## Syntheses and Circular Dichroism Studies of Oligopeptides Having Tri- or Tetra-L-prolyl Sequences

Kazuki Sato,\* Masato Tanaka, and Tetsuo Hayase†

Mitsubishi-Kasei Institute of Life Sciences, 11 Minamiooya, Machida, Tokyo 194 Research Center, Mitsubishi Chemical Industries Ltd., Kamoshida, Midori-ku, Yokohama 227 (Received May 23, 1983)

Synopsis. CD spectra of many oligopeptides having tri- or tetra-L-prolyl sequences showed that a penta- or hexapeptide having a tetra-L-prolyl sequence took poly(L-proline) II helix irrespective of amino acid residues at N- or C-terminal of a L-prolyl cluster and that as to a size of the cluster, tri-Lprolyl sequence was enough to form the helix, when glycine was used as substituent.

It has been reported that oligo(L-proline)s begin to form poly(L-proline) II (PPII) helix at the tetramer, and that the pentamer or the greater has higher population of the helix. 1-4) However, there has been no report in detail on the PPII helix of oligopeptides containing a variety of amino acids other than L-proline. This paper describes syntheses and CD measurements of many oligopeptides having tri- or tetra-L-prolyl sequences (1b-19b, 12c, and 14c; see Table 1) to study effects of amino acid residues bound to the L-prolyl cluster on the formation of PPII helix.5)

First, we synthesized sequential peptides 1b-6b. Compound **1b** showed a negative CD band at 201 nm. Both 3b and 5b showed larger CD bands at longer wavelength of 203 nm (Table 1). Marked difference was not observed between 3b and 5b. Okabayashi et al. reported that Aoc-oligo(L-proline)s having some populations of PPII helix showed a CD minimum at longer wavelength than 203 nm.<sup>2)</sup> So, the CD spectra suggests that both 3b and 5b have ordered conformation such as PPII helix and that 1b has unordered one. All of Boc-[(Pro)<sub>4</sub>-Gly]<sub>m</sub>-OH showed large negative CD bands at 203.5 nm suggesting that all these peptides took also PPII helix. Therefore, prolyl peptides with more than five amino acid residues and at least one tri-L-prolyl sequence seem to take PPII helix. CD spectra of 7b and 8b support this consideration; 8b showed a negative CD band at 203 nm similar to those of 3b and **5b**, whereas **7b** showed a weak CD band at shorter

TABLE 1. CD SPECTRAL DATA OF SYNTHETIC PEPTIDES<sup>a)</sup>

Peptides	[θ] <sup>min</sup> ~200 nm	[θ] <sup>max</sup> =225 nm		
replides	10³ degree cm² dmol⁻¹			
Boc-(Pro) <sub>3</sub> -Gly-OH (1b)	-23.3 (201.0)	0.9 (223.0)		
Boc-(Pro) <sub>4</sub> -Gly-OH (2b)	-26.2 (203.5)	1.3 (226.5)		
$Boc-[(Pro)_3-Gly]_2-OH(3b)$	-28.9 (203.0)	0.9 (227.0)		
Boc-[(Pro)4-Gly]2-OH (4b)	-31.8(203.5)	0.9 (227.5)		
Boc-[(Pro) <sub>3</sub> -Gly] <sub>3</sub> -OH (5b)	-31.1(203.0)	0.6 (227.0)		
Boc-[(Pro)4-Gly]3-OH (6b)	-32.8(203.5)	0.7 (229.0)		
Boc-Gly-(Pro)2-Gly-OH (7b)	-17.0(200.5)	not observed		
Boc-Gly-(Pry) <sub>3</sub> -Gly-OH (8b)	-23.1 (203.0)	0.6 (228.0)		
Boc-Gly-(Pro)4-OH (9b)	-27.6(203.5)	2.0 (229.0)		
Boc-Ala-(Pro)4-OH (10b)	-25.9(203.5)	1.6 (225.0)		
Boc-Val-(Pro)4-OH (11b)	-25.8 (204.0)	0.8 (229.0)		
Boc-Ser-(Pro)4-OH (12b)	-23.7 (204.0)	0.8 (230.0)		
Boc-Thr-(Pro) <sub>4</sub> -OH (13b)	-21.4(206.5)	0.4 (230.0)		
Boc-Tyr-(Pro) <sub>4</sub> -OH (14b)	-20.8 (206.5)	4.3 (228.5)		
Boc-Hyp-(Pro) <sub>4</sub> -OH (15b)	-28.8 (203.5)	2.6 (226.0)		
Boc-Ala-Ser-(Pro)4-OH (16b)	-25.0 (203.0)	0.4 (235.0)		
Boc-(Pro) <sub>4</sub> -Ser-OH (17b)	-25.2 (203.0)	1.1 (226.0)		
Boc-(Pro) <sub>4</sub> -Val-OH (18b)	-29.8 (203.5)	0.4 (227.0)		
Boc-Ser-(Pro) <sub>4</sub> -Gly-OH (19b)	-25.7 (203.0)	1.1 (227.5)		
Boc-Ser-(Pro)4-NH2 (12c)	-25.9(204.0)	1.0 (227.0)		
Boc-Tyr-(Pro) <sub>4</sub> -NH <sub>2</sub> (14c)	-22.7(206.0)	4.2 (228.0)		

