#### Tetrahedron: Asymmetry 23 (2012) 852-858

Contents lists available at SciVerse ScienceDirect

Tetrahedron: Asymmetry

journal homepage: www.elsevier.com/locate/tetasy

# Stereospecific inversion of secondary tosylates to yield chiral methyl-branched building blocks, applied to the asymmetric synthesis of leafminer sex pheromones

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#### ARTICLE INFO

Article history: Received 17 April 2012 Accepted 15 May 2012 Available online 11 July 2012

# ABSTRACT

All four of the possible stereoisomers of 5,9-dimethylheptadecane, the major sex pheromone component secreted by female moths of the mountain-ash bentwing (*Leucoptera scitella*), were synthesized by the coupling of two chiral blocks with a methyl branch at the 2- or 3-position. The blocks were prepared by applying the stereospecific inversion of secondary tosylates, which were derived from (R)- and (S)-propylene oxide, and their enantiopurities were confirmed by chiral HPLC analysis.

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#### 1. Introduction

Species-specific sex pheromones have been identified from adult females of more than 600 lepidopteran species.<sup>1</sup> While pheromones are mainly composed of fatty alcohols, polyenyl hydrocarbons, and their derivatives with a straight chain, several species utilize methyl-branched compounds for mating communication.<sup>2</sup> In addition to the well-known methyl-branched alkanes, alkenes, and esters of a primary alcohol or acid, we recently found ketones and a secondary alcohol with a branched skeleton in the pheromone glands of female moths,<sup>3</sup> indicating the diverse chemical structures of lepidopteran sex pheromones. With regard to the large number of insect species, however, knowledge of the lepidopteran species secreting branched compounds remains very limited. Almost all methyl-branched compounds are chiral because a tertiary carbon is a stereogenic center, except for that at the 2-position. In order to understand the communication systems of moths, it is important to supply not only a sufficient amount of enantiopure known pheromones but also their analogues, which could possibly be produced by taxonomically related species. We investigated the possibilities of making chiral building blocks with high enantiomeric excess, which could be applied for the systematic synthesis of dimethyl hydrocarbons, such as 5,9-dimethylheptadecane 1 secreted by Leucoptera scitella females as a sex pheromone component.<sup>4</sup>

In addition to some unique synthetic routes,<sup>5</sup> the asymmetric synthesis of dimethyl hydrocarbons has reliably been accomplished by the coupling of two chiral methyl-branched building blocks derived from citronellal, 3-hydroxy-2-methylpropanoic

acid, and 3-methyl-4-butanolide.<sup>6</sup> The starting materials are commercially available but rather expensive. For another chiral source, we selected (R)- and (S)-propylene oxide (R)-**2** and (S)-**2**, which could be converted into the tosylates of 2-alkanols with a different chain length, as shown in Scheme 1. If an alkylating reagent attacks the tosylate via an S<sub>N</sub>2 reaction, the desired building block is formed after the inversion of configuration at the reaction site. By selecting an appropriate carbanion, it might be possible to produce chiral blocks **A** and **B** with a methyl group at the 3- and 2-positions, respectively. The two building blocks could be useful for the stereospecific construction of many n,(n+4)-dimethyl compounds, and so we began applying them to the asymmetric synthesis of all of the stereoisomers of **1**.

#### 2. Results and discussion

First, chiral blocks (*S*)-**A** and (*R*)-**A**  $[R^1 = n-C_4H_9, X = I; (S)-8$  and (R)-8] were synthesized starting from (R)-2 and (S)-2 (>99% ee), respectively, as shown in Scheme 2. The reaction between (R)-2 and  $n-C_3H_7MgBr$  with Li<sub>2</sub>CuCl<sub>4</sub> gave secondary alcohol (R)-3, which was converted into tosylate (R)-4. Antipodal tosylate (S)-4 was prepared from (*S*)-**2**. By HPLC equipped with a normal phase chiral column (Daicel Chiralpak AS-H), (R)-4 and (S)-4 could be detected separately. The chiral HPLC analysis confirmed their high enantiomeric excess (>99% ee), as shown in Figure 1A. In order to achieve an S<sub>N</sub>2 reaction associated with chain elongation, we attempted to utilize an enolate of tert-butyl acetate as the nucleophile because an ester elongated by two carbons would be produced directly. However, the expected product was not obtained. Therefore, we examined a reaction with an enolate of dimethyl malonate. While some studies have reported that the S<sub>N</sub>2 reaction of 2-bromo or 2mesyl compounds with enolates of dialkyl malonate proceed with





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**Scheme 1.** Synthetic strategy for  $n_i(n+4)$ -dimethylalkanes using stereospecific inversion of secondary tosylates.



Scheme 2. Synthetic routes to (*S*)-8 (chiral building block A, R<sup>1</sup> = *n*-C<sub>4</sub>H<sub>9</sub>, X = I) and the antipode for the synthesis of 5,9-dimethylheptadecane 1. Reagents and conditions: (i) *n*-C<sub>3</sub>H<sub>7</sub>MgBr, Li<sub>2</sub>CuCl<sub>4</sub> (0.1 mol/L, 3 mol %), THF (84%); (ii) TsCl, Et<sub>3</sub>N, Me<sub>3</sub>N·HCl (10 mol %), CH<sub>2</sub>Cl<sub>2</sub> (93%); (iii) NaCH(CO<sub>2</sub>Me)<sub>2</sub>, 1,2-dimethoxyethane (94%); (iv) LiCl, DMSO, H<sub>2</sub>O, 170–180 °C (82%); (v) LiAlH<sub>4</sub>, THF (89%); (vi) I<sub>2</sub>, PPh<sub>3</sub>, imidazole, THF (91%).



**Figure 1.** Chiral HPLC analyses of stereoisomers of 2-hexanyl tosylate **4** (A), 3methylheptan-1-ol **7** (B), and 2-methyl-1-(tolylsulfonyl)decane **12** (C). The HPLC was conducted with a Chiralpack AS-H column (4.6 mm ID × 25 cm) and a UV detector (254 nm) using 10% 2-propanol in hexane as an eluent (flow rate, 0.5 mL/min) for **4** and **12** and with a Chiralpack AY-H column (4.6 mm ID × 25 cm) and an RI detector using 0.4% 2-propanol in hexane as an eluent (flow rate, 0.5 mL/min) for **7**.

considerable racemization,<sup>7</sup> it has also been reported that the reaction of a secondary tosylate with an enolate forms a new C-C-bond along with the complete inversion of stereochemistry.<sup>8</sup> Based on this result. (R)-**4** was treated with the enolate of dimethyl malonate to vield geminal ester (S)-5, which was decarboxylated to ester (S)-6 by the Krapcho's method.<sup>9</sup> The reduction of (S)-6 with LiAlH<sub>4</sub> produced primary alcohol (S)-7, which had been synthesized from (S)- $\beta$ -citronellol in related research.<sup>3b</sup> The antipode (R)-7 was prepared in the same manner. The two chiral alcohols could be separated by another normal phase chiral HPLC column (Daicel Chiralpak AY-H), as shown in Figure 1B, and their high enantiomeric purity (>99% ee) indicated that the nucleophile attacked the tosylates 4 via an S<sub>N</sub>2 reaction. By treatment with a mixture of I<sub>2</sub>, triphenylphosphine (PPh<sub>3</sub>), and imidazole, (S)-7 and (R)-7 were converted into iodides (S)-**8** and (R)-**8**, respectively. The overall yield of (S)-8 was 49% from (R)-2, and that of (R)-8 was 53% from (S)-2.

