

STRUCTURE OF DESACYLSAPONINS OBTAINED FROM THE BARK OF *QUILLAJA SAPONARIA*

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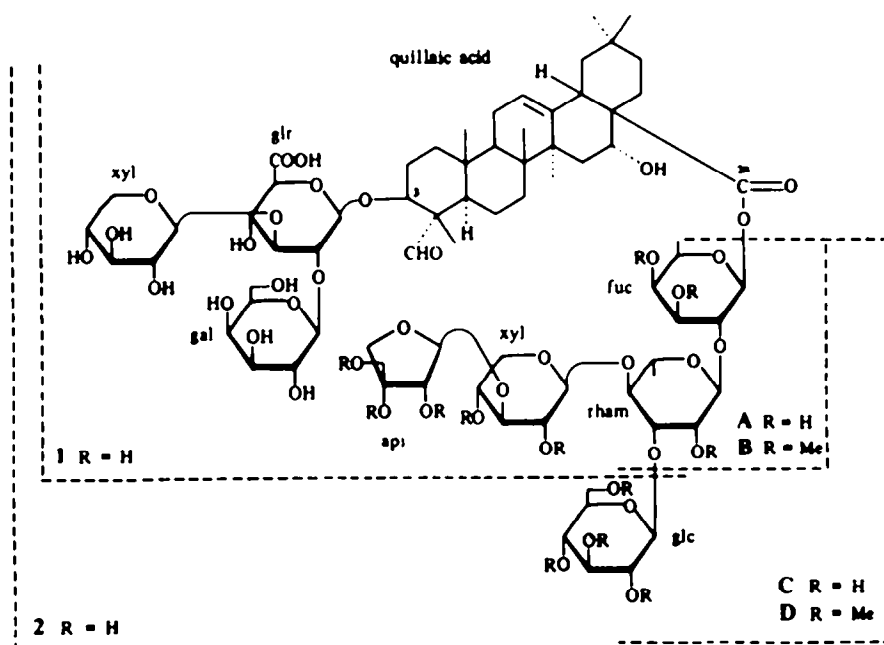
Abstracts—A triterpenoid saponin mixture (so-called quillajasaponin) obtained from the bark of *Quillaja saponaria* was treated with weak alkali and two major desacylsaponins were isolated. On the basis of chemical and spectral evidence, they were determined as 3-O-β-D-galactopyranosyl-(1 → 2)-[β-D-xylopyranosyl-(1 → 3)]-β-D-glucopyranosyl quillaic acid 28-O-β-D-apiofuranosyl-(1 → 3)-β-D-xylopyranosyl-(1 → 4)-α-L-rhamnopyranosyl-(1 → 2)-β-D-fucopyranoside and 28-O-β-D-apiofuranosyl-(1 → 3)-β-D-xylopyranosyl-(1 → 4)-[β-D-glucopyranosyl-(1 → 3)]-α-L-rhamnopyranosyl-(1 → 2)-β-D-fucopyranoside. Diazomethane degradation providing selectively the 28-O-glycoside from the 3,28-O-bisglycoside was a useful method for the structure elucidation.

INTRODUCTION

The bark of *Quillaja saponaria* Molina, named quillaja bark (cortex quillajae), is known as a saponin crude drug and has been used as a detergent, dentifrice and expectorant [1]. The existence of a saponin mixture (designated as quillajasaponin), which was recently reported to have a

strong adjuvant activity [2, 3] and a plasma cholesterol lowering effect [4], was recognized but as for the constituents of the crude saponin, little was known except for quillaic acid [5, 6] and its monoglucuronide [7], which were obtained upon acid hydrolysis of the saponin. A study on the constituents of the quillajasaponin has been conducted in an attempt to isolate the physiologically active triterpenoid compounds. We report in this paper the structures of two desacylsaponins (quillaic acid 3,28-O-bisglycosides), DS-1 (1) and DS-2 (2), obtained by mild alkaline hydrolysis of the quillajasaponin.

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RESULTS AND DISCUSSION

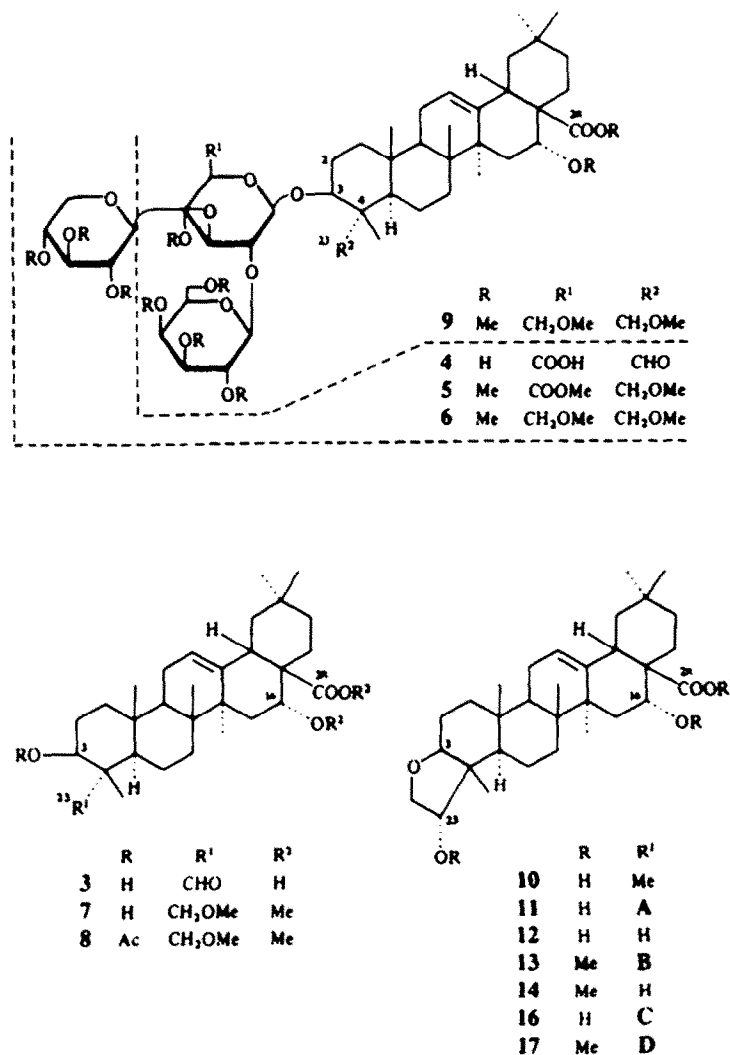
The methanol extract of the bark was fractionated by the ordinary procedure, as described in the Experimental, to give a saponin fraction (quillajasaponin). On treatment with 6% NaHCO_3 in 50% methanol, the fraction afforded a desacylsaponin mixture, which was separated by normal and reverse phase column chromatography to give two major compounds, DS-1 (1) and DS-2 (2). Compound 1, as well as 2, showed a single spot on normal and reverse phase TLC, respectively.

