N-Alkylation of Aniline and Its Derivatives by Alcohols in the Presence of Copper Compounds

A. R. Bayguzina^{a,b},* Ch. F. Musina^b, and R. I. Khusnutdinov^{a,b}

^a Institute of Petrochemistry and Catalysis of the Russian Academy of Sciences, pr. Oktyabrya 141, Ufa, 450075 Bashkortostan, Russia *e-mail: inklab4@gmail.com

^b Ufa State Petroleum Technological University, ul. Kosmonavtov 1, Ufa, 450062 Bashkortostan, Russia

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Abstract—N-Alkyl- and N,N-dialkyl-substituted anilines were obtained in the reaction of aniline and its derivatives with primary and secondary alcohols in the presence of catalysts CuCl₂·2H₂O, CuBr₂ and halomethanes as promoters.

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N-Alkyl- and N,N-dialkyl derivatives of aniline find wide practical applications. In particular, N-methylaniline is used as antidetonating additive to gasoline, and N,N-dimethylaniline finds application in the preparation of paints, explosives, and polyester resins [1]. N-Alkyl derivatives of aniline with functional groups are used in production of explosives, photography reagents, pesticides, stabilizers, paints, medications, and polymers [2].

N-Alkylation of derivatives of aniline is mainly performed with alkyl halides at high temperature. However this process has its disadvantages connected with low availability and toxicity of alkyl halides and the necessity of utilization of forming hydrogen halides [3]. Other important method of synthesis of N-alkyl derivatives is based on N-alkylation of aniline with alcohols in the presence of complexes of palladium, ruthenium, iridium, silver, gold, platinum, nickel, cobalt, rhodium, and osmium [4, 5]. Important benefit of alcohols application for N-alkylation of aniline is the generation of water as the side product of the reaction. Among the drawbacks of the mentioned

metal complex catalysts is their expensiveness, the necessity to apply additional reagents (usually bases), and low activity with respect to secondary alcohols [6].

We developed a general method of synthesis of Nalkyl- and N,N-dialkyl-substituted products by the reaction of aniline and its derivatives with alcohols in the presence of copper compounds. Active catalysts of *N*-alkylation of aniline and its derivatives with alcohols are CuBr₂, Cu(acac)₂, CuCl₂·2H₂O, Cu(CH₃COO)₂, and Cu(C₆H₅COO)₂. The regularities of N-alkylation were investigated by an example of the reaction of aniline with methanol, optimal conditions of reaction were as follows: 180°C, 6 h at the molar ratio Cu-aniline-MeOH 10: 100: 200. The reaction occurs with the formation of N-methylaniline 1 and N,N-dimethylaniline 2, whose yield and ratio depend of the catalyst nature (Scheme 1). The best ones in the series of investigated catalysts are CuBr₂ and CuCl₂·2H₂O, in the presence of which the conversion of aniline is 61 and 46% respectively. Without the catalyst the reaction does not occur.

G-4-14	Molar ratio	Aniline con-	Yield, %	
Catalyst–promoter	CuBr ₂ –P–aniline–MeOH	version, %	1	2
CuBr ₂	10:0:100:200	61	13	48
CuBr ₂ –CBr ₄	10:5:100:200	94	32	62
CuBr ₂ –HCl	10:5:100:200	98	26 6 33	72 94 60
CuBr ₂ –CHBr ₃	10:5:100:200	100		
CuBr ₂ –HBr	10:5:100:200	93		
CuBr ₂ –CCl ₄	10: 5: 100: 200	100	0	99
	1:5:100:200	100	0	99
	0.1 : 5 : 100 : 200	100	0	99
	0.1 : 5 : 100 : 200 ^b	60	44	16
	0.1 : 5 : 100 : 200°	77	42	35
	0.1 : 5 : 100 : 200 ^d	100	0	0
	0.1:3:100:200	65	33	33
	0.1:1:100:200	45	21	24
$CuCl_2 \cdot 2H_2O$	10:100:200	46	22	24
$CuCl_2 \cdot 2H_2O$ – CCl_4	0.1:5:100:200	100	0	99
CuBr ₂ –CCl ₄	0.1 : 5 : 100 : 200 ^e	100	2	98

Table 1. Effect of promoter character on the yield of *N*-methylaniline 1 and *N*,*N*-dimethylaniline 2^a

Aiming at improving the yield of *N*-methyl and *N*,*N* -dimethyl aniline derivatives we tested as promoters a series of compounds: CBr₄, CCl₄, CHBr₃, HBr, and HCl (Table 1).

The most effective catalytic systems were CuBr₂–CCl₄ and CuCl₂·2H₂O–CCl₄ at a molar ratio Cu–CCl₄–aniline–alcohol 0.1 : 5 : 100 : 200. Yield of *N*,*N*-dimethylaniline **2** was 99%.

In discovered optimal conditions in the presence of CuBr₂–CCl₄ 3-chloroaniline, 4-chloroaniline, 2-

Scheme 2.

$$R + MeOH = \frac{CuBr_2-CCl_4}{180^{\circ}C, 6 \text{ h}} + R = 3-Cl (3), 4-Cl (4), 2-OMe (5), 4-OMe (6).$$

methoxyaniline, 4-methoxyaniline reacted with methanol to afford the corresponding *N*,*N*-dimethyl derivatives in quantitative yields (Scheme 2).

The reaction of *N*-alkylation of aniline in the presence of catalytic systems CuCl₂·2H₂O, CuBr₂ and CCl₄ proceeds as well with the other primary alcohols (EtOH, PrOH, BuOH, AmOH, HexOH). One of the benefits of copper compounds as catalysts is the possibility of introduction of secondary alcohols into the reaction of *N*-alkylation of aniline derivatives, in particular, of cyclopentanol, cyclohexanol, cycloheptanol, butan-2-ol and propan-2-ol (Table 2).

