

Synthesis of Bisarylmethyl-Substituted Pyrimidines and Quinolines

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Abstract—The condensation of 2-chloro-8-methylquinoline-3-carbaldehyde, 2-chloro- and 2-chloro-7,8,9,10-tetrahydrobenzo[*h*]quinoline-3-carbaldehydes, 2,4-dioxo-1,2,3,4-tetrahydropyrimidine-5-carbaldehyde, 6-methyl-2,4-dioxo-1,2,3,4-tetrahydropyrimidine-5-carbaldehyde, and 2-chloropyrido[1,2-*a*]pyrimidine-3-carbaldehyde with *N*-substituted anilines gave the corresponding diaryl(hetaryl)methanes.

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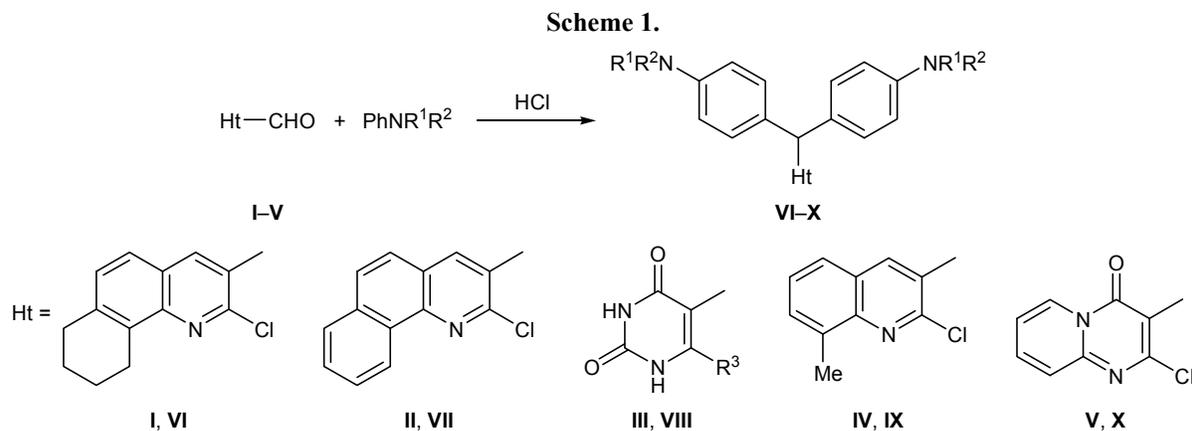
It is known that substituted triphenylmethanes (leuco bases of the corresponding triphenylmethane dyes) are potent chemotherapeutic and photodynamic agents [1]; they also exhibit nonlinear optical properties due to specific π -electron density distribution in their molecules [2].

Aryl and hetaryl triphenylmethane analogs containing five- and six-membered heterocycles and fused heterocyclic systems have recently been synthesized [3]. It should be noted that only a few examples of pyrimidine analogs of triphenylmethane and triphenylmethanol have been reported [4, 5]; some of them were tested as antiestrogens (aromatase inhibitors) [5].

In order to find new potential biologically active compounds in the series of diaryl(hetaryl)methanes,

a number of heterocyclic aldehydes were brought into condensation with *N*-substituted anilines in the presence of various catalysts, in particular H_2SO_4 , polyphosphoric acid, HCl , and ZnCl_2 . The best results were obtained when a suspension of aldehyde **I–V** and the corresponding aniline in 30% aqueous HCl was heated under reflux for 12 h (Scheme 1). As a result, diaryl(hetaryl)methanes **VI–X** were isolated in good yields. However, 6-styryl-2,4-dioxo-1,2,3,4-tetrahydropyrimidine-5-carbaldehyde [6] failed to react with *N,N*-dimethyl- and *N,N*-diethylanilines under the above conditions, and the initial aldehyde was recovered from the reaction mixture.

