



5-Hydroxypent-2-enitriles as precursors toward dihydropyranones, dienenitriles and functionalized naphthalenes

Xinying Zhang^{*}, Xuefei Jia, Nan Liu, Xiaojie Guo, Yunping Song, Xuesen Fan^{*}

School of Chemistry and Chemical Engineering, Henan Key Laboratory of Environmental Pollution Control, Key Laboratory of Green Chemical Media and Reactions, Ministry of Education, Henan Normal University, Xinxiang, Henan 453007, PR China

ARTICLE INFO

Article history:

Received 11 January 2013

Received in revised form 16 April 2013

Accepted 26 April 2013

Available online 9 May 2013

Keywords:

5-Hydroxypent-2-enitriles

Dihydropyranones

Dienenitriles

Functionalized naphthalenes

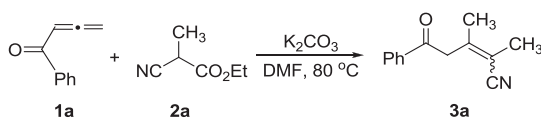
ABSTRACT

5-Hydroxypent-2-enitriles were found to be precursors toward several classes of compounds with synthetic and biological interests. To be specific, upon treatment with sulfuric acid in dichloromethane and depending on the configuration of the C–C double bond and the nature of substituents attached on the C–C double bond, 5-hydroxypent-2-enitriles could afford 5,6-dihydropyran-2-ones, 2,4-dienenitriles, or functionalized naphthalenes through intramolecular Pinner reaction, β -elimination or intramolecular Friedel–Crafts reaction, respectively.

© 2013 Elsevier Ltd. All rights reserved.

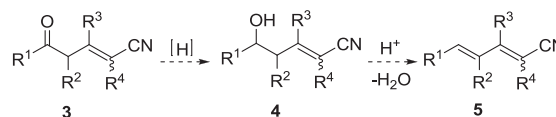
1. Introduction

As part of our recent studies on functionalized allenes,¹ we have developed a synthetic pathway toward δ -oxo- α,β -unsaturated nitrile (**3**) via the tandem reaction of 1,2-allenic ketone (**1**) with α -substituted cyanoacetate (**2**) (Scheme 1).^{1b}



Scheme 1. Synthesis of δ -oxo- α,β -unsaturated nitrile (**3a**).

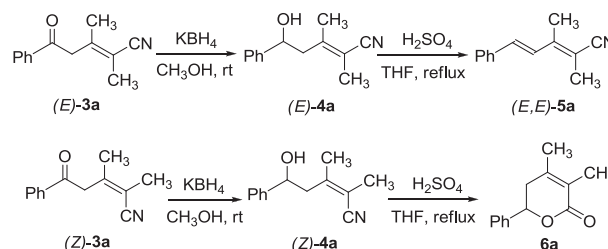
As both carbonyl and cyano groups are exceptionally versatile in organic transformations, we expected δ -oxo- α,β -unsaturated nitriles (**3**) to serve as valuable precursors toward compounds with synthetic and biological interests. In this regard, 2,4-dienenitriles emerged as attractive targets since they are not only essential building blocks of biologically active natural products, but also intermediates to high-electron affinity polymers used to produce light emitting diodes (LEDs).^{2–4} Hence, we hypothesized a synthetic method toward 2,4-dienenitrile (**5**) starting from δ -oxo- α,β -unsaturated nitrile (**3**) and intermediating with 5-hydroxypent-2-enitrile (**4**) as shown in Scheme 2.



Scheme 2. Proposed synthesis of 2,4-dienenitrile (**5**) from **3**.

2. Results and discussion

To check the feasibility of the envisioned pathway to 2,4-dienenitrile (**5**), (*E*)-5-hydroxy-2,3-dimethyl-5-phenylpent-2-enitrile ((*E*)-**4a**), prepared through reduction of (*E*)-2,3-dimethyl-5-oxo-5-phenylpent-2-enitrile ((*E*)-**3a**),^{1b} was treated with H₂SO₄ in THF under reflux for 8 h. To our delight, it gave the expected (2*E*,4*E*)-2,3-dimethyl-5-phenylpenta-2,4-dienenitrile ((*E,E*)-**5a**) in a yield of 84% (Scheme 3). However, when (*Z*)-5-hydroxy-2,3-dimethyl-5-phenylpent-2-enitrile ((*Z*)-**4a**) was subjected to similar

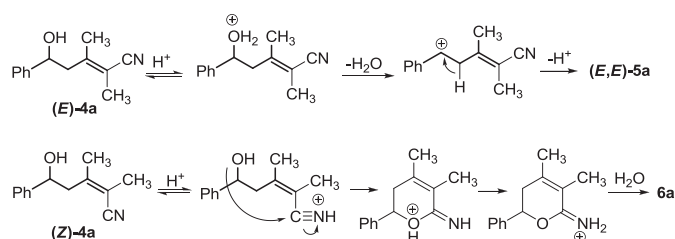


Scheme 3. Formation of (*E,E*)-**5a** from (*E*)-**3a** and **6a** from (*Z*)-**3a**.

^{*} Corresponding authors. E-mail address: xinyingzhang@htu.cn (X. Zhang).

conditions, the corresponding (2*Z*,4*E*)-2,3-dimethyl-5-phenylpenta-2,4-dienitrile ((*Z,E*)-**5a**) was formed only in trace amount. Instead, 5,6-dihydro-3,4-dimethyl-6-phenylpyran-2-one (**6a**) was isolated in a yield of 66% (Scheme 3).

Plausible pathways accounting for the formation of (*E,E*)-**5a** and **6a** were described in Scheme 4. While an acid-promoted dehydration of (*E*)-**4a** affords (*E,E*)-**5a**, an alternative reaction pathway featured with an intramolecular Pinner reaction⁵ occurs with (*Z*)-**4a** resulting in the formation of **6a**.



Scheme 4. Plausible pathways toward (*E,E*)-**5a** and **6a**.

Literature searching revealed that 5,6-dihydropyran-2(1*H*)-one is a privileged structural scaffold frequently found in both natural products and pharmaceutical agents.⁶ This promoted us to carry out thorough studies on the transformation of (*Z*)-**4a** with the aim to develop it into a novel synthetic method toward 5,6-dihydropyran-2(1*H*)-ones.

Thus, the reaction of (*Z*)-**4a** toward **6a** was optimized with regard to solvents and acidic promoters and the results were included in Table 1. With H₂SO₄ as the catalyst, the choice of solvent turned out to be essential to the proceeding of the reaction. Among commonly used solvents, such as THF, DMF, CH₃CN, EtOH, toluene, CH₂Cl₂, and H₂O, CH₂Cl₂ appeared to be the most efficient in mediating this tandem process. Of the acidic catalysts screened, H₂SO₄ exhibited the highest catalytic activity. Finally, the optimal conditions were identified as follows: treating (*Z*)-**4a** with 1 equiv of H₂SO₄ in CH₂Cl₂ under reflux for 3 h. Under such conditions, **6a** was obtained in a yield of 75%.

Table 1
Optimization for the formation of **6a** from (*Z*)-**4a**^a

Entry	Catalyst	Solvent	<i>T</i> (°C)	<i>t</i> (h)	Yield ^b (%)
1	H ₂ SO ₄	THF	Reflux	10	66
2	H ₂ SO ₄	DMF	80	10	Trace
3	H ₂ SO ₄	CH ₃ CN	Reflux	5	66
4	H ₂ SO ₄	EtOH	Reflux	10	58
5	H ₂ SO ₄	Toluene	80	10	22
6	H ₂ SO ₄	CH ₂ Cl ₂	Reflux	3	75
7	H ₂ SO ₄	H ₂ O	80	10	NR
8	BF ₃ ·Et ₂ O	CH ₂ Cl ₂	Reflux	10	70
9	CF ₃ CO ₂ H	CH ₂ Cl ₂	Reflux	15	63

^a Reaction conditions: **4a** (0.5 mmol), acid (0.5 mmol), solvent (5 mL).

^b Isolated yield.

With the optimized conditions in hand, we then examined the scope and generality of the synthetic procedure toward **6**. The results included in Table 2 show that most of the substrates studied undergo the intramolecular Pinner reaction smoothly to afford the corresponding 5,6-dihydropyran-2(1*H*)-ones (**6**) with moderate to good yields. Notably, the reactions are compatible with a variety of functional groups such as chloro, bromo, fluoro, trifluoromethyl, allyl, and benzyl groups. By varying the substituents attached on

Table 2
Scope of the reaction leading to **6** from (*Z*)-**4a**

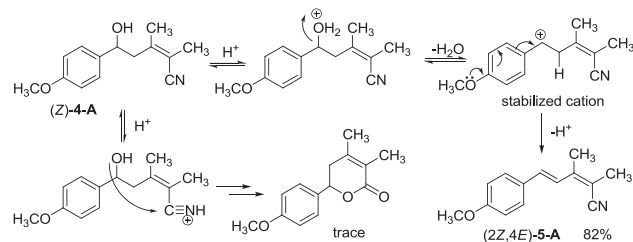
Entry	R ¹	R ²	R ³	R ⁴	Products	Yield ^b (%)
1	C ₆ H ₅	H	CH ₃	CH ₃	6a	75
2	<i>m</i> -ClC ₆ H ₄	H	CH ₃	CH ₃	6b	81
3	<i>p</i> -ClC ₆ H ₄	H	CH ₃	CH ₃	6c	85
4	<i>m</i> -BrC ₆ H ₄	H	CH ₃	CH ₃	6d	78
5	<i>p</i> -BrC ₆ H ₄	H	CH ₃	CH ₃	6e	86
6	<i>m</i> -CH ₃ C ₆ H ₄	H	CH ₃	CH ₃	6f	72
7	<i>p</i> -CF ₃ C ₆ H ₄	H	CH ₃	CH ₃	6g	88
8	C ₆ H ₅	H	CH ₃	Bn	6h	68
9	<i>m</i> -ClC ₆ H ₄	H	CH ₃	Bn	6i	80
10	<i>p</i> -BrC ₆ H ₄	H	CH ₃	Bn	6j	79
11	<i>p</i> -CH ₃ C ₆ H ₄	H	CH ₃	Bn	6k	51
12	C ₆ H ₅	H	CH ₃	Allyl	6l	72
13	<i>p</i> -FC ₆ H ₄	H	CH ₃	Allyl	6m	79
14	<i>p</i> -CF ₃ C ₆ H ₄	H	CH ₃	Allyl	6n	86
15	C ₆ H ₅	CH ₃	CH ₃	CH ₃	6o	80
16	4-BrC ₆ H ₄	CH ₃	CH ₃	CH ₃	6p	82
17	C ₆ H ₅	H	CH ₃	Ph	6q	68
18	3-ClC ₆ H ₄	H	CH ₃	Ph	6r	73
19	4-ClC ₆ H ₄	H	CH ₃	Ph	6s	78
20	4-BrC ₆ H ₄	H	CH ₃	Ph	6t	85
21	3-CH ₃ C ₆ H ₄	H	CH ₃	Ph	6u	51
22	4-CH ₃ C ₆ H ₄	H	CH ₃	Ph	6v	48
23	4-CF ₃ C ₆ H ₄	H	CH ₃	Ph	6w	90
24	4-BrC ₆ H ₄	CH ₃	CH ₃	Ph	6x	86
25	CH ₃	H	Bn	CH ₃	6y	81
26	CH ₃	H	Bn	Bn	6z	76
27	Ph	H	Bn	CH ₃	6aa	72

^a Reaction conditions: **4** (0.5 mmol), H₂SO₄ (0.5 mmol), CH₂Cl₂ (5 mL), reflux, 3 h.

