

## TWO LIGNANS FROM *SCHISANDRA SPHENANTHERA*\*

YUKINOBU IKEYA, KO SUGAMA, MINORU OKADA and HIROSHI MITSUHASHI

Research Institute for Biology & Chemistry, Tsumura & Co., 3586 Yoshiwara Ami-machi, Inashiki-gun, Ibaraki 300-11, Japan

(Received 2 July 1990)

**Key Word Index**—*Schisandra sphenanthera*; Schisandraceae; dibenzocyclooctadiene; lignan; benzoylgomisin U; tigloylgomisin O; isoschizandrin.

**Abstract**—Two new dibenzocyclooctadiene lignans, benzoylgomisin U and tigloylgomisin O were isolated from the fruits of *Schisandra sphenanthera* together with known lignans, gomisin U and epigomisin O. Their structures were determined by chemical and spectral studies. The structure of isoschizandrin was also revised by advanced chemical and spectral studies.

### INTRODUCTION

The fruits of *Schisandra sphenanthera* Rehd. et Wils. are used as an antitussive, tonic, and sedative agent under the name of Wuweizi in Chinese traditional medicine together with the fruits of *S. chinensis* Baill. Eleven dibenzocyclooctadiene lignans [1, 2] have been isolated from this plant. In this paper we report the isolation of two new lignans, benzoylgomisin U (**1a**) and tigloylgomisin O (**2a**) together with two known lignans, gomisin U (**1b**) and epigomisin O (**2b**) from *S. sphenanthera* collected in the province of Shangxi in China, and the revised stereostructure of isoschizandrin (**3a**) [3] from *S. chinensis*.

### RESULTS AND DISCUSSION

Compound **1b**, named gomisin U was identified as 6 $\beta$ -hydroxy compound [4] of (–)-gomisin K<sub>1</sub> (**1c**) by direct comparison with an authentic sample. Compound **2b** was identified as epigomisin O [5] by direction comparison with an authentic sample obtained from *S. chinensis*.

Benzoylgomisin U (**1a**) was obtained as needles, C<sub>30</sub>H<sub>34</sub>O<sub>8</sub>, [ $\alpha$ ]<sub>D</sub> –61.6° (CHCl<sub>3</sub>) and possessed the characteristic UV spectrum ( $\lambda_{max}$  215.2, 255 sh, and 289 sh nm) of dibenzocyclooctadiene lignans [6]. Its CD spectrum ([ $\theta$ ]<sub>213.1</sub> +86 200, [ $\theta$ ]<sub>235.0</sub> –98 500, and [ $\theta$ ]<sub>248.6</sub> –55 100 sh) indicated that **1a** has a *S*-biphenyl configuration [7]. The <sup>1</sup>H NMR spectral analyses (Table 1) showed that **1a** has one phenolic hydroxyl and five methoxyl groups on the aromatic rings, and a benzoyloxy and two secondary methyl groups. The mass spectrum with peaks at *m/z* 522 [M]<sup>+</sup>, 400 [M – C<sub>6</sub>H<sub>5</sub>CO<sub>2</sub>H]<sup>+</sup> and 105 [C<sub>6</sub>H<sub>5</sub>CO]<sup>+</sup> [8], and <sup>13</sup>C NMR spectrum (Table 2) also supported the presence of a benzoyloxy group in **1a**.

On hydrolysis with 3% ethanolic potassium hydroxide, **1a** afforded a benzoic acid and **1b**. The doublet at  $\delta$ 4.42 in the <sup>1</sup>H NMR spectrum of **1b**, which appeared at

$\delta$ 6.01 in **1a**, was assigned to the C-6 benzylic methine. This showed that the benzoyl group in **1a** is linked to the C-6 hydroxyl group in **1b**.

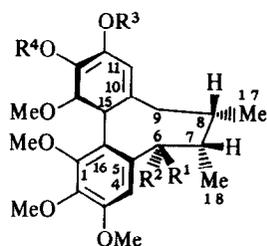
The structure of **1a** was confirmed by the two-dimensional nuclear Overhauser effects spectroscopy (NOESY) spectrum of **1a** in chloroform-*d* (Fig. 1). The NOESY spectrum showed appreciable NOE between the C-3 methoxyl signal at  $\delta$ 3.89 and the lower-field aromatic proton (H-4) signal at  $\delta$ 6.80, and between the H-4 and the C-6 benzylic methine signal, but no NOE between the higher-field aromatic proton (H-11) signal at  $\delta$ 6.67 and any methoxyl signal. These findings indicated that the phenolic hydroxyl group is located adjacent to H-11, and that the C-6 benzylic methine proton is  $\alpha$ -oriented. The NOESY spectrum also showed appreciable NOE between the C-7 methyl signal at  $\delta$ 0.84 and the H-4 signal, and between the H-8 signal and the H-11 signal, indicating that both C-7 and C-8 methyl groups are  $\alpha$ -oriented. On the basis of the above results, the structure of benzoylgomisin U was determined as (6*R*,7*S*,8*S*,*S*-biar)-6-benzoyloxy-6,7,8,9-tetrahydro-1,2,3,13,14-heptamethoxy-7,8-dimethyl-12-dibenzo [*a*, *c*]-cyclooctenol (**1a**) (Fig. 1).