The reaction of aniline with butan-1,4-diol in the optimal conditions (180°C, 6 h, molar ratio Cu–CCl₄–aniline–alcohol 0.1 : 5 : 100 : 200) affords 1-phenylpyrrolidine **22** (yield 98%), a product of *N*-heterocyclization (Scheme 3).

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a Reaction conditions: 180°C, 6 h.

b 150°C.

c 170°C.

^d 200°C (tarring).

e Reaction duration 5 h.

Table 2. N-Alkylation of aniline with primary and secondary alcohols in the presence of the catalytic system CuBr₂–CCl₄

ROH	Aniline conversion, %	Yield of compounds 7–21, %		
		C ₆ H ₅ NHR	C ₆ H ₅ NR ₂	
EtOH	91	78 (7)	13 (17)	
PrOH	75	67 (8)	8 (18)	
BuOH	75	65 (9)	10 (19)	
AmOH	46	44 (10)	2 (20)	
HexOH	44	42 (11)	2 (21)	
OH	71	71 (12)	-	
OH	33	33 (13)	-	
OH	41	41 (14)	-	
<i>i</i> -PrOH	53	53 (15)	_	
s-BuOH	55	55 (16)	_	

Table 3. N-Alkylation of aniline and its derivatives with benzyl alcohol at 180°Ca

RNH ₂	[Cu]–CCl ₄	Amine conversion, %	Yield, %		
			RC ₆ H ₄ NHBn	RC ₆ H ₄ NBn ₂	
NH ₂	CuBr ₂ –CCl ₄	85	75 (23)	10 (27)	
-//-	CuBr ₂ –CCl ₄ ^b	100	80 (23)	19 (27)	
//	CuCl ₂ ·2H ₂ O–CCl ₄ ^b	100	79 (23)	20 (27)	
//	CuCl ₂ ·2H ₂ O–CCl ₄ ^c	100	66 (23)	32 (27)	
NH ₂	CuCl ₂ ·2H ₂ O–CCl ₄ ^b	100	74 (24)	24 (28)	
NH ₂	CuCl ₂ ·2H ₂ O-CCl ₄ ^b	100	62 (25)	34 (29)	
NH ₂	CuCl ₂ ·2H ₂ O–CCl ₄ ^b	100	78 (26)	20 (30)	

^a Molar ratio [$\overline{\text{Cu}}$]– $\overline{\text{CCl}}_4$ –amine– $\overline{\text{BnOH 0.1}}: 5:100:100.$ ^b The reaction was performed in benzene (0.5 mL).

N-alkylation of aniline derivatives with benzyl alcohol was investigated in order to obtain N-benzyl and N,N-dibenzyl derivatives applicable to the synthesis of drugs, pesticides, antioxidants, and lubricants [7].

N-Alkylation of aniline and its substituted derivatives with benzyl alcohol to give N-benzyl (23-26) and N,N-dibenzyl (27–30) products is effectively catalyzed with CuCl₂·2H₂O and CuBr₂ in the presence

^c The reaction was performed in toluene (0.5 mL).

Scheme 4.

NH₂

$$+ BnOH$$

$$- \frac{[Cu]-CCI_4}{180^{\circ}C, 6 h, solvent}$$

$$+ MeOH$$

$$- \frac{CuBr_2}{-H_2O}$$

$$+ MeOH$$

$$- \frac{CuBr_2-HCI}{-H_2O}$$

$$+ MeOH$$

$$- \frac{CuBr_2-HCI}{-H_2O}$$

$$+ MeOH$$

$$- \frac{CuBr_2-HCI}{-H_2O}$$

$$+ MeOH$$

$$- \frac{CuBr_2-HCI}{-H_2O}$$

$$- \frac{NHMe}{-H_2O}$$

of promoter CCl_4 . The reaction occurs in benzene and toluene at a molar ratio $Cu-CCl_4-C_6H_5NH_2-C_6H_5CH_2OH~0.1:5:100:100$, conversion of aniline is 100%, and yields of *N*-benzyl (23–26) and *N*,*N*-dibenzyl (27–30) derivatives are 64–80 and 20–36% respectively (Scheme 4). Without solvent the conversion of aniline in the reaction is not full (Table 3).

To find the explanation for the favorable effect of promoters on the course of the reaction an additional analysis of the reaction mass was performed. According to the analysis data, in the course of the reaction CCl₄ promoter undergoes hydrolysis and transforms into CO₂ and HCl. The presence of HCl in the reaction mixture was confirmed by the mercurimetric titration. The concentration of HCl was 48.72 (after 3 h), 44.16 (after 4 h), 42.46 (after 5 h), 41.4 mg/mL (after 6 h). The discovery of HCl in the reaction mixture allows a suggestion that HCl forms with copper salts Gustavson complexes [CuBr₂·HCl] and [CuCl₂·HCl] that catalyze *N*-alkylation of aniline with methanol to form *N,N*-dimethylaniline **2** (Scheme 5).

In all experiments water forms in concentration corresponding to the theoretical amount that is confirmed by Fischer titration.

Hence, CuBr₂ and CuCl₂·2H₂O promoted by CCl₄ actively catalyze *N*-alkylation of aniline and its derivatives with primary and secondary alcohols providing *N*-alkyl and *N*,*N*-dialkyl derivatives.