Next, the chiral building block **B** was synthesized starting from (S)-2 and (R)-2, respectively, as shown in Scheme 3. The reaction between (S)-2 and *n*-C<sub>7</sub>H<sub>15</sub>MgBr with Li<sub>2</sub>CuCl<sub>4</sub> gave secondary alcohol (S)-9, which was converted into tosylate (S)-10. The antipodal tosylate (R)-10 was prepared from (R)-2. Chiral HPLC analysis confirmed their high enantiomeric purity (>99% ee). It has been reported that the cvanide ion and the base-generated carbanion of methylthiomethyl p-tolyl sulfone can be utilized as a nucleophile in an S<sub>N</sub>2 reaction,<sup>10</sup> but information on other reactions for one chain elongation associated with the complete inversion of configuration is to the best of our knowledge very limited. Therefore, we examined other reactions to make an optically active 2-methylundecanyl derivative. One idea was a coupling reaction between the secondary tosylate 10 and a sulfonyl acetate with an active methylene to yield a methyl-branched sulfonyl compound 11, which might be converted into building block **B** ( $R^2 = n - C_8 H_{17}$ , X = Ts,



**Scheme 3.** Synthetic routes to (*R*)-**12** (chiral building block **B**,  $R^2 = n-C_8H_{17}$ , X = Ts) and the antipode, which were coupled with block **A** to make all stereoisomers of 5,9-dimethylheptadecane **1**. Reagents and conditions: (i)  $n-C_7H_{15}$ MgBr, Li<sub>2</sub>CuCl<sub>4</sub> (0.1 mol/L, 3 mol %), THF (92%); (ii) TsCl, Et<sub>3</sub>N, Me<sub>3</sub>N·HCl (10 mol %), CH<sub>2</sub>Cl<sub>2</sub> (93%); (iii)  $p-Me(C_6H_4)SO_2CH_2CO_2Me/base;$  (iv)  $p-Me(C_6H_4)SCH_2CO_2Me/base;$  (v)  $p-Me(C_6H_4)SO_2Me/BuLi$ , THF (72%); (vi) BuLi, THF, HMPA (86%); (vii) Mg, MeOH (79%).

**12**) after decarboxylation. Methyl (toluene-4-sulfonyl)acetate, which was prepared with sodium *p*-toluenesulfinate and methyl 2-chloroacetate, was treated with NaH, but the base-generated carbanion did not couple to **10** even when heating at 85 °C for 24 h. Another reaction of **10** with the dianion of sulfonylacetate, which was prepared by treatment with NaH and butyllithium (BuLi), produced a complicated mixture. Next, we attempted the reaction with a carbanion of methyl *p*-tolylsulfanylacetate, which was synthesized by the reaction with 4-methylbenzenethiol and methyl 2-chloroacetate. When lithium diisopropylamide (LDA), so-dium hexamethyldisilazide (NaHMDS), or potassium hexamethyl-disilazide (KHMDS) were used as a base, the desired sulfanyl product **13** was not obtained.

However, the carbanion of 1-methanesulfonyl-4-methylbenzene (methyl *p*-tolyl sulfone), which was generated by treatment with BuLi, successfully attacked (*S*)-**10**, and (*R*)-**12** was produced. We examined the best reaction conditions by changing the type of base, solvent, temperature, and substrate concentration, as shown in Table 1. The highest yield (72%) was achieved with BuLi in THF at 55 °C and a low concentration of methyl *p*-tolyl sulfone. When a THF solution of the carbanion was stirred without (*S*)-**10** for several minutes, the yield of (*R*)-**12** decreased. The coupling reaction between an anion of the sulfone and an alkyl halide usually required the addition of a very polar solvent, such as hexamethylphosphoric triamide (HMPA),<sup>11</sup> but the co-solvent had a

Table 1

Synthesis of methyl (R)-2-methyl-1-(4-methylphenylsulfonyl)decane (**12**) by coupling of a tosylate [(S)-**10**] and a carbanion of methyl p-tolyl sulfone (S)-**10** 

p-Me(C <sub>6</sub> H <sub>4</sub> )SO <sub>2</sub> Me + base (R)-12·					
Entry	Base	Solvent	Temp (°C)	Concentration (M)	Yield (ee)
1	BuLi	THF	55	0.5	41% (>99%)
2	BuLi	THF	55	0.07	72% (>99%)
3	BuLi	THF	20	0.07	None
4	BuLi	THF-HMPA	55	0.07	None
5	BuLi	THF-HMPA	20	0.07	23% (>99%)
6	BuLi	THF-NMP	20	0.07	None
7	LDA	THF	55	0.07	18%
8	NaHMDS	THF	55	0.07	30%
9	KHMDS	THF	55	0.07	None

negative effect on the synthesis of (R)-12. The S<sub>N</sub>2 reaction of a primary alkyl halide by the anion of methyl sulfone has been more often reported than that of an alkyl tosylate.<sup>6a,d</sup> To the best of our knowledge, the  $S_N2$  reaction of **10** is the first successful example of an intermolecular coupling between an optically active secondarv tosvlate and an anion of a sulfonyl compound, although an intramolecular  $S_N2$  reaction compound with a secondary tosyl and methanesulfonyl groups has been reported.<sup>11</sup> The desired intermolecular S<sub>N</sub>2 reaction produced successfully the methylbranched sulfone 12 without racemization and E2 elimination of the secondary tosylate. The absolute configuration was confirmed by comparison with (R)-12, which was synthesized starting from methyl (S)-3-bromoisobutylate 14 by the route shown in Scheme 4. By using *N*-methylpyrrolidinone (NMP) as the co-solvent,<sup>12</sup> bromide **14** was treated with *n*-C<sub>7</sub>H<sub>15</sub>MgBr and Li<sub>2</sub>CuCl<sub>4</sub> to yield methyl 2-methyloctadecanoate 15. Reduction of 15 with LiAlH<sub>4</sub>, iodination of the produced alcohol, and the substitution of the iodine with a sulfonyl group yielded (R)-12. The sulfones synthesized by two different routes showed almost the same specific rotation values. Compound (S)-12 was prepared from (R)-10, and the high enantiomeric purity of (R)-12 and (S)-12 (>99% ee) was confirmed by chiral HPLC analyses using an AS-H column (Fig. 1C). The total yield of (*R*)-12 was 62% from (*S*)-2, while that of (*S*)-12 was 63% from (R)-2.

