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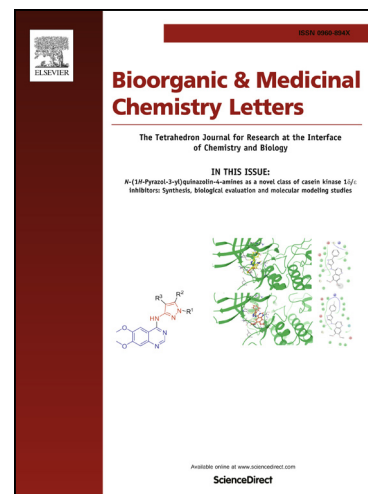
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Discovery of Potent and Selective CDK8 Inhibitors through FBDD Approach

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Abstract: A fragment library screen was carried out to identify starting points for novel CDK8 inhibitors. Optimization of a fragment hit guided by co-crystal structures led to identification of a novel series of potent CDK8 inhibitors which are highly ligand efficient, kinase selective and cellular active. Compound **16** was progressed to a mouse pharmacokinetic study and showed good oral bioavailability.

Keywords:

CDK8; inhibitor; fragment-based drug discovery (FBDD)

CDK8 is a Cyclin C-dependent serine-threonine kinase and a subunit of the Mediator complex that functions as a transcriptional regulator. In the past several years, multiple substrates of CDK8 have been identified, including histone H3, the RNA polymerase II (RNAPII) C-terminal domain (CTD), subunits of general transcription factors (GTFs) and certain transactivators.^{1,2} CDK8 plays important roles in oncogenic signalling pathways, including the Wnt- β -catenin pathway,^{3,4} the TGF β signalling pathway,⁵ the p53 pathway,^{6,7} the serum and hypoxia response network,^{8,9} the Notch and STAT1 signaling,^{10,11} and those governed by SMADs and the thyroid hormone receptor.^{5, 12} Furthermore, CDK8 was reported to be frequently dysregulated in breast cancer,¹³ colon cancer,³ gastric cancer¹⁴ and melanoma¹⁵. Inhibition of CDK8 by short hairpin RNA (shRNA) suppresses proliferation in cancer cells, and induces cell cycle arrest and apoptosis in *in vitro* and *in vivo* models.³ Thus,

targeting gene transcription through CDK8 is a potential approach for cancer therapy. Recently, there has been increased interest in targeting CDK8 and a number of CDK8 inhibitors have been reported by pharmaceutical companies and academic groups.¹⁶ Here, we disclose the identification of a novel class of potent and selective CDK8 inhibitors through fragment-based drug discovery (FBDD) approach.

Our approach began with a fragment screening of ~6500 compounds (heavy atom count less than 19 atoms) against CDK8 in a biochemical assay at a concentration of 100 μM to identify 403 primary hits showing > 70% inhibition. Evaluation of 227 selected primary hits by dose response narrowed down the list to 48 interesting fragments showing $\text{IC}_{50} < 50$ μM with ligand efficiency (LE) > 0.3.

In this manuscript we outline our efforts to optimize one fragment hit (**1**) which had an IC_{50} of 4.63 μM against CDK8 with an LE of 0.49.¹⁷ Initial exploration was conducted in an “SAR by catalog” approach utilizing similar compounds in Roche collection. This effort led to the identification of several compounds with improved potency, and accordingly higher ligand efficiency. As shown in Figure 1, removal of the ethyl group (compound **2**) resulted in a 16-fold increase in potency. Replacement of the cyano in compound **2** with a primary amide led to compound **3** showing slightly improved potency. *N*-monomethylation of the carbamoyl group (compound **4**) resulted in a ~2-fold loss in potency.

