis1) and tumor growth inhibition.13 The recent quest for Dot1L inhibitors has been spearheaded by Epizyme and culminated in the discovery of EPZ-5676, a SAM-competitive, nucleosidecontaining Dot1L inhibitor, which is currently being evaluated in MLL patients in Phase 1b clinical trials.¹⁴ The agent is administered by uninterrupted, continuous intravenous (i.v.) infusion due to its physicochemical properties.15 Other research groups, based at Baylor College and the Structural Genomics Consortium, have identified additional Dot1L inhibitors structurally related to EPZ-5676.¹⁶⁻¹⁹

Herein, we describe the discovery of SAM-competitive, structurally novel Dot1L inhibitors which bind to an induced pocket adjacent to the SAM binding site. The compounds display cellular potency and molecular properties suitable for further optimization.

A variety of approaches to discover chemical starting points in the Dot1L program were taken and included virtual, biophysical, and biochemical screens. In our highthroughput screening (HTS) campaign, a luminescencebased coupled assay was used,²⁰ which had been adapted for measuring Dot1L-catalyzed nucleosome methylation and miniaturized for allowing screening of the Novartis Compound Collection in 1536-well plates (primary hit rat

o.47%, threshold 50% inhibition at 26.7 uM compound concentration). For IC₅₀ and K_i determinations, a scintillation proximity assay (SPA) format was employed as a more direct readout, quantifying enzyme activity via the amount of the transfer of ³H-labelled methyl group from SAM to biotinylated nucleosomes. Experiments described herein were carried out at SAM concentration equal to K_M while nucleosome concentration was in excess relative to K_M (Supporting Information). In the HTS, a commercial compound sample, assigned to structure 1, was identified as an inhibitor of Dot1L (IC₅₀ = 14 µM) (Figure 1). Analysis of the screening sample revealed that the active principle is a regioisomeric impurity, compound 2 (IC₅₀ = 4.4 µM), whereas purified 1 had no inhibitory activity up to 100 µM.



Figure 1. Protein structure-guided optimization of HTS hit 1.

The crystal complex structure with compound 2 was solved using the catalytic domain of Dot1L (Figure 2, PDB code 5drt). To our surprise, 2 binds to a newly formed pocket adjacent to the SAM binding site, inducing a fundamentally different conformation of DotiL compared to the conformation of SAM-bound Dot1L (Figure 2B).²¹ A flexible loop comprised of amino acids 126-140, which forms the "lid" of the SAM binding pocket in the SAMbound state, is completely rearranged, partially disordered and forms a part of the induced binding pocket of 2. We and others subsequently observed that only SAM and SAH are able to induce a closed lid (e.g. PDB code 3uwp), whereas inhibitors rearrange and often engage the flexible loop.^{16,22} Other residues, such as Leu143, Met147, Phe239 and Tyr312, also undergo conformational shifts compared to the SAM-bound complex and are forming the bottom lining of the binding pocket of 2. Remarkably, 2 forms five direct hydrogen bonds with Dot1L (Figure 2A). The central urea NH's are coordinating Asp161, whose side chain rotation is cushioned by crystal waters. The interaction of the urea carbonyl with Asn241 varies between 3.2 and 3.8 Å across several cogeneric cocrystal structures. The urea of 2 interacts with Dot1L in a comparable fashion to the urea-containing Dot1L inhibitor EPZ-4777.¹⁶ The second hydrazide amide forms a hydrogen bond to Gly163 perpendicular to the urea plane and stacks

onto the Ser164-Gly165 backbone amide. The tetrazole forms a hydrogen bond with Gln168 and the tetrazolephenoxy motif engages the flexible loop through hydrophobic contacts, while the *p*-chlorophenyl substituent slides between the Asn241-containing loop and an α -helix, occupying a similar position as the *tert*-Bu group of EPZ-5676 (Figure 2C).



Figure 2. (A) X-ray cocrystal structure of DotL (grey) with **2** (blue) (PDB code 5drt). Amino acid side chains engaged in key interactions with the ligand are illustrated as sticks. Key polar interactions of DotL and **2** are shown as red dotted red lines, the tetrahedral coordination of the two waters ligating Asp161 as grey dotted lines. (B) Comparison of the flexible loop (amino acids 126-140) in DotL bound to **2** (blue, protein grey) (PDB code 5drt) and SAM (green) (PDB code 3qow). (C) Overlay of ligands **2** (blue) (PDB 5drt) and EPZ-5676 (green) (PDB code 4hra) bound to DotL in the same view as 2A.

