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### A NOVEL ROUTE TO SPIRO PHOSPHORUS-HETEROCYCLE VIA LAWESSON'S REAGENT

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## A NOVEL ROUTE TO SPIRO PHOSPHORUS-HETEROCYCLE VIA LAWESSON'S REAGENT

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### ABSTRACT

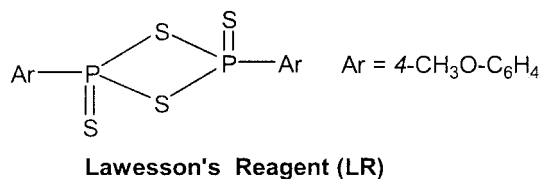
2-Aryl-spiro-[5,5]-1,3,2-dioxaphosphorinane-2-sulfides were prepared by the cyclization reaction of Lawesson's reagent with pentaerythritol and monobenzalpenterythritol, respectively.

*Key Words:* Cyclization; Lawesson's reagent; Pentaerythritol; Spiro phosphorus heterocycle

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In recent years, our research interest has been focused on the development of new synthetic methodology centered around biologically active phosphorus heterocycles,<sup>1</sup> because functionalized phosphorus heterocycles and their derivatives are bioactive substances of great interest with various properties.<sup>2-6</sup> In the preceding papers,<sup>1</sup> we disclosed a methodology to prepare 5-membered and 6-membered phosphorus heterocycles with biological activity via cyclization of Lawesson's reagent, 2,4-bis(4-methoxyphenyl)-1,3,2,4-dithiadiphosphetane-2,4-disulfide, with bifunctional substrates. The fact that this cyclization reaction is readily applicable to other bifunctional compounds to form different kinds of phosphorus-heterocycles prompted us to develop a new method for synthesis of spiro phosphorus-heterocycles. The preparation of spiro crown ethers and rings are of importance since their complexes play an important role in biological systems and coordination chemistry.<sup>7</sup> We report here the synthesis of 2-aryl-spiro-[5,5]-1,3,2-dioxaphosphorinane-2-sulfides **3** and **4** starting from pentaerythritol **1** and monobenzalpenterythritol **2** using Lawesson's reagent.

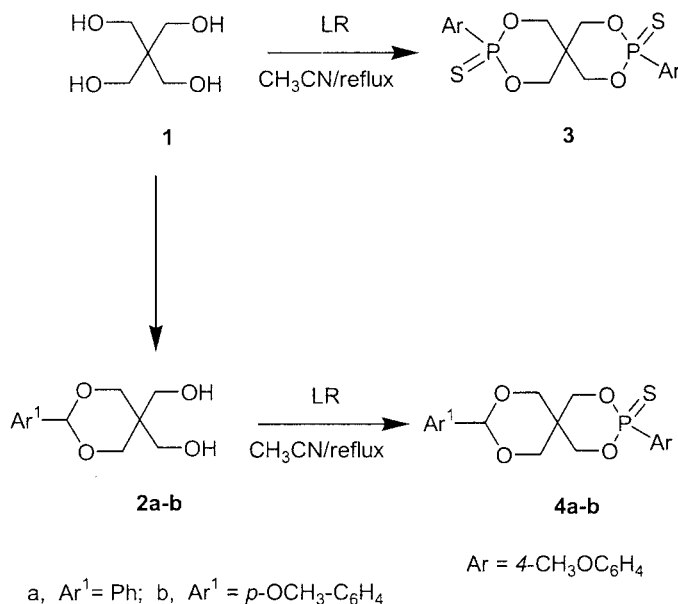


## RESULTS AND DISCUSSION

Lawesson's reagent reacted with pentaerythritol **1** in anhydrous acetonitrile as a solvent using 2.2:1 molar ratio at reflux under dry nitrogen for 6–8 h to afford the spiro heterocycles **3** in significant yields, as well as thioacetamide (m.p. 113°C) as a side product, as depicted in Scheme 1. Similarly, 1.1 molar equivalent of Lawesson's reagent was treated with monobenzalpenterythritol **2a–b** to provide the ring **4a–b**. All new compounds were identified satisfactorily by elemental analyses and spectral data (IR, NMR, and MS) (Scheme 1).

