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Reaction of 1-Bromovinyl and 1-Bromo-2-phenylvinyl Sulfones with Some CH Acids

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Abstract—Benzyl, methyl, and phenyl α -bromovinyl sulfones reacted with malononitrile and dimethyl malonate sodium enolates in THF to give sulfonyl-substituted cyclopropanes. Reactions of the same sulfones with methyl acetoacetate sodium enolate afforded the corresponding sulfonyl-substituted cyclopropanes as mixtures of *cis* and *trans* isomers with a small impurity of 5-sulfonyl-4,5-dihydrofuran derivative. Phenyl and *p*-tolyl 1-bromo-2-phenylvinyl sulfones reacted with methyl acetoacetate sodium salt to produce a mixture of *trans*-isomeric 5-sulfonyl-4,5-dihydrofuran and Michael adduct of the CH acid with acetivated acetylene generated by concurrent 1,2-dehydrobromination of the initial α -bromovinyl sulfone.

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α-Halo α,β-unsaturated esters, nitriles, and ketones are known to react with dialkyl malonates and malononitrile in the presence of a base to give substituted cyclopropanecarboxylates, cyclopropanecarbonitriles, and cyclopropyl ketones, respectively [1–6]. Reactions of the same activated alkenes with alkyl acetoacetate or 1,3-diketones lead to the formation of mixtures of the corresponding cyclopropanes and dihydrofuran derivative; in some cases, the latter is formed as the major product [7–9]. These transformations can be regarded as MIRC reactions (Michael-Induced Ring Closure) where the Michael addition is followed by 1,3- and/or 1,5-cyclization.

A few published data are available on analogous transformations involving α -halovinyl sulfones. As shown in [10, 11], 1-halo-2-phenylvinyl sulfones **Ia** and **Ib** react with dialkyl malonates in the presence of sodium hydride according to the MIRC reaction pat-



I, R = Ph, Hlg = Cl(a); R = Me, Hlg = Br(b); II, IV, R = Ph, R' = Et(a); R = R' = Me(b).

tern with formation of cyclopropyl sulfones **IIa** and **IIb** (Scheme 1). The reaction of 1-bromo-2-phenylvinyl sulfone (**Ib**) with methyl acetoacetate gave a mixture of cyclopropane and dihydrofuran derivatives **IIIb** and **IVb** at a ratio of 3:1, whereas 1-chloro-2-phenylvinyl sulfone (**Ia**) reacted with ethyl acetoacetate to produce substituted dihydrofuran **IVa** as the only product. Unexpectedly, in the reaction of α -bromo sulfone **Ib** with malononitrile we isolated allyl sulfone **Vb**, presumably as a result of synchronous process including 1,2-dehydrobromination, Michael addition, and isomerization [12].

In the present work we studied reactions of 1-bromovinyl sulfones VIa-VIc with malononitrile, dimethyl malonate, and methyl acetoacetate with a view to comparison of their behavior with the behavior of 1-halo-2-phenylvinyl sulfones I with different substitution pattern of the C=C double bond. Reactions of sulfones VIa and VIc with CH acids have already been reported. For example, the MIRC reaction of sulfone VIa with diethyl malonate and ethyl acetoacetate in the presence of sodium hydride in THF gave the corresponding substituted cyclopropyl sulfone [13]. Interesting results were obtained in the reaction of sulfone VIc with dimethyl 2-methylmalonate [14]. In this case, MIRC reaction is impossible, and MIRB reaction (Michael-Induced Ramberg-Bäcklund rearrangement) leads to diethyl 2-cinnamoyl-2-methylmalonate.

The reactions of bromovinyl sulfones **VIa–VIc** with CH acids were carried out in THF at 20–50°C using sodium hydride as base to generate the corresponding CH acid enolate. In the reactions with dimethyl malonate and malononitrile we isolated in all cases the only product, the corresponding cyclopropyl sulfone **VIIa–VIIc** or **VIIIa–VIIIc** (Scheme 2). Much more complicated pattern was observed in the reactions with methyl acetoacetate. Bromovinyl sulfone

VIa gave rise to a mixture of stereoisomeric cyclopropanes IXa and IXa' at a ratio of 1:1 with a small impurity of dihydrofuran Xa. This result differs from the data of [13], according to which the reaction of sulfone VIa with ethyl acetoacetate afforded an analog of IXa as a single stereoisomer whose configuration was not determined. From vinyl sulfone VIb we obtained cyclopropanes IXb and IXb' at a ratio of 1:2 and a small amount (3.5%) of dihydrofuran Xb. In the reaction of vinyl sulfone VIc with methyl acetoacetate stereoisomeric cyclopropanes IXc and IXc' were formed at a ratio of 1:3. In addition, the reaction mixture contained small amounts of dihydrofuran derivative Xc and cyclopropane XIc which were detected by ¹H NMR. Presumably, the latter was formed as a result of partial decomposition of acetyl-substituted cyclopropane IVc' under alkaline conditions during the isolation procedure (treatment with water). No products that could arise from *a priori* possible MIRB reactions were identified in the reactions of vinyl sulfones VIb and VIc with all the CH acids used.

