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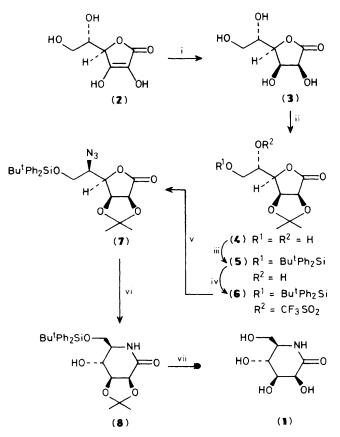
Enantiospecific Synthesis of P-Mannono-δ-lactam from Vitamin C

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An eight-step synthesis of the glycosidase inhibitor p-mannono-δ-lactam from vitamin C is described.

Very recently, the potential of polyhydroxylated δ -lactams as a new class of glycosidase inhibitors has been indicated.¹ D-Mannono- δ -lactam (1), previously formed by microbial



 $\begin{array}{l} \textbf{Scheme 1. } \textit{Reagents: i, Pd, H_2, H_2O(100\%); ii, Me_2CO/HCl then aq.} \\ AcOH (68\%); iii, ButPh_2SiCl, pyridine (95\%); iv, (CF_3SO_2)_2O, CH_2Cl_2, pyridine (92\%); v, Bu_4^nNN_3, tetrahydrofuran (76\%); vi, H_2, Pd(OH)_2, EtOAc (88\%); vii, aq. CF_3CO_2H (85\%). \\ \end{array}$

oxidation of nojirimycin B, is a potent inhibitor of rat epididymal α -mannosidase and of apricot β -glucosidase.² This communication reports the first enantiospecific synthesis of (1) from a non-carbohydrate precursor, vitamin C (2).

The strategy involves a stereoselective hydrogenation from the less hindered side of the enediol (2) to secure the chirality of the resultant C-2 and C-3 hydroxy groups and introduction of nitrogen with inversion of configuration at C-5. Thus hydrogenation of (2) gave stereospecifically the saturated lactone (3),³ which was bis-acetonated and then partially hydrolysed to the diol (4), † m.p. 130–132 °C (Scheme 1). The primary hydroxy group in (4) was protected by O-silylation to give (5), $[\alpha]_D^{20} + 61.1^{\circ}$ (c 1.4 in CHCl₃), which was subsequently esterified to give the trifluoromethanesulphonate (6). A nucleophilic displacement reaction of (6) with tetrabutylammonium azide afforded the azido lactone (7), m.p. 97–98 °C; $[\alpha]_{D^{20}}$ +6.5° (c 1.4 in CHCl₃), which was hydrogenolysed to form the protected lactam (8), $[\alpha]_{D}^{22}$ $+35.8^{\circ}$ (c 0.6 in CHCl₃). Hydrolysis then furnished D-mannono- δ -lactam (1), m.p. 170–172 °C; $[\alpha]_D^{22} + 2.0^\circ$ (c 0.8 in water) {lit.² m.p. 169–170 °C, $[\alpha]_{D^{20}} + 1.6^{\circ} (c \ 1.0 \text{ in water})$ }.

Synthetic applications of vitamin C, enhanced by its cheapness and availability, should attract more attention from organic chemists.

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References

- 1 G. W. J. Fleet, N. G. Ramsden, R. A. Dwek, T. W. Rademacher, L. E. Fellows, R. J. Nash, D. St. C. Green, and B. Winchester, J. Chem. Soc., Chem. Commun., 1988, 483.
- 2 T. Niwa, T. Tsuruoka, H. Goi, Y. Kodama, J. Itoh, S. Inouye, Y. Yamada, T. Niida, M. Nobe, and Y. Ogawa, J. Antibiot., 1984, 37, 1579.
- 3 G. C. Andrews, T. C. Crawford, and B. E. Bacon, J. Org. Chem., 1981, 46, 2976.

[†] All new compounds gave satisfactory analytical and spectral data.