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Bioorganic & Medicinal Chemistry Letters

Bioorganic & Medicinal Chemistry Letters 17 (2007) 239–242

New antifungal flavonoid glycoside from *Vitex negundo*^{\approx}

B. Sathiamoorthy,^a Prasoon Gupta,^a Manmeet Kumar,^a Ashok K. Chaturvedi,^b P. K. Shukla^b and Rakesh Maurya^{a,*}

^aMedicinal and Process Chemistry Division, Central Drug Research Institute, Lucknow 226 001, India ^bDivision of Fermentation Technology, Central Drug Research Institute, Lucknow 226 001, India

> Received 4 July 2006; revised 15 September 2006; accepted 19 September 2006 Available online 5 October 2006

Abstract—Flavonoids are ubiquitous in photosynthesizing cells and are common part of human diet. For centuries, preparations containing these compounds as the principal physiologically active constituents have been used to treat human diseases. Increasingly, this class of natural products is becoming the subject of anti-infective research. Our bioactivity guided fractionation of ethanolic extract of leaves of *Vitex negundo* resulted in the isolation of new flavone glycoside (4) along with five known compounds 1–3, 5 and 6. All the isolated compounds were evaluated for their antimicrobial activities. The new flavone glycoside 4 and compound 5 were found to have significant antifungal activity against *Trichophyton mentagrophytes* and *Cryptococcus neoformans* at MIC 6.25 μ g/ml. © 2006 Elsevier Ltd. All rights reserved.

Resistance to antimicrobial agents has become a global problem since last three decades. About 2 million people in US acquire bacterial and fungal infections each year, of the 65% patients have resistance to at least one drug.^{1,2} A similar cause of concern in other countries including UK that leads to repeated use of antibiotic and insufficient control of the disease.³ Second adequate treatment of mycotic infections is difficult since fungi are eukaryotic organisms with a structure and metabolism that is similar to those of eukaryotic host. For this reason, substantial researches in the field of anti-infectives are now desperately needed to develop new prototype antimicrobial agents to avert this situation.

Development of new drugs from plants is not a new phenomenon. Plants and plant derived agents have long history to clinical relevance as source of potential chemotherapeutic agents.⁴ In our continuing efforts to identify antifungal agents from plant sources, we have studied antimicrobial activity of *Vitex negundo* (Leaves) against a variety of microorganisms including Gram-positive and Gram-negative bacteria and fungi (Fig. 1). *Vitex negundo* (Linn.) is a large shrub grown throughout the India. It is one of the common plants used in traditional medicine and reported to have variety of biological activities.⁵ The preparation of leaves used in catarrhal fever and applied to sinuses and scrofulous sores.⁶ Aqueous extract and oil of seeds possessed anti-oxidant and anti-inflammatory property.^{7–9} Antigenotoxic effects were reported from the leaves of *V. negundo* in combination with other constituents.¹⁰ Flavonoid-rich fraction of the seeds showed anti-fertility effect in adult dogs.¹¹ Plant also exhibited CNS depressant activity¹² and antihistamine release property.¹³ Flavonoids,^{14–16} iridoids,^{17–20} terpenes,^{21,22} and steroids²³ are the major classes of compounds isolated from this plant.

In our study, we have found significant antifungal activity in ethanolic extract against *Cryptococcus neoformans* and *Trichophyton mentagrophytes*. The extract of the leaves and twigs was reported to show antibacterial activity against *Micrococcus pyogenes* and *Escherichia coli*.⁶ On bioassay guided fractionation of ethanolic extract, the activity was localized in *n*-butanol fraction. Further work led to the isolation of compound (1–6) including a new flavone glycoside (4). All the six compounds were evaluated for antifungal and antibacterial activity.

Powdered leaves of the plant (5.0 kg) were extracted with ethanol by percolation, affording (590 g) extract.

Keywords: Vitex negundo; Antimicrobial activity; Fluconazole. * CDRI communication no. 7053.

