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Synthesis of the C₁₄ Chromanyl Moiety of Natural α-Tocopherol (Vitamin E)

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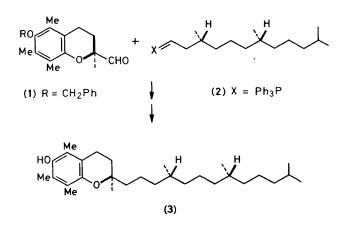
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The synthesis of the C_{14} (2S) aldehyde (1) from the C_5 ketone (7) via the C_{16} adduct (10) is reported.

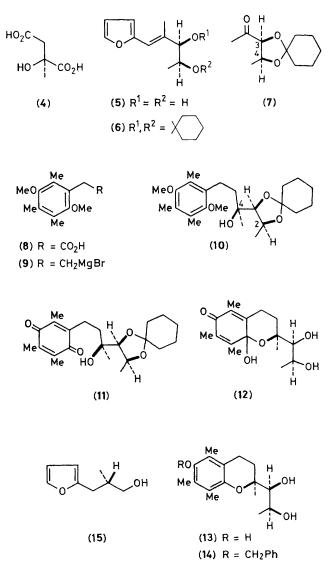
One approach to the total synthesis of natural α -tocopherol (vitamin E) (3) is based on chiral synthons obtained by microbial transformations as starting materials. By this method, the two C_{14} and C_{15} key building blocks (1) and (2) have been prepared from (+)-(S)-citramalic acid¹ (4) and from a variety of C_4 and C_5 intermediates,²⁻⁴ respectively. The synthesis of the C_{14} chromanyl aldehyde (1) from (4) involves the direct incorporation of the chirality present in (4) through C-C bond formation between a suitable C₅ unit derived from (4) and a C_{9} aromatic fragment, followed by ring closure.

We present now a synthesis of enantiomerically pure (1) based on the use as optically active starting material of the C₅ ketone (7), obtained from the diol (5), prepared, in turn, from α -methyl- β -(2-furyl)acrolein and fermenting bakers' yeast.⁵

To this end, the (2S,3R)-diol (5), prepared in 15–20% yield in fermenting bakers' yeast from α -methyl- β -(2-furyl)acrolein, †



† Typically, 60 g of α-methyl-β-(2-furyl)acrolein in 301 of tap water, containing 2.5 kg of commercial bakers' yeast, 0.8 kg of D-glucose, and 0.1 kg of Na₂HPO₄, under stirring at room temperature gave within 6–8 h, after extraction with ethyl acetate and SiO₂ column chromatography, 40–45 g of (15), containing some α-methyl-β-(2-furyl)prop-2-en-1-ol, and ca. 10 g of (5) of (5).



was quantitatively converted (cyclohexanone, toluene-psulphonic acid, benzene) into (6). Ozonolysis of (6) at -30 °C in CH₂Cl₂, followed by treatment with 1 mol. equiv. of Ph₃P, gave the ketone (7), b.p. 70 °C at 20 mmHg, $[\alpha]_{D}^{20} - 51^{\circ}$ (c 1, CHCl₃), in ca. 70% yield. The (3R, 4S) absolute configuration of (7) is based on its conversion into N-benzoyl-2,3,6trideoxy-3-C-methyl-3-amino-L-arabinohexose.⁶ The C₉-C₂ Grignard reagent (9), prepared from the acid (8)⁷ by standard methods, added in tetrahydrofuran at -30 °C to the ketone (7) to give⁸ in ca. 75% yield the C₁₆ adduct (10), $[\alpha]_D^{20} - 20^\circ$ (c 1, $CHCl_3$). The conversion of (10) into the chromanyl moiety (14) was achieved by known methods.^{1,9} Thus, compound (10), upon oxidation with Ce(NH₄)₂(NO₃)₆ in MeCNwater (1:1), yielded the quinone (11), $[\alpha]_{D}^{20} - 20^{\circ}$ (c 1, CHCl₃), the ¹H n.m.r. spectrum (90 MHz; CDCl₃) of which showed a signal due to the C-4 methyl group at δ 1.50. Upon acid treatment, compound (11) gave rise to (12), $[\alpha]_{D}^{20} - 80^{\circ}$ (c 1, CHCl₃), in ca. 70% overall yield from (10). Hydrogenation of (12) over 10% Pd-C at room temperature gave quantitatively the vicinal diol (13), which, following benzylation to (14) (PhCH₂Br; Me₂NCHO; K₂CO₃) and HIO₄ oxidation in dry tetrahydrofuran (85%), yielded the required C_{14} aldehyde (1), $[\alpha]_{D}^{20}$ 12.3° (c 5, MeOH), in good agreement with the literature¹ value. In this last step, the two chiral centres of (7) which induced the chirality at C-4 in (10) are destroyed.

The significance of the present result is further supported by the fact that compound (5) is accompanied by the chiral alcohol (15) in the yeast fermentation of α -methyl- β -(2-furyl)acrolein. The alcohol (15) obtained was optically pure since on ozonolysis and oxidative work-up it gave (S)-3-methyl- γ butyrolactone, $[\alpha]_{D}^{20} - 24.5^{\circ}$ (c 4, MeOH) (lit.³ - 24.7°), and,

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on chain elongation through a procedure similar to that previously described⁴ for the (2S)-2-methyl-5-phenylpent-4-en-1ol, it gave optically pure (2). In this way, the two key intermediates (1) and (2) for the synthesis of enantiomerically pure (3) can be prepared from the optically active products (5) and (15), easily obtained from α -methyl- β -(2-furyl)acrolein in a single microbial transformation using inexpensive commercial bakers' yeast, in 15–20 and 40–60% yields, respectively.

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