PALLADIUM-CATALYZEDCROSS-COUPLINGREACTIONSOFBROMINATEDPORPHYRINSWITHFUNCTIONALIZEDORGANOMAGNESIUMREAGENTS:DIRECTPREPARATIONOFFUNCTIONAL-GROUP-BEARING FREEBASE PORPHYRINS

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Abstract – The direct preparation of free-base porphyrins possessing reactive functional groups such as esters, halides, amide, nitriles, and acetals is described. This method relies on a simple one-pot procedure that involves sequential palladium-catalyzed cross-coupling of brominated free-base porphyrins with functionalized aryl and alkenylmagnesium reagents, which are readily prepared by iodine–magnesium exchange of the corresponding organic iodides with the turbo Grignard reagent *i*-PrMgCl·LiCl, followed by demetallation of the resulting magnesium porphyrins under very mild conditions involving either a 1.2 M aqueous methanol solution of HCl or a 0.1 M methanol solution of citric acid at ambient temperature.

INTRODUCTION

Porphyrins are becoming increasingly important target structures since compounds of this class have widespread use in many fields¹ including catalysis,² medicine,³ materials,⁴ and molecular recognition/sensing.⁵ The physical, chemical, and biological properties of porphyrins can be fine-tuned or dramatically changed by the selection of the peripheral substituents having different electronic and steric properties and by the central metal ions coordinated to the nitrogen core.¹ Accordingly, significant interest has arisen in developing new synthetic intermediates and strategies for preparing porphyrin derivatives that possess various peripheral substituents.^{6,7} In a related contribution, we previously reported an approach to prepare functionalized porphyrins via palladium-catalyzed Negishi cross-coupling of bromoporphyrins H_2 -1 with functional-group-bearing organozinc reagents 2 (Scheme 1).⁸ The functionalized zinc porphyrin products **Zn-3** of the reaction, where the peripheral substituents of

This work is dedicated to Professor Dr. Lutz F. Tietze on the occasion of his 75th birthday.



Scheme 1. Preparation of functional-group-bearing free-base porphyrins

Zn-3 bear chemically reactive function- alities such as ester, amide, halogen, and nitrile groups, could serve as synthetic intermediates for more complicated porphyrin derivatives because these reactive functional groups can provide a convenient handle for further functionalization. However, when the desired product is not zinc porphyrin complexes but other metalloporphyrins, multiple chemical transformations are necessary to remove the central zinc metal from Zn-3 or its zincated derivatives and install other metals into the resulting free base. In addition, the demetallation of zinc porphyrins typically requires the use of strong acidic conditions such as those afforded by CF₃CO₂H and conc. H₂SO₄, which might lead to undesired decomposition of the substrates. Therefore, from the perspective of synthetic divergence and step economy, functional-group-bearing free-base porphyrins H₂-3, a demetallated analog of zinc porphyrins Zn-3, would be more attractive as an efficient and versatile synthetic intermediate in the construction of complicated porphyrin derivatives. The use of these free-base porphyrins H_{2-3} as a synthetic building block for the construction of complex porphyrin derivatives should enable the insertion of desired metal ions even in the case of the late-stage functionalization of porphyrins in a multistep synthetic sequence in order to avoid involving a dezincation process under harsh conditions. Despite recent significant advances in the transition-metal-catalyzed porphyrin functionalization with organometallic reagents, a majority of the reported methods afford metallated porphyrins, such as zinc(II), nickel(II), and copper(II) porphyrins, as the products even when free-base porphyrins are used as starting

substrates.⁶⁻⁹ Thus, a novel, mild, and direct method that can provide functional-group-bearing free-base

porphyrins including H₂-3 remains in demand.

Recently, Knochel's aryl and alkenylmagnesium reagents **4**, which are readily prepared in situ from the corresponding aryl and alkenyl iodides with the turbo Grignard reagent *i*-PrMgCl·LiCl, have garnered considerable interest in synthetic chemistry because of their high synthetic potential on the basis of their facile availability and significant functional tolerability.¹⁰ However, despite their numerous elegant applications in laboratory synthesis and up-scaled industrial processes, to the best of our knowledge, the literature contains no account of the functionalization of porphyrins utilizing these organomagnesium reagents **4**.⁷ We expected that the transition-metal-catalyzed cross-coupling reaction of halogenated free-base porphyrin **H**₂-**1** with a functionalized organomagnesium reagent **4** would initially provide the corresponding functionalized magnesium porphyrin **Mg-3** (Scheme 1). Unlike zinc, nickel, and copper porphyrins,¹¹ magnesium porphyrins are highly sensitive toward acidic conditions.^{9,12,13} Therefore, the resulting magnesium complex **Mg-3** would be easily transformed into the corresponding free-base porphyrins **4**₂-**3** under very mild conditions without destruction of the acid-sensitive peripheral functional groups.

Here, we present an efficient, direct method for preparing functional-group-bearing free-base porphyrins H_2 -3 via a simple one-pot procedure involving palladium-catalyzed cross-coupling reaction of brominated free-base porphyrins H_2 -1 with functionalized aryl and alkenylmagnesium reagents 4, followed by demetallation of the resulting magnesium porphyrins Mg-3 under very mild acidic conditions using either a 1.2 M aqueous methanol solution of HCl or a 0.1 M methanol solution of citric acid at ambient temperature. The exceptional mildness of these reaction conditions offers high functional tolerability to the present method, thereby providing the desired free-base porphyrins H_2 -3 bearing reactive functional groups such as esters, nitriles, amides, silyls, and acetals in good to high yields.

RESULTS AND DISCUSSION

In a preliminary experiment, we examined the cross-coupling of 5-bromo-10,15,20-triphenylporphyrin **H**₂-1**a** with 3-(ethoxycarbonyl)phenylmagnesium chloride lithium chloride 4**a**, which was generated by a metallation of ethyl 3-iodobenzoate 5**a** using the turbo Grignard reagent *i*-PrMgCl·LiCl, in the presence of Pd(Ph₃P)₄ as a catalyst in THF at 25 °C (Scheme 2). Upon completion of the reaction, the mixture was concentrated in vacuo and chromatographed on silica gel to give the desired *meso*-functionalized free-base porphyrin **H**₂-3**aa** along with the debrominated free base **H**₂-6**a** and their magnesium complexes **Mg-3aa** and **Mg-6a**. These magnesium complexes **Mg-3aa** and **Mg-6a** were quantitatively transformed to the corresponding free bases **H**₂-3**aa** and **H**₂-6**a** by hydrolysis upon treatment with a 1.2 M aqueous



Scheme 2. Palladium-catalyzed coupling of free-base bromoporphyrins H₂-1 with functionalized arylmagnesium reagents 4a and demagnesiation of Mg-3aa and Mg-6a with a 1.2 M aqueous methanol solution of HCl

Table 1. Ligand screening for the $Pd(OAc)_2$ -catalyzed coupling of meso-bromoporphyrin H2-1a withfunctionalized aryImagnesium reagent 4a

Ph-{	Ph Ph Br Ph Br Ph H ₂ -1a	5a D ₂ C ↓ <i>i</i> -PrMgCI•LiCI →-MgCI•LiCI 4a (10 equiv) 5 mol% Pd(OAc) ₂ 10 mol% Ligand THF, 25 °C, 12 h then 1.2M HCI in H ₂ O/MeOH 25 °C, 5 min	Ph NH N= Ph N HN- Ph H ₂ -3aa	CO ₂ Et + F	Ph NH N= NHN Ph Hn H_2-6a
entry	ligand	yield ^a (%)	yield ^a (%)	\sim	
		of H2-3aa	of H2-6a		
1	rac-BIN.	AP 27	61	$H_{2}CO$	<i>i</i> -Pr
2	Cy ₃ P	38	58		
3	$(t-Bu)_3P$	51	38	SPhos	s <i>i-</i> Pr XPhos
4	Ph ₃ P	45	49		
5	SPhos	75	21		P(<i>t</i> -Bu) ₂
6	XPhos	62	24	/-Pr	<i>I-</i> P I
7	t-BuXPh	los 83	12	i-P	<i>t-</i> BuXPhos
^a Isolated	l vield				

^{*a*} Isolated yield.

methanol solution of HCl. As expected, these cross-coupling and demetallation processes could be conducted in a one-pot manner. Thus, the sequential one-pot treatment of *meso*-bromoporphyrin H₂-1a with 10 equiv of arylmagnesium reagent 4a in the presence of Pd(Ph₃P)₄ (5 mol%) in THF followed by the addition of a 1.2 M aqueous methanol solution of HCl at 25 °C afforded the desired functionalized base porphyrin H₂-3aa and debrominated free base H₂-6a in 55% and 38% yields, respectively, without their magnesium complexes Mg-3aa and Mg-6a.

Having established demetallation conditions, we explored the catalyst systems on the basis of the combination of $Pd(OAc)_2$ and different phosphine ligands for optimizing the formation of the functionalized base porphyrin **H**₂-**3aa**. As shown in Table 1, ligand screening indicated that bidentate phosphine ligand was less effective (entry 1) and the steric hindrance of monodentate phosphine ligands greatly influenced the catalytic reactivity (entries 2–6). Ultimately, *t*-BuXPhos was the most effective ligand among those screened as it afforded the desired free-base porphyrin **H**₂-**3aa** in 83% isolated yield without erosion of the ester functionality and significantly reduced the formation of the debrominated free-base byproduct **H**₂-**6a** to 12% yield.

Having developed an optimum catalyst system, we examined the substrate scope of the method for the preparation of functional-group-bearing free-base porphyrins. The examples described in Table 2 indicate that the sequential method involving the palladium-catalyzed cross-coupling followed by demetallation with a 1.2 M aqueous methanol solution of HCl can be readily applied to the cross-coupling reactions of the functionalized arylmagnesium reagent **4a** with a variety of di- and tri-substituted *meso*-bromoporphyrins **H**₂-**1**. The reaction tolerated well a range of substituents on the phenyl ring, including alkyl, vinyl, silylethynyl, and alkoxy groups, to give the corresponding functionalized free bases **H**₂-**3** in good to high yields (entries 5–10). Alkyl-substituted substrates **H**₂-**1b** and **H**₂-**1m** could also undergo the reaction to afford the corresponding functionalized free bases in good yields (entries 2 and 13). The present catalytic conditions were suitable for C₆F₅ and thienyl substituents on the *meso*-brominated porphyrins, and the desired products were obtained in good yields (entries 11 and 12). As shown in Scheme 3, performing the reaction with β -bromoporphyrin **H**₂-**1n** gave the desired β -functionalized product **H**₂-**3n** in 86% yield. Under similar conditions, double functionalization at the *meso* positions could also proceed smoothly using 5,15-dibromoporphyrin **H**₂-**1o** as the substrate (Scheme 4).

