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> SHORT COMMUNICATIONS

Synthesis of *meta*-Carboranyl-(S)-homocysteine Sulfoxide

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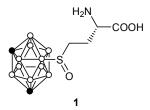
Abstract—New (S)-homocysteine derivatives containing a *meta*-carborane fragment were synthesized. *m*-Carboranyl-(S)-homocysteine sulfoxide was obtained as a mixture of diastereoisomers. The reduction of the side-chain carboxy group of *N*-*tert*-butoxycarbonyl-(S)-aspartic acid α -*tert*-butyl ester with sodium tetra-hydridoborate was not accompanied by racemization.

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Amino acid derivatives containing polyhedral dicarba-*closo*-dodecaborane (carborane) fragments in the side chain and unsubstituted functional groups in the α -position are promising agents for boron neutron capture therapy (BNCT) of cancer. Amino acid derivatives are efficiently transported to tumor cells [1–4], and boron-containing amino acid and peptide derivatives are capable of selectively accumulating in tumors [5–8].

We previously synthesized a series of lysine, glutamine, and asparagine derivatives containing an *o*-carborane fragment in the side chain and free carboxy and amino groups in the α -position [9, 10]. In this work we were the first to synthesize *S*-(*m*-carboran-9-yl)-(*S*)homocysteine sulfoxide [1, (2*S*)-2-amino-4-(1,7-dicarba-*closo*-dodecaboran-9-yl)sulfinylbutanoic acid] as a mixture of diastereoisomers.

Sulfur-substituted homocysteine and methionine derivatives are potent glutamine synthetase inhibitors [11, 12]. This enzyme is very important for the vital



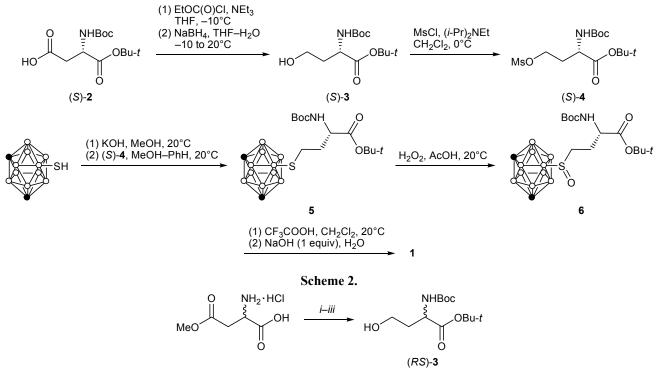
Hereinafter, light circles denote BH groups and boron atoms, and dark circles denote CH groups.

activity of *Mycobacterium tuberculosis*, so that it is the target of tuberculostatic agents [13, 14]. We believe that compound **1** could be used as a BNCT agent and also as a potential mycobacterial glutamine synthetase inhibitor.

By analogy with the previously reported procedure [15], the reduction of *N*-Boc-(*S*)-aspartic acid α -tertbutyl ester (*S*)-**2** afforded Boc-(*S*)-homoserine tertbutyl ester (*S*)-**3** without loss of enantiomeric purity (Scheme 1). According to the HPLC data (chiral stationary phase), the *ee* value of (*S*)-**3** was 98% [in comparison to racemate (*RS*)-**3**]. Protected homoserine (*RS*)-**3** was prepared starting from racemic aspartic acid methyl ester through the same reaction sequence as in the synthesis of (*S*)-**3** [10, 15] (Scheme 2).

Sulfonylation of (S)-3 and subsequent nucleophilic substitution gave *m*-carboranylhomocysteine derivative 5 in an overall yield of 47% (Scheme 1). Analogous approach to homocysteine derivatives was described in [16, 17]. By oxidation of 5 with hydrogen peroxide in acetic acid at 20°C we obtained sulfoxide 6, and deprotection of the latter by the action of trifluoroacetic acid afforded *m*-carboranyl-(S)-homocysteine sulfoxide 1 which was isolated in a moderate yield after purification by preparative reversed-phase chromatography. Some signals in the NMR spectra of Boc-amino acid *tert*-butyl esters were doubled due to the presence of stable conformers. Compounds 1 and 6 possess an asymmetric sulfur atom, and they were isoScheme 1.