Structural analysis of the Dot1L-compound **2** complex revealed a large hydrophobic pocket, which is only partially filled by the *p*-chlorophenyl substituent. Structurebased chemistry optimization led to the discovery of the significantly more potent compound **3** (IC₅₀ = 20 nM) in which the *p*-chlorophenyl substituent was replaced by an *N*-arylated indole system (Figure 1). The indole core is deeply buried in a hydrophobic cleft while the *N*-phenyl moiety engages in a face-edge interaction with Phe131 thereby further stabilizing the flexible loop of Dot1L (Figure **3**, PDB code 5dry). The o-Cl substituent does not only pre-organize the bioactive conformation of **3**, but also fills a hydrophobic nook of Dot1L.



Figure 3. Overlay of X-ray cocrystal structures of Dot1L with **2** (ligand blue, protein grey) (PDB code 5drt) and **3** (ligand green, parts of the protein green) (PDB code 5dry). Prominent amino acid residues are illustrated as sticks.



Figure 4. Fragmentation of 3 to urea 4. Discovery of urea replacements 5-9.

Inhibitor 3 shows low permeability as assessed in the parallel artificial membrane permeation (PAMPA) assay, which might not be surprising, considering its number of hydrogen bond acceptors and donors (Supporting Information). It is likely that the weak flux of 3 contributes to the lack of oral bioavailability in rats and weak cellular activity.

Considering the poor physicochemical properties, the complex functional group arrangement and high molecular weight (MW = 521), we regarded 3 unsuitable for further stepwise optimization. Therefore, we decided to pursue a more radical fragmentation approach in which the entire aryl-tetrazole system was eliminated leading to urea 4, which showed residual biochemical potency (IC_{50} = 40 μ M) (Figure 4). Compound 4 was viewed a more suitable starting point for additional chemistry activities. Initial work focused on the identification of urea replacements with the goal to reduce the risk of low solubility and poor permeability often associated with ureas. A series of functional groups were discovered as urea isosters capable of coordinating the central Asp161 side chain of Dot1L. Diamino-pyrimidines 5 and 6 (in their protonated forms), amino-pyrimidone 7, aminopyridazine 8 (with its polarized C-H group) and aminoimidazolone 9 inhibit Dot1L with improved potency compared to urea 4 (Figure 4). It is worth noting that the methyl group in the 2-position of the indole and the insertion of a nitrogen atom into the N-phenyl substituent typically contribute to the potency by a 5-fold reduction of the IC₅₀. Remarkably, the *N*-methyl group of **6** increases inhibitory activity against Dot1L by almost 20-fold over 5, possibly through a pre-organization effect (discussed below).

The optimization opportunities for urea replacements **6** and **7** were evaluated in detail. In line with our structural understanding and using molecular modeling, the diamino-pyrimidine of **6** and the amino-pyrimidone of **7** undergo polar interactions with Asp161 (Figure 5). The *N*-aryl 2-methylindole core of both analogs is shape-

complementary to the hydrophobic pocket formed by the Asn241-containing loop and the α -helix. Importantly, the carbonyl group of amino-pyrimidone 7 and the *N*-methyl group of diamino-pyrimidine 6 interact with Dot1L in distinct ways: The *N*-methyl group of 6 does not undergo any specific contacts with Dot1L, but rather plays a role in pre-organizing the ligand by pointing into the *N*-aryl moiety (van der Waals contact). In contrast, the carbonyl group of 7 engages in polar interactions with Asn241 and the backbone NH of Tyr136, thereby tightly stabilizing the flexible loop of Dot1L (Figure 5). Consequently, potential growth vectors in ortho- and meta-position of the *N*-pyridyl moiety of pyrimidone 7 are blocked by the loop, whereas they are open for pyrimidine 6 (Figure 5).



Figure 5. Overlay of X-ray cocrystal structures of Dot1L with **6** (ligand green, parts of the protein green) and **7** (ligand blue, protein grey) bound to Dot1L.²³ Amino acid side chains engaging in key interactions with the ligands are illustrated as sticks, polar contacts are highlighted as dotted red lines.