Tigloylgomisin O (**2a**) was obtained as an amorphous powder, C<sub>28</sub>H<sub>34</sub>O<sub>8</sub>, [ $\alpha$ ]<sub>D</sub> –22° (CHCl<sub>3</sub>) and possessed the characteristic UV spectrum ( $\lambda_{max}$  215.2, 258 sh, and 293 sh nm) of dibenzocyclooctadiene lignan. Its CD spectrum ([ $\theta$ ]<sub>229.4</sub> +5 700, [ $\theta$ ]<sub>241.1</sub> –57 200, [ $\theta$ ]<sub>249.1</sub> –47 200 sh) indicated that **2a** has an *S*-biphenyl configuration [7]. The <sup>1</sup>H NMR spectrum (Table 1) showed that **2a** has a methylenedioxy and four methoxyl groups on the aromatic rings, and a tigloyloxy and two secondary methyl groups. The <sup>13</sup>C NMR spectrum (Table 2) also supported the presence of tigloyl group [ $\delta$ 11.2 ( $\alpha$ -Me), 14.2 ( $\beta$ -Me), 128.4, 137.1 (C=C), and 166.9 (C=O)].

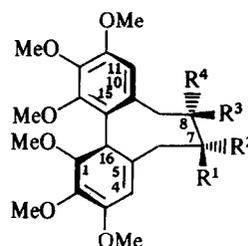
On hydrolysis with 3% ethanolic potassium hydroxide, **2a** afforded a tiglic acid and **2c**, which was identified as gomisin O [5] by direct comparison with an authentic sample. The doublet at  $\delta$ 4.34 in the <sup>1</sup>H NMR spectrum of **2c**, which appeared at  $\delta$ 5.76 in **2a**, was assigned to the C-6 benzylic methine. This showed that the tigloyl group in **2a** is linked to the C-6 hydroxyl group in **2c**.

The structure of **2a** was confirmed by the NOESY spectrum in chloroform-*d* (Fig. 2). The NOESY spectrum

\* Part 17 in the series 'The Constituents of *Schisandra* Species'. For Part 16 see ref. [2].



- 1a**  $R^1 = \text{OCO}-\text{C}_6\text{H}_5$ ,  $R^2 = R^3 = \text{H}$ ,  $R^4 = \text{Me}$     **3a**  $R^1 = R^4 = \text{Me}$ ,  $R^2 = \text{OH}$ ,  $R^3 = \text{H}$   
**1b**  $R^1 = \text{OH}$ ,  $R^2 = R^3 = \text{H}$ ,  $R^4 = \text{Me}$     **3b**  $R^1 = R^3 = \text{Me}$ ,  $R^2 = \text{OH}$ ,  $R^3 = \text{H}$   
**1c**  $R^1 = R^2 = R^3 = \text{H}$ ,  $R^4 = \text{Me}$     **4a**  $R^1 = \text{OH}$ ,  $R^2 = R^4 = \text{Me}$ ,  $R^3 = \text{H}$   
**4c**  $R^1 + R^2 = \text{CH}_2$ ,  $R^3 = \text{H}$ ,  $R^4 = \text{Me}$



- 2a**  $R^1 = \text{OCO}-\text{C}_6\text{H}_5$ ,  $R^2 = \text{H}$ ,  $R^3 + R^4 = \text{CH}_2$

- 2b**  $R^1 = \text{H}$ ,  $R^2 = \text{OH}$ ,  $R^3 + R^4 = \text{CH}_2$   
**2c**  $R^1 = \text{OH}$ ,  $R^2 = \text{H}$ ,  $R^3 + R^4 = \text{CH}_2$

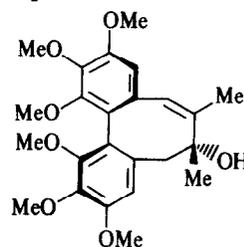
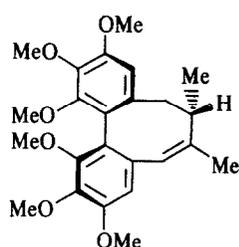
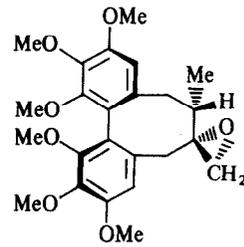
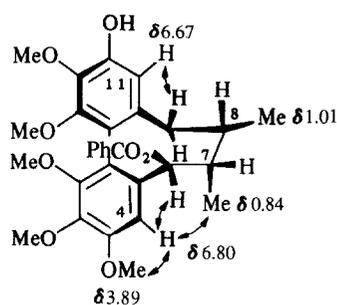
**4b****4d****4e**

Fig. 1. NOE ( $\leftrightarrow$ ) in the NOESY spectrum of compound **1a** (in  $\text{CDCl}_3$ ).