a) Figures in parentheses are wavelength at  $[\theta]^{\text{extreme}}$ 

wavelength of 200.5 nm.

To study the relation between amino acid sequences and CD spectra, many peptides having a variety of amino acids at either or both of N- and C-termini of tetra-L-prolyl sequence (9b—19b, 12c, and 14c) were synthesized. N-terminal substitution caused various patterns of CD spectra (9b-15b), however, all these peptides showed CD bands at longer wavelength than 203 nm suggesting that they took PPII helix. A tendency was observed; the longer wavelength a CD band shifted the weaker its intensity decreased. Close similarity of CD spectra between Boc-Gly-(Pro)4-OH (9b) and Boc-(Pro)5-OH (which showed a CD minimum of  $-31.0\times10^3$  degree cm<sup>2</sup> dmol<sup>-1</sup> at 203.5 nm)<sup>6)</sup> suggested that Gly residue was suitable for PPII helix. As to a C-terminal substitution, clear-cut tendency was not observed in the range of this experiment (2b, 17b, and 18b). Compounds 12b, 16b, and 19b, which had the same Ser-(Pro)<sub>4</sub> sequence, showed similar CD spectra to each other suggesting that chain elongation from pentapeptide (12b) to hexapeptides (16b and 19b) did not affect significantly to PPII helix. Amide derivatives (12c and 14c) showed a little larger CD bands than the corresponding peptides with free carboxyl groups (12b and 14b). It seems that charged group near L-prolyl cluster is not suitable for the helix.

From CD spectra of various prolyl peptides mentioned above, we concluded that a penta- or hexapeptide having a tetra-L-prolyl sequence took PPII helix irrespective of the amino acid residues at N- or C-terminal of a L-prolyl cluster, and that as to a size of the cluster tri-L-prolyl sequence was enough to form PPII helix, when glycine is used as substituent.

## **Experimental**

Melting points were uncorrected. Optical rotations were measured on an Union automatic polarimeter PM-201. Yields and analytical data of synthetic peptides except for some oily products are summarized in Table 2; compounds 1a-19a with benzyl ester group at C-termini are precursors of 1b-19b.

 $Boc-(Pro)_3-Gly-OBzl$  (1a). To a chilled solution of  $Boc\text{-}Pro\text{-}OH \hspace{0.2cm} \textbf{(4.30\,g, \hspace{0.1cm} 20\hspace{0.1cm}mmol),} \hspace{0.2cm} \textbf{H}\text{-}(Pro)_2\text{-}Gly\text{-}OBzl\cdot HCl}$ (7.92 g, 20 mmol), and TEA (2.8 ml, 20 mmol) in CHCl<sub>3</sub> (80 ml) was added EDC  $\cdot$  HCl (3.83 g, 20 mmol), and the reaction mixture was stirred for 1 h at 0 °C and overnight at room temperature and evaporated. The residue was dissolved in EtOAc (120 ml), and the solution was washed successively with 10% citric acid, 4% NaHCO<sub>3</sub>, and water, and then evaporated to a small volume after being dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Addition of petroleum ether to the solution gave amorphous solid which was reprecipitated from EtOAc-petroleum ether; yield, 8.46 g (76%).