The structure of compounds **VI–X** was confirmed by elemental analyses and IR and ^1H NMR spectra.



III, $\text{R}^3 = \text{H}$ (**a**), Me (**b**); **VI, VII, IX**, $\text{R}^1 = \text{R}^2 = \text{Me}$ (**a**), Et (**b**); **VIII**, $\text{R}^3 = \text{H}$, $\text{R}^1 = \text{R}^2 = \text{Me}$ (**a**), Et (**b**);
 $\text{R}^1 = \text{R}^3 = \text{H}$, $\text{R}^2 = i\text{-Bu}$ (**c**); $\text{R}^3 = \text{H}$; $\text{R}^1 = \text{R}^2 = \text{Me}$ (**d**), Et (**e**); **X**, $\text{R}^1 = \text{R}^2 = \text{Me}$.

Diaryl(hetaryl)methanes VI–X were tested for antibacterial activity against *S. aureus* 209, p 1, *Sh. dysenteriae Flexneri* 6858, and *E. coli* 0-55. Compound VIIa turned out to be weakly active against Gram-negative bacteria, compounds VIIIb, VIIIc, and VIIIe were weakly active against Gram-positive bacteria, and compound VIIIa displayed activity against all the examined bacterial strains. Pyrimidines VIIIc and VIIIe at a dose of 150 mg/kg inhibited by 30–40% the growth of sarcoma 37 in mice.

EXPERIMENTAL

The IR spectra were recorded in mineral oil on a Nicolet Avatar 330 spectrometer. The ^1H NMR spectra were measured on a Varian Mercury-300 instrument at 300 MHz using tetramethylsilane as internal reference. Silufol UV-254 were used for thin-layer chromatography; spots were visualized by treatment with iodine vapor.

2-Chloro-7,8,9,10-tetrahydrobenzo[*h*]quinoline-3-carbaldehyde (I). Vilsmeier reagent prepared from 18.3 g (0.25 mol) of dimethylformamide and 107.5 g (0.7 mol) of phosphoryl chloride was cooled to -5 to 0°C , 18.9 g (0.1 mol) of *N*-(1,2,3,4-tetrahydronaphthalen-5-yl)acetamide [7] was added in portions, and the mixture was allowed to warm up to room temperature, heated for 2 h under reflux, and evaporated under reduced pressure. The residue was poured onto 0.5 kg of ice, the mixture was left to stand for 3 h in the cold, and the precipitate was filtered off and dried. Yield 15.6 g (64%), mp 160 – 162°C (from EtOH), R_f 0.70 (EtOH–dichloroethane, 1:10). IR spectrum, ν , cm^{-1} : 1682 (C=O), 1608 s (C=C_{arom}). ^1H NMR spectrum (DMSO- d_6 -CCl₄, 1:3), δ , ppm: 1.85–2.00 m (4H, CH₂CH₂CH₂CH₂), 2.96 br.t and 3.24 br.t (2H each, CH₂CH₂CH₂CH₂, $J = 5.4$ Hz), 7.34 d and 7.79 d (1H each, 5-H, 6-H, $J = 8.4$ Hz), 8.69 s (1H, 4-H), 10.45 s (1H, CHO). Found, %: N 5.86. C₁₄H₁₂ClNO. Calculated, %: N 5.70.

Diaryl(hetaryl)methanes VI–X (general procedure). A mixture of 0.01 mol of aldehyde I–V, 2.4 g (0.02 mol) of 30% aqueous HCl, and 0.03 mol of the corresponding *N*-substituted aniline was stirred and heated for 12 h under reflux. The resulting suspension was made alkaline by treatment with aqueous ammonia, excess *N*-substituted aniline was removed by steam distillation, and the precipitate was filtered off, washed with water, and dried.