Compound 1 was hydrolysed with acid to yield glucuronic acid (Glr), galactose (Gal), xylose (Xyl), fucose (Fuc), rhamnose (Rham), apiose (Api) and quillaic acid (3). Compound 2 yielded the same products as those of 1 and in addition glucose (Glc). The ^{13}C NMR spectra of 1 and 2 showed seven and eight anomeric carbon signals respectively. The FAB mass spectra of 1 and 2 revealed the molecular ion peaks as a cationized cluster ion at m/z 1589 $[\text{M} + 2\text{K} - \text{H}]^+$ and 1751 $[\text{M} + 2\text{K} - \text{H}]^+$, respectively. These data indicate 1 to consist of 1 mol each of 3, Glr, Gal, Xyl, Fuc, Rham, Api and another one, Xyl or Api, and 2 to consist of the same components as those of 1 and 1 mol of Glc. Both compounds were suggested to have a 28-*O*-glycosidic linkage since in their ^{13}C NMR spectra

the signals due to C-28 of the aglycone part were observed at δ 176.0 (in 1) and 176.3 (in 2) [8].

When compounds 1 and 2 were treated with 2% potassium hydroxide in 50% ethanol, they afforded the same prosapogenin (4), which was hydrolysed with acid to yield 3, Glr, Gal and Xyl. Compound 4 showed in the ^{13}C NMR spectrum three anomeric carbon signals and three carbon signals at δ 24.6, 54.9 and 84.2 due to C-2, C-4 and C-3 of quillaic acid having a sugar moiety at its 3-position [9], and showed the molecular ion peak at m/z 995 $[\text{M} + \text{K}]^+$ in the FAB mass spectrum. These data indicate that 4 is the 3-*O*-glycoside of 3 and consists of 1 mol each of 3, Glr, Gal and Xyl. Since the aldehyde group and Glr unit in 4 were thought to complicate its structure elucidation, compound 4 was converted to the more stable compound 6 as follows.

Reduction of 4 with sodium borohydride followed by methylation of the product by the Hakomori method [10] afforded compound 5. Compound 5 was again reduced and methylated to yield compound 6, which was methanolysed to give an aglycone (7) and three methylated sugars. An acetate (8) of 7 showed in the ^1H NMR spectrum the signals of one acetoxy and three methoxy groups, together with triplet-like signals at δ 4.92 ascribable to the proton at C-3 bearing the acetoxy group



[11]. Therefore 7 and 8 were regarded as the 16,23-di-*O*-methyl ether of 16 α -hydroxyhederagenin methyl ester and its 3-*O*-acetate, respectively. The methylated sugars were identified as methyl pyranosides of 2,3,4-tri-*O*-methyl-xylose (S-1), 2,3,4,6-tetra-*O*-methyl-galactose (S-2) and 4,6-di-*O*-methyl-glucose (S-3). Therefore, compound 4 has a branched trisaccharide, xylopyranosyl-[galactopyranosyl]-glucuronopyranose, combined with the 3-hydroxyl group of 3, and the Xyl and Gal units are attached to the 2- and 3-, or 3- and 2-hydroxyl groups of Glc, respectively.

When compound 6 was treated with dilute hydrochloric acid in methanol, a major product was obtained. Methylation of the product afforded compound 9, which gave on methanolysis compound 7, S-2 and methyl 3,4,6-tri-*O*-methyl-glucopyranose (S-4). This indicated compound 9 to be the permethylate of the galactosyl-(1 \rightarrow 2)-glucoside of 16 α -hydroxyhederagenin. Therefore, if Xyl, Gal and Glc are assumed to be the most commonly found D-series, the oligosaccharide moiety of 4 is D-galactopyranosyl-(1 \rightarrow 2)-[D-xylopyranosyl-(1 \rightarrow 3)]-D-glucuronopyranose. The ^1H NMR spectrum of 5 showed three anomeric proton signals of sugar units as doublets with large *J* values (7, 7 and 8 Hz) indicating their β -linkage [12].

Consequently, compound 4 is quillaic acid 3-*O*- β -D-galactopyranosyl-(1 \rightarrow 2)-[β -D-xylopyranosyl-(1 \rightarrow 3)]- β -D-glucuronopyranoside, and therefore compounds 1 and 2 are thought to be 28-*O*-glycosides of 4.