Sulfone (*R*)-12 was alkylated with (*S*)-8 to give (5*S*,9*R*)-16 with a 5,9-dimethyl skeleton by using a known procedure.<sup>6a,d</sup> Reductive removal of the sulfonyl moiety of (5S,9R)-16 was achieved smoothly by using magnesium turnings in methanol,<sup>13</sup> and the desired final product (5S,9R)-1 was obtained in 68% yield from (R)-12. Before the reaction, magnesium was activated with a catalytic amount of methylmagnesium bromide (MeMgBr) in THF. No olefinic by-products were formed, as reported by Kuwahara et al. using a method involving lithium in liquid ethylamine.<sup>6d</sup> In the same manner, three other stereoisomers were synthesized by the following combination of two synthetic building blocks; (55,95)-1 from (S)-8 and (S)-12, (5R,9R)-1 from (R)-8 and (R)-12, and (5*R*,9*S*)-1 from (*R*)-8 and (*S*)-12. The chemical data for each isomer were almost identical to those published with regard to previous syntheses,<sup>6</sup> indicating that racemization did not occur after the alkylation. While the <sup>13</sup>C NMR spectra of the enantiomers were identical, the diastereomers showed small but discernible differ-



Scheme 4. Synthetic route to (*R*)-12 starting from methyl (*S*)-3-bromoisobutyrate 14. Reagents and conditions: (i) *n*-C<sub>7</sub>H<sub>15</sub>MgBr, Li<sub>2</sub>CuCl<sub>4</sub>, NMP–THF (91%); (ii) LiAlH<sub>4</sub>, THF; (iii) l<sub>2</sub>, PPh<sub>3</sub>, imidazole, THF; (iv) *p*-Me(C<sub>6</sub>H<sub>4</sub>)SO<sub>2</sub>Na, PEG400, MeOH (44% from 15).

ences of chemical shifts for some <sup>13</sup>C signals as indicated for 5,9dimethylpentadecane.<sup>6d</sup> We tentatively assigned each <sup>13</sup>C signal, and the assignments indicated the differences for carbons of the two branched-methyl groups and carbons around the branches, reflecting the conformational differences of the diastereomers (see Section 4).

# 3. Conclusion

In many syntheses of chiral methyl-branched insect pheromones, the original configuration at a stereogenic center in the starting material was retained in the final product.<sup>14</sup> Stereospecific conversion has been rarely used in pheromone syntheses because of the risk of racemization. Our synthetic strategy, however, utilized the S<sub>N</sub>2 reaction of a tosylate of a 2-alkanol. The reason was that we expected the possibility of enantiomeric separation of methyl-branched synthetic intermediates by a chiral HPLC column.<sup>3</sup> In fact, we could confirm, by chiral HPLC analyses, the stereospecific conversion of a chiral secondary tosylate after an S<sub>N</sub>2 reaction with an alkylating reagent, such as the enolate of dimethyl malonate and the anion of methyl sulfone. In the reaction, elongation by two or one carbon(s) was accomplished, in association with complete inversion of the original stereochemistry, which successfully produced two chiral methyl-branched building blocks, A and **B**, with high enantiomeric purity (>99% ee). In addition to 5,9-dimethylheptadecane **1**, these building blocks can be utilized for the synthesis of other 5,9-dimethylalkanes with a different carbon chain identified from Lyonetiidae species as well as 7,11-dimethylalkanes identified from Geometridae species. The synthetic pheromones will be utilized as a monitoring tool for pest control and an authentic standard for the identification of new pheromones.<sup>2</sup>

### 4. Experimental

#### 4.1. General

The specific rotation of each CHCl<sub>3</sub> solution was measured on a JASCO DIP-4 polarimeter. IR spectra were recorded as a thin film (neat liquid) with a Jasco FT/IR-350 (JASCO, Tokyo, Japan). <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded by a JNM-ECA 500 FT-NMR spectrometer (JEOL, Tokyo, Japan) at 500.16 and 125.77 MHz, respectively, for CDCl<sub>3</sub> solutions containing TMS as the internal standard. GC-MS was conducted in EI mode (70 eV) with an HP5973 mass spectrometer system (Hewlett-Packard) equipped with a split/splitless injector and a DB-23 column (0.25 mm  $ID \times 30$  m, 0.25  $\mu$ m film, J & W Scientific, Folsom, CA, USA). The column temperature program was 50 °C for 2 min, 4 °C/min to 150 °C, 10 °C/min to 220 °C, and 220 °C for 10 min. The carrier gas was He. High-resolution MS (HR-MS) analysis was performed with a IMS-MS700V mass spectrometer (JEOL). Enantiomeric excess (ee) was determined by a Chiral HPLC analysis, which employed a system composed of a pump (PU-980, JASCO), an RI detector (RI-98SCOPE, Labo System, Tokyo, Japan) or a UV detector (UV-970, JASCO), an integrator (807-IT, JASCO), and a normal phase chiral column, Chiralpak AS-H or AY-H (4.6 mm ID  $\times$  25 cm, Daicel Chemical Industry, Osaka, Japan). The resolution of tosylates 4 and 10 and sulfone 12 was achieved with the AS-H column using 10% 2-propanol in hexane as an eluent, and each enantiomer was detected by UV 254 nm. The resolution of alcohol **7** was achieved with the AY-H column using 0.4% 2-propanol in hexane, and each enantiomer was detected by RI. The flow rate of the eluents was 0.5 mL/min.