Reagents and conditions: *i*: (1) *t*-BuOAc, HClO₄, 20°C; (2) Boc₂O, NEt₃, DMF, 20°C; *ii*: LiOH, THF–H₂O, 0–20°C; *iii*: (1) EtOCOCl, NEt₃, THF, -10°C; (2) NaBH₄, THF–H₂O, -10 to 20°C.

lated as mixtures of diastereoisomers distinguishable by NMR and HPLC.

tert-Butyl (2S)-2-(tert-butoxycarbonylamino)-4-hydroxybutanoate (S)-3. A solution of 0.65 g (2.25 mmol) of N-Boc-(S)-aspartic acid α -tert-butyl ester [10] in 5 mL of THF was cooled to -10° C, 0.24 mL of ethyl chloroformate and 0.35 mL of triethylamine were added, and the mixture was stirred for 30 min at -10° C. The precipitate was filtered off, a solution of 0.18 g (4.73 mmol) of sodium tetrahydridoborate in 5 mL of water was added to the filtrate, and the mixture was stirred for 22 h at 20°C. The mixture was then acidified with 1 M aqueous HCl to pH 3 and extracted with ethyl acetate $(3 \times 10 \text{ mL})$. The combined extracts were washed with 5% aqueous sodium hydrogen carbonate (2×15 mL) and brine (2×15 mL), dried over Na₂SO₄, and evaporated, and the residue was purified by flash chromatography using hexane-ethyl acetate (85:15 to 70:30) as eluent. Yield 0.46 g (70%), colorless oily material, $[\alpha]_D^{20} = -14.7^\circ$ (c = 0.63, EtOH); published data: $\left[\alpha\right]_{D}^{25} = -39.9^{\circ}$ (c = 1.0, EtOH) $[18], -37.5^{\circ}$ (c = 1, EtOH) [15]; ee 98%. HPLC (Chiralcel OD-H, hexane-*i*-PrOH-CF₃CO₂H, 10:1:0.02), $\tau = 8.4$ min. The ¹H and ¹³C NMR spectra of (S)-3 were identical to those reported in [18].

N-tert-Butoxycarbonyl-(*RS*)-aspartic acid α -tertbutyl ester (*RS*)-2 was synthesized as described in [10] for (*S*)-2 [10] from (*RS*)-aspartic acid β -methyl ester [19]. White powder, mp 111°C. The ¹H and ¹³C NMR spectra were identical to those of (*S*)-2 [10].

tert-Butyl (2*RS*)-2-(*tert*-butoxycarbonylamino)-4-hydroxybutanoate (*RS*)-3 was synthesized as described above for (*S*)-3 from 0.32 g (1.11 mmol) of (*RS*)-2. Yield 0.21 g (70%), colorless oily material. HPLC (Chiralcel OD-H, hexane–*i*-PrOH–CF₃CO₂H, 10:1:0.02): $\tau_{(R)-3} = 7.3$ min, $\tau_{(S)-3} = 8.4$ min. The ¹H and ¹³C NMR spectra were identical to those of (*S*)-3.

tert-Butyl (2*S*)-2-(*tert*-butoxycarbonylamino)-4-(methanesulfonyloxy)butanoate (*S*)-4. A solution of 0.4 g (1.38 mmol) of (*S*)-3 in 10 mL of methylene chloride was cooled to 0°C, 0.12 mL (1.53 mmol) of methanesulfonyl chloride and 0.72 mL (4.15 mmol) of ethyl(diisopropyl)amine were added, and the mixture was stirred for 2 h at 0°C. The mixture was washed with a 10% solution of citric acid (3×10 mL) and water (2×10 mL), the organic phase was dried over Na₂SO₄ and evaporated, and the residue was purified by flash chromatography using hexane–ethyl acetate (6:4 to 1:1) as eluent. Yield 0.44 g (89%), white powder, mp 87–90°C, $[\alpha]_D^{20} = +12.0^\circ$ (c = 0.51, CHCl₃); published data [20]: mp 85–87°C, $[\alpha]_D^{20} =$ +11.6° (*c* = 1.03, CHCl₃). ¹H NMR spectrum (CDCl₃), δ , ppm: 1.45 s (9H, *t*-Bu), 1.48 s (9H, *t*-Bu), 2.00– 2.10 m and 2.25–2.38 m (1H each, 3-H), 3.03 s (3H, MeSO₂), 4.25–4.36 m (2H, 4-H), 5.16 d (1H, NH, *J* = 6.7 Hz). ¹³C NMR spectrum (CDCl₃), δ_C , ppm: 27.94 (3C), 28.27 (3C), 32.21, 37.28, 50.90, 66.13, 80.11, 82.77, 155.35, 170.77. Found, %: C 47.72; H 7.83; N 4.12; S 9.31. C₁₄H₂₇NO₇S. Calculated, %: C 47.58; H 7.70; N 3.96; S 9.07.