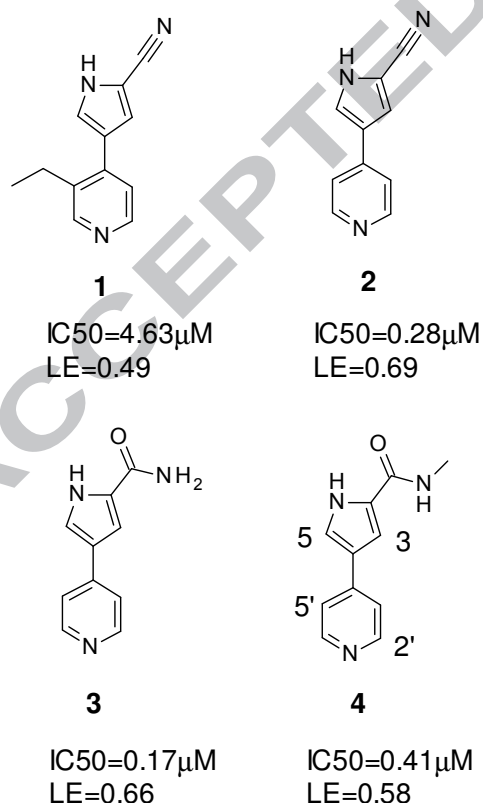


Figure 1. Potency and ligand efficiency of compounds **1~4**

To get a better understanding of the binding mode to assist compound design, we tried to co-crystallize the fragments with CDK8/cyclin C complex, and a co-crystal structure of compound **4** was obtained (Figure 2). The X-ray structure revealed that compound **4** occupied the ATP binding region of the CDK8/cyclin C complex which adopted an active conformation. In the hinge region, the pyridine nitrogen atom establishes a hydrogen-bonding interaction with the residue of Ala100, and the C-H at position 2' of pyridine (H-2', Figure 1) forms C-H \cdots O interaction with Asp98. Another direct hydrogen bond is formed between the amide carbonyl and the side chain of Lys52. A water molecule forms a hydrogen bond with the pyrrole nitrogen and thus indirectly links the compound **4** to the protein. The structure suggested that a small hydrophobic group at position 5 of the pyrrole or position 5' of the pyridine would potentially fill a hydrophobic site in the protein, and a small group would be tolerated at position 2' of the pyridine.

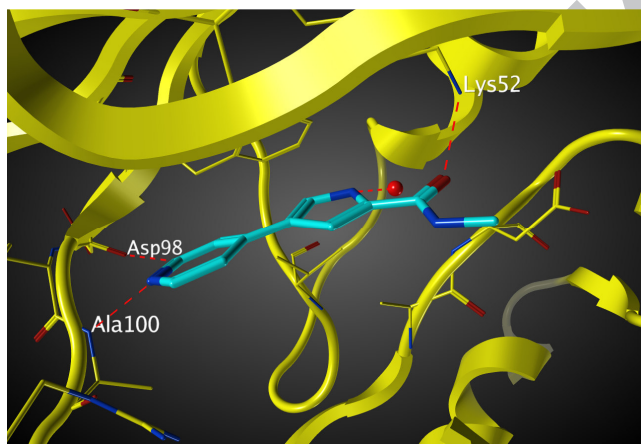
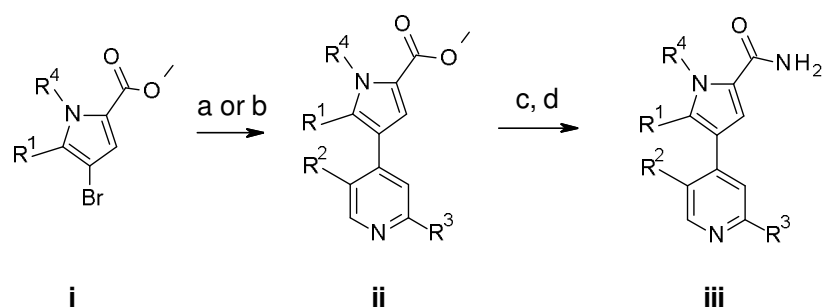


Figure 2. Co-crystal structure of compound **4** in CDK8/cyclin C (PDB code: 5XQX).

Compound **4** is highlighted in cyan sticks. Red dash lines represent the key hydrogen bonds and C-H \cdots O interaction.

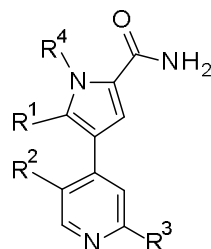
Based on these observations, we first synthesized compounds **5~11** to explore the SAR of these positions with the primary amide group fixed at position 2 of the pyrrole. The general synthetic strategy is depicted in Scheme 1. The key intermediate **ii** was synthesized by Suzuki coupling of substituted 4-Br pyrroles with appropriate 4-pyridylboronic acids, or converting the 4-Br pyrroles into the corresponding boronate ester followed by coupling with appropriate 4-Br pyridines.