In the course of experiments concerning the structure elucidation of 4, we found and reported [13] that the sugar aglycone linkage of 4 was cleaved to give the aglycone 10 (methyl 3 β ,16 α ,23 α -trihydroxy-3-*O*,23-methylenolean-12-en-28-oate) and the corresponding oligosaccharide residue by only treatment of 4 with diazomethane ether in methanol. If this procedure (diazomethane degradation) is applied to compounds 1 and 2, the 28-*O*-glycoside of 12 must be obtained.

Treatment of compound 1 with diazomethane ether in methanol afforded a less polar compound (11). Compound 11 revealed an ester carbonyl absorption (1735 cm^{-1}) in the IR spectrum, the signals of four anomeric carbons in the ^{13}C NMR spectrum and a molecular ion peak at m/z 1079 [$\text{M} + \text{Na}$] $^+$ in the FAB mass spectrum, and gave on acid hydrolysis Fuc, Rham, Xyl, Api and an aglycone (12), which was converted to 10 by methylation with diazomethane. These data indicated that 11 is the 28-*O*-tetraglycoside of 12 obtained by cleavage of the 3-*O*-glycosidic linkage in 1, and that the sugar moiety of 11 consisted of 1 mol each of Fuc, Rham, Xyl and Api.

Methanolysis of the permethylate (13) of 11 gave 14 (the 16,23-di-*O*-methyl ether of 12), methyl 2,3,5-tri-*O*-methyl-apiofuranoside (S-5) and methyl pyranosides of 2,4-di-*O*-methyl-xylose (S-6), 2,3-di-*O*-methyl-rhamnose (S-7) and 3,4-di-*O*-methyl-fucose (S-8). This indicated that the sugar moiety of 11 is linear in structure and that apiofuranose is located at the terminal. Compound 11 was treated with 2% hydrochloric acid in methanol to yield 15 together with 12 and methyl apiofuranoside. Compound 15 showed a molecular ion peak at m/z 479 [$\text{M} + \text{Na}$] $^+$ in the FAB mass spectrum and revealed one methoxyl and three anomeric carbon signals in the ^{13}C NMR spectrum, and afforded Xyl, Rham and Fuc on acid hydrolysis. These data indicated that 15 was a methyl glycoside of a trisaccharide consisting of Xyl, Rham and Fuc. The

detailed assignment of the ^{13}C NMR spectrum of 15 was made by taking the glycosylation shift [14, 15] into account and by comparison with the spectra of methyl fucopyranoside and gleditsia saponin [16] as shown in Table 1. The result of methanolysis of 13 indicated that if Xyl, Fuc and Rham are assumed to be the most commonly found D, D and L-series, compound 15 is the methyl glycoside of β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 2)- α -D-fucopyranose. The above facts suggested that the sugar moiety of 11 is D-apiofuranosyl-(1 \rightarrow 3)- β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 2)-D-fucopyranose since apiose in glycosides is also usually found as the D-series.

The configuration of the D-fucopyranose unit was regarded to be β by the *J* values of its anomeric proton signal (doublet, *J* = 8 Hz) [12] in the ^1H NMR spectrum of 13. The D-apiofuranose unit was considered to have the β -configuration by comparison of its anomeric carbon signal (δ 111.1) in the ^{13}C NMR spectrum of 11 with those of the methyl β - and α -D-apiofuranosides (β -anomer: δ 111.3; α -anomer: δ 104.4). Therefore, it follows that compound 11 is the 28-*O*- β -D-apiofuranosyl-(1 \rightarrow 3)- β -D-

Table 1. ^{13}C NMR spectral data ($\text{C}_6\text{D}_6\text{N}$) of compounds 15, 18 and reference compounds

C	15	18	Reference compounds*
			[F]
1'	100.5	100.4	101.6
2'	78.4	77.9†	70.0
3'	69.9	69.9	71.6
4'	73.3	73.2	73.2
5'	66.6	66.6	66.9
6'	17.0	17.0	17.1
OMe	54.9	54.9	55.1
			[R]
1"	104.2	104.0	101.2
2"	71.6	71.1	71.9
3"	72.7	77.7†	72.6
4"	84.5	82.7	83.9
5"	68.1	68.1	68.4
6"	18.4	18.7	18.7
			[X]
1"	107.0	105.1‡	106.9
2"	76.0	75.4§	76.0
3"	78.5	78.7	78.6
4"	70.9	71.1	71.0
5"	67.4	67.0	67.4
			[G]
1"		104.7‡	105.5
2"		75.2§	74.9
3"		78.7	78.3
4"		71.1	71.6
5"		78.7	78.3
6"		62.2	62.7

* [F]: Methyl α -D-fucopyranoside; [R] and [X]: rhamnose and xylose parts in β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl moiety in gleditsia saponin [16]; [G]: methyl β -D-glucopyranoside.

†,‡,§ Assignments may be reversed in each vertical column.

xylopyranosyl-(1 → 4)- α -L-rhamnopyranosyl-(1 → 2)- β -D-fucopyranoside of 12.

Consequently, due to the structures of compounds 4 and 11, compound 1 is characterized as 3-O- β -D-galactopyranosyl-(1 → 2)-[β -D-xylopyranosyl-(1 → 3)]- β -D-glucuronopyranosyl quillaic acid 28-O- β -D-apiofuranosyl-(1 → 3)- β -D-xylopyranosyl-(1 → 4)- α -L-rhamnopyranosyl-(1 → 2)- β -D-fucopyranoside.