tert-Butyl (2S)-2-(tert-butoxycarbonylamino)-4-[(1,7-dicarba-closo-dodecaboran-9-yl)sulfanyl]butanoate (5). 1,7-Dicarba-closo-dodecaborane-9thiol, 0.243 g (1.38 mmol), was added to a solution of 0.079 g (1.38 mmol) of potassium hydroxide in 3 mL of methanol, and the mixture was stirred under argon until it became homogeneous. A solution of 0.400 g (1.1 mmol) of (S)-4 in 3 mL of benzene was then added, and the mixture was stirred for 1.5 h at 20°C. Ethyl acetate, 20 mL, was added, and the mixture was washed with a 5% solution of NaHCO₃ (3×15 mL), water (15 mL), and brine (15 mL), dried over Na₂SO₄, and evaporated. The residue was purified by flash chromatography using hexane-ethyl acetate (90:10 to 80:20) as eluent. Yield 0.317 g (53%), colorless oily material, $[\alpha]_{D}^{20} = +17.0^{\circ}$ (c = 0.49, CHCl₃). ¹H NMR spectrum (DMSO- d_6), δ , ppm (a mixture of conformers A and B at a ratio of 8:2): 1.36 s (1.8H, *t*-Bu, B), 1.38 s (7.2H, t-Bu, A), 1.39 s (7.2H, t-Bu, A), 1.41 s (1.8H, t-Bu, B), 1.76–1.92 m (2H, 3-H), 1.00–3.20 m (9H, BH), 3.30–3.39 m (2H, 4-H), 3.83–3.90 m (0.2H, 2-H, B), 3.88–3.94 m (0.8H, 2-H, A), 4.09 br.s (2H, CH, carborane), 6.80 d (0.2H, NH, **B**, J = 7.5 Hz), 7.15 d (0.8H, NH, A, J = 7.9 Hz). ¹³C NMR spectrum (DMSO-*d*₆), δ_C, ppm: 27.57 (3C), 28.12 (3C), 28.61, 32.66, 53.31, 55.38 (2C), 77.98, 80.25, 155.44, 171.38. Found, %: C 41.42; H 8.32; N 3.23. C₁₅H₃₅B₁₀NO₄S. Calculated, %: C 41.55; H 8.14; N 3.23.

tert-Butyl (2S)-2-(*tert*-butoxycarbonylamino)-4-[(1,7-dicarba-*closo*-dodecaboran-9-yl)sulfinyl]butanoate (6, mixture of diastereoisomers). Compound 5, 0.300 g (0.69 mmol), was dissolved in 1.3 mL of acetic acid, 0.26 mL of 30% hydrogen peroxide was added, and the mixture was stirred for 30 min at 20°C. The mixture was treated with 10 mL of methylene chloride, the organic layer was separated, and the aqueous layer was adjusted to pH 8 by adding NaHCO₃ and extracted with methylene chloride (3×10 mL). The combined extracts were washed with a 5% solution of NaHCO₃ (3×15 mL) and water (2×15 mL), dried over Na₂SO₄, and evaporated, and

