Tandem asymmetric allylboration-alkene metathesis: a novel strategy for the synthesis of *trans*-disubstituted homoallylic alcohols

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Tandem use of Brown's asymmetric allylboration technology and crossed-alkene metathesis provides exclusively *trans*-disubstituted homoallylic alcohols in good yields.

The condensation reaction of allylborane reagents with aldehydes represents a strategy of considerable merit in asymmetric synthesis (Scheme 1). The reaction of an aldehyde 1 with an allylborane 2 proceeds via a 6-centre chair transition state to provide the adduct 3 and subsequently the homoallylic alcohol 4 on work-up. There is considerable versatility in this process. Variation of the substituent R² affords polyfunctional homoallylic alcohols $[R^2 = H, Me, OMe, OCH_2OMe, NPh_2,$ $N = CPh_2$, SiMe₂(NPrⁱ₂), *etc.*].^{1,2} The absolute stereochemistry of the reaction can be controlled by the choice of boron ligands, L, and the relative stereochemistry by the geometry of the allylborane 2: Z-allylboranes provide syn-\beta-substituted homoallylic alcohols, and E-allylboranes give rise to the antiisomers. There is currently a serious limitation to this methodology. For a given chiral α -substituted allylborane 2 (R³ \neq H), control of the alkene geometry in the product 4 requires that the absolute stereochemistry at C1 in reagent 2 be fixed.³ Whilst this is possible, preparation of chiral α -substituted allylboranes is by no means experimentally easy and the applications are not general; the existing methods are more appropriate for preparing Z-homoallylic alcohols.

The molybdenum alkene metathesis catalyst $Mo(OR_{F6})_2$ -(NAr)(CHCMe₂)Ph [R_{F6} = C(CF₃)₂Me, Ar = 2,6-Pri₂C₆H₃] **5** (Fig. 1) has found widespread use in ring opening metathesis polymerisation and ring closing metathesis of functionalised terminal alkenes.^{4,5} Until recently, this catalyst had not been applied to the cross-metathesis of unlike alkenes as the selectivity of the cross-coupling reaction had yet to be ascertained. Crowe has demonstrated that very high crossmetathesis selectivities can be achieved during the crosscouplings of both functionalised aryl alkenes and aryl/alkyl-



substituted alkenes using 1 mol% catalyst.⁶ Furthermore, these reactions are mild, operationally facile, and, most strikingly, proceed with >95% *trans* selectivity making crossed-alkene metathesis an enormously useful synthetic tool. Indeed, crossed-metathetic exchange could become quite powerful if used in tandem with other synthetic transformations. We now report that tandem asymmetric allylboration–alkene metathesis provides a convenient method for the preparation of *E*-homoallylic alcohols **4** from simple allylboranes **2** ($\mathbb{R}^3 = \mathbb{H}$).

Asymmetric allylboration reactions were conducted on a variety of achiral aldehydes at -100 °C using (-)-*B*-methoxydiisopinocampheylborane **6** under Brown's 'salt-free' conditions⁷ to furnish secondary homoallylic alcohols **7** in excellent ees.[†] To eliminate the Brønsted acidity, and thus permit tolerance by molybdenum catalyst **5**, homoallylic alcohols **7** were quantitatively converted to their *tert*-butyldimethylsilyl ethers (Scheme 2). Cross-metathesis of homoallylic silyl ethers **8** with a variety of *para*-substituted styrenes at room temperature using 1 mol% of catalyst **5** provided exclusively *trans*- disubstituted homoallylic alcohols in good yields (Scheme 3).[‡] The results are summarised in Table 1.§ It is again evident that the chemical yields are correlated with the degree of styrene activation.⁶

In conclusion, tandem asymmetric allylboration–alkene metathesis is an excellent procedure for the exclusive preparation of *trans*-disubstituted homoallylic alcohols. Investigations into the



Scheme 2 Reagents and conditions: i, CH₂=CHCH₂MgBr, Et₂O, room temp., 1 h; ii, anhydrous filtration; iii, RCHO, Et₂O, -100 °C, 1 h; iv, NaOH, H₂O₂; v, Bu⁴Me₂SiOSO₂CF₃, 2,6-lutidine, CH₂Cl₂, 0 °C, 15 min



Table 1 Cross-metathesis reactions of homoallylic silyl ethers with p-substituted styrenes

Entry	R	Ar	Yield (%)
1	Phenyl	Phenyl	72
2	Phenyl	p-ClC ₆ H ₄	54
3	Phenyl	p-MeOC ₆ H ₄	52
4	Cyclohexyl	Phenyl	65
5	Cyclohexyl	p-ClC ₆ H ₄	60
6	Cyclohexyl	p-MeOC ₆ H ₄	50
7	Propyl	Phenyl	54
8	Propyl	p-ClC ₆ H ₄	65
9	Propyl	p-MeOC ₆ H ₄	43

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Footnotes

[†] The purification procedure and spectral data for homoallylic alcohols derived from benzaldehyde and butyraldehyde are available from ref.7.

‡ A typical experimental procedure is as follows: to a solution of styrene (2 equiv.) and **5** (1 mol%) in CH_2Cl_2 (0.1 ml) was added a CH_2Cl_2 solution of **8** (0.5 M). The solution was allowed to stand for 3 h then adsorbed onto Na₂SO₄. Purification was accomplished by either flash or radial chromatography providing **9** as a clear, colourless oil.

§ All new compounds were fully characterised by spectroscopic data and combustion analysis or HRMS.

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