The structure of the 28-O-oligosaccharide moiety in compound 2 was also determined by using the diazomethane degradation method as follows. On treatment with diazomethane, compound 2 afforded a less polar compound (16) which was presumed to be the 28-O-pentaglycoside of 12. On acid hydrolysis 16 gave 12, Fuc, Rham, Xyl, Api and Glc, and showed the molecular ion peak at m/z 1241 [$M + Na$]⁺ in the FAB mass spectrum, five anomeric carbon signals in the ¹³C NMR spectrum and an ester carbonyl absorption (1735 cm⁻¹) in the IR spectrum. These data indicated 16 is the 28-O-glycoside of 12 consisting of 1 mol each of 12, Fuc, Rham, Xyl, Api and Glc. Methanolysis of the permethylate (17) of 16 afforded 14 and five methylated sugars, S-5, S-6, S-8 and the methyl pyranosides of 2-O-methyl-rhamnose (S-9) and 2,3,4,6-tetra-O-methyl-glucose (S-10). These facts and the co-existence of 1 and 2 in the same plant material suggested that the sugar moiety of 16 is a pentasaccharide in which a glucopyranose is located on the 3-hydroxyl group of the rhamnopyranose unit in the sugar moiety of 11.

When compound 16 was treated with 2% hydrochloric acid in methanol in the same manner as 11, compound 18 was obtained together with 12 and methyl apiofuranoside. Since 18 gave on acid hydrolysis Xyl, Rham, Fuc and Glc and showed the signals of one methoxyl and four anomeric carbons in the ¹³C NMR spectrum and the molecular ion peak at m/z 619 [$M + H$]⁺ in the FAB mass spectrum, compound 18 was regarded as the methyl glycoside of tetraose derived from the sugar moiety of 16. The ¹³C NMR signals of 18 were assigned as shown in Table 1 by taking the glycosylation shift into account and by comparison with the spectra of 15 and methyl glucopyranoside. The permethylate (19) of 18 was methanolysed to give S-8, S-9, S-10 and S-1. Therefore, if Xyl, Glc, Fuc and Rham are assumed to be the commonly found D, D, D and L-series, compound 18 must be methyl β -D-xylopyranosyl-(1 → 4)-[β -D-glucopyranosyl-(1 → 3)]- α -L-rhamnopyranosyl-(1 → 2)- α -D-fucopyranoside. The site

of linkage of the Glc and Xyl units to the 3- and 4-hydroxyl groups of the rhamnose unit in 18 was confirmed as follows. Compound 19 was hydrolysed with 5% hydrochloric acid in methanol and the major product was methylated to give a compound (20) which showed no hydroxyl absorption in the IR spectrum and three anomeric proton signals in the ¹H NMR spectrum. Methanolysis of 20 afforded three methylated sugars, S-8, S-10 and methyl 2,4-di-O-methyl-rhamnopyranoside (S-11), which indicated that 20 is the permethylate of glucosyl-(1 → 3)-rhamnosyl-(1 → 2)-fucose.

Accordingly, compound 16 is the 28-O-apiofuranosyl-(1 → 3)- β -D-xylopyranosyl-(1 → 4)-[β -D-glucopyranosyl-(1 → 3)]- α -L-rhamnopyranosyl-(1 → 2)-D-fucopyranoside of 12. The configurations of both D-fucopyranose and D-apiofuranose were suggested to be β by the anomeric proton signal of fucopyranose (doublet, $J = 8$ Hz) [12] in the ¹H NMR spectrum of 17 and the anomeric carbon signals of apiofuranose (δ 111.2) in the ¹³C NMR spectrum of 16.

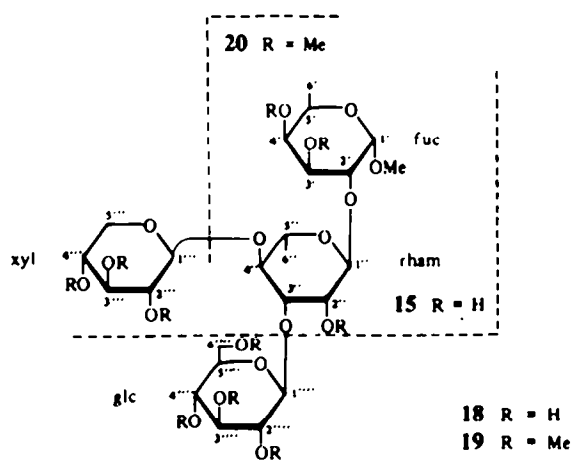
Based on the structure of 16, compound 2 was regarded to be 3-O- β -D-galactopyranosyl-(1 → 2)-[β -D-xylopyranosyl-(1 → 3)]- β -D-glucuronopyranosyl quillaic acid 28-O- β -D-apiofuranosyl-(1 → 3)- β -D-xylopyranosyl-(1 → 4)-[β -D-glucopyranosyl-(1 → 3)]- α -L-rhamnopyranosyl-(1 → 2)- β -D-fucopyranoside.

To our knowledge, two quillaic acid 3,28-O-bisglycosides have been reported [9, 17], but the two desacylsaponins, 1 and 2, reported here are different in their sugar moieties from those described so far. Diazomethane degradation was a useful method for structure elucidation of the triterpene 3,28-O-bisglycoside as described in this paper.