^b Isolated yield.

different positions of the substrates, diversely substituted 5,6-dihydropyran-2(1*H*)-ones could be obtained.

It is also noted, however, for substrate with strong electron-donating group on the phenyl ring of R¹, namely (*Z*)-5-hydroxy-5-(4-methoxyphenyl)-2,3-dimethylpent-2-enitrile ((*Z*)-**4-A**), the reaction mainly give the corresponding dehydration product, (2*Z*,4*E*)-5-(4-methoxyphenyl)-2,3-dimethylpenta-2,4-dienitrile ((*Z,E*)-**5-A**, Scheme 5). The expected 5,6-dihydropyran-2(1*H*)-one was formed only in trace amount. It might be explained by the stabilizing effect of the electron-donating methoxy group on the proposed cationic intermediate toward 2,4-dienitrile. The increased stability of the cation intermediate makes the dehydration more favored than the competing intramolecular Pinner reaction.

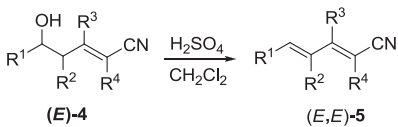


Scheme 5. Dehydration favored with electron-donating effect.

Having established a new protocol for the preparation of 5,6-dihydropyran-2(1*H*)-ones (**6**) from (*Z*)-5-hydroxypent-2-enitrile ((*Z*)-**4**), we moved our focus back to the synthesis of (2*E*,4*E*)-2,4-dienitriles ((*E,E*)-**5**) from (*E*)-5-hydroxypent-2-enitrile ((*E*)-**4**). Optimization study with (*E*)-**4a** as a model substrate turned out that treatment of (*E*)-**4a** with 1 equiv of H₂SO₄ in CH₂Cl₂ under reflux for 2 h afforded (*E,E*)-**5a** in 86% yield.

The scope and generality of the synthetic strategy toward 2,4-dienitriles (**5**) were then studied and the results were listed in Table 3. Generally, good to excellent yields of (2*E*,4*E*)-2,4-dienitriles were observed and the reaction was found to be compatible with various functional groups including the electron-deficient trifluoromethyl and the electron-rich methoxy groups, thus resulting a general and efficient approach toward 2,4-dienitrile derivatives.

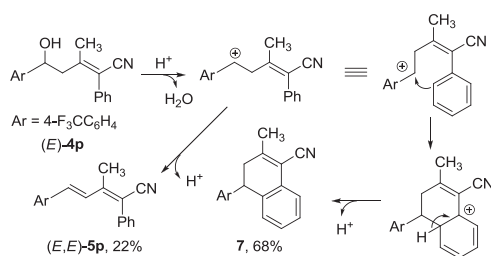
Table 3

Scope of the reaction leading to (E,E)-**5** from (E)-**4**^a


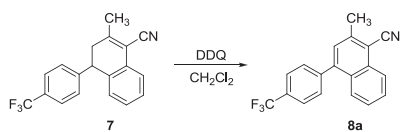
Entry	R ¹	R ²	R ³	R ⁴	Products	Yield ^b (%)
1	C ₆ H ₅	H	CH ₃	CH ₃	5a	86
2	<i>m</i> -ClC ₆ H ₄	H	CH ₃	CH ₃	5b	88
3	<i>p</i> -ClC ₆ H ₄	H	CH ₃	CH ₃	5c	92
4	<i>m</i> -BrC ₆ H ₄	H	CH ₃	CH ₃	5d	91
5	<i>p</i> -BrC ₆ H ₄	H	CH ₃	CH ₃	5e	93
6	<i>m</i> -CH ₃ C ₆ H ₄	H	CH ₃	CH ₃	5f	82
7	<i>p</i> -CH ₃ OC ₆ H ₄	H	CH ₃	CH ₃	5g	86
8	<i>p</i> -CF ₃ C ₆ H ₄	H	CH ₃	CH ₃	5h	85
9	C ₆ H ₅	H	CH ₃	Bn	5i	83
10	<i>m</i> -ClC ₆ H ₄	H	CH ₃	Bn	5j	86
11	<i>p</i> -BrC ₆ H ₄	H	CH ₃	Bn	5k	85
12	C ₆ H ₅	H	CH ₃	Allyl	5l	82
13	<i>p</i> -FC ₆ H ₄	H	CH ₃	Allyl	5m	88
14	<i>p</i> -CF ₃ C ₆ H ₄	H	CH ₃	Allyl	5n	85
15	C ₆ H ₅	CH ₃	CH ₃	CH ₃	5o	81

^a Reaction conditions: **4** (0.5 mmol), H₂SO₄ (0.5 mmol), CH₂Cl₂ (5 mL), reflux.^b Isolated yield.

To further extend the scope of the above reaction, (E)-5-hydroxy-3-methyl-2-phenyl-5-(4-(trifluoromethyl)phenyl)pent-2-enitrile ((E)-**4p**), with a phenyl group on the α-position of the cyano, was treated with H₂SO₄ in CH₂Cl₂. Interestingly, it gave 2-methyl-4-(4-(trifluoromethyl)phenyl)-3,4-dihydronaphthalene-1-carbonitrile (**7**) as a major product rather than the expected (2*E*,4*E*)-3-methyl-2-phenyl-5-(4-(trifluoromethyl)phenyl)-penta-2,4-dienitrile ((E,E)-**5p**). A mechanism rationalizing the formation of **7** featured with an intramolecular Friedel–Crafts reaction is depicted in Scheme 6.

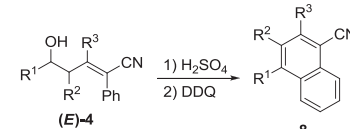
Scheme 6. A plausible pathway for the formation of **7**.

In following studies, we found that upon treatment with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (DDQ), compound **7** underwent an oxidative aromatization to give 2-methyl-4-(4-(trifluoromethyl)phenyl)-1-naphthonitrile (**8a**) in 87% yield (Scheme 7).

Scheme 7. Oxidative aromatization of **7** with DDQ.

It is well known that naphthalene skeleton occurs in many biologically important natural products and building blocks for the synthesis of pharmaceuticals and polycyclic aromatic electronic materials.^{7,8} To date, several efficient methods for the synthesis of substituted naphthalenes have been reported.⁹ In spite of their success, some of these synthetic methods still suffer from expensive catalysts, harsh reaction conditions or low efficiency. Based on the above facts, we proceeded to study the feasibility of a one-pot preparation of **8a** directly from (E)-**4p**. For this purpose, (E)-**4p** was firstly treated with H₂SO₄ in refluxing CH₂Cl₂ for 2 h. Then DDQ was added and the mixture was stirred under reflux for another 4 h. Separation of the resulting mixture gave **8a** in a total yield of 63%. The generality and limitation of the one-pot process leading to α-aryl naphthalene (**8**) was then studied. We found that a variety of (E)-5-hydroxypent-2-enitriles with a phenyl group on the α-position of the cyano participated in this reaction to give the corresponding α-aryl naphthalenes in moderate yields (Table 4). It is notable that this sequence of reactions provides an attractive alternative to cross-coupling protocols in some cases for the preparation of biaryls, including the 1,1'-binaphthyl-4-carbonitrile (**8h**, entry 8).

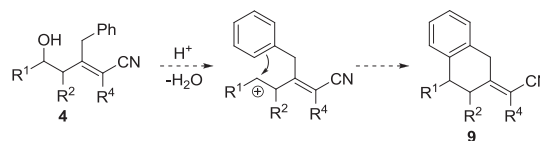
Table 4

Scope of the one-pot synthesis of α-aryl naphthalenes (**8**)^a


Entry	R ¹	R ²	R ³	Products	Yield ^b (%)
1	4-CF ₃ C ₆ H ₄	H	CH ₃	8a	63
2	3-ClC ₆ H ₄	H	CH ₃	8b	52
3	4-ClC ₆ H ₄	H	CH ₃	8c	53
4	4-BrC ₆ H ₄	H	CH ₃	8d	56
5	3-CH ₃ C ₆ H ₄	H	CH ₃	8e	42
6	4-CH ₃ C ₆ H ₄	H	CH ₃	8f	40
7	C ₆ H ₅	H	CH ₃	8g	61
8	1-Naphthyl	H	CH ₃	8h	35
9	4-BrC ₆ H ₄	CH ₃	CH ₃	8i	62

^a Reaction conditions: **4** (0.5 mmol), H₂SO₄ (0.5 mmol), CH₂Cl₂ (5 mL), reflux, 2 h; then DDQ (0.5 mmol), reflux, 4 h.^b Isolated yield.

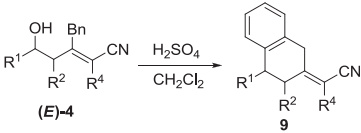
In a further aspect, inspired by the reaction of (E)-5-hydroxypent-2-enitriles with a phenyl group on the α-position of cyano affording α-substituted 3,4-dihydronaphthalenes (**7**), we envisioned that tetrahydronaphthalene (**9**)^{10–13} might be obtained from (E)-5-hydroxypent-2-enitriles having a benzyl group on the β-position of cyano via an intramolecular Friedel–Crafts reaction as shown in Scheme 8.

Scheme 8. Proposed synthesis of tetrahydronaphthalene (**9**).

In deed, when (E)-3-benzyl-5-hydroxy-2-methylhex-2-enitrile was treated with H₂SO₄ in CH₂Cl₂, the expected 2-(4-methyl-3,4-dihydronaphthalen-2(1*H*)-ylidene)propanenitrile (**9a**) was obtained in a yield of 56%. The success of this strategy was further demonstrated by the reactions with other substrates and the results were shown in Table 5.

Finally, since effective separation of (E)-**3** and (Z)-**3** or (E)-**4** and (Z)-**4** usually requires considerable efforts due to their structural similarity, the feasibility of one-pot preparation of (2*E*,4*E*)-**5a** and **6a** by

Table 5
Synthesis of 2-methylene-1,2,3,4-tetrahydronaphthalene (**9**)^a

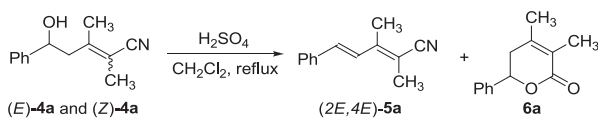


Entry	R ¹	R ²	R ⁴	Products	Yield ^b (%)
1	CH ₃	H	CH ₃	9a	56
2	CH ₃	H	Bn	9b	53
3	CH ₃	H	Ph	9c	43
4	Ph	H	CH ₃	9d	48

^a Reaction conditions: **4** (0.5 mmol), H₂SO₄ (0.5 mmol), CH₂Cl₂ (5 mL), reflux, 3 h.

^b Isolated yield.

using a mixture of (*E*)-**4a** and (*Z*)-**4a** as substrates was investigated. To our delight, treating a mixture of (*E*)-**4a** and (*Z*)-**4a**, obtained through the reduction of a mixture of (*E*)-**3a** and (*Z*)-**3a**, with H₂SO₄ in CH₂Cl₂ resulted in efficient formation of (*2E,4E*)-**5a** and **6a** (Scheme 9). In addition, these two products were separated conveniently as they are remarkably different in their structural characteristics.



Scheme 9. One-pot preparation of (*2E,4E*)-**5a** and **6a** from **4a**.