^{*} Corresponding author. Tel.: +91 522 2612411 18x4235; fax: +91 522 2623405/2623938/2629504; e-mail: mauryarakesh@rediffmail.com

⁰⁹⁶⁰⁻⁸⁹⁴X/\$ - see front matter @ 2006 Elsevier Ltd. All rights reserved. doi:10.1016/j.bmcl.2006.09.051



Figure 1. Chemical structure of isolated compounds 1-6.

Ethanolic extract (550 g) was fractionated into n-hexane (180 g), CHCl₃ (70 g) and *n*-butanol (240 g) soluble fractions. The *n*-butanol soluble fraction was subjected to column chromatography over flash silica gel (230-400 mesh), eluted with a gradient of chloroform/methanol (95:05) to methanol/water (95:05) to afford six fractions (F-1 to F-6). Flash column chromatography of fraction F-1 using CHCl₃/MeOH (97:3) afforded compound 1 (50 mg). Purification of F-2 using CHCl₃/ MeOH (95:5) afforded compound 2 (350 mg) and with CHCl₃/MeOH (88:12) gave compound 3 (35 mg). Successive purification of F-3 by reverse-phase flash chromatography, eluted with a gradient of MeOH/H₂O (30:70), afforded 4 (25 mg). Column chromatography purification of F-4, using CHCl₃/MeOH (80:20), afforded 5 (500 mg). Repeated column chromatography of F-5, eluted fraction with EtOAc/acetone/H₂O (80:18:2), yielded compound 6 (50 mg).

Compound 4 was obtained as yellow colored solid. Positive Shinoda and Fiegel's test indicated that compound could be a flavonoid glycoside. The FAB-MS exhibited molecular ion peak $[M+H]^+$ at m/z 477 corresponding to the molecular formula $C_{22}H_{20}O_{12}$. IR absorption bands at 3250, 1746, 1640, and 1498 cm⁻¹ indicated the presence of hydroxyl group, ester moiety, conjugated carbonyl, and aromatic ring, respectively. UV spectrum showed λ_{max} at 336, 329, 266, and 209 nm (MeOH). A bathochromic shift (λ_{max} 329, 266, 216 nm) of 7 nm of the band II with NaOAc indicated the presence of free 7-OH group. Addition of NaOMe resulted in the bathochromic shift (λ_{max} 391, 324, 272 nm) of 55 nm of the band I confirmed the presence of free 4'-OH group. The UV spectrum of 4 was unaffected by the addition of NaOAc + H₃BO₃ (λ_{max} 328, 266, 210 nm) which clearly suggested the absence of ortho dihydroxyl system.24

¹H and ¹³C NMR (Table 1) spectrum showed the characteristic signal for H-3 proton at $\delta_{\rm H}$ 6.80 (s); $\delta_{\rm C}$ 104.1. A sharp singlet observed at $\delta_{\rm H}$ 12.8 was assigned to the C-5 chelated hydroxyl proton. The aromatic protons were resolved as one ABX system at $\delta_{\rm H}$ 7.60 (d,

Table 1. ¹H and ¹³C data of compound 4 in DMSO-*d*₆ (200 MHz)

Position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	HMBC ^a
2	_	164.1	H-3, H-2', H-6'
3	6.80 (s)	104.1	_
4	_	182.6	H-3
5		162.2	H-6
6	6.19 (d, $J = 2.0$)	99.7	H-8
7		165.0	H-6, H-8
8	6.45 (d, $J = 2.0$)	94.8	H-6
9	_	158.2	H-8
10	_	104.6	H-3, H-6, H-8
1'	_	122.4	H-2', H-6', H-5'
2'	7.60 (d, $J = 1.5$)	114.7	H-6′
3'		145.9	H-1", H-5', H-2'
4′	_	151.6	H-2', H-6', H-5'
5'	6.97 (d, J = 8.2)	117.5	H-6′
6′	7.62 (dd, <i>J</i> = 8.2, 1.5)	122.8	H-2', H-5'
1″	5.21 (d, $J = 6.2$)	101.6	H-3", H-5"
2″	3.38 (m)	73.7	H-1", H-3", H-4"
3″	3.30 (m)	76.1	H-2", H-3", H-5"
4″	3.42 (m)	72.2	H-3", H-5"
5″	4.19 (d, <i>J</i> = 8.9)	75.9	H-1", H-2", H-4"
6″		170.0	H-5", 6"-OMe
6"-OMe	3.67 (s)	52.9	_
5-OH	12.8 (s)	_	_

^a HMBC experiment in DMSO-*d*₆ (600 MHz).