EXPERIMENTAL

All mps are uncorr. Optical rotations were recorded at 18–28° using a 1 dm cell. ¹H NMR spectra were taken at 100 MHz in CDCl₃ soln unless otherwise specified, using TMS as internal standard. ¹³C NMR spectra were recorded at 25 MHz in C₃D₈N (TMS as internal standard) unless otherwise noted, employing the FT mode. The EI- and FABMS were measured on a double focusing mass spectrometer. The former were taken with an accelerating potential of 3–6.5 kV and an ionizing potential of 30–75 eV, and the latter at 1.5–3 kV for the ion source and 6 kV for an Ar beam source; the spectra were obtained from glycerol solns unless otherwise specified. FDMS were taken at 2–3 kV for the field anode and at –5 kV for the slotted cathode plate, at an ion source pressure of ca 10⁻⁶ Torr and an emitter heating current of 18–24 mA. Conditions of GLC (FID mode): (a) glass column (1.2 m × 3 mm) packed with 10% 1,4-butanediol succinate on Shimalite W (60–80 mesh), column temp. 155°; (b) glass column (1.2 m × 3 mm) packed with 1% neopentyl glycol succinate polyester on Chromosorb W(AW) DMCS (60–80 mesh), column temp. 130°. Solvent systems of TLC [silica gel, C-8 (reversed phase) and Avicel]: (a) CHCl₃-MeOH-HOAc-H₂O (15:9:1:2); (b) EtOAc-MeOH-H₂O (8:1:0.1); (c) CHCl₃-MeOH (10:1); (d) C₆H₆-Me₂CO (2:1); (e) C₆H₆-Me₂CO (4:1); (f) *n*-hexane-EtOAc (2:1); (g) 60% MeOH; (h) upper layer of *n*-BuOH-pyridine-H₂O (6:2:3) + pyridine (1); (i) upper layer of *n*-BuOH-HOAc-H₂O (4:1:5).

Isolation of quillajasaponin. Commercial quillaja bark (2.5 kg) (a voucher specimen of the crude drug is on file at the Faculty of Pharmaceutical Sciences, Kyushu University) was extracted with MeOH and the MeOH was evaporated *in vacuo* to leave the MeOH extract (420 g). The latter was partitioned between EtOAc



and H₂O, and the H₂O layer was passed through an Amberlite XAD-2 column and eluted with H₂O and MeOH. Crude glycosides (85 g) obtained by evaporation of the MeOH eluate were treated with Sephadex LH-20 CC (eluate MeOH) to give two fractions, fraction 1 (39 g) (quillajasaponin) and 2 (42 g). Fraction 1 showing 7 spots (giving a dark-green colour after spraying with 5% H₂SO₄-MeOH followed by heating) on TLC [silica gel (a)] was refluxed with 3.5% HCl in 50% EtOH for 3 hr and the reaction mixture revealed a spot of quillaic acid (*R_f* 0.40) on TLC [silica gel (c)] (identified with an authentic sample). Fraction 2 showed two major spots (yellow colour by 5% H₂SO₄-MeOH and heating) on TLC [silica gel (a)] and further investigation was not carried out.

Alkaline hydrolysis of quillajasaponin. Quillajasaponin (18 g) was boiled with 6% NaHCO₃ in 50% MeOH (300 ml) for 1 hr, and the reaction mixture was neutralized with Dowex 50W-X8 and filtered. The filtrate showing spots of less polar and polar compounds on TLC [silica gel (a)] was evaporated *in vacuo* and the residue was chromatographed on Sephadex LH-20 (MeOH) to give two fractions, fractions 1 (5.0 g) (polar compounds, desacylsaponins) and 2 (2.2 g) (less polar compounds, eliminated acyl groups). Fraction 1 showing two major spots (*R_f* 0.11, 0.07) on TLC [silica gel (a)] was chromatographed on silica gel (eluant CHCl₃-MeOH-HOAc-H₂O, 15:9:1:2) to give two fractions, fractions 1' (*R_f* 0.11) and 2' (*R_f* 0.07). Each fraction showed a major and a few minor spots on TLC [C-8(g)] and each was chromatographed on C-8 (eluant 50% MeOH) to give a major component, DS-1 (1, 1.4 g) and DS-2 (2, 2.0 g) [*R_f* 0.11 (1) and 0.07 (2), silica gel (a), *R_f* 0.29 (1) and 0.35 (2), C-8 (g)].

DS-1 (1). White powder (from *n*-BuOH-H₂O-MeOH), mp 255–258° (decomp.), [α]_D²⁰ –19.6° (H₂O; *c* 0.67). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1730 (carbonyl), 1610 (COO⁻). FABMS (using KI as additive) *m/z*: 1589 [M(C₃₀H₁₀₈O₃₆) + 2K - H]⁺ = [(C₃₀H₁₀₈O₃₆ - COOK) + K]⁺. ¹³C NMR: δ 94.8, 101.0, 103.1, 103.9, 104.5, 106.9, 111.0 (each *d*, anomeric C \times 7), 176.0 (*s*, C-28), 209.8 (*d*, C-23). Compound 1 (40 mg) was refluxed with 2 N H₂SO₄ in 50% EtOH for 8 hr and then diluted with H₂O, and the ppt. was collected by filtration. The ppt. (crude aglycone) (10 mg) was chromatographed on silica gel (CHCl₃-MeOH, 30:1 \rightarrow 15:1) to provide 3 as colourless needles (from MeOH-H₂O), mp 256–260°, [α]_D²⁰ +66.7° (MeOH; *c* 1.02). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3450 (OH), 1730, 1715 (carbonyl). EIMS *m/z*: 486 [M(C₃₀H₄₆O₃)]⁺, 264, 246. Compound 3 was identified with an authentic sample of quillaic acid, by mmp, TLC, IR and ¹³C NMR. The H₂O layer was neutralized with Ba(OH)₂ soln, filtered and the filtrate was concentrated. The residue was examined by TLC [Avicel (h) and (i)], and Glc, Gal, Xyl, Fuc, Rham and Api were detected.