Scheme 1. General synthetic approach for compounds **5~13**: (a) 4-pyridylboronic acids, Cs_2CO_3 , $\text{Pd}(\text{dppf})\text{Cl}_2$, EtOH/water, reflux; (b) $\text{Pd}_2(\text{dba})_3$, bis(pinacolato)diboron, KOAc, butyldi-1-adamantylphosphine, DME, 100 °C 1 h, then add 4-bromo-pyridines, K_2CO_3 , dioxane/ water, microwave 110 °C 1.5 h; (c) NaOH or LiOH, ethanol/water; (d) HATU or EDCI/HOBt, NH_3 .

Table 1

Potency and Ligand efficiency for compounds **5~13**



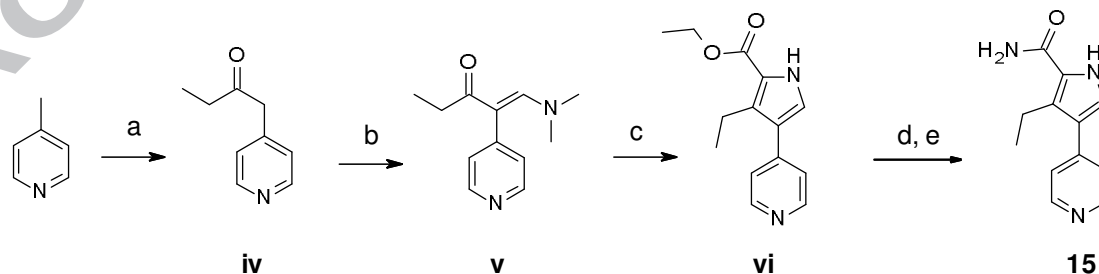
Compd	R ¹	R ²	R ³	R ⁴	CDK8 IC ₅₀ (μM)	LE
5	Me	H	H	H	0.24	0.60
6	Cl	H	H	H	0.16	0.62
7	H	Cl	H	H	0.019	0.70
8	H	Me	H	H	0.087	0.64
9	H	Br	H	H	0.037	0.68
10	H	H	Me	H	0.48	0.58
11	H	H	Cl	H	0.35	0.59
12	H	H	H	Me	4.95	0.48
13	H	H	H	Bn	0.17	0.44

As shown in Table 1, the addition of a small hydrophobic group at the R² position generated compounds **7~9** with improved potency. Specifically, compound **7** showed the

best potency with an IC_{50} of 0.019 μ M and LE up to 0.70. Methyl and chloro substitutions were well tolerated at the R^1 position (compounds **5** and **6**), but they led to diminished potency at the R^3 position (compounds **10** and **11**).

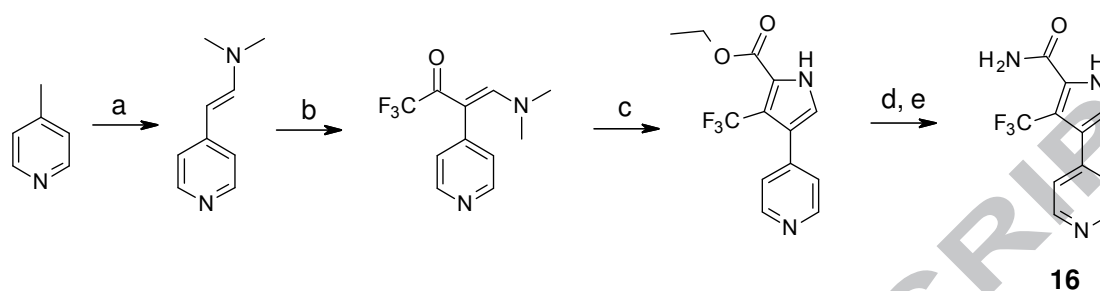
Compounds **12** and **13** (Table 1) were also synthesized using the route in Scheme 1 to explore the position 1 of the pyrrole. Methylation of the pyrrole nitrogen resulted in compound **12** with a ~29-fold loss in potency and a concomitant reduction in LE from 0.66 to 0.48, possibly due to the blockage of the hydrogen bonding interaction mediated by water. Unexpectedly, benzylation of the pyrrole nitrogen as in compound **13** did not diminish the potency. As the benzyl group also blocked the water mediated hydrogen bond, and might not be tolerated due to its relatively large size based on the co-crystal structure of compound **4** in CDK8, it was hypothesized that compound **13** and some other analogues might adopt a flip-over binding mode, in which the pyrrole nitrogen pointed outward while the hinge interactions were kept. The proposed flip-over binding mode also suggested that a small hydrophobic group at the position 3 of the pyrrole (R^5 , Table 2) would potentially enhance the hydrophobic interaction.