Compounds 2a—19a were prepared as described for 1a; in such cases that a product was soluble in water, column chromatographies with silica gel or Sephadex LH-20 were used in stead of acid and alkali washings.

TABLE 2. YIELDS AND ANALYTICAL DATA OF SYNTHETIC PEPTIDES

Com-	Yield	$Mp \; \theta_m$	$[\alpha]_{\mathrm{D}}^{22}$ a)		Found (%)		Calcd (%)		
pound	%	°C	0	С	Н	N	С	Н	N
la	76	72—73	-171.2	61.54	7.42	9.87	61.57	7.31	9.90 <sup>b)</sup>
2a	68	73—75	-215.4	62.23	6.95	10.71	62.46	7.25	10.70
3a	60	125—128	-243.6	59.52	7.33	11.70	59.28	7.25	$12.02^{c)}$
<b>4</b> a	54	155—157	-284.6	59.27	7.50	12.31	59.25	7.28	$12.34^{d}$
5a	34	162 - 165	-273.6	57.52	6.98	12.91	57.87	7.24	12.86 <sup>e)</sup>
6a	34	191—193	-314.8	59.63	7.12	13.31	59.95	7.16	13.44 <sup>f)</sup>
10a	69	212-213	-264.6	63.18	7.38	10.54	62,95	7.40	10.49
lla	73	198-200	-259.6	63.99	7.54	9.99	63.86	7.68	10.06
12a	<b>7</b> 5	192—193	-249.6	61.36	7.05	10.12	61.48	7.22	10.24
13a	71	93— 95	-233.4	60.19	7.27	9.71	60.40	7.46	$9.78^{\mathrm{f}}$
14a	68	120—123	-196.2	63.52	6.92	9.24	63.30	7.13	9.00 <sup>f)</sup> 9.62 <sup>f)</sup>
15a	57	118-124	-250.6	61.00	7.06	9.53	61.06	7.34	$9.62^{f)}$
16a	66	102-104	-232.4	58.46	7.05	10.86	58.37	7.35	$10.75^{c)}$
17a	69	105—108	-213.8	59.46	6.98	9.96	59.14	7.37	9.85 <sup>c)</sup> 10.82 <sup>d)</sup>
19a	53	105-108	-221.6	56.89	6.91	10.95	57.20	7.27	$10.82^{d)}$
1b	93	115—117	-184.4	55.91	7.40	12.13	56.10	7.38	$11.90^{g}$
<b>2</b> b	75	175—177	-208.4	59.06	7.60	11.37	55.75	7.45	$12.04^{f}$
3b	87	178-180	-254.0	55.65	7.19	13.31	55.64	7.30	13.31 <sup>c)</sup>
<b>4</b> b	90	203 - 205	-303.0	56.57	7.29	13.34	56.31	7.33	13.40 <sup>d)</sup>
5b	84	208-210	-278.0	55.74	7.16	13.85	55.66	7.26	13.91 <sup>h)</sup>
<b>6</b> b	88	223 - 225	-312.2	56.25	7.21	13.78	56.52	7.28	13.93 <sup>e)</sup>
7b	80	90- 93	-142.0	51.12	6.97	12.45	51.34	7.26	12.60 <sup>f)</sup>
<b>8</b> b	61	123-128	-206.4	52.54	6.92	12.72	52.35	7.32	12.72 <sup>c)</sup> 11.68 <sup>d)</sup> 11.25 <sup>h)</sup>
9b	67	155-157	-224.2	53.96	6.82	11.52	54.08	7.56	$11.68^{d}$
10b	58	157—161	-242.2	53.89	7.60	11.21	54.01	7.77	11.25 <sup>h)</sup>
11b	85	143-146	-248.8	56.67	7.53	10.66	56.95	7.97	11.07°)
12b	67	173—175	-235.6	53.42	7.35	11.09	53.41	7.52	11.12 <sup>d)</sup>
13b	72	165-168	-230.0	52.97	7.49	10.67	52.64	7.77	10.58 <sup>e)</sup>
14b	69	186-191	-186.0	56.67	7.10	9.73	56.42	7.38	$9.68^{e)}$
15b	74	174—181	-251.8	54.86	7.43	10.53	54.94	7.53	$10.68^{d}$
16b	80	156-159	-232.6	52.56	6.88	11.87	52.46	7.53	9.68 <sup>e)</sup> 10.68 <sup>d)</sup> 11.84 <sup>h)</sup>
17b	72	164—166	-216.0	53.18	7.25	11.04	53.41	7.52	11.12 <sup>d)</sup> 10.91 <sup>d)</sup>
18b	85	143-146	-223.6	56.09	7.86	10.77	56.15	8.02	10.91 <sup>d)</sup>
19b	56	172—175	-244.6	52.66	6.93	12.20	52.47	7.34	12.24 <sup>d)</sup>
12c	77	147—151	-240.0	53.70	7.44	13.43	53.49	7.69	12.24 <sup>d)</sup> 13.37 <sup>d)</sup>
14c	70	163—166	-193.4	57.27	7.42	11.71	57.21	7.48	11.77 <sup>c)</sup>