DS-2 (2). White powder (from *n*-BuOH-H₂O-MeOH), mp 258–261° (decomp.), [α]_D²⁰ –24.7° (H₂O; *c* 1.00). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1730 (carbonyl), 1610 (COO⁻). FABMS (using KI as additive) *m/z*: 1751 [M(C₃₅H₁₁₈O₄₁) + 2K - H]⁺ = [(C₃₅H₁₁₈O₄₁ - COOK) + K]⁺. ¹³C NMR: δ 95.0, 101.9, 103.1, 103.6, 104.3, 104.5, 104.9, 110.8 (each *d*, anomeric C \times 8), 176.3 (*s*, C-28), 210.2 (*d*, C-23). On hydrolysis with acid under the same conditions as those for 1, compound 2 gave 3 and a sugar mixture. The sugar mixture was found to consist of Glc, Gal, Xyl, Fuc, Rham, Api and Glc [TLC, Avicel (h) and (i)].

Alkaline hydrolysis of 1 and 2 providing 4. Compound 1 (200 mg) was refluxed with 2% KOH in 50% EtOH (10 ml) for 7 hr and the reaction mixture was neutralized with dilute HCl and concentrated *in vacuo*. The residue showing a spot (*R_f* 0.40) on TLC [silica gel (a)] was passed through a Sephadex G-15 column (H₂O) and a silica gel column (CHCl₃-MeOH-HOAc-H₂O, 15:9:1:2) to give a white powder (4) (from *n*-BuOH-H₂O-MeOH) (80 mg), mp > 300° (decomp.), [α]_D²⁰ +18.6° (H₂O; *c* 0.73). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1710 (carbonyl),

1610 (COO⁻). FABMS *m/z*: 995 [M(C₄₇H₇₂O₅₀) + K]⁺. ¹³C NMR (C₅D₅N + D₂O): δ 24.6 (*t*, C-2), 54.9 (*s*, C-4), 84.2 (*d*, C-3), 102.7, 102.9, 103.6 (each *d*, anomeric C \times 3), 180.1 (*s*, C-28), 210.1 (*d*, C-23). Compound 4 was hydrolysed with acid in the same manner as for 1 to provide 3 and sugars. The sugars were identified as Glc, Gal and Xyl [TLC, Avicel (h) and (i)]. Compound 2 (500 mg) was hydrolysed with alkali and worked up in the same manner as that for 1 to provide a white powder (200 mg) which was identical with 4 (mmp, TLC, IR and ¹³C NMR).

Preparation of 6 from 4. NaBH₄ (1.6 g) was added to a soln of 4 (250 mg) in 50% MeOH (25 ml) under ice-cooling and the reaction mixture was stirred for 2 hr at room temp. Me₂CO (2 ml) was added to the mixture and concentrated *in vacuo*, and the residue was passed through a Diaion HP20AG column (eluant H₂O \rightarrow MeOH). The MeOH eluate showing a spot (*R_f* 0.36) on TLC [silica gel (a)] was evaporated *in vacuo* to give a white powder (230 mg), mp 275–277° (decomp.), [α]_D²⁰ +3.8° (H₂O; *c* 0.9). ¹³C NMR: δ 64.2 (*t*, C-23), 175.0 (*s*, COOH of Glc), 182.4 (*s*, C-28), no CHO. The product (200 mg) was treated with NaH (500 mg) and MeI (5 ml) in DMSO (20 ml) (Hakomori method) and the reaction mixture was diluted with H₂O, extracted with Et₂O and the Et₂O layer was washed, dried and evaporated. The residue was chromatographed on silica gel (C₆H₆-Me₂CO, 7:1) to give a major product as a white powder (5) (150 mg). IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1760 (Glc-COOMe) [11], 1720 (28-COOMe) [18], no OH. ¹H NMR: δ 4.35 (1H, *d*, *J* = 7 Hz, anomeric H), 4.87 (1H, *d*, *J* = 7 Hz, anomeric H), 4.96 (1H, *d*, *J* = 8 Hz, anomeric H). FDMS *m/z*: 1127 [M(C₃₉H₉₈O₃₀) + H]⁺. Compound 5 (100 mg) in MeOH (10 ml) was reduced with NaBH₄ (1.2 g) as for 4. Me₂CO (2 ml) and H₂O (2 ml) were added to the mixture and evaporated *in vacuo*, and the residue was diluted with excess H₂O and extracted with CHCl₃. The CHCl₃ layer was washed, dried and concentrated, and the residue showing a spot (*R_f* 0.53) (*R_f* of 5, 0.77) on TLC [silica gel (d)] was methylated by the Hakomori method. The crude methylated product was chromatographed on silica gel (C₆H₆-Me₂CO, 7:1) to give 6 as a white powder (23 mg). IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1720 (28-COOMe), no OH. ¹H NMR: δ 3.30 (6H, *s*, OMe \times 2), 3.35, 3.37, 3.46 (each *s*, 3H, OMe \times 3), 3.50 (9H, *s*, OMe \times 3), 3.53, 3.55 (each *s*, 3H, OMe \times 2), 3.59 (6H, *s*, OMe \times 2), 4.24 (1H, *d*, *J* = 7 Hz, anomeric H), 4.92 (1H, *d*, *J* = 7 Hz, anomeric H), 5.02 (1H, *d*, *J* = 8 Hz, anomeric H).