### 4.2. 2-Hexanols (R)- and (S)-3

To an ice-cooled and stirred mixture of  $n-C_3H_7Br$  (7.38 g, 60 mmol) and magnesium turnings (2.92 g, 120 mmol) in dry THF (60 mL), a catalytic amount of MeMgBr (3.0 M, two drops) was added as an initiator for making a Grignard reagent under an argon atmosphere. After stirring for 1 h at 0 °C, Li<sub>2</sub>CuCl<sub>4</sub> in THF (0.1 M, 18 mL, 1.8 mmol) was added to a solution with a cloudy gray color at -78 °C. Then, (*R*)-(+)-propylene oxide (*R*)-2 (2.90 g, 50 mmol) dissolved in dry THF (5 mL) was added dropwise through a syringe, and stirring of the mixture was continued for 30 min at -78 °C and an additional 1 h at room temperature. The reaction mixture was then poured into an ice-cooled aqueous solution of NH<sub>4</sub>Cl (300 mL), and the crude products were extracted with Et<sub>2</sub>O. The extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (50 g). Elution with hexane/EtOAc (4:1) afforded 4.28 g (42 mmol, 84%) of (*R*)-**3** as an oil,  $[\alpha]_D^{23} = -8.1$  (*c* 1.5, CHCl<sub>3</sub>);  $v_{max}$ (cm<sup>-1</sup>): 3354 (s, O-H), 2925 (s), 2854 (s), 1458 (m), 1377 (m), 1115 (m);  $\delta_{\rm H}$ : 0.91 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 7.0 Hz), 1.19 (3H, d, CH<sub>3</sub>CHOH I = 6.2 Hz, ~1.3 (4H, m), ~1.4 (2H, m), 3.80 (1H, m, CH<sub>3</sub>CHOH);  $\delta_{C}$ : 14.0, 22.7, 23.4, 28.0, 39.0, 68.1. In the same manner, (S)-(-)-propylene oxide (*S*)-**2** (2.90 g, 50 mmol) was reacted with *n*-C<sub>3</sub>H<sub>7</sub>MgBr to yield 4.59 g (45 mmol, 90%) of (*S*)-**3** as an oil,  $[\alpha]_D^{23} = +7.9$  (*c* 1.2, CHCl<sub>3</sub>).

#### 4.3. 2-Hexanyl tosylates (R)- and (S)-4

Triethylamine (8.10 g, 80 mmol) was added dropwise to a stirred solution of tosyl chloride (TsCl, 11.4 g, 60 mmol) and Me<sub>3</sub>N·HCl (0.38 g, 4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (70 mL) at 0 °C under an argon atmosphere. Then (R)-3 (4.08 g, 40 mmol) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (8 mL) was added to the solution through a syringe. After stirring at 0 °C for 1.5 h, 1,3-diaminopropane (1.48 g, 20 mmol) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (2 mL) was added to decompose the excess TsCl. The mixture was stirred for 10 min and poured into water, after which the crude product was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The extract was washed with dil HCl, saturated NaHCO3 solution, and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (50 g). Elution with hexane/EtOAc (9:1) afforded 9.49 g (37 mmol, 93%) of (*R*)-**4** as an oil,  $[\alpha]_D^{23} = -3.7$  (*c* 2.0, CHCl<sub>3</sub>); v<sub>max</sub> (cm<sup>-1</sup>): 2931 (s), 2856 (s), 1599 (m, benzene ring), 1363 (s, S=0), 1176 (s, S=0);  $\delta_{\text{H}}$ : 0.81 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.9 Hz), ~1.2 (4H, m), 1.25 (3H, d,  $CH_3$ CHOTs, J = 6.3 Hz), 1.50 and 1.59 (2H, m), 2.45 (3H, s, CH<sub>3</sub>Ph), 4.60 (1H, tq, J = 6.3, 6.3 Hz, CH<sub>3</sub>CHOTs), 7.33 (2H, d, CH=CH, I = 8.2 Hz), 7.80 (2H, d, CH=CH, I = 8.2 Hz);  $\delta_C$ : 13.8, 20.8, 21.6, 22.2, 27.0, 36.2, 80.7, 127.7 (×2), 129.7 (×2), 134.6, 144.4; GC-MS: t<sub>R</sub> 20.32 min, m/z 256 (M<sup>+</sup>, 1%), 199 (18%), 155 (100%). In the same manner, (S)-3 (4.08 g, 40 mmol) was treated with tosyl chloride to yield 10.0 g (39 mmol, 98%) of (S)-4 as an oil,  $[\alpha]_{D}^{23} = +3.8$  (*c* 2.1, CHCl<sub>3</sub>). The two enantiomers were analyzed by chiral HPLC equipped with a Chiralpak AS-H column and a UV

detector (254 nm). Peaks of the two enantiomers were detected at different  $t_{\rm R}$  values [(*R*)-**4**, 18.8 min and (*S*)-**4**, 16.0 min] by elution with hexane/2-propanol (9:1) at a flow rate (0.5 mL/min), and chromatograms showed their enantiomeric purity (>99% ee).

# 4.4. 2-(1-Methylpentyl)malonic acid dimethyl esters (S)- and (R)-5

Dimethyl malonate (7.13 g, 54 mmol) was added dropwise to a suspension of sodium hydride (60%, 2.16 g, 54 mmol) in 1,2-dimethoxyethane at 0 °C and stirred for 15 min. Then (R)-4 (9.23 g, 36 mmol) was added to the mixture, which produced sodium ptoluenesulfonate as a white solid after stirring at 80-85 °C for 12 h. The suspension was concentrated in vacuo and a saturated NH<sub>4</sub>Cl solution was poured into the residue. The product was extracted with EtOAc, and the extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, concentrated in vacuo, and chromatographed over SiO<sub>2</sub> (50 g). Elution with hexane/EtOAc (9:1) afforded 7.35 g (34 mmol, 94%) of (*S*)-**5** as an oil,  $[\alpha]_D^{23} = -2.9$  (*c* 1.2, CHCl<sub>3</sub>);  $\nu_{max}$  (cm<sup>-1</sup>): 2956 (s), 1737 (s, C=O), 1435 (s), 1151 (s), 1022 (m);  $\delta_H$ : 0.89 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.8 Hz), 0.97 (3H, d, CH<sub>3</sub>CH, J = 6.8 Hz), ~1.3 (6H, m), 2.25 (1H, m, CH<sub>3</sub>CH), 3.27 (1H, d, CHCO<sub>2</sub>CH<sub>3</sub>, I = 8.2 Hz, 3.73 (6H, s, CO<sub>2</sub>CH<sub>3</sub>);  $\delta_{C}$ : 14.0, 17.0, 22.7, 29.0, 33.5, 34.0, 52.2, 52.3, 57.6, 169.3, 169.5; GC-MS: t<sub>R</sub> 12.04 min, m/z 185  $([M-32]^+, 10\%)$ , 132 (100%). In the same manner, (S)-4 (9.74 g, 38 mmol) was treated with sodium malonic ester to yield 7.35 g (34 mmol, 89%) of (*R*)-**5** as an oil,  $[\alpha]_{D}^{23} = +3.0$  (*c* 0.74, CHCl<sub>3</sub>).