J = 1.5 Hz), $\delta_{\rm C}$ 114.7 for H-2', $\delta_{\rm H}$ 6.97 (d, J = 8.2 Hz); $\delta_{\rm C}$ 117.5 for H-5' and $\delta_{\rm H}$ 7.62 (dd, J = 8.2, 1.5 Hz) for H-6'. The other AX system at $\delta_{\rm H}$ 6.19 (d, J = 2.0 Hz), $\delta_{\rm C}$ 99.7 and $\delta_{\rm H}$ 6.45 (d, J = 2.0 Hz), $\delta_{\rm C}$ 94.8 was assigned to H-6 and H-8 protons, respectively. A singlet at $\delta_{\rm H}$ 3.67 ($\delta_{\rm C}$ 52.9) was assigned to methoxy group of the esterified sugar. In case of 3', 4' dihydroxy flavone system, studies revealed that $\delta_{\rm C}$ value for 4' appeared downfield²⁴ than for 3'. Hence $\delta_{\rm C}$ 145.9, 151.6 were assigned to C-3' and C-4', respectively. The anomeric proton of the sugar appeared at $\delta_{\rm H}$ 5.21 (d, J = 6.2 Hz; $\delta_{\rm C}$ 101.6). The coupling constant of the anomeric proton (J = 6.2 Hz) confirmed the β -linkage of the sugar.²⁵ Using anomeric proton as starting point in ${}^{1}H^{-1}H$ COSY spectrum other sugar protons were assigned, respectively (Table 1). Downfield shift and splitting pattern of H-5 sugar proton with absence of C-6 methylene

Table 2. Antimicrobial activity of extract, fractions, and isolated compounds

Extract/fraction/pure compound	Minimum inhibitory concn (MIC) in µg/ml against										
	Bacteria			Fungi							
	1	2	3	4	5	6	7	8	9	10	11
Ethanolic extract	125	250	250	500	250	>500	62.5	>500	125	>500	>500
<i>n</i> -Hexane fraction	>500	_	500	>500	>500	>500	125	>500	>500	>500	>500
Chloroform fraction	250	500	500	>500	>500	>500	>500	>500	>500	>500	>500
<i>n</i> -Butanol fraction	250	250	250	500	125	>500	62.5	>500	250	>500	>500
1*	>50	25	50	50	25	>50	12.5	>50	25	50	25
2*	25	25	50	25	25	>50	25	>50	50	>50	25
3*	>50	50	>50	>50	>50	>50	12.5	>50	12.5	>50	>50
4*	>50	>50	>50	>50	>50	>50	6.25	>50	6.25	>50	>50
5*	>50	25	25	25	50	>50	6.25	>50	6.25	50	>50
6*	25	25	25	>50	25	>50	25	>50	12.5	50	>50
Flu	ND	ND	ND	ND	ND	0.5	1.0	2.0	2.0	2.0	1.0
Amp	0.02	0.09	0.02	50.0	0.09	ND	ND	ND	ND	ND	ND

1, Streptococcus faecalis; 2, Klebsiella pneumoniae; 3, Escherichia coli; 4, Pseudomonas aeruginosa; 5, Staphylococcus aureus; 6, Candida albicans; 7, Cryptococcus neoformans; 8, Sporothrix schenckii; 9, Trichophyton mentagrophytes; 10, Aspergillus funigatus; 11, Candida parapsilosis (ATCC-22019) Flu, fluconazole; Amp, ampicillin; ND, not done; *, pure compound.