Methanolysis of 6. Compound 6 (35 mg) was boiled with 10% HCl in MeOH (4 ml) for 2 hr, the mixture was neutralized with Ag₂CO₃ and filtered. The filtrate was evaporated and the residue (methanolysate) was examined by TLC [silica gel (f) and (d)] and GLC [conditions (a) and (b)]; one aglycone (7) and three methylated sugars were detected. The sugars were identified as S-1, S-2 and S-3 by direct comparison with authentic samples. The methanolysate was chromatographed on silica gel (C₆H₆-Me₂CO, 80:1 \rightarrow Me₂CO) to give 7 (10 mg). Compound 7 was acetylated with Ac₂O-pyridine as usual to give an acetate (8) (5 mg) as colourless needles (from MeOH-Me₂CO), mp 207–210°. IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1730 (ester), no OH. EIMS *m/z*: 572 [M(C₃₅H₅₆O₆)]⁺, 292. ¹H NMR: δ 2.04 (3H, *s*, OAc), 2.93 and 3.10 (1H each, *d*, *J* = 10 Hz, H₂-23), 3.25, 3.29, 3.59 (each *s*, 3H, OMe \times 3), 3.91 (1H, *br s*, H-16), 4.92 (1H, *t*-like, *J* = 8 Hz, H-3).

Preparation and methanolysis of 9. Compound 6 was heated with 2% HCl-MeOH for 8 hr at 60° and worked up as before. The mixture showing two major spots [*R_f* 0.28(6), 0.14] on TLC [silica gel; C₆H₆-Me₂CO, 5:1] was chromatographed on silica gel (*n*-hexane-Me₂CO, 5:1) to give a compound (*R_f* 0.14) which was methylated by the Hakomori method and worked up as before to yield 9 as a white powder. IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1720 (ester), no OH. ¹H NMR: δ 3.28 (6H, *s*, OMe \times 2), 3.32, 3.35, 3.39 (each *s*, 3H, OMe \times 3), 3.51 (9H, *s*, OMe \times 3), 3.60, 3.65 (each *s*, 3H, OMe \times 2),

4.24 (1H, *d*, *J* = 8 Hz, anomeric H), 4.62 (1H, *d*, *J* = 7 Hz, anomeric H). Compound 9 was subjected to methanolysis as for 6, and 7: S-2 and S-4 were obtained [TLC silica gel (f), GLC (a)].

Diazomethane degradation of 1 providing 11. A soln of CH₂N₂ in Et₂O (150 ml) was poured into a soln of 1 (1.1 g) in MeOH (300 ml) under cooling with ice. The mixture was left to stand for 1 hr at room temp., excess CH₂N₂ was decomposed with HOAc and the solvents were removed by distillation. The crude reaction mixture revealing two major spots (*R_f* 0.51 and 0.27) in TLC [silica gel (a)] was chromatographed on silica gel (CHCl₃-MeOH-H₂O, 8:3:0.3) to give a substance (white powder, *R_f* 0.27) (a mixture of the related compounds of the methyl ester of 1 and trisaccharides derived from the 3-*O*-sugar residue in 1 [13]) and 11 (white powder, 164 mg, *R_f* 0.51), mp 198–201° (decomp.), [α]_D –30.4° (MeOH; *c* 1.65). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3400 (OH), 1735 (ester). FABMS *m/z*: 1079 [M(C₃₃H₅₄O₂₁) + Na]⁺. ¹³C NMR: δ 94.8, 101.2, 106.9, 111.1 (each *d*, anomeric C \times 4), 176.1 (*s*, C-28). Compound 11 was boiled with 2 N H₂SO₄ for 2 hr and the reaction mixture was extracted with CHCl₃. The CHCl₃ layer showing a spot (*R_f* 0.21) on TLC [silica gel (e)] was washed, dried and evaporated to give 12, which was methylated with CH₂N₂ Et₂O in MeOH affording a compound corresponding with an authentic sample of 10 [TLC (e), IR, ¹H NMR and ¹³C NMR]. The aq. layer was treated as for 1 and Fuc, Rham, Xyl and Api were detected [TLC, Avicel (h) and (i)].

Synthesis and methanolysis of the permethylate (13) of 11. Compound 11 (40 mg) was methylated by the Hakomori method as before, and the reaction mixture was diluted with H₂O, extracted with CHCl₃ and the CHCl₃ layer was washed, dried and evaporated. The residue was passed through silica gel (*n*-hexane-Me₂CO, 2:1) and a Sephadex LH-20 (MeOH) column to give 13 as a white powder (16 mg). IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{–1}: 1750 (ester), no OH. FDMS *m/z*: 1210 [M(C₆₄H₁₀₆O₂₁)]⁺. ¹H NMR: δ 4.63 (1H, *d*, *J* = 8 Hz, anomeric H of Xyl), 5.24 (1H, *d*, *J* = 1 Hz, anomeric H of Rham), 5.37 (1H, *d*, *J* = 8 Hz, anomeric H of ester glycosidic fucose [19]), 5.46 (1H, *d*, *J* = 2 Hz, anomeric H of Api [20]) (anomeric H signals were assigned by comparison with those observed in the ¹H NMR spectrum of 17, and with the reported δ and *J* values [19, 20]). Compound 13 was methanolysed and worked up in the same manner as that for 6. The methanolysate was examined by TLC [silica gel (d)] and GLC [condition (a)], and S-5, S-6, S-7 and S-8 were detected. Silica gel CC of the methanolysate (*n*-hexane-Me₂CO, 6:1) afforded an aglycone (14). EIMS *m/z*: 528 [M(C₃₃H₅₂O₃)]⁺, 278.