### 4.5. Methyl 3-methylheptanoates (S)- and (R)-6

Water (2.34 g, 130 mmol) was added to a solution of (S)-5 (7.14 g, 33 mmol) and lithium chloride (2.80 g, 66 mmol) in DMSO (170 mL), and the mixture was stirred at 170-180 °C for 3.5 h. After cooling, the mixture was poured into NaCl solution (12%, 500 mL) and the product was extracted with EtOAc. The extract was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, concentrated in vacuo, and chromatographed over SiO<sub>2</sub> (50 g). Elution with hexane/EtOAc (9:1) afforded 4.27 g (27 mmol, 82%) of (S)-**6** as an oil,  $[\alpha]_{D}^{23} = -3.1$ (c 0.82, CHCl<sub>3</sub>); v<sub>max</sub> (cm<sup>-1</sup>): 2956 (s), 2858 (s), 1741 (s, C=O), 1437 (s), 1173 (s), 1009 (m);  $\delta_{\rm H}$ : 0.89 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.8 Hz), 0.93 (3H, d, CH<sub>3</sub>CH, J = 6.6 Hz), ~1.3 (6H, m), 1.95 (1H, m, CH<sub>3</sub>CH), 2.11 (1H, dd, CHHCO<sub>2</sub>CH<sub>3</sub>, *I* = 14.5, 8.1 Hz), 2.31 (1H, dd, CHHCO<sub>2</sub>CH<sub>3</sub>, I = 14.5, 6.1 Hz), 3.67 (3H, s, CO<sub>2</sub>CH<sub>3</sub>);  $\delta_{C}$ : 14.1, 19.8, 22.8, 29.1, 30.3, 36.4, 41.7, 51.4, 173.9; GC-MS:  $t_{\rm R}$  6.53 min, m/z 158 (M<sup>+</sup>, 1%), 127 (11%), 74 (100%). In the same manner, decarboxylation of (*R*)-**5** (7.57 g, 35 mmol) produced 4.59 g (29 mmol, 83%) of (*R*)-**6** as an oil,  $[\alpha]_D^{23} = +2.9$  (*c* 0.63, CHCl<sub>3</sub>).

#### 4.6. 3-Methylheptan-1-ols (S)- and (R)-7

A solution of (S)-6 (4.27 g, 27 mmol) in dry THF (10 mL) was added dropwise through a syringe to an ice-cooled and stirred suspension of LiAlH<sub>4</sub> (1.02 g, 27 mmol) in dry THF (50 mL) under an argon atmosphere. The mixture was stirred for 30 min at 0 °C, after which water (50 mL) was added slowly to the mixture. After acidification with dilute HCl solution, the product was extracted with Et<sub>2</sub>O. The extract was washed with saturated NaHCO<sub>3</sub> solution and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (50 g), and elution with hexane/EtOAc (4:1) to afford 3.13 g (24 mmol, 89%) of (S)-7 as an oil,  $[\alpha]_{D}^{23} = -2.6$  (c 1.4, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 3336 (s, O-H), 2925 (s), 1460 (s), 1379 (m), 1057 (s);  $\delta_{\rm H}$ : 0.89 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.8 Hz), 0.89 (3H, d, CH<sub>3</sub>CH, J = 6.6 Hz),  $\sim$ 1.3 (7H, m),  $\sim$ 1.6 (2H, m), 3.68 (2H, m, CH<sub>2</sub>OH); δ<sub>C</sub>: 14.2, 19.7, 23.0, 29.2, 29.5, 36.8, 40.0, 61.3; GC-MS:  $t_{\rm R}$  8.00 min, m/z 112 ([M-18]<sup>+</sup>, 2%), 84 (72%), 55 (100%). In the same manner,  $LiAlH_4$  reduction of (*R*)-6 (4.43 g, 28 mmol)

produced 3.26 g (25 mmol, 89%) of (*R*)-**7** as an oil,  $[\alpha]_D^{23} = +2.7$  (*c* 1.2, CHCl<sub>3</sub>). The two enantiomers were analyzed by chiral HPLC equipped with a Chiralpak AY-H column and a RI detector. The peaks of the two enantiomers were detected at different  $t_R$  values [(*S*)-**7**, 29.6 min and (*R*)-**7**, 28.4 min] by elution with hexane/2-propanol (99.6:0.4) at a flow rate (0.5 mL/min), and chromatograms showed their enantiomeric purity (>99% ee).

#### 4.7. 1-Iodo-3-methylheptanes (S)- and (R)-8

Iodine (7.11 g, 28 mmol) was added to a solution of triphenylphosphine (6.56 g, 25 mmol) and imidazole (3.74 g, 55 mmol) in dry THF (150 mL) at 0 °C under an argon atmosphere. After stirring for 15 min, (S)-7 (3.00 g, 23 mmol) dissolved in THF (5 mL) was added through a syringe. The mixture was stirred at 0 °C for 1 h, and then poured into saturated Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> solution and the product was extracted with EtOAc. The extract was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (50 g), and elution with hexane/EtOAc (9:1) to afford 5.04 g (21 mmol, 91%) of (S)-8 as an oil,  $[\alpha]_{D}^{23} = -12.6$  (c 1.4, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 2956 (s), 2925 (s), 2856 (s), 1464 (s), 1379 (m), 1178 (s), 602 (w); δ<sub>H</sub>: 0.87 (3H, d, CH<sub>3</sub>CH, J = 6.5 Hz), 0.89 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.5 Hz), 1.14 (1H, m) ~1.3 (5H, m), 1.54 (1H, m, CHCH<sub>2</sub>), 1.64 (1H, ddd, CHHCH<sub>2</sub>I, *J* = 21.3, 8.1, 5.8 Hz), 1.87 (1H, ddd, CHHCH<sub>2</sub>I, J = 21.3, 8.1, 5.8 Hz), 3.17 (1H, ddd, CHHI, J = 9.0, 8.0, 8.0 Hz), 3.25 (1H, ddd, CHHI, J = 9.0, 9.0, 5.8 Hz);  $\delta_{\rm C}$ : 5.3, 14.1, 18.8, 22.9, 29.0, 33.9, 35.9, 41.0; GC-MS:  $t_{\rm R}$ 9.72 min, m/z 240 (M<sup>+</sup>, 3%), 113 (23%), 57 (100%). In the same manner, iodization of (*R*)-**7** (3.26 g, 25 mmol) produced 5.52 g (23 mmol, 92%) of (*R*)-**8** as an oil,  $[\alpha]_D^{23} = +12.5$  (*c* 0.60, CHCl<sub>3</sub>).