The scope of functionalized arylmagnesium reagents prepared in situ from the corresponding aryl iodides and *i*-PrMgCl·LiCl was then examined. As shown in Table 3, the mild conditions employed with the $Pd(OAc)_2/t$ -BuXPhos catalyst system and a 1.2 M aqueous methanol solution of HCl afforded a range of functional groups including ester, nitrile, halogenic, amino, and silyl groups on the phenyl ring of arylmagnesium reagents **4a**—**4l** to be tolerated in the reaction with *meso*-bromoporphyrin **H**₂-**1a**. The *ortho*-substitution was not consequential with respect to reaction yield (products **H**₂-**3ab** and **H**₂-**3ah**,

 $R^{2} + NHN = R^{1} + R^{1} + R^{1} + R^{1} + R^{2} + R^{2}$

entry	R^1	R ²	substrate	product	yield ^a (%)
1	Ph	Ph	1a	H ₂ -3aa	83
2	Ph	<i>n</i> -Bu	1b	H2-3ba	66
3	Ph	C_6F_5	1c	H ₂ -3ca	69
4	Ph	Н	1d	H ₂ -3da	75
5	<i>p</i> -tolyl	Н	1e	H2-3ea	70
6	3-(CH ₂ =CH)C ₆ H ₄	Н	1 f	H ₂ -3fa	81
7	$4-(i-\Pr_3SiC\equiv C)C_6H_4$	Н	1g	H ₂ -3ga	62
8	3-(MeO)C ₆ H ₄	Н	1h	H2-3ha	72
9	3,5-(<i>t</i> -Bu) ₂ C ₆ H ₃	Н	1i	H ₂ -3ia	64
10	2,4,6-Me ₃ C ₆ H ₂	Н	1j	H2-3ja	70
11	C_6F_5	Ph	1k	H ₂ -3ka	73
12	2-thienyl	Н	11	H2-3la	77
13	<i>i</i> -Bu	Н	1m	H2-3ma	69

^{*a*} Isolated yield.

62% and 64% yields). As expected, heteroarylmagnesium reagents **4k** and **4l** could also be used to access free-base porphyrins **H₂-3ak** and **H₂-3al** bearing pyridyl and thienyl groups, respectively, in good yields. To further demonstrate the generality of this novel preparation of functional-group-bearing free-base porphyrins, we sought to employ functional-group-bearing alkenylmagnesium reagents, which are readily prepared from alkenyl iodides and *i*-PrMgCl·LiCl in a similar manner for the preparation of arylmagnesium reagents (Table 4). With acyclic alkenylmagnesium reagents **4m**–**4r**, the present cross-coupling/demetallation method was again very effective in producing good yields of the desired free-base alkenylporphyrins **H₂-3am**–**H₂-3ar** containing various functional groups such as nitrile, ester, amide, and silyl functionalities, although a longer reaction time and an increased amount of the magnesium

Table 2. Palladium-catalyzed coupling of 4a with various *meso*-bromoporphyrins H₂-1



Scheme 3. Preparation of β -(3-ethoxycarbonylphenyl)-substituted porphyrin H₂-3na



Scheme 4. Preparation of bis(3-ethoxycarbonylphenyl)porphyrin H2-30a

reagents were required to complete the reaction. Notably, the stereochemistry of the parent alkenyl iodides were completely preserved during the cross-coupling/demetallation (entries 1–5); in contrast, a previous method for preparing these alkenylporphyrins via the palladium-catalyzed Heck reaction of *meso*-bromoporphyrins with vinyl derivatives such as acrylonitrile has resulted in the formation of an inseparable mixture of *trans* and *cis* isomers.¹⁴ As expected, bromo-substituted cyclic alkenylmagnesium reagent **4s** was also suitable for the reaction, affording the desired free base **H**₂**-3as** in 65% yield; the bromo functionality on the olefinic moiety was well tolerated (entry 7).

Finally, we examined the coupling reactions of *meso*-bromoporphyrin H₂-1a with arylmagnesium reagents 4t and 4u, which possess acetal units; these units are more challenging functionalities because of their highly sensitive nature toward acidic conditions. As illustrated in Schemes 5 and 6, arylmagnesium reagents of the protected phenol and aldehyde 4t and 4u, respectively, underwent the sequential cross-coupling/demetallation under optimal conditions using the $Pd(OAc)_2/t$ -BuXPhos catalyst system and a 1.2 M aqueous methanol solution of HCl. However, the free-base products obtained were not the

desired acetal derivatives H₂-3at and H₂-3au but deprotected phenol and aldehyde derivatives H₂-7 and H₂-8. We subsequently reinvestigated the demetallation conditions and observed that a 0.1 M methanol





solution of citric acid could be sufficient for the demetallation of the initially produced magnesium complexes yet compatible with the acetal functional groups. Thus, exposure of *meso*-bromoporphyrin

H₂-3a to arylmagnesium reagents **4t** and **4u** in the presence of $Pd(OAc)_2$ and *t*-BuXPhos followed by treatment with a 0.1 M methanol solution of citric acid afforded the corresponding adducts **H₂-3at** and **H₂-3au** in 78% and 76% isolated yields, respectively, leaving the highly acid-sensitive acetal groups untouched.







Scheme 5. Preparation of free-base porphyrin H₂-3at bearing a protected phenol functional group



Scheme 6. Preparation of free-base porphyrin H2-3au bearing a protected aldehyde functional group

In summary, we developed a facile, direct method for the preparation of free-base porphyrins that include reactive functionalities such as esters, nitriles, amides, halides, silyls, and acetals. The procedure involves a sequential palladium-catalyzed cross-coupling of free-base bromoporphyrins with functionalized aryl and alkenylmagnesium reagents, which are readily available from their iodide precursor and the turbo Grignard reagent *i*-PrMgCl·LiCl, followed by demagnesiation of the initially formed magnesium porphyrins with either a 1.2 M aqueous methanol solution of HCl or a 0.1 M methanol solution of citric acid at room temperature to give the corresponding functional-group-bearing free-base porphyrins will be a valuable addition to the building blocks available in the total synthesis of more complicated porphyrin systems, thereby allowing the insertion of desired metal ions for later transformation in a multistep synthetic sequence while preventing the demetallation process under harsh conditions. We are currently conducting further synthetic studies using these functional-group-bearing free-base porphyrins, and the results will be reported in due course.

EXPERIMENTAL

General ¹H and ¹³C NMR spectra were recorded at rt on 300, 400, and 500 MHz spectrometers using perdeuterated solvents as internal standards. Chemical shifts of ¹H and ¹³C spectra are given in ppm relative to residual protiated solvent and relative to the solvent respectively. ¹⁹F NMR spectra were recorded at rt on a 500 MHz spectrometer using benzotrifluoride as an external standard. The chemical shift values are expressed as δ values (ppm) and the couple constants values (*J*) are in Hertz (Hz). The following abbreviations were used for signal multiplicities: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; and br, broad. UV–visible spectra were recorded using a dual-beam grating spectrophotometer

with a 1 cm quartz cell. The melting point data were not available for the porphyrin derivatives obtained because these compounds are infusible below 300 °C.

Reactions involving moisture sensitive reagents were carried out under an argon atmosphere using standard vacuum line techniques and glassware that was flame-dried and cooled under argon before use. Dry THF was purchased for the reactions and used without further desiccation. Bromoporphyrin derivatives, H₂-1a-H₂-1b,⁸ H₂-1d-H₂-1h,⁹ H₂-1i,⁸ H₂-1j,⁹ H₂-1m,⁸ H₂-1n,⁹ and H₂-1o,⁸ were prepared according to the method described in literature. Other chemicals were purchased from commercial sources and used as received unless stated otherwise.

Preparation of 5,15-diphenyl-10-pentafluorophenylporphyrin. [5,15-Diphenyl-10-pentafluorophenylporphyrinato]zinc(II)¹⁵ (830 mg, 1.200 mmol) was dissolved in a mixture of CH₂Cl₂ (60 mL) and MeOH (9 mL) and the solution was treated with trifluoroacetic acid (15 mL) at rt for 30 min. The solution was diluted with CH₂Cl₂ (300 mL) and washed with saturated NaHCO₃ (150 mL \times 2) and H₂O (150 mL). The solution dried over MgSO₄ and removal of the solvent organic was offered 5,15-diphenyl-10-pentafluorophenylporphyrin (754 mg, quant.) as a purple solid. $R_{\rm f} = 0.55$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.21 (1H, s), 9.29 (2H, d, J = 4.8 Hz), 9.02 (2H, d, J =4.8 Hz), 9.01 (2H, d, J = 4.8 Hz), 8.84 (2H, d, J = 4.8 Hz), 8.29–8.20 (4H, m), 7.85–7.73 (6H, m), -2.90 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 147.8, 147.3, 146.8 (2C, d, J_{CF} = 246.0 Hz), 145.9, 145.5, 142.0 $(1C, d, J_{CF} = 255.9 \text{ Hz}), 141.5, 137.6 (2C, d, J_{CF} = 254.7 \text{ Hz}), 134.7, 132.3, 131.8, 131.5, 129.2, 128.0,$ 126.9, 120.5, 117.4, 106.6, 100.7; ¹⁹F NMR (CDCl₃, 466 MHz) δ –138.3 (2F, ddd, J_{FF} = 23.1, 8.0, 5.9), – 154.5 (1F, tt, $J_{\text{FF}} = 20.5$, 8.1 Hz), -163.8 (2F, ddd, $J_{\text{FF}} = 23.1$, 20.5, 8.5 Hz); IR (KBr) 3298, 3116, 3062, 1496, 1068, 984, 922, 856, 791, 725 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 417.0 (5.7), 546.5 (4.3) nm ; HRMS (EI) calcd for C₃₈H₂₂F₅N₄ 629.1765, found 629.1769.

Preparation of 5-bromo-10,20-diphenyl-15-pentafluolophenylporphyrin (H2-1c). To a solution of 5,15-diphenyl-10-pentafluorophenylporphyrin (629 mg, 1.000 mmol) in CHCl₃ (200 mL) was added NBS (196 mg, 1.100 mmol, 1.1 equiv) at rt. After stirring for 15 min, the solution was concentrated in vacuo. Column chromatography on silica gel (5:1 hexane/toluene) followed by recrystallization from hexane/CH₂Cl₂ gave the product **H2-1c** (672 mg, 95%) as a purple solid. $R_{\rm f}$ = 0.65 (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 9.65 (2H, d, *J* = 4.9 Hz), 8.89 (2H, d, *J* = 4.9 Hz), 8.87 (2H, d, *J* = 4.9 Hz), 8.75 (2H, d, *J* = 4.9 Hz), 8.21–8.14 (4H, m), 7.81–7.73 (6H, m), -2.63 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 147.2 (8C, br), 146.7 (2C, d, *J*_{CF} = 243.3 Hz), 142.1 (1C, d, *J*_{CF} = 256.6 Hz), 141.5, 137.7 (2C, d, *J*_{CF} = 254.9 Hz), 134.6, 133.0, 132.6, 132.4, 129.2, 128.1, 126.9, 121.6, 116.7, 105.2, 101.1; ¹⁹F NMR (CDCl₃, 466 MHz) δ –136.6 (2F, ddd, *J*_{FF} = 24.0, 8.9, 6.8), -152.3 (1F, tt, *J*_{FF} = 21.4, 6.8 Hz), -161.8 (2F, ddd, *J*_{FF} = 24.0, 21.4, 8.0 Hz); IR (KBr) 3321, 3059, 3028, 1489, 1346, 980, 926, 791, 717 cm⁻¹; UV/vis

(CHCl₃) λ_{max} (log ε) 419.0 (5.6), 516.5 (4.2), 549.5 (3.6), 595.0 (3.4), 649.5 (2.9) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₃₈H₂₁BrF₅N₄ 707.0870, found 707.0875.

