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Synthesis of 2-oxo- and 2-thioxo-5-(benzofuran-2-yl)tetrahydropyrimidines

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New 5,6-dicyanobenzofurans bearing a 2-(thi)oxotetrahydropyrimidine moiety at the 2-position were synthesized from 2-(2-dimethyl-aminovinyl)-5,6-dicyanobenzofurans, benzaldehydes and (thio)urea in AcOH.

Recently, the area of heterocyclic chemistry dealing with the syntheses of substituted tetrahydropyrimidin-2-ones and their thio analogues was growing rapidly; these syntheses are based on the three-component Biginelli reaction¹ catalysed both by acids and bases.^{2,3} The resulting 6-aryl-2-(thi)oxotetrahydropyrimidines show various biological activities, *e.g.*, they can act as calcium channel modulators and be employed for the therapy of cardio-vascular diseases.⁴ Compounds of benzofuran series capable of uric acid liberation, such as amiodarone – a third class anti-arhythmic compound, benzbromarone and benziodarone,⁵ are also widely used in practical medicine. Thus, combining benzofuran and pyrimidine moieties in the same molecule creates promising opportunities for development of new pharmaceuticals.

Here, we describe a synthesis of new substituted 5,6-dicyanobenzofurans bearing a dihydropyrimidin-2-one(thione) moiety



Scheme 1

at the 2-position. Syntheses of similar compounds based on amino enones available from the corresponding R-carboxymethyl derivatives are known,^{6,7} but data on the use of catalysts reported in those publications are contradictory.

In our experiments, substituted 2-(2-dimethylaminovinyl)-5,6-dicyanobenzofurans are used for the first time as the substrate analogous to amino enones, with glacial acetic acid being used as the catalyst and solvent (Scheme 1).

The starting compounds 1a,b were obtained by heating of 3-aroyl-5,6-dicyano-2-methylbenzofurans⁸ in excess of dimethylformamide dimethylacetal.⁹ Tetrahydropyrimidin-2-ones(thiones) 4a-e were synthesized in 50–75% yield by heating (thio)urea 2 and aromatic aldehydes 3a,b for 8–18 h with equimolar amounts of compounds 1 in glacial acetic acid. The process occurs selectively to give one regioisomer. The reaction with urea proceeds more quickly than that with thiourea and the yields of the corresponding derivatives are generally higher than those of the thioxo analogues, since the longer heating results in greater resinification.

The structures of the products **4** were confirmed by IR and NMR spectroscopy and mass spectrometry.[†] ¹H NMR spectra

[†] IR spectra were measured on a Perkin-Elmer RX-1 spectrometer in the range of 700–4000 cm⁻¹ using suspensions of substances in Nujol. Mass spectra were obtained using a FINNIGAN MAT.INCOS 50 mass spectrometer; the ionization energy was 70 eV. NMR spectra were recorded on a Bruker DRX-500 instrument at 30 °C for solutions in DMSO-*d*₆. Signals of residual protons of the solvent in ¹H NMR spectra ($\delta_{\rm H}$ 2.50) or the signal of DMSO-*d*₆ in ¹³C spectra ($\delta_{\rm C}$ 39.5) were used as references for chemical shift measurements.

The starting benzofurans were synthesized according to procedure published elsewhere,⁸ whereas aminovinyl compounds 1a,b were obtained by another reported method.⁹

General procedure for the synthesis of compounds 4a-e. A mixture of corresponding compounds 1-3 (0.01 mol each) was refluxed in glacial acetic acid (20 ml) for 8–18 h at 100 °C, cooled and diluted with water (20 ml); the precipitate formed was filtered off and dissolved in dichloromethane. The solution was washed with water; flash chromatography on silica gel was carried out, the solvent was evaporated, and the precipitate was filtered off and washed with ethanol.

3-(4-Methylbenzoyl)-2-(2-oxo-4-phenyl-1,2,3,4-tetrahydropyrimidin-5-yl)-1-benzofuran-5,6-dicarbonitrile **4a**: yield 71%, mp 208–210°C. IR (ν/cm⁻¹): 3287 (N–H), 2234 (C≡N), 1656, 1640 (C=O), 1286 (C–O–C). ¹H NMR, δ: 2.40 (s, 3H, Me), 5.42 (d, 1H, H-4', J 3.0 Hz), 7.16–7.28 (m, 8H), 7.43 (d, 2H, H-2", H-6", J 8.1 Hz), 7.72 (s, 1H, H-7), contain signals of all protons of the benzofuran and pyrimidine moieties; furthermore, pyrimidine ring protons are manifested by vicinal coupling constants, and the ${}^{4}J_{\rm NH^{1}-NH^{3}}$ long-range constant can sometimes be observed for urea protons.