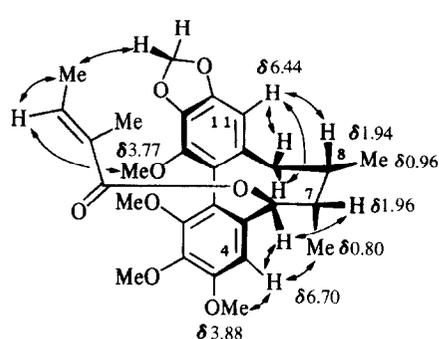


Fig. 2. NOE ( $\leftrightarrow$ ) in the NOESY spectrum of compound **2a** (in  $\text{CDCl}_3$ ).

showed appreciable NOE between the C-3 methoxyl signal at  $\delta$  3.88 and the lower-field aromatic proton (H-4) signal at  $\delta$  6.70, and between the H-4 and the C-6 benzylic methine signal, indicating that the C-6 benzylic methine proton is  $\alpha$ -oriented. The NOESY spectrum also showed

appreciable NOE between the C-7 methyl signal at  $\delta$  0.80 and the H-4 signal, and between the H-8 signal and the H-11 signal, indicating that both C-7 and C-8 methyl groups are  $\alpha$ -oriented. On the basis of above results, the structure of tigloylgomisins O was determined as (6*R*,7*S*,8*S*,*S*-biar)-

Table 1. <sup>1</sup>H NMR spectral data of compounds **1a**, **1b**, **2a**, **2b**, **2c**, **3a**, **4c**, and **4e** (CDCl<sub>3</sub>, 500 MHz)

Compound	H-4, s H-11, s	H-6 $\alpha$ ( <i>J</i> = Hz)	H-6 $\beta$ ( <i>J</i> = Hz)	H-9 $\alpha$ , <i>dd</i> ( <i>J</i> = Hz)	H-9 $\beta$ , <i>dd</i> ( <i>J</i> = Hz)	H-7 ( <i>J</i> = Hz)	H-8 ( <i>J</i> = Hz)	C-7-Me ( <i>J</i> = Hz)	C-8-Me <i>d</i> ( <i>J</i> = Hz)	OMe <i>s</i>
<b>1a</b> *	6.80 6.67	6.01 <i>d</i> (7.7)		2.22 <i>br t</i> † †	2.30 <i>br d</i> † †	2.09 <i>m</i>	2.09 <i>m</i>	0.84 <i>d</i> (6.8)	1.01 (6.9)	3.38, 3.53, 3.75, 3.89 3.92, 5.80 1H, <i>s</i> (OH)
<b>1b</b>	6.55 6.59	4.42 <i>d</i> (7.7)		2.06 (12.7, 10.0)	2.27 (12.7, 4.0)	1.76 <i>m</i>	1.88 <i>m</i>	0.88 <i>d</i> (6.9)	0.94 (7.1)	3.53, 3.71, 3.89, 3.90 3.94, 5.78 1H, <i>s</i> (OH)
<b>2a</b> *	6.70 6.44	5.76 <i>d</i> (7.8)		2.15 (13.4, 9.4)	2.22 (13.4, 3.8)	1.96 <i>m</i>	1.94 <i>m</i>	0.80 <i>d</i> (6.9)	0.96 (7.1)	3.52, 3.77, 3.89 ( $\times 2$ ) 5.92 and 5.96 each 1H, <i>d</i> (1.5 Hz, OCH <sub>2</sub> O)
<b>2b</b>	7.01 6.44	1.09 <i>br s</i> (OH)	4.57 <i>d</i> (1.2)	2.12 (13.4, 4.9)	1.95 <i>d</i> (13.4)	1.85–2.05 <i>m</i>	1.85–2.05 <i>m</i>	0.70 <i>d</i> (6.8)	1.00 (6.8)	3.55, 3.85, 3.91 ( $\times 2$ ) 5.94 2H, <i>s</i> (OCH <sub>2</sub> O)
<b>2c</b>	6.57 6.42	4.34 <i>d</i> (8.3)		2.02 (13.0, 4.9)	2.32 (13.0, 5.5)	1.67 <i>m</i>	1.85 <i>m</i>	0.92 <i>d</i> (5.1)	0.90 (5.6)	3.53, 3.89, 3.90, 3.91 5.95 and 5.97 each 1H, <i>d</i> (1.5 Hz, OCH <sub>2</sub> O)
<b>3a</b>	6.61 6.54	2.82 <i>d</i> (13.1)	2.32 <i>dd</i> (13.1, 1.1)	2.52 (14.1, 6.5)	2.55 (14.1, 3.2)		1.90 <i>m</i>	1.19 <i>s</i>	0.88 (7.1)	3.55, 3.56, 3.88 ( $\times 2$ ) 3.89 ( $\times 2$ )
<b>4c</b>	6.65 6.55	2.93 <i>d</i> <sup>a</sup> (12.3)	3.00 <i>d</i> <sup>a</sup> (12.3)	2.52 (13.5, 5.4)	2.56 (13.5, 2.7)		2.70 <i>ddd</i> (7.2, 5.4, 2.7)	4.72 1H, <i>d</i> (2.0) 4.86 1H, <i>d</i> (2.0)	1.02 (7.2)	3.60, 3.62, 3.87, 3.88 ( $\times 2$ ), 3.90
<b>4e</b>	6.48 <sup>a</sup> 6.58 <sup>a</sup>	3.02 <i>dd</i> (13.3, 1.5)	1.95 <i>dd</i> (13.3, 0.9)	2.59 (13.7, 6.9)	2.69 (13.7, 1.7)		1.52 <i>m</i>	2.54 1H, <i>dd</i> (4.5, 1.5) 2.98 1H, <i>d</i> (4.5)	1.00 (7.1)	3.59, 3.60, 3.88 ( $\times 2$ ) 3.90 ( $\times 2$ )