Growth vectors (solid red arrow) open in ortho and metaposition for **6**, but blocked for **7** due to tighter engagement of the flexible loop.

Both growth vectors, extruding from either the orthoor the meta-position were explored for **6**. Guided by abovementioned protein-structural considerations, the direct attachment of a pyridone in meta-position, leading to compound **10** (Figure 6), turned out to have a beneficial impact on potency ($IC_{50} = 10 \text{ nM}$). Growing **6** from the ortho-position was best achieved with an aryloxy group while eliminating the Cl substituent in the "other" orthoposition, as e.g. in compound **11** ($IC_{50} = 150 \text{ nM}$) (Figure 6).



Figure 6. Optimization of **6** by growing from the metaposition, leading to **10**, or from the ortho-position, leading to **11**.

Analysis of the X-ray cocrystal structures of **10** and **11** bound to Dot1L revealed some noteworthy aspects (Figure 7, PDB codes 5dsx and 5dt2): both compounds stack tightly with Phe243 and engage the flexible loop through their diaryl and diaryl ether groups, respectively. Furthermore, the pyridone carbonyl of **10** forms a hydrogen bond to the side-chain NH_2 of Asn241. The *N*-phenyl group of **11** moves deep into the hydrophobic pocket, taking the place of the o-Cl substituent of **10**.



Figure 7. Overlay of ligands **10** (ligand blue, protein grey) (PDB code 5dsx) and **11** (ligand green, parts of the protein green) (PDB code 5dt2) bound to Dot1L based on X-ray cocrystal structures. Both aromatic extensions engage Phe243 and the Pro130-Phe131 motif of the flexible loop.

Chemistry optimization continued on both structural subtypes exemplified by compounds 10 ("meta") and 11 ("ortho"). A number of opportunities were discovered to further improve potency. For example, the 2-methyl indole core of 10 can be replaced by a 2-chloro benzothiophene core, leading to the highly potent Dot1L inhibitor 12 (IC₅₀ = 1.8 nM) (Table 1). Along different lines, the aromatic stacking capacity of 11 can be dramatically improved by substituting the outer phenyl group of **11** by an aza-benzimidazole group, resulting in 400-fold more potent compound 13 (IC₅₀ = 0.4 nM). Both compounds 12 and 13 are characterized by long on-target residence times as assessed by surface plasmon resonance experiments (τ = 45 min for 12 and >240 min for 13 (the detection limit of our internal SPR assay). Most importantly, 12 and 13 potently suppress H₃K₇₉ dimethylation (IC₅₀ = 23 nM and 16nM, respectively), the direct product of the Dot1Lcatalyzed reaction, as well as the activity of the HoxA9 promoter (IC₅₀ = 384 nM and 340 nM, respectively) in cellular systems (Table 1). We speculate that the significantly higher SAM concentration in cells ($\approx 200 \ \mu M$)²⁴ compared to our biochemical assay (200 nM) is contributing to the considerable cell-enzyme IC_{50} shift. Finally, both compounds effectively inhibit proliferation of the human MLL-rearranged leukemia cell line MV4-11 carrying the oncogenic MLL-AF4 fusion (IC₅₀ = 85 nM and 128 nM, respectively) (Table 1). Importantly, 12 and 13 display a favorable selectivity profile against a panel of 22 PKMTs and PRMTs with no inhibitory activity up to 50 µM.

Table 1. Biochemical, biophysical and cellular characterization of lead compounds 12 and 13



All data are the results of at least two assay runs with the mean value reported. The coefficient of variation was less than 60% in all cases. Biochemical IC_{50} values were determined at K_M for SAM. K_i values were determined by applying the Morrison tight binding model (Supporting Information).

By virtue of its in vitro potency and selectivity as well as good molecular properties (e.g. improved permeability vs 3, Supporting Information), 12 was evaluated in PK experiments in rats (Table 2). After intravenous bolus administration (1 mg/kg), the compound showed high total blood clearance, high volume of distribution and a moderate half-life. After oral administration in suspension (15 mg/kg), blood levels were detected up to 24 h. The maximum total concentration (ca. 225 nM) was reached after 3.5 h, total exposure was 2370 nM·h. Oral bioavailability at 15 mg/kg was medium, reaching 40%.