Preparation of [5,15-bis(pentafluorophenyl)-10-bromoporphyrinato]zinc(II). To a solution of [5,15-bis(pentafluorophenyl)porphyrinato]zinc(II)¹⁶ (690 mg, 0.978 mmol) in CHCl₃ (210 mL) and pyridine (0.8 mL) was added NBS (192 mg, 1.076 mmol, 1.1 equiv) at room temperature. After stirring for 15 min, the solution was concentrated in vacuo. Column chromatography on silica gel (5:1 hexane/toluene) followed by recrystallization from hexane/CH₂Cl₂ gave the [5,15-bis(pentafluorophenyl)-10-bromoporphyrinato]zinc(II) (430 mg, 56%) as a red-purple solid. $R_{\rm f} =$ 0.57 (2:1 hexane/THF); ¹H NMR (THF- d_8 , 300 MHz) δ 10.31 (1H, s), 9.84 (2H, d, J = 4.8 Hz), 9.47 (2H, d, J = 4.8 Hz), 9.09 (2H, d, J = 4.8 Hz), 9.08 (2H, d, J = 4.8 Hz); ¹³C NMR (THF- d_8 , 100 MHz) δ 152.4, 150.91, 150.86, 150.8, 147.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 252.8$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 248.5$ Hz), 143.0 (2C, d, $J_{CF} = 252.8$ Hz), 138.7 (4C, d, $J_{CF} = 252.8$ Hz), 143.0 (2C, d, J_{CF} = 252.8 Hz), 143.0 (252.2 Hz), 134.7, 134.4, 131.92, 131.87, 118.4, 107.9, 105.9, 103.2; ¹⁹F NMR (THF- d_8 , 466 MHz) δ – 142.9 (4F, ddd, $J_{FF} = 24.3$, 8.0, 6.1 Hz), -160.1 (2F, tt, $J_{FF} = 20.7$, 6.1 Hz), -168.7 (4F, ddd, $J_{FF} = 24.3$, 20.7, 8.2 Hz); IR (KBr) 3097, 3028, 2966, 2924, 2862, 2800, 2750, 1493, 1288, 1061, 987, 856, 775 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.5 (5.8), 550.0 (4.4), 584.0 (3.7) nm; HRMS-FAB⁺ (M⁺) calcd for C₃₂H₉BrF₁₀N₄Zn 781.9142, found 781.9146.

Preparation of 5,15-bis(pentafluorophenyl)-10-bromoporphyrin (H₂-1k). [5,15-Bis(pentafluorophenyl)-10-bromoporphyrinato]zinc(II) (300 mg, 0.382 mmol) was dissolved in a mixture of CH₂Cl₂ (20 mL) and MeOH (3 mL) and the solution was treated with trifluoroacetic acid (5 mL) at rt for 1.5 h. The solution was diluted with CH₂Cl₂ (100 mL) and washed with saturated NaHCO₃ (50 mL × 2) and H₂O (50 mL). The organic solution was dried over MgSO₄ and removal of the solvent offered pure compound H₂-1k (276 mg, quant.) as a purple solid. R_f = 0.65 (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.21 (1H, s), 9.80 (2H, d, *J* = 4.9 Hz), 9.35 (2H, d, *J* = 4.9 Hz), 8.88 (2H, d, *J* = 4.9 Hz), 8.86 (2H, d, *J* = 4.9 Hz), -3.00 (2H, br s); ¹³C NMR data was not available due to their poor solubility; ¹⁹F NMR (CDCl₃, 466 MHz) δ -138.4 (4F, ddd, J_{FF} = 23.2, 8.9, 6.8 Hz), -153.4 (2F, tt, J_{FF} = 19.6, 6.8 Hz), -163.3 (4F, ddd, J_{FF} = 23.2, 19.6, 8.7 Hz); IR (KBr) 3313, 3101, 3039, 2927, 1500, 1323, 1142, 1045, 980, 918, 771 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 412.5 (5.6), 507.5 (4.4), 540.5 (3.8) 585.0 (3.9), 640.5 (3.4) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₃₂H₁₂BrF₁₀N₄ 721.0086, found 721.0089; Anal. calcd for C₃₂H₁₁BrF₁₀N₄: C, 53.28; H, 1.54; N, 7.77, found: C, 53.32; H, 1.72; N, 7.52.

Preparation of 10-bromo-5,15-di(2-thienyl)porphyrin (H2-11). To a solution of 5,15-di-2-thienylporphyrin¹⁷ (382 mg, 0.805 mmol) in CHCl₃ (160 mL) was added NBS (158 mg, 0.886 mmol, 1.1 equiv) at rt. After stirring for 15 min, the solution was concentrated in vacuo. Column chromatography on silica gel (3:1 hexane/toluene) followed by recrystallization from hexane/CH₂Cl₂ gave the product **H**₂-**11** (275 mg, 62%) as a purple solid. $R_f = 0.55$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 10.08 (1H, s), 9.70 (2H, d, J = 4.9 Hz), 9.22 (2H, d, J = 4.9 Hz), 9.13 (2H, d, J = 4.9 Hz), 9.12 (2H, d, J = 4.9 Hz), 7.90 (2H, dd, J = 3.4, 1.3 Hz), 7.87 (2H, dd, J = 5.4, 1.3 Hz), 7.51 (2H, dd, J = 5.4, 3.4 Hz), -2.89 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 148.3 (4C, br), 146.8 (br), 145.8 (br), 142.2, 134.0, 132.7, 131.9, 131.8 (4C), 128.1, 126.3, 112.0, 106.1, 104.5; IR (KBr) 3309, 3105, 1400, 1323, 1242, 1049, 972, 791, 714 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.5 (5.6), 515.5 (4.3), 552.5 (4.0), 591.0 (3.8), 650.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₂₈H₁₈BrN₄S₂ 553.0156, found 553.0157.

Preparation of Functionalized Organomagnesium Reagents (4). All the organomagnesium reagents used were prepared according to the method described in literature¹⁸ as follows. To a well-stirred mixture of aryl or alkenyl iodide **5** (1.620 mmol) in 6.5 mL of THF was slowly added a solution of *i*PrMgCl·LiCl (1.250 mL, 1.620 mmol, 1.3 M in THF) at -20 °C. The mixture was stirred for 30 min at this temperature and was used immediately.

General Procedure for the Palladium-Catalyzed Reaction of Bromoporphyrins with Functionalized Organomagnesium Reagents. An oven-dried 100 mL two-necked flask equipped with a magnetic stirring bar and rubber septum was charged with a free base bromoporphyrin H₂-1 (0.162 mmol), Pd(OAc)₂ (1.8 mg, 8.1 μ mol, 5 mol%), and *t*-BuXPhos (6.9 mg, 16.2 μ mol, 10 mol%). The reaction vessel was evacuated and flushed with argon (three times), and then dry THF (30 mL) was added. To the solution was slowly added 0.2 M THF solution of an organomagnesium reagent 4 (8.0 mL, 1.620 mmol, 10 equiv) via cannula at rt. The mixture was stirred at rt (or 40 °C) for several hours (1–24 h), having been monitored by TLC (2:1 hexane/THF). To the solution of was added 5 mL of a 1.2 M aqueous methanol solution of HCl (a 1:9 mixture of conc. HCl and MeOH), and the solution was stirred for 5 min. The solvent was evaporated and the resulting solid was dissolved in CH₂Cl₂ (50 mL) and washed with water and brine. The organic layer was dried over MgSO₄ and concentrated in vacuo. Column chromatography on silica gel (1:1 toluene/hexane) followed by recrystallization from MeOH/CH₂Cl₂ gave the pure product H₂-3.

5-(3-Ethoxycarbonylphenyl)-10,15,20-triphenylporphyrin (H₂-**3aa**): Prepared from bromoporphyrin H₂-**1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Purple solid; 92.8 mg, 83% yield; $R_f = 0.31$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.90 (1H, s), 8.85 (6H, d, J = 4.9 Hz), 8.77 (2H, d, J = 4.9 Hz), 8.48 (1H, d, J = 7.9 Hz), 8.39 (1H, d, J = 7.5 Hz), 8.27– 8.15 (6H, m), 7.82 (1H, dd, J = 7.9, 7.5 Hz), 7.80–7.68 (9H, m), 4.46 (2H, q, J = 7.1 Hz), 1.39 (3H, t, J =7.1 Hz), -2.71 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.8, 146.9 (8C, br), 142.6, 142.24, 142.21, 138.3, 134.8, 134.6 (6C), 131.2 (6C), 130.8, 129.4, 129.0, 127.8 (3C), 126.8, 126.7 (6C), 120.5, 120.4, 118.7, 61.2, 14.4; IR (KBr) 3317, 3055, 3028, 2981, 1724, 1473, 1281, 1223, 972, 798, 737, 702 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.7), 515.5 (4.3), 550.0 (3.9), 645.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₇H₃₅N₄O₂ 687.2760, found 687.2756; Anal. Calcd for C₄₇H₃₄N₄O₂: C, 82.19; H, 4.99; N, 8.16. Found: C, 82.17; H, 5.32; N, 7.90.

5-(*n*-Butyl)-10,20-diphenyl-15-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ba): Prepared from bromoporphyrin H₂-1b (96.8 mg, 0.162 mmol) and arylmagnesium reagent 4a following the general procedure; Purple solid; 71.4 mg, 66% yield; $R_f = 0.35$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 9.47 (2H, d, *J* = 4.9 Hz), 8.94 (2H, d, *J* = 4.9 Hz), 8.93 (1H, s), 8.84 (2H, d, *J* = 4.9 Hz), 8.76 (2H, d, *J* = 4.9 Hz), 8.49 (1H, d, *J* = 8.3 Hz), 8.38 (1H, d, *J* = 7.4 Hz), 8.27–8.19 (4H, m), 7.82 (1H, dd, *J* = 8.3, 7.4 Hz), 7.81–7.69 (6H, m), 4.98 (2H, t, *J* = 7.8 Hz), 4.49 (2H, q, *J* = 7.0 Hz), 2.56 (2H, tt, *J* = 7.8, 7.5 Hz), 1.82 (2H, tq, *J* = 7.5, 7.3 Hz), 1.42 (3H, t, *J* = 7.0 Hz), 1.15 (3H, t, *J* = 7.3 Hz), -2.58 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.8, 146.6 (8C, br), 142.6, 142.5, 138.3, 134.9, 134.5, 131.6, 131.5, 130.4, 129.5, 129.0, 128.1, 127.7, 126.8, 126.6, 121.0, 119.8, 117.9, 61.2, 40.9, 35.3, 23.6, 14.3, 14.1; IR (KBr) 3317, 3101, 3055, 2958, 2924, 2862, 1720, 1473, 1281, 976, 795, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.9), 516.5 (4.5), 551.0 (4.1), 592.5 (3.9), 648.5 (3.9) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₅H₃₉N₄O₂ 667.3073, found 667.3067; Anal. Calcd for C₄₅H₃₈N₄O₂: C, 81.05; H, 5.74; N, 8.40. Found: C, 81.09; H, 5.72; N, 8.44.