4,4'-(2-Chloro-7,8,9,10-tetrahydrobenzo[*h*]quinolin-3-yl)methylene]bis(*N,N*-dimethylaniline) (VIa)

was synthesized from aldehyde I and *N,N*-dimethylaniline. Yield 68%, mp 255 – 256°C (from 2-ethoxyethanol), R_f 0.80 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum: ν 1611 cm^{-1} , s (C=C_{arom}). ^1H NMR spectrum (DMSO- d_6 -CCl₄, 1:3), δ , ppm: 1.84–1.96 m (4H, 8-H, 9-H), 2.90 m (2H, 7-H or 10-H), 2.93 s (12H, NCH₃), 3.21 m (2H, 10-H or 7-H), 5.71 s (1H, CH), 6.56–6.67 m and 6.83–6.89 m (4H each, C₆H₄), 7.17 d and 7.40 d (1H each, 5-H, 6-H, $J = 8.4$ Hz), 7.51 s (1H, 4-H). Found, %: N 8.74. C₃₀H₃₂ClN₃. Calculated, %: N 8.94.

4,4'-(2-Chloro-7,8,9,10-tetrahydrobenzo[*h*]quinolin-3-yl)methylene]bis(*N,N*-diethylaniline) (VIIb) was synthesized from aldehyde I and *N,N*-diethylaniline. Yield 72%, mp 201 – 202°C (from DMF–2-ethoxyethanol), R_f 0.73 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum: ν 1610 cm^{-1} , s (C=C_{arom}). ^1H NMR spectrum (DMSO- d_6 -CCl₄, 1:3), δ , ppm: 1.15 t (12H, CH₂CH₃, $J = 7.0$ Hz), 1.84–1.96 m (4H, 8-H, 9-H), 2.90 m and 3.21 m (2H each, 7-H, 10-H), 3.33 q (8H, CH₂CH₃, $J = 7.0$ Hz), 5.66 s (1H, CH), 6.49–6.56 m and 6.78–6.85 m (4H each, C₆H₄), 7.16 d and 7.42 d (1H each, 5-H, 6-H, $J = 8.4$ Hz), 7.55 s (1H, 4-H). Found, %: N 7.82. C₃₄H₄₀ClN₃. Calculated, %: N 7.99.

4,4'-(2-Chlorobenzo[*h*]quinolin-3-yl)methylene]bis(*N,N*-dimethylaniline) (VIIa) was synthesized from aldehyde II [8] and *N,N*-dimethylaniline. Yield 60%, mp 221 – 222°C (from 2-ethoxyethanol), R_f 0.64 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum: ν 1610 cm^{-1} , s (C=C_{arom}). ^1H NMR spectrum (DMSO- d_6 -CCl₄, 1:3), δ , ppm: 2.94 s (12H, CH₃), 5.79 s (1H, CH), 6.63 m and 6.90 m (4H each, C₆H₄), 7.62 d and 7.80 d (1H each, 5-H, 6-H, $J = 8.8$ Hz), 7.63–7.72 m (2H, 8-H, 9-H), 7.71 s (1H, 4-H), 7.90 m (1H, 7-H), 9.10 m (1H, 10-H). Found, %: N 8.86. C₃₀H₂₈ClN₃. Calculated, %: N 9.02.

4,4'-(2-Chlorobenzo[*h*]quinolin-3-yl)methylene]bis(*N,N*-diethylaniline) (VIIb) was synthesized from aldehyde II and *N,N*-diethylaniline. Yield 66%, mp 224 – 225°C (from 2-ethoxyethanol), R_f 0.61 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum: ν 1610 cm^{-1} , s (C=C_{arom}). ^1H NMR spectrum (DMSO- d_6 -CCl₄, 1:3), δ , ppm: 1.16 t (12H, CH₂CH₃, $J = 7.0$ Hz), 3.34 q (8H, CH₂CH₃, $J = 7.0$ Hz), 5.74 s (1H, CH), 6.54 m and 6.87 m (4H each, C₆H₄), 7.63 d and 7.80 d (1H each, 5-H, 6-H, $J = 8.9$ Hz), 7.64–7.72 m (2H, 8-H, 9-H), 7.75 s (1H, 4-H), 7.89 m (1H, 7-H), 9.11 m (1H, 10-H). Found, %: N 8.26. C₃₄H₃₆ClN₃. Calculated, %: N 8.05.