\*Other signals: **1a**, C<sub>6</sub>H<sub>5</sub>CO—: 7.29 (2H, *m*), 7.45 (1H, *m*), 7.57 (2H, *br*); **2a**, Tigloyl: 1.62 (3H, *quin*, *J* = 1.2 Hz,  $\alpha$ -Me), 1.67 (3H, *br q*, *J* = 7.0, 1.2 Hz,  $\beta$ -Me), 6.27 (1H, *br q*, *J* = 7.0 Hz).†In case of 200 MHz <sup>1</sup>H NMR spectrum, these signals were shown at  $\delta$ 2.20 (*dd*, *J* = 13.0, 10.0 Hz) and 2.30 (*dd*, *J* = 13.0, 4.0 Hz).<sup>a</sup>Assignments of these signals may be reversed.

Table 2.  $^{13}\text{C}$  NMR spectral data of compounds **1a**, **1b**, **2a**, and **2c** ( $\text{CDCl}_3$ , 50 MHz)

C	<b>1a</b> *	<b>1b</b>	<b>2a</b> *†	<b>2c</b>
1	152.0	152.0 <sup>a</sup>	152.0	151.9
2	142.0	141.6	141.8 <sup>a</sup>	141.7 <sup>a</sup>
3	151.8	151.9 <sup>a</sup>	151.7	152.1
4	111.0	110.1	110.8	110.1
5	132.1	136.4 <sup>b</sup>	132.6	137.0
6	81.4	81.4	80.9	81.4
7	37.6	40.0	37.4	40.0
8	36.4	36.8	36.5	37.0
9	36.8	37.3	36.9	38.0
10	137.6	137.7 <sup>b</sup>	135.5	135.5
11	109.6	109.8	102.4	102.5
12	148.9	149.4	148.6	149.2
13	137.5	137.7	134.5	134.6
14	150.3	150.6	141.4 <sup>a</sup>	141.5 <sup>a</sup>
15	122.2 <sup>a</sup>	122.0 <sup>c</sup>	121.9 <sup>b</sup>	122.2 <sup>b</sup>
16	123.2 <sup>a</sup>	120.8 <sup>c</sup>	123.2 <sup>b</sup>	120.6 <sup>b</sup>
17	20.3	18.3	19.2	17.5 <sup>c</sup>
18	14.2	15.7	14.2	16.5 <sup>c</sup>
C-1, 14	60.5 <sup>b</sup> , 59.7	60.4, 60.2	60.5, 59.2	60.4, 59.5
OMe C-2, 13	60.9, 60.6 <sup>b</sup>	60.9, 60.9	60.9, —	60.9, —
C-3, 12	56.0,	56.0, —	56.0, —	56.0, —
OCH <sub>2</sub> O	—	—	100.6	100.8

\* Other signals: **1a**, 128.1 (3', 5'), 129.7 (2', 6'), 130.3 (1'), 132.8 (4'), 165.5 (C=O) ( $\text{C}_6\text{H}_5\text{CO}-$ ); **2a**, 11.7 ( $\alpha$ -Me), 14.2 ( $\beta$ -Me), 128.4 ( $\alpha$ -olephin), 137.1 ( $\beta$ -olephin), 166.9 (C=O) (Tigloyl).

