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New Efficient Catalysts for Enantioselective Transfer Hydrogenations

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Summary: New C₂-symmetrical diaminoferrocenyl derivatives 3 and 2-amino(sulfonamido)cyclohexanes 4 were found to be highly active ligands for the ruthenium catalyzed asymmetric transfer hydrogenation of ketones. Contrary to many existing catalytic systems, the ligands 3 show a high activity at 25 °C and operate even at -30 °C (up to 90 % ee). On the other hand, the slightly less active ligands 4 are very easily prepared and are highly enantioselective at 30 °C in HCOOH/Et₃N (up to 96 % ee). Copyright © 1996 Elsevier Science Ltd

The catalytic enantioselective reduction of ketones has been extensively studied during the last decade.¹ A especially useful method is the catalytic transfer hydrogenation² using *i*-PrOH³ or a HCOOH/Et₃N mix-ture⁴ as hydride source and a chiral ruthenium catalyst bearing ligands such as 1 or 2. In the course of our work on the preparation of new chiral C_2 -symmetrical ferrocenyl derivatives for asymmetric catalysis,⁵ we have now discovered that the ruthenium complexes of the diaminoferrocenyl derivatives of type 3 are a new class of highly efficient transfer hydrogenation catalysts operating even at -30 °C with high conversion and good enantioselectivity.



Herein we report our preliminary results with the new chiral ligand system as well as a highly enantioselective transfer hydrogenation using *N*-monosulfonylated 1,2-diaminocyclohexane derivatives of type **4**. The ferrocenes **3** were prepared from ferrocene in four steps (Scheme 1). Acylation of ferrocene (R¹COCl, CH₂Cl₂, AlCl₃, 0 °C to 25 °C, 4 h) provides ferrocenyldiketones (71 - 92 %) which were reduced enantioselectively with an oxazaborolidine catalyst (CBS-reduction) providing 1,1'-ferrocenyl diols in 90 % - 96 % yield and > 99 % *ee.*⁵ Acetylation of these diols (Ac₂O, pyridine, 25 °C, 12 h) gives quantitatively the corresponding diacetates **5** which undergo a substitution with an excess of primary amines (THF-H₂O, 25 °C, 12 h) furnishing the diaminoferrocenes **3a-e** (56 % - 93 %) with retention of configuration.⁶ Hydrogenolysis of **3e** (H₂, Pd(OH)₂ cat) leads to the unprotected diaminoferrocene derivative **3f** (96 %).⁷Monotosylation of **3f** (TosCl (1 equiv), CH₂Cl₂, 0 °C, 1 h) affords the aminosulfonamide **3g** in 36 % yield.⁸ The aminoferrocenes **3a-d,f,g** were tested in the catalytic reduction of acetophenone (**6**). Thus, a *i*-PrOH solution of the catalytic system prepared from the ferrocene derivatives **3a-d,f,g** (2 mol %) and



Scheme 1

[Ru(p-cymene)Cl₂]₂ (0.5 mol %) was treated at temperatures between -30 °C and 22 °C with a *i*-PrOH solution of acetophenone (6) in the presence of KOH (5 mol %) affording (R)-1-phenylethanol (7) with excellent conversions and enantioselectivities up to 80 % *ee* (Scheme 2 and Table 1).⁹ No decrease of the enantioselectivity is observed with time.



Similarly 1-acetylnaphthalene (8) is reduced to (R)-1-naphthyl-1-ethanol (9) with an enantioselectivity up to 90 % ee. Entries 1-6 of Table 1 compare the results of the different ligands 3a-d,f,g at 22 °C. All except the monosulfonamido derivative 3g are highly effective and high conversions are reached within a few minutes to a few hours at rt. The best enantioselectivities are obtained with 3a and 3b, both bearing an aryl substituent ($R^1 = Ph$ or o-Tol) and a N-methylamino substituent. The free diaminoferrocene derivative 3f shows a good catalytic activity but leads to a moderate enantioselectivity (52 % ee, see entry 5). The increase of the size of the aryl substituent R^1 (phenyl to 1-naphthyl) doubles the catalytic activity but lowers the enantioselectivity (71 % ee to 60 % ee, compare entries 1 and 3). Remarkably, the high activity of our catalytic system allows us to lower further the reaction temperature. Carrying out the acetophenone reduction at -14 °C with the ligand 3a increases the enantioselectivity from 71 % ee to 79 % ee. Lowering the temperature further to -30 °C (120 h) furnishes the alcohol 7 with 80 % ee (compare entries 1, 7 and 8). A similar temperature effect is observed with 1-acetylnaphthalene (8). By using the most selective ligands 3a and 3b, enantioselectivities of 78 % ee and 85 % ee were obtained at rt. Lowering the reaction

entry	ketone	ligand	T (°C)	reaction time (h)	conversion (%)[a]	ee (%)[a]
1	6	3a	22	0.5	98	71
2	6	3b	22	1.5	97	80
3	6	3c	22	0.25	98	60
4	6	3d	22	3	94	62
5	6	3f	22	1	98	52
6	6	3g	22	24	97	56
7	6	3a	-14	41	96	79
8	6	3a	-30	120	95	80
9	6	3g	30	120	42	83[b]
10	8	3a	22	1	99	78
11	8	3 a	-30	120	91	90
12	8	3b	22	1	97	85
13	8	3b	0	12	96	88

Table 1. Enantioselective transfer hydrogenation of ketones **6** and **8** in *i*-PrOH in the presence of 0.5 mol % of [Ru(*p*-cymene)Cl₂]₂ and 2 mol % of the aminoferrocenes **3a-d**, **f**, **g**.