Except for Xb, compounds VII-XI were isolated in the pure state by silica gel column chromatography and/or crystallization. Dihydrofuran Xb was detected in the product mixture by ¹H NMR. Chromatographic purification of cyclopropane IXa was accompanied by its partial transformation into dihydrofuran Xa (cf. [15]). The structure of the isolated compounds was determined on the basis of their IR, ¹H and ¹³C NMR, and mass spectra. Cyclopropanes VIIIa-VIIIc characteristically showed in the ¹³C NMR spectra anomalously upfield position of the C¹ signal ($\delta_{\rm C} \sim 6$ ppm) despite two cyano groups attached to that carbon atom. Signals from carbon nuclei were assigned using ¹³C⁻¹H HMBC technique. In the ¹H NMR spectra of VII-IX multiplet signals from protons on the cyclopropane ring were reliably recognized as an AMX spin system. These protons were assigned, and the con-



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Fig. 1. A fragment of the ¹H–¹H NOESY spectrum of methyl 1-acetyl-2*c*-(phenylmethanesulfonyl)cyclopropane-1*r*-carboxylate (IXc).

figuration of stereoisomers IX and IX', was determined with account taken of the known relation for spin–spin coupling constants in cyclopropanes: ${}^{3}J_{cis} >$ ${}^{3}J_{trans} > {}^{2}J_{gauche}$ [16, 17]. The assumed configurations of IXa-IXc and IXa'-IXc' were consistent with the observed differences in the chemical shifts of protons in the methoxycarbonyl and acetyl group due to deshielding effect of the oppositely faced sulfonyl group. The ¹H–¹H NOESY data (Figs. 1, 2) provided additional proofs for the assigned configurations of IXc and IXc'. In the NOESY spectrum of IXc, cross-peaks a correspond to the interaction between H_A and H_X , crosspeaks b reflect coupling of protons in the acetyl group with H_A , correlations H_A/H_M and H_X/H_M give rise to cross-peaks c and d, respectively, and cross-peaks e belong to coupling of H_X with methyl protons in the

acetyl group. In the ¹H–¹H NOESY spectrum of **XIc'** (Fig. 2) we observed the following correlations: H_{M-} MeCO (*a*), H_A-H_M (*b*), H_X-H_A (*c*), and H_X-H_M (*d*). Cyclopropane **XIc** was assigned *trans* configuration [16] on the basis of the spin–spin coupling constant between H_X and H_Y (³J = 7.4 Hz), which was determined by decoupling from H_A and H_M .

Comparison of our results with the data of [13] on the reactions of α -bromovinyl sulfones **VIa–VIc** with CH acids and with the results of the reactions of sulfones **Ia** and **Ib** with the same CH acids [10–12] shows almost complete similarity with respect to the reaction pattern, except for the reaction of **Ib** with malononitrile. One more notable feature is the behavior of chlorostyryl sulfone **Ia** in the reaction with alkyl acetoacetate, which, unlike the reactions with bromovinyl



Fig. 1. A fragment of the ¹H–¹H NOESY spectrum of methyl 1-acetyl-2*t*-(phenylmethanesulfonyl)cyclopropane-1*r*-carboxylate (IXc').

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sulfones **Ib** and **VIa–VIc**, led to the formation of dihydrofuran derivative **IVa** as the only product.

We also examined reactions of analogs of Ia and Ib, aryl 1-bromo-2-phenyl sulfones Ic and Id, with methyl acetoacetate. The products of these reactions were mixtures of dihydrofurans IVc and IVd and vinyl sulfones XIIc and XIId at ratios of 2:1 and 1:1.2, respectively (Scheme 3). Pure compounds IVc, IVd, XIIc, and XIId were isolated by silica gel column chromatography. The configuration of IVc and IVd was assumed to be the same as that of their analogs IVa and IVb [10] on the basis of the expected similarity of their ¹H NMR spectra. Sulfones XIIc and XIId were identified by comparison with authentic samples [18].



Presumably, the presence of a phenyl group in sulfone **Ib** facilitates its 1,2-dehydrobromination which becomes preferred to Michael addition; therefore, the reaction of **Ib** with malononitrile yields allylic derivative **Vb** in contrast to sulfones **VIa–VIc** whose reaction with malononitrile leads to the formation of cyclopropyl sulfones **VIIIa–VIIIc**.

The different behaviors of halostyryl sulfone Ia, on the one hand, and Ic and Id, on the other, can also be rationalized taking into account that dehydrobromination is more favorable than dehydrochlorination. Then, bromo sulfones Ic and Id are involved in both MIRC reaction and tandem dehydrobromination–Michael addition with formation of compounds Vc, Vd and XIIc, XIId, respectively, whereas chloro sulfone Ia reacts exclusively according to the MIRC pattern to produce dihydrofuran derivative IVa.

Finally, somewhat different stereoselectivities in the MIRC reactions of vinyl sulfones **VIa–VIc** may be interpreted as follows. The reaction with sulfone **VIa**

is not selective. The reaction with sulfone VIc containing a bulkier (than $PhSO_2$) $PhCH_2SO_2$ group clearly shows *trans* selectivity, i.e., the major product is cyclopropane **IXc'** in which the $PhCH_2SO_2$ is faced to smaller acetyl group rather than ester group (cf. conformational energies of the COMe and CO₂Me substituents, 1.02 and 1.2 kcal/mol, respectively [19]).

In summary, we can state that Michael-induced tandem reactions of 1-bromovinyl sulfones and 1-bromo-2-phenylvinyl sulfones with CH acids should be regarded as a convenient method for the preparation of functionally substituted cyclopropane derivatives which are difficult to obtain by other methods. On the other hand, there exist some limitations related to specific structural features of the initial α -bromo sulfones.

EXPERIMENTAL

The ¹H and ¹³C NMR spectra were recorded from solutions in CDCl₃ on a Jeol JNM-ECX400 spectrometer at 400 and 100 MHz, respectively; the chemical shifts were measured relative to the residual proton and carbon signals of the solvent (CHCl₃, δ 7.26 ppm; CDCl₃, $\delta_{\rm C}$ 77.0 ppm). The IR spectra were recorded from thin films or KBr pellets on an InfraLYuM FT-02 spectrometer with Fourier transform. The elemental compositions were determined on a VarioMICRO CHNS-analyzer. The mass spectra (electron impact, 70 eV) were obtained using a KONIK RBK-HRGC5000B-MSQ12 instrument (Spain). Silufol UV-254 plates were used for analytical TLC; eluent light petroleum ether-ethyl acetate (2:1), development with iodine vapor. Silica gel L (40-100 µm) was used for column chromatography; eluent light petroleum etherethyl acetate (5:1).

α-Bromovinyl sulfones **VIa**, **VIc** [20], and **VIb** [21] were synthesized by known procedures.

