132 Communications SYNTHESIS

2-(1,3-Dioxan-2-yl)-ethylidenetriphenylphosphorane, A New Three-Carbon Homologating Agent

John C. Stowell*, David R. Keith

Department of Chemistry, University of New Orleans, New Orleans, Louisiana 70122, U.S.A.

Several organometallic reagents are available for extending carbon chains by three carbons with an aldehyde function at the new terminus¹⁻⁶. We wish to introduce a convenient new reagent for this purpose which also offers the alternatives of providing α,β - or β,γ -unsaturation as well. We have found that the Wittig reagent 1 derived from the readily available 2-(2-bromoethyl)-1,3-dioxane is thermally stable⁷, and reacts with aldehydes to give cyclic acetals of *cis*-3-alkenals (2) in yields up to 76 %. The stereoselectivity is greater than 95 % when the phosphorane 1 is generated using potassium *t*-butoxide⁸.

Certain cis-3-alkenals are common volatile flavor constituents in foods. The earlier general, but lengthy synthetic routes to these compounds are based on chain extension of cis-2-butenediol^{9,10}, or of butadiyne via 1-methoxy-1butene-3-yne¹¹. trans-3-Alkenals have been prepared using methoxycyclopropyllithium12. We have prepared free cis-3alkenals (4) from their 1,3-propanediol acetals (2) by a twostep process. The six-membered ring acetal has a high equilibrium stability such that simple acid hydrolysis is unfavorable. However, trans-acetalization in acidic methanol gives the dimethyl acetal 3 in high yield without isomerization of the β , γ -unsaturation. The dimethyl acetal can then be hydrolyzed at room temperature in aqueous acetic acid to give the cis-3-alkenal 4 containing none of the α,β -unsaturated isomer^{13, 14} (2-alkenal). The product is quite sensitive to heat. For instance, simple steam distillation of cis-3decenal caused rapid isomerization to the α,β -unsaturated isomer. If the stable 2-alkenals (e.g. 5) are desired, the cyclic acetals 2 may be hydrolyzed with concomitant isomerization¹⁵ using hot aqueous acid plus chromium(III) chloride to complex the diol¹⁶.

Finally, the saturated aldehydes (e.g. 6) may be obtained by hydrogenation of 2 followed by hydrolysis again using aqueous acid plus chromium(III) chloride.

For the α , β -unsaturated and the saturated aldehydes, a mixture of cis and trans Wittig products is suitable. This is made conveniently using butyllithium in tetrahydrofuran. This mixture was used also for comparisons to show the exclusive cis nature of the products 2 above. The cis/trans isomer ratio in the cyclic acetals can be estimated by examination of the 100 MHz ¹H-N.M.R. signal (triplet) for the acetal proton at $\delta \approx 4.5$ ppm. The signal of the trans isomer appears 0.025 ppm upfield from the cis signal. The configurational assignment is based on ¹³C-N.M.R. spectra where the signals of the allyl C atoms in cis-2b appear 4.5 and 7.1 ppm upfield from the signals of the corresponding trans isomer ¹⁷. Also, the cis isomer is normally expected from a Wittig reaction in the absence of lithium salts⁸.

2-(2-Bromoethyl)-1,3-dioxane:

An excess of hydrogen bromide is bubbled into a solution of acrolein (56 g, 1.0 mol) in dichloromethane (300 ml) at room temperature. Excess hydrogen bromide is removed with a stream of nitrogen. 1,3-Propanediol (76 g, 1.0 mol) and p-toluenesulfonic acid hydrate (1 g) are added to the crude 3-bromopropanal and the mixture is stirred under nitrogen for 8 h. The resulting solution is neutralized with saturated sodium hydrogen carbonate, washed with water, and dried with potassium carbonate. The solvent is evaporated and the residual product distilled in vacuo; yield: 155 g (80 %); b.p. 68-70°/3 torr (Ref. 18, b.p. 68-74°/4.8 torr).