To test the hypothesis we synthesized compounds **14**~**19**. Synthesis of compounds **14** and **15** started from 4-methylpyridine. Scheme 2 depicts the synthesis of compound **15** as a representative procedure. Deprotonation of the 4-methylpyridine by LDA, followed by treatment with acyl chloride afforded ketone **iv**. The ketone was heated with DMF-DMA to give intermediate **v**, which was then heated with diethyl aminomalonate to give pyrrole **vi** with an ester group. Saponification, followed by coupling with ammonia afforded the desired compound **15**. The trifluoromethyl analogue **16** was prepared starting from enamine formation of 4-methylpyridine with 1-*tert*-butoxy-*N,N,N',N'*-tetramethylmethanediamine. The route is shown in Scheme 3. A similar approach as the one shown in Scheme 1 was used to synthesize compounds **17**, **18** and **19**.



Scheme 2. Synthesis of compound **15**: (a) LDA, THF, -78 °C, then propanoyl chloride, 21%; (b) DMF-DMA, dioxane, reflux, 8 h; (c) diethyl aminomalonate, acetic acid, reflux, 14 h, 20

% for 2 steps; (d) NaOH, ethanol/water, reflux, 4 h, 69%; (e) EDCI/HOBt, NH₃, DMF, rt, 5 h, 98%.



Scheme 3: (a) 1-*tert*-butoxy-*N,N,N',N'*- tetramethylmethanediamine, DMF, 130 °C, 16 h, 65%; (b) (CF₃CO)₂O, Et₃N, DCM, rt, overnight, 41%; (c) diethyl aminomalonate, acetic acid, 140 °C, 1 h, 11%; (d) conc. HCl, 80 °C, 4 h, 6%; (e) EDCI/HOBt, NH₃, DMF 1.5 h, 43%.

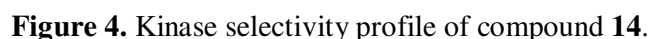
As can be seen from Table 2, the compounds with introduction of R⁵ achieved very good potency, in which compound **16** was the most potent one (IC₅₀ = 0.003 μM) and compound **17** achieved the highest ligand efficiency (LE = 0.76). The compounds were then tested in *in vitro* cell proliferation assay on AGS human gastric adenocarcinoma cells¹⁸ (CDK8 wild-type) and showed sub-micromolar to low micromolar activity.

Table 2

Potency, Ligand efficiency, and cellular activity for compounds **14~19**

Compd	R ²	R ³	R ⁵	CDK8 IC ₅₀ (μM)	LE	AGS IC ₅₀ (μM)
14	H	H	Me	0.008	0.74	0.80
15	H	H	Et	0.013	0.67	1.12
16	H	H	CF ₃	0.003	0.65	0.27
17	H	H	Cl	0.005	0.76	0.37
18	H	Me	Me	0.019	0.66	NA

In order to investigate the kinase selectivity, compound **14** was measured against a panel of 43 kinases and this compound demonstrated an excellent kinase selectivity profile. At a concentration of 1 μ M of the compound **14**, the inhibition against all kinases was less than 20% (Figure 4).



To understand the *in vivo* pharmacokinetic properties of this chemical series, compound **16** was selected for a mouse PK study. This compound displayed low systemic clearance, very good exposure and oral bioavailability (Table 3).

Table 3

Mouse pharmacokinetic profile of compound **16**.

Route	IV	PO
Dose level (mg/kg)	5	25
AUC	9378	25952
CL (mL/min/kg)	8.87	NA
C _{max} (ug/L)	9940	12740
T _{max} (h)	NA	0.25
T _{1/2} (h)	0.84	7.91
V _{ss} (L/kg)	0.49	NA
Bioavailability (%)	NA	57.5

In the work presented herein, using fragment based screening followed by “SAR by catalog” approach, we have discovered low molecular weight compounds to inhibit CDK8. With the guidance of co-crystal structures, optimization of the compounds led to the identification of a group of potent and selective CDK8 inhibitors which are cellular active and highly ligand efficient. The potent compound **16** with an IC₅₀ of 0.003 μ M against CDK8 represents >1500-fold improvement in potency over the initial fragment hit. Compound **16** also showed good oral bioavailability in a mouse PK study. As the molecular weight is still low, this chemical series has ample space for further optimization of *in vitro* and *in vivo* properties.

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