1-Bromo-2-phenylvinyl sulfones Ic and Id (general procedure). (E)-1-Benzenesulfonyl-2-phenylethene [22] or (E)-1-(4-methylbenzenesulfonyl)-2phenylethene [23], 12 mmol, was dissolved in 12 mL of anhydrous methylene chloride, a solution of 13.2 mmol (2.11 g) of bromine in 4 mL of the same solvent was gradually added, and the mixture was stirred for 40 h at 20°C. The solvent was distilled off under reduced pressure (water-jet pump). The oily residue was treated with 30 mL of aqueous dioxane (1:1) and 1.53 g (14 mmol) of sodium carbonate, the mixture was stirred for 25 h at 40–45°C and diluted with 150 mL of water, and the crystals were filtered off, dried in air, and recrystallized from diethyl ether-hexane (1:4).

(*Z*)-1-(Benzenesulfonyl)-1-bromo-2-phenylethene (Ic). Yield 83%, R_f 0.56, mp 101–102°C; published data [24]: mp 101–103°C. ¹H NMR spectrum, δ , ppm: 7.40–7.47 m (3H, H_{arom}), 7.54–7.60 m (2H, H_{arom}), 7.63–7.68 m (1H, H_{arom}), 7.77–7.87 m (2H, H_{arom}), 7.98–8.00 m (2H, H_{arom}), 8.36 s (1H, 2-H). ¹³C NMR spectrum, δ_C , ppm: 120.6 weak, 128.7 (2C), 129.05 (2C), 129.11 (2C), 130.1 (2C), 131.1, 131.8 weak, 134.0, 137.3 weak, 138.9. Found, %: C 52.14; H 3.51; S 9.89. C₁₄H₁₁BrO₂S. Calculated, %: C 52.03; H 3.43; S 9.92.

(*Z*)-1-Bromo-1-(4-methylbenzenesulfonyl)-2-phenylethene (Id). Yield 85%, R_f 0.68, mp 105– 106°C; published data [25]: mp 109–111°C. IR spectrum, v, cm⁻¹: 1593 m, 1323 s, 1312 m, 1150 v.s, 756 m, 687 m, 667 s, 606 m, 544 v.s. ¹H NMR spectrum, δ , ppm: 2.46 s (3H, CH₃), 7.36 d (2H, H_{arom}, *J* = 8.2 Hz), 7.38–7.47 m (3H, H_{arom}), 7.79–7.84 m (2H, H_{arom}), 7.88 d (2H, H_{arom}, *J* = 8.2 Hz), 8.34 s (1H, 2-H). ¹³C NMR spectrum, δ_C , ppm: 21.7 (CH₃), 121.0 weak, 128.5 (2C), 129.1 (2C), 129.8 (2C), 130.0 (2C), 131.0, 131.9 weak, 134.2 weak, 138.5, 145.1. Found, %: C 53.45; H 3.68; S 9.39. C₁₃H₁₄O₆S. Calculated, %: C 53.42; H 3.89; S 9.51.

Reaction of a-bromovinyl sulfones VIa-VIc with dimethyl malonate and malononitrile (general procedure). Sodium hydride, 6 mmol (prepared by washing a 60% suspension of NaH in mineral oil with hexane), was added under argon to 5 mL of anhydrous THF, a solution of 4.5 mmol of dimethyl malonate or malononitrile in 10 mL of THF was added dropwise over a period of 15 min under stirring at 20°C, and the mixture was stirred for 1 h at 20°C. A solution of 4.0 mmol of α -bromovinyl sulfone VIa–VIc in 10 mL of anhydrous THF was added dropwise over a period of 15 min under stirring at 20°C, and the mixture was stirred for 15 h at 20°C, diluted with water (200 mL), neutralized with dilute (1:1) aqueous HCl, and extracted with methylene chloride $(3 \times 30 \text{ mL})$. The combined extracts were washed with water and dried over MgSO₄, the solvent was removed under reduced pressure, and the residue was subjected to column chromatography on silica gel.

Dimethyl 2-(benzenelsulfonyl)cyclopropane-1,1dicarboxylate (VIIa). Yield 55%, mp 118–119°C (from Et₂O). IR spectrum, v, cm⁻¹: 3029 w, 1748 v.s, 1736 v.s, 1443 m, 1346 m, 1308 s, 1262 s, 1208 m, 1157 s, 725 m, 613 m. ¹H NMR spectrum, δ , ppm: 1.73 d.d (1H, H_A, J = 5.6, 9.2 Hz), 2.24 d.d (1H, H_M, J = 5.6, 7.1 Hz), 3.35 d.d (1H, H_X, J = 7.1, 9.2 Hz), 3.72 s and 3.85 s (3H each, OCH₃), 7.57 t (2H, H_{arom}, J = 7.9 Hz), 7.67 t (1H, H_{arom}, J = 7.5 Hz), 7.92 d (2H, H_{arom}, J = 7.3 Hz). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 17.7 (C³), 36.5 weak (C¹), 44.6 (C²), 53.4 and 53.5 (OCH₃), 127.9 (2C, C_{arom}), 129.3 (2C, C_{arom}), 134.1 (C_{arom}), 139.6 weak (C_{arom}), 164.7 and 167.9 (C=O). Found, %: C 52.44; H 4.61; S 10.79. C₁₃H₁₄O₆S. Calculated, %: C 52.34; H 4.73; S 10.75.