† This compound was measured at 125 MHz and assignments were confirmed by  $^1\text{H}-^{13}\text{C}$  COSY spectrum.

<sup>a,b,c</sup> Assignments within any vertical column may be reversed.

6,7,8,9-tetrahydro-1,2,3,14-tetramethoxy-7,8-dimethyl-12,13-methylenedioxy-6-tigloyloxy-dibenzo [*a, c*] cyclooctene (**2a**) (Fig. 2).

Previously, the structure of isoschizandrin isolated from *S. chinensis* was proposed as **3b** having a C-8 $\beta$  secondary methyl group on the basis of spectral studies and chemical correlation with **4b** derived from schizandrin (**4a**) [3]. However, this structural re-investigation showed that the structure of isoschizandrin is **3a** having the C-8 $\alpha$  secondary methyl group as follows. The NOESY spectrum of **3a** (Fig. 3) showed appreciable NOE between the H-11 signal at  $\delta$ 6.54 and the C-8 secondary methyl signal at  $\delta$ 0.88, indicating that the C-8 secondary methyl group is  $\alpha$ -oriented. The configuration of the C-8 position in **3a** was confirmed by the transformation of schizandrin (**4a**) to **3a**.

Treatment of **4a** with hydrochloric acid afforded **4c** and **4d** and some minor compounds. The physical and spectroscopic properties of **4c** and **4d** were identical with those of authentic samples, respectively [3]. Oxidation of **4c** with *m*-chloroperbenzoic acid afforded an epoxide **4e**. The presence of a methylene signal ( $\delta$ 2.54, 1H, *dd*,  $J = 4.5$ , 1.5 Hz; 2.98, 1H, *d*,  $J = 4.5$  Hz) and no exo-methylene signal in the  $^1\text{H}$  NMR spectrum of **4e** indicated that an epoxide ring in **4e** is formed by oxidation of exo-methylene group in **4c**. Reduction of **4e** with  $\text{LiAlH}_4$  afforded **3a**. These findings indicate that isoschizandrin is the C-7 epimer of **4a**. From these spectral and chemical data, the

structure of isoschizandrin was revised to (7*R*,8*S*,*R*-biar)-6,7,8,9-tetrahydro-1,2,3,12,13,14-hexamethoxy-7,8-dimethyl-7-dibenzo [*a, c*] cyclooctenol (**3a**).

## EXPERIMENTAL

*General.* See ref. [2]. 10%  $\text{AgNO}_3$ -impregnated silica gel (Merck Kieselgel 60F<sub>254</sub>) was used for argentic prep. TLC.

*2D-NOESY.* 2D-NOE data were obtained from the phase sensitive NOESY using the standard DISNMR software package (Bruker): data were collected in 2K  $t_2$  data with 4 scans and 1K  $t_1$  increments. A mixing time of 2.5 sec was randomly modulated by  $\pm 10\%$  in order to eliminate coherent magnetization transfer. The total experimental time was 10 hr. The data were filtered through a squared sinebell window filter (SSB2=2) and doubly transformed in a  $2 \times 1\text{K}$  data matrix (real + imaginary) with a digital resolution of 2.68 Hz per point in the  $\omega_2$  and  $\omega_1$  domain.

*Plant material.* Dried fruits of *Schisandra sphenanthera* Rehd. et Wils. collected in October 1987, were provided from the Institute of Medicine and Pharmacy Shanxi Province.

*Extraction and isolation.* The dried fruits (475 g) of *S. sphenanthera* were pulverized and extracted with *n*-hexane (1.5 l  $\times$  3, 3 hr each) under reflux. The *n*-hexane extracts were concd to give a brown mass (79.5 g). This afforded 12 frs (frs 1–12) on silica gel CC with *n*-hexane–EtOAc. Fr. 8 (2.87 g) was rechromatographed on silica gel (3 cm i.d.  $\times$  16 cm) with  $\text{C}_6\text{H}_6$ – $\text{Et}_2\text{O}$  (9:1). The first eluate (200 ml) was concd to give a residue. This residue was