a) c 1, MeOH. b) 0.5H<sub>2</sub>O. c) 1.5H<sub>2</sub>O. d) 2H<sub>2</sub>O. e) 3H<sub>2</sub>O. f) H<sub>2</sub>O. g) 0.25H<sub>2</sub>O. h) 2.5H<sub>2</sub>O.

Boc-(Pro)<sub>3</sub>-Gly-OH (1b). Compound la (5.57 g, 10 mmol) dissolved in MeOH (20 ml) was hydrogenated in the presence of 5% paradium carbon (ca. 2 g) for 4 h at room temperature. The product was recrystallized from EtOAcpetroleum ether; yield, 4.33 g (93%).

Compounds 2b—19b were prepared as describe for 1b.

*H*–(*Pro*)<sub>3</sub>–*Gly*–*OBzl*·*HCl*. Compound **1a** (2.78 g, 5 mmol) dissolved in 0.1 mol dm<sup>-3</sup> hydrogen chloride in formic acid (60 ml) was allowed to stand for 30 min at room temperature and evaporated. Addition of ether to the residue gave highly hygroscopic crystals, which were used for the next reaction without further treatment.

Boc groups in other peptides (2a-4a, and 12a) were removed as described above.

 $Boc\text{-}Ser\text{-}(Pro)_4\text{-}NH_2$  (12c). Compound 12a (137 mg, 0.2 mmol) dissolved in MeOH (3 ml) saturated with NH<sub>3</sub> was allowed to stand at room temperature for 3 d. After the evaporation of solvent, the residue was purified by silica gel (CHCl<sub>3</sub>-MeOH; 10:1, v/v) and Sephadex LH-20 (MeOH) column chromatographies; yield, 91 mg (77%).

Compound 14c was prepared as described for 12c.

CD Measurements. CD spectra were measured on a JASCO J-40A spectropolarimeter at 23 °C in  $H_2O$  solution with quartz cells of 1-mm path length. The ellipticity was expressed as mean residual molar ellipticity [ $\theta$ ] in degree cm<sup>2</sup> dmol<sup>-1</sup>.

We thank Drs. Tsuneko Uchida and Ukon Nagai of this institute for helpful discussion, and the members of Analysis Center, Mitsubishi Chemical Industries Ltd. for elemental analysis.

## References

- 1) T. Isemura, H. Okabayashi, and S. Sakakibara, *Biopolymers*, **6**, 307 (1968).
- 2) H. Okabayashi, T. Isemura, and S. Sakakibara, *Biopolymers*, **6**, 323 (1968).
- 3) H. Okabayashi and T. Isemura, *Bull. Chem. Soc. Jpn.*, **43**, 20 (1970).
- 4) H. Okabayashi and T. Isemura, *Bull. Chem. Soc. Jpn.*, **43**, 359 (1970).
- 5) The abbreviations used in this paper are those recommended by IUPAC-IUB: *J. Biol. Chem.*, **247**, 977 (1972). Additional abbreviations: Boc, *t*-butoxycarbonyl; OBzl, benzyl ester; Aoc, *t*-pentyloxycarbonyl; EDC, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide; TEA, triethylamine. Symbols of amino acids denote the L-configuration.
- 6) M. Tanaka, K. Sato, and T. Uchida, J. Biol. Chem., **256**, 11397 (1981).