Partial methanolysis of 11 yielding 15. Compound 11 (300 mg) in 2% HCl-MeOH (20 ml) was left to stand for 2.5 hr at 40°, and the reaction mixture was treated with Ag₂CO₃, filtered and evaporated. The residue giving three spots [*R_f* 0.84(12), 0.61 (methyl β -D-apiofuranoside) and 0.31] on TLC [silica gel (a)] was chromatographed on silica gel (CHCl₃-MeOH-HOAc-H₂O, 15:8:1:1) and Sephadex LH-20 (MeOH) to yield 15 (*R_f* 0.31) (31 mg) as a hygroscopic powder, [α]_D +11.8° (MeOH; *c* 1.00). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3400 (OH). FABMS *m/z*: 479 [M(C₁₈H₃₂O₁₃) + Na]⁺. ¹³C NMR: see Table 1. Compound 15 was boiled with 2 N H₂SO₄ for 2 hr and treated as usual. The hydrolysate was examined by TLC [Avicel (h) and (i)], and Xyl, Rham and Fuc were detected.

Diazomethane degradation of 2 affording 16. Compound 2 (2.0 g) in MeOH (150 ml) was treated with CH₂N₂ Et₂O (130 ml) and worked up as for 1. The crude reaction mixture showing three spots (*R_f* 0.42, 0.27 and 0.21) on TLC [silica gel (a)] was chromatographed on silica gel (CHCl₃-MeOH-H₂O, 6:4:0.3) to give substances of *R_f* 0.27 (3-*O*-trisaccharide residues in 2), *R_f* 0.21 (related compounds of the methyl ester of 2) and *R_f* 0.42 (16) (a white powder, 508 mg), mp 211–214° (decomp.), [α]_D –40.4° (MeOH; *c* 3.23). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3420 (OH), 1735 (ester)

FAB MS *m/z*: 1241 [M(C₅₉H₉₄O₂₆) + Na]⁺. ¹³C NMR: δ 95.1, 102.0, 104.8, 105.2, 111.2 (each *d*, anomeric C \times 5), 175.9 (*s*, C-28). On hydrolysis with acid under the same conditions as for 11, 16 gave 12 and a sugar mixture. The sugar mixture was found to consist of Fuc, Rham, Xyl, Api and Glc [TLC, Avicel (h) and (i)].

Preparation and methanolysis of 17. Compound 16 (300 mg) was methylated by the Hakomori method and worked up in the same manner as for 11 to provide a permethylate (17) (88 mg) as a white powder. IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{–1}: 1755 (ester), no OH. FDMS *m/z*: 1414 [M(C₇₃H₁₂₂O₂₆)]⁺. ¹H NMR: δ 4.68 (1H, *d*, *J* = 7 Hz, anomeric H of xylose), 4.82 (1H, *d*, *J* = 7 Hz, anomeric H of Glc), 5.05 (1H, *d*, *J* = 2 Hz, anomeric H of Rham), 5.39 (1H, *d*, *J* = 8 Hz, anomeric H of Fuc), 5.41 (1H, *d*, *J* = 2 Hz, anomeric H of Api) (anomeric H signals were assigned by comparison with the ¹H NMR spectrum of 13). Compound 17 was methanolysed and the methanolysate was examined as for 13. Compound 14, S-5, S-6, S-8, S-9 and S-10 were detected [TLC silica gel (d), GLC (a)].

Partial methanolysis of 16 yielding 18. Compound 16 (450 mg) was treated with 2% HCl-MeOH for 1.5 hr and worked up as for 11. The residue showing the spots of 12, methyl apiofuranoside and 18 (*R_f* 0.22) on TLC [silica gel (a)] was chromatographed on silica gel (EtOAc-MeOH-H₂O, 6:4:0.4) and Sephadex LH-20 (MeOH) to give 18 (145 mg) as a white powder, mp 178–182°, [α]_D –3.2° (MeOH; *c* 3.73). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3380 (OH). FABMS *m/z*: 619 [M(C₂₄H₄₂O₁₀) + H]⁺. ¹³C NMR: see Table 1. Compound 18 was hydrolysed with acid and worked up as for 15; Xyl, Glc, Rham and Fuc were obtained [TLC Avicel (h) and (i)].

Synthesis of 19 and its methanolysis. In the same manner as that for 11, compound 19 (63 mg) was prepared from 18 (120 mg) by the Hakomori method. IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{–1}: no OH. EIMS *m/z*: 758 [M(C₃₄H₆₂O₁₆)]⁺. Compound 19 was methanolysed as for 13 to give S-8, S-9, S-10 and S-11 [TLC silica gel (d), GLC (a)].

Preparation of 20 and its methanolysis. Compound 19 (50 mg) was methanolysed with 5% HCl-MeOH (3 ml) for 3 hr at 50° and worked up as before, and the residue (partial methanolysate) was chromatographed on silica gel (*n*-hexane-Me₂CO, 3:1) to give a major product (*R_f* 0.09) [TLC silica gel (*n*-hexane-Me₂CO, 3:1)] (*R_f* of 19, 0.18). The major product was methylated by the Hakomori method and treated as usual to yield 20 (8 mg) as a colourless syrup. IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{–1}: no OH. ¹H NMR: δ 3.33, 3.34, 3.47, 3.53, 3.57, 3.60, 3.62 (each *s*, 3H, OMe \times 7), 3.52 (6H, *s*, OMe \times 2), 4.45 (1H, *d*, *J* = 7 Hz, anomeric H), 4.77 (1H, *d*, *J* = 4 Hz, anomeric H), 4.95 (1H, *d*, *J* = 2 Hz, anomeric H). Methanolysis of 20 as before afforded S-8, S-10 and S-11 [TLC silica gel (d), GLC (a)].

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