3. Conclusion

In conclusion, novel synthetic methods for the preparation of 5,6-dihydropyran-2(1*H*)-ones, 2,4-dienitriles, and functionalized naphthalenes via acid-promoted reactions of 5-hydroxypent-2-enenitriles with various substitution patterns have been developed. Notable features of the results reported herein include: (1) various kinds of target compounds could be obtained from substrates originated from the same reaction; (2) different transformations could be performed under almost equally mild conditions without using any noble catalysts. In addition, a mixture of *E/Z* isomers of 5-hydroxypent-2-enenitrile could be used directly and two different kinds of products can be obtained conveniently from the corresponding *E* or *Z* isomer via a one-pot procedure. With the above-described advantages, these synthetic methods are expected to find wide applications in synthetic, medicinal chemistry, and related areas.

4. Experimental section

4.1. General

¹H and ¹³C NMR spectra were recorded at 400 and 100 MHz, respectively. High-resolution mass spectra (HRMS) were performed on a time-of-flight (ESI-TOF) mass spectrometer. Melting points were measured with a micro melting point apparatus. All reactions were monitored by thin-layer chromatography (TLC) using silica gel plates (silica gel 60 F₂₅₄ 0.25 mm) and components were visualized by observation under UV light (254 and 365 nm). δ-Oxo-α,β-unsaturated nitriles (**3**) were prepared based on our previously published procedure.^{1b} 5-Hydroxypent-2-enenitriles (**4**) were obtained by reduction of the corresponding δ-oxo-α,β-unsaturated nitriles (**3**) with KBH₄ in methanol at ambient temperature.

4.2. Typical procedure for the preparation of (*2E,4E*)-2,3-dimethyl-5-phenylpenta-2,4-dienitrile ((*2E,4E*)-**5a**) (Table 3)

To a flask containing (*E*)-5-hydroxy-2,3-dimethyl-5-phenylpent-2-enenitrile (101 mg, 0.5 mmol) in CH₂Cl₂ (5 mL) was added H₂SO₄

(27 μL, 0.5 mmol). The solution was stirred under reflux. Upon completion, the reaction was quenched with aqueous Na₂CO₃. The resulting mixture was extracted with ethyl acetate (5 mL×3). The combined organic phases were washed with brine, dried, filtered, and concentrated under vacuum. The residue was purified by column chromatography on silica gel eluting with ethyl acetate/hexane (1:10) to give (*2E,4E*)-2,3-dimethyl-5-phenylpenta-2,4-dienitrile ((*2E,4E*)-**5a**) in a yield of 86% (79 mg, 0.43 mmol). Other (*2E,4E*)-2,4-dienitrile derivatives were obtained in a similar manner.

4.2.1. (*2E,4E*)-2,3-Dimethyl-5-phenylpenta-2,4-dienitrile (5a**).** Yield: 79 mg, 86%. White solid, mp: 104–105 °C; ¹H NMR (400 MHz, CDCl₃) δ: 2.09 (s, 3H), 2.28 (s, 3H), 6.88 (d, *J*=16.0 Hz, 1H), 7.11 (d, *J*=16.0 Hz, 1H), 7.32–7.40 (m, 3H), 7.49 (d, *J*=7.2 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ: 15.9, 18.5, 105.9, 120.8, 123.8, 127.2, 128.9, 129.0, 135.2, 136.2, 147.9. IR (KBr) ν=2919, 2211, 1531, 772 cm⁻¹; MS: *m/z* 184 (MH)⁺. HRMS calcd for C₁₃H₁₄N: 184.1126 [M+H], found: 184.1134.

4.2.2. (*2E,4E*)-5-(3-Chlorophenyl)-2,3-dimethylpenta-2,4-dienitrile (5b**).** Yield: 95 mg, 88%. White solid, mp: 91–93 °C; ¹H NMR (400 MHz, CDCl₃) δ: 2.10 (s, 3H), 2.26 (s, 3H), 6.80 (d, *J*=16.0 Hz, 1H), 7.09 (d, *J*=16.0 Hz, 1H), 7.26–7.35 (m, 3H), 7.46 (s, 1H). ¹³C NMR (100 MHz, CDCl₃) δ: 16.0, 18.5, 107.0, 120.5, 125.0, 125.5, 126.8, 128.8, 130.1, 133.6, 134.8, 138.0, 147.4. IR (KBr) ν=2924, 2201, 1510, 957, 780 cm⁻¹; MS: *m/z* 218 (MH)⁺. HRMS calcd for C₁₃H₁₃ClN: 218.0737 [M+H], found: 218.0745.

4.2.3. (*2E,4E*)-5-(4-Chlorophenyl)-2,3-dimethylpenta-2,4-dienitrile (5c**).** Yield: 100 mg, 92%. White solid, mp: 122–124 °C; ¹H NMR (400 MHz, CDCl₃) δ: 2.10 (s, 3H), 2.27 (s, 3H), 6.83 (d, *J*=16.0 Hz, 1H), 7.08 (d, *J*=16.0 Hz, 1H), 7.33 (d, *J*=7.2 Hz, 2H), 7.41 (d, *J*=7.2 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ: 15.9, 18.4, 106.5, 120.6, 124.3, 128.3, 129.1, 133.8, 134.68, 134.71, 147.5. IR (KBr) ν=2923, 2202, 1487, 814 cm⁻¹; MS: *m/z* 218 (MH)⁺. HRMS calcd for C₁₃H₁₃ClN: 218.0737 [M+H], found: 218.0729.

4.2.4. (*2E,4E*)-5-(3-Bromophenyl)-2,3-dimethylpenta-2,4-dienitrile (5d**).** Yield: 119 mg, 91%. White solid, mp: 112–114 °C; ¹H NMR (400 MHz, CDCl₃) δ: 2.10 (s, 3H), 2.26 (s, 3H), 6.78 (d, *J*=16.0 Hz, 1H), 7.08 (d, *J*=16.0 Hz, 1H), 7.23 (t, *J*=8.0 Hz, 1H), 7.38 (d, *J*=7.6 Hz, 1H), 7.43 (d, *J*=7.6 Hz, 1H), 7.62 (s, 1H). ¹³C NMR (100 MHz, CDCl₃) δ: 16.0, 18.4, 107.1, 120.4, 123.0, 125.1, 125.9, 129.8, 130.3, 131.7, 133.5, 138.3, 147.3. IR (KBr) ν=2924, 2204, 779 cm⁻¹; MS: *m/z* 262 (MH)⁺. HRMS calcd for C₁₃H₁₃BrN: 262.0231 [M+H], found: 262.0235.

4.2.5. (*2E,4E*)-5-(4-Bromophenyl)-2,3-dimethylpenta-2,4-dienitrile (5e**).** Yield: 122 mg, 93%. White solid, mp: 156–157 °C; ¹H NMR (400 MHz, CDCl₃) δ: 2.09 (s, 3H), 2.27 (s, 3H), 6.80 (d, *J*=16.0 Hz, 1H), 7.08 (d, *J*=16.0 Hz, 1H), 7.34 (d, *J*=8.4 Hz, 2H), 7.48 (d, *J*=8.0 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ: 15.9, 18.5, 106.5, 120.6, 122.9, 124.4, 128.6, 132.0, 133.9, 135.1, 147.6. IR (KBr) ν=2918, 2203, 1502, 819 cm⁻¹; MS: *m/z* 262 (MH)⁺. HRMS calcd for C₁₃H₁₃BrN: 262.0231 [M+H], found: 262.0238.

4.2.6. (*2E,4E*)-5-(3-Methylphenyl)-2,3-dimethylpenta-2,4-dienitrile (5f**).** Yield: 81 mg, 82%. Syrup; ¹H NMR (400 MHz, CDCl₃) δ: 2.10 (s, 3H), 2.28 (s, 3H), 2.39 (s, 3H), 6.86 (d, *J*=15.6 Hz, 1H), 7.09–7.16 (m, 2H), 7.27–7.31 (m, 3H). ¹³C NMR (100 MHz, CDCl₃) δ: 15.9, 18.5, 21.4, 105.7, 120.8, 123.6, 124.4, 127.9, 128.8, 129.9, 135.4, 136.1, 138.5, 148.0. IR (KBr) ν=2919, 1328, 785 cm⁻¹; MS: *m/z* 198 (MH)⁺. HRMS calcd for C₁₄H₁₆N: 198.1283 [M+H], found: 198.1276.

4.2.7. (*2E,4E*)-5-(4-Methoxyphenyl)-2,3-dimethylpenta-2,4-dienitrile (5g**).** Yield: 92 mg, 86%. White solid, mp: 86–88 °C; ¹H NMR (400 MHz, CDCl₃) δ: 2.07 (s, 3H), 2.26 (s, 3H), 3.83 (s, 3H), 6.82 (d,

$J=15.6$ Hz, 1H), 6.89 (d, $J=8.0$ Hz, 2H), 6.97 (d, $J=16.0$ Hz, 1H), 7.42 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 15.8, 18.4, 55.3, 104.6, 114.3, 120.1, 121.8, 128.6, 129.0, 134.8, 148.1, 160.4. IR (KBr) $\nu=2929$, 2197, 1252, 805 cm^{-1} ; MS: m/z 214 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{16}\text{NO}$: 214.1232 [M+H], found: 214.1238.

4.2.8. (2E,4E)-2,3-Dimethyl-5-(4-(trifluoromethyl)phenyl)penta-2,4-dienitrile (5h). Yield: 107 mg, 85%. White solid, mp: 116–118 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 2.12 (s, 3H), 2.29 (s, 3H), 6.90 (d, $J=16.0$ Hz, 1H), 7.18 (d, $J=16.0$ Hz, 1H), 7.57 (d, $J=8.4$ Hz, 2H), 7.62 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 16.0, 18.4, 107.6, 120.3, 125.75, 125.78, 126.1, 127.3, 130.3, 133.5, 139.6, 147.2. IR (KBr) $\nu=2928$, 2201, 1324, 827 cm^{-1} ; MS: m/z 252 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{13}\text{F}_3\text{N}$: 252.1000 [M+H], found: 252.0995.

4.2.9. (2E,4E)-2-Benzyl-3-methyl-5-phenylpenta-2,4-dienitrile (5i). Yield: 108 mg, 83%. White solid, mp: 99–101 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 2.39 (s, 3H), 3.83 (s, 2H), 7.00 (d, $J=16.0$ Hz, 1H), 7.29–7.42 (m, 9H), 7.50 (d, $J=7.2$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.8, 35.5, 110.6, 120.1, 123.7, 127.0, 127.3, 128.4, 128.87, 128.94, 129.2, 136.1, 136.2, 137.5, 148.4. IR (KBr) $\nu=2928$, 2206, 760 cm^{-1} ; MS: m/z 260 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{18}\text{N}$: 260.1439 [M+H], found: 260.1448.

4.2.10. (2E,4E)-2-Benzyl-5-(3-chlorophenyl)-3-methylpenta-2,4-dienitrile (5j). Yield: 126 mg, 86%. White solid, mp: 122–123 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 2.36 (s, 3H), 3.82 (s, 2H), 6.90 (d, $J=16.0$ Hz, 1H), 7.25–7.38 (m, 9H), 7.45 (s, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.9, 35.5, 111.6, 119.9, 124.9, 125.5, 127.0, 127.1, 128.4, 128.9, 129.0, 130.2, 134.7, 134.9, 137.2, 137.9, 147.9. IR (KBr) $\nu=2931$, 2203, 705 cm^{-1} ; MS: m/z 294 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{17}\text{ClN}$: 294.1050 [M+H], found: 294.1058.