2-(1,3-Dioxan-2-yl)-ethyltriphenylphosphonium Bromide:

A solution of 2-(2-bromoethyl)-1,3-dioxane (94.1 g, 0.482 mol) and triphenylphosphine (200 g, 0.762 mol) in cyclohexane (250 ml) is heated at reflux with stirring for one day. The resulting solid is ground to a powder, washed with ether and then pentane (1000 ml each), and dried under vacuum; yield: 212 g (96%); m.p. 205-208°. If seed crystals of the product are added at the beginning of the reaction, the product is more easily isolated.

Table. 2-(cis-2-Alkenyl)-1,3-dioxanes (2):

2	R	Yield [%]	b.p./torr	Molecular formula ^a	¹H-N.M.R. δ [ppm]
a	C ₂ H ₅	41	31-34°/0.4	C ₉ H ₁₆ O ₂ (156.2)	(60 MHz, CCl ₄): 0.93 (t, 3H); 1.32 (m, 1H); 1.8–2.3 (m, 5H); 3.3–4.2 (m, 4H); 4.36 (t, 1H); 5.31 (m, 2H)
b ^b	n-C ₄ H ₉	60	103°/4	$C_{11}H_{20}O_2$ (184.3)	(100 MHz, CDCl ₃): 0.86 (t, 3H); 1.30 (m, 5H); 1.99 (m, 3H); 2.30 (m, 2H); 3.72 (dt, 2H); 4.03 (dd, 2H); 4.45 (t, 1H); 5.40 (m, 2H)
c	n-C ₆ H ₁₃	72	134°/5	$C_{13}H_{24}O_2$ (212.3)	(60 MHz, CCl ₄): 0.88 (t, 3H); 1.30 (m, 7H); 1.8–2.5 (m, 5H); 3.4–4.2 (m, 4H); 4.49 (t, 1H); 5.44 (m, 2H)
d	<i>n</i> -C ₈ H ₁₇	69	140-142°/1.2	C ₁₅ H ₂₈ O ₂ (240.4)	(100 MHz, CDCl ₃): 0.85 (t, 3H); 1.22 (m, 13H); 1.99 (m, 3H); 2.30 (m, 2H); 3.66 (dt, 2H); 4.01 (dd, 2H); 4.43 (t, 1H); 5.35 (m, 2H)

^a The microanalyses of compounds 2a, c, d were in good agreement with the calculated values: C, ± 0.18 ; H, ± 0.14 . 2b: C, -0.42;

^b ¹³C-N.M.R. (25.2 MHz, CDCl₃): δ = 14.0, 22.4, 25.8, 27.2, 31.75, 33.5, 67.0, 102.1, 123.0, 132.6 ppm.

February 1979 Communications 133

2-(cis-2-Alkenyl)-1,3-dioxanes (2); General Procedure:

A 0.5 molar solution of potassium t-butoxide in tetrahydrofuran (100 ml, 50 mmol) is added to a stirred solution of 2-(1,3-dioxan-2-yl)-ethyltriphenylphosphonium bromide (22.8 g, 50 mmol) in dry dimethyl sulfoxide or tetrahydrofuran (50 ml) to give the orange phosphorane 1. Stirring is continued at room temperature for 30 min and the aldehyde (100 mmol) is added dropwise. The mixture is stirred for 2h and then poured into water (500 ml). The product is extracted with ether, the extract dried, the solvent evaporated, and the residue distilled in vacuo.

cis- and trans-2-(2-Alkenyl)-1,3-dioxanes; General Procedure:

A 1.6 molar solution of butyllithium in hexane (11 ml, 17.5 mmol) is added to a stirred suspension of 2-(1,3-dioxan-2-yl)-ethyltriphenylphosphonium bromide (7.98 g, 17.5 mmol) in dry tetrahydrofuran (50 ml) at -20° . The resultant solution of the orange phosphorane 1 is stirred at -20° for 1 h and then the aldehyde (18 mmol) is added. Stirring is continued for 12 h at room temperature, the mixture is poured into water (500 ml), and extracted with ether. The extract is dried, the solvent evaporated, and the residue distilled in vacuo to give compounds 2a-d containing up to 40 % of the corresponding trans isomers; yield: 65–76 %. The 13 C-N.M.R. spectrum of 2-(2-heptenyl)-1,3-dioxane thus obtained showed the signals for the cis isomer as listed above along with the following peaks for the trans isomer: δ =14.1, 22.3, 31.7, 32.4, 38.9, 102.3, 123.9, 133.6 ppm.