5,15-Diphenyl-10-(3-ethoxycarbonylphenyl)-20-pentafluorophenylporphyrin (H₂-3ca): Prepared from bromoporphyrin H₂-1c (114.6 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Purple solid; 85.8 mg, 69% yield; $R_{\rm f} = 0.42$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.93 (2H, d, J = 4.9 Hz), 8.88 (1H, s), 8.84 (2H, d, J = 4.9 Hz), 8.77 (2H, d, J = 4.9 Hz), 8.76 (2H, d, J = 4.9 Hz), 8.48 (1H, d, J = 8.1 Hz), 8.36 (1H, d, J = 7.8 Hz), 8.23–8.17 (4H, m), 7.82 (1H, dd, J = 8.1, 7.8 Hz), 7.80–7.71 (6H, m), 4.45 (2H, q, J = 7.0 Hz), 1.39 (3H, t, J = 7.0 Hz), -2.73 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.7, 146.69 (2C, d, $J_{\rm CF} = 247.5$ Hz), 146.66 (8C, br), 142.3, 141.9 (1C, d, $J_{\rm CF} = 247.5$ Hz), 141.7, 138.2, 137.6 (2C, d, $J_{\rm CF} = 249.1$ Hz), 134.7, 134.6, 132.7, 131.5 (4C), 129.4, 129.2, 129.0, 128.0, 126.8 (5C), 121.1, 120.6, 116.8 (1C, t, $J_{\rm CF} = 20.3$ Hz), 100.5, 61.3, 14.3; ¹⁹F NMR (CDCl₃, 466 MHz) δ –138.3 (2F, ddd, $J_{\rm FF} = 24.9$, 8.5, 6.4 Hz), –154.3 (1F, tt, $J_{\rm FF} = 20.6$, 6.4 Hz), –163.8 (2F, ddd, $J_{\rm FF} = 24.9$, 20.6, 8.6 Hz); IR (KBr) 3321, 3101, 3059, 2985, 2924, 1716, 1508, 1281, 984, 922, 798, 748 cm⁻¹; UV/vis (CHCl₃) $\lambda_{\rm max}$ (log ε) 417.5 (5.6), 513.5 (4.2), 547.0 (3.5), 588.5 (3.6), 642.0 (3.2) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C4₇H₃₀F₅N₄O₂ 777.2289, found 777.2296.

5,15-Diphenyl-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3da): Prepared from bromoporphyrin H₂-1d (87.7 mg, 0.162 mmol) and arylmagnesium reagent 4a following the general procedure; Purple solid; 74.2 mg, 75% yield; $R_f = 0.32$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.23 (1H, s), 9.34 (2H, d, J = 4.8 Hz), 9.02 (2H, d, J = 4.8 Hz), 8.90 (2H, d, J = 4.8 Hz), 8.84 (1H, s), 8.78 (2H, d, J =4.8 Hz), 8.45 (1H, d, J = 7.9 Hz), 8.37 (1H, d, J = 7.5 Hz), 8.28–8.19 (4H, m), 7.85–7.72 (7H, m), 4.40 (2H, q, J = 7.1 Hz), 1.35 (3H, t, J = 7.1 Hz), -3.02 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.8, 146.9 (4C, br), 146.0 (4C, br), 142.8, 141.6, 138.2, 134.7, 131.5 (4C), 131.2, 131.0, 129.0 (2C), 127.8 (3C), 126.8, 126.6, 119.8, 119.0, 105.1, 61.2, 14.3; IR (KBr) 3055, 2924, 1720, 1593, 1477, 1404, 1288, 1080, 968, 849, 795, 733 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 413.0 (5.6), 509.5 (4.2), 543.0 (3.7), 583.0 (3.7), 638.5 (3.2) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₁H₃₁N₄O₂ 611.2447, found 611.2446.

5,15-Di(*p*-tolyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ea): Prepared from bromoporphyrin H₂-1e (92.3 mg, 0.162 mmol) and arylmagnesium reagent 4a following the general procedure; Purple solid; 72.4 mg, 70% yield; $R_f = 0.35$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 10.18 (1H, s), 9.30 (2H, d, J = 4.8 Hz), 9.03 (2H, d, J = 4.8 Hz), 8.92 (2H, d, J = 4.8 Hz), 8.88 (1H, s), 8.77 (2H, d, J = 4.8 Hz), 8.48 (1H, d, J = 8.0 Hz), 8.37 (1H, d, J = 7.6 Hz), 8.11 (4H, d, J = 7.8 Hz), 7.80 (1H, dd, J = 8.0, 7.6 Hz), 7.57 (4H, d, J = 7.8 Hz), 4.45 (2H, q, J = 7.1 Hz), 2.71 (6H, s), 1.38 (3H, t, J = 7.1 Hz), -2.91 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.9, 147.4 (4C, br), 146.1 (br), 145.9 (br), 143.1, 138.9, 138.2, 137.5, 134.7, 134.5, 131.5, 131.3, 131.0 (4C), 129.0, 127.6, 127.3, 126.6, 120.0, 119.8, 104.9, 61.2, 21.5, 14.3; IR (KBr) 3309, 3116, 3024, 2978, 2924, 2870, 1720, 1469, 1404, 1284, 1215, 1103, 968, 795, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 415.0 (5.7), 511.5 (4.3), 546.5 (3.8), 587.0 (3.8), 641.0 (3.4) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₃H₃₅N₄O₂ 639.2760, found 639.2769.

5,15-Bis(3-vinylphenyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-**3fa**): Prepared from bromoporphyrin H₂-**1f** (96.1 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Brown-purple solid; 87.1 mg, 81% yield; $R_{\rm f} = 0.35$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 10.16 (1H, s), 9.29 (2H, d, J = 4.8 Hz), 9.06 (2H, d, J = 4.8 Hz), 9.00 (2H, d, J = 4.8 Hz), 8.99 (1H, s), 8.87 (2H, d, J = 4.8 Hz), 8.54 (1H, d, J = 7.8 Hz), 8.43 (1H, d, J = 7.3 Hz), 8.34 (2H, s), 8.17 (2H, d, J =6.8 Hz), 7.86 (2H, d, J = 7.5 Hz), 7.84 (1H, dd, J = 7.8, 7.3 Hz), 7.74 (2H, dd, J = 7.5, 6.8 Hz), 7.03 (2H, dd, J = 17.6, 10.7 Hz), 6.00 (2H, d, J = 17.6 Hz), 5.43 (2H, d, J = 10.7 Hz), 4.51 (2H, q, J = 7.0 Hz), 1.43 (3H, t, J = 7.0 Hz), -2.83 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.8, 147.2 (4C, br), 146.2 (br), 146.1 (br), 143.0, 142.1, 138.2, 136.9, 136.3, 134.8, 134.2, 132.7, 131.4 (4C), 131.2, 131.0, 129.3, 129.1, 127.0, 126.7, 125.7, 119.63, 119.2, 114.8, 105.1, 61.2, 14.3; IR (KBr) 3309, 3097, 3051, 2981, 2927, 1716, 1589, 1473, 1400, 1281, 1099, 976, 910, 791, 733 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 414.0 (5.8), 509.5 (4.4), 543.0 (3.9), 582.5 (3.9), 636.5 (3.4) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C4₅H₃₅N₄O₂ 663.2760, found 663.2769; Anal. Calcd for C4₅H₃₄N₄O₂: C, 81.55; H, 5.17; N, 8.45. Found: C, 81.73; H, 5.28; N, 8.39.

5,15-Bis[4-{2-(triisopropylsilyl)ethynyl}phenyl]-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ga): Prepared from bromoporphyrin H₂-1g (146.2 mg, 0.162 mmol) and arylmagnesium reagent 4a following the general procedure; Brown-purple solid; 97.7 mg, 62% yield; $R_f = 0.43$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 10.19 (1H, s), 9.31 (2H, d, J = 4.4 Hz), 9.00 (2H, d, J = 4.4 Hz), 8.90 (2H, d, J = 4.9Hz), 8.88 (1H, s), 8.80 (2H, d, J = 4.4 Hz), 8.49 (1H, d, J = 8.3 Hz), 8.37 (1H, d, J = 7.3 Hz), 8.18 (4H, d, J = 7.8 Hz), 7.90 (4H, d, J = 7.8 Hz), 7.82 (1H, dd, J = 8.3, 7.3 Hz), 4.46 (2H, q, J = 7.1 Hz), 1.39 (3H, t, J = 7.1 Hz), 1.35–1.18 (42H, m), –2.94 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.8, 147.0 (4C, br), 146.2 (4C, br), 142.9, 141.9, 138.2, 134.6, 134.4, 131.6, 131.4, 131.3, 130.8, 130.6, 130.3, 129.1, 126.7, 123.3, 119.4, 119.2, 107.3, 105.3, 92.2, 61.2, 18.8, 14.3, 11.6; IR (KBr) 3309, 3109, 3066, 3032, 2943, 2866, 1724, 1466, 1396, 1281, 1242, 1107, 972, 798, 737, 671 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 417.0 (5.7), 512.0 (4.4), 547.0 (4.0), 586.0 (3.9), 640.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₆₃H₇₁N₄O₂Si₂ 971.5116, found 971.5114; Anal. Calcd for C₆₃H₇₀N₄O₂ Si₂: C, 77.89; H, 7.26; N, 5.77. Found: C, 77.54; H, 7.24; N, 5.43.

5,15-Bis(3-methoxyphenyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ha): Prepared from bromoporphyrin H₂-1h (97.4 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Purple solid; 78.1 mg, 72% yield; $R_f = 0.27$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.18 (1H, s), 9.30 (2H, d, J = 4.8 Hz), 9.06 (2H, d, J = 4.8 Hz), 8.97 (2H, d, J = 4.8 Hz), 8.91 (1H, s), 8.80 (2H, d, J = 4.8 Hz), 8.50 (1H, d, J = 7.9 Hz), 8.39 (1H, d, J = 7.7 Hz), 7.90–7.75 (5H, m), 7.66 (2H, dd, J = 8.0, 7.6 Hz), 7.35 (2H, d, J = 8.0 Hz), 4.47 (2H, q, J = 7.1 Hz), 4.00 (6H, s), 1.40 (3H, t, J = 7.1 Hz), -2.91 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.8, 158.3, 147.1 (br), 147.1 (br), 146.2 (br), 146.0 (br), 143.2, 143.0, 138.2, 134.8, 131.5, 131.4, 131.1, 131.0, 129.3, 129.1, 127.8, 127.7, 126.7, 120.9, 119.6, 119.1, 113.8, 105.0, 61.2, 55.6, 14.3; IR (KBr) 3309, 3101, 3059, 2947, 2835, 1716, 1589, 1466, 1277, 1038, 791 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 414.0 (5.8), 509.5 (4.4), 543.0 (3.8), 582.0 (3.9), 636.0 (3.3) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C4₃H₃₅N₄O₄ 671.2658, found 671.2659; Anal. Calcd for C4₃H₃₅N₄O₄: C, 77.00; H, 5.11; N, 8.35. Found: C, 76.91; H, 5.23; N, 8.16.