Dimethyl 2-(methanesulfonyl)cyclopropane-1,1dicarboxylate (VIIb). Yield 59%, mp 81–82°C (from acetone–hexane). IR spectrum, v, cm⁻¹: 1747 s, 1740 v.s, 1724 s, 1443 m, 1319 s, 1311 v.s, 1257 v.s, 1215 m, 1142 s, 791 m. ¹H NMR spectrum, δ , ppm: 1.81 d.d (1H, H₄, J = 5.8, 9.2 Hz), 2.26 d.d (1H, H_M, J = 5.8, 6.8 Hz), 3.12 s (3H, CH₃SO₂), 3.30 d.d (1H, H_X, J = 6.8, 9.2 Hz), 3.79 s and 3.81 s (3H each, OCH₃). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 15.8 (C³), 35.7 weak (C¹), 41.6 (C²), 42.4 (SO₂CH₃), 53.4 and 53.7 (OCH₃), 164.7 and 167.8 (C=O). Mass spectrum, m/z($I_{\rm rel}$, %): 236 (0.7) [M]⁺⁺, 205 (90), 157 (92), 156 (84), 125 (84), 98 (90), 83 (90), 69 (93), 67 (92), 59 (99), 55 (100). Found, %: C 40.64; H 5.21; S 13.64. C₈H₁₂O₆S. Calculated, %: C 40.67; H 5.12; S 13.57.

Dimethyl 2-(phenylmethanesulfonyl)cyclopropane-1,1-dicarboxylate (VIIc). Yield 68%, mp 110– 111°C (from EtOAc–hexane). IR spectrum, v, cm⁻¹: 1755 s, 1732 v.s, 1443 m, 1338 s, 1311 s, 1273 s, 1211 m, 1200 m, 1142 s, 1126 s, 1011 m, 694 m, 505 m. ¹H NMR spectrum, δ , ppm: 1.72 d.d (1H, H_A, J = 5.7, 9.3 Hz), 2.20 d.d (1H, H_M, J = 5.7, 6.9 Hz), 2.94 d.d (1H, H_X, J = 6.9, 9.3 Hz), 3.77 s and 3.85 s (3H each, OCH₃), 4.36 d and 4.53 d (1H each, SO₂CH₂, J = 13.6 Hz), 7.37–7.41 m (3H, H_{arom}), 7.42– 7.47 m (2H, H_{arom}). ¹³C NMR spectrum, δ_{C} , ppm: 15.8 (C³), 35.5 weak (C¹), 39.8 (C²), 53.5 and 53.6 (OCH₃), 60.1 (SO₂CH₂), 127.3 weak (C_{arom}), 129.0 (2C, C_{arom}), 129.1 (C_{arom}), 130.9 (2C, C_{arom}), 165.0 and 167.6 (C=O). Found, %: C 53.75; H 5.12; S 10.29. C₁₄H₁₆O₆S. Calculated, %: C 53.84; H 5.16; S 10.26.

2-(Benzenesulfonyl)cyclopropane-1,1-dicarbonitrile (VIIIa). Yield 43%, mp 126–127°C (from acetone–hexane). IR spectrum, v, cm⁻¹: 3040 m, 2253 w, 1346 m, 1319 s, 1165 v.s, 1157 s, 1084 m, 744 m, 687 m, 548 m. ¹H NMR spectrum, δ , ppm: 2.33 d.d (1H, H_A, J = 6.7, 9.3 Hz), 2.66 d.d (1H, H_M, J = 6.7, 7.9 Hz), 3.46 d.d (1H, H_X, J = 7.9, 9.3 Hz), 7.69 t (2H, H_{arom}, J = 7.4 Hz), 7.82 t (1H, H_{arom}, J = 7.5 Hz), 8.05 d (2H, H_{arom}, J = 7.3 Hz). ¹³C NMR spectrum, δ_{C} , ppm: 6.2 weak (C¹), 20.5 (C³), 45.2 (C²), 110.2 and 112.5 (CN) 128.4 (2C, C_{arom}), 130.1 (2C, C_{arom}), 135.6 (C_{arom}), 137.2 weak (C_{arom}). Found, %: C 56.76; H 3.40; N 11.96; S 13.86. $C_{11}H_8N_2O_6S$. Calculated, %: C 56.89; H 3.47; N 12.06; S 13.80.

2-(Methanesulfonyl)cyclopropane-1,1-dicarbonitrile (VIIIb). Yield 61%, mp 129–130°C (from EtOH). IR spectrum, v, cm⁻¹: 2256 w, 1335 m, 1311 s, 1165 m, 1146 v.s, 1099 m, 1088 m, 783 m, 536 m, 509 m. ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 2.43 d.d (1H, H_M, *J* = 6.8, 8.2 Hz), 2.71 d.d (1H, H_A, *J* = 6.8, 9.3 Hz), 3.34 s (3H, CH₃), 4.45 d.d (1H, H_X, *J* = 8.2, 9.3 Hz). ¹³C NMR spectrum (DMSO-*d*₆), δ_{C} , ppm: 5.6 weak (C¹), 20.3 (C³), 42.0 (SO₂CH₃), 43.4 (C²), 112.2 and 114.2 (C≡N). Found, %: C 42.42; H 3.50; N 16.54; S 18.72. C₆H₆N₂O₂S. Calculated, %: C 42.35; H 3.55; N 16.46; S 18.84.

2-(Phenylmethanesulfonyl)cyclopropane-1,1-dicarbonitrile (VIIIc). Yield 56%, mp 106–107°C (from EtOH). IR spectrum, v, cm⁻¹: 3024 m, 2256 w, 1331 s, 1315 v.s, 1153 m, 1126 v.s, 698 m, 540 m, 517 s. ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 2.25 d.d (1H, H_M, J = 6.8, 8.3 Hz), 2.66 d.d (1H, H_A, J = 6.8, 9.5 Hz), 4.40 d.d (1H, H_X, J = 8.3, 9.5 Hz), 4.85 d and 4.90 d (1H each, SO₂CH₂, J = 14.0 Hz), 7.41–7.48 m (3H, H_{arom}), 7.48–7.52 m (2H, H_{arom}). ¹³C NMR spectrum (DMSO-*d*₆), δ_{C} , ppm: 6.2 weak (C¹), 19.6 (C³), 41.6 (C²), 59.9 (SO₂CH₂), 111.5 and 113.9 (C≡N), 127.7 weak (C_{arom}), 129.4 (2C, C_{arom}), 129.5 (C_{arom}), 131.8 (2C, C_{arom}). Found, %: C 58.49; H 4.15; N 11.39; S 12.99. C₁₂H₁₀N₂O₂S. Calculated, %: C 58.52; H 4.09; N 11.37; S 13.02.