EXPERIMENTAL

¹H and ¹³C NMR spectra were registered on a spectrometer Bruker Avance-400 (400.13 and 100.62 MHz respectively) in CDCl₃. Mass spectra were recorded on a chromato-mass-spectrometer Shimadzu GCMS-QP2010Plus (capillary column SPB-5 30 m × 0.25 mm, carrier gas helium, temperature programmed from 40 to 300°C at a rate 8 deg/min, evaporation temperature 280°C, temperature of ion source 200°C, ionization energy 70 eV). Chromatographic analysis was performed on a chromatograph Shimadzu GC-9A, GC-2014 [column 2 m × 3 mm, stationary phase silicone SE-30 (5%) on Chromaton N-AW-HMDS, ramp from 50 to 270°C, 8 deg/min, carrier gas helium (47 mL/min)].

Initial reagents were commercially available aniline, 3-chloroaniline, 4-chloroaniline, 2-methoxyaniline, 4-methoxyaniline, methanol, ethanol, propan-1-ol, butan-1-ol, penthan-1-ol, hexan-1-ol, cyclopentanol, cyclohexanol, cycloheptanol, butan-2-ol, propan-2-ol, butan-1,4-diol, benzyl alcohol, CCl₄, CBr₄, and CHBr₃, and also catalysts CuBr₂, CuCl₂·2H₂O, Cu(acac)₂, Cu(CH₃COO)₂, Cu(C₆H₅COO)₂.

Alkylation of aniline derivatives. General method. Reactions were carried out in glass ampules (12 mL) placed into pressure microreactor of stainless steel of 17 mL capacity. Into the ampule in an argon flow was loaded 0.1 mmol of CuBr₂, 1 mmol (0.2 mL) of aniline, 2 mmol of primary or 1 mmol of secondary alcohol (0.18 mL of methanol, 0.26 mL of ethanol,

0.32 mL of propan-1-ol, 0.4 mL of butan-1-ol, 0.46 mL of penthan-1-ol, 0.54 mL of hexan-1-ol, 0.2 mL of cyclopentanol, 0.46 mL of cyclohexanol, 0.26 mL of cycloheptanol, 0.19 mL of butan-1,4-diol, 0.2 mL of butan-2-ol, 0.17 mL of propan-2-ol, 0.2 mL of benzyl alcohol), 5 mol (0.01 mL) of CCl₄, and 0.5 mL of benzene (toluene) as solvent. The sealed ampule was placed into the reactor, it was hermetically closed and heated during 6 h at 180°C at constant stirring, then the reactor was cooled to ~20°C, the ampule was opened, the reaction mixture was neutralized with 10% aqueous solution of Na₂CO₃, the reaction products were extracted with methylene chloride, the solvent was distilled off, the residue was distilled in a vacuum or subjected to column chromatography on silica gel, elution with a mixture petroleum ether-ethyl acetate, 1:1 (column 21×1.2 cm).

The structure of *N*-alkyl and *N*,*N*-dialkyl derivatives **1–30** was proved by NMR and mass spectra, as well as by comparison with authentic samples and published data. To identify compounds **1–30** we used samples obtained in 3–6 experiments.

N-Methylaniline (1). Yield 44%, light-yellow liquid, bp 78–80°C (10 mmHg) {76–77°C (10 mmHg) [8]}. ¹H NMR spectrum, δ, ppm: 3.00 s (3H, CH₃), 3.89 s (1H, NH), 6.66 d (2H, $C^{2.6}$ H, *J* 8 Hz), 6.70–6.85 m (1H, C^4 H), 7.15–7.35 m (2H, $C^{3.5}$ H). ¹³C NMR spectrum, δ, ppm: 30.90 (CH₃), 112.59 ($C^{2.6}$), 116.99 (C^4), 129.22 ($C^{3.5}$), 149.36 (C^I). Mass spectrum, *m/z* (I_{rel} , %): 107 (87) [M]⁺, 106 (100), 93 (7), 77 (30), 65 (13).

N,*N*-Dimethyl-3-chloroaniline (3). Yield 99%, light-yellow oily liquid, bp 105–106°C (10 mmHg) $\{231-232^{\circ}\text{C}\ (740 \text{ mmHg})\ [9]\}$. ¹H NMR spectrum, δ, ppm: 3.00 s (6H, CH₃), 6.80 d (2H, C^{2,6}H, *J* 7.2 Hz), 6.80–6.82 m (1H, C⁴H), 7.25–7.35 m (3H, C^{3,5}H). ¹³C NMR spectrum, δ, ppm: 40.66 (CH₃), 112.70 (C^{2,6}), 116.66 (C⁴), 129.11 (C^{3,5}), 150.70 (C¹). Mass spectrum, m/z (I_{rel} , %): 121 (90) [M]⁺, 120 (100), 104 (15), 91 (8), 77 (25).

N,*N*-Dimethyl-3-chloroaniline (3). Yield 99%, light-yellow oily liquid, bp 105–106°C (10 mmHg) $\{231-232^{\circ}\text{C}\ (740 \text{ mmHg})\ [9]\}$. ¹H NMR spectrum, δ, ppm: 2.98 s (6H, CH₃), 6.80 d (1H, C^{4,6}H, *J* 7.4 Hz), 6.88 s (2H, C²H), 7.19 t (1H, C⁵H, *J* 7.8 Hz). ¹³C NMR spectrum, δ, ppm: 41.52 (CH₃), 112.15 (C⁶), 113.85 (C²), 117.67 (C⁴), 130.29 (C⁵), 135.14 (C³), 150.06 (C¹). Mass spectrum, $m/z\ (I_{\text{rel}}, \%)$: 155 (80) $[M]^+$, 139 (8), 118 (15), 111 (8), 77 (13).

