19. 1,7-Trimethylenenorbornane. A Novel Member of the 'Adamantaneland'

Preliminary communication

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Dedicated to Professor Dr. Edgardo Giovannini on the occasion of his seventieth birthday

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Summary

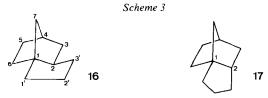
A synthesis of the novel $C_{10}H_{16}$ hydrocarbon 1,7-trimethylenenorbornane (13), one of the 19 members of the adamantane family, is described.

1,7-Trimethylenenorbornane (13)¹) is a member of the 'adamantaneland' [1], a set of 19 isomeric tricyclic $C_{10}H_{16}$ compounds. In the present communication we describe a synthesis of this hithertho unknown member²)³). The dihydropentalene 2, readily available by the reaction of 8,8-dibromobicyclo [5.1.0]octa-2,4-diene (1) with methyl lithium according to *Baird & Reese* [3], was chosen as suitable starting material. An obvious approach to construct the 1,7-trimethylenenorbornane skeleton is by a [4+2] cycloaddition of 2 and a ketene equivalent [4]. Addition of a-acetoxyacrylonitrile to the reaction mixture of 1 and methyl lithium in diethyl ether at -40 to 0° yielded with full regio- and stereoselectivity in a 9:1 ratio 6^4) (an adduct of 2) and 4 (an adduct of the isomerized triene 3 [3])⁵). Hydrogenation

- Tricyclo [4.2.2.0^{1,5}]decane or 1,3a-Ethanoperhydropentalene.
- An analogous synthesis of 13 by L. Skattebøl et al. (University of Oslo) as well as studies on the Lewis acid catalyzed rearrangement of 13 by Schleyer et al. (University of Erlangen-Nürnberg) will be described by these authors elsewhere. We thank them for discussing results prior to publication.
- For a recent synthesis of two other representatives see e.g. [2].
- 4) 6 can be separated from the reaction mixture of 6 and 4 by filtration on silicagel, subsequent recrystallisation and sublimation.
- 5) Addition of dimethylacetylenedicarboxylate to the dihydropentalenes 2 and 3 yielding the [4+2] cycloaddition products 18 and 19, respectively, was already reported by *Baird & Reese* [3]. However no further transformations were described.

in the presence of 5% Pd/CaCO₃ as a catalyst and subsequent hydrolysis in KOH/CH₃OH led to the saturated ketones 9 and the already known isomer 5 [5], which were easily separated by chromatography on silicagel. The ketone 9 was also obtained by inverting the reaction sequence, i.e. by first hydrolyzing 6 to the unsaturated ketone 8 and subsequently hydrogenating the latter. Conversion of ketone 9 to the title hydrocarbon 13 was achieved either by Wolff-Kishner reduction (approximately 30% overall yield relative to the starting material 1) or by photochemical reduction of the acetate 12 in hexamethylphosphoric triamide (HMPT)/ water 95:5 according to a procedure described by Pète et al. [6]. Reduction of ketone 9 with sodium borohydride in ethanol gave the endo- and the exo-alcohols 10 and 11 in a ratio of 1:4. Both were reconverted to ketone 9 on oxidation with pyridinium chlorochromate. The acetate 12 was formed from the alcohol 11 by standard treatment with acetic anhydride in pyridine.

The ¹³C-NMR.-spectrum of **13** shows 10 signals: 7 triplets, 2 doublets and 1 singlet⁶). Only two other members of the adamantane family are also consistent with this multiplicity. As **13** they are trimethylenenorbornanes, the 1,2exo-hydrocarbon **16** [1b]⁷) and its 1,2endo-isomer **17**⁸). Further evidence for the 1,7-trimethylene-



norbornane skeleton of the compounds 6-13 is given by the ¹H- and ¹³C-NMR.-spectra of the primary cycloaddition product 6⁹) indicating 4 olefinic C-atoms each bearing one H-atom in contrast to the spectral data of adduct 4¹⁰).

The orientation of the trimethylene bridge towards C(2) is deduced from the result of reducing ketone 9, where a preferred attack of sodium borohydride from below was observed (10/11 1:4)¹¹) and the ease of the quantitative rearrangement of the exo-alcohol 11 in thionylchloride at room temperature to 14, which on reduction with magnesium and subsequent hydrolysis yielded the well known 2exo, 3exo-trimethylenenorbornane (15). On the other hand, the endo-alcohol 10 reacted much slower under analogous reaction conditions and gave a mixture (approx. 10:1) of the same chloride 14 and some not further identified chlorides.

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- 6) 13C-NMR. (CDCl₃) of **13**: 21.51, 28.77, 29.04, 30.32, 32.30, 32.84 and 37.56 (7t, C(2), C(3), C(5), C(6), C(1'), C(2') and C(3')); 36.87 (d, C(4)); 60.47 (d, C(7)); 58.66 (s, C(1)).
- 7) 13C-NMR. (CDCl₃) of **16**: 26.58, 28.76, 29.61, 33.74, 33.74, 40.04 and 41.13 (7t, C(3), C(5), C(6), C(7), C(1'), C(2') and C(3')); 39.29 (d, C(4)); 48.14 (d, C(2)); 56.49 (s, C(1)).
- 8) The synthesis of **17** will be described in a separate communication by *F.J. Jäggi & C. Ganter*, in preparation. ¹³C-NMR. (CDCl₃): 23.26, 27.54, 27.68, 28.23, 29.87, 32.26 and 40.41 (7t, C(3), C(5), C(6), C(7), C(1'), C(2') and C(3')); 43.48 (d, C(4)); 53.62 (d, C(2)); 55.86 (s, C(1)).
- 9) 13C-NMR. (CDCl₃) of 6: 131.91, 133.45, 137.47 and 144.23 (4 d).
- ¹⁰) ¹³C-NMR. (CDCl₃) of 4: 112.12, 124.01 and 144.93 (3*d*); 164.06 (*s*).
- 11) In an unhindered case a ratio of approximately 9:1 in favour of an *endo*-alcohol would be expected; see *e.g.* [7].