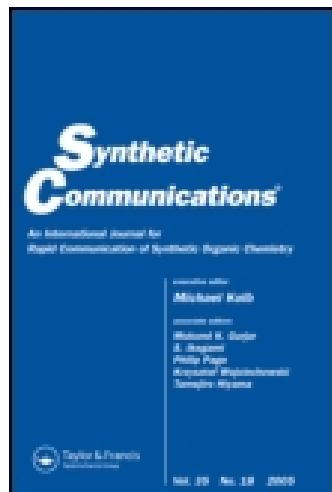




Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lsec20>

Biginelli-Like Cyclocondensation Reaction: Efficient Synthesis of 4,6-Diarylpyrimidin-2(1H)-one Under Solvent-Free Conditions

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Published online: 11 Jul 2011.

To cite this article: Yisi Dai, Minxiang Cao, Mengjie Zhuang, Sheng Xia, Shujiang Tu & Liangce Rong (2011) Biginelli-Like Cyclocondensation Reaction: Efficient Synthesis of 4,6-Diarylpyrimidin-2(1H)-one Under Solvent-Free Conditions, *Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry*, 41:20, 3039-3044, DOI: [10.1080/00397911.2010.516619](https://doi.org/10.1080/00397911.2010.516619)

To link to this article: <http://dx.doi.org/10.1080/00397911.2010.516619>

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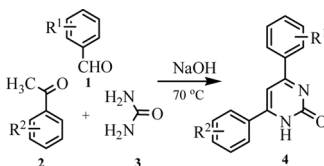
BIGINELLI-LIKE CYCLOCONDENSATION REACTION: EFFICIENT SYNTHESIS OF 4,6-DIARYLPYRIMIDIN-2(1H)-ONE UNDER SOLVENT-FREE CONDITIONS

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GRAPHICAL ABSTRACT



Abstract An efficient and facile synthesis of 4,6-diarylpyrimidin-2(1H)-one via a Biginelli-like reaction of aromatic aldehydes, aromatic ketones, and urea in the presence of NaOH under solvent-free conditions using a heating method has been developed. Compared with the classical reaction conditions, this new synthetic method has the advantages of excellent yields, shorter reaction time, and mild reaction conditions.

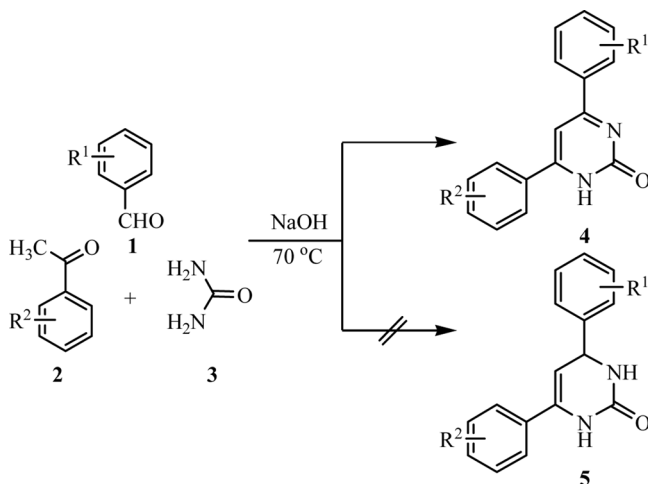
Keywords Biginelli-like reaction; 4,6-diarylpyrimidin-2(1H)-one; green chemistry; solvent-free synthesis

INTRODUCTION

The *N*-heterocyclic compounds, such as hydropyrimidinones, are very useful intermediates for the development of molecules of pharmaceutical or biological interest. These compounds show variable biological activities, such as antimalarial, anticonvulsant, anesthetic, antioxidant, antibacterial, and antiparasitic properties.^[1–3] In addition, hydropyrimidinones have also attracted much attention in previous years because of their large is range of biological activities, including as calcium channel blockers.^[4,5] So, synthesis of these compounds is an important topic in organic chemistry.

Received July 16, 2009.

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Scheme 1. Reaction of aromatic aldehydes, aromatic ketones, and urea.