N,*N*-Dimethyl-4-chloroaniline (4). Yield 99%, light-yellow oily liquid, bp 120–121°C (20 mmHg) {233–236°C (740 mmHg) [9]}. ¹H NMR spectrum, δ, ppm: 2.91 s (6H, CH₃), 6.63 d (1H, $C^{2.6}$ H, *J* 8 Hz), 7.16 d (1H, $C^{3.5}$ H, *J* 12 Hz). ¹³C NMR spectrum, δ, ppm: 40.42 (CH₃), 113.49 ($C^{2.6}$), 120.97 (C^4), 129.65 ($C^{3.5}$), 149.24 (C^I). Mass spectrum, *m/z* (I_{rel} , %): 155 (82) [M]⁺, 154 (100), 139 (17), 118 (15), 111 (8), 77 (13).

N,*N*-Dimethyl-2-methoxyaniline (5). Yield 99%, light-yellow oily liquid, bp 113°C (18 mmHg) {86–90°C (16 mmHg) [10]}. ¹H NMR spectrum, δ, ppm: 2.80 s (6H, CH₃), 3.90 s (3H, OCH₃), 6.88 d (1H, C⁶H, *J* 8 Hz), 6.90–7.03 m (3H, C^{3,4,5}H). ¹³C NMR spectrum, δ, ppm: 43.37 [N(CH₃)₂], 55.27 (OCH₃), 110.90 (C³), 118.16 (C⁶), 120.76 (C⁴), 122.48 (C⁵), 142.43 (C¹), 152.40 (C²). Mass spectrum, m/z (I_{rel} , %): 151 (100) [M]⁺, 136 (66), 120 (24), 107 (20), 93 (10), 77 (12).

N,*N*-Dimethyl-4-methoxyaniline (6). Yield 99%, light-yellow oily liquid, bp 124–125°C (20 mmHg) {70–75°C (5 mmHg) [11]}. ¹H NMR spectrum, δ, ppm: 2.87 s (6H, CH₃), 3.37 s (3H, OCH₃), 6.77 d (1H, C^{2,6}H, *J* 12 Hz), 6.85 d (1H, C^{3,5}H, *J* 8 Hz). ¹³C NMR spectrum, δ, ppm: 41.82 [N(CH₃)₂], 55.72 (OCH₃), 114.60 (C^{3,5}), 114.92 (C^{2,6}), 145.89 (C¹), 151.86 (C⁴). Mass spectrum, m/z (I_{rel} , %): 151 (65) [M]⁺, 136 (100), 120 (10), 108 (18), 92 (8), 77 (5).

N-Ethylaniline (7). Yield 78%, light-yellow liquid, bp 85–87°C (12 mmHg) (204°C [12]). ¹H NMR spectrum, δ, ppm: 1.29 t (3H, C^8H_3 , J 8 Hz), 3.20 q (2H, C^7H_2), 3.66 s (1H, NH), 6.65 d (2H, $C^{2.6}H$, J 8 Hz), 6.70–6.77 m (1H, C^4H), 7.72 t (2H, $C^{3.5}H$, J 8 Hz). ¹³C NMR spectrum, δ, ppm: 14.91 (C^8), 38.47 (C^7), 112.76 ($C^{2.6}$), 117.22 (C^4), 129.24 ($C^{3.5}$), 148.48 (C^1). Mass spectrum, m/z ($I_{\rm rel}$, %): 121 (33) [M]⁺, 107 (8), 106 (100), 93 (5), 77 (18).

N,*N*-**Diethylaniline** (17). Yield 13%, light-yellow liquid, bp 96–98°C (10 mmHg) (216°C [13]). ¹H NMR spectrum, δ, ppm: 1.20 t (3H, $C^{8.8'}$ H₃, *J* 8 Hz), 3.39 q (2H, $C^{7.7'}$ H₂, *J* 8 Hz), 6.55–6.57 m (2H, $C^{2.6}$ H), 6.57–6.59 m (1H, C^4 H), 7.05–7.15 m (2H, $C^{3.5}$ H). ¹³C NMR spectrum, δ, ppm: 12.58 ($C^{8.8'}$), 44.22 ($C^{7.7'}$), 113.73 ($C^{2.6}$), 115.12 (C^4), 129.05 ($C^{3.5}$), 148.33 (C^{1}). Mass spectrum, m/z (I_{rel} , %): 149 (35) [M]⁺, 134 (100), 120 (5), 106 (45), 93 (5), 77 (20).

N-Propylaniline (8). Yield 67%, light-yellow liquid, bp 112–114°C (20 mmHg) (222°C [12]). ¹H

NMR spectrum, δ , ppm: 1.09 t (3H, C^9H_3 , J 8 Hz), 1.65–1.77 m (2H, C^8H_2), 3.16 t (2H, C^7H_2 , J 8 Hz), 3.71 s (1H, NH), 6.69 d (2H, $C^{2,6}H$, J 8 Hz), 6.70–6.85 m (1H, C^4H), 7.22–7.35 m (2H, $C^{3,5}H$). ¹³C NMR spectrum, δ , ppm: 11.74 (C^9), 22.79 (C^8), 45.85 (C^7), 112.78 ($C^{2,6}$), 117.12 (C^4), 129.30 ($C^{3,5}$), 148.63 (C^1). Mass spectrum, m/z ($I_{\rm rel}$, %): 136 (5) [M]⁺, 135 (30), 118 (5), 106 (100), 77 (18).

N,*N*-**Dipropylaniline** (18). Yield 8%, light-yellow oily liquid, bp 112–114°C (10 mmHg) {62–66°C (0.23 mmHg) [14]}. ¹H NMR spectrum, δ, ppm: 1.02 t (3H, $C^{9,9'}$ H₃, *J* 4 Hz), 1.80–1.92 m (2H, $C^{8,8'}$ H₂), 3.30–3.35 m (2H, $C^{7,7'}$ H₂), 6.58–6.64 m (2H, $C^{2,6}$ H), 6.65–6.70 m (1H, C^4 H), 7.05–7.20 m (2H, $C^{3,5}$ H). ¹³C NMR spectrum, δ, ppm: 11.55 ($C^{9,9'}$), 20.49 ($C^{8,8'}$), 52.94 ($C^{7,7'}$), 111.77 ($C^{2,6}$), 115.19 (C^4), 129.08 ($C^{3,5}$), 148.30 (C^I). Mass spectrum, *m/z* ($I_{\rm rel}$, %): 177 (20) [M]⁺, 162 (5), 148 (100), 120 (10), 106 (72), 77 (18).

