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### HYPERVALENT IODINE IN SYNTHESIS. 77. AN EFFICIENT METHOD FOR THE SYNTHESIS OF N-ARYLINDOLES BY THE COPPER-CATALYZED N-ARYLATION OF INDOLE WITH DIARYLIODONIUM SALTS

Tao Zhou<sup>a</sup> & Zhen-Chu Chen<sup>a</sup>

<sup>a</sup> Department of Chemistry, Zhejiang University, Xixi Campus, Hangzhou, 310028, P.R. China

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**HYPERVALENT IODINE IN  
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METHOD FOR THE SYNTHESIS  
OF *N*-ARYLINDOLES BY  
THE COPPER-CATALYZED  
*N*-ARYLATION OF INDOLE  
WITH DIARYLIODONIUM SALTS**

**Tao Zhou and Zhen-Chu Chen\***

Department of Chemistry, Zhejiang University  
(Xixi Campus), Hangzhou, 310028, P.R. China

**ABSTRACT**

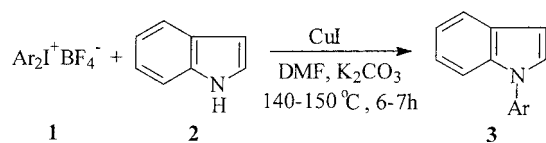
*N*-arylindoles were prepared in good yields by copper-catalyzed *N*-arylation of indole with diaryliodonium salts.

In continuation of our interest in exploring the utility of transition metal-catalyzed arylation reaction,<sup>1</sup> we examined *N*-arylation of indole with diaryliodonium salts in the presence of copper catalyst (Scheme 1). The results are summarized in Table 1.

We found that copper-catalyzed *N*-arylation of indole with diaryliodonium salts could proceed smoothly at 140–150°C and gave *N*-arylindoles in good yields. Compared with the traditional Ullmann reaction<sup>2</sup>

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\*Corresponding author.



Scheme 1.

Table 1. Copper-Catalyzed *N*-Arylation of Indole

Entry	$\text{Ar}_2\text{I}^+\text{BF}_4^-$	Copper Catalyst	Product	Yield <sup>a</sup> (%)
1	$\text{Ph}_2\text{I}^+\text{BF}_4^-$ <b>1a</b>	$\text{Cu}(\text{OAc})_2$	<b>3a</b>	52
2	<b>1a</b>	$\text{Cu}(\text{acac})_2$	<b>3a</b>	83
3	<b>1a</b>	$\text{CuI}$	<b>3a</b>	83
4	$(p\text{-Tol})_2\text{I}^+\text{BF}_4^-$ <b>1b</b>	$\text{CuI}$	<b>3b</b>	87
5	$(p\text{-ClC}_6\text{H}_4)_2\text{I}^+\text{BF}_4^-$ <b>1c</b>	$\text{CuI}$	<b>3c</b>	92
6	$(p\text{-CH}_3\text{OC}_6\text{H}_4)_2\text{I}^+\text{BF}_4^-$ <b>1d</b>	$\text{CuI}$	<b>3d</b>	80
7	$(p\text{-BrC}_6\text{H}_4)_2\text{I}^+\text{BF}_4^-$ <b>1e</b>	$\text{CuI}$	<b>3e</b>	81
8	$(m\text{-NO}_2\text{C}_6\text{H}_4)_2\text{I}^+\text{BF}_4^-$ <b>1f</b>	$\text{CuI}$	<b>3f</b>	76
9	$(p\text{-CH}_3\text{CONHC}_6\text{H}_4)_2\text{I}^+\text{BF}_4^-$ <b>1g</b>	$\text{CuI}$	<b>3g</b>	72

<sup>a</sup>Yield of isolated pure product.

(190–210°C), the reaction temperature was lower greatly. Furthermore, stoichiometric copper powder was not needed. Among the tested copper catalysts including  $\text{CuI}$ ,  $\text{Cu}(\text{OAc})_2$  and  $\text{Cu}(\text{acac})_2$ , though each of them can catalyze this reaction,  $\text{CuI}$  is the most effective and practical. Several diaryliodonium salts with various substituents such as methyl, chloro, methoxy, bromo, nitro and acetamido groups were tested. It was found that the reaction is general for these substrates.

In conclusion, we provide an effective synthesis for *N*-arylindoles by copper-catalyzed *N*-arylation of indole with diaryliodonium salts. It has some advantages over others<sup>3–10</sup> such as mild reaction conditions, simplicity of procedure, using nonpoisonous, accessible reagents and good yields.

## EXPERIMENTAL

Melting points were uncorrected.  $^1\text{H}$ NMR data were recorded on Avance 400 spectrometer using  $\text{CDCl}_3$  as the solvent with TMS as an

internal standard. IR spectra were determined on Vector 22 infrared spectrometer with KBr pallet. MS spectra were recorded on HP5859B mass spectrometer. Elemental analyses were performed on EA1110.