The Biginelli reaction is one of the most important reactions to synthesize 3,4-dihydropyrimidin-2-(1*H*)-one in a one-pot reaction of aldehyde, α -ketoester, and urea. In past years, many methods have been reported to improve this reaction because of its harsh reaction conditions, such as long reaction times, high temperatures, low yields, expensive catalysts, and harmful and difficult-to-handle reagents, especially on a large scale.^[6] At the same time, many Biginelli-like reactions also have been reported and many pyridine-like derivatives have been synthesized.^[7]

Because it avoids using an organic solvent in the reactions, solvent-free synthesis is regarded as one of the important methods in current green chemistry. Herein, we report a Biginelli-like cyclocondensation reaction to prepare 4,6-diarylpyrimidin-2(1*H*)-one derivatives from aromatic aldehydes, aromatic ketones, and urea using NaOH as catalyst under solvent-free conditions (Scheme 1).

The reaction process could be depicted as follows: the aromatic aldehydes **1**, aromatic ketones **2**, and urea **3** were put in a mortar to blend together in the presence

Table 1. Synthesis of 4,6-diarylpyrimidin-2(1*H*)-one

Entry	R ¹	R ²	Product	Yields (%)
1	H	4-CH ₃ O	4a	90
2	H	3-Cl	4b	85
3	H	4-Cl	4c	91
4	3,4-(CH ₃) ₂	4-CH ₃	4d	80
5	4-CH ₃	H	4e	85
6	4-F	H	4f	88
7	4-Br	H	4g	92
8	2-Cl	H	4h	82
9	3-Cl	H	4i	86
10	4-Cl	H	4j	92
11	2,4-Cl ₂	H	4k	90
12	3,4-Cl ₂	H	4l	88

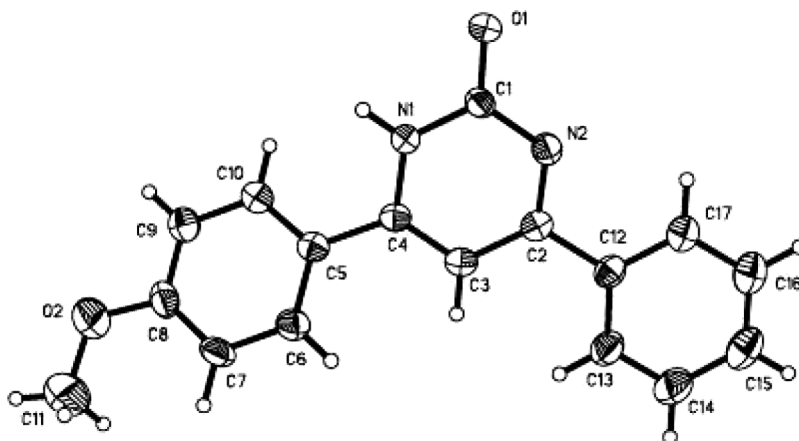


Figure 1. Structure of compound **4a**.

of a small amount of NaOH (0.2 g) as catalyst, and then the mixture was placed into a round flask and left at 70 °C to react. The reactions could be finished in about 10–15 min and the product, 4,6-diarylpyrimidin-2(1H)-one **4**, could be gained with excellent yields. The results of reactions are listed in Table 1. Previously,^[7a] 3,4-dihydropyrimidin-2(1H)-ones **5** has been prepared from the reaction of aromatic aldehydes, aromatic ketones, and urea using a Lewis acid ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ -TMSCl) as catalyst; however, in our research, we did not find 3,4-dihydropyrimidin-2(1H)-ones **5**. Instead, 4,6-diarylpyrimidin-2(1H)-one **4** was obtained with good yields. We think the different catalysts resulted in the different results of the reaction.

From Table 1, we saw that the all reactions were carried out smoothly, and we also found that the nature of substitute groups in the aldehydes or ketones has no significant effect on this reaction. The structures of **4** were characterized by ^1H NMR, infrared (IR) and high-resolution mass spectrometry (HRMS). In particular, the measured data of HRMS was in accord with the corresponding calculated data, so the deduced structures of products were right. The structure of **4a** was additionally confirmed by x-ray diffraction analysis. The crystal structure of **4a** is shown in Fig. 1.