5-{Bis[4-(dimethylamino)phenyl]methyl}pyrimidine-2,4(1*H*,3*H*)-dione (VIIIa) was synthesized from

aldehyde **IIIa** [9] and *N,N*-dimethylaniline. Yield 76%, mp 299–301°C (from EtOH), R_f 0.53 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum, ν , cm⁻¹: 3343 br (OH), 1707, 1672 (C=O), 1607 (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 2.84 s (12H, CH₃), 5.05 s (1H, CH), 6.48 d (1H, 6-H, J = 5.8 Hz), 6.64 m and 6.89 m (4H each, C₆H₄), 10.52 d.d (1H, 1-H, J = 5.8, 1.5 Hz), 11.02 d (1H, 3-H, J = 1.5 Hz). Found, %: N 15.26. C₂₁H₂₄N₄O₂. Calculated, %: N 15.37.

5-{Bis[4-(diethylamino)phenyl]methyl}pyrimidine-2,4(1H,3H)-dione (VIIIb) was synthesized from aldehyde **IIIa** and *N,N*-diethylaniline. Yield 85%, mp 280–281°C (from EtOH), R_f 0.43 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum, ν , cm⁻¹: 3232 br, 3155 br (NH), 1713, 1665 (C=O), 1607 (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.13 t (12H, CH₂CH₃, J = 7.0 Hz), 3.31 q (8H, CH₂CH₃, J = 7.0 Hz), 5.04 s (1H, CH), 6.46–6.58 m (5H, 6-H, *m*-H), 6.85 m (4H, *o*-H), 10.11 br.d (1H, 1-H, J = 5.7 Hz), 10.74 br.s (1H, 3-H). Found, %: N 12.97. C₂₅H₃₂N₄O₂. Calculated, %: N 13.32.

5-{Bis[4-(isobutylamino)phenyl]methyl}pyrimidine-2,4(1H,3H)-dione (VIIIc) was synthesized from aldehyde **IIIa** and *N*-isobutylaniline. Yield 65%, mp 237–238°C (from EtOH), R_f 0.58 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum, ν , cm⁻¹: 3287, 3170, 3140 (NH), 1709, 1653 (C=O), 1607 (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 0.96 d (12H, CH₃, J = 6.6 Hz), 1.86 non (2H, CHCH₂, J = 6.6 Hz), 2.80 d (4H, CH₂, J = 6.6 Hz), 4.96 br.s (2H, NHCH₂), 4.99 s (1H, 5-CH), 6.43 m (4H, *m*-H), 6.44 d (1H, 6-H, J = 5.6 Hz), 6.76 m (4H, *o*-H), 10.38 d.d (1H, 1-H, J = 5.6, 1.8 Hz), 10.72 d (1H, 3-H, J = 1.8 Hz). Found, %: N 13.48. C₂₅H₃₂N₄O₂. Calculated, %: N 13.32.

5-{Bis[4-(dimethylamino)phenyl]methyl}-6-methylpyrimidine-2,4(1H,3H)-dione (VIIId) was synthesized from aldehyde **IIIb** [10] and *N,N*-dimethylaniline. Yield 74%, mp 184–186°C (from EtOH), R_f 0.30 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum, ν , cm⁻¹: 1718 s (C=O), 1615 s (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.79 s (3H, 6-CH₃), 2.85 s (12H, NCH₃), 5.42 s (1H, CH), 6.64 m (4H, *m*-H), 6.91 m (4H, *o*-H), 10.55 br.s (1H, 1-H), 10.90 br.s (1H, 3-H). Found, %: N 14.58. C₂₂H₂₆N₄O₂. Calculated, %: N 14.80.