cis-3-Decenal Dimethyl Acetal (3c):

A solution of 2-(cis-2-nonenyl)-1,3-dioxane (2c; 4.64 g, 21.9 mmol) and p-toluenesulfonic acid (0.15 g, 0.8 mmol) in methanol (200 ml) is heated at reflux for 2h, cooled, and neutralized with anhydrous sodium carbonate. After evaporation of the methanol, the product is taken up in dichloromethane, washed with water, rotary-evaporated, and distilled; yield: 3.83 g (88%); b.p. 109-111°/11 torr (Ref. 11, b.p. 108-110°/14 torr).

¹H-N.M.R. (60 MHz, CDCl₃): δ =0.9 (t, 3H); 1.24 (m, 8H); 2.0 (m, 2H); 2.3 (t, 2H); 3.28 (s, 6H); 4.32 (t, 1H); 5.38 ppm (m, 2H).

The N.M.R. shows a trace of cyclic acetal remaining, but if necessary this can be removed by a second treatment with acidic methanol.

cis-3-Dodecenal Dimethyl Acetal (3d):

Prepared as above from **2d** (2.4 g, 0.1 mol); yield: $2.03 \, g$ (89 %); b.p. $84^{\circ}/0.4$ torr (Ref. 11 , b.p. $62-63^{\circ}/0.001$ torr); n_D^{22} : 1.4442 (Ref. 11 , n_D^{20} : 1.4446).

¹H-N.M.R. (100 MHz, CDCl₃): δ =0.86 (t, 3H); 1.24 (m, 12H); 1.98 (m, 2H); 2.30 (t, 2H); 3.24 (s, 6H); 4.27 (t, 1H); 5.33 ppm (m, 2H).

cis-3-Decenal (4c):

A solution of cis-3-decenal dimethyl acetal (3c; 1.21 g, 6.05 mmol) in acetic acid (24 ml) and water (6 ml) is stirred at room temperature for 4 h, neutralized with cold saturated sodium hydrogen carbonate, taken up in dichloromethane, and dried with potassium carbonate. The solvent is rotary-evaporated to give 4c containing a small amount of unhydrolyzed 3c, but none of the isomeric 2-decenal; yield: 0.683 g (73%).

I.R. (neat): $v_{\text{max}} = 1730$, 740 cm^{-1} .

¹H-N.M.R. (60 MHz, CDCl₃): δ = 0.9 (t, 3 H); 1.25 (m, 8 H); 2.0 (m, 2 H); 3.1 (m, 2 H); 5.2-5.7 (m, 2 H); 9.51 ppm (t, 1 H).

trans-2-Decenal (5):

A solution of chromium(III) chloride hexahydrate (2.6 g, 9.8 mmol) in 4 molar hydrochloric acid (75 ml) is added to 2-(cis-2-nonenyl)-1,3-dioxane (2c; 0.59 g, 2.8 mmol), the mixture is heated at reflux for 15 min, and then steam-distilled. The wet mixture of 2-decenal and 3-decenal thus obtained is added to 4 molar hydrochloric acid (75 ml), the mixture is heated at reflux for 30 min, and then steam-distilled. The distillate is extracted with pentane, the extract is dried with potassium carbonate; and rotary evaporated;

yield of 5 (homogeneous by G.L.C.): 0.37 g (86 %); semicarbazone, m.p. 167° (Ref. ¹⁹, m.p. 168.5°).

¹H-N.M.R. (60 MHz, CDCl₃): δ = 0.8 (t, 3 H); 1.2 (m, 10 H); 2.2 (m, 2 H); 5.90 (ddt, 1 H); 6.72 (dt, 1 H); 9.30 ppm (d, 1 H).