5,15-Bis[3,5-di(*t*-butyl)phenyl]-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ia): Prepared from bromoporphyrin H₂-1i (124.1 mg, 0.162 mmol) and arylmagnesium reagent 4a following the general procedure; Brown-purple solid; 86.6 mg, 64% yield; $R_{\rm f} = 0.38$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.21 (1H, s), 9.33 (2H, d, J = 4.8 Hz), 9.06 (2H, d, J = 4.8 Hz), 8.95 (2H, d, J = 4.8 Hz), 8.91 (1H, s), 8.80 (2H, d, J = 4.8 Hz), 8.48 (1H, d, J = 8.1 Hz), 8.40 (1H, d, J = 7.7 Hz), 8.13 (4H, d, J = 1.5 Hz), 7.85 (2H, d, J = 1.5 Hz), 7.82 (1H, dd, J = 8.1, 7.7 Hz), 4.46 (2H, q, J = 7.1 Hz), 1.57 (36H, s), 1.39 (3H, t, J = 7.1 Hz), -2.83 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.9, 149.1, 147.6 (br), 147.5 (br), 146.0 (br), 145.8 (br), 143.3, 140.9, 138.2, 134.7, 131.7, 131.2 (4C), 131.0, 130.0, 129.7, 129.0, 126.6, 121.3, 121.2, 118.8, 104.8, 61.2, 35.1, 31.8, 14.3; IR (KBr) 3313, 3066, 2962, 2870, 1720, 1589, 1481, 1365, 1246, 1103, 972, 914, 795, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 416.0 (5.7), 511.5 (4.3), 547.0 (3.9), 586.0 (3.8), 641.5 (3.5) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₅₇H₆₃N₄O₂ 835.4951, found 835.4952; Anal. Calcd for C₅₇H₆₂N₄O₂: C, 81.98; H, 7.48; N, 6.71. Found: C, 81.77; H, 7.51; N, 6.57.

5,15-Bis(2,4,6-trimethylphenyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ja): Prepared from bromoporphyrin H₂-1j (101.3 mg, 0.162 mmol) and arylmagnesium reagent 4a following the general

procedure; Purple solid; 78.9 mg, 70% yield; $R_f = 0.38$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.13 (1H, s), 9.27 (2H, d, J = 4.8 Hz), 8.93 (1H, s), 8.85 (2H, d, J = 4.8 Hz), 8.77 (4H, d, J = 4.8 Hz), 8.49 (1H, d, J = 7.9 Hz), 8.41 (1H, d, J = 7.5 Hz), 7.82 (1H, dd, J = 7.9, 7.5 Hz), 7.33 (4H, s), 4.48 (2H, q, J = 7.0 Hz), 2.66 (6H, s), 1.88 (12H, s), 1.42 (3H, t, J = 7.0 Hz), -2.76 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.9, 146.9 (4C, br), 146.3 (4C, br), 142.9, 139.5, 138.2, 138.1, 137.9, 134.7, 131.6, 131.4, 130.4, 130.0, 129.3, 129.0, 127.9, 126.6, 118.4, 118.1, 104.4, 61.2, 21.6, 21.4, 14.3; IR (KBr) 3309, 3097, 2970, 2920, 2866, 1724, 1570, 1466, 1277, 1111, 972, 798, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 416.0 (5.6), 512.5 (4.3), 547.5 (3.7), 589.0 (3.7), 647.5 (3.3) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₇H₄₃N₄O₂ 695.3386, found 695.3386.

5,15-Bis(pentafluorophenyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H2-3ka): Prepared from bromoporphyrin **H2-1k** (116.9 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Purple solid; 93.6 mg, 73% yield; $R_{\rm f} = 0.43$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.22 (1H, s), 9.37 (2H, d, J = 4.8 Hz), 8.95 (2H, d, J = 4.6 Hz), 8.91 (2H, d, J = 4.8 Hz), 8.90 (1H, s), 8.86 (2H, d, J = 4.8 Hz), 8.53 (1H, d, J = 7.9 Hz), 8.36 (1H, d, J = 7.5 Hz), 7.84 (1H, dd, J = 7.9, 7.5 Hz), 4.48 (2H, q, J = 7.1 Hz), 1.40 (3H, t, J = 7.1 Hz), -2.99 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.6, 147.0 (4C, br), 146.7 (4C, d, $J_{\rm CF} = 252.2$ Hz), 146.4 (4C, br), 142.2 (2C, d, $J_{\rm CF} = 259.0$ Hz), 142.0, 138.2, 137.7 (4C, d, $J_{\rm CF} = 253.5$ Hz), 134.7 , 132.9, 132.8, 130.0, 129.6, 129.5 (3C), 126.9, 120.6, 116.2 (2C, t, $J_{\rm CF} = 19.6$ Hz), 106.1, 102.0, 61.3, 14.3; ¹⁹F NMR (CDCl₃, 466 MHz) δ –138.3 (4F, ddd, $J_{\rm FF} = 24.1$, 8.1, 6.2 Hz), -153.8 (2F, tt, $J_{\rm FF} = 21.4$, 6.2 Hz), -163.4 (4F, ddd, $J_{\rm FF} = 24.1$, 21.4, 8.0 Hz); IR (KBr) 3317, 3105, 3043, 2981, 2935, 1720, 1512, 1281, 987, 914, 733 cm⁻¹; UV/vis (CHCl₃) $\lambda_{\rm max}$ (log ε) 409.5 (5.7), 504.5 (4.4), 536.0 (3.9), 579.0 (3.9), 634.0 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₁H₂₁F₁₀N₄O₂ 791.1505, found 791.1501. Anal. Calcd for C₄₁H₂₀F₁₀N₄O₂: C, 62.29; H, 2.55; N, 7.09. Found: C, 62.38; H, 2.84; N, 6.82.

5,15-Di(2-thienyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H2-3la): Prepared from bromoporphyrin H2-11 (89.7 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Purple solid; 77.8 mg, 77% yield; $R_f = 0.30$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.15 (1H, s), 9.29 (2H, d, J = 4.8 Hz), 9.21 (2H, d, J = 4.8 Hz), 9.11 (2H, d, J = 4.8 Hz), 8.88 (1H, s), 8.78 (2H, d, J = 4.8 Hz), 8.49 (1H, d, J = 7.9 Hz), 8.36 (1H, d, J = 7.5 Hz), 7.93 (2H, dd, J = 3.5, 1.3 Hz), 7.86 (2H, dd, J = 5.1, 1.3 Hz), 7.82 (1H, dd, J = 7.9, 7.5 Hz), 7.51 (2H, dd, J = 5.1, 3.5 Hz), 4.46 (2H, q, J = 7.1 Hz), 1.39 (3H, t, J = 7.1 Hz), -2.83 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.8, 148.1 (br), 148.0 (br), 146.4 (br), 146.2 (br), 142.8, 142.5, 138.2, 134.8, 133.9, 131.6, 131.43, 131.38, 130.9, 129.4, 129.1, 128.0, 126.7, 126.2, 119.9, 111.5, 105.7, 61.2, 14.3; IR (KBr) 3305, 3105, 3074, 2962, 2927, 2866, 1712, 1589, 1365, 1288, 1103, 949, 795, 702 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 417.5 (5.7), 512.5 (4.3), 550.5 (4.0), 587.5 (3.9), 643.0 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₃₇H₂₇N₄O₂S₂ 623.1575, found 623.1579.

5,15-Di(*i*-butyl)-10-(3-ethoxycarbonylphenyl)porphyrin (H₂-3ma): Prepared from bromoporphyrin H₂-1m (81.2 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Brown-purple solid; 63.8 mg, 69% yield; $R_f = 0.33$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.05 (1H, s), 9.51 (2H, d, J = 4.8 Hz), 9.41 (2H, d, J = 4.8 Hz), 9.32 (2H, d, J = 4.8 Hz), 8.90 (1H, s), 8.80 (2H, d, J = 4.8 Hz), 8.52 (1H, d, J = 7.9 Hz), 8.36 (1H, d, J = 7.5 Hz), 7.82 (1H, dd, J = 7.9, 7.5 Hz), 4.83 (4H, d, J = 7.2 Hz), 4.49 (2H, q, J = 7.0 Hz), 2.92–2.69 (2H, m), 1.41 (3H, t, J = 7.0 Hz), 1.22 (12H, d, J = 6.4 Hz), -2.74 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.9, 148.1, 147.9, 145.0, 144.9, 143.6, 138.1, 134.6, 131.6, 131.5, 129.1, 128.9, 128.7, 128.1, 126.5, 118.4, 118.0, 104.2, 61.2, 43.4, 36.7, 23.3, 14.3; IR (KBr) 3290, 3116, 2954, 2870, 1716, 1581, 1466, 1281, 1107, 787 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 412.5 (5.6), 511.5 (4.3), 544.5 (3.8), 586.0 (3.8), 642.0 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₃₇H₃₉N₄O₂ 571.3073, found 571.3073.

2-(3-Ethoxycarbonylphenyl)-5,10,15,20-tetraphenylporphyrin (H2-3na): Prepared from bromoporphyrin H2-1n (112.4 mg, 0.162 mmol) and arylmagnesium reagent **4a** following the general procedure; Purple solid; 106.3 mg, 86% yield; $R_{\rm f} = 0.32$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.95 (1H, d, J = 4.9 Hz), 8.94 (2H, d, J = 4.9 Hz), 8.92 (1H, d, J = 4.9 Hz), 8.88 (1H, s), 8.86 (1H, d, J = 4.9 Hz), 8.81 (1H, d, J = 4.9 Hz), 8.36–8.27 (6H, m), 8.07 (1H, s), 8.03–7.95 (2H, m), 7.90 (1H, d, J = 8.3 Hz), 7.86–7.76 (9H, m), 7.67 (1H, d, J = 7.8 Hz), 7.36–7.22 (4H, m), 4.47 (2H, q, J = 7.2 Hz), 1.48 (3H, t, J = 7.2 Hz), -2.49 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.6, 148.1 (br), 147.3 (2C, br), 146.9 (3C, br), 144.8, 144.1 (br), 143.4 (br), 142.6, 142.4, 142.0, 140.6, 139.1, 135.7 (2C), 134.6 (2C), 134.6 (2C), 134.6 (2C), 134.3, 134.3, 131.8, 131.6 (2C), 131.4, 131.0 (2C), 130.8, 129.7, 127.8 (3C), 127.5, 127.4, 127.1, 126.8 (2C), 126.7 (4C), 126.1 (2C), 121.2, 120.5, 120.2, 120.0, 60.8, 14.4; IR (KBr) 3321, 3055, 3028, 2981, 2931, 2908, 1716, 1597, 1473, 1281, 1107, 987, 798, 725 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 422.0 (5.8), 518.5 (4.4), 553.5 (3.9), 593.0 (3.7), 650.5 (3.4) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for Cs₃H₃₉N₄O₂ 763.3073, found 763.3067.