4.2.11. (2E,4E)-2-Benzyl-5-(4-bromophenyl)-3-methylpenta-2,4-dienitrile (5k). Yield: 144 mg, 85%. White solid, mp: 130–131 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 2.36 (s, 3H), 3.80 (s, 2H), 6.89 (d, $J=16.0$ Hz, 1H), 7.23–7.36 (m, 8H), 7.49 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.8, 35.5, 111.1, 120.0, 123.2, 124.3, 127.1, 128.3, 128.6, 128.9, 132.1, 134.8, 135.0, 137.2, 148.1. IR (KBr) $\nu=3023$, 2206, 1486, 723 cm^{-1} ; MS: m/z 338 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{17}\text{BrN}$: 338.0544 [M+H], found: 338.0547.

4.2.12. (2E,4E)-2-Allyl-3-methyl-5-phenylpenta-2,4-dienitrile (5l). Yield: 86 mg, 82%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 2.33 (s, 3H), 3.20–3.22 (m, 2H), 5.18–5.22 (m, 2H), 5.84–5.92 (m, 1H), 6.93 (d, $J=16.0$ Hz, 1H), 7.12 (d, $J=15.6$ Hz, 1H), 7.31–7.40 (m, 3H), 7.47 (d, $J=7.6$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.7, 33.8, 109.1, 117.5, 119.9, 123.7, 127.2, 128.9, 129.1, 133.2, 135.8, 136.1, 148.6. IR (KBr) $\nu=2933$, 2210, 910, 768 cm^{-1} ; MS: m/z 210 (MH) $^+$. HRMS calcd for $\text{C}_{15}\text{H}_{16}\text{N}$: 210.1283 [M+H], found: 210.1286.

4.2.13. (2E,4E)-2-Allyl-5-(4-fluorophenyl)-3-methylpenta-2,4-dienitrile (5m). Yield: 100 mg, 88%. Yellow solid, mp: 84–86 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 2.31 (s, 3H), 3.18–3.20 (m, 2H), 5.17–5.21 (m, 2H), 5.83–5.89 (m, 1H), 6.88 (d, $J=16.0$ Hz, 1H), 7.00–7.08 (m, 3H), 7.43–7.46 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.7, 33.7, 109.1, 115.9, 116.1, 117.6, 119.9, 123.4, 128.86, 128.94, 132.30, 132.33, 133.1, 134.5, 148.4, 161.8, 164.4. IR (KBr) $\nu=2931$, 2198, 912, 823 cm^{-1} ; MS: m/z 228 (MH) $^+$. HRMS calcd for $\text{C}_{15}\text{H}_{15}\text{FN}$: 228.1189 [M+H], found: 228.1187.

4.2.14. (2E,4E)-2-Allyl-3-methyl-5-(4-(trifluoromethyl)phenyl)penta-2,4-dienitrile (5n). Yield: 118 mg, 85%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 2.34 (s, 3H), 3.21–3.23 (m, 2H), 5.18–5.22 (m, 2H), 5.83–5.89 (m, 1H), 6.93 (d, $J=15.6$ Hz, 1H), 7.18 (d, $J=16.0$ Hz, 1H), 7.56 (d, $J=8.0$ Hz, 2H), 7.62 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.7, 33.8, 110.8, 117.8, 119.6, 125.8, 125.9, 127.3,

131.2, 132.9, 134.0, 139.5, 147.9. IR (KBr) $\nu=2929$, 2201, 915, 822 cm^{-1} ; MS: m/z 278 (MH) $^+$. HRMS calcd for $\text{C}_{16}\text{H}_{15}\text{F}_3\text{N}$: 278.1157 [M+H], found: 278.1152.

4.2.15. (2E,4E)-2,3,4-Trimethyl-5-phenylpenta-2,4-dienitrile (5o). Yield: 80 mg, 81%. White solid, mp: 114–115 $^{\circ}\text{C}$; ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ : 1.57 (s, 3H), 1.92 (s, 3H), 2.11 (s, 3H), 6.40 (s, 1H), 7.16–7.22 (m, 3H), 7.29–7.33 (m, 2H). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$) δ : 16.7, 21.2, 23.1, 104.7, 119.6, 127.5, 127.6, 127.7, 129.0, 136.0, 136.7, 155.7. IR (KBr) $\nu=2921$, 2210, 772 cm^{-1} ; MS: m/z 198 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{16}\text{N}$: 198.1283 [M+H], found: 198.1288.

4.2.16. (2E,4E)-3-Methyl-2-phenyl-5-(4-(trifluoromethyl)phenyl)penta-2,4-dienitrile (5p). Yield: 35 mg, 22%. White solid, mp: 161–163 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 2.50 (s, 3H), 6.98 (d, $J=16.0$ Hz, 1H), 7.14 (d, $J=16.0$ Hz, 1H), 7.38–7.47 (m, 7H), 7.56 (d, $J=8.0$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 19.0, 114.3, 119.2, 125.7, 125.8, 127.3, 127.5, 127.8, 128.7, 128.8, 129.0, 129.4, 129.9, 130.3, 133.4, 134.2, 139.4, 148.6. IR (KBr) $\nu=2929$, 2195, 822, 766 cm^{-1} ; MS: m/z 314 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{15}\text{F}_3\text{N}$: 314.1157 [M+H], found: 314.1166.

4.3. Typical procedure for the preparation of 5,6-dihydro-3,4-dimethyl-6-phenylpyran-2-one (6a) (Table 2)

To a flask containing (Z)-5-hydroxy-2,3-dimethyl-5-phenylpenta-2-enitrile (101 mg, 0.5 mmol) in CH_2Cl_2 (5 mL) was added H_2SO_4 (27 μL , 0.5 mmol). The solution was stirred under reflux. Upon completion, the reaction was quenched with aqueous Na_2CO_3 . The resulting mixture was extracted with ethyl acetate (5 mL $\times$ 3). The combined organic phases were washed with brine, dried, filtered, and concentrated under vacuum. The residue was purified by column chromatography on silica gel eluting with ethyl acetate/hexane (1:10) to give 5,6-dihydro-3,4-dimethyl-6-phenylpyran-2-one (6a) in a yield of 75% (76 mg, 0.38 mmol). Other 5,6-dihydropyran-2-one derivatives were obtained in a similar manner.

4.3.1. 3,4-Dimethyl-6-phenyl-5,6-dihydro-2H-pyran-2-one (6a). Yield: 76 mg, 75%. White solid, mp: 98–99 $^{\circ}\text{C}$ (lit.⁶¹ 97–97.5 $^{\circ}\text{C}$); ^1H NMR (400 MHz, CDCl_3) δ : 1.94 (s, 3H), 1.97 (s, 3H), 2.40–2.45 (m, 1H), 2.66–2.74 (m, 1H), 5.32–5.36 (m, 1H), 7.31–7.40 (m, 5H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.5, 20.2, 38.2, 122.4, 126.0, 128.4, 128.5, 138.9, 148.6, 166.2. IR (KBr) $\nu=2919$, 1692, 1293, 772 cm^{-1} ; MS: m/z 203 (MH) $^+$. HRMS calcd for $\text{C}_{13}\text{H}_{15}\text{O}_2$: 203.1072 [M+H], found: 203.1077.

4.3.2. 3,4-Dimethyl-6-(3-chloro)phenyl-5,6-dihydro-2H-pyran-2-one (6b). Yield: 96 mg, 81%. White solid, mp: 90–91 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 1.89 (s, 3H), 1.94 (s, 3H), 2.37–2.42 (m, 1H), 2.56–2.64 (m, 1H), 5.24–5.28 (m, 1H), 7.23–7.26 (m, 3H), 7.37 (s, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.5, 20.2, 38.0, 122.3, 124.1, 126.1, 128.4, 129.9, 134.4, 140.9, 148.7, 165.9. IR (KBr) $\nu=2917$, 1700, 786 cm^{-1} ; MS: m/z 237 (MH) $^+$. HRMS calcd for $\text{C}_{13}\text{H}_{14}\text{ClO}_2$: 237.0682 [M+H], found: 237.0678.

4.3.3. 3,4-Dimethyl-6-(4-chloro)phenyl-5,6-dihydro-2H-pyran-2-one (6c). Yield: 100 mg, 85%. White solid, mp: 110–111 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 1.92 (s, 3H), 1.96 (s, 3H), 2.38–2.43 (m, 1H), 2.60–2.68 (m, 1H), 5.28–5.32 (m, 1H), 7.32–7.35 (m, 4H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.5, 20.2, 38.1, 122.5, 127.3, 128.7, 134.1, 137.4, 148.4, 165.9. IR (KBr) $\nu=2925$, 1711, 814 cm^{-1} ; MS: m/z 237 (MH) $^+$. HRMS calcd for $\text{C}_{13}\text{H}_{14}\text{ClO}_2$: 237.0682 [M+H], found: 237.0676.

4.3.4. 3,4-Dimethyl-6-(3-bromo)phenyl-5,6-dihydro-2H-pyran-2-one (6d). Yield: 110 mg, 78%. White solid, mp: 97–98 $^{\circ}\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ : 1.92 (s, 3H), 1.96 (s, 3H), 2.40–2.44 (m, 1H),

2.60–2.68 (m, 1H), 5.27–5.31 (m, 1H), 7.21–7.26 (m, 1H), 7.30 (d, $J=7.6$ Hz, 1H), 7.44 (d, $J=7.2$ Hz, 1H), 7.55 (s, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.5, 20.2, 38.1, 122.5, 122.6, 124.5, 129.0, 130.2, 131.4, 141.1, 148.5, 165.8. IR (KBr) $\nu=2920, 1701, 780\text{ cm}^{-1}$; MS: m/z 281 (MH) $^+$. HRMS calcd for $\text{C}_{13}\text{H}_{14}\text{BrO}_2$: 281.0177 [M+H], found: 281.0186.

4.3.5. 3,4-Dimethyl-6-(4-bromophenyl)-5,6-dihydro-2H-pyran-2-one (6e). Yield: 121 mg, 86%. White solid, mp: 119–120 °C; ^1H NMR (400 MHz, CDCl_3) δ : 1.86 (s, 3H), 1.91 (s, 3H), 2.33–2.38 (m, 1H), 2.53–2.60 (m, 1H), 5.21–5.25 (m, 1H), 7.20 (d, $J=8.4$ Hz, 2H), 7.42 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.5, 20.2, 38.0, 122.2, 122.3, 127.7, 131.6, 138.0, 148.8, 165.9. IR (KBr) $\nu=2919, 1711, 1135, 819\text{ cm}^{-1}$; MS: m/z 281 (MH) $^+$. HRMS calcd for $\text{C}_{13}\text{H}_{14}\text{BrO}_2$: 281.0177 [M+H], found: 281.0183.

4.3.6. 3,4-Dimethyl-6-(3-methylphenyl)-5,6-dihydro-2H-pyran-2-one (6f). Yield: 78 mg, 72%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 1.92 (s, 3H), 1.95 (s, 3H), 2.33 (s, 3H), 2.37–2.42 (m, 1H), 2.62–2.70 (m, 1H), 5.26–5.30 (m, 1H), 7.10–7.16 (m, 2H), 7.21–7.25 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.5, 20.2, 21.4, 38.2, 122.3, 123.0, 126.6, 128.4, 129.1, 138.2, 138.8, 148.9, 166.3. IR (KBr) $\nu=2928, 1708, 785\text{ cm}^{-1}$; MS: m/z 217 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{17}\text{O}_2$: 217.1229 [M+H], found: 217.1236.