N-Butylaniline (9). Yield 65%, light-yellow oily liquid, bp 112–114°C (10 mmHg) {60–61°C (0.5 mmHg) [15]}. ¹H NMR spectrum, δ, ppm: 0.97 t (3H, C^{10} H₃, J 8 Hz), 1.40–1.50 m (2H, C^{9} H₂), 1.55–1.65 m (2H, C^{8} H₂), 3.11 t (2H, C^{7} H₂, J 4 Hz), 3.66 s (1H, NH), 6.61 d (2H, $C^{2.6}$ H, J 8 Hz), 6.65–6.70 m (1H, C^{4} H), 7.15–7.20 m (2H, $C^{3.5}$ H). ¹³C NMR spectrum, δ, ppm: 13.89 (C^{10}), 19.30 (C^{9}), 31.65 (C^{8}), 43.65 (C^{7}), 112.64 ($C^{2.6}$), 116.99 (C^{4}), 129.18 ($C^{3.5}$), 148.56 (C^{1}). Mass spectrum, m/z (I_{rel} , %): 150 (5) [M]⁺, 149 (20), 107 (10), 106 (100), 93 (3), 77 (10).

N,*N*-**Dibutylaniline (19).** Yield 10%, light-yellow oily liquid, bp 146–148°C (15 mmHg) {82–84°C (0.23 mmHg) [14]}. H NMR spectrum, δ, ppm: 0.95–1.10 m (6H, $C^{10,10'}$ H₃), 1.40–1.60 m (3H, $C^{9,9'}$ H₂), 1.60–1.72 m (4H, $C^{8,8'}$ H₂), 3.35 t (4H, $C^{7,7'}$ H₂, *J* 8 Hz), 6.56–6.60 m (2H, $C^{2,6}$ H), 6.61–6.65 m (1H, C^{4} H), 7.05–7.18 m (2H, $C^{3,5}$ H). 13 C NMR spectrum, δ, ppm: 14.02 ($C^{10,10'}$), 20.40 ($C^{9,9'}$), 29.40 ($C^{8,8'}$), 50.84 ($C^{7,7'}$), 111.35 ($C^{2,6}$), 115.15 (C^{4}), 131.61 ($C^{3,5}$), 148.26 (C^{1}). Mass spectrum, m/z (I_{rel} , %): 205 (34) [M]⁺, 162 (100), 120 (79), 106 (60), 93 (5), 77 (18).

N-Pentylaniline (10). Yield 44%, light-yellow oily liquid, bp 148–150°C (20 mmHg) {130°C (11 mmHg) [16]}. ¹H NMR spectrum, δ, ppm: 0.96 t (3H, $C^{II}H_3$, *J* 4 Hz), 1.32–1.48 m (2H, $C^{I0}H_2$), 1.32–1.48 m (2H, C^9H_2), 1.55–1.68 m (2H, C^8H_2), 3.62 t (2H, C^7H_2 , *J* 4 Hz), 3.80 s (1H, NH), 6.62 d (2H, $C^{2.6}H$, *J* 8 Hz), 6.67–6.73 m (1H, C^4H), 7.12–7.22 m (2H, $C^{3.5}H$). ¹³C NMR spectrum, δ, ppm: 14.08 (C^{II}), 22.56 (C^{I0}), 28.02 (C^9), 32.44 (C^8), 44.05 (C^7), 112.86 ($C^{2.6}$),

117.14 (C⁴), 129.25 (C^{3,5}), 148.63 (C¹). Mass spectrum, m/z (I_{rel} , %): 163 (20) [M]⁺, 132 (3), 118 (5), 106 (100), 93 (7), 77 (14).

N-Hexylaniline (11). Yield 42%, light-yellow oily liquid, bp 157–159°C (20 mmHg) {88–89°C (0.1 mmHg) [15]}. ¹H NMR spectrum, δ, ppm: 0.91 t (3H, $C^{12}H_3$, J 4 Hz), 1.25–1.45 m (6H, $C^{9,I0,II}H_2$), 1.55–1.65 m (2H, C^8H_2), 3.62 t (2H, C^7H_2 , J 4 Hz), 3.80 s (1H, NH), 6.62 d (2H, $C^{2,6}H$, J 8 Hz), 6.67–6.73 m (1H, C^4H), 7.12–7.22 m (2H, $C^{3,5}H$). ¹³C NMR spectrum, δ, ppm: 14.03 (C^{12}), 22.65 (C^{1I}), 25.47 (C^{10}), 26.87 (C^9), 29.53 (C^8), 44.04 (C^7), 112.76 ($C^{2,6}$), 117.10 (C^4), 129.20 ($C^{3,5}$), 148.56 (C^I). Mass spectrum, m/z (I_{rel} , %): 177 (18) [M]⁺, 132 (3), 118 (3), 106 (100), 93 (5), 77 (10).