# 4.8. 2-Decanols (S)- and (R)-9

In the same manner as that described for the preparation of (*S*)-**3**, (*S*)-**2** (3.48 g, 60 mmol) was reacted with n-C<sub>7</sub>H<sub>15</sub>MgBr, which was prepared from n-C<sub>7</sub>H<sub>15</sub>Br (16.1 g, 90 mmol), to produce 8.71 g (55 mmol, 92%) of (*S*)-**9** as an oil,  $[\alpha]_D^{23} = +8.1$  (*c* 1.1, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 3354 (s, O-H), 2925 (s), 2854 (s), 1458 (m), 1377 (m), 1115 (m);  $\delta_{H}$ : 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, *J* = 6.7 Hz), 1.19 (3H, d, CH<sub>3</sub>CHOH *J* = 6.2 Hz), ~1.3 (12H, m), ~1.4 (2H, m), 3.79 (1H, m, CH<sub>3</sub>CHOH);  $\delta_C$ : 14.1, 22.7, 23.5, 25.8, 29.3, 29.6, 29.7, 31.9, 39.4, 68.2; GC-MS: *t*<sub>R</sub> 9.90 min, *m/z* 140 (M-18, 3%), 112 (7%), 45 (100%). In the same manner, (*R*)-**2** (4.07 g, 70 mmol) was reacted with *n*-C<sub>7</sub>H<sub>15</sub>MgBr to yield 10.9 g (69 mmol, 99%) of (*R*)-**9** as an oil,  $[\alpha]_D^{23} = -8.0$  (*c* 0.61, CHCl<sub>3</sub>).

### 4.9. 2-Decanyl tosylates (S)- and (R)-10

In the same manner as that described for the preparation of (S)-4, (S)-9 (7.28 g, 46 mmol) was treated with tosyl chloride (13.2 g, 69 mmol) to produce 13.4 g (43 mmol, 93%) of (S)-10 as an oil,  $[\alpha]_{D}^{23} = +3.5$  (c 2.0, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 2927 (s), 2856 (s), 1599 (m, benzene ring), 1363 (s, S=O), 1176 (s, S=O);  $\delta_{H}$ : 0.87 (3H, t,  $CH_3CH_2$ , J = 6.9 Hz), ~1.2 (12H, m), 1.26 (3H, d,  $CH_3CHOTs$ , J = 6.2 Hz), 1.50 and 1.59 (2H, m), 2.44 (3H, s, CH<sub>3</sub>Ph), 4.59 (1H, tq, J = 6.3, 6.3 Hz, CH<sub>3</sub>CHOTs), 7.33 (2H, d, CH=CH, J = 8.2 Hz), 7.80 (2H, d, CH=CH, J = 8.2 Hz);  $\delta_C$ : 14.1, 20.9, 21.6, 22.6, 24.9, 29.07, 29.09 (×2), 31.7, 36.5, 80.7, 127.7 (×2), 129.7 (×2), 134.6, 144.4. In the same manner, (R)-9 (8.86 g, 56 mmol) was treated with tosyl chloride to yield 15.9 g (51 mmol, 91%) of (R)-10 as an oil,  $[\alpha]_D^{23}=-3.5$  (c 2.0, CHCl\_3). The two enantiomers were analyzed by chiral HPLC equipped with a Chiralpak AS-H column and a UV detector (254 nm). The peaks of the two enantiomers were detected at different  $t_{\rm R}$  values [(S)-10, 11.5 min and (R)-10, 14.5 min] and the chromatograms showed their enantiomeric purity (>99% ee).

#### 4.10. 2-Methyl-1-(4-methylphenylsulfonyl)decanes (*R*)- and (*S*)-12 from 10

In a two necked flask, methyl p-tolyl sulfone (2.04 g, 12 mmol) was dissolved in dry THF (180 mL). A hexane solution of BuLi (2.69 M, 4.7 mL, 12 mmol) and (S)-10 (2.50 g, 8.0 mmol) dissolved in THF (4 mL) were added successively into the flask through a syringe at 0 °C under an argon atmosphere. The reaction mixture was stirred at 55 °C for 4 h and then poured into an aqueous solution of NH₄Cl (100 mL). The crude product was extracted with EtOAc and the extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (40 g). Elution with hexane/EtOAc (14:1) afforded 1.79 g (5.8 mmol, 72%) of (*R*)-**12** as an oil,  $[\alpha]_D^{23} = -1.1$  (*c* 2.5, CHCl<sub>3</sub>);  $v_{max}$ (cm<sup>-1</sup>): 2925 (s), 2854 (s), 1599 (m, benzene ring), 1313 (s, S=O), 1147 (s, S=0), 1088 (s);  $\delta_{\rm H}$ : 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.9 Hz), 1.05  $(3H, d, CH_3CH, I = 6.2 \text{ Hz}), \sim 1.2 (12H, m), 2.04 (3H, m, CH_3CH),$ 2.45 (3H, s, CH<sub>3</sub>Ph), 2.90 (1H, dd, J = 14.2, 7.9 Hz, CHHTs), 3.06 (1H, dd, *J* = 14.2, 4.6 Hz, CHHTs), 7.35 (2H, d, CH=CH, *J* = 8.2 Hz), 7.79 (2H, d, CH=CH, I = 8.2 Hz);  $\delta_{C}$ : 14.1, 19.9, 21.6, 22.7, 26.3, 28.6, 29.2, 29.5 (×2), 31.8, 36.7, 62.7, 127.9 (×2), 129.9 (×2), 137.3, 144.4. In the same manner, (R)-10 (2.50 g, 8.0 mmol) was treated with the carbanion of methyl p-tolyl sulfone to yield 1.74 g (5.6 mmol, 70%) of (S)-**12** as an oil,  $[\alpha]_D^{23} = +1.1$  (c 2.5, CHCl<sub>3</sub>). The two enantiomers were analyzed by chiral HPLC (AS-H column) and detected at different  $t_R$  values [(R)-12, 53.4 min and (S)-12, 28.2 min] by elution with hexane/2-propanol (9:1). The chromatograms showed their enantiomeric purity (>99% ee).

#### 4.11. Methyl (R)-2-methylundecanoate (R)-15

In the same manner as that described for the preparation of (*S*)-**3**, methyl (*S*)-3-bromoisobutylate (*S*)-**14** (1.99 g, 11 mmol) was reacted with n-C<sub>7</sub>H<sub>15</sub>MgBr, which was prepared from n-C<sub>7</sub>H<sub>15</sub>Br (4.48 g, 25 mmol), to produce 2.00 g (10 mmol, 91%) of (*R*)-**15** as an oil,  $[\alpha]_D^{23} = -16.0$  (*c* 1.0, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 2925 (s), 2856 (s), 1739 (s, C=O), 1464 (s), 1196 (s), 1167 (s);  $\delta_{\rm H}$ : 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, *J* = 6.7 Hz), 1.14 (3H, d, CH<sub>3</sub>CH, *J* = 6.9 Hz), ~1.3 (12H, m), 1.64 (1H, m, CHHCH), 2.43 (1H, tq, *J* = 6.9, 6.9 Hz, CH<sub>3</sub>CH), 3.67 (3H, s, OCH<sub>3</sub>);  $\delta_{\rm C}$ : 14.1, 17.1, 22.7, 27.3, 29.3, 29.46, 29.54, 31.9, 33.8, 39.5, 51.5, 177.5.