of glucose indicated presence of carbonyl group at this position. This observation was well supported with presence of carbonyl carbon at $\delta_{\rm C}$ 170.0 in ¹¹C NMR spectrum. The HMBC spectrum was utilized to identify position of sugar, a long range correlation (Table 1) between H-1"/C-3' confirmed the attachment of sugar at C-3', other useful correlation between H-5"/C-6", -OCH3/C-6", H-2'/C-3',4', and H-5'/C-1',3' confirmed the position of methoxyl and carbonyl group. Further acid hydrolysis of 4 followed by co-TLC with authentic samples glycone (methyl ester of glucuronic acid) and aglycone (luteolin) was confirmed.²⁶ Further D-configuration of glycone was confirmed with optical rotation of acetate derivative $[\alpha]_{D}^{30}$ +11.4° (CHCl₃) with those reported in the literature.²⁷ Thus based on the foregoing evidence, the structure of compound 4 was elucidated as 4', 5, 7-trihydroxy-3'-O-β-D-glucuronic acid-6"-methyl ester, a new naturally occurring compound named vitegnoside. Other five known compounds. 5'-hvdroxy-3'. 4', 3, 6, 7-pentamethoxyflavone $(1)^{28}$, luteolin $(2)^{24}$, agnuside $(3)^{17}$, negundoside $(5)^{17,18}$, and *iso*-orientin $(\mathbf{6})^{29}$, were characterized by comparing their spectroscopic data with those reported in the literature.

All the isolated compounds 1-6 were evaluated for antifungal activity (Table 2). The compounds 4 and 5 showed promising activity against T. mentagrophytes and C. neoformans of MIC 6.25 µg/ml as compared to standard antifungal drug fluconazole, having MIC of 2 µg/ml against T. mentagrophytes. Compounds 1 and 3 showed activity of MIC 12.5 μ g/ml against C. neoformans, whereas compounds 3 and 6 showed activity against T. mentagrophytes. at similar concentration. Compound 2 showed moderate activity (25 µg/ml) against each test organism. The minimum inhibitory concentration (MIC) of standard drug fluconazole and test compounds was determined against test isolates by broth micro-dilution technique as per guidelines of NCCLS using 96-well tissue culture plates using RPMI 1640 media buffered with MOPS (3-[N-Morpholino] propanesulfonic acid) (Sigma Chemical Co.).^{30–33}

In conclusion, we have isolated five known compounds 1–3, 5 and 6 and a new flavonoid glycoside 4 from *V*. *negundo* through activity guided fractionation and discovered the potent antifungal activity of isolated compounds. The compounds 4 and 5 were found to be most active at MIC 6.25 μ g/ml among all the isolated compounds. The further chemical transformation of compounds 4 and 5 is under progress to improve their biological profile.

Acknowledgments

Authors are thankful to ICMR, New Delhi, for financial assistance. We also thank Dr. Ashish Arora for the HMBC spectral data and S. C. Tiwari for technical assistance.

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- 33. The fungal and the bacterial strains were grown on Sabouraud's dextrose agar and nutrient agar media, respectively. After the incubation fungal and bacterial growth were suspended in normal saline and maintained at $1.0-5.0 \times 10^3$ cfu/ml. The activity of compounds was determined by the NCCLS method for fungus using RPMI-1640 media buffered with MOPS (3-[N-Morpholino] propanesulfonic acid) (Sigma Chemical Co.) and Mueller-Hinton broth for bacteria. The 96-well tissue culture plates were used for 2-fold serial dilution. The proper growth control, drug control, and the blank were adjusted onto the plate. Compounds were dissolved in DMSO at a concentration of 1 mg/ml and 20 µl of this was added to 96-well tissue culture plate having 180 µl RPMI-1640 so the maximum concentration of the compound became 50 µg/ml. From here the solution was serially diluted resulting into the half of the concentration of test compounds and then inoculum was added and kept for incubation. Microtiter plates were incubated at 35 °C in a moist, dark chamber and MICs were recorded spectrophotometrically.