In conclusion, we have successfully developed an efficient and facile method to prepare a variety of 4,6-diarylpyrimidin-2(1H)-one derivatives via the multicomponent reactions of different aromatic aldehydes, aromatic ketones, and urea under solvent-free conditions. Because no toxic organic solvent was used and the basic catalyst (NaOH) was employed, this method presents an efficient and facile route for the synthesis of these compounds.

EXPERIMENTAL

Melting points were uncorrected. IR spectra were recorded on a Tensor 27 spectrometer in KBr. ^1H NMR spectra were obtained in dimethylsulfoxide ($\text{DMSO}-d_6$) solution with Me $_4\text{Si}$ as internal standard using a Bruker-400 spectrometer. HRMS spectra were obtained with a Bruker microOTOF-Q 134 instrument. X-ray diffractions were recorded on a Siemens P4 or Simart-1000 diffractometer.

General Procedure for the Syntheses of 4,6-Diarylpyrimidin-2(1*H*)-one Derivatives

Aromatic aldehyde **1** (2 mmol), aromatic ketone **2** (2 mmol), urea **3** (3 mmol), and NaOH (0.2 g) were added to a mortar. The mixture was blended together and then shifted into a round flask. The reagent was heated about 70 °C under atmospheric conditions, and the reaction could be finished within 10–15 min. The reaction mixture was poured into water, and the product was filtered, dried, and recrystallized from 95% ethanol.

Data

6-(4-Methoxyphenyl)-4-phenylpyrimidin-2(1*H*)-one (4a). Mp 252–253 °C; IR (KBr, ν , cm^{-1}): 3279, 3095, 2998, 2834, 1618, 1514, 1459, 1420, 1394, 1339, 1262, 1174, 1038, 991, 821, 776 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO-}d_6$) (δ , ppm): 3.86 (3H, s, OCH_3), 7.10 (2H, d, $J=8.8$ Hz, ArH), 7.50 (1H, s, $\text{C}^5\text{-H}$), 7.54–7.59 (3H, m, ArH), 8.14–8.19 (4H, m, ArH), 11.96 (1H, s, NH); HRMS m/z calculated for $\text{C}_{17}\text{H}_{14}\text{N}_2\text{O}_2$ [$\text{M} + \text{H}$]: 279.1134; found: 279.1144.

6-(3-Chlorophenyl)-4-phenylpyrimidin-2(1*H*)-one (4b). Mp 219–220 °C; IR (KBr, ν , cm^{-1}): 3305, 3069, 2886, 1625, 1496, 1456, 1398, 1329, 1197, 995, 858, 764, 682 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO-}d_6$) (δ , ppm): 7.50–7.57 (5H, m, ArH), 7.62–7.65 (1H, m, ArH), 8.18–8.19 (3H, m, ArH), 8.30 (1H, s, $\text{C}^5\text{-H}$), 12.16 (1H, s, NH); HRMS m/z calculated for $\text{C}_{16}\text{H}_{11}\text{ClN}_2\text{O}$ [$\text{M} + \text{H}$]: 283.0638; found: 283.0643.

6-(4-Chlorophenyl)-4-phenylpyrimidin-2(1*H*)-one (4c). Mp 192–194 °C; IR (KBr, ν , cm^{-1}): 3324, 3060, 1653, 1556, 1489, 1358, 1175, 1091, 1013, 998, 820, 700, 694, 648 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO-}d_6$) (δ , ppm): 7.20 (1H, s, $\text{C}^5\text{-H}$), 7.46–7.48 (3H, m, ArH), 7.52 (2H, dd, $J=1.6$ Hz, $J=1.6$ Hz, ArH), 8.10–8.12 (2H, m, ArH), 8.17 (2H, d, $J=8.0$ Hz, ArH), 12.12 (1H, s, NH); HRMS m/z calculated for $\text{C}_{16}\text{H}_{11}\text{ClN}_2\text{O}$ [$\text{M} + \text{H}$]: 283.0638; found: 283.0636.