Reaction of sulfones Ic, Id, and VIa-VIc with methyl acetoacetate (general procedure). The reactions of α -bromo sulfones Ic, Id, and VIa–VIc with methyl acetoacetate were carried out at the same reactant ratio as in the reactions with malonic acid derivatives. The mixture was stirred for 22–24 h at 50°C and treated as described above. The crude product was analyzed by ¹H NMR. From bromovinyl sulfone Ic we obtained dihydrofuran IVc and vinyl sulfone XIIc at a ratio of 2:1: from sulfone Id a mixture of dihydrofuran IVd and vinyl sulfone XIId at a ratio of 1:1.2 was formed. The reaction with VIa gave a 1:1 mixture of IXa and IXa' and <8% of dihydrofuran Xa. Sulfone VIb gave rise to compounds IXb and IXb' at a ratio of 1:2 and 3.5% of dihydrofuran Xb; and the reaction with VIc produced sulfones IXc and IXc' at a ratio of 1:3 and $\sim 5\%$ of **Xc** and **XIc** each. The pure products were isolated by column chromatography on silica gel.

Methyl trans-5-(benzenesulfonyl)-2-methyl-4phenyl-4,5-dihydrofuran-3-carboxylate (IVc). Yield

(3H, OCH₃), 4.91 d.q (1H, 4-H, *J* = 1.4, 3.2 Hz), 5.11 d (1H, 5-H, J = 3.2 Hz), 7.19 d (2H, H_{arom}, J = 7.0 Hz), 7.25 t (1H, H_{arom}, J = 7.3 Hz), 7.32 t (2H, H_{arom}, J =6.9 Hz), 7.58 t (2H, H_{arom}, J = 7.8 Hz), 7.70 t (1H, H_{arom} , J = 7.5 Hz), 7.94 d (2H, H_{arom} , J = 8.4 Hz). ^{13}C NMR spectrum, δ_C , ppm: 13.8 (CH₃), 48.9 (C⁴), 51.2 (OCH₃), 99.9 (C⁵), 107.6 (C³); 127.1 (2C, C_{arom}), 127.7 (C_{arom}), 129.0 (2C, C_{arom}), 129.2 (2C, C_{arom}), 129.5 (2C, Carom), 134.6 (Carom), 135.5 weak (Carom), 140.5 weak (C_{arom}), 164.3 (C^2), 167.4 (C=O). Found, %: C 63.61; H 5.13; S 8.99. C₁₉H₁₈O₅S. Calculated, %: C 63.67; H 5.06; S 8.95. Methyl (2Z,3Z)-4-(benzenesulfonyl)-2-(1-hydroxyethylidene)-3-phenylbut-2-enoate (XIIc). Yield 17%, R_f 0.24, mp 118–119°C (from acetone– hexane). The NMR spectra of XIIc were identical to those given in [18]. Methyl trans-2-methyl-5-(4-methylbenzenesul-

49%, R_f 0.37, mp 148–149°C (from EtOAc-hexane).

IR spectrum, v, cm⁻¹: 1701 v.s, 1659 s, 1458 m,

1451 m, 1439 m, 1327 s, 1308 m, 1238 m, 1208 m, 1188 m, 1150 m, 1088 m, 1022 m, 992 m, 814 m,

760 m, 721 m, 706 m, 687 m, 594 s, 540 m. ¹H NMR

spectrum, δ , ppm: 2.28 d (3H, CH₃, J = 1.4 Hz), 3.55 s

fonyl)-4-phenyl-4,5-dihydrofuran-3-carboxylate (IVd). Yield 34%, R_f 0.54, mp 153–154°C (from acetone-hexane). IR spectrum, v, cm⁻¹: 1697 m, 1659 v.s. 1435 m, 1335 m, 1323 m, 1311 m, 1215 m, 1184 s, 1153 s, 1088 s, 1018 s, 694 m, 660 v.s, 590 v.s, 528 m, 521 m. ¹H NMR spectrum, δ , ppm: 2.28 d (3H, CH_3 , J = 1.4 Hz), 3.54 s (3H, OCH₃), 4.89 d.q (1H, 4-H, J = 1.4, 3.9 Hz), 5.08 d (1H, 5-H, J = 3.9 Hz), 7.19 d (2H, H_{arom} , J = 7.0 Hz), 7.23–7.27 m (1H, Harom), 7.29-7.33 m (2H, Harom), 7.36 d and 7.81 d (2H each, H_{arom}, J = 8.2 Hz). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 13.8 (CH₃), 21.7 (CH₃C₆H₄), 48.8 (C⁴), 51.2 (OCH₃), 99.9 (C⁵), 107.6 (C³), 127.1 (2C, C_{arom}), 127.6 (C_{arom}), 128.9 (2C, Carom), 129.5 (2C, Carom), 129.9 (2C, Carom), 132.4 weak (Carom), 140.6 weak (Carom), 145.8 weak (C_{arom}), 164.4 (C²), 167.4 (C=O). Found, %: C 64.57; H 5.53; S 8.68. C₂₀H₂₀O₅S. Calculated, %: C 64.50; H 5.41; S 8.61.

Methyl (2Z,3Z)-2-(1-hydroxyethylidene)-4-(4-methylbenzenesulfonyl)-3-phenylbut-2-enoate (XIId). Yield 35%, R_f 0.38, mp 109–110°C (from acetone–hexane). The NMR spectra of XIId were identical to those given in [18].