5,15-Diphenyl-10,20-di(3-ethoxycarbonylphenyl)porphyrin (H₂-**3**oa): The general procedure with dibromoporphyrin H₂-**1o** (100.5 mg, 0.162 mmol) and 20 equiv instead of 10 equiv of arylmagnesium reagent **4a** (16 mL of its ca. 0.2 M solution in THF, ca. 3.2 mmol) gave the title compound as a purple solid (80.0 mg, 65% yield); $R_f = 0.22$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.96 (2H, s), 8.91 (4H, d, J = 4.9 Hz), 8.83 (4H, d, J = 4.9 Hz), 8.52 (2H, d, J = 7.8 Hz), 8.43 (2H, d, J = 7.3 Hz), 8.30– 8.21 (4H, m), 7.84 (2H, dd, J = 7.8, 7.3 Hz), 7.81–7.71 (6H, m), 4.50 (4H, q, J = 7.2 Hz), 1.43 (6H, t, J = 7.2 Hz), -2.61 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.8, 146.8 (8C, br), 142.6, 142.2, 138.3, 134.9, 134.6, 131.5, 130.9, 129.5, 129.1, 127.8, 126.8, 126.7, 120.6, 119.0, 61.2, 14.3; IR (KBr) 3317, 3097, 3059, 3032, 2985, 2931, 2908, 1712, 1435, 1358, 1281, 1107, 972, 802, 744 cm⁻¹; UV/vis (CHCl₃)

 λ_{max} (log ε) 419.0 (5.7), 505.0 (4.2), 549.0 (3.6), 588.5 (3.4), 644.5 (2.9) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₅₀H₃₉N₄O₄ 759.2971, found 759.2968.

5-(2-Ethoxycarbonylphenyl)-10,15,20-triphenylporphyrin (H2-3ab): Prepared from bromoporphyrin **H2-1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4b** following the general procedure; Purple solid; 69.1 mg, 62% yield; $R_f = 0.32$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.86 (2H, d, J = 4.9 Hz), 8.85 (2H, d, J = 4.9 Hz), 8.83 (2H, d, J = 4.9 Hz), 8.70 (2H, d, J = 4.9 Hz), 8.41 (1H, d, J = 7.8 Hz), 8.28–8.20 (6H, m), 8.18 (1H, d, J = 7.3 Hz), 7.87 (1H, dd, J = 7.8, 7.6 Hz), 7.81 (1H, dd, J = 7.6, 7.3 Hz), 7.78–7.71 (9H, m), 3.21 (2H, q, J = 7.0 Hz), -0.42 (3H, t, J = 7.0 Hz), -2.56 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 167.5, 146.8 (8C, br), 142.5, 142.44, 142.35, 135.8, 135.1, 134.60, 134.57, 131.2, 131.0 (4C), 130.4, 129.6, 129.5, 128.3, 127.7 (3C), 126.7 (6C), 120.20, 120.16, 119.3, 60.0, 12.4; IR (KBr) 3317, 3059, 3028, 2981, 2924, 1720, 1469, 1254, 1080, 968, 798, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.7), 516.0 (4.3), 550.5 (3.7), 591.0 (3.6), 646.5 (3.2) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₇H₃₅N₄O₂ 687.2760, found 687.2757; Anal. Calcd for C₄₇H₃₄N₄O₂: C, 82.19; H, 4.99; N, 8.16. Found: C, 82.38; H, 5.07; N, 8.03.

5-(4-Ethoxycarbonylphenyl)-10,15,20-triphenylporphyrin (H₂-**3ac**): Prepared from bromoporphyrin H₂-**1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4c** following the general procedure; Purple solid; 90.2 mg, 81% yield; $R_f = 0.32$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.86 (6H, d, J =4.9 Hz), 8.80 (2H, d, J = 4.9 Hz), 8.45 (2H, d, J = 8.2 Hz), 8.31 (2H, d, J = 8.2 Hz), 8.25–8.17 (6H, m), 7.83–7.66 (9H, m), 4.59 (2H, q, J = 7.1 Hz), 1.56 (3H, t, J = 7.1 Hz), -2.67 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 166.8, 147.1, 146.9 (8C, br), 142.2 (3C), 134.6 (8C), 131.4, 131.2 (4C), 130.7, 130.2, 127.9, 127.8 (3C), 126.7 (6C), 120.6, 120.4, 118.7, 61.3, 14.5; IR (KBr) 3317, 3105, 3055, 3028, 2978, 2927, 2858, 1720, 1604, 1481, 1358, 1273, 1180, 1107, 964, 802, 706 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.7), 515.5 (4.3), 551.5 (4.0), 590.5 (3.8), 645.0 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₇H₃₅N₄O₂ 687.2760, found 687.2756; Anal. Calcd for C₄₇H₃₄N₄O₂: C, 82.19; H, 4.99; N, 8.16. Found: C, 82.01; H, 5.25; N, 7.91.

5-(3-Cyanophenyl)-10,15,20-triphenylporphyrin (H₂-3ad): Prepared from bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent 4d following the general procedure; Purple solid; 75.6 mg, 73% yield; $R_f = 0.35$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.90 (2H, d, J = 4.8 Hz), 8.86 (4H, d, J = 4.8 Hz), 8.71 (2H, d, J = 4.8 Hz), 8.54 (1H, s), 8.45 (1H, d, J = 7.9 Hz), 8.27–8.16 (6H, m), 8.07 (1H, d, J = 7.7 Hz), 7.84 (1H, dd, J = 7.9, 7.7 Hz), 7.80–7.70 (9H, m), –2.68 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 147.0 (8C, br), 143.9, 142.2, 142.1, 138.2, 137.1, 134.6 (6C), 131.7, 131.5, 131.3 (4C), 130.2, 127.9 (3C), 127.6, 126.8, 126.7, 121.0, 120.7, 118.8, 116.6, 111.5; IR (KBr) 3317, 3101, 3059, 3028, 2233, 1562, 1473, 1349, 1184, 972, 798, 729, 702 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε)

419.0 (5.7), 515.0 (4.3), 550.0 (3.9), 591.0 (3.8), 647.5 (3.6) nm; HRMS-FAB⁺ ($[M+H]^+$) calcd for C₄₅H₃₀N₅ 640.2501, found 640.2504.

5-(4-Chlorophenyl)-10,15,20-triphenylporphyrin (H₂-3ae): Prepared from bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4e** following the general procedure; Purple solid; 84.0 mg, 80% yield; $R_f = 0.65$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.88 (4H, d, J = 4.8 Hz), 8.86 (2H, d, J = 4.8 Hz), 8.83 (2H, d, J = 4.8 Hz), 8.28–8.19 (6H, m), 8.15 (2H, d, J = 8.2 Hz), 7.82–7.67 (11H, m), -2.66 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 146.9 (8C, br), 142.3 (3C), 140.9, 135.6 (6C), 134.6, 134.3, 131.2 (6C), 130.7, 127.8 (3C), 127.0, 126.7 (6C), 120.5, 120.4, 118.4; IR (KBr) 3317, 3055, 1558, 1473, 1350, 1180, 1088, 968, 798, 729 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.7), 515.0 (4.4), 550.0 (4.1), 591.0 (4.0), 645.5 (3.9) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₄H₃₀ClN₄ 649.2159, found 649.2159; Anal. Calcd for C₄₄H₂₉ClN₄: C, 81.41; H, 4.50; N, 8.63. Found: C, 81.78; H, 4.70; N, 8.70.

5-(3-Bromophenyl)-10,15,20-triphenylporphyrin (H₂-3af): Prepared from bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4f** following the general procedure; Purple solid; 76.7 mg, 68% yield; $R_f = 0.62$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.89 (2H, d, J = 4.9 Hz), 8.86 (4H, d, J = 4.9 Hz), 8.85 (2H, d, J = 4.9 Hz), 8.41 (1H, s), 8.28–8.19 (6H, m), 8.17 (1H, d, J = 7.5 Hz), 7.92 (1H, d, J = 8.2 Hz), 7.82–7.70 (9H, m), 7.61 (1H, dd, J = 8.2, 7.5 Hz), -2.77 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 146.9 (8C, br), 144.5, 142.3, 142.2, 137.3, 134.6 (6C), 133.2, 131.4, 131.2 (4C), 131.0, 130.7, 128.1, 127.8 (3C), 126.7 (6C), 121.2, 120.6, 120.4, 118.0; IR (KBr) 3317, 3101, 3055, 3028, 1585, 1558, 1469, 1350, 972, 798, 702 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.8), 515.5 (4.4), 550.5 (4.0), 590.0 (3.9), 645.5 (3.8) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₄H₃₀BrN₄ 693.1654, found 693.1645.

5-[3,5-Bis(trifluoromethyl)phenyl]-10,15,20-triphenylporphyrin (H₂-3ag): Prepared from bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent 4g following the general procedure; Purple solid; 87.6 mg, 72% yield; $R_f = 0.70$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.98 (2H, d, J = 4.9 Hz), 8.92 (2H, d, J = 4.9 Hz), 8.91 (2H, d, J = 4.9 Hz), 8.78 (2H, s), 8.75 (2H, d, J = 4.9 Hz), 8.41 (1H, s), 8.30–8.22 (6H, m), 7.84–7.74 (9H, m), -2.62 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 147.1 (8C, br), 144.8, 142.11, 142.05, 134.6 (6C), 133.9, 132.1, 131.6, 131.5, 130.5 (2C, q, $J_{CF} = 33.7$ Hz), 129.9, 128.0, 127.9, 126.81, 126.78, 123.8 (2C, q, $J_{CF} = 273.1$ Hz), 121.9–121.6 (1C, m), 121.3, 120.9, 115.5; ¹⁹F NMR (CDCl₃, 466 MHz) δ –63.9 (6F, s); IR (KBr) 3317, 3051, 1381, 1346, 1273, 1176, 1138, 968, 910, 798, 717 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.7), 515.0 (4.4), 551.0 (4.0), 590.5 (3.9), 645.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₆H₂₉F₆N₄ 751.2296, found 751.2302.

5-(2-Methoxyphenyl)-10,15,20-triphenylporphyrin (H_2 -3ah): Prepared from bromoporphyrin H_2 -1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent 4h following the general procedure; Purple solid;

66.8 mg, 64% yield; $R_f = 0.48$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.81 (6H, d, J = 4.9 Hz), 8.78 (2H, d, J = 4.9 Hz), 8.28–8.13 (6H, m), 8.01 (1H, d, J = 7.3 Hz), 7.81–7.62 (10H, m), 7.39–7.29 (2H, m), 3.58 (3H, s), -2.64 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 159.7, 146.8 (8C, br), 142.5, 142.4, 135.7, 134.6, 131.4, 131.1, 130.9 (6C), 129.9, 127.7 (3C), 126.6 (6C), 120.2, 119.9, 119.6, 116.2, 111.3, 56.0; IR (KBr) 3321, 3132, 3055, 3024, 2939, 1593, 1473, 1250, 972, 798, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.7), 515.0 (4.3), 549.5 (3.7), 589.0 (3.6), 644.5 (3.1) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₅H₃₃N₄O 645.2654, found 645.2656.