N-Cyclopentylaniline (12). Yield 71%, lightyellow oily liquid, bp 140–142°C (15 mmHg) {137°C (12 mmHg) [17]}. ¹H NMR spectrum, δ, ppm: 1.45–1.55 m (2H^a, C^{3',4'}H), 1.60–1.70 m (2H^a, C^{2',5'}H), 1.70–1.80 m (2H^e, C^{3',4'}H), 2.00–2.10 m (2H^e, C^{2',5'}H), 3.67 br.s(1H, NH), 3.75–3.9 m (1H, C^{1'}H), 6.63 d (2H, C^{2,6}H, *J* 8 Hz), 6.71 t (1H, C⁴H, *J* 8 Hz), 7.19 t (2H, C^{3,5}H, *J* 8 Hz). ¹³C NMR spectrum, δ, ppm: 24.08 (C^{3',4'}), 33.61 (C^{2',5'}), 54.65 (C^{1'}), 113.16 (C^{2,6}), 116.89 (C⁴), 129.19 (C^{3,5}), 148.06 (C¹). Mass spectrum, *m/z* (I_{rel} , %): 161 (50) [M]⁺, 133 (13), 132 (100), 119 (30), 93 (18).

N-Cyclohexylaniline (13). Yield 23%, light-yellow oily liquid, bp 140–142°C (10 mmHg) {92–96°C (2 mmHg) [11]}. ¹H NMR spectrum, δ, ppm: 1.20–1.50 m (2H, C⁴'H), 1.50–1.70 m (4H, C^{3′,5′}H), 2.00–2.20 m (4H, C^{2′,6′}H), 3.31–3.37 m (1H, C^{1′}H), 3.37 s (1H, NH), 6.64 d (2H, C^{2,6}H, *J* 8 Hz), 6.75–6.85 m (2H, C^{3,5}H), 7.10–7.30 m (1H, C⁴H). ¹³C NMR spectrum, δ, ppm: 25.09 (C^{3′,5′}), 25.99 (C^{4′}), 35.49 (C^{2′,6′}), 51.72 (C^{1′}), 113.24 (C^{2,6}), 117.87 (C⁴), 129.30 (C^{3,5}), 147.46 (C¹). Mass spectrum, m/z ($I_{\rm rel}$, %): 175 (30) [M]⁺, 146 (5), 132 (100), 106 (15), 93 (17).

N-Cycloheptylaniline (14). Yield 41%, lightyellow oily liquid, bp 126–128°C (5 mmHg) {156°C (11 mmHg) [17]}. ¹H NMR spectrum, δ, ppm: 1.20–1.80 m (8H, $C^{3'-6'}H_2$), 2.00–2.10 m (4H, $C^{2',7'}H_2$), 3.44–3.52 m (1H, $C^{I'}H$), 3.65 br.s (1H, NH), 6.59 d (2H, $C^{2,6}H$, *J* 8 Hz), 6.70–6.75 m (1H, C^4H), 7.20 t (2H, $C^{3,5}H$, *J* 8 Hz). ¹³C NMR spectrum, δ, ppm: 24.44 ($C^{4',5'}$), 28.39 ($C^{3',6'}$), 34.86 ($C^{2',7'}$), 54.66 ($C^{I'}$), 113.23 ($C^{2,6}$), 116.77 (C^4), 129.29 ($C^{3,5}$), 147.38 (C^{I}). Mass spectrum, m/z (I_{rel} , %): 189 (38) [M]⁺, 160 (5), 146 (17), 132 (100), 106 (17), 93 (20), 77 (12).

N-Isopropylaniline (15). Yield 53%, yellow oily liquid, bp 68–70°C (15 mmHg) {105–106°C (27 mmHg) [18]}. 1 H NMR spectrum, δ, ppm: 1.37 t (6H, $C^{7,9}$ H₃, J 8 Hz), 3.60–3.70 m (1H, C^{8} H), 4.05 br.s (1H, NH), 6.54 d (2H, $C^{2.6}$ H, J 8 Hz), 6.55–6.75 m (1H, C^{4} H), 7.18 t (2H, $C^{3.5}$ H, J 8 Hz). 13 C NMR spectrum, δ, ppm: 22.80 ($C^{7,9}$), 44.19 (C^{8}), 113.22 ($C^{2.6}$), 116.93 (C^{4}), 129.04 ($C^{3.5}$), 146.92 (C^{I}). Mass spectrum, m/z (I_{rel} , %): 135 (30) [M]⁺, 121 (8), 120 (100), 120 (100), 106 (10), 93 (18), 77 (12).

N-(**Butan-2-yl)aniline** (16). Yield 55%, light-yellow oily liquid, bp 99–100°C (10 mmHg) {74–75°C (0.15 mmHg) [15]}. ¹H NMR spectrum, δ, ppm: 0.98 t (3H, C^{10} H₃, J 8 Hz), 1.20 d (3H, C^{7} H₃, J 4 Hz), 1.55–1.70 m (2H, C^{9} H), 4.05–4.20 m (1H, C^{8} H), 4.05 s (1H, NH), 6.60 d (2H, $C^{2,6}$ H, J 8 Hz), 6.65–6.75 m (1H, C^{4} H), 7.18 t (2H, $C^{3,5}$ H, J 8 Hz). ¹³C NMR spectrum, δ, ppm: 10.20 (C^{10}), 20.29 (C^{7}), 29.66 (C^{9}), 49.86 (C^{8}), 113.09 ($C^{2,6}$), 116.47 (C^{4}), 129.26 ($C^{3,5}$), 147.33 (C^{I}). Mass spectrum, m/z (I_{rel} , %): 149 (19) [M]⁺, 134 (11), 121 (13), 120 (100), 103 (5), 93 (7), 77 (10).

1-Phenylpyrrolidine (22). Yield 98%, colorless oily liquid, bp 105–106°C (5 mmHg) {89–90°C (1 mmHg) [19]}. ¹H NMR spectrum, δ, ppm: 2.00–2.05 m (4H, $C^{2',3'}$ H), 3.31 t (4H, $C^{I',4'}$ H), 6.60 d (2H, $C^{2,6}$ H, J 8 Hz), 6.66–6.74 m (1H, C^4 H), 7.26 t (2H, $C^{3,5}$ H, J 8 Hz). ¹³C NMR spectrum, δ, ppm: 25.47 ($C^{I',2'}$), 47.57 ($C^{3',4'}$), 111.63 ($C^{2,6}$), 115.35 (C^4), 129.14 ($C^{3,5}$), 148.04 (C^I). Mass spectrum, m/z (I_{rel} , %): 147 (82) [M]⁺, 146 (100), 119 (10), 106 (23), 91 70), 77 (31).