Decanal (6):

2-Nonyl-1,3-dioxane: A solution of 2-(cis-2-nonenyl)-1,3-dioxane (2c; 0.68 g, 3.2 mmol) in ethanol (40 ml) is hydrogenated over 5% palladium on carbon (4.9 mg) at 1 atm for 25 min. The resultant mixture is filtered and evaporated; yield: 0.68 g (99%); b.p. 99–100°/0.6 torr.

Decanal (6): A solution of chromium(III) chloride hexahydrate (15.6 g, 58.5 mmol) in 4 molar hydrochloric acid (100 ml) is added to 2-nonyl-1,3-dioxane (2.59 g, 12.1 mmol), the mixture is heated at reflux for 15 min, and then steam-distilled. The distillate is extracted with pentane, the extract is dried with potassium carbonate and evaporated to give 6; yield: 1.75 g (92%); semicarbazone, m.p. 102° (Ref. 19, m.p. 102°).

We are grateful to J. B. Stothers of the University of Western Ontario for the ¹³C-N.M.R. spectra, and to H. E. Ensley of Tulane University for the 100 MHz ¹H-N.M.R. spectra.

Received: August 21, 1978

- ¹ G. Büchi, H. Wüest, J. Org. Chem. 34, 1122 (1969).
- ² D. N. Brattesani, C. H. Heathcock, J. Org. Chem. 40, 2165 (1975).
- ³ J. C. Stowell, J. Org. Chem. 41, 560 (1976).
- ⁴ K. Kondo, D. Tunemoto, Tetrahedron Lett. 1975, 1397.
- ⁵ G. K. Cooper, L. J. Dolby, Tetrahedron Lett. 1976, 4675.
- ⁶ E. J. Corey, I. Vlattas, N. H. Andersen, K. Harding, J. Am. Chem. Soc. 90, 3247 (1968).
- Other reagents with an O-function attached on a position β to the nucleophilic site may undergo intramolecular attack near room temperature:
- C. Feugeas, Bull. Soc. Chim. Fr. 1963, 2568.
- P. E. Eaton et al., J. Am. Chem. Soc. 99, 2751 (1977).
- A lower homologue of 1, the phosphorane derived from 2-bromomethyl-1,3-dioxolane, is unstable and is best formed *in situ:* T. M. Cresp, M. V. Sargent, P. Vogel, *J. Chem. Soc. Perkin Trans. 1* 1974, 37.
- ⁸ R. J. Anderson, C. A. Henrick, J. Am. Chem. Soc. 97, 4327 (1975).
- ⁹ T. Kajiwara, T. Harada, A. Hatanaka, Agr. Biol. Chem. 39, 243 (1975).
- ¹⁰ T. Kajiwara, Y. Odake, A. Hatanaka, Agr. Biol. Chem. 39, 1617 (1975).
- T. Herbertz, Chem. Ber. 85, 475 (1952).
 M. Winter, Helv. Chim. Acta 46, 1792 (1963).
- ¹² E. J. Corey, P. Ulrich, Tetrahedron Lett. 1975, 3685.
- These conditions were reported recently for the hydrolysis of ethylene glycol acetals of β,γ-unsaturated ketones without double-bond migration: J. H. Babler, N. C. Malek, M. J. Coghlan, J. Org. Chem. 43, 1821 (1978).
- The conditions reported by Winter¹¹ and also by Corey¹² for the hydrolysis of the β , γ -unsaturated dimethyl acetals gave a considerable amount of α , β -unsaturated aldehyde in our hands.
- A related process was reported recently which utilized methoxyallene: S. F. Martin, P. J. Garrison, *Tetrahedron Lett.* 1977, 3875.
- ¹⁶ R. A. Barnes, J. Polymer Sci. 27, 285 (1958).
- ¹⁷ J. W. de Haan, L. J. M. van de Ven, *Org. Magn. Res.* 5, 147 (1973).
- ¹⁸ D. C. Kriesel, O. Gisvold, J. Pharm. Sci. 60, 1250 (1971).
- Dictionary of Organic Compounds, Vol. 2, I. Heilbron, ed., 4th Ed., Oxford University Press, New York, 1965, p. 822, 825.