4.3.7. 3,4-Dimethyl-6-(4-(trifluoromethyl)phenyl)-5,6-dihydro-2H-pyran-2-one (6g). Yield: 119 mg, 88%. White solid, mp: 124–125 °C; ^1H NMR (400 MHz, CDCl_3) δ : 1.91 (s, 3H), 1.96 (s, 3H), 2.42–2.47 (m, 1H), 2.59–2.67 (m, 1H), 5.36–5.40 (m, 1H), 7.50 (d, $J=7.6$ Hz, 2H), 7.60 (d, $J=8.0$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.4, 20.1, 38.0, 122.5, 122.6, 125.3, 125.48, 125.51, 126.2, 130.3, 130.6, 142.9, 148.5, 165.7. IR (KBr) $\nu=2918, 1693, 1330, 836\text{ cm}^{-1}$; MS: m/z 271 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{14}\text{F}_3\text{O}_2$: 271.0946 [M+H], found: 271.0952.

4.3.8. 3-Benzyl-4-methyl-6-phenyl-5,6-dihydro-2H-pyran-2-one (6h). Yield: 95 mg, 68%. White solid, mp: 85–86 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.05 (s, 3H), 2.48–2.53 (m, 1H), 2.75–2.83 (m, 1H), 3.74–3.90 (m, 2H), 5.36–5.39 (m, 1H), 7.18–7.30 (m, 5H), 7.32–7.39 (m, 5H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.6, 32.3, 38.3, 77.5, 126.0, 126.1, 126.4, 128.3, 128.4, 128.6, 138.7, 139.6, 150.2, 165.7. IR (KBr) $\nu=2918, 1698, 770\text{ cm}^{-1}$; MS: m/z 279 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{19}\text{O}_2$: 279.1385 [M+H], found: 279.1388.

4.3.9. 3-Benzyl-6-(3-chlorophenyl)-4-methyl-5,6-dihydro-2H-pyran-2-one (6i). Yield: 125 mg, 80%. White solid, mp: 101–102 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.04 (s, 3H), 2.46–2.51 (m, 1H), 2.68–2.75 (m, 1H), 3.73–3.88 (m, 2H), 5.30–5.34 (m, 1H), 7.18–7.32 (m, 8H), 7.41 (s, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.7, 32.3, 38.1, 76.7, 124.1, 126.2, 126.3, 128.3, 128.5, 128.6, 128.8, 130.0, 134.5, 139.4, 140.7, 150.4, 165.4. IR (KBr) $\nu=3026, 2937, 1703, 724\text{ cm}^{-1}$; MS: m/z 313 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{18}\text{ClO}_2$: 313.0995 [M+H], found: 313.0988.

4.3.10. 3-Benzyl-6-(4-bromophenyl)-4-methyl-5,6-dihydro-2H-pyran-2-one (6j). Yield: 141 mg, 79%. White solid, mp: 120–121 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.05 (s, 3H), 2.46–2.51 (m, 1H), 2.69–2.76 (m, 1H), 3.73–3.88 (m, 2H), 5.31–5.35 (m, 1H), 7.17–7.27 (m, 7H), 7.50 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.6, 32.3, 38.2, 122.4, 126.1, 126.5, 127.6, 128.3, 128.5, 131.7, 137.8, 139.4, 150.0, 165.4. IR (KBr) $\nu=2925, 1712, 1123, 819\text{ cm}^{-1}$; MS: m/z 357 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{18}\text{BrO}_2$: 357.0490 [M+H], found: 357.0479.

4.3.11. 3-Benzyl-4-methyl-6-*p*-tolyl-5,6-dihydro-2H-pyran-2-one (6k). Yield: 75 mg, 51%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 2.05 (s, 3H), 2.37 (s, 3H), 2.45–2.51 (m, 1H), 2.75–2.82 (m, 1H), 3.75–3.90 (m, 2H), 5.33–5.37 (m, 1H), 7.18–7.30 (m, 9H). ^{13}C NMR (100 MHz,

CDCl_3) δ : 20.6, 21.2, 32.3, 38.3, 77.5, 126.0, 126.1, 126.3, 128.3, 128.5, 129.2, 135.8, 138.2, 139.6, 150.4, 165.9. IR (KBr) $\nu=2928, 1710, 1118, 815\text{ cm}^{-1}$; MS: m/z 293 (MH) $^+$. HRMS calcd for $\text{C}_{20}\text{H}_{21}\text{O}_2$: 293.1542 [M+H], found: 293.1552.

4.3.12. 3-Allyl-4-methyl-6-phenyl-5,6-dihydro-2H-pyran-2-one (6l). Yield: 82 mg, 72%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 1.97 (s, 3H), 2.42–2.47 (m, 1H), 2.67–2.75 (m, 1H), 3.10–3.24 (m, 2H), 5.00–5.06 (m, 2H), 5.32–5.36 (m, 1H), 5.81–5.88 (m, 1H), 7.26–7.39 (m, 5H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.0, 30.9, 38.3, 77.7, 115.2, 124.6, 126.0, 128.4, 128.5, 134.9, 138.8, 150.5, 165.5. IR (KBr) $\nu=2930, 1699, 773\text{ cm}^{-1}$; MS: m/z 229 (MH) $^+$. HRMS calcd for $\text{C}_{15}\text{H}_{17}\text{O}_2$: 229.1229 [M+H], found: 229.1221.

4.3.13. 3-Allyl-6-(4-fluorophenyl)-4-methyl-5,6-dihydro-2H-pyran-2-one (6m). Yield: 97 mg, 79%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 2.00 (s, 3H), 2.41–2.47 (m, 1H), 2.70–2.77 (m, 1H), 3.11–3.25 (m, 2H), 5.02–5.07 (m, 2H), 5.33–5.36 (m, 1H), 5.82–5.88 (m, 1H), 7.05–7.09 (m, 2H), 7.37–7.40 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.1, 30.9, 38.4, 77.1, 115.3, 115.4, 115.6, 124.8, 127.8, 127.9, 134.5, 134.8, 150.3, 165.4. IR (KBr) $\nu=2931, 1711, 825\text{ cm}^{-1}$; MS: m/z 247 (MH) $^+$. HRMS calcd for $\text{C}_{15}\text{H}_{16}\text{FO}_2$: 247.1134 [M+H], found: 247.1122.

4.3.14. 3-Allyl-4-methyl-6-(4-(trifluoromethyl)phenyl)-5,6-dihydro-2H-pyran-2-one (6n). Yield: 127 mg, 86%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 2.00 (s, 3H), 2.47–2.52 (m, 1H), 2.66–2.74 (m, 1H), 3.11–3.24 (m, 2H), 5.01–5.06 (m, 2H), 5.41–5.45 (m, 1H), 5.79–5.89 (m, 1H), 7.53 (d, $J=8.4$ Hz, 2H), 7.64 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.0, 30.8, 38.2, 76.8, 115.4, 124.8, 125.6, 126.2, 134.7, 142.7, 150.2, 165.0. IR (KBr) $\nu=2931, 1693, 838\text{ cm}^{-1}$; MS: m/z 297 (MH) $^+$. HRMS calcd for $\text{C}_{16}\text{H}_{16}\text{F}_3\text{O}_2$: 297.1102 [M+H], found: 297.1108.

4.3.15. 3,4,5-Trimethyl-6-phenyl-5,6-dihydro-2H-pyran-2-one (6o). Yield: 86 mg, 80%. White solid, mp: 94–95 °C; ^1H NMR (400 MHz, CDCl_3) δ : 1.14 (d, $J=7.2$ Hz, 3H), 1.88 (s, 6H), 2.68 (t, $J=6.8$ Hz, 1H), 5.05 (d, $J=6.4$ Hz, 1H), 7.27–7.35 (m, 5H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.8, 15.9, 18.2, 39.5, 83.1, 122.1, 126.6, 128.2, 128.4, 138.8, 152.2, 165.3. IR (KBr) $\nu=2911, 1699, 771\text{ cm}^{-1}$; MS: m/z 217 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{17}\text{O}_2$: 217.1229 [M+H], found: 217.1222.

4.3.16. 6-(4-Bromophenyl)-3,4,5-trimethyl-5,6-dihydro-2H-pyran-2-one (6p). Yield: 121 mg, 82%. White solid, mp: 152–154 °C; ^1H NMR (400 MHz, CDCl_3) δ : 1.07 (d, $J=7.2$ Hz, 3H), 1.83 (s, 6H), 2.60 (t, $J=6.8$ Hz, 1H), 4.97 (d, $J=6.8$ Hz, 1H), 7.12 (d, $J=8.4$ Hz, 2H), 7.41 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ : 12.7, 15.8, 18.3, 39.4, 82.3, 122.0, 122.2, 128.4, 131.5, 137.9, 152.3, 165.1. IR (KBr) $\nu=2918, 1698, 770\text{ cm}^{-1}$; MS: m/z 295 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{16}\text{BrO}_2$: 295.0334 [M+H], found: 295.0341.

4.3.17. 4-Methyl-3,6-diphenyl-5,6-dihydro-2H-pyran-2-one (6q). Yield: 90 mg, 68%. White solid, mp: 106–107 °C; ^1H NMR (400 MHz, CDCl_3) δ : 1.91 (s, 3H), 2.60–2.65 (m, 1H), 2.86–2.93 (m, 1H), 5.53–5.57 (m, 1H), 7.32–7.48 (m, 10H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.5, 38.2, 77.8, 126.1, 127.7, 128.1, 128.5, 128.6, 128.9, 130.0, 134.6, 138.7, 151.4, 164.9. IR (KBr) $\nu=2915, 1701, 773\text{ cm}^{-1}$; MS: m/z 265 (MH) $^+$. HRMS calcd for $\text{C}_{18}\text{H}_{17}\text{O}_2$: 265.1229 [M+H], found: 265.1232.

4.3.18. 6-(3-Chlorophenyl)-4-methyl-3-phenyl-5,6-dihydro-2H-pyran-2-one (6r). Yield: 109 mg, 73%. White solid, mp: 120–121 °C; ^1H NMR (400 MHz, CDCl_3) δ : 1.88 (s, 3H), 2.58–2.83 (m, 2H), 5.47–5.50 (m, 1H), 7.24 (d, $J=6.8$ Hz, 2H), 7.32–7.40 (m, 6H), 7.48 (s, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.5, 38.0, 77.0, 124.2, 126.2, 127.8, 128.2, 128.60, 128.64, 130.01, 130.04, 134.4, 134.5, 140.8, 151.7, 164.6.

IR (KBr) ν =3058, 2909, 1703, 788 cm^{-1} ; MS: m/z 299 (MH)⁺. HRMS calcd for C₁₈H₁₆ClO₂: 299.0839 [M+H], found: 299.0838.

4.3.19. 6-(4-Chlorophenyl)-4-methyl-3-phenyl-5,6-dihydro-2H-pyran-2-one (6s). Yield: 116 mg, 78%. White solid, mp: 128–130 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.90 (s, 3H), 2.58–2.63 (m, 1H), 2.80–2.87 (m, 1H), 5.49–5.53 (m, 1H), 7.23–7.25 (m, 2H), 7.32–7.41 (m, 7H). ¹³C NMR (100 MHz, CDCl₃) δ : 21.5, 38.1, 127.4, 127.8, 128.2, 128.8, 129.9, 134.3, 134.4, 137.2, 151.3, 164.6. IR (KBr) ν =2922, 1703, 827 cm^{-1} ; MS: m/z 299 (MH)⁺. HRMS calcd for C₁₈H₁₆ClO₂: 299.0839 [M+H], found: 299.0832.