5-(4-*N*,*N***-Dimethylaminophenyl)-10,15,20-triphenylporphyrin** (H₂-**3ai**): Prepared from bromoporphyrin H₂-**1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4i** following the general procedure; Purple solid; 84.2 mg, 79% yield; $R_f = 0.24$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.98 (2H, d, J = 4.8 Hz), 8.86 (4H, d, J = 4.6 Hz), 8.85 (2H, d, J = 4.8 Hz), 8.29–8.19 (6H, m), 8.10 (2H, d, J = 8.6 Hz), 7.81–7.70 (9H, m), 7.10 (2H, d, J = 8.6 Hz), 3.22 (6H, s), -2.58 (2H, br s); ¹³C NMR (CDCl₃, 00 MHz) δ 150.2, 146.8 (8C, br), 142.54, 142.46, 135.7, 134.6 (6C), 131.5, 131.0, 130.8 (4C), 130.4, 127.6 (3C), 126.6 (6), 121.5, 120.0, 119.6, 110.8, 40.7; IR (KBr) 3317, 3028, 2897, 2854, 2800, 1604, 1516, 1473, 1354, 1188, 968, 802, 741 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 420.0 (5.5), 519.5 (4.3), 558.5 (4.1), 596.0 (3.9), 651.0 (3.9) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₆H₃₆N₅ 658.2971, found 658.2971; Anal. Calcd for C₄₆H₃₅N₅: C, 83.99; H, 5.36; N, 10.65. Found: C, 83.98; H, 5.35; N, 10.56.

5-(4-Trimethylsilylphenyl)-10,15,20-triphenylporphyrin (H₂-**3a**j): Prepared from bromoporphyrin H₂-**1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4j** following the general procedure; Purple solid; 90.2 mg, 81% yield; $R_f = 0.75$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.91 (2H, d, J =4.9 Hz), 8.87 (4H, d, J = 4.9 Hz), 8.87 (2H, d, J = 4.9 Hz), 8.29–8.20 (8H, m), 7.92 (2H, d, J = 7.8 Hz), 7.82–7.69 (9H, m), 0.55 (9H, s), -2.62 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 146.9 (8C, br), 142.7, 142.41 (3C), 139.7, 134.61 (6C), 134.1, 131.7, 131.1 (8C), 127.7 (3C), 126.7 (6C), 120.4, 120.2 (3C), – 0.81; IR (KBr) 3321, 3059, 3024, 2954, 1597, 1473, 1350, 1250, 972, 841, 798, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.8), 516.0 (4.4), 551.0 (4.0), 590.5 (3.8), 646.0 (3.7) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₇H₃₉N₄Si 687.2944, found 687.2946; Anal. Calcd for C₄₇H₃₈N₄Si: C, 82.18; H, 5.58; N, 8.16. Found: C, 82.47; H, 5.81; N, 7.86.

5-(3-Pyridyl)-10,15,20-triphenylporphyrin (H₂-3ak): Prepared from bromoporphyrin **H₂-1a** (100.0 mg, 0.162 mmol) and heteroarylmagnesium reagent **4k** following the general procedure; Purple solid; 50.9 mg, 51% yield; $R_{\rm f} = 0.28$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 9.50 (1H, s), 9.04 (1H, d, J = 3.8 Hz), 8.91 (2H, d, J = 4.8 Hz), 8.88 (2H, d, J = 4.8 Hz), 8.87 (2H, d, J = 4.8 Hz), 8.80 (2H, d, J = 4.8 Hz), 8.51 (1H, d, J = 7.7 Hz), 8.28–8.18 (6H, m), 7.83–7.66 (10H, m), -2.65 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 153.8, 149.1, 147.0 (8C, br), 142.18, 142.16, 140.8, 138.3, 134.6 (6C), 131.7, 131.3 (4C), 130.4, 127.8 (3C), 126.7 (6C), 121.9, 120.8, 120.6, 115.5; IR (KBr), 3317, 3055, 3028, 1558, 1473, 1404,

1350, 1184, 972, 798, 729, 706 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.8), 515.5 (4.4), 550.0 (4.0), 590.5 (3.9), 645.5 (3.7) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₃H₃₀N₅ 616.2501, found 616.2501.

5-(3-Thienyl)-10,15,20-triphenylporphyrin (H₂-3al): Prepared from bromoporphyrin **H₂-1a** (100.0 mg, 0.162 mmol) and heteroarylmagnesium reagent **4l** following the general procedure; Purple solid; 55.4 mg, 55% yield; $R_f = 0.40$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.98 (2H, d, J = 4.9 Hz), 8.86 (2H, d, J = 4.9 Hz), 8.85 (4H, d, J = 4.9 Hz), 8.25–8.20 (6H, m), 8.02 (1H, dd, J = 2.9, 1.0 Hz), 8.00 (1H, ddd, J = 4.9, 2.9, 1.0 Hz), 7.81–7.72 (9H, m), 7.70 (1H, ddd, J = 4.9, 2.9, 1.0 Hz), -2.65 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 146.9 (8C, br), 142.5, 142.4 (3C), 134.6 (6C), 131.1 (6C), 130.8, 128.3, 127.8 (3C), 126.7 (6C), 123.4 (2C), 120.4, 120.3, 114.5; IR (KBr) 3321, 3101, 3055, 3028, 1473, 1346, 1176, 972, 798, 741, 702 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.4), 516.0 (4.3), 552.5 (4.0), 590.5 (3.8), 647.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₂H₂₉N₄S 621.2113, found 621.2114.

(*Z*)-5-(2-Ethoxycarbonyl)vinyl-10,15,20-triphenylporphyrin (H₂-3am): The general procedure with bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and 40 equiv instead of 10 equiv of alkenykmagnesium reagent 4m (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (56.7 mg, 55%); $R_f = 0.34$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 10.33 (1H, d, J = 15.6 Hz), 9.55 (2H, d, J = 4.9 Hz), 8.99 (2H, d, J = 4.9 Hz), 8.87 (2H, d, J = 4.9 Hz), 8.85 (2H, d, J = 4.9 Hz), 8.31–8.21 (6H, m), 7.86–7.72 (9H, m), 6.92 (1H, d, J = 15.6 Hz), 4.60 (2H, q, J = 7.0 Hz), 1.59 (3H, t, J = 7.0 Hz), -2.36 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 166.3, 146.8 (8C, br), 145.7, 142.1, 141.9, 134.54, 134.49, 132.1, 131.4, 131.3 (4C), 128.8, 127.9 (3C), 126.8 (6C), 121.8, 121.1, 112.3, 60.9, 14.5; IR (KBr) 3317, 3055, 2981, 2904, 1712, 1620, 1469, 1362, 1273, 1165, 972, 725 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 425.0 (5.6), 522.0 (4.1), 565.5 (4.1), 657.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₃H₃₃N₄O₂ 637.2604, found 637.2599; Anal. Calcd for C₄₃H₃₂N₄O₂: C, 81.11; H, 5.07; N, 8.80. Found: C, 81.17; H, 5.19; N, 8.58.

(*Z*)-5-(2-Cyano)vinyl-10,15,20-triphenylporphyrin (H₂-3an): The general procedure with bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and 40 equiv instead of 10 equiv of alkenykmagnesium reagent 4n (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (49.7 mg, 52%); $R_f = 0.37$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 9.64 (1H, d, J = 11.2 Hz), 9.24 (2H, d, J = 4.9 Hz), 8.93 (2H, d, J = 4.9 Hz), 8.80 (2H, d, J = 4.9 Hz), 8.78 (2H, d, J = 4.9 Hz), 8.28–8.12 (6H, m), 7.84–7.67 (9H, m), 6.59 (1H, d, J = 11.2 Hz), -2.55 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 151.2, 146.9 (8C, br), 142.0 (3C), 134.6, 134.5, 132.4, 131.5, 131.2, 128.6, 127.9 (3C), 126.8, 126.7, 122.1, 121.1, 116.5, 109.6, 106.5; IR (KBr) 3321, 3055, 3032, 2954, 2920, 2222, 1601, 1558, 1469, 1354, 1157, 1072, 972, 798, 706 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 423.0 (5.7), 519.5 (4.4), 558.5 (4.2), 594.5 (4.0), 652.5 (3.5) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C4₁H₂₈N₅ 590.2345, found 590.2350. (*E*)-5-(2-Cyano)vinyl-10,15,20-triphenylporphyrin (H₂-3ao): The general procedure with bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and 40 equiv, instead of 10 equiv, of alkenykmagnesium reagent 4o (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (58.3 mg, 61%); $R_f = 0.37$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 9.92 (1H, d, J = 16.1Hz), 9.32 (2H, d, J = 4.9 Hz), 8.92 (2H, d, J = 4.9 Hz), 8.79 (2H, d, J = 4.9 Hz), 8.76 (2H, d, J = 4.9 Hz), 8.22–8.12 (6H, m), 7.84–7.68 (9H, m), 6.20 (1H, d, J = 16.1 Hz), -2.46 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 151.8, 146.8 (8C, br), 141.9, 141.7, 134.51, 134.47, 132.6, 131.6, 131.4, 128.1, 128.03, 127.95, 126.9, 126.8, 122.6, 121.6, 118.2, 110.4, 107.8; IR (KBr) 3317, 3105, 3059, 3028, 2214, 1604, 1477, 1354, 1165, 968, 798, 733 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 427.5 (5.5), 525.5 (4.1), 566.5 (4.1), 659.0 (3.8) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₁H₂₈N₅ 590.2345, found 590.2350; Anal. Calcd for C₄₁H₂₇N₅: C, 83.51; H, 4.61; N, 11.88. Found: C, 83.73; H, 4.72; N, 12.02.

(*E*)-5-(2-*N*,*N*-Dimethylaminocarbonyl)vinyl-10,15,20-triphenylporphyrin (H₂-3ap): The general procedure with bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and 40 equiv instead of 10 equiv of alkenykmagnesium reagent 4p (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (57.8 mg, 56%); $R_f = 0.24$ (1:1 hexane/toluene); ¹H NMR (THF-*d*₈, 400 MHz) δ 10.08 (1H, d, *J* = 15.1 Hz), 9.48 (2H, d, *J* = 4.9 Hz), 8.85 (2H, d, *J* = 4.9 Hz), 8.76 (2H, d, *J* = 4.9 Hz), 8.75 (2H, d, *J* = 4.9 Hz), 8.23–8.08 (6H, m), 7.79–7.63 (9H, m), 7.26 (1H, d, *J* = 15.1 Hz), 3.18 (6H, br s), -2.45 (2H, br s); ¹³C NMR (THF-*d*₈, 100 MHz) δ 165.0, 146.7 (8C, br), 142.4, 142.2, 141.8, 134.37, 134.35, 132.9, 131.4, 130.8 (4C), 128.9, 127.7 (3C), 126.6 (6C), 120.8, 120.5, 114.7, 36.3 (br), 34.9 (br); IR (KBr) 3317, 3105, 3055, 2920, 2858, 1647, 1608, 1473, 1396, 1350, 1142, 972, 802, 741 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 425.5 (5.6), 524.5 (4.2), 563.0 (4.2), 657.0 (3.8) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₃H₃₄N₅O 636.2763, found 636.2766.

