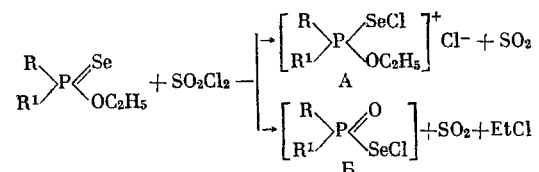


REACTION OF ESTERS OF PHOSPHORUS SELENOACIDS
WITH SULFURYL CHLORIDE

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In contrast to the data given in [1], we found that the esters of phosphorus selenoacids react with SO_2Cl_2 to give the chloroselenenylphosphonium salts A, and not the phosphorylated selenenyl chlorides B



In the case of (I, R = R¹ = C₆H₅O) and (II, R = C₆H₅O, R¹ = C₂H₅O) we obtained crystalline compounds with mp 38–39 and 33–34°, with the corresponding $\delta_{31\text{P}}$ –31 ppm for (IA) and –34 ppm for (IIA) relative to 85% H₃PO₄. The elemental analysis for P, C, and H corresponds to compounds (IA)–(IIA). The obtained compounds are soluble in CHCl₃ and CH₂Cl₂, and insoluble in ether and toluene. The reaction of (III, R = R¹ = OC₂H₅) with SO₂Cl₂ gives (IIIa) ($\delta_{31\text{P}}$ –40 ppm at –20°), which is unstable and decomposes in the solvent medium at 20° to give (C₂H₅O)₂P(O)Cl ($\delta_{31\text{P}}$ –3 ppm), C₂H₅Cl, and selenium.

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