the residue was purified by flash chromatography using benzene-ethyl acetate (9:1 to 7:3) as eluent. Yield 0.215 g (65%), colorless amorphous powder. HPLC (Chiralcel OD-H, hexane-i-PrOH, 40:1), $\tau_1 = 13.6 \text{ min}, \tau_2 = 16.0 \text{ min}.$ ¹H NMR spectrum (DMSO- d_6), δ , ppm (a mixture of diastereoisomers A and **B** at a ratio of 85:15): 1.36 s (1.35H, t-Bu, **B**), 1.38 s (7.65H, t-Bu, A), 1.40 s (7.65H, t-Bu, A), 1.42 s (1.35H, t-Bu, B), 1.95–2.20 m (2H, 3-H), 1.50–3.20 m (9H, BH), 2.58–2.87 m (2H, 4-H), 3.93–4.03 m (1H, 2-H), 4.32 br.s (2H, CH, carborane), 6.95 br.s (0.15H, NH, B), 7.28–7.36 m (0.85H, NH, A). ¹³C NMR spectrum (DMSO- d_6), δ_C , ppm (a mixture of diastereoisomers A and B): 25.81 (A), 26.43 (B), 27.56 (3C, A, B), 28.09 (3C, A, B), 48.13 (A), 48.32 (B), 52.96 (A), 53.56 (**B**), 56.53 (2C, **A**, **B**), 78.12 (**A**), 78.15 (**B**), 80.57 (A, B), 155.39 (A), 155.41 (B), 170.84 (A), 170.89 (**B**). Mass spectrum: m/z 456.3396 $[M + \text{Li}]^+$. $C_{15}H_{35}^{10}B_2^{11}B_8LiNO_5S$. Calculated: M + Li 456.3394.

(2S)-2-Amino-4-[(1,7-dicarba-closo-dodecaboran-9-yl)sulfinyl|butanoic acid (1, mixture of diastereoisomers). Trifluoroacetic acid, 1.5 mL, was added to a solution of 0.195 g (0.42 mmol) of compound 6 in 1.5 mL of methylene chloride, and the mixture was stirred for 4 h at 20°C. The mixture was evaporated, the residue was dissolved in 2 mL of methanol, a 2 N solution of NaOH was added to pH 6-7, the resulting solution was evaporated, and the residue was purified first by flash chromatography using chloroformmethanol (9:1 to 7:3) as eluent and then by preparative HPLC. Yield 0.072 g (55%), colorless powder, mp 260°C (decomp.). ¹H NMR spectrum (D_2O+DCl), δ , ppm (a mixture of diastereoisomers **A** and **B**): 1.70– 3.40 m (9H, BH), 2.02–2.10 m (0.5H, 3-H, A), 2.11– 2.17 m (1H; 3-H, A; 3-H, B), 2.20–2.27 m (0.5H, 3-H, **B**), 2.76–2.82 m (0.5H, 4-H, **A**), 2.84–2.92 m (1H; 4-H, A; 4-H, B), 2.93-3.01 m (0.5H, 4-H, B), 3.29-3.36 m (1H, 2-H, A, B), 4.32 br.s (2H, CH, carborane), 7.74 br.s (3H, NH₂, COOH). ¹³C NMR spectrum $(DMSO-d_6)$, δ_C , ppm (A/B): 26.65 (A), 27.06 (B), 48.23 (A), 48.48 (B), 52.86 (A), 53.22 (B), 56.54 (2C, A, B), 168.88 (A), 169.05 (B). Found, %: C 24.60; H 6.38; N 4.57. C₆H₁₉B₁₀NO₃S. Calculated, %: C 24.56; H 6.53; N 4.77.

The ¹H and ¹³C NMR spectra were recorded at 25°C on a Bruker Avance 500 spectrometer (500 and 125 MHz, respectively) using tetramethylsilane as internal standard. The melting points were measured on a Stuart SMP3 melting point apparatus (Barloworld Scientific, UK). The optical rotations were determined with a Perkin Elmer Model 341 polarimeter. The

elemental analyses were obtained on a Perkin Elmer 2400 II automated CHNS-O analyzer. The high-resolution mass spectrum was recorded on a Bruker maXis impact HD instrument (electrospray ionization, positive ion detection; nebulizer gas nitrogen, flow rate 4 L/min, nebulizer pressure 0.4 bar; capillary voltage 4.5 kV). HPLC analysis of compounds 3 and 6 was performed on a Shimadzu LC-20 Prominence instrument (Japan) using a Chiralcel OD-H column, 250× 4.6 mm, grain size 5 µm (Daicel, Japan); detection at λ 220 nm; eluent flow rate 1.0 mL/min. Compound 1 was purified by preparative HPLC on a Reprosil-Pur C18-AO column, 250×20 mm, 10 µm (Dr. Maisch, Germany); detection at λ 210 nm; eluent acetonitrile– water, 9:1, flow rate 10 mL/min. Silica gel 60 (0.063-0.040 mm, Alfa Aesar, UK) was used for column flash chromatography.