**General procedure for preparation of *N*-arylindoles.** A mixture of diaryliodonium salt **1** (1 mmol), indole **2** (1 mmol),  $K_2CO_3$  (3 mmol), CuI (10 mol%) and DMF (5 ml) was stirred under a nitrogen atmosphere at 140–150°C for 6–7 h. After cooling, the reaction mixture was diluted with saturated  $NH_4Cl$  aqueous (20 ml), and extracted with diethyl ether ( $3 \times 15$  ml). The combined organic layer was washed with brine, dried over anhydrous sodium sulfate. After removal of the solvent under vacuum, the residue was chromatographed on silica gel plate using petroleum ether for **3a–3e**, *c*-hexane/ethyl acetate (20:1) for **3f** and *c*-hexane/ethyl acetate (1:1) for **3g** as a developer to give pure product.

#### Physical and Spectroscopic Data

***N*-Phenylindole 3a.** Oil (lit<sup>2(c)</sup> oil).  $^1H$  NMR  $\delta_H$  6.67 (m, 1H), 7.20 (m, 2H), 7.33 (m, 2H), 7.49 (m, 4H), 7.55 (m, 1H), 7.68 (m, 1H). IR  $\nu_{max}/cm^{-1}$  1597, 1515, 1498, 1457, 1332, 1234, 1213, 1136, 774, 742, 696.

***N*-(4-Methylphenyl)indole 3b:** Oil (lit<sup>2(c)</sup> oil).  $^1H$  NMR  $\delta_H$  2.42 (s, 3H), 6.65 (m, 1H), 7.15 (m, 1H), 7.20 (m, 1H), 7.29 (m, 3H), 7.37 (m, 2H), 7.52 (m, 1H), 7.68 (m, 1H). IR  $\nu_{max}/cm^{-1}$  1519, 1458, 1334, 1234, 1213, 1135, 822, 762, 741.

***N*-(4-Chlorophenyl)indole 3c.** M.p. 66–67°C (lit<sup>2(c)</sup> m.p. 64–66°C).  $^1H$  NMR  $\delta_H$  6.69 (m, 1H), 7.20 (m, 2H), 7.29 (m, 1H), 7.43–7.52 (m, 5H), 7.68 (m, 1H). IR  $\nu_{max}/cm^{-1}$  1592, 1516, 1492, 1331, 1233, 1211, 1132, 1088, 1011, 831, 762, 737.

***N*-(4-Methoxyphenyl)indole 3d.** M.p. 56–58°C (lit<sup>2(c)</sup> m.p. 57–58°C).  $^1H$  NMR  $\delta_H$  3.88 (s, 3H), 6.65 (m, 1H), 7.03 (m, 2H), 7.15–7.20 (m, 2H), 7.27 (m, 1H), 7.39 (m, 2H), 7.44 (m, 1H), 7.68 (m, 1H). IR  $\nu_{max}/cm^{-1}$  1611, 1517, 1457, 1333, 1301, 1248, 1213, 1179, 1134, 1105, 835, 762, 747.

***N*-(4-Bromophenyl)indole 3e.** M.p. 64–65°C (lit<sup>9</sup> m.p. was not reported).  $^1H$  NMR  $\delta_H$  6.69 (m, 1H), 7.17 (m, 1H), 7.24 (m, 1H), 7.29 (m, 1H), 7.39 (m, 2H), 7.51 (m, 1H), 7.63 (m, 2H), 7.68 (m, 1H). IR  $\nu_{max}/cm^{-1}$  1588, 1516, 1491, 1455, 1334, 1310, 1233, 1211, 1133, 1101, 1072, 1009, 828, 762, 738, 722. MS  $m/z$  271 ( $M^+$ , 100), 273 ( $M^+ + 2$ , 98.14). Anal. Calcd for  $C_{14}H_{10}BrN$ : C, 61.79; H, 3.70; N, 5.14. Found: C, 61.56; H, 3.74; N, 5.28.

***N*-(3-Nitrophenyl)indole 3f.** M.p. 65–67°C (lit<sup>2(b)</sup> m.p. 67–68°C).  $^1H$  NMR  $\delta_H$  6.76 (m, 1H), 7.24 (m, 1H), 7.29 (m, 1H), 7.38 (m, 1H), 7.59 (m, 1H), 7.71 (m, 2H), 7.87 (m, 1H), 8.19 (m, 1H), 8.40 (m, 1H). IR  $\nu_{max}/cm^{-1}$  1607, 1534, 1487, 1455, 1349, 1236, 1211, 1137, 742.

***N*-(4-Acetamidophenyl)indole 3g.** M.p. 161–162°C.  $^1\text{H}$  NMR  $\delta_{\text{H}}$  2.23 (s, 3H), 6.67 (m, 1H), 7.16–7.21 (m, 2H), 7.29 (m, 1H), 7.38 (s, 1H, N-H), 7.45 (m, 2H), 7.51 (m, 1H), 7.64–7.69 (m, 3H). IR  $\nu_{\text{max}}/\text{cm}^{-1}$  3243, 3052, 1659, 1612, 1519, 1461, 1415, 1371, 1316, 1266, 1236, 1215, 1139, 1014, 969, 840, 764, 744. MS  $m/z$  250 ( $\text{M}^+$ , 100). Anal. Calcd for  $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}$ : C, 76.78; H, 5.64; N, 11.19. Found: C, 76.62; H, 5.68; N, 11.27.

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