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[a] Determined by GC analysis (Chirasil-DEX CB) [b] A HCOOH/Et₃N mixture was used.

Table 2. Enantioselective transfer hydrogenation of ketones 6, 8 and 12 in *i*-PrOH or HCOOH/Et₃N in the presence of 0.5 mol % of [Ru(*p*-cymene)Cl₂]₂ and 2 mol % of the aminosulfonamides 4a-d.

entry	ketone	ligand	T (°C)	reaction time (h)	conversion (%)[a]	<i>ee</i> (%)[a]
1	6	4a	22 (30)	24	97 (>99)	89 (94)
2	6	4b	22 (30)	24	96 (>99)	90 (89)
3	6	4c	22 (30)	24	97 (>99)	90 (95)
4	6	4d	22 (30)	24	96 (>99)	92 (96)
5	8	4d	22 (30)	36	99 (>99)	92 (96)
6	12	4d	22 (30)	65	60 (54)	23 (67)

[a] Determined by GC analysis (Chirasil-DEX CB). Conversions and % ee in HCOOH-Et₃N are indicated in parenthesis.

temperature with the ligand **3a** to -30 °C and for **3b** to 0 °C allows to increase the enantioselectivity respectively to 90 % *ee* and 88 % *ee* (compare entries 10 to 13). The high catalytic activity observed using the ligand **3a**, even allows the reduction of sterically hindered ketones. Thus *t*-butylphenylketone was reduced with high conversion (> 93 %) within 17 h at 22 °C leading to (S)-2,2-dimethyl-1-phenyl-1-propanol with 38 % *ee*. Interestingly, the less active transfer hydrogenation ligand the monosulfonamide **3g** reacts with high enantioselectivity by replacing the *i*-PrOH/KOH reaction medium by the solvent mixture HCOOH/Et₃N (5:2). Under these conditions, a moderate conversion is obtained (42 %, 120 h) but the enantioselectivity increases from 56 % *ee* to 83 % *ee* (compare entries 6 and 9). This enantioselectivity increase led us to prepare several monosulfonamides derived from the readily available (1R, 2R)-1,2diaminocyclohexane (10). Thus, treatment of commercially available 10 with various arylsulfonyl chlorides 11 (1 equiv, 0 °C, CH₂Cl₂) furnishes highly enantioselective transfer hydrogenation ligands 4a-d (Scheme 3 and Table 2).¹⁰



The reduction of acetophenone (6) proceeds smoothly with the aminosulfonamides 4a-d. High enantioselectivities (90-92 % ee) are obtained with 4a-d in i-PrOH. Switching to the HCOOH/Et₃N solvent system leads to a further increase of the enantioselectivity (94-96 % ee) and to almost quantitative conversions (see entries 1-4 of Table 2).¹¹ 1-Acetylnaphthalene (8) behaves as expected in the same way and furnishes the alcohol 9 in 96 % ee (see entry 5 of Table 2). Interestingly, the sterically hindered isopropylphenylketone 12 is reduced in 23 % ee using the i-PrOH/KOH system, whereas an enantioselectivity of 67 % ee is obtained with the HCOOH/Et₃N system.

In summary, we have reported two new classes of highly active transfer hydrogenation catalysts derived from C₂-symmetrical diaminoferrocene derivatives or readily available 2-(sulfonamido)cyclohexanes. Extension of this work is currently underway.

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- (8) 3a: ¹H NMR (CDCl₃, 300 MHz): δ 7.30-7.15 (m, 10H), 4.33-4.30 (m, 2H), 4.28 (s, 2H), 4.08-4.05 (m, 4H), 4.00-3.99 (m, 2H), 2.36 (s, 6H), 2.0 (s, br, 2H). [a]_D +56 (c 0.55, CHCl₃). **3b**: [a]_D -34 (c 2.00, CHCl3). 3c: $[\alpha]_D$ -5 (c 2.33, CHCl3). 3d: $[\alpha]_D$ -6 (c 1.73, CHCl3). 3e: $[\alpha]_D$ -98 (c 0.71, CHCl₃). **3f**: [*α*]_D +30 (*c* 2.39, CHCl₃). **3g**: [*α*]_D -32 (*c* 2.52, CHCl₃).
- (9) The experiments were carried out according to Noyori's procedure (ref. 2 h) using a 0.05 M solution of the ketone.
- (10) Compounds 4a-d were isolated as hydrochlorides. 4a: $[\alpha]_D$ +57 (c 3.50, MeOH). 4b: $[\alpha]_D$ +54 (c
- 0.98, MeOH). 4c: [α]_D +31 (c 1.26, MeOH). 4d: [α]_D +164 (c 0.98, MeOH).
 (11) A mixture of [Ru(p-cymene)Cl₂]₂ (4.5 mg, 7.3 μmol) and 4a (7.9 mg, 29 μmol) was heated in MeOH at 80 °C for 30 min under an argon atmosphere. After the solvent had been removed, 6 (0.17 mL, 1.5 mmol) and HCOOH/Et₃N (5:2) (0.73 mL) were added and the mixture stirred at 30 °C for 24 h.