The structure of compound 4a was ultimately proved by X-ray diffraction (Figure 1).[‡] A unit cell contains a molecule of compound 4a and two disordered THF molecules. The benzofuran

7.82 (d, 1H, H-3', J 3.0 Hz), 8.49 (s, 1H, H-4), 9.41 (br. s, 1H, H-1'). ^{13}C NMR, δ : 21.22, 54.07, 100.34, 108.98, 109.67, 112.50, 116.07, 116.18, 117.22, 126.06, 126.34 (2 C), 127.78, 128.55 (2 C), 129.22 (2 C), 129.31 (2 C), 132.35, 134.14, 134.61, 142.56, 144.35, 150.77, 160.61, 188.80. MS, m/z (%): 458 [M⁺] (14.7), 339 [M⁺ – MeC₆H₄CO] (16.6), 119 [MeC₆H₄CO⁺] (100), 91 [MeC₆H₅⁺] (61.5). Found (%): C, 73.22; H, 4.01; N, 12.15. Calc. for C₂₈H₁₈N₄O₃ (%): C, 73.35; H, 3.96; N, 12.22.

2-[4-(4-Methoxyphenyl)-2-oxo-1,2,3,4-tetrahydropyrimidin-5-yl]-3-(4-methylbenzoyl)-1-benzofuran-5,6-dicarbonitrile **4b**: yield 75%, mp 227–229 °C. IR (ν /cm⁻¹): 3328, 3200 (N–H), 2236 (C≡N), 1656, 1644 (C=O), 1282 (C–O–C). ¹H NMR, δ : 2.40 (s, 3H, Me), 3.65 (s, 3H, OMe), 5.36 (d, 1H, H-4', J 2.9 Hz), 6.67 (d, 2H, J 8.6 Hz), 7.04 (d, 2 H, J 8.6 Hz), 7.25 (s, 1H, H-6'), 7.26 (d, 2 H, H-3", H-5", J 8.2 Hz) 7.37 (d, 2 H, H-2", H-6", J 8.2 Hz), 7.57 (s, 1H, H-7), 7.56 (d, 1H, H-3', J 2.9 Hz), 8.39 (s, 1H, H-4), 9.37 (s, 1H, H-1'). MS, m/z (%): 488 [M⁺] (3.8), 309 (23.3), 285 (21.7), 134 (54.8), 119 [MeC₆H₄CO⁺] (95), 91 [MeC₆H₅] (100). Found (%): C, 71.10; H, 3.97; N, 11.52. Calc. for C₂₉H₂₀N₄O₄ (%): C, 71.30; H, 4.13; N, 11.47.

3-(4-Methylbenzoyl)-2-(4-phenyl-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl)-1-benzofuran-5,6-dicarbonitrile **4c**: yield 62%, mp 198–201 °C. IR (ν/cm⁻¹): 3285 (N–H), 2238 (C≡N), 1651, 1637 (C=O), 1299 (C–O–C). ¹H NMR, δ: 2.40 (s, 3 H, Me), 5.41 (s, 1H, H-4'), 7.14–7.28 (m, 8 H), 7.44 (d, 2 H, H-2", H-6", J 8.1 Hz), 7.73 (s, 1H, H-7), 8.54 (s, 1H, H-4), 9.66 (br. s, 1H, H-3'), 10.60 (br. s, 1H, H-1'). MS, *m*/z (%): 474 [M⁺] (1.5), 119 [MeC₆H₄CO⁺] (100), 91 [MeC₆H₅⁺] (56). Found (%): C, 70.42; H, 3.57; N, 11.72. Calc. for C₂₈H₁₈N₄O₂S (%): C, 70.87; H, 3.82; N, 11.81.

