UNUSUAL ACTION OF CARBOXYLIC ACID HALIDES

ON DIALKYLACETYLDITHIOPHOSPHITES

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It is known that O,O-dialkylacylphosphites react with carboxylic acid halides according to the Arbuzov reaction scheme to give ketophosphonates [1, 2].

We found that, in contrast, S,S-dialkylacetyldithiophosphites (Ia, b) react with acyl halides with the substitution of the acetoxy group for the halogen atom, giving the corresponding dithiophosphorous acid halide and acetic anhydride.

$$(RS)_2POCOCH_3 + CH_3C(O)X_{\rightarrow}(RS)_2PX + (CH_3CO)_2O$$

 (Ia, b) (IIa, b)
 $R = Et, X = Cl (a); R = Pr, X = Br (b)$

In an Arbuzov flask, 2.9 g of (Ia) and 1.07 g of acetyl chloride were mixed. The mixture was left to stand for 30 min at ~20°C. Distillation gave 1.87 g (72%) of (IIa), bp 52-53°C (0.02 mm), $\rm np^{20}$ 1.5792, $\rm \delta P$ 188 ppm (cf. [3]). Distillation of the liquid condensed in a liquid nitrogen trap gave 1.26 g (89%) of acetic anhydride, bp 137°C, $\rm np^{20}$ 1.3903, which corresponds to the data in [4].

The reaction of (Ib) with acetyl bromide was carried out similarly. From 4.55 g of (Ib) and 2.33 g of acetyl bromide, 1.6 g (8.2%) of acetic anhydride, bp 138-139°C, n_D^{20} 1.4009, and 3.45 g (69%) of (IIb), bp 74-75°C (0.02 mm), n_D^{20} 1.5836, δP 198 ppm, were obtained. Found: Br 30.37; P 11.62%. $C_6H_{14}BrPS_2$. Calculated: Br 30.65; P 11.88%.

LITERATURE CITED

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