N-Benzylaniline (23). Yield 80%, light-yellow oily liquid, bp 140–142°C (6 mmHg) {125°C (3 mmHg) [11]}. ¹H NMR spectrum, δ, ppm: 4.45 s (2H, C^7H_2), 4.80 s (1H, NH), 6.50–6.80 m (2H, $C^{2.6}$ H), 6.85–6.95 m (1H, C^4 H), 7.20–7.40 m (2H, $C^{3.5}$ H), 7.45–7.50 m (5H, $C^{2'-6'}$ H). ¹³C NMR spectrum, δ, ppm: 48.40 (C^7), 112.81 ($C^{2.6}$), 117.68 (C^4), 127.37 (C^4), 127.65 ($C^{3.5}$), 129.42 ($C^{2'.6'}$), 128.79 ($C^{3'.5'}$), 139.61 ($C^{1'}$), 148.30 (C^{1}). Mass spectrum, m/z (I_{rel} , %): 184 (14) [M]⁺, 183 (58), 106 (18), 91 (100), 77 (15), 65 (17).

N,N-Dibenzylaniline (27). Yield 20%, mp 59–60°C (60°C [20]). Mass spectrum, m/z (I_{rel} , %): 273 (63) $[M]^+$, 196 (17), 182 (18), 91 (100), 77 (20), 65 (13).

N-Benzyl-3-chloroaniline (24). Yield 74%, light-yellow oily liquid, bp 150–152°C (2 mmHg) {182–186°C (5 mmHg) [21]}. ¹H NMR spectrum, δ, ppm: 4.07 s (1H, NH), 4.36 s (2H, C^7H_2), 6.72 s (1H, C^2H), 6.50–7.20 m (3H, $C^{4-6}H$), 7.20–7.50 m (5H, $C^{2'-6'}H$).

¹³C NMR spectrum, δ, ppm: 48.30 (C^7), 111.64 ($C^{2.6}$), 117.87 (C^4), 127.59 (C^4), 127.62 ($C^{2',6'}$), 128.76 ($C^{3.5}$), 130.26 (C^5), 135.02 (C^3), 137.86 (C^1), 148.72 (C^1). Mass spectrum, m/z (I_{rel} , %): 217 (40) [M]⁺, 182 (5), 140 (10), 91 (100), 77 (5).

N,*N*-**Dibenzyl-3-chloroaniline** (28). Yield 24%, yellow oily liquid, bp 165–167°C (0.02 mmHg). 1 H NMR spectrum, δ, ppm: 4.70 s (2H, C7,8 H), 6.72 s (1H, C2 H), 6.50–7.20 m (1H, $^{C4-6}$ H), 7.20–7.50 m (5H, $^{C2-6',2"-6''}$ H). 13 C NMR spectrum, δ, ppm: 47.89 (C7,8), 110.69 (C6), 112.26 (C2), 116.71 (C4), 127.53 ($^{C4',4"}$), 127.72 ($^{C2',2",6',6"}$), 128.76 ($^{C3',3",5',5"}$), 129.04 (C5), 135.18 (C3), 138.44 ($^{C1',1"}$), 148.72 (C1). Mass spectrum, $^{m/z}$ (I _{rel}, %): 309 (25) [M]⁺, 307 (65), 230 (15), 216 (30), 111 (10), 91 (100), 77 (5).

N-Benzyl-2-methoxyaniline (25). Yield 62%, yellow oily liquid, bp 155–157°C (2 mmHg) {183–187°C (5 mmHg) [21]}. ¹H NMR spectrum, δ, ppm: 3.90 s (3H, OCH₃), 4.41 s (2H, C⁷H₂), 4.62 s (1H, NH), 6.55–6.75 m (1H, C⁶H), 6.75–6.92 m (3H, C³–5H), 7.20–7.50 m (5H, C^{2′-6′}H). ¹³C NMR spectrum, δ, ppm: 48.16 (C⁷), 55.51 (OCH₃), 109.51 (C³), 110.25 (C⁶), 116.80 (C⁴), 121.38 (C⁵), 127.62 (C^{2′,6′}), 127.67 (C^{3′,5′}), 127.72 (C^{4′}), 138.17 (C¹), 139.65 (C^{1′}), 146.92 (C²). Mass spectrum, m/z ($I_{\rm rel}$, %): 213 (75) [M]⁺, 198 (23), 120 (10), 91 (100), 77 (13).

N,*N*-**Dibenzyl-2-methoxyaniline** (29). Yield 34%, yellow oily liquid, bp 144–146°C (0.02 mmHg). 1 H NMR spectrum, δ, ppm: 3.90 s (3H, OCH₃), 4.41 s (2H, $C^{7.8}$ H₂), 6.55–6.70 m (4H, C^{3-6} H), 7.20–7.50 m (5H, $C^{2-6',2''-6''}$ H). 13 C NMR spectrum, δ, ppm: 48.41 ($C^{7.8}$), 55.51 (OCH₃), 109.51 (C^{3}), 110.53 (C^{6}), 116.80 (C^{4}), 121.53 (C^{5}), 127.66 ($C^{2',2'',6',6''}$), 127.75 ($C^{4',4''}$), 128.87 ($C^{3',3'',5',5''}$), 138.17 (C^{1}), 139.65 ($C^{1',1''}$), 146.92 (C^{2}). Mass spectrum, m/z (I_{rel} , %): 303 (100) [M]⁺, 213 (10), 180 (7), 91 (88), 77 (6).

