



# An efficient palladium catalyzed synthesis of 2-arylbenzothiazoles

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**Abstract**—A novel and convergent palladium catalyzed synthesis of 2-arylbenzothiazoles has been investigated. The key step in the synthesis is a Suzuki biaryl coupling of 2-bromobenzothiazole with aryl boronic acids to provide a variety of 2-arylbenzothiazole derivatives in good yield. The synthetic utility of this methodology is demonstrated by the synthesis of 2-(4-aminophenyl)-6-methoxybenzothiazole, a PET probe precursor for the in vivo imaging of Alzheimer's disease.

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2-Arylbenzothiazoles are an important class of compounds owing to their potent utility as imaging agents for  $\beta$ -amyloid, antitumor agents, antituberculosics, antiparasitics, calcium channel antagonists, chemiluminescent agents and also as photosensitizers.<sup>1–7</sup> The reported syntheses of 2-arylbenzothiazoles involve the condensation of 2-aminobenzenethiol with 4-substituted phenyl derivatives of nitrile, aldehyde, acid, acid chlorides or esters and by the use of Jacobson's cyclization of thiobenzanilides.<sup>8–10</sup> Other general methods include microwave mediated reaction of *o*-aminothiophenol with  $\beta$ -chlorocinnamaldehydes, reaction of dibenzyl disulfides with *o*-aminothiophenol, reduction of *o,o'*-dinitrodiphenyl disulfide, reaction of *S*-aryl thiobenzoate with arylhaloamines, from 1,2,3-benzodithiazole-2-oxides, radical cyclization of benzyne intermediates and Grignard reactions of arylisothiocyanates.<sup>11–16</sup> However, most of the above methods require multistep synthesis and therefore, we have sought to develop a convergent strategy for the synthesis of 2-arylbenzothiazoles. Our method utilizes a Suzuki cross coupling of the common intermediates 2-bromobenzothiazole (**1**) and 2-bromo-6-methoxybenzothiazole (**2**) with various aryl boronic acids/ esters as the key step (Table 1). Since a wide variety of aryl boronic acids are commercially available, or can be easily prepared from the corresponding bromides,<sup>17</sup> this methodology would offer a higher degree of flexibility with regard to functional groups that can be placed on the 2-aryl moiety, thereby providing a better understanding of the structure activity relationship (SAR) of the target compounds.

**Keywords:** boronic acid; benzothiazole; Suzuki coupling; palladium.  
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Benzothiazole was brominated using *n*BuLi and CBr<sub>4</sub> to provide corresponding 2-bromo benzothiazole (**1**) in 60% yield.<sup>18</sup> 2-bromo-6-methoxybenzothiazole (**2**) (65%) was synthesized by Sandmeyer reaction of 2-amino-6-methoxybenzothiazole by heating with isoamyl nitrite and CuBr<sub>2</sub> in presence of PEG.<sup>19</sup> The biaryl coupling was initially attempted with **1** and phenylboronic acid under the optimized reaction condition using Pd<sub>2</sub>(dba)<sub>3</sub> in DME-water with aqueous K<sub>2</sub>CO<sub>3</sub> to provide 2-phenylbenzothiazole (**4a**) in 50% yield.<sup>20</sup> Under identical conditions, coupling of **1** and **2** with 4-acetylboronic acid, 4-aminophenylboronic ester, Boc protected 4-aminophenylboronic ester, 3-thiopheneboronic acid and 2-naphthaleneboronic acid provided the corresponding 2-arylbenzothiazole derivatives in moderate to good yield (Table 1). Benzothiazole derivative **4i** and its demethylated analogue are positron emission tomography (PET) probe precursors for the in vivo quantification of  $\beta$ -amyloid.<sup>1,21</sup> However, our attempts to couple 2,4-dimethoxyphenylboronic acid with **1** did not afford the desired product probably due to steric hindrance, dehydroboration of the boronic acid and recovery of **1** was the major process observed under the optimized reaction conditions.

In summary, we have utilized a novel chemistry of 2-bromobenzothiazoles for the facile synthesis of 2-arylbenzothiazoles using Suzuki biaryl coupling. The synthetic utility of these reactions are demonstrated by the one step synthesis of 2-amino-6-hydroxybenzothiazole, a potent PET probe precursor for the in vivo imaging of  $\beta$ -amyloid. Future work will be undertaken to develop a combinatorial version of this synthesis for the SAR of 2-arylbenzothiazoles for various pharmaceutical applications.

**Table 1.** Biaryl coupling of 2-bromobenzothiazole (**1**) and 2-bromo-6-methoxy benzothiazole (**2**) with various boronic acids and esters

$\text{X} = \text{H}, \mathbf{1}$   
 $\text{X} = \text{OMe}, \mathbf{2}$

$\text{ArB(OH)}_2$   
 $\mathbf{3a-i}$

$\text{Pd}_2(\text{dba})_3, \text{DME-Water}$   
 $\text{K}_2\text{CO}_3, 100^\circ\text{C}, 6\text{h}$

$\mathbf{4a-n}$

S. No	Aryl halide	Boronic acid	Product	Yield (%) <sup>a</sup>
1	<b>1</b>		<b>4a</b>	51
2	<b>1</b>		<b>4b</b>	40
3	<b>1</b>		<b>4c</b>	72
4	<b>1</b>		<b>4d</b>	48
5	<b>1</b>		<b>4e</b>	---
6	<b>1</b>		<b>4f</b>	74
7	<b>1</b>		<b>4g</b>	55
8	<b>1</b>		<b>4h</b>	35
9	<b>2</b>		<b>4i</b>	55
10	<b>2</b>		<b>4j</b>	58
11	<b>2</b>		<b>4k</b>	63
12	<b>2</b>		<b>4l</b>	54
13	<b>2</b>		<b>4m</b>	68
14	<b>2</b>		<b>4n</b>	62

a. The yield represent the isolated yields of the product after column chromatography based on aryl bromide.

b. Pinacolate ester of the corresponding boronic acid was used for the coupling.