(*E*)-5-(2-Trimethylsilyl)vinyl-10,15,20-triphenylporphyrin (H₂-3aq): The general procedure with bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and 40 equiv instead of 10 equiv of alkenykmagnesium reagent 4q (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (76.2 mg, 74%); $R_f = 0.68$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 9.49 (1H, d, J = 18.9 Hz), 9.47 (2H, d, J = 4.8 Hz), 8.94 (2H, d, J = 4.8 Hz), 8.84 (4H, d, J = 4.8 Hz), 8.30–8.20 (6H, m), 7.84–7.71 (9H, m), 6.97 (1H, d, J = 18.9 Hz), 0.60 (9H, s), -2.48 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 148.3, 146.7 (8C, br), 144.2, 142.5, 142.3, 134.6, 134.5, 131.2, 131.1, 131.0, 129.4, 127.7 (3C), 126.7 (6C), 120.24, 120.17, 119.6, -0.9; IR (KBr) 3317, 3055, 2951, 2924, 2854, 1593, 1469, 1250, 976, 864, 795, 729 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 421.5 (5.6), 520.5 (4.2), 558.5 (4.0), 598.0 (3.8), 654.0 (3.7) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₃H₃₇N₄Si 637.2787, found 637.2797; Anal. Calcd for C₄₃H₃₆N₄Si: C, 81.09; H, 5.70; N, 8.80. Found: C, 80.76; H, 5.66; N, 8.62.

5-(1-Trimethylsilyl)vinyl-10,15,20-triphenylporphyrin (H₂-**3ar**): The general procedure with bromoporphyrin H₂-**1a** (100.0 mg, 0.162 mmol) and 40 equiv, instead of 10 equiv, of alkenykmagnesium reagent **4r** (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (87.8 mg, 85%); $R_f = 0.70$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 9.22 (2H, d, J = 4.8 Hz), 8.87 (2H, d, J = 4.8 Hz), 8.83 (2H, d, J = 4.8 Hz), 8.81 (2H, d, J = 4.8 Hz), 8.87-8.06 (6H, m), 7.84–7.64 (9H, m), 6.84 (1H, d, J = 3.3 Hz), 6.69 (1H, d, J = 3.3 Hz), 0.18 (9H, s), -2.70 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 155.8, 146.7 (6C), 145.0, 142.49, 142.45, 134.6 (6C), 132.4, 131.2, 130.9, 130.7, 130.6, 127.7 (3C), 126.7 (6C), 124.1, 119.9, 119.3, -0.7; IR (KBr) 3317, 3055, 2954, 2897, 1469, 1346, 1250, 972, 845, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.0 (5.7), 517.5 (4.3), 553.0 (4.0), 593.5 (3.7), 648.5 (3.7) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₃H₃₇N₄Si 637.2787, found 637.2789; Anal. Calcd for C₄₃H₃₆N₄Si: C, 81.09; H, 5.70; N, 8.80. Found: C, 81.01; H, 5.66; N, 8.80.

5-(2-Bromocyclopent-1-enyl)-10,15,20-triphenylporphyrin (H2-3as): The general procedure with bromoporphyrin H2-1a (100.0 mg, 0.162 mmol) and 40 equiv instead of 10 equiv of alkenylmagnesium reagent 4s (32 mL of its ca. 0.2 M solution in THF, ca. 6.4 mmol) gave the title compound as a purple solid (72.0 mg, 65%); $R_f = 0.68$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 9.21 (2H, d, J = 4.9 Hz), 8.96 (2H, d, J = 4.9 Hz), 8.86 (2H, d, J = 4.9 Hz), 8.85 (2H, d, J = 4.9 Hz), 8.32–8.19 (6H, m), 7.83–7.70 (9H, m), 3.62 (2H, t, J = 7.3 Hz), 3.48 (2H, t, J = 7.6 Hz), 2.68 (2H, tt, J = 7.6, 7.3 Hz), -2.66 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 146.9 (8C, br), 143.2, 142.4, 142.2, 134.6, 134.5, 131.8, 131.2, 131.0, 129.4, 127.7 (3C), 126.7, 126.6, 125.5, 120.6, 112.0, 114.6, 45.1, 41.8, 23.9; IR (KBr) 3321, 3093, 3055, 3024, 2943, 2850, 1558, 1473, 1350, 1184, 972, 798, 733 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.7), 514.5 (4.3), 549.5 (3.9), 590.5 (3.8), 645.5 (3.6) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C4₃H₃₁BrN₄: C, 75.55; H, 4.57; N, 8.20. Found: C, 75.44; H, 4.56; N, 8.03.

5-[4-{(Tetrahydro-2H-pyran-2-yl)oxy}phenyl]-10,15,20-triphenylporphyrin (H2-3at): The general procedure with bromoporphyrin H2-1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent 4t, followed by demetalation with 5 mL of a 0.1 M methanol solution of citric acid, instead of a 1.2 M aqueous MeOH solution of HCl (a 1:9 mixture of conc. HCl and MeOH), gave the title compound as a purple solid (90.2 mg, 78%); $R_f = 0.42$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 400 MHz) δ 8.93 (2H, d, J = 4.9 Hz), 8.86 (6H, d, J = 4.9 Hz), 8.28–8.20 (6H, m), 8.14 (2H, d, J = 8.3 Hz), 7.81–7.70 (9H, m), 7.45 (2H, d, J = 8.3 Hz), 5.73 (1H, t, J = 3.4 Hz), 4.20 (1H, td, J = 10.4, 3.8 Hz), 3.85–3.77 (1H, m), 2.26–2.12 (1H, m), 2.11–2.02 (2H, m), 1.88–1.69 (3H, m), –2.67 (2H, br s); ¹³C NMR (CDCl₃, 100 MHz) δ 157.2, 146.9 (8C, br), 142.4, 135.6 (3C), 135.5, 134.6 (6C), 131.0 (8C, br), 127.7 (3C), 126.7 (6C), 120.2, 120.1, 120.0, 114.8, 97.0, 62.5, 30.7, 25.4, 19.1; IR (KBr) 3321, 3101, 3055, 3028, 2943, 2873, 1601, 1470, 1350, 1230, 964, 798, 741 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.5 (5.8), 516.0 (4.3), 552.0 (3.9), 591.5 (3.6), 647.0 (3.5)

nm; HRMS-FAB⁺ ($[M+H]^+$) calcd for C₄₉H₃₉N₄O₂ 715.3073, found 715.3080; Anal. Calcd for C₄₉H₃₈N₄O₂: C, 82.33; H, 5.36; N, 7.84. Found: C, 82.28; H, 5.42; N, 7.74.

5-(4-Hydroxyphenyl)-10,15,20-triphenylporphyrin (H₂-7): Prepared from bromoporphyrin H₂-1a (100.0 mg, 0.162 mmol) and arylmagnesium reagent 4t following the general procedure; Purple solid; 74.6 mg, 73% yield; $R_f = 0.22$ (1:1 hexane/toluene); ¹H NMR (THF- d_8 , 400 MHz) δ 8.92 (2H, d, J = 4.9 Hz), 8.81 (6H, d, J = 4.9 Hz), 8.58 (1H, br s), 8.24-8.15 (6H, m), 8.03 (2H, d, J = 8.3 Hz), 7.75-7.65 (9H, m), 7.17 (2H, d, J = 8.3 Hz), -2.58 (2H, br s); ¹³C NMR (THF- d_8 , 100 MHz) δ 158.8, 147.7 (8C, br), 143.5, 143.4, 136.4, 135.3 (6C), 134.1, 131.9, 131.5 (6C, br), 128.5 (3C), 127.5 (6C), 121.6, 120.9, 120.7, 114.6; IR (KBr) 3317, 3051, 1701, 1601, 1562, 1473, 1350, 1211, 968, 798, 702 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 420.0 (5.6), 516.5 (4.4), 552.5 (3.9), 592.0 (3.6), 647.5 (3.5) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₄H₃₁N₄O 631.2498, found 631.2496; Anal. Calcd for C₄₄H₃₀N₄O: C, 83.79; H, 4.79; N, 8.88. Found: C, 83.62; H, 5.00; N, 8.71.

5-[4-(1,3-Dioxolan-2-yl)phenyl]-10,15,20-triphenylporphyrin (H₂-3au): The general procedure with bromoporphyrin **H₂-1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **2u**, followed by demetalation with 5 mL of a 0.1 M MeOH solution of citric acid, instead of a 1.2 M aqueous MeOH solution of HCl (a 1:9 mixture of conc. HCl and MeOH), gave the title compound as a purple solid (84.8 mg, 76%); $R_f = 0.43$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 8.87 (8H, d, J = 4.8 Hz), 8.26 (2H, d, J = 8.2 Hz), 8.24–8.19 (6H, m), 7.88 (2H, d, J = 8.2 Hz), 7.82–7.67 (9H, m), 6.15 (1H, s), 4.35 (2H, t, J = 7.0 Hz), 4.21 (2H, t, J = 7.0 Hz), -2.69 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 146.8 (8C, br), 143.3, 142.3 (3C), 137.4, 134.6 (8C), 131.1 (8C, br), 127.7 (3C), 126.7 (6C), 124.9, 120.3, 120.2, 119.6, 104.1, 65.6; IR (KBr) 3317, 3055, 3028, 2954, 2924, 2877, 1601, 1473, 1350, 1219, 1076, 972, 802, 737 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 418.5 (5.8), 515.5 (4.3), 550.0 (3.8), 589.5 (3.6), 645.5 (3.3) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C₄₇H₃₅N₄O₂ 687.2760, found 687.2756; Anal. Calcd for C₄₇H₃₄N₄O₂: C, 82.19; H, 4.99; N, 8.16. Found: C, 82.10; H, 5.05; N, 7.86.

5-(4-Formylphenyl)-10,15,20-triphenylporphyrin (H2**-8)**: Prepared from bromoporphyrin H2**-1a** (100.0 mg, 0.162 mmol) and arylmagnesium reagent **4u** following the general procedure; Purple solid; 73.9 mg, 71% yield; $R_{\rm f} = 0.41$ (1:1 hexane/toluene); ¹H NMR (CDCl₃, 300 MHz) δ 10.34 (1H, s), 8.89 (4H, d, J = 4.8 Hz), 8.88 (2H, d, J = 4.8 Hz), 8.78 (2H, d, J = 4.8 Hz), 8.39 (2H, d, J = 7.9 Hz), 8.27-8.19 (8H, m), 7.80-7.71 (9H, m), -2.68 (2H, br s); ¹³C NMR (CDCl₃, 75 MHz) δ 192.2, 148.7, 146.7 (8C, br), 142.09, 142.07, 135.7, 135.2, 134.5 (6C), 131.6, 131.3 (4C, br), 130.5, 127.9, 127.8 (3C), 126.7 (6C), 120.8, 120.5, 118.1; IR (KBr) 3579, 3518, 3425, 3317, 3055, 3028, 1601, 1473, 1350, 1265, 1176, 972, 802, 706 cm⁻¹; UV/vis (CHCl₃) λ_{max} (log ε) 419.5 (5.8), 516.5 (4.3), 551.5 (3.8), 590.5 (3.6), 647.0 (3.3) nm; HRMS-FAB⁺ ([M+H]⁺) calcd for C4₅H₃₁N₄O 643.2498, found 643.2495; Anal. Calcd for C4₅H₃₀N₄O: C, 84.09; H, 4.70; N, 8.72. Found: C, 84.39; H, 4.80; N, 8.48.

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