4.3.20. 6-(4-Bromophenyl)-4-methyl-3-phenyl-5,6-dihydro-2H-pyran-2-one (6t). Yield: 146 mg, 85%. White solid, mp: 149–150 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.90 (s, 3H), 2.58–2.63 (m, 1H), 2.78–2.82 (m, 1H), 5.49–5.52 (m, 1H), 7.22–7.24 (m, 2H), 7.34–7.40 (m, 5H), 7.52–7.55 (m, 2H). ¹³C NMR (100 MHz, CDCl₃) δ : 21.5, 38.1, 122.4, 127.7, 127.8, 128.2, 128.8, 130.0, 131.8, 134.4, 137.8, 151.4, 164.6. IR (KBr) ν =2916, 1699, 826 cm^{-1} ; MS: m/z 343 (MH)⁺. HRMS calcd for C₁₈H₁₆BrO₂: 343.0334 [M+H], found: 343.0349.

4.3.21. 4-Methyl-3-phenyl-6-m-tolyl-5,6-dihydro-2H-pyran-2-one (6u). Yield: 71 mg, 51%. White solid, mp: 116–118 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.91 (s, 3H), 2.39 (s, 3H), 2.59–2.64 (m, 1H), 2.85–2.93 (m, 1H), 5.50–5.54 (m, 1H), 7.18 (d, J =7.2 Hz, 1H), 7.25–7.42 (m, 8H). ¹³C NMR (100 MHz, CDCl₃) δ : 21.4, 21.5, 38.3, 77.9, 123.1, 126.7, 127.7, 128.1, 128.5, 128.8, 129.2, 130.0, 134.6, 138.4, 138.6, 151.4, 164.9. IR (KBr) ν =2930, 1708, 771 cm^{-1} ; MS: m/z 279 (MH)⁺. HRMS calcd for C₁₉H₁₉O₂: 279.1385 [M+H], found: 279.1381.

4.3.22. 4-Methyl-3-phenyl-6-p-tolyl-5,6-dihydro-2H-pyran-2-one (6v). Yield: 67 mg, 48%. White solid, mp: 129–130 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.90 (s, 3H), 2.38 (s, 3H), 2.58–2.63 (m, 1H), 2.85–2.92 (m, 1H), 5.50–5.53 (m, 1H), 7.21–7.26 (m, 4H), 7.32–7.42 (m, 5H). ¹³C NMR (100 MHz, CDCl₃) δ : 21.2, 21.5, 38.2, 77.8, 126.0, 127.7, 128.1, 128.8, 129.3, 130.0, 134.6, 135.7, 138.3, 151.4, 165.0. IR (KBr) ν =2922, 1698, 718 cm^{-1} ; MS: m/z 279 (MH)⁺. HRMS calcd for C₁₉H₁₉O₂: 279.1385 [M+H], found: 279.1376.

4.3.23. 4-Methyl-3-phenyl-6-(4-(trifluoromethyl)-phenyl)-5,6-dihydro-2H-pyran-2-one (6w). Yield: 150 mg, 90%. White solid, mp: 123–124 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.92 (s, 3H), 2.63–2.68 (m, 1H), 2.82–2.89 (m, 1H), 5.60–5.63 (m, 1H), 7.24 (d, J =6.8 Hz, 1H), 7.35–7.43 (m, 3H), 7.60 (d, J =8.0 Hz, 2H), 7.68 (d, J =8.0 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ : 21.6, 38.2, 76.9, 125.6, 125.7, 126.2, 127.9, 128.2, 128.9, 129.9, 134.2, 142.6, 151.2, 164.4. IR (KBr) ν =2947, 2867, 1710, 1130, 836, 701 cm^{-1} ; MS: m/z 333 (MH)⁺. HRMS calcd for C₁₉H₁₆F₃O₂: 333.1102 [M+H], found: 333.1089.

4.3.24. 6-(4-Bromophenyl)-4,5-dimethyl-3-phenyl-5,6-dihydro-2H-pyran-2-one (6x). Yield: 153 mg, 86%. White solid, mp: 138–140 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.26 (d, J =6.8 Hz, 3H), 1.79 (s, 3H), 2.80–2.84 (m, 1H), 5.21 (d, J =6.8 Hz, 1H), 7.15 (d, J =7.2 Hz, 2H), 7.26–7.39 (m, 5H), 7.53 (d, J =8.4 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ : 15.9, 19.7, 39.7, 82.5, 122.4, 127.8, 128.2, 128.4, 128.6, 129.9, 131.7, 134.7, 137.7, 155.0, 163.8. IR (KBr) ν =2921, 1711, 1010, 697 cm^{-1} ; MS: m/z 357 (MH)⁺. HRMS calcd for C₁₉H₁₈BrO₂: 357.0490 [M+H], found: 357.0498.

4.3.25. 4-Benzyl-3,6-dimethyl-5,6-dihydro-2H-pyran-2-one (6y). Yield: 88 mg, 81%. Syrup; ¹H NMR (400 MHz, CDCl₃) δ : 1.32 (d, J =6.0 Hz, 3H), 2.06 (s, 3H), 2.15–2.20 (m, 2H), 3.54–3.67 (m, 2H), 4.38–4.42 (m, 2H), 7.14 (d, J =7.2 Hz, 1H), 7.26–7.34 (m, 3H). ¹³C NMR (100 MHz, CDCl₃) δ : 12.8, 20.6, 35.1, 39.5, 73.0, 122.9, 126.9, 128.6, 128.9, 137.0,

150.2, 166.8. IR (KBr) ν =3020, 2926, 1708, 708 cm^{-1} ; MS: m/z 217 (MH)⁺. HRMS calcd for C₁₄H₁₇O₂: 217.1229 [M+H], found: 217.1226.

4.3.26. 3,4-Dibenzyl-6-methyl-5,6-dihydro-2H-pyran-2-one (6z). Yield: 111 mg, 76%. Syrup; ¹H NMR (400 MHz, CDCl₃) δ : 1.33 (d, J =6.4 Hz, 3H), 2.15–2.25 (m, 2H), 3.68 (s, 2H), 3.89–3.99 (m, 2H), 4.41–4.46 (m, 1H), 7.01–7.03 (m, 2H), 7.18–7.31 (m, 8H). ¹³C NMR (100 MHz, CDCl₃) δ : 20.6, 32.2, 35.1, 39.6, 72.9, 126.2, 126.7, 127.0, 128.3, 128.5, 128.78, 128.85, 136.6, 139.8, 152.0, 166.4. IR (KBr) ν =3024, 2929, 1709, 706 cm^{-1} ; MS: m/z 293 (MH)⁺. HRMS calcd for C₂₀H₂₁O₂: 293.1542 [M+H], found: 293.1546.

4.3.27. 4-Benzyl-3-methyl-6-phenyl-5,6-dihydro-2H-pyran-2-one (6aa). Yield: 100 mg, 72%. White solid, mp: 112–113 °C; ¹H NMR (400 MHz, CDCl₃) δ : 2.14 (s, 1H), 2.36–2.42 (m, 1H), 2.52–2.60 (m, 1H), 3.62–3.71 (m, 2H), 5.28–5.32 (m, 1H), 7.14 (d, J =7.6 Hz, 2H), 7.24–7.35 (m, 8H). ¹³C NMR (100 MHz, CDCl₃) δ : 12.9, 35.8, 39.6, 78.0, 123.3, 126.0, 127.0, 128.4, 128.5, 128.6, 128.9, 136.8, 138.7, 150.2, 166.4. IR (KBr) ν =3025, 2932, 1700, 708 cm^{-1} ; MS: m/z 279 (MH)⁺. HRMS calcd for C₁₉H₁₉O₂: 279.1385 [M+H], found: 279.1375.

4.4. Typical procedure for the preparation of 7

To a flask containing (*E*)-5-hydroxy-3-methyl-2-phenyl-5-(4-(trifluoromethyl)phenyl)pent-2-enenitrile (166 mg, 0.5 mmol) in CH₂Cl₂ (5 mL) was added H₂SO₄ (27 μ L, 0.5 mmol). The solution was stirred under reflux. Upon completion, the mixture was neutralized with aqueous Na₂CO₃, and then extracted with ethyl acetate (5 mL $\times$ 3). The combined organic phases were washed with brine, dried, filtered, and concentrated under vacuum. The residue was purified by column chromatography on silica gel eluting with ethyl acetate/hexane (1:20) to give 2-methyl-4-(4-(trifluoromethyl)phenyl)-3,4-dihydronaphthalene-1-carbonitrile (7) in a yield of 68% (107 mg, 0.34 mmol).

4.4.1. 2-Methyl-4-(4-(trifluoromethyl)phenyl)-3,4-dihydronaphthalene-1-carbonitrile (7). Yield: 107 mg, 68%. Syrup; ¹H NMR (400 MHz, CDCl₃) δ : 2.24 (s, 3H), 2.69–2.86 (m, 2H), 4.24 (t, J =7.6 Hz, 1H), 6.86 (d, J =7.2 Hz, 1H), 7.19–7.24 (m, 3H), 7.34 (t, J =7.6 Hz, 1H), 7.56–7.59 (m, 3H). ¹³C NMR (100 MHz, CDCl₃) δ : 23.4, 38.2, 42.7, 109.6, 116.2, 124.9, 125.7, 127.9, 128.4, 128.6, 129.2, 129.6, 134.6, 146.9, 152.9. MS: m/z 314 (MH)⁺. HRMS calcd for C₁₉H₁₅F₃N: 314.1157 [M+H], found: 314.1158.

4.5. Typical procedure for the one-pot preparation of 8a (Table 4)

To a flask containing (*E*)-5-hydroxy-3-methyl-2-phenyl-5-(4-(trifluoromethyl)phenyl)pent-2-enenitrile (166 mg, 0.5 mmol) in CH₂Cl₂ (5 mL) was added H₂SO₄ (27 μ L, 0.5 mmol). The solution was stirred under reflux. Upon complete consumption of the starting material as indicated by TLC analysis, DDQ (114 mg, 0.5 mmol) was added. The mixture was stirred under reflux. Upon completion, the mixture was filtered through Celite and the filter cake was washed twice with CH₂Cl₂ (5 mL). The combined filtrate was neutralized with aqueous Na₂CO₃, and then extracted with ethyl acetate (5 mL $\times$ 3). The combined organic phases were washed with brine, dried, filtered, and concentrated under vacuum. The residue was purified by column chromatography on silica gel eluting with ethyl acetate/hexane (1:20) to give 2-methyl-4-(4-(trifluoromethyl)phenyl)-1-naphthonitrile (8a) in a yield of 63% (98 mg, 0.32 mmol). Other naphthalene-1-carbonitriles were obtained in a similar manner.

4.5.1. 2-Methyl-4-(4-(trifluoromethyl)phenyl)-1-naphthonitrile (8a). Yield: 98 mg, 63%. White solid, mp: 156–157 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.80 (s, 3H), 7.35 (s, 1H), 7.52 (t, $J=7.6$ Hz, 1H), 7.59 (d, $J=7.6$ Hz, 2H), 7.70 (t, $J=7.6$ Hz, 1H), 7.78–7.80 (m, 3H), 8.30 (d, $J=8.0$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.3, 109.5, 116.9, 125.38, 125.45, 125.49, 126.2, 127.0, 128.6, 128.7, 129.4, 130.1, 130.6, 133.2, 142.3, 142.8, 143.4. IR (KBr) $\nu=2921, 2216, 818\text{ cm}^{-1}$; MS: m/z 312 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{13}\text{F}_3\text{N}$: 312.1000 [M+H], found: 312.1011.