3-(4-Methylbenzoyl)-2-[4-(4-methoxyphenyl)-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl]-1-benzofuran-5,6-dicarbonitrile **4d**: yield 53%, mp 258–260°C. IR (ν/cm⁻¹): 3387, 3146 (N–H), 2236 (C≡N), 1650, 1632 (C=O), 1273 (C–O–C). ¹H NMR, δ: 2.40 (s, 3 H, Me), 3.63 (s, 3 H, OMe), 5.32 (d, 1H, H-4', J 3.3 Hz), 6.71 (d, 2H, J 8.7 Hz), 7.03 (d, 2H, J 8.7 Hz), 7.19 (d, 1H, H-6', J 5.7 Hz), 7.26 (d, 2H, H-3'', H-5'', J 8.1 Hz), 7.41 (d, 2H, H-2'', H-6'', J 8.1 Hz), 7.71 (s, 1H, H-7), 8.53 (s, 1H, H-4), 9.59 (dd, 1H, H-3', J 3.3 Hz, J 1.7 Hz), 10.54 (dd, 1H, H-1', J 5.7 Hz, J 1.7 Hz). MS, *m*/z (%): 504 [M⁺] (3.5), 119 [MeC₆H₄CO⁺] (100), 91 [MeC₆H₅⁺] (58). Found (%): C, 68.86; H. 3.84; N, 11.02. Calc. for C₂₉H₂₀N₄O₃S (%): C, 69.03, H 4.00, N, 11.10.

3-(4-Methoxybenzoyl)-2-[4-(4-methoxyphenyl)-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl]-1-benzofuran-5,6-dicarbonitrile **4e**: yield 58%, mp 262–264°C, IR (ν/cm⁻¹): 3156 (N–H), 2236 (C≡N), 1650, 1632 (C=O), 1263 (C–O–C). ¹H NMR, δ: 3.65 (s, 3 H, OMe), 3.88 (s, 3 H, OMe), 5.33 (br. s, 1H, H-4'), 6.65 (d, 2 H, J 8.2 Hz), 6.91 (d, 2 H, J 8.5 Hz), 7.02 (d, 2 H, J 8.2 Hz), 7.19 (br. s, 1H, H-6'), 7.46 (d, 2 H, J 8.5 Hz), 7.63 (s, 1H, H-7), 8.41 (s, 1H, H-4), 9.47 (br. s, 1H, H-3'), 10.43 (br. s, 1H, H-1'). Found (%): C, 66.62; H, 3.74; N, 10.62. Calc. for C₂₉H₂₀N₄O₄S (%): C, 67.00; H, 3.85; N, 10.79.

^{*} *X-ray diffraction study.* Single crystals were obtained as very thin needles characterized by extremely weak reflection. Reflections for compound **4a** were collected with a SMART APEX2 diffractometer [λ (MoK α) = 0.71073 Å, graphite monochromator, ω -scans] at 100 K. The structure was solved by direct methods and refined by the full-matrix least-squares procedure against F^2 in anisotropic approximation (except for solvated THF molecules which were both disordered over two positions and refined isotropically). Hydrogen atoms of NH groups were found in the difference Fourier synthesis, normalized at the standard X-ray value (0.9 Å) and refined within the riding model. All the other hydrogen atoms were placed in geometrically calculated positions and refined within the riding model.

4a·2(C₄H₈O): C₂₈H₂₀N₄O₃·2(C₄H₈O), monoclinic, space group $P2_1/n$, a = 9.331(2), b = 18.448(4) and c = 18.476(4) Å, $\beta = 103.077(3)^\circ$, V = 3098.0(10) Å³, Z = 4, M = 604.69, $d_{calc} = 1.296$ g cm⁻³, $\mu = 0.087$ mm⁻¹, F(000) = 1280, $wR_2 = 0.2418$, GOF = 1.068 for 6076 independent reflections with $2\theta < 52^\circ$, $R_1 = 0.0963$ for 2098 reflections with $I > 2\sigma(I)$.



Figure 1 General view of molecule **4a**. The atoms are presented as thermal vibration ellipsoids at 50% probability level; the second component of the disordered 4-MeC₆H₄ moiety is shown by dashed lines and dots (hydrogen atoms are not shown).

ring is planar; the pyrimidine ring also has a planar structure and is coplanar to benzofuran, which results in π -conjugation between the rings. This is indicated by the shortening of the C(2)–C(16) bond [1.439(7) Å] and by the C(16)=C(17) bond length [1.334(7) Å] typical of conjugated systems (the average statistical lengths of similar bonds are 1.478 and 1.330 Å, respectively¹⁰).

In the crystal structure, one of the amide protons is involved in H-bonding [N(3)–H(3)···O(3), H···O 1.88 Å, N···O 2.782(6) Å, \angle NHO 176°], which leads to formation of centrosymmetrical dimers, whereas the N(4)–H(4) fragment is bound to one of the THF molecules.

In conclusion, new benzofurans containing 2-positioned tetrahydropyrimidin-2-one(thione) moiety have been selectively synthesized from substituted 5,6-dicyanobenzofurans with acetic acid as the catalyst.

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CCDC 805524 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk/data_request/cif. For details, see 'Notice to Authors', *Mendeleev Commun.*, Issue 1, 2011.