Methyl 1-acetyl-2*c*-(benzenesulfonyl)cyclopropane-1*r*-carboxylate (IXa). Yield 30%, R_f 0.25, mp 83–84°C (from EtOAc–hexane). IR spectrum, v, cm⁻¹: 1748 v.s, 1709 s, 1443 m, 1362 m, 1350 m, 1331 m, 1312 s, 1296 s, 1223 s, 1204 s, 1173 s, 1154 s, 1127 m, 737 m, 725 m, 598 v.s. ¹H NMR spectrum, δ, ppm: 1.63 d.d (1H, H_A , J = 5.2, 9.0 Hz), 2.22 d.d (1H, H_M , J = 5.2, 7.1 Hz), 2.33 s (3H, CH₃), 3.35 d.d (1H, H_X , J = 7.1, 9.0 Hz), 3.92 s (3H, OCH₃), 7.59 t (2H, H_{arom} , J = 7.2 Hz), 7.68 t (1H, H_{arom} , J = 7.9 Hz), 7.92 d (2H, H_{arom}, J = 6.3 Hz). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 19.5 (C³), 28.4 (CH₃), 43.0 weak (C¹), 45.8 (C²), 53.5 (OCH₃), 127.8 (2C, C_{arom}), 129.3 (2C, C_{arom}), 134.1 (Carom), 139.6 weak (Carom), 166.8 (C=O), 199.0 (MeC=O). Mass spectrum, m/z (I_{rel} , %): 282 (0.7) $[M]^+$, 267 (13), 240 (43), 209 (29), 208 (100), 141 (25), 125 (34), 116 (20), 105 (15), 99 (16), 77 (51), 59 (13), 51 (16), 43 (63). Found, %: C 55.37; H 5.05; S 11.46. C₁₃H₁₄O₅S. Calculated, %: C 55.31; H 5.00; S 11.36.

Methyl 1-acetyl-2t-(benzenesulfonyl)cyclopropane-1*r*-carboxylate (IXa'). Yield 36%, $R_{\rm f}$ 0.15, mp 80–81°C (from EtOAc-hexane). IR spectrum, v, cm⁻¹: 2924 weak, 1747 v.s, 1716 s, 1450 m, 1357 m, 1292 s, 1265 s, 1257 v.s, 1153 v.s, 1061 s, 729 s, 698 s, 544 s. ¹H NMR spectrum, δ , ppm: 1.66 d.d (1H, H₄, J = 5.5, 8.8 Hz), 2.16 d.d (1H, H_M, J = 5.5, 7.1 Hz), 2.57 s (3H, CH₃), 3.39 d.d (1H, H_X , J = 7.1, 8.8 Hz), 3.76 s (3H, OCH₃), 7.59 t (2H, H_{arom} , J = 7.0 Hz), 7.69 t (1H, H_{arom}, J = 7.5 Hz), 7.90 d (2H, H_{arom}, J =8.3 Hz). ¹³C NMR spectrum, δ_{C} , ppm: 18.1 (C³), 30.7 (CH₃), 42.1 (C¹), 46.0 (C²), 53.4 (OCH₃), 127.8 (2C, Carom), 129.4 (2C, Carom), 134.2 (Carom), 139.4 weak (Carom), 168.5 (C=O), 197.1 (MeC=O). Found, %: C 55.40; H 5.15; S 11.41. C₁₃H₁₄O₅S. Calculated, %: C 55.31; H 5.00; S 11.36.

Methyl 5-(benzenesulfonyl)-4,5-dihydrofuran-3carboxylate (Xa). Yield 10%, Rf 0.32, mp 122-123°C (acetone-hexane). IR spectrum, v, cm⁻¹: 2985 m, 1701 v.s. 1662 v.s. 1446 s. 1435 s. 1342 s. 1327 v.s. 1254 s, 1153 s, 1142 s, 1084 s, 1018 s, 733 s, 586 v.s. ¹H NMR spectrum, δ , ppm: 2.11 t (3H, CH₃, J =1.8 Hz), 3.20-3.30 m and 3.38-3.48 m (1H each, CH₂), 3.67 s (3H, OCH₃), 5.34 d.d (1H, 5-H, J = 5.1, 10.9 Hz), 7.57 t (2H, H_{arom} , J = 7.8 Hz), 7.69 t (1H, H_{arom} , J = 7.5 Hz), 7.92 d (2H, H_{arom} , J = 8.4 Hz). ¹³C NMR spectrum, δ_{C} , ppm: 13.5 (CH₃), 30.8 (C⁴), 51.2 (OCH₃); 93.0 (C⁵), 102.9 (C³), 129.2 (2C, C_{arom}), 129.5 (2C, Carom), 134.6 (Carom), 135.2 weak (Carom), 164.6 (C²), 166.4 (C=O). Mass spectrum, m/z (I_{rel} , %): 268 (0.6), 251 (22), 250 (13), 209 (14), 208 (44), 186 (32), 176 (13), 171 (16), 141 (100), 140 (22), 125 (42), 116 (14), 115 (11), 109 (20), 78 (11), 77 (83), 67 (14), 59 (23), 43 (81). Found, %: C 55.38; H 5.09; S 11.50. C₁₃H₁₄O₅S. Calculated, %: C 55.31; H 5.00; S 11.36.

Methyl 1-acetyl-2*c***-(methanesulfonyl)cyclopropane-1***r***-carboxylate (IXb). Yield 21%, R_f 0.43 (light petroleum ether–EtOAc, 1:1), mp 93–94°C (from EtOAc–hexane). IR spectrum, v, cm⁻¹: 1716 s, 1662 s, 1446 m, 1385 m, 1338 m, 1311 s, 1257 s, 1219 m, 1196 m, 1161 m, 1142 v.s, 1080 s, 1026 m, 1003 m, 914 s, 779 m, 760 m, 513 m, 482 m. ¹H NMR spectrum, \delta, ppm: 1.64 d.d (1H, H_A, J = 5.6, 8.9 Hz), 2.19 d.d (1H, H_M, J = 5.6, 6.6 Hz), 2.33 s (3H, CH₃), 3.03 s (3H, SO₂CH₃), 3.28 d.d (1H, H_X, J = 6.6, 8.9 Hz), 3.82 s (3H, OCH₃). ¹³C NMR spectrum, \delta_C, ppm: 16.3 (C³), 30.5 (CH₃), 40.3 (C¹), 42.0 (C²), 43.6 (SO₂CH₃), 53.6 (OCH₃), 168.4 (C=O), 198.3 (MeC=O). Found, %: C 43.54; H 5.54; S 14.62. C₈H₁₂O₅S. Calculated, %: C 43.63; H 5.49; S 14.56.**