Table 2. Pharmacokinetic parameters of 12 in maleSprague Dawley rats following IV and PO dosing

	1 mg/kg iv ^a	15 mg/kg po ^b
Cl [mL/min/kg]	84	n.a.
V _d [L/kg]	10	n.a.
t _{1/2} [h]	1.7	n.a.
$AUC_{\text{o-inf}}[nM \cdot h]$	396	2370
C _{max} [nM]	n.a.	225
BAV [%]	n.a.	40

^aNMP:PEG200 (30:70), ^bMEPC5:Water (10:90).

Compound 12 can be conveniently prepared from commercially available 5-nitrobenzo[b]thiophene-2carboxylic acid 14 (Scheme 1). Thermal decarboxylation, followed by bromination and a lithiation-chlorination sequence, yielded intermediate 15. Nitro reduction to 16 and nucleophilic aromatic substitution with chloropyrimidine 24 delivered key intermediate 17, suitable for late state derivatization in the 3-position of the benzothiophene core. Suzuki coupling of 17 with the customized boronic ester 20 produced lead compound 12 in a convergent fashion (Scheme 1).

Scheme 1. Synthetic route to compound 12.^a



^aReagents and conditions: (a) Cu, quinoline, 2 h, 150 °C, 81%, (b) Br₂, CHCl₃, 30 h, 60 °C, 93%, (c) NCS, TMPMgCl*LiCl, THF, toluene, -75 °C, 80%, (d) RaNi, H₂, EtOH, rt, 96%, (e) **24**, *p*-TsOH*H₂O, DMF, 15 h, 80 °C, 66%, (f) **20**, Na₂CO₃ 2 M aq, Pd(amphos)Cl₂, MeCN, 30 min, 80 °C, 25%, (g) **22**, Na₂CO₃ 2 M aq, PdCl₂(dppf)*DCM, DME, 80 °C, 68%, (h) NaNO₂, KI, HCl 4 M, H₂O, MeCN, 5 °C to rt, 36%, (i) Bis(pinacolato)diboron, KOAc, PdCl₂(dppf)*DCM, dioxane, 10 h, 110 °C, 49%, (j) Bis(pinacolato)diboron, KOAc, PdCl₂(dppf)*DCM, dioxane, 3 h, 100 °C, 80%, (k) Boc₂O, DMAP, DCM, rt, 98%.

In conclusion, we have discovered potent, selective and orally bioavailable Dot1L inhibitors which have a novel binding mode and are structurally unique. Future work on further improving potency and *in vivo* exposure based on lead compounds **12** and **13** is warranted.

ASSOCIATED CONTENT

PyMol was used for structural visualization and figure preparation.²⁵ Synthetic procedures, compound characterization and assay protocols. This material is available free of charge via the Internet at http://pubs.acs.org.

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Author Contributions

The manuscript was written through contributions of all authors. / All authors have given approval to the final version of the manuscript.

ACKNOWLEDGMENT

We are grateful to our medicinal chemistry colleagues Ross Strang, Christian Ragot, Rainer Tschan, Michael Hediger, and Stephan Kläusler for their scientific contributions. We thank Kim Twesten and Anton Kessler for developing and running the high-throughput screening assay, and acknowledge supply of Dot1L enzyme and nucleosome substrate by the laboratories of Lukas Leder, Dirk Erdmann and Kehao Zhao, respectively. Jidong Zhu is thanked for valuable and inspiring discussions on target biology, Steffen Renner for in silico hit list analysis and Justin Gu for verification of screening hits in an orthogonal LC-MS assay. We thank Paul Westwood, Kerstin Pollehn, Shin Numao, Claudio Thoma and Andreas Theuer for developing and supporting projectrelated biophysical, biochemical and cellular assays. The authors would like to acknowledge Elke Koch, Céline Be, Aurelie Winterhalter, Aude Izaac, Julia Klopp and Patrick Graff for their contributions to the protein structural work. The crystallographic experiments were performed on the X10SA beamline at the Swiss Light Source, Paul Scherrer Institut, Villigen, Switzerland supported by the team of Expose GmbH. We thank Sandrine Desrayaud and her team for performing the rat PK studies as well as Bernard Faller and his team for contributing to the in vitro ADME studies.

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