4.5.2. 4-(3-Chlorophenyl)-2-methyl-1-naphthonitrile (8b). Yield: 72 mg, 52%. White solid, mp: 114–115 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.78 (s, 3H), 7.34 (s, 2H), 7.45–7.46 (m, 3H), 7.51 (t, $J=8.0$ Hz, 1H), 7.67 (t, $J=7.6$ Hz, 1H), 7.84 (d, $J=8.8$ Hz, 1H), 8.27 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.3, 109.2, 117.1, 125.3, 126.4, 126.9, 127.9, 128.3, 128.6, 128.7, 129.4, 129.7, 129.8, 133.2, 134.5, 140.9, 142.3, 143.4. IR (KBr) $\nu=3059, 2921, 2218, 799, 765\text{ cm}^{-1}$; MS: m/z 278 (MH) $^+$. HRMS calcd for $\text{C}_{18}\text{H}_{13}\text{ClN}$: 278.0737 [M+H], found: 278.0745.

4.5.3. 4-(4-Chlorophenyl)-2-methyl-1-naphthonitrile (8c). Yield: 74 mg, 53%. White solid, mp: 148–149 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.78 (s, 3H), 7.33 (s, 1H), 7.39 (d, $J=8.4$ Hz, 2H), 7.48–7.52 (m, 3H), 7.68 (t, $J=7.6$ Hz, 1H), 7.84 (d, $J=8.4$ Hz, 1H), 8.27 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.3, 109.0, 117.1, 125.3, 126.4, 126.8, 128.5, 128.68, 128.74, 129.5, 131.0, 133.3, 134.3, 137.5, 142.3, 143.7. IR (KBr) $\nu=2923, 2218, 813\text{ cm}^{-1}$; MS: m/z 278 (MH) $^+$. HRMS calcd for $\text{C}_{18}\text{H}_{13}\text{ClN}$: 278.0737 [M+H], found: 278.0743.

4.5.4. 4-(4-Bromophenyl)-2-methyl-1-naphthonitrile (8d). Yield: 90 mg, 56%. White solid, mp: 164–165 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.78 (s, 3H), 7.32–7.34 (m, 3H), 7.51 (t, $J=7.6$ Hz, 1H), 7.64–7.70 (m, 3H), 7.83 (d, $J=8.4$ Hz, 1H), 8.28 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.3, 109.1, 117.1, 122.5, 125.3, 126.4, 126.8, 128.5, 128.6, 129.5, 131.3, 131.7, 133.3, 138.0, 142.3, 143.7. IR (KBr) $\nu=2946, 2210, 819\text{ cm}^{-1}$; MS: m/z 322 (MH) $^+$. HRMS calcd for $\text{C}_{18}\text{H}_{13}\text{BrN}$: 322.0231 [M+H], found: 322.0241.

4.5.5. 2-Methyl-4-*m*-tolyl-1-naphthonitrile (8e). Yield: 54 mg, 42%. White solid, mp: 126–128 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.46 (s, 3H), 2.78 (s, 3H), 7.25–7.51 (m, 6H), 7.66 (t, $J=7.6$ Hz, 1H), 7.92 (d, $J=8.4$ Hz, 1H), 8.28 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.3, 21.5, 108.5, 117.3, 125.2, 126.5, 126.8, 126.9, 128.3, 128.6, 128.9, 129.8, 130.3, 133.3, 138.2, 139.1, 142.3, 145.2. IR (KBr) $\nu=2931, 2218, 810\text{ cm}^{-1}$; MS: m/z 258 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{16}\text{N}$: 258.1283 [M+H], found: 258.1282.

4.5.6. 2-Methyl-4-*p*-tolyl-1-naphthonitrile (8f). Yield: 52 mg, 40%. White solid, mp: 134–135 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.47 (s, 3H), 2.78 (s, 3H), 7.32–7.37 (m, 5H), 7.48 (t, $J=7.6$ Hz, 1H), 7.66 (t, $J=7.6$ Hz, 1H), 7.93 (d, $J=7.6$ Hz, 1H), 8.27 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.26, 21.29, 108.4, 117.3, 125.2, 126.5, 126.8, 128.3, 128.7, 129.2, 129.6, 129.8, 133.3, 136.2, 138.0, 142.3, 145.1. IR (KBr) $\nu=2959, 2918, 2214, 1286, 834\text{ cm}^{-1}$; MS: m/z 258 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{16}\text{N}$: 258.1283 [M+H], found: 258.1273.

4.5.7. 2-Methyl-4-phenyl-1-naphthonitrile (8g). Yield: 74 mg, 61%. White solid, mp: 114–115 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.79 (s, 3H), 7.36 (s, 1H), 7.45–7.54 (m, 6H), 7.67 (t, $J=7.6$ Hz, 1H), 7.91 (d, $J=8.4$ Hz, 1H), 8.28 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.3, 108.6, 117.2, 125.2, 126.6, 126.8, 128.1, 128.4, 128.5, 128.7, 129.67, 129.74, 133.3, 139.1, 142.3, 145.1. IR (KBr) $\nu=3053, 2921, 2217, 766, 704\text{ cm}^{-1}$; MS: m/z 244 (MH) $^+$. HRMS calcd for $\text{C}_{18}\text{H}_{14}\text{N}$: 244.1126 [M+H], found: 244.1136.

4.5.8. 3-Methyl-1,1'-binaphthyl-4-carbonitrile (8h). Yield: 52 mg, 35%. White solid, mp: 120–122 °C; ^1H NMR (400 MHz, CDCl_3) δ :

2.83 (s, 3H), 7.33–7.36 (m, 3H), 7.45–7.54 (m, 4H), 7.60–7.67 (m, 2H), 7.99 (d, $J=8.0$ Hz, 1H), 8.02 (d, $J=8.0$ Hz, 1H), 8.36 (d, $J=8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 21.4, 109.1, 117.3, 125.1, 125.4, 126.0, 126.2, 126.5, 126.7, 127.3, 127.6, 128.4, 128.5, 128.7, 129.7, 131.0, 132.2, 133.0, 133.5, 136.8, 142.4, 143.7. IR (KBr) $\nu=3055, 2918, 2219, 804\text{ cm}^{-1}$; MS: m/z 294 (MH) $^+$. HRMS calcd for $\text{C}_{22}\text{H}_{16}\text{N}$: 294.1283 [M+H], found: 294.1285.

4.5.9. 4-(4-Bromophenyl)-2,3-dimethyl-1-naphthonitrile (8i). Yield: 104 mg, 62%. White solid, mp: 162–163 °C; ^1H NMR (400 MHz, CDCl_3) δ : 2.17 (s, 3H), 2.74 (s, 3H), 7.10 (d, $J=8.4$ Hz, 2H), 7.33 (d, $J=8.4$ Hz, 1H), 7.39 (t, $J=8.0$ Hz, 1H), 7.57 (t, $J=7.6$ Hz, 1H), 7.66 (d, $J=8.0$ Hz, 2H), 8.20 (d, $J=8.0$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ : 18.0, 20.1, 109.7, 117.6, 121.9, 124.9, 126.6, 126.8, 127.5, 131.0, 131.2, 131.3, 131.9, 133.1, 137.8, 142.0, 142.6. IR (KBr) $\nu=2952, 2211, 817, 770\text{ cm}^{-1}$; MS: m/z 336 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{15}\text{BrN}$: 336.0388 [M+H], found: 336.0382.

4.6. Typical procedure for the preparation of 2-(4-methyl-3,4-dihydronaphthalene-2(1H)-ylidene)propanenitrile (9a) (Table 5)

To a flask containing (*E*)-3-benzyl-5-hydroxy-2-methylhex-2-enenitrile (108 mg, 0.5 mmol) in CH_2Cl_2 (5 mL) was added H_2SO_4 (27 μL , 0.5 mmol). The solution was stirred under reflux. Upon completion, the reaction was quenched with aqueous Na_2CO_3 and the mixture was extracted with ethyl acetate (5 mL $\times$ 3). The combined organic phases were washed with brine, dried, filtered, and concentrated under vacuum. The residue was purified by column chromatography on silica gel eluting with ethyl acetate/hexane (1:10) to give 2-(4-methyl-3,4-dihydronaphthalene-2(1H)-ylidene)propanenitrile (**9a**) in a yield of 56% (55 mg, 0.28 mmol). Other 2-methylene-1,2,3,4-tetrahydronaphthalenes were obtained in a similar manner.

4.6.1. (Z)-2-(4-Methyl-3,4-dihydronaphthalen-2(1H)-ylidene)propanenitrile (9a). Yield: 55 mg, 56%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 1.31 (d, $J=6.8$ Hz, 3H), 1.93 (s, 3H), 2.29–2.35 (m, 1H), 2.62–2.67 (m, 1H), 3.02–3.07 (m, 1H), 3.71–4.01 (m, 2H), 7.18–7.26 (m, 4H). ^{13}C NMR (100 MHz, CDCl_3) δ : 15.7, 20.2, 33.2, 35.9, 37.2, 102.5, 119.4, 125.8, 126.56, 126.63, 128.0, 134.0, 141.4, 154.8. IR (KBr) $\nu=2962, 2209, 1309, 758\text{ cm}^{-1}$; MS: m/z 198 (MH) $^+$. HRMS calcd for $\text{C}_{14}\text{H}_{16}\text{N}$: 198.1283 [M+H], found: 198.1276.

4.6.2. (Z)-2-(4-Methyl-3,4-dihydronaphthalen-2(1H)-ylidene)-3-phenylpropanenitrile (9b). Yield: 73 mg, 53%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 1.35 (d, $J=7.2$ Hz, 3H), 2.48–2.54 (m, 1H), 2.78–2.84 (m, 1H), 3.05–3.10 (m, 1H), 3.61–3.71 (m, 2H), 3.83–4.10 (m, 2H), 7.22–7.39 (m, 9H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.4, 33.3, 35.5, 36.1, 37.4, 107.6, 118.8, 126.0, 126.7, 126.8, 127.0, 128.2, 128.4, 128.8, 133.7, 137.3, 141.3, 155.6. IR (KBr) $\nu=3062, 2963, 2209, 758\text{ cm}^{-1}$; MS: m/z 274 (MH) $^+$. HRMS calcd for $\text{C}_{20}\text{H}_{20}\text{N}$: 274.1596 [M+H], found: 274.1605.

4.6.3. (Z)-2-(4-Methyl-3,4-dihydronaphthalen-2(1H)-ylidene)-2-phenylacetoneitrile (9c). Yield: 56 mg, 43%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 1.21 (d, $J=6.8$ Hz, 3H), 2.45–2.50 (m, 1H), 2.70–2.75 (m, 1H), 2.96–3.01 (m, 1H), 3.96–4.19 (m, 2H), 7.22–7.26 (m, 4H), 7.31–7.44 (m, 5H). ^{13}C NMR (100 MHz, CDCl_3) δ : 20.7, 33.7, 36.6, 37.1, 110.0, 118.5, 126.2, 126.6, 126.7, 128.3, 128.3, 128.4, 128.7, 128.8, 129.0, 129.1, 133.3, 141.3, 157.2. IR (KBr) $\nu=3061, 2960, 2213, 759, 699\text{ cm}^{-1}$; MS: m/z 260 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{18}\text{N}$: 260.1439 [M+H], found: 260.1442.