4-(3,4-Dimethylphenyl)-6-*p*-tolylpyrimidin-2(1*H*)-one (4d). Mp > 300 °C; IR (KBr, ν , cm^{-1}): 3330, 3100, 2918, 1623, 1538, 1506, 1455, 1418, 1392, 1338, 1250, 1185, 990, 915, 808, 715, 596 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO-}d_6$) (δ , ppm): 2.31 (3H, s, CH_3), 2.33 (3H, s, CH_3), 2.40 (3H, s, CH_3), 7.32 (1H, d, $J=8.0$ Hz, ArH), 7.37 (2H, d, $J=8.0$ Hz, ArH), 7.47 (1H, m, ArH), 7.88–7.90 (1H, m, ArH), 7.97 (1H, s, $\text{C}^5\text{-H}$), 8.06–8.08 (2H, m, ArH), 11.98 (1H, s, NH); HRMS m/z calculated for $\text{C}_{19}\text{H}_{18}\text{N}_2\text{O}$ [$\text{M} + \text{H}$]: 291.1497; found: 291.1496.

6-Phenyl-4-*p*-tolylpyrimidin-2(1*H*)-one (4e). Mp 271–273 °C (lit.^[6e] 287–290 °C); IR (KBr, ν , cm^{-1}): 3298, 3097, 3007, 2901, 1616, 1460, 1393, 1339, 1183, 994, 919, 809, 774, 689, 596 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO-}d_6$) (δ , ppm): 2.41 (3H, s, CH_3), 7.37 (2H, d, $J=8.4$ Hz, ArH), 7.54–7.60 (4H, m, ArH, $\text{C}^5\text{-H}$), 8.09 (2H, s, ArH), 8.16 (2H, br, ArH), 12.08 (1H, s, NH); HRMS m/z calculated for $\text{C}_{17}\text{H}_{14}\text{N}_2\text{O}$ [$\text{M} + \text{H}$]: 263.1184; found: 263.1192.

4-(4-Fluorophenyl)-6-phenylpyrimidin-2(1*H*)-one (4f). Mp 159–161 °C; IR (KBr, ν , cm^{-1}): 3301, 3059, 2906, 1614, 1557, 1509, 1356, 1231, 1155, 993, 821, 766, 690 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO-}d_6$) (δ , ppm): 7.27 (2H, t, $J=8.4$ Hz,

$J=8.4$ Hz, ArH), 7.35 (1H, s, C⁵-H), 7.46–7.47 (3H, m, ArH), 8.15 (2H, d, $J=4.0$ Hz, ArH), 8.23 (2H, dd, $J=6.0$ Hz, $J=6.0$ Hz, ArH), 11.99 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₁FN₂O [M + H]: 267.0934; found: 267.0945.

4-(4-Bromophenyl)-6-phenylpyrimidin-2(1H)-one (4g). Mp 255–256 °C (lit.^[6e] 251–254 °C); IR (KBr, ν , cm⁻¹): 3401, 3058, 1647, 1610, 1488, 1447, 1072, 1010, 821, 768, 697 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) (δ , ppm): 7.60 (4H, t, $J=8.0$ Hz, $J=8.0$ Hz, ArH), 7.66 (1H, d, $J=8.0$ Hz, ArH), 7.71 (1H, m, ArH), 8.18 (3H, d, $J=6.8$ Hz, ArH), 8.27 (1H, s, C⁵-H), 12.17 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₁BrN₂O [M + H]: 327.0133; found: 327.0137.

4-(2-Chlorophenyl)-6-phenylpyrimidin-2(1H)-one (4h). Mp 220–223 °C; IR (KBr, ν , cm⁻¹): 3342, 3030, 2926, 1659, 1577, 1538, 1455, 1396, 1337, 1053, 994, 758, 689 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) (δ , ppm): 7.24 (1H, s, C⁵-H), 7.54–7.57 (5H, m, ArH), 7.64 (2H, d, $J=7.2$ Hz, ArH), 8.14 (2H, d, $J=7.2$ Hz, ArH), 12.24 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₁ClN₂O [M + H]: 283.0638; found: 283.0643.