Methyl 1-acetyl-2*t*-(**methanesulfonyl**)**cyclopropane-1***r*-**carboxylate (IXb').** Yield 40%, R_f 0.35 (light petroleum ether–EtOAc, 1:1), oily substance. IR spectrum, v, cm⁻¹: 1747 m, 1740 s, 1732 s, 1724 v.s, 1716 s, 1311 s, 1257 s, 1142 s. ¹H NMR spectrum, δ , ppm: 1.80 d.d (1H, H_A, J = 5.4, 9.0 Hz), 2.18 d.d (1H, H_M, J = 5.4, 6.7 Hz), 2.42 s (3H, CH₃), 3.08 s (3H, SO₂CH₃), 3.33 d.d (1H, H_X, J = 6.7, 9.0 Hz), 3.77 s (3H, OCH₃). ¹³C NMR spectrum, δ , ppm: 17.6 (C³), 28.3 (CH₃), 41.5 weak (C¹), 42.3 (C²), 43.4 (SO₂CH₃), 53.5 (OCH₃), 166.8 (C=O), 198.8 (MeC=O). Found, %: C 43.59; H 5.53; S 14.64. C₈H₁₂O₅S. Calculated, %: C 43.63; H 5.49; S 14.56.

Methyl 5-(methanesulfonyl)-4,5-dihydrofuran-3-carboxylate (Xb). ¹H NMR spectrum (from the spectrum of the reaction mixture), δ , ppm: 2.24 d (3H, CH₃, J = 1.8 Hz), 2.91 s (3H, SO₂CH₃), 3.70 s (3H, OCH₃), 5.31 d.d (1H, 5-H, J = 5.5, 11.0 Hz).

Methyl 1-acetyl-2*c*-(phenylmethanesulfonyl)cyclopropane-1*r*-carboxylate (IXc). Yield 10%, R_f 0.22, oily substance. IR spectrum, v, cm⁻¹: 1740 v.s, 1720 v.s, 1335 s, 1308 v.s, 1257 v.s, 1153 v.s, 1122 s, 598 m, 544 m. ¹H NMR spectrum, δ , ppm: 1.57 d.d (1H, H_A, J = 5.4, 9.1 Hz), 2.13 d.d (1H, H_M, J = 5.4, 6.9 Hz), 2.33 s (3H, CH₃), 2.97 d.d (1H, H_X, J = 6.9, 9.1 Hz), 3.87 s (3H, OCH₃), 4.30 d and 4.44 d (1H each, CH₂SO₂, J = 13.7 Hz), 7.38 br.s (5H, H_{arom}). ¹³C NMR spectrum, δ_C , ppm: 17.4 (C³), 28.1 (COMe), 35.6 (C²), 42.0 weak (C¹), 53.5 (OMe), 60.0 (CH₂SO₂), 127.1 (C_{arom}), 128.9 (2C, C_{arom}), 129.1 (C_{arom}), 130.7 (2C, C_{arom}), 167.0 (C=O), 198.5 (MeC=O). Found, %: C 56.57; H 5.38; S 10.80. C₁₄H₁₆O₅S. Calculated, %: C 56.74; H 5.44; S 10.82.

Methyl 1-acetyl-2t-(phenylmethanesulfonyl)cyclopropane-1r-carboxylate (IXc'). Yield 36%, $R_{\rm f}$ 0.20, mp 87–88°C (from EtOAc-hexane). IR spectrum, v, cm⁻¹: 3048 m, 1736 s, 1728 s, 1709 s, 1439 s, 1335 s, 1316 s, 1308 v.s, 1300 v.s, 1265 v.s, 1258 v.s, 1123 s, 1115 v.s, 795 m, 698 s, 536 s, 478 s. ¹H NMR spectrum, δ , ppm: 1.69 d.d (1H, H_A, J = 5.5, 9.1 Hz), 2.06 d.d (1H, H_M, J = 5.5, 6.7 Hz), 2.43 s (3H, CH₃), 3.01 d.d (1H, H_X, J = 6.7, 9.1 Hz), 3.76 s (3H, OCH₃), 4.34 d and 4.47 d (1H each, CH₂SO₂, J = 13.7 Hz), 7.35–7.41 m (3H, H_{arom}), 7.42–7.46 (2H, H_{arom}). ¹³C NMR spectrum, δ_{C} , ppm: 16.4 (C³), 30.5 (COMe), 40.0 (C¹), 41.1 (C²), 53.5 (OMe), 60.6 (CH₂SO₂), 127.2 weak (C_{arom}), 129.0 (2C, C_{arom}), 129.2 (C_{arom}), 130.9 (2C, C_{arom}), 68.3 (C=O), 198.4 (MeC=O). Found, %: C 56.55; H 5.35; S 10.71. C₁₄H₁₆O₅S. Calculated, %: C 56.74; H 5.44; S 10.82.

Methyl 2t-(phenylmethanesulfonyl)cyclopropane-1r-carboxylate (XIc). Yield 2%, R_f 0.13, mp 116– 117°C (from EtOAc–hexane). IR spectrum, v, cm⁻¹: 2947 w, 1732 v.s, 1319 s, 1296 s, 1223 s, 1130 s, 910 m, 729 m, 702 m. ¹H NMR spectrum, δ , ppm: 1.32–1.38 m (1H, H_A), 1.96–2.01 m (1H, H_Y), 2.19– 2.25 m (1H, H_X), 2.38–2.44 m (1H, H_M), 3.80 s (3H, CH₃), 4.30 d and 4.42 d (1H each, CH₂SO₂, J =13.7 Hz), 7.35–7.42 m (3H, H_{arom}), 7.46–7.53 (2H, H_{arom}). ¹³C NMR spectrum, δ_C , ppm: 8.3 (C³), 15.4 (C¹), 22.3 (C²), 52.9 (OMe), 60.2 (CH₂SO₂), 127.8 weak (C_{arom}), 129.0 (3C, C_{arom}), 130.9 (2C, C_{arom}), 167.7 (C=O). Found, %: C 56.62; H 5.58; S 12.69. C₁₂H₁₄O₄S. Calculated, %: C 56.68; H 5.55; S 12.61.