5-{Bis[4-(diethylamino)phenyl]methyl}-6-methylpyrimidine-2,4(1H,3H)-dione (VIIIe) was synthesized from aldehyde **IIIb** and *N,N*-diethylaniline. Yield 79%, mp 286–287°C (from EtOH), R_f 0.57 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum, ν , cm⁻¹: 1715 s (C=O), 1612 s (C=C_{arom}). ¹H NMR spectrum

(DMSO-*d*₆), δ , ppm: 1.06 t (12H, CH₂CH₃, J = 7.0 Hz), 1.79 s (3H, CH₃), 3.28 q (8H, CH₂CH₃, J = 7.0 Hz), 5.37 s (1H, CH), 6.56 m and 6.88 m (4H each, C₆H₄), 10.54 d (1H, 1-H, J = 1.5 Hz), 10.90 d (1H, 3-H, J = 1.5 Hz). Found, %: N 12.68. C₂₆H₃₄N₄O₂. Calculated, %: N 12.89.

4,4'-(2-Chloro-8-methylquinolin-3-yl)methylene]bis(*N,N*-dimethylaniline) (IXa) was synthesized from aldehyde **IV** [11] and *N,N*-dimethylaniline. Yield 65%, mp 205–206°C (from EtOH), R_f 0.44 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum: ν 1612 cm⁻¹, s (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆–CCl₄, 1:3), δ , ppm: 2.75 s (3H, 8-CH₃), 2.93 s (12H, NCH₃), 5.72 s (1H, CH), 6.60 m and 6.87 m (4H each, C₆H₄), 7.36 d.d (1H, J = 7.9, 7.3 Hz) and 7.48–7.55 m (2H) (5-H, 6-H, 7-H), 7.59 s (1H, 4-H). Found, %: N 9.50. C₂₇H₂₈ClN₃. Calculated, %: N 9.77.

4,4'-(2-Chloro-8-methylquinolin-3-yl)methylene]bis(*N,N*-diethylaniline) (IXb) was synthesized from aldehyde **IV** and *N,N*-diethylaniline. Yield 73%, mp 177–178°C (from EtOH), R_f 0.48 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum: ν 1612 cm⁻¹, s (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆–CCl₄, 1:3), δ , ppm: 1.15 t (12H, CH₃CH₂, J = 7.0 Hz), 2.71 s (3H, 8-CH₃), 3.33 q (8H, CH₂CH₃, J = 7.0 Hz), 5.68 s (1H, CH), 6.53 m and 6.83 m (4H each, C₆H₄); 7.36 d.d (1H, J = 8.0, 7.1 Hz), 7.49 m (1H), and 7.55 m (1H) (5-H, 6-H, 7-H), 7.63 s (1H, 4-H). Found, %: N 8.50. C₃₁H₃₆ClN₃. Calculated, %: N 8.64.

2-Chloro-3-{bis[4-(dimethylamino)phenyl]methyl}-4H-pyrido[1,2-*a*]pyrimidin-4-one (X) was synthesized from aldehyde **V** [12] and *N,N*-dimethylaniline. Yield 68%, mp 199–200°C (from EtOH), R_f 0.47 (*i*-BuOH–H₂O–AcOH, 2:2:1). IR spectrum, ν , cm⁻¹: 1782 (C=O), 1608 s (C=C_{arom}). ¹H NMR spectrum (DMSO-*d*₆–CCl₄, 1:3), δ , ppm: 2.90 s (12H, CH₃), 5.74 s (1H, CH), 6.57 m and 7.09 m (4H each, C₆H₄), 7.27 d.d.d (1H, 7-H, J = 7.3, 6.7, 1.3 Hz), 7.57 d.d.d (1H, 9-H, J = 8.8, 1.3, 0.8 Hz), 7.88 d.d.d (1H, 8-H, J = 8.8, 6.7, 1.6 Hz), 8.89 d.d.d (1H, 6-H, J = 7.3, 1.6, 0.8 Hz). Found, %: N 12.65. C₂₅H₂₅ClN₄O. Calculated, %: N 12.94.