## EXPERIMENTAL

Melting points were determined with a model X<sub>4</sub> apparatus and are uncorrected. <sup>1</sup>H NMR spectra and <sup>31</sup>P NMR spectra were recorded on a

*Scheme 1.*

Varian XL-200 MHz spectrometer. Mass spectra were measured on a HP 5988A spectrometer. Elemental analyses were measured with a PE-2400 elementary analyzer. The IR spectra were measured by using a Shimadzu-408 instrument. Column chromatography was performed on silica gel II (10–40  $\mu$ , Hai Yang Chemical Factory of Qingdao). All solvents and materials were reagent grade and purified as required. Monobenzalpenterythritol **2a–b** were obtained from pentaerythritol and benzaldehyde in the usual way.<sup>8</sup> Lawesson's reagent was prepared in a yield of 75% according to published procedure.<sup>9</sup>

**General procedure for synthesis of 3 and 4a–b:** A three-necked flask equipped with a dropping funnel, stirrer, drying CaCl<sub>2</sub> tube and nitrogen gas inlet was charged with 10 ml of anhydrous acetonitrile and 1 mmol of pentaerythritol **1** or monobenzalpenterythritol **2a–b**. Then Lawesson's reagent (2.2 mmol, 1.1 mmol for preparation of **4a–b**) was added to the flask at room temperature. After complete addition, the stirred reaction mixture was refluxed under dry nitrogen for 6–8 h until no starting materials could be detected (TLC). Evaporation of the solvent followed by column chromatography on silica gel (column size: 4 cm in diameter  $\times$  15 cm long; 10 grams of silica gel, *R<sub>f</sub>* = 0.55) using light petroleum ether

(b.p. 40–60°C)-dry ethyl ether as eluent yielded the corresponding heterocycles **3** and **4a–b**. Yields were determined after separation on silicon gel column. The structures of the new compounds were confirmed by elemental analyses and spectral results.

**3**: m.p. 197–198°C; 0.169 g (yield 35.8%);  $^1\text{H}$  NMR  $\delta_{\text{H}}$  (DMSO- $d_6$ ): 3.84 (s, 6H,  $2 \times \text{CH}_3\text{O}$ ), 4.1–4.2 (m, 8H,  $4 \times \text{CH}_2$ ), 7.0–8.0 (m, 8H, aromatic protons, Ar-H).  $^{31}\text{P}$  NMR  $\delta_{\text{P}}$  (DMSO- $d_6$ ): 89.09. IR  $\nu$  (KBr,  $\text{cm}^{-1}$ ): 750 (P=S), 1050 (P-O-C). EI-MS (int. rel)  $m/z$  (%): 472 ( $\text{M}^+$ , 88). Anals. calcd for  $\text{C}_{19}\text{H}_{22}\text{O}_6\text{P}_2\text{S}_2$ : C, 48.31; H, 4.66. Found: C, 48.22; H, 4.48.

**4a**: m.p. 131–132°C; 0.127 g (yield 32.4%);  $^1\text{H}$  NMR  $\delta_{\text{H}}$  (DMSO- $d_6$ ): 3.87 (s, 3H,  $\text{CH}_3\text{O}$ ), 4.0–4.2 (m, 8H,  $4 \times \text{CH}_2$ ), 7.0–7.9 (m, 9H, aromatic protons, Ar-H).  $^{31}\text{P}$  NMR  $\delta_{\text{P}}$  (DMSO- $d_6$ ): 88.34. IR  $\nu$  (KBr,  $\text{cm}^{-1}$ ): 750 (P=S), 1060 (P-O-C). EI-MS (int. rel)  $m/z$  (%): 392 ( $\text{M}^+$ , 48). Anals. calcd for  $\text{C}_{19}\text{H}_{21}\text{O}_5\text{PS}$ : C, 58.16; H, 5.36. Found: C, 58.34; H, 5.12.

**4b**: m.p. 137–138°C; 0.162 g (yield 38.4%);  $^1\text{H}$  NMR  $\delta_{\text{H}}$  (DMSO- $d_6$ ): 3.89 (s, 6H,  $2 \times \text{CH}_3\text{O}$ ), 4.1–4.2 (m, 8H,  $4 \times \text{CH}_2$ ), 7.1–7.8 (m, 8H, aromatic protons, Ar-H). IR  $\nu$  (KBr,  $\text{cm}^{-1}$ ): 760 (P=S), 1060 (P-O-C). EI-MS (int. rel)  $m/z$  (%): 422 ( $\text{M}^+$ , 32). Anals. calcd for  $\text{C}_{20}\text{H}_{23}\text{O}_6\text{PS}$ : C, 56.87; H, 5.45. Found: C, 56.63; H, 5.61.

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