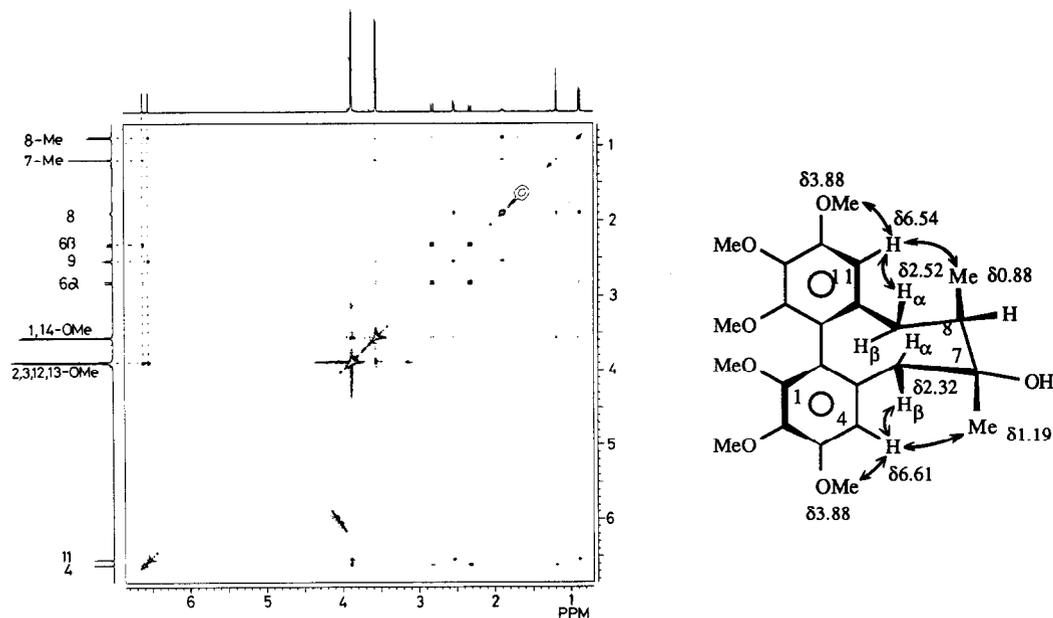


Fig. 3. NOE (↔) in the NOESY spectrum of compound **3a** (in  $\text{CDCl}_3$ ).

purified by prep. TLC [ $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (4:1),  $R_f$  0.48] to give **2a** (24 mg, yield 0.0051%). Fr. 10 (0.81 g) was purified by repeated prep. TLC [1st:  $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (3:2),  $R_f$  0.69; 2nd: *n*-hexane- $\text{CHCl}_3$ - $\text{EtOH}$  (5:14:1),  $R_f$  0.65] to give **1a** (22 mg, yield 0.0046%). Fr. 11 (2.985 g) was rechromatographed on silica gel (3 cm i.d.  $\times$  19 cm) with  $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$ . The  $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (85:15) eluate (1.74 g) was purified by repeated prep. TLC [1st:  $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (3:2),  $R_f$  0.40; 2nd: *n*-hexane- $\text{CHCl}_3$ - $\text{EtOH}$  (10:10:1),  $R_f$  0.47] to give **2b** (22 mg, yield 0.0046%). Fr. 12 (0.173 g) was purified by repeated prep. TLC [1st:  $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (3:2),  $R_f$  0.19; 2nd: *n*-hexane- $\text{Me}_2\text{CO}$  (7:2),  $R_f$  0.25] to give **1a** (12.5 mg, yield 0.00026%).

Compounds **1b** and **2b** were identified by direct comparison with authentic samples [4, 5].

**Benzoylgomisin U (1a)**. Needles from *n*-hexane- $\text{Et}_2\text{O}$ , mp 166–167.5°,  $[\alpha]_D^{24} -61.6^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.730). IR  $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$ : 3404 (OH), 1708 (C=O), 1582, 720, 712 (aromatic ring). UV  $\lambda_{\text{max}}^{\text{EtOH}} \text{nm}$  (log  $\epsilon$ ): 215.2 (4.72), 255 (sh 4.09), 289 (sh 3.49). EIMS  $m/z$  (rel. int.): 522 [ $\text{M}^+$ ] (37), 400 (100), 385 (13), 122 (23), 105 (75), 77 (37). High resolution MS  $m/z$ : 522.2248 (calc. for  $\text{C}_{30}\text{H}_{34}\text{O}_8$ : 522.2254). CD ( $c$  0.0122, MeOH)  $[\theta]^{24}$  (nm): 0 (203.7), +86 200 (213.1), 0 (222.6), -98 500 (235.0), -55 100 sh (248.6).

**Tigloylgomisin O (2a)**. Amorphous powder,  $[\alpha]_D^{24} -22.0^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.830). IR  $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$ : 1702 (C=O), 1650 (C=C), 1620, 1598 (aromatic ring). UV  $\lambda_{\text{max}}^{\text{EtOH}} \text{nm}$  (log  $\epsilon$ ): 215.2 (4.74), 258 (sh 4.94), 293 (sh 3.42). EIMS  $m/z$  (rel. int.): 498 [ $\text{M}^+$ ] (100), 399 (33), 398 (73), 342 (13), 83 (79), 55 (68). High resolution MS  $m/z$ : 498.2263 (calc. for 498.2254). CD ( $c$  0.0117, MeOH)  $[\theta]^{24}$  (nm): -4 800 (217.1), +5 700 (229.4), -57 200 (241.1), -47 200 sh (249.1), -11 700 sh (273.8).