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REFERENCES

- Mibu, N., Yokomizo, K., Uyeda, M., and Sumoto, K., *Chem. Pharm. Bull.*, 2003, vol. 51, p. 1325; Al-Qawasme, R.A., Lee, Y., Cao, M.Y., Gu, X., Vassilakos, A., Wright, J.A., and Young, A., *Bioorg. Med. Chem. Lett.*, 2004, vol. 14, p. 347; Wainwright, M., *Photosensitizers in*

- Biomedicine*, Chichester: Wiley, 2009, p. 81; Mibu, N., Yokomizo, K., Miyata, T., and Sumoto, K., *J. Heterocycl. Chem.*, 2010, vol. 47, p. 1434; Lavrenov, S.N., Luzikov, Y.N., Bykov, E.E., Reznikova, M.I., Stepanova, E.V., Glazunova, V.A., Volodina, Y.L., Tatarsky, V.V., Jr., Shtil, A.A., and Preobrazhenskaya, M.N., *Bioorg. Med. Chem.*, 2010, vol. 18, p. 6905.
- Noack, A., Schroder, A., and Hartmann, H., *Angew. Chem., Int. Ed.*, 2001, vol. 40, p. 3008.
 - Nair, V., Thomas, S., Mathew, S.C., and Abhilash, K.G., *Tetrahedron*, 2006, vol. 62, p. 6731; Mibu, N., Yokomizo, K., Uyeda, M., and Sumoto, K., *Chem. Pharm. Bull.*, 2005, vol. 53, p. 1171; Parai, M.K., Panda, G., Chaturvedi, V., Manju, Y.K., and Sinha, S., *Bioorg. Med. Chem. Lett.*, 2008, vol. 18, p. 289.
 - Johnson, T.B., *J. Am. Chem. Soc.*, 1929, vol. 51, p. 1274; Budesinsky, Z., Vavrina, J., Longsaddl, L., and Holubek, J., *Collect. Czech. Chem. Commun.*, 1980, vol. 45, p. 539.
 - Taylor, H., Jones, C.D., Davenport, J.D., Hirsh, K.S., Kress, T.J., and Weaver, D., *J. Med. Chem.*, 1987, vol. 30, p. 1359; Jones, C.D., Winter, M.A., Hirsh, K.S., Stamm, N., Taylor, H.M., Holden, H.E., Davenport, J.D., Krumkalns, E.V., and Suhr, R.G., *J. Med. Chem.*, 1990, vol. 33, p. 416.
 - Arutyunyan, A.A., Saakyan, A.G., Mamyan, S.S., and Melik-Ogandzhanyan, R.G., *Khim. Zh. Arm.*, 2008, vol. 61, p. 104.
 - Beilsteins Handbuch der organischen Chemie*, vol. H12, p. 1197.
 - Roopan, S.M., Nawaz Khan, F., Subashini, R., Hathwar, V.R., and Ng, S.W., *Acta Crystallogr., Sect. E*, 2009, vol. 65, p. 2711.
 - Wiley, R.H. and Yamamoto, Y., *J. Org. Chem.*, 1960, vol. 25, p. 1906.
 - Wiley, P.F. and MacKellar, F.A., *J. Org. Chem.*, 1976, vol. 41, p. 1858.
 - Meth-Cohn, O., Narine, B., and Tarnowski, B., *J. Chem. Soc., Perkin Trans. 1*, 1981, p. 1520.
 - Abass, M., Ismail, M.M., Abdel-Monem, W.R., and Mayas, A.S., *Chem. Pap.*, 2010, vol. 64, p. 72.