# 4.12. (*R*)-2-Methyl-1-(4-methylphenylsulfonyl)decane (*R*)-12 from (*R*)-15

In the same manner as that described for the preparation of (S)-**7**, (*R*)-**15** (2.00 g, 10 mmol) was treated with  $\text{LiAlH}_4$  (0.38 g, 10 mmol) to produce 1.38 g (8.0 mmol, 80%) of (R)-2-methyl-1decanol as an oil;  $\delta_{\rm H}$ : 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.9 Hz), 0.91 (3H, d,  $CH_3CH$ , J = 6.7 Hz),  $\sim 1.3 (14H, m)$ , 1.61 (1H, m,  $CH_3CH$ ), 3.41 (1H, dd, CHHOH, J = 10.5, 6.6 Hz), 3.51 (1H, dd, CHHOH, J = 10.5, 5.7 Hz); δ<sub>C</sub>: 14.1, 16.6, 22.7, 27.0, 29.4, 29.6, 30.0, 31.9, 33.2, 35.8. Next, in the same manner as that described for the preparation of (S)-**8**, the alcohol was treated with a mixture of iodine (2.28 g, 9.0 mmol), triphenylphosphine (2.62 g, 10 mmol), and imidazole (1.36 g, 20 mmol) to produce 2.05 g (7.3 mmol, 91%) of (R)-1iodo-2-methyldecane as an oil;  $\delta_{\rm H}$ : 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.6 Hz), 0.97 (3H, d, CH<sub>3</sub>CH, J = 6.4 Hz), ~1.3 (14H, m), 1.45 (1H, m, CH<sub>3</sub>CH), 3.15 (1H, dd, CHHI, J = 9.5, 5.9 Hz), 3.23 (1H, dd, CHHI, J = 9.5, 4.6 Hz);  $\delta_C$ : 14.1, 18.1, 20.6, 22.7, 26.9, 29.3, 29.6, 29.7, 31.9, 34.7, 36.5. The iodo compound was added into a stirred mixture of sodium p-toluenesulfinate (1.78 g, 10 mmol), PEG-400 (10 mL), and DMSO (5 mL) at 80 °C. The reaction mixture was stirred for 2 h at 80 °C and then poured into a 10% NaCl solution (75 mL) after cooling to room temperature. The crude product was extracted with EtOAc and the extract was washed with water and brine, dried over  $Na_2SO_4$ , and concentrated in vacuo. The residue was chromatographed over  $SiO_2$  (50 g). Elution with hexane/ EtOAc (14:1) afforded 1.37 g (4.4 mmol, 60%) of (*R*)-**12** as an oil, which was analyzed by chiral HPLC.

# 4.13. 5,9-Dimethyl-8-(4-methylphenylsulfonyl)heptadecanes (5*s*,8*sR*,9*r*)-, (5*s*,8*sR*,9*s*)-, (5*r*,8*sR*,9*s*)-, and (5*r*,8*sR*,9*r*)-16

A hexane solution of BuLi (1.63 M, 3.7 mL, 6.0 mmol) was added through a syringe to a stirred solution of (R)-12 (1.55 g, 5.0 mmol) in THF (10 mL) at -78 °C under an argon atmosphere. The mixture was stirred at -78 °C for 10 min and then at -35 °C for 30 min. After cooling to  $-78 \circ C$ , (S)-8 (1.44 g, 6.0 mmol) dissolved in HMPA (3.5 mL) was added dropwise. The mixture was then allowed to warm gradually to room temperature and stirred for 8 h. The resulting mixture was poured into 1 M HCl (50 mL) and the crude products were extracted with EtOAc, washed with a saturated NaHCO<sub>3</sub> solution and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (50 g). Elution with hexane/EtOAc (19:1) afforded 1.82 g (4.3 mmol, 86%) of (5S,8SR,9R)-16 as an oil,  $v_{max}$  (cm<sup>-1</sup>): 2925 (s), 2854 (s), 1597 (m, benzene ring), 1464 (s), 1300 (s, S=0), 1146 (s, S=0), 1086 (s);  $\delta_{\rm H}$ : 0.77 and 0.78 (3H, d, CH<sub>3</sub>CH, J = 6.1 Hz), 0.87 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.3 Hz), 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 6.3 Hz), 1.00 and 1.03 (3H, d,  $CH_3CH$ , J = 7.1 Hz,  $\sim 1.2 (22H, m)$ , 1.67 (2H, m), 1.82 (1H, m), 2.19 (1H, m), 2.44 (3H, s, CH<sub>3</sub>Ph), 2.85 (1H, m, CHTs), 7.33 (2H, d, CH=CH, J = 8.0 Hz), 7.76 (2H, d, CH=CH, J = 8.0 Hz). In the same manner, the syntheses of (5S,8SR,9S)-16 from (S)-8 and (S)-12, (5R,8SR,9S)-16 from (R)-8 and (S)-12, and (5R,8SR,9R)-16 from (*R*)-8 and (*R*)-12 were accomplished in similar yields (80–85%).