## References

- Mathis, C. A.; Wang, Y.; Holt, D. P.; Huang, G.-F.; Debnath, M. L.; Klunk, W. E. *J. Med. Chem.* **2003**, *46* (13), 2740–2754.
- Hutchinson, I.; Jennings, S. A.; Vishnuvajjala, B. R.; Westwell, A. D.; Stevens, M. F. G. *J. Med. Chem.* **2002**, *45* (3), 744–747.
- Stevens, M. F. G.; Wells, G.; Westwell, A. D.; Poole, T. D. *PCT Int. Appl.* 2003, WO 0304479.
- Caujolle, R.; Loiseau, P.; Payard, M.; Gayral, P.; Kerhir, M. N. *Ann. Pharma. Fr.* **1989**, *47* (2), 68–73.
- Yamamoto, K.; Fujita, M.; Tabashi, K.; Kawashima, Y.; Kato, E.; Oya, M.; Iso, T.; Iwao. *J. Med. Chem.* **1983**, *31* (5), 919–930.
- Yoshida, H.; Nakao, R.; Nohta, H.; Yamaguchi, M. *Dyes and Pigments* **2000**, *47* (3), 239–245.
- Petkov, I.; Deligeorgiev, T.; Markov, P.; Evstatiev, M.; Fakirov, S. *Polym. Degrad. Stab.* **1991**, *33* (1), 53–66.
- Shi, D.-F.; Bradshaw, T. D.; Wrigley, S.; McCall, C. J.; Lelieveld, I. F.; Stevens, M. F. G. *J. Med. Chem.* **1996**, *39*, 3375–3384.
- Hein, D. W.; Alheim, R. J.; Leavitt, J. J. *J. Am. Chem. Soc.* **1957**, *79*, 427–429.
- Ben-Alloum, A.; Bakkas, S.; Soufiaoui, M. *Tetrahedron Lett.* **1997**, *38*, 6395–6396.

11. Paul, S.; Gupta, M.; Gupta, R. *Synth. Commun.* **2002**, *32* (23), 3541–3547.
12. Shirinian, V. Z.; Melkova, S. Yu.; Belen'kii, L. I.; Krayushkin, M. M.; Zelinsky, N. D. *Russ. Chem. Bull.* **2000**, *49* (11), 1859–1862.
13. Zhong, W. H.; Zhang, Y. M.; Chen, X. Y. *J. Indian Chem. Soc.* **2001**, *78* (6), 316–318.
14. Roe, A.; Tucker, W. P. *J. Heterocycl. Chem.* **1965**, *2*, 148–151.
15. Stanetty, P.; Krumpak, B. *J. Org. Chem.* **1996**, *61*, 5130–5133.
16. Ares, J. J. *Synth. Commun.* **1991**, *21* (5), 625–633.
17. Murata, M.; Oyama, T.; Watanabe, Y.; Matsuda, Y. *J. Org. Chem.* **2000**, *65*, 164.
18. Boga, C.; Vecchio, E. D.; Forlani, L.; Todeso, P. E. *J. Organomet. Chem.* **2000**, *601*, 233–236.
19. Suzuki, N.; Nomoto, T.; Toya, Y.; Yoda, B.; Saeki, A. *Chem. Express* **1992**, *7* (9), 717–720.
20. Typical experimental procedure for the Suzuki coupling: A suspension of 2-bromobenzothiazole (86 mg, 0.4 mmol), thiophene-3-boronic acid (67 mg, 0.52 mmol) and Pd<sub>2</sub>(dba)<sub>3</sub> (18 mg, 5 mol%) in 1,2-dimethoxyethane (1 mL) was deaerated and stirred under argon. Deionized water (0.1 mL) and aqueous K<sub>2</sub>CO<sub>3</sub> (2 M, 400 μL, 0.8 mmol) were added to the reaction mixture and heated at 100°C for 6 h. The reaction mixture was diluted with EtOAc, dried over MgSO<sub>4</sub>, passed through a short pad of celite, concentrated under vacuum and column chromatographed (97:3 hexane: ethyl acetate) to yield 2-thiophene-3-yl-benzothiazole (**4f**) as a colorless solid (65 mg) in 75% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.04 (d, 1 H, *J*=8.2 Hz), 8.02 (m, 1 H), 7.88 (1 H, d, *J*=8.0 Hz), 7.77 (1 H, d, *J*=5.0 Hz), 7.48 (t, 1 H, *J*=8.0 Hz), 7.44 (dd, 1 H, *J*=3, 5 Hz), 7.38 (t, 1 H, *J*=8.0 Hz); HRMS: calcd. for C<sub>11</sub>H<sub>8</sub>NS<sub>2</sub> (MH<sup>+</sup>): 218.0098 found: 218.0108; mp 110°C (lit.<sup>22</sup> 111.5–112.5°C).
21. Kumar, J. S. D.; Wang, T. S.; Arango, V.; Underwood, M. D.; Parsey, R. V.; Simpson, N. R.; Kassir, S.; Cooper, A.; Arcement J.; Van Heertum, R. L.; Mann, J. J. 226th National ACS meeting, Sept 7–11, New York, USA, 2003.
22. Lindley, J. M.; Meth-Cohn, O.; Suschitzky, H. *J. Chem. Soc., Perkin Trans. 1* **1978**, 1198.