4-(3-Chlorophenyl)-6-phenylpyrimidin-2(1H)-one (4i). Mp 210–212 °C; IR (KBr, ν , cm⁻¹): 3214, 3078, 2888, 1625, 1576, 1497, 1339, 1330, 1198, 995, 765, 683, 593 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) (δ , ppm): 7.56–7.62 (4H, m, ArH), 7.66 (2H, d, $J=8.0$ Hz, ArH), 8.20 (3H, s, ArH), 8.30 (1H, s, C⁵-H), 12.17 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₁ClN₂O [M + H]: 283.0638; found: 283.0645.

4-(4-Chlorophenyl)-6-phenylpyrimidin-2(1H)-one (4j). Mp 242–244 °C (lit.^[6e] 258–260 °C); IR (KBr, ν , cm⁻¹): 3311, 3105, 3059, 2900, 1614, 1538, 1438, 1337, 1178, 1090, 994, 821, 764, 685 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) (δ , ppm): 7.37 (1H, s, C⁵-H), 7.47–7.48 (3H, m, ArH), 7.51 (2H, d, $J=8.0$ Hz, ArH), 8.14 (2H, d, $J=8.4$ Hz, ArH), 8.20 (2H, d, $J=8.4$ Hz, ArH), 12.18 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₁ClN₂O [M + H]: 283.0638; found: 283.0644.

4-(2,4-Dichlorophenyl)-6-phenylpyrimidin-2(1H)-one (4k). Mp 223–225 °C; IR (KBr, ν , cm⁻¹): 3315, 3027, 2932, 1664, 1587, 1448, 1393, 1142, 1098, 995, 777, 689 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) (δ , ppm): 7.55–7.69 (6H, m, ArH), 7.85 (1H, s, C⁵-H), 8.12–8.13 (2H, br, ArH), 12.26 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₀Cl₂N₂O [M + H]: 317.0248; found: 317.0248.

4-(3,4-Dichlorophenyl)-6-phenylpyrimidin-2(1H)-one (4l). Mp > 290 °C; IR (KBr, ν , cm⁻¹): 3280, 3094, 2893, 1616, 1578, 1536, 1497, 1475, 1442, 1398, 1331, 1269, 1229, 1202, 1163, 1133, 1097, 1026, 997, 909, 855, 832, 767, 684, 645, 595 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) (δ , ppm): 7.56–7.59 (3H, m, ArH), 7.80 (1H, s, C⁵-H), 7.84 (1H, d, $J=8.8$ Hz, ArH), 8.19–8.24 (3H, m, ArH), 8.51 (1H, d, $J=2.0$ Hz, ArH), 12.16 (1H, s, NH); HRMS m/z calculated for C₁₆H₁₀Cl₂N₂O [M + H]: 317.0248; found: 317.0243.

X-ray crystallography for 4a. Empirical formula C₁₇H₁₄N₂O₂, $F_w=278.30$, $T=298(2)$ K, orthorhombic, $a=7.3081(17)$ Å, $b=13.462(3)$ Å, $c=27.986(7)$ Å, $\alpha=90^\circ$, $\beta=90^\circ$, $\gamma=90^\circ$, $V=2753.2(11)$ Å³, $Z=8$, $D_c=1.343$ Mg/m³, $\lambda(\text{MoK}\alpha)=0.71073$ Å, $\mu=0.090$ mm⁻¹, $F(000)=1168$, $1.46^\circ < \theta < 25.01^\circ$, $R=0.0671$, $wR=0.2225$, $S=1.034$, largest diff. peak and hole: 0.633 and -0.345 e Å⁻³.

ACKNOWLEDGMENTS

We thank the Natural Science Foundation of Jiangsu Education Department (No. 08KJB150017) and the Foundation of Xuzhou Normal University (No. 10XLS02) for financial support.

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