# 4.14. 5,9-Dimethylheptadecanes (5*S*,9*R*)-, (5*S*,9*S*)-, (5*R*,9*S*)-, and (5*R*,9*R*)-1

To magnesium turnings (1.46 g, 60 mmol) stirred in dry THF (2.5 mL), a catalytic amount of MeMgBr (3.0 M, two drops) was added to activate the metal under an argon atmosphere. After stirring for 15 min at room temperature, a solution of (5S,8SR,9R)-16 (1.65 g, 3.9 mmol) in dry MeOH (60 mL) was added through a syringe, and the mixture was stirred at 50 °C for 4 h. Next, the mixture was evaporated to remove MeOH and an ice-cooled 1 M HCl solution (20 mL) was added to dissolve the white residues. The crude products were extracted with Et<sub>2</sub>O, washed with saturated NaH-CO<sub>3</sub> solution and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was chromatographed over SiO<sub>2</sub> (20 g). Elution with hexane afforded 0.84 g (3.1 mmol, 79%) of (5S,9R)-1 as an oil,  $[\alpha]_{D}^{23} = +1.1$  (c 0.63, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 2924 (s), 2854 (s), 1464 (m), 1377 (m), 723 (w);  $\delta_{\rm H}$ : 0.841 and 0.843 (6H, d, CH<sub>3</sub>CH, J = 6.7 Hz), 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 7.2 Hz), 0.89 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 7.2 Hz), 1.0-1.4 (28H, m);  $\delta_{C}$ : 14.14 and 14.19 (C-1, 17), 19.78 (×2, CHCH<sub>3</sub>), 22.73 (C-16), 23.09 (C-2), 24.50 (C-7), 27.12 (C-11), 29.38 and 29.41 (C-13, 14), 29.73 (C-12), 30.08 (C-3), 31.98 (C-15), 32.80 (×2, C-5, 9), 36.79 (C-10), 37.10 (C-4), 37.47 (×2, C-6, 8); GC-MS: t<sub>R</sub> 18.95 min, m/z 268 (M<sup>+</sup>, 1%), 211 (9%), 155 (15%), 140 (13%), 85 (83%), 57 (100%); HR-MS calcd for  $C_{19}H_{40}$ 268.3130; found 268.3126. In the same manner, the syntheses of (55,95)-1 from (55,85R,95)-16, (5R,95)-1 from (5R,85R,95)-16, and (5R,9R)-1 from (5R,8SR,9R)-16 were accomplished in similar yields (75–80%). (55,95)-**1**,  $[\alpha]_{D}^{23} = +2.2$  (*c* 1.3, CHCl<sub>3</sub>);  $v_{max}$  (cm<sup>-1</sup>): 2924 (s), 2854 (s), 1464 (m), 1377 (m), 723 (w);  $\delta_{\rm H}$ : 0.839 and 0.841 (6H, d, CH<sub>3</sub>CH, J = 6.3 Hz), 0.88 (3H, t, CH<sub>3</sub>CH<sub>2</sub>, J = 7.0 Hz), 0.89 (3H, t,  $CH_3CH_2$ , J = 7.0 Hz), 1.0–1.4 (28H, m);  $\delta_C$ : 14.14 and 14.19 (C-1, 17), 19.72 (×2, CHCH<sub>3</sub>), 22.73 (C-16), 23.09 (C-2), 24.50 (C-7), 27.13 (C-11), 29.39 and 29.41 (C-13, 14), 29.73 (C-12), 30.08 (C-3), 31.98 (C-15), 32.77 (×2, C-5, 9), 36.89 (C-10), 37.19 (C-4),

37.41 (×2, C-6, 8) GC-MS:  $t_R$  18.93 min, m/z 268 (M<sup>+</sup>, 1%), 211 (8%), 155 (13%), 140 (13%), 85 (83%), 57 (100%); HR-MS calcd for  $C_{19}H_{40}$  268.3130; found 268.3125. (5*R*,9S)-1,  $[\alpha]_D^{23} = -1.1$  (*c* 1.2, CHCl<sub>3</sub>); HR-MS calcd for  $C_{19}H_{40}$  268.3130; found 268.3125. (5*R*,9*R*)-1,  $[\alpha]_D^{23} = -2.1$  (*c* 1.3, CHCl<sub>3</sub>); HR-MS calcd for  $C_{19}H_{40}$  268.3130; found 268.3126.

#### Acknowledgments

This work was supported in part by grant-aid for scientific research from the Ministry of Education, Culture, Sports, Science, and Technology (MEXT) of Japan [Grant-in-Aid for Scientific Research (C) No. 24580158].

#### References

- (a) Ando, T. Internet database 2011. http://www.tuat.ac.jp/~antetsu/ LepiPheroList.htm.; (b) El-Sayed, A. M. Internet database 2011. http:// www.pherobase.com/.
- 2. Ando, T.; Inomata, S.; Yamamoto, M. Top. Curr. Chem. 2004, 239, 51–96.
- (a) Adachi, Y.; Do, N. D.; Kinjo, M.; Makisako, S.; Yamakawa, R.; Mori, K.; Ando, T. J. Chem. Ecol. 2010, 36, 814–823; (b) Yamakawa, R.; Kiyota, R.; Taguri, T.; Ando, T. Tetrahedron Lett. 2011, 52, 5808–5811.

- Francke, W.; Franke, S.; Toth, M.; Szöcs, G.; Guerin, P.; Arn, H. Naturwissenschaften 1987, 74, 143–144.
- (a) Díaz, D. D.; Martín, V. S. J. Org. Chem. 2000, 65. 7896-7801; (b) Moreira, J. A.; Corrêa, A. G. Tetrahedron: Asymmetry 2003, 14, 3787-3795; (c) Chow, S.; Koenig, W. A.; Kitching, W. Eur. J. Org. Chem. 2004, 1198-1201; (d) Zarbin, P. H. G.; Princival, J. L.; de Lima, E. R.; dos Santos, A. A.; Ambrogio, B. G.; de Oliveira, A. R. M. Tetrahedron Lett. 2004, 45, 239-241; (e) van Summeren, R. P.; Reijmer, S. J. W.; Feringa, B. L.; Minnaard, A. J. Chem. Commun. 2005, 1387-1389.
- (a) Mori, K.; Wu, J. Liebigs Ann. Chem. 1991, 439–443; (b) Mori, K.; Horikiri, H. Liebigs Ann. 1996, 501–505; (c) Tamagawa, H.; Takikawa, H.; Mori, K. Eur. J. Org. Chem. 1999, 973–978; (d) Kuwahara, S.; Liang, T.; Leal, W. S.; Ishikawa, J.; Kodama, O. Biosci., Biotechnol., Biochem. 2000, 64, 2723–2726; (e) Mori, K. Tetrahedron: Asymmetry 2008, 19, 857–861.
- (a) Cason, J.; Coad, R. A. J. Am. Chem. Soc. 1950, 72, 4695–4697; (b) Sato, T.; Otera, J. J. Org. Chem. 1995, 60, 2627–2629.
- 8. Sorger, K.; Stohrer, J. US 2007/0225519 A1.
- Krapcho, A. P.; Weimaster, J. F.; Eldridge, J. M.; Jahngen, E. G. E., Jr.; Lovey, A. J.; Stephens, W. P. J. Org. Chem. **1978**, 43, 138–147.
- (a) Tsunoda, T.; Nagaku, M.; Nagino, C.; Kawamura, Y.; Ozaki, F.; Hioki, H.; Ito, S. *Tetrahedron Lett.* **1995**, *36*, 2531–2534; (b) Tsunoda, T.; Ozaki, F.; Shirakata, N.; Tamaoka, Y.; Yamamoto, H.; Ito, S. *Tetrahedron Lett.* **1996**, *37*, 2463–2466.
- 11. Lebel, N. A.; Balasubramanian, N. J. Am. Chem. Soc. 1989, 111, 3363-3368.
- 12. Cahiez, G.; Chaboche, C.; Jézéquel, M. Tetrahedron 2000, 56, 2733-2737.
- 13. Brown, A. C.; Carpino, L. A. J. Org. Chem. 1985, 50, 1749–1750.
- (a) Mori, K.; Tashiro, T. Curr. Org. Synth. 2004, 1, 11–29; (b) Mori, K. Bioorg. Med. Chem. 2007, 15, 7505–7523.