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REFERENCES

- 1. Krasnov, V.P., Zhdanova, E.A., and Smirnova, L.I., *Russ. Chem. Rev.*, 1995, vol. 64, p. 1049.
- 2. Levit, G.L., Radina, L.B., and Krasnov, V.P., *Pharm. Chem. J.*, 1995, vol. 29, no. 8, p. 518.
- Singh, V.K. and Subudhi, B.B., Med. Chem. Res., 2015, vol. 24, p. 624.
- Abet, V., Filace, F., Recio, J., Alvarez-Builla, J., and Burgos, C., *Eur. J. Med. Chem.*, 2017, vol. 127, p. 810.

- 5. Kabalka, G.W. and Yao, M.-L., Anti-Cancer Agents Med. Chem., 2006, vol. 6, p. 111.
- Kimura, S., Masunaga, S., Harada, T., Kawamura, Y., Ueda, S., Okuda, K., and Nagasawa, K., *Bioorg. Med. Chem.*, 2011, vol. 19, p. 1721.
- El-Zaria, M.E., Genady, A.R., Janzen, N., Petlura, C.I., Beckford-Vera, D.R., and Valliant, J.F., *Dalton Trans.*, 2014, vol. 43, p. 4950.
- Kikuchi, S., Kanoh, D., Sato, S., Sakurai, Y., Suzuki, M., and Nakamura, H., *J. Controlled Release*, 2017, vol. 237, p. 160.
- Gruzdev, D.A., Levit, G.L., Bazhov, I.V., Demin, A.M., Sadretdinova, L.Sh., Ol'shevskaya, V.A., Kalinin, V.N., Krasnov, V.P., and Chupakhin, O.N., *Russ. Chem. Bull.*, *Int. Ed.*, 2010, vol. 59, p. 110.
- 10. Gruzdev, D.A., Levit, G.L., Olshevskaya, V.A., and Krasnov, V.P., *Russ. J. Org. Chem.*, 2017, vol. 53, p. 769.
- Eisenberg, D., Gill, H.S., Pfluegl, G.M.U., and Rotstein, S.H., *Biochim. Biophys. Acta*, 2000, vol. 1477, p. 122.
- 12. Berlicki, Ł., Mini-Rev. Med. Chem., 2008, vol. 8, p. 869.
- 13. Harth, G. and Horwitz, M.A., *Infect. Immun.*, 2003, vol. 71, p. 456.
- Mowbray, S.L., Kathiravan, M.K., Pandey, A.A., and Odell, L.R., *Molecules*, 2014, vol. 19, p. 13161.
- 15. Ramsamy, K., Olsen, R.K., and Emery, T., *Synthesis*, 1982, p. 42.
- Dalhoff, C., Hüben, M., Lenz, T., Poot, P., Nordhoff, E., Köster, H., and Weinhold, E., *ChemBioChem*, 2010, vol. 11, p. 256.
- 17. Zhang, Y., Pan, Y., Yang, W., Liu, W., Zou, H., and Zhao, Z.K., *ChemBioChem*, 2013, vol. 14, p. 1438.
- Qu, W., Zha, Z., Ploessl, K., Lieberman, B.P., Zhu, L., Wise, D., Thompson, C., and Kung, H.F., *J. Am. Chem. Soc.*, 2011, vol. 133, p. 1122.
- 19. Yoshioka, R., Ohtsuki, O., Senuma, M., and Tosa, T., *Chem. Pharm. Bull.*, 1989, vol. 37, p. 883.
- 20. Flohr, A., Aemissegger, A., and Hilvert, D., J. Med. Chem., 1999, vol. 42, p. 2633.