## Monosubstituted Azamines. Generation and Trapping Reactions

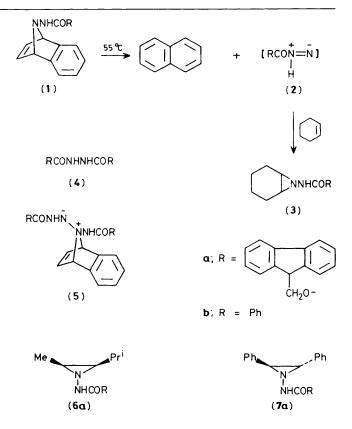
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Thermolysis of 7-acylamino-7-azabenzonorbornadienes and 1-(acylamino)-2,3-diphenylaziridines leads to the generation of monosubstituted azamines as shown by trapping reactions.

We report the first examples of the generation and capture by olefins of monosubstituted azamines (aminonitrenes, 1,1diazenes) (2). Thermolysis of  $(1a)^{\dagger}$  in cyclohexene at 55 °C led to the isolation of (3a) in 44% yield [(3b), 18%]. These results show that under these conditions (2) does not suffer 1,2hydrogen migration to the corresponding monosubstituted diimide.<sup>1</sup> In the absence of an olefinic trapping agent the major insoluble product along with naphthalene is  $(4a)^2$ (23%) or (4b) (35%). In the latter case the hydrazide (4b) is accompanied by 25% of 1-benzoyl-2-(1-naphthyl)hydrazine,<sup>3</sup> a product which must arise *via* simple ring-opening of the strained azabicyclic ring system. Hydrazide (4) may arise *via* interception of (2) by (1) to give (5) followed by thermolysis of

<sup>†</sup> Selected characterization data: (1a): <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>) δ 4.00– 4.55 (m, 3H, CHCH<sub>2</sub>), 4.70 (t, major) and 4.90 (br s, minor) [totalling 2H, bridgehead protons of two invertomers], 5.80 (br s, major) and 6.50 (br s, minor) [totalling 1H, NH of two invertomers], 6.75–7.90 (m, 14H, olefinic + aromatic), m.p. 84 °C (decomp.); (3a): <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>) δ 0.95–1.45 (br m, 4H, CH<sub>2</sub>CH<sub>2</sub>), 1.55–1.95 (br m, 4H, CH<sub>2</sub>) overlapping with following peak, 1.95–2.20 (br m, 2H, CHCH) overlapping with preceding peak, 4.00–4.60 (m, 3H, CHCH<sub>2</sub>), 6.30 (br s, 1H, NH), m.p. 163.5–165 °C; (6a): <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>) δ 0.90 (d, J 6 Hz, 3H, one Me of Pr<sup>i</sup>), 1.10–1.75 (m, 8H, ring Me + 2nd Me of Pr<sup>i</sup> + ring H + CH of Pr<sup>i</sup>), 1.85–2.15 (m, 1H, ring H), 4.10–4.55 (m, 3H, CHCH<sub>2</sub>), 6.20 (br s, 1H, NH), 7.15–7.90 (m, 8H, aromatic), m.p. 153–5 °C (decomp.); (7a): <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>) δ 3.55 (AB, J 5 Hz, 2H, CHCH), 4.00–4.50 (m, 3H, CHCH<sub>2</sub>), 5.73 (br s, 1H, NH), 7.00–7.90 (m, 18H, aromatic), m.p. 99–100.5 °C (decomp.) All new compounds gave satisfactory elemental analyses.



Novel techniques were developed for the synthesis of the 7-amino-7-azabenzonorbornadienes. The parent hydrazine (1; COR = H) was generated from the corresponding amine by direct amination via O-mesitylenesulphonylhydroxylamine<sup>5</sup> at -10 °C followed by immediate acylation with fluoren-9-ylmethyl chloroformate or benzoyl chloride. In view of the instability of the parent hydrazine it was conveniently stored as the fluoren-9-ylmethoxycarbonyl derivative (1a) from which the free hydrazine could be quickly regenerated by deblocking<sup>2</sup> via diethylamine in MeCN-CH<sub>2</sub>Cl<sub>2</sub> at 0 °C. Although of significant theoretical interest, this route to (2) is somewhat tedious for preparative purposes. A more practical route involves the readily-available<sup>6</sup> aziridine (7a) which in cyclohexene (cis-4-methylpent-2-ene) at 55 °C for 5.5 h gives 58% (3a) [76%, (6a)]. Thermolysis of the free hydrazine (1; COR = H) will be described separately. Schultz and coworkers<sup>7</sup> have reported the extrusion of azamine fragments from a variety of N-di-, mono-, and un-substituted 7-amino-7azanorbornadienes. In the case of the N-monosubstituted derivatives, the fate of the extruded fragment was not View Article Online

specifically considered. The thermolysis of arenesulphenyl analogues of (1) was also reported recently.<sup>8</sup>

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