N-Benzyl-2-chloroaniline (26). Yield 78%, yellow oily liquid, bp 150–152°C (2 mmHg) {162–163°C (4 mmHg) [22]}. ¹H NMR spectrum, δ, ppm: 4.00 s (1H, NH), 4.47 s (2H, C^7 H₂), 6.50–7.80 m (4H, C^{3-6} H), 6.50–7.80 m (5H, $C^{2'-6'}$ H). ¹³C NMR spectrum, δ, ppm: 47.97 (C^7), 111.82 (C^6), 117.68 (C^3), 119.30 ($C^{2,4}$), 127.39 (C^4), 127.46 ($C^{2',6'}$), 127.91 (C^5), 128.81 ($C^{3',5'}$), 138.72 ($C^{1'}$), 143.78 (C^{1}). Mass spectrum, *m/z* (I_{rel} , %): 217 (35) [M]⁺, 180 (5), 140 (10), 91 (100), 77 (10).

N,*N*-**Dibenzyl-2-chloroaniline** (30). Yield 20%, yellow oily liquid, bp 160–162°C (0.02 mmHg). ¹H

NMR spectrum, δ , ppm: 4.47 s (2H, $C^{7,8}H$), 6.50–7.80 m (4H, $C^{3-6}H$), 6.50–7.80 m (5H, $C^{2'-6',2''-6''}H$). ¹³C NMR spectrum, δ , ppm: 48.20 ($C^{7,8}$), 111.82 (C^6), 117.68 (C^3), 119.30 ($C^{2,4}$), 127.39 ($C^{4',4''}$), 127.46 ($C^{2',2'',6',6''}$), 127.91 (C^5), 129.21 ($C^{3',3'',5',5''}$), 134.55 ($C^{I',I''}$), 143.78 (C^I). Mass spectrum, m/z (I_{rel} , %): 309 (25) [M]⁺, 307 (68), 230 (10), 216 (5), 180 (7), 91 (100), 77 (3).

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REFERENCES

- Lawrence, S.A., Amines: Synthesis, Properties and Applications, Cambridge: Cambridge University Press, 2004.
- Van der Vlugt, J.I., Chem. Soc. Rev., 2010, vol. 39, p. 2302.
- 3. Salvatore, R.N., Yoon, C.H., and Jung, K.W., *Tetrahedron*, 2001, vol. 57, p. 7785.
- 4. Bähn, S., Imm, S., Neubert, L., Zhang, M., Neumann, H., and Beller, M., Chem. Cat. Chem., 2011, vol. 3, p. 1853.
- Guillena, G., Ramyn, D.J., and Yus, M., Chem. Rev., 2009, vol. 110, p. 1611.
- 6. Naskar, S. and Bhattacharjee, M., *Tetrahedron Lett.*, 2007, vol. 48, p. 3367.
- 7. Bamoniri, A., Zolfigol, M.A., Mirjalili, B.F., and Fallah, F., *Russ. J. Org. Chem.*, 2007, vol. 43, p. 1393. doi 10.1134/S1070428007090229

- 8. Khusnutdinov, R.I., Bayguzina, A.R., and Aminov, R.I., *Russ. J. Org. Chem.*, 2013, vol. 49, p. 1447. doi 10.1134/S1070428013100072
- 9. Thomas, D.G., Billman, J.H., and Davis, C.E., *J. Am. Chem. Soc.*, 1946, vol. 68, p. 895. doi 10.1021/ja01209a058
- 10. Slocum, D.W. and Jennings, C.A., *J. Org. Chem.*, 1976, vol. 41, p. 3653.
- 11. Kim, S., Oh, Ch.H., Ko, J.S., Ahn, K.H., Kim, Y.J., *J* . *Org. Chem.*, 1985, vol. 50, p. 1927. doi 10.1021/jo00211a028
- 12. Petyunin, P.A., Shklyaev, V.S., and Konshin, M.E., *Zh. Prikl. Khim.*, 1960, vol. 33, p. 1428.
- 13. Shulgin, A.T. and Baker, A.W., *J. Org. Chem.*, 1963, vol. 28, p. 2468.
- 14. Watanabe, Y., Suzuki, N., Tsuji, Y., Shim, S.C., and Mitsudo, T., *Bull. Chem. Soc. Jpn.*, 1982, vol. 55, p. 1116.
- 15. Barbero, M., Degani, I., Dughera, S., and Fochi, R., *Synthesis*, 2003, vol. 5, p. 742.
- 16. Meyers, A.I. and Hellring, S., *Tetrahedron Lett.*, 1981, vol. 22, p. 5119.
- 17. Loevenich, J., Utsch, H., Moldrick, P., and Schaefer, E., *Ber. Dtsch. Chem. Ges. B*, 1929, vol. 62, p. 3084.
- 18. Botta, M., De Angelis, F., and Nicoletti, R., *Synthesis*, 1977, p. 722. doi 10.1055/s-1977-24553
- 19. Khusnutdinov, R.I., Bayguzina, A.R., Aminov, R.I., Asylbaeva, R.S., and Dzhemilev, U.M., *Arkivoc*, 2014, vol. v, p. 341.
- 20. Zhang, Y., Qi, X., Cui, X., Shi, F., and Deng, Y., *Tetrahedron Lett.*, 2011, vol. 52, p. 1334. doi 10.1016/j.tetlet.2011.01.059
- Hörlein, U., Chem. Ber., 1954, vol. 87, p. 463. doi 10.1002/cber.19540870404
- Stavrovskaya, V.I., Zh. Obshch. Khim., 1954, vol. 24, p. 1038.