4.6.4. (Z)-2-(4-phenyl-3,4-dihydronaphthalen-2(1H)-ylidene)propanenitrile (9d). Yield: 62 mg, 48%. Syrup; ^1H NMR (400 MHz, CDCl_3) δ : 1.68 (s, 3H), 2.83–2.86 (m, 2H), 3.84–4.29 (m, 3H), 6.91 (d,

$J=7.6$ Hz, 1H), 7.08 (d, $J=7.6$ Hz, 2H), 7.12–7.34 (m, 6H). ^{13}C NMR (100 MHz, CDCl_3) δ : 15.4, 36.2, 37.3, 45.5, 103.1, 119.4, 126.6, 126.8, 127.1, 128.2, 128.3, 128.4, 128.6, 134.6, 138.8, 143.2, 153.5. IR (KBr) $\nu=3055, 2953, 2212, 760\text{ cm}^{-1}$; MS: m/z 260 (MH) $^+$. HRMS calcd for $\text{C}_{19}\text{H}_{18}\text{N}$: 260.1439 [M+H], found: 260.1441.

Acknowledgements

We are grateful to the National Natural Science Foundation of China (21172057, 21272058), RFDP (20114104110005), and PCSIRT (IRT 1061) for financial support.

Supplementary data

The X-ray crystal structures of **5e**, **6b**, **8i** and the copies of ^1H and ^{13}C NMR spectra of **5a–5p**, **6a–6aa**, **7**, **8a–8i**, and **9a–9d**. Supplementary data associated with this article can be found in the online version, at <http://dx.doi.org/10.1016/j.tet.2013.04.118>.

References and notes

- (a) Fan, X. S.; Wang, Y. Y.; Qu, Y. Y.; Xu, H. Y.; He, Y.; Zhang, X. Y.; Wang, J. J. *J. Org. Chem.* **2011**, *76*, 982; (b) Zhang, X. Y.; Jia, X. F.; Fang, L. L.; Liu, N.; Wang, J. J.; Fan, X. S. *Org. Lett.* **2011**, *13*, 5024; (c) Xu, H. Y.; Zhang, X. Y.; He, Y.; Guo, S. H.; Fan, X. S. *Chem. Commun.* **2012**, 3121.
- (a) Mandai, T.; Gotoh, J.; Otera, J.; Kawada, M. *Chem. Lett.* **1980**, 313; (b) Chikhalikar, S.; Bhawe, V.; Ghotekar, B.; Jachak, M.; Ghagare, M. *J. Org. Chem.* **2011**, *76*, 3829; (c) Nakao, Y.; Yada, A.; Ebata, S.; Hiyama, T. *J. Am. Chem. Soc.* **2007**, *129*, 2428; (d) Chow, T. J.; Lin, R.; Ko, C. W.; Tao, Y. T. *J. Mater. Chem.* **2002**, *12*, 42.
- Manimekalai, A.; Balamurugan, A. *J. Mol. Struct.* **2011**, *995*, 78.
- Recent examples on the preparation of 2,4-dienitriles: (a) Tomioka, T.; Sankranti, R.; Vaughan, T. G.; Maejima, T.; Yanase, T. *J. Org. Chem.* **2011**, *76*, 8053; (b) Qin, C.; Jiao, N. *J. Am. Chem. Soc.* **2010**, *132*, 15893; (c) Krishna, C. V.; Bhonde, V. R.; Devendar, A.; Maitra, S.; Mukkanti, K.; Iqbal, J. *Tetrahedron Lett.* **2008**, *49*, 2013; (d) Huang, H. Y.; Wang, H. M.; Kang, I. J.; Chen, L. C. *J. Chin. Chem. Soc.* **2003**, *50*, 1053; (e) Clary, K. N.; Back, T. G. *Synlett* **2007**, 2995.
- Recent examples on intramolecular Pinner reaction, see: (a) Grossman, R. B.; Rasne, R. M. *Org. Lett.* **2001**, *3*, 4027; (b) Silvina, V. J.; Pellegrinet, S. C.; Colombo, M. I. *J. Org. Chem.* **2007**, *72*, 3702.
- (a) Fisyuk, A. S.; Bundel, Y. G. *Chem. Heterocycl. Compd.* **1999**, *35*, 125; (b) Fisyuk, A. S.; Poendaev, N. V. *Molecules* **2002**, *7*, 124; (c) Fisyuk, A. S.; Bogza, Y. P.; Poendaev, N. V.; Goncharov, D. S. *Chem. Heterocycl. Compd.* **2010**, *46*, 844; (d) Tolmachova, N. A.; Dolovanyuk, V. G.; Geras, I. I.; Kondratov, I. S.; Polovinko, V. V.; Bergander, K.; Haufe, G. *Synthesis* **2011**, 1149; (e) Golovach, N. M.; Sukach, V. A.; Vovk, M. V. *Russ. J. Org. Chem.* **2010**, *46*, 1070; (f) Fernández-García, J. M.; Fernández-Rodríguez, M. A.; Aguilar, E. *Org. Lett.* **2011**, *13*, 5172; (g) Cardillo, G.; Fabbri, S.; Gentilucci, L.; Perciaccante, R.; Piccinelli, F.; Tolomelli, A. *Tetrahedron* **2004**, *60*, 5031 and references therein; (h) Hanessian, S.; Seid, M.; Nilsson, I. *Tetrahedron Lett.* **2002**, *43*, 1991; (i) Carlson, R. M.; Oyler, A. R.; Peterson, J. R. *J. Org. Chem.* **1975**, *40*, 1610.
- (a) Seong, B. L.; Han, M. L. *Chem. Lett.* **1982**, 627; (b) Trujillo, J. M.; Jorge, R. E.; Boada, J. *Phytochemistry* **1990**, *29*, 2991; (c) Ward, R. S. *Nat. Prod. Rep.* **1995**, *12*, 183; (d) Ukita, T.; Nakamura, Y.; Kubo, A.; Yamamoto, Y.; Takahashi, M.; Kotera, J.; Ikeo, T. *J. Med. Chem.* **1999**, *42*, 1293; (e) Xie, X.; Kozlowski, M. C. *Org. Lett.* **2001**, *3*, 2661; (f) Silva, O.; Gomes, E. T. *J. Nat. Prod.* **2003**, *66*, 447; (g) Dai, J.; Liu, Y.; Zhou, Y.-D.; Nagle, D. G. *J. Nat. Prod.* **2007**, *70*, 1824; (h) Brasholz, M.; Sörgel, S.; Azap, C.; Reissig, H.-U. *Eur. J. Org. Chem.* **2007**, 3801; (i) Krohn, K.; Kounam, S. F.; Cludius-Brandt, S.; Draeger, S.; Schulz, B. *Eur. J. Org. Chem.* **2008**, 3615; (j) Lowell, A. N.; Fennie, M. W.; Kozlowski, M. C. *J. Org. Chem.* **2008**, *73*, 1911.
- Watson, M. D.; Fechtenkotter, A.; Mullen, K. *Chem. Rev.* **2001**, *101*, 1267.
- (a) Kim, S. H.; Kim, Y. M.; Lee, H. S.; Kim, J. N. *Tetrahedron Lett.* **2010**, *51*, 1592; (b) Balamurugan, R.; Gudla, V. *Org. Lett.* **2009**, *11*, 3116; (c) Duan, S.; Sinha-Mahapatra, D. K.; Herndon, J. W. *Org. Lett.* **2008**, *10*, 1541; (d) Zhang, X.; Sarkar, S.; Larock, R. C. *J. Org. Chem.* **2006**, *71*, 236; (e) Wang, Y.; Xu, J.; Burton, D. J. *J. Org. Chem.* **2006**, *71*, 7780; (f) Lewis, K. D.; Matzger, A. J. *J. Am. Chem. Soc.* **2005**, *127*, 9968; (g) Nishii, Y.; Yoshida, T.; Asano, H.; Wakasugi, K.; Morita, J.; Aso, Y.; Yoshida, E.; Motoyoshiya, J.; Aoyama, H.; Tanabe, Y. *J. Org. Chem.* **2005**, *70*, 2667; (h) de Koning, C. B.; Rousseau, A. L.; van Otterlo, W. A. L. *Tetrahedron* **2003**, *59*, 7; (i) Kabalka, G. W.; Ju, Y.; Wu, Z. *J. Org. Chem.* **2003**, *68*, 7915; (j) Barluenga, J.; Vázquez-Villa, H.; Ballesteros, A.; González, J. M. *Org. Lett.* **2003**, *5*, 4121; (k) Viswanathan, G. S.; Wang, M.; Li, C.-J. *Angew. Chem., Int. Ed.* **2002**, *41*, 2138.
- For examples in which tetrahydronaphthalene derivatives act as scaffolds of biological agents or parts of naturally occurring products, please see: (a) Pinto, O.; Sardinha, J.; Vaz, P. D.; Piedade, F.; Calhorda, J. J.; Abramovitch, R.; Zazareth, N.; Pinto, M.; Nascimento, M. S. J.; Rauter, A. P. *J. Med. Chem.* **2010**, *54*, 3175; (b) Barker, M.; Clackers, M.; Copley, R.; Demaine, D. A.; Humphreys, D.; Inglis, G. G. A.; Johnston, M. J.; Jones, H. T.; Haase, M. V.; House, D.; Loiseau, R.; Nisbet, L.; Pacquet, F.; Skone, P. A.; Shanahan, S. E.; Tape, D.; Vinader, V. M.; Washington, M.; Uings, I.; Upton, R.; McLay, I. M.; Macdonald, S. J. *F. J. Med. Chem.* **2006**, *49*, 4216; (c) Kim, I.-H.; Par, Y.-K.; Hammock, B. D.; Nishi, K. *J. Med. Chem.* **2011**, *54*, 1752; (d) Pu, J.; Deng, K. W.; Butera, J.; Chlenov, M.; Gilbert, A.; Kagan, M.; Mattes, J.; Resnick, L. *Tetrahedron* **2010**, *66*, 1963; (e) Ahmad, S.; Doweyko, L.; Ashfaq, A.; Ferrara, F. N.; Bisaha, F. S.; Schmidt, J. B.; DiMarco, J.; Conder, M. L.; Jenkins-West, T.; Normandin, D. E.; Russell, A. D.; Smith, M. A.; Levesque, P. C.; Lodge, N. J.; Lloyd, J.; Stein, P. D.; Atwal, K. S. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 99.
- For a recent Friedel–Crafts reaction of tetrahydronaphthols, see: Penick, M. A.; Mahindaratne, P. D.; Guiterrez, R. D.; Smith, T. D.; Tiekinck, R. T.; Negrete, G. R. *J. Org. Chem.* **2008**, *73*, 6378.
- For recent examples for the preparation tetrahydronaphthalenes, see: (a) Grisé, C. M.; Barriault, L. *Org. Lett.* **2006**, *8*, 5905; (b) Riou, M.; Barriault, L. *J. Org. Chem.* **2008**, *73*, 7436; (c) Huang, W.; Zheng, P. Z.; Zhang, Z. X.; Liu, R. T.; Chen, Z. X.; Zhou, X. G. *J. Org. Chem.* **2008**, *73*, 6845; (d) Kuwano, R.; Shige, T. *J. Am. Chem. Soc.* **2007**, *129*, 3802; (e) Faul, M. M.; Ratz, A. M.; Sullivan, K. A.; Trankle, W. G.; Winneroske, L. L. *J. Org. Chem.* **2011**, *66*, 5772.
- For 2-methylene-1,2,3,4-tetrahydronaphthalene used as starting material for the synthesis of non-racemic Fixolide, see: Ciappa, A.; Scrivant, A.; Matteoli, U. *J. Mol. Catal. A: Chem.* **2004**, *219*, 217.