Methyl 5-(phenylmethanesulfonyl)-4,5-dihydrofuran-3-carboxylate (Xc). Yield 7%, Rf 0.27, mp 133-134°C (from EtOAc-hexane). IR spectrum, v, cm⁻¹: 2955 w, 1701 v.s, 1662 v.s, 1435 m, 1323 s, 1273 s, 1223 s, 1134 s, 1088 s, 1018 m, 760 m, 505 m. ¹H NMR spectrum, δ , ppm: 2.29 d.d (3H, CH₃, J = 1.4, 1.6 Hz), 3.04–3.15 m and 3.37–3.45 m (1H each, 4-H), 3.71 s (3H, OCH₃), 4.23 d and 4.48 d (1H each, CH_2SO_2 , J = 14.1 Hz), 5.19 d.d (1H, 5-H, J = 5.5, 11.0 Hz), 7.37-7.45 m (5H, H_{arom}). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 13.7 (CH₃), 28.9 (C⁴), 51.3 (OCH₃), 55.7 (CH₂SO₂), 87.6 (C⁵), 103.3 (C³), 126.9 weak (Carom), 129.1 (2C, Carom), 129.2 (Carom), 130.9 (2C, C_{arom}), 164.7 (C²), 166.2 (C=O). Found, %: C 56.82; H 5.41; S 10.71. C₁₄H₁₆O₅S. Calculated, %: C 56.74; H 5.44; S 10.82.

REFERENCES

- 1. Verhé, R. and De Kimpe, N., *The Chemistry of the Cyclopropyl Group*, Patai, S. and Rappoport, Z., Eds., Chichester: Wiley, 1987, vol. 1, chap. 9, p. 445.
- Joucla, M. and Le Brun, J., *Tetrahedron Lett.*, 1985, vol. 26, p. 3001.

- 3. Crossland, I., Bock, K., and Norrestam, R., *Acta Chem. Scand.*, 1985, vol. 39, p. 7.
- Vardapetyan, A.A., Khachatryan, D.S., Panosyan, G.A., and Morlyan, N.M., *Zh. Org. Khim.*, 1986, vol. 22, p. 2262.
- Arai, S., Nakayama, K., Hatano, K., and Shioiri, T., J. Org. Chem., 1998, vol. 63, p. 9572.
- 6. Arai, S., Nakayama, K., Ishida, T., and Shioiri, T., *Tetrahedron Lett.*, 1999, vol. 40, p. 4215.
- Farina, F., Maestro, M.C., Martin, M.V., and Soria, M.L., *Tetrahedron*, 1987, vol. 43, p. 4007.
- 8. Hagiwara, H., Sato, K., Suzuki, T., and Ando, M., *Tetrahedron Lett.*, 1997, vol. 38, p. 2103.
- 9. Hagiwara, H., Sato, K., Nishino, D., Hoshi, T., Suzuki, T., and Ando, M., J. Chem. Soc., Perkin Trans. 1, 2001, p. 2946.
- Yamamoto, I., Sakai, T., Ohto, K., Matsuzaki, K., and Fukuyama, K., J. Chem. Soc., Perkin Trans. 1, 1985, p. 2785.
- 11. Vasin, V.A., Bolusheva, I.Yu., Razin, V.V., and Somov, N.V., *Russ. J. Org. Chem.*, 2014, vol. 50, p. 993.
- 12. Vasin, V.A., Bolusheva, I.Yu., and Razin, V.V., *Chem. Heterocycl. Compd.*, 2008, vol. 44, no. 4, p. 419.
- Yamamoto, I., Saluchi, N., Futaesaku, N., Fujimoto, K., Fujimoto, T., and Ohta, K., *J. Chem. Res. (M)*, 1992, p. 656.
- 14. Evans, P. and Taylor, R.J.K., Synlett, 1997, p. 1043.
- 15. Alonso, M.E. and Morales, A., J. Org. Chem., 1980, vol. 45, p. 4530.
- 16. Günther, H., *NMR Spectroscopy: An Introduction*, Chichester: Wiley, 1980, chap. IV.2.
- 17. Solladie-Cavallo, A. and Isarno, T., *Tetrahedron Lett.*, 1999, vol. 40, p. 1579.
- 18. Vasin, V.A., Bolusheva, I.Yu., Razin, V.V., and Somov, N.V., *Russ. J. Org. Chem.*, 2013, vol. 49, p. 1257.
- Eliel, E.L., Wilen, S.H., and Doyle, M.P., Basic Organic Stereochemistry, New York: Wiley, 2001. Translated under the title Osnovy organicheskoi stereokhimii, Moscow: Binom. Laboratoriya Znanii, 2007, p. 453.
- Evans, P. and Taylor, R.J.K., J. Sulfur Chem., 2005, vol. 26, p. 481.
- Gaillot, J.-M., Gelas-Mialhe, Y., and Vessière, R., *Can. J. Chem.*, 1979, vol. 57, p. 1958.
- 22. Harwood, L.M., Julia, M., and Le Thuillier, G., *Tetrahedron*, 1980, vol. 36, p. 2483.
- Inomata, K., Sasaoka, S., Kobayashi, T., Tanaka, Y., Igarashi, S., Ohtani, T., Kinoshita, H., and Kotake, H., *Bull. Chem. Soc. Jpn.*, 1987, vol. 60, p. 1767.
- 24. Carlier, P., Gelas-Mialhe, Y., and Vessière, R., *Can. J. Chem.*, 1977, vol. 55, p. 3190.
- Makosza, M., Podraza, R., and Bialecki, M., *Gazz. Chim. Ital.*, 1995, vol. 125, p. 601.

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