**Hydrolysis of compound 1a**. A soln of **1a** (10 mg) in 3% KOH- $\text{EtOH}$  (1 ml) was kept at 65° for 3 hr, then diluted with  $\text{H}_2\text{O}$  (15 ml), and extracted with  $\text{Et}_2\text{O}$ . The  $\text{Et}_2\text{O}$  extract was washed with  $\text{H}_2\text{O}$ , dried over  $\text{Na}_2\text{SO}_4$ , and concd. The residue was purified by prep. TLC [ $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (2:1)] to give **1c** (6.5 mg) as an amorphous powder,  $[\alpha]_D^{24} -43.9^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.210). High resolution MS  $m/z$ : 418.1994 (calc. for  $\text{C}_{23}\text{H}_{30}\text{O}_7$ : 418.1992). This compound was identified as gomisin U (**1c**) by

direct comparison with an authentic sample ( $[\alpha]_D$ , IR, EIMS, and  $^1\text{H}$ NMR). The aq. soln was acidified with 1 M HCl and extracted with  $\text{Et}_2\text{O}$ . The  $\text{Et}_2\text{O}$  extract was washed with  $\text{H}_2\text{O}$ , dried over  $\text{Na}_2\text{SO}_4$ , and evapd. The residue was purified by prep. TLC [*n*-hexane- $\text{EtOAc}$  (3:1)] to give benzoic acid as an amorphous powder, which was identical with an authentic sample by HPLC [HPLC conditions: column, YMC Pack A-312ODS (6 mm i.d.  $\times$  150 mm); mobile phase, 20 mM  $\text{KH}_2\text{PO}_4$ - $\text{MeCN}$  (4:1); flow rate, 1 ml  $\text{min}^{-1}$ ; detection, UV 215 nm; benzoic acid, RR, (min), 11.8].

**Hydrolysis of compound 2a**. A soln of **2a** (14 mg) in 3% KOH- $\text{EtOH}$  (2 ml) was kept at 65° for 4 hr, then extracted with  $\text{Et}_2\text{O}$ . The  $\text{Et}_2\text{O}$  extract was washed with  $\text{H}_2\text{O}$ , dried over  $\text{Na}_2\text{SO}_4$ , and evapd. The residue was purified by prep. TLC [*n*-hexane- $\text{Me}_2\text{CO}$  (7:3)] to give **2c** (8 mg) as prisms from *n*-hexane- $\text{Et}_2\text{O}$ , mp 144–145.5°,  $[\alpha]_D^{24} -35.3^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.340). High resolution MS  $m/z$ : 416.1842 (calc. for  $\text{C}_{23}\text{H}_{28}\text{O}_7$ : 416.1835). This compound was identified as gomisin O (**2c**) by direct comparison with an authentic sample ( $[\alpha]_D$ , IR, EIMS, and m mp). The aq. soln was treated as described for hydrolysis of **1a** to give tiglic acid (1 mg) as an amorphous powder, which was identical with an authentic sample by GC [GC conditions: column, 20% FFAP on Chromosorb WAW 80–100 mesh (3 mm i.d.  $\times$  2 m); column temp., 60°; inj. temp., 180°; carrier gas, He, 50 ml  $\text{min}^{-1}$ ; tiglic acid, RR, (min), 6.4].

**Dehydration of schizandrin (4a) with HCl**. Schizandrin (**4a**) (2.5 g) was dissolved in a mixture of 1 M HCl (50 ml) and dioxane (50 ml), and the soln was heated at 90° for 4 hr. After cooling, the reaction mixture was diluted with  $\text{Et}_2\text{O}$  and washed with  $\text{H}_2\text{O}$ . The  $\text{Et}_2\text{O}$  soln was dried over  $\text{Na}_2\text{SO}_4$  and evapd. The residue was chromatographed on silica gel (4 cm i.d.  $\times$  18 cm) with *n*-hexane- $\text{Me}_2\text{CO}$ . The fraction eluted with *n*-hexane- $\text{Me}_2\text{CO}$  (4:1) was evapd to give **4a** (441 mg). The fraction eluted with *n*-hexane-acetone (17:3) was evapd to give a residue, which was purified by argentic prep. TLC [*n*-hexane- $\text{Et}_2\text{O}$  (1:1)] to give **4c** (63 mg) and **4d** (137 mg) [9]. Compound **4c**: prisms from *n*-hexane- $\text{Et}_2\text{O}$ , mp 115–117°,  $[\alpha]_D^{24} +178^\circ$  ( $\text{CHCl}_3$ ;  $c$  1.06) (found: C, 69.33; H, 7.42. Calc. for  $\text{C}_{24}\text{H}_{30}\text{O}_6$ : C, 69.54; H,

7.30%). This compound was identical ( $[\alpha]_D$ , IR,  $^1\text{H NMR}$ , and  $m$  mp) with an authentic sample [3]. Compound **4d**: needles from *n*-hexane–Et<sub>2</sub>O, mp 122–122.5°,  $[\alpha]_D^{24} -104^\circ$  (CHCl<sub>3</sub>; *c* 1.36) (found: C, 69.45; H, 7.31. Calc. for C<sub>24</sub>H<sub>30</sub>O<sub>6</sub>: C, 69.54; H, 7.30%). This compound was identical ( $[\alpha]_D$ , IR,  $^1\text{H NMR}$ , and  $m$  mp) with an authentic sample [3].

*Oxidation of 4c with m-chloroperbenzoic acid.* Compound **4c** (60 mg) and *m*-chloroperbenzoic acid (60 mg) were dissolved in CHCl<sub>3</sub> (2 ml) and the soln was stirred at room temp. for 3 hr. The reaction mixture was diluted with EtOAc, washed 5% NaHCO<sub>3</sub> and H<sub>2</sub>O, and evapd. The residue was purified by prep. TLC [*n*-hexane–Et<sub>2</sub>O (2:3)] to give **4e** (22 mg) as an amorphous powder,  $[\alpha]_D^{24} +123^\circ$  (CHCl<sub>3</sub>; *c* 0.72). IR  $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$ : 1598 (aromatic ring). EIMS  $m/z$  (rel. int.): 430 [M]<sup>+</sup> (100), 402 (27), 400 (14), 399 (17), 360 (11), 359 (14). High resolution MS  $m/z$ : 430.1984 (calc. for C<sub>24</sub>H<sub>30</sub>O<sub>7</sub>: 430.1991).

*Reduction of 4e with LiAlH<sub>4</sub>.* LiAlH<sub>4</sub> (15 mg) was added to a soln of **4e** (16 mg) in dry THF (3 ml). The reaction mixture was stirred at room temp. for 3 hr, and then wet Et<sub>2</sub>O added. The reaction mixture was purified by prep. TLC [*n*-hexane–EtOAc (3:2)] to give **3a** (7 mg) as an amorphous powder,  $[\alpha]_D^{24} +85^\circ$  (CHCl<sub>3</sub>; *c* 0.275). High resolution MS  $m/z$ : 432.2125 (calc. for C<sub>24</sub>H<sub>32</sub>O<sub>7</sub>: 432.2147). This compound was identified as isoschizandrin by direct comparison with an authentic sample ( $[\alpha]_D$ , IR, EIMS, and  $^1\text{H NMR}$ ).

*Acknowledgements*—The authors wish to express their gratitude to Prof. A. I. Meyers (Colorado State University), for his kind

advice on structure determination of isoschizandrin and also to Dr M. Tanaka (Research Institute for Biology & Chemistry, Tsumura & Co.) for valuable discussions. Thanks are also due to Mr Kano and Mrs N. Kobayashi (Research Institute for Biology & Chemistry, Tsumura & Co.), for elemental analysis, and for CD and mass spectra.

#### REFERENCES

1. Liu, J.-S., Fang, S.-D., Huang, M.-D., Kao, Y.-L. and Hsu, J.-S. (1976) *Acta Chim. Sinica* **34**, 229.
2. Ikeya, Y., Miki, E., Okada, M., Mitsuhashi, H. and Chai, J.-G. (1990) *Chem. Pharm. Bull.* **38**, 1408.
3. Ikeya, Y., Taguchi, H., Mitsuhashi, H., Takeda, S., Kase, Y. and Aburada, M. (1988) *Phytochemistry* **27**, 569.
4. Ikeya, Y., Kanatani, H., Hakozaiki, M., Taguchi, H. and Mitsuhashi, H. (1988) *Chem. Pharm. Bull.* **36**, 3974.
5. Ikeya, Y., Taguchi, H., Yosioka, I. and Kobayashi, H. (1979) *Chem. Pharm. Bull.* **27**, 2695.
6. Kochetkov, N. K., Khorlin, A., Chizhov, O. S. and Sheichenko, V. I. (1961) *Tetrahedron Letters* 730.
7. Ikeya, Y., Taguchi, H., Yosioka, I. and Kobayashi, H. (1979) *Chem. Pharm. Bull.* **27**, 1383.
8. Zhai, H.-B. and Cong, P.-Z. (1990) *Acta Pharm. Sinica* **25**, 110.
9. Li, L.-N., Huang, X. and Tan